



Programme: Tuesday 19th June
Room Vermeulen

Materials processing session Chairman: Tetsuo Uchikoshi			
9h20-9h45	Invited	Kenjiro Fujimoto	Materials innovation using combinatorial processing and high-throughput measurement system
9h45-10h05		Jef Bergmans	Recycling of spent Cu-based oxygen carriers into high-strength ceramic
10h05-10h25		Juha-Pekka Nikkanen	Utilization of CO ₂ in modification of galvanized steel surface
Sustainability and catalysis session Chairman : Oliver Kappe			
10h50-11h15	Invited	Pascal Van Der Voort	Hybrid Materials for Sustainability
11h15-11h35		Fransesc Sastre-Calabuig	CO ₂ photoassisted methanation on Ru supported Al ₂ O ₃ under solar light irradiation
11h35-11h55		Carlos Mendoza	Rose Bengal@MSNs as heterogeneous photocatalysts in photooxygenation reactions
Environment-related materials Chairman: Yoshikazu Kameshima			
14h25-14h50	Invited	Kazuya Nakata	Biological applications of photocatalysis
14h50-15h10		Daniela Meroni	Structural and ran properties of N-doped TiO ₂ /SnO ₂ photocatalysts for air pollutant remediation
15h10-15h30		Giuseppe Granata	Removal Mechanism of Se(IV) and Cr(VI) from Aqueous Solution by Sulfate Green Rust
Biomaterials session Chairman: Hirotaka Maeda			
16h40-17h05	Invited	Kimihiko Yamashita	Introduction of hydroxyapatite electret for novel biointerface engineering
17h05—17h25		Sander Leeuwenburgh	Reinforcement of bioceramics with polymeric particles to obtain self-healing and tough polymer-ceramic composite biomaterials

Structural and electronic properties of N-doped TiO₂/SnO₂ photocatalysts for air pollutant remediation

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Introduction

TiO₂/SnO₂ heterojunctions bear relevance to numerous research fields, such as gas sensors, fuel cells and photocatalysis [1]. Actually, the isomorphism of SnO₂ cassiterite and TiO₂ rutile (both P4₂/mmn) makes them suitable for developing stable junctions. Moreover, the relative positions of the two semiconductors' Fermi energies favour electron transfer from the TiO₂ conduction band to the SnO₂ one, promoting charge carrier separation and increasing the e⁻-h⁺ life-time. In turn, this improves the device efficiency in photo-electrochemical applications with respect to N-doped TiO₂, one of the most investigated visible-light active photocatalysts, which suffers from enhanced recombination of photo-generated charges [2].

Results and Discussion

N-doped TiO₂-SnO₂ photocatalysts were prepared in a broad range of Sn:Ti ratios (0-20%) by three different synthetic approaches (mechanical mixing, seeded growth and co-synthesis), followed by calcination at 400°C. Samples were characterized from the structural, morphological and electronic point of view, also via *in situ* synchrotron radiation-based XAS studies. Structural properties are remarkably modified by the adopted synthetic procedure (Fig.1a). Mechanical mixing leads to mixed phases in almost stoichiometric ratios. The seeded growth gives rise to partial segregation of SnO₂, while for co-synthesis, the only appreciable component is rutile TiO₂, despite the low calcination temperature. Such a tailoring of the structural features reflects in tuneable optical properties (Fig.1b), which show synergistic effects when both N and Sn diffuse in bulk TiO₂. The samples' photocatalytic activity was tested toward the gas phase degradation of VOCs, under both UV and visible light irradiation, monitoring the pollutant disappearance, the main intermediates' formation and the mineralization degree.

Conclusions

By tuning the synthetic procedure, mixed phases or Sn doping within the TiO₂ matrix could be obtained, leading to different optical and electronic properties. High-resolution XRPD and XAS analyses shed light on (i) the actual location and chemical nature of Sn species, (ii) the defectivity of the TiO₂ lattice, and (iii) the interplay between the dopants. Structural and spectroscopic results were correlated with the photocatalytic activity in the visible region.

References

- [1] G. Kelp et al., *Nanoscale*, **2016**, 8, 7056; H. An et al., *Electrochim. Acta*, **2013**, 92, 176.
 [2] Asahi et al. *Chem Rev*, **2014**, 114, 9824; Rimoldi et al. *J Phys Chem C*, **2015**, 119, 24410.

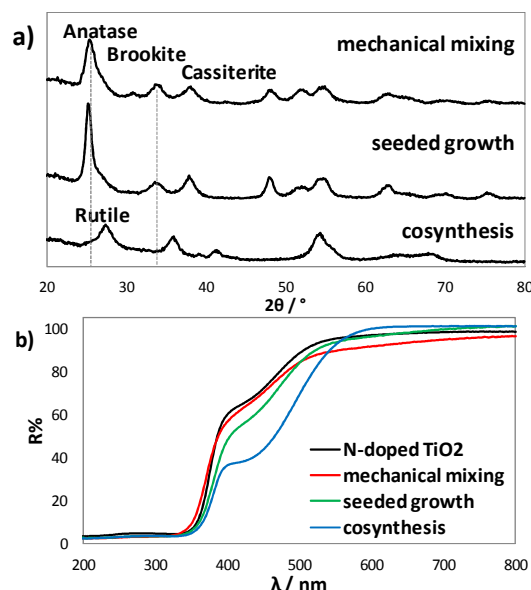


Figure 1: a) XRD and b) DRS of TiNSn20 samples from the three synthetic procedures.