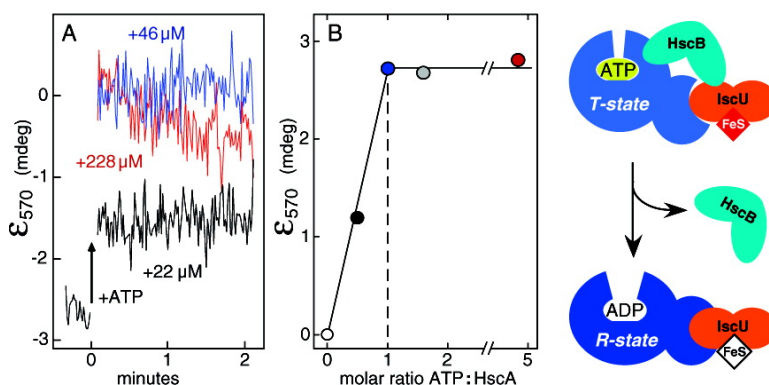


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# Studies on the Mechanism of Catalysis of Iron–Sulfur Cluster Transfer from IscU[2Fe2S] by HscA/HscB Chaperones<sup>†</sup>

Francesco Bonomi,<sup>\*,‡</sup> Stefania Iametti,<sup>‡</sup> Anna Morleo,<sup>‡</sup> Dennis Ta,<sup>§</sup> and Larry E. Vickery<sup>\*,§</sup>

Section of Biochemistry, DISMA, University of Milan, Celoria 2, 20133 Milan, Italy, and Department of Physiology and Biophysics, University of California, Irvine, California 92617

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**ABSTRACT:** The HscA/HscB chaperone/cochaperone system accelerates transfer of iron–sulfur clusters from the FeS-scaffold protein IscU (IscU<sub>2</sub>[2Fe<sub>2</sub>S], holo-IscU) to acceptor proteins in an ATP-dependent manner. We have employed visible region circular dichroism (CD) measurements to monitor chaperone-catalyzed cluster transfer from holo-IscU to apoferredoxin and to investigate chaperone-induced changes in properties of the IscU<sub>2</sub>[2Fe<sub>2</sub>S] cluster. HscA-mediated acceleration of [2Fe<sub>2</sub>S] cluster transfer exhibited an absolute requirement for both HscB and ATP. A mutant form of HscA lacking ATPase activity, HscA(T212V), was unable to accelerate cluster transfer, suggesting that ATP hydrolysis and conformational changes accompanying the ATP (T-state) to ADP (R-state) transition in the HscA chaperone are required for catalysis. Addition of HscA and HscB to IscU<sub>2</sub>[2Fe<sub>2</sub>S] did not affect the properties of the [2Fe<sub>2</sub>S] cluster, but subsequent addition of ATP was found to cause a transient change of the visible region CD spectrum, indicating distortion of the IscU-bound cluster. The dependence of the rate of decay of the observed CD change on ATP concentration and the lack of an effect of the HscA(T212V) mutant were consistent with conformational changes in the cluster coupled to ATP hydrolysis by HscA. Experiments carried out under conditions with limiting concentrations of HscA, HscB, and ATP further showed that formation of a 1:1:1 HscA–HscB–IscU<sub>2</sub>[2Fe<sub>2</sub>S] complex and a single ATP hydrolysis step are sufficient to elicit the full effect of the chaperones on the [2Fe<sub>2</sub>S] cluster. These results suggest that acceleration of iron–sulfur cluster transfer involves a structural change in the IscU<sub>2</sub>[2Fe<sub>2</sub>S] complex during the T → R transition of HscA accompanying ATP hydrolysis.

The biosynthesis of iron–sulfur proteins is a multistep process involving a number of specialized proteins that mediate iron–sulfur cluster formation and delivery to acceptor proteins (reviewed in refs 1–5). The most widely distributed and conserved system involves a scaffold protein employed for preassembly of FeS complexes and a Hsp70 class chaperone and J-protein cochaperone system that interacts with the scaffold protein (reviewed in ref 6). Specialized forms of Hsp70 and their cochaperones have evolved in bacteria (HscA and HscB) and in certain fungi (Ssq1 and Jac1), whereas most eukaryotes employ a multifunctional mitochondrial Hsp70 (mtHsp70) together with a specialized cochaperone homologous to HscB/Jac1. HscA and Ssq1 have been shown to specifically bind to a conserved sequence present in the FeS-scaffold protein designated IscU in bacteria (7, 8) and Isu in fungi (9, 10), and the crystal structure of a complex of an IscU-derived peptide with the substrate binding domain of HscA has been determined (11). The interaction of IscU/Isu with HscA/Ssq1 is regulated by the cochaperone HscB/Jac1 that binds the scaffold protein

to assist its delivery to the chaperone. The cochaperone further stabilizes the chaperone–scaffold complex by enhancing chaperone ATPase activity to convert the low-affinity chaperone–ATP complex to the higher affinity chaperone–ADP complex (12).

The exact functions chaperone–scaffold interactions play in cluster assembly/transfer events remain to be established. *In vivo* and *in vitro* studies of yeast Ssq1 and Jac1 indicated that the chaperones are not required for FeS cluster assembly on Isu, suggesting that the interactions instead play a role in cluster maturation and/or transfer to acceptor proteins (13, 14). *In vitro* studies using purified forms of bacterial HscA, HscB, and IscU have provided evidence that the chaperones act to facilitate cluster transfer. Bonomi et al. (15) found that *Escherichia coli* HscA/HscB enhanced the rate of cluster transfer from IscU<sub>2</sub>[2Fe<sub>2</sub>S] to apoferredoxin in an ATP-dependent manner; moreover, stimulation occurred at low chaperone/scaffold ratios, suggesting that the chaperones act catalytically. Chandramouli and Johnson (16) carried out studies with the HscA/HscB/IscU system from *Azotobacter vinelandii* and found that stoichiometric concentrations of the chaperones together with excess ATP stimulated cluster transfer to apoferredoxin. Analysis of the HscA/HscB/IscU chaperone cycle (17) led Chandramouli and Johnson (16) to suggest that enhancement could result from interaction of acceptor proteins with the HscA(ATP)–HscB–IscU<sub>2</sub>[2Fe<sub>2</sub>S] complex or alternatively with the HscA(ADP)–IscU(2Fe<sub>2</sub>S)

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\* To whom correspondence should be addressed. F.B.: e-mail, francesco.bonomi@unimi.it; tel, 39-02-50316819. L.E.V.: e-mail, LVickery@uci.edu; tel, 949-824-6580; fax, 949-824-8540.

<sup>‡</sup> University of Milan.

<sup>§</sup> University of California, Irvine.

complex formed following ATP hydrolysis. Because the conformational states of HscA with ATP bound (T-state) or ADP bound (R-state) differ, cluster release could be facilitated by structural alterations of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  imposed by interactions with the HscA(ATP) T-state complex or by alterations coupled to conformational changes in HscA occurring during the  $T \Rightarrow R$  transition following ATP hydrolysis. The rapid rate of ATP hydrolysis by the HscA(ATP)–HscB– $\text{IscU}_2[2\text{Fe}_2\text{S}]$  complex makes it difficult to distinguish between these two possible mechanisms.

In the studies described herein we have used a mutant form of HscA lacking ATPase activity to investigate the role of ATP hydrolysis in facilitating cluster transfer. The findings suggest that structural changes in the  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  complex that occur during the  $T \Rightarrow R$  conformational change of HscA accompanying ATP hydrolysis are coupled to  $[2\text{Fe}_2\text{S}]$  cluster activation and subsequent transfer.

## EXPERIMENTAL PROCEDURES

**Proteins.** Recombinant forms of *E. coli* IscU (12), ferredoxin (18), HscA (20), and HscB (20) were prepared according to previously published procedures, and protein concentrations were determined as described therein. HscA(T212V) was prepared by site-specific mutagenesis (QuikChange; Stratagene) using the pTrcHsc66 plasmid (20), and mutations were confirmed by DNA sequencing (Laragen Inc.). The HscA(T212V) protein was expressed and purified as for wild-type HscA (20). Concentrations of the holo forms of IscU and ferredoxin are reported in terms of the  $[2\text{Fe}_2\text{S}]$  complex of each protein. Under the conditions employed IscU behaves as a dimer, and concentrations are therefore reported in terms of the  $(\text{IscU})_2[2\text{Fe}_2\text{S}]$  complex (12, 21); ferredoxin is monomeric and contains a single  $[2\text{Fe}_2\text{S}]$  cluster (18).

**Sample Handling and Cluster Reconstitution.** Sample manipulations, including reagent preparation and chromatography, were carried out anaerobically under argon in septum-capped vials, cuvettes, or columns, and stainless steel needles, tubing, and cannulae were used for sample transfers as described previously (21). The buffer used was 0.1 M Tris-HCl, pH 8.0, containing 5 mM dithiothreitol (TD buffer).<sup>1</sup> Where indicated, in experiments involving HscA and HscB, 10 mM  $\text{MgCl}_2$  and 150 mM KCl were added to TD buffer (referred to as TDMK buffer). Unless otherwise specified, the temperature was maintained at 23 °C.

$\text{IscU}_2[2\text{Fe}_2\text{S}]$  was prepared by three sequential additions at 5 min intervals of 0.32 equiv of ferric ammonium citrate and lithium sulfide to 0.5 mM apo-IscU in TD buffer on ice. The final solution thus contained slightly substoichiometric concentrations of iron and sulfide required for formation of holo-IscU. Formation of the  $[2\text{Fe}_2\text{S}]$  cluster was monitored by visible region CD spectroscopy. At concentrations in the 0.5 mM range reconstituted  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  was found to be stable for at least 18 h under anaerobic conditions in TD buffer at 0 °C.

Native ferredoxin was isolated as the  $[2\text{Fe}_2\text{S}]$  complex (18). Apoferreredoxin was prepared by precipitating the

holoprotein in 10% trichloroacetic acid containing 10 mM DTT for 10 min at 0 °C. The apoferreredoxin pellet was collected by centrifugation, washed twice in cold water under anaerobic conditions, and dissolved anaerobically in TD buffer.

**Cluster Transfer Studies.** Unless otherwise specified, reactions were initiated by addition of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to a solution containing a 1.05–1.10 equiv of apoferreredoxin in TDMK buffer at 23 °C. Where indicated, HscA and/or HscB were (was) added to the buffer prior to other proteins. ATP was added from a 100 mM stock solution in TDMK buffer immediately after the addition of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ . Mixtures were prepared directly in septum-capped 1 mL anaerobic cuvettes. Initial cluster transfer rates were calculated from the increase in ellipticity at 435 nm during the first 30–60 s.

**Chaperone-Induced Spectral Modifications.** Unless otherwise specified, experiments were performed in TDMK buffer at 23 °C in septum-capped 1 mL anaerobic cuvettes. Spectral changes resulting from addition of HscA, HscB, and/or ATP to  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  were monitored by CD by recording spectra at specified time intervals or by continuous monitoring of ellipticity at 570 nm. Typically,  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  was mixed with a variable amount of HscA and HscB, and the reaction was initiated by the addition of ATP. In some experiments ATP was added to mixtures of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  and HscA, and the reaction was initiated shortly thereafter by the addition of HscB.

**Analytical Methods.** ATPase assays (12), analyses for iron and sulfide (21), isothermal titration calorimetry measurements (22), and difference spectroscopy (22) were carried out as described previously. Circular dichroism measurements were recorded at 23 °C in 1 cm path length anaerobic cuvettes using a Jasco J-810 spectropolarimeter and analyzed using Jasco software.

## RESULTS

**Circular Dichroism Spectra of Holo-IscU and Holoferredoxin.** In the experiments described below we used *E. coli* apoferreredoxin as an acceptor protein for FeS clusters provided by  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  and employed CD spectroscopy to monitor cluster transfer. Figure 1 shows the visible region CD spectra of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ , native holoferredoxin, and holoferredoxin reconstituted by cluster transfer from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  in the presence of HscA, HscB, and ATP. The spectra of native and reconstituted holoferredoxin are virtually identical, suggesting that the structure of the  $[2\text{Fe}_2\text{S}]$  cluster in the reconstituted protein is the same as that of the native protein. This finding is similar to the earlier results involving FeS cluster transfer using  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  in the absence of chaperones (21) and indicates that under these conditions HscA and HscB do not alter the type of cluster transferred. In subsequent experiments cluster transfer from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to apoferreredoxin was followed by monitoring ellipticity changes at 435 nm.

**FeS Cluster Transfer Studies.** The rates of cluster transfer from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to apoferreredoxin were determined under conditions including or lacking each chaperone protein and/or ATP (Figure 2). Acceleration of cluster transfer was observed only when HscA, HscB, and ATP were all present, and omission of any component gave rates comparable to

<sup>1</sup> Abbreviations: CD, circular dichroism; TD buffer, 0.1 M Tris-HCl, pH 8.0, and 5 mM dithiothreitol; TDMK buffer, 0.1 M Tris-HCl, pH 8.0, 5 mM dithiothreitol, 10 mM  $\text{MgCl}_2$ , and 150 mM KCl.

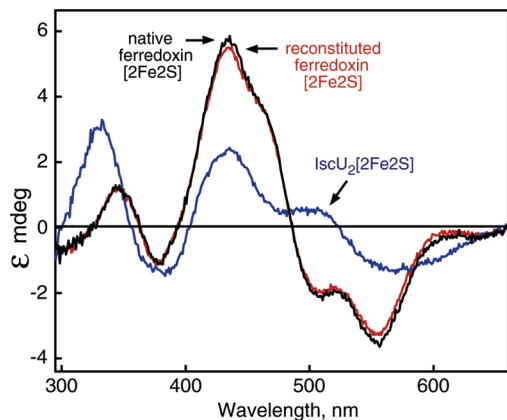


FIGURE 1: Circular dichroism spectra of holo-IscU and holoferredoxin.  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ ,  $50 \mu\text{M}$ , was prepared by reconstitution of the apoprotein. Native ferredoxin[ $2\text{Fe}_2\text{S}$ ],  $48 \mu\text{M}$ , was purified directly from cultures of *E. coli*. Reconstituted ferredoxin[ $2\text{Fe}_2\text{S}$ ] was prepared by incubating  $48 \mu\text{M}$  apoferredoxin,  $38 \mu\text{M}$  HscA,  $38 \mu\text{M}$  HscB, and  $50 \mu\text{M}$   $\text{IscU}_2[2\text{Fe}_2\text{S}]$  with 2 mM ATP for 4 h in TDMK buffer at  $23^\circ\text{C}$ .

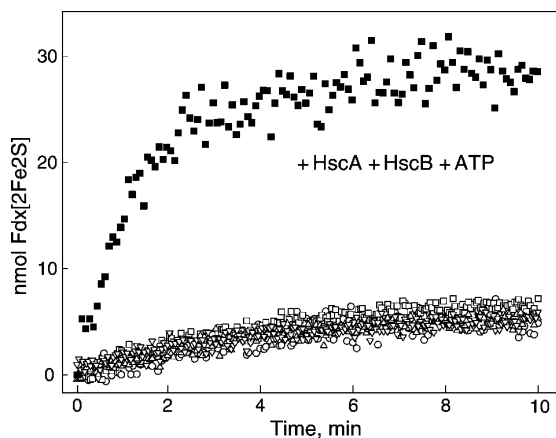


FIGURE 2: Rates of  $[2\text{Fe}_2\text{S}]$  cluster transfer from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to apoferredoxin. Cluster transfer was assessed by monitoring CD ellipticity increases at 435 nm following the addition of  $50 \mu\text{M}$   $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to  $48 \mu\text{M}$  apoferredoxin. Other additions: none ( $\circ$ ); HscA, HscB, and ATP ( $\blacksquare$ ); HscA and HscB ( $\triangle$ ); HscA and ATP ( $\nabla$ ); ATP ( $\square$ ). When present, HscA and HscB were  $149 \mu\text{M}$  each; ATP was added at a final concentration of 2 mM immediately before the addition of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ .

the background rate of transfer observed using  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  alone. The finding that no inhibition was observed in the presence of HscA with or without HscB in the absence of ATP indicates that simple reversible binding of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to the chaperone(s) does not affect the rate of cluster transfer. Although these findings establish that ATP is required for chaperone enhancement of cluster transfer, they do not allow determination of whether ATP hydrolysis is required or whether ATP binding to the HscA–HscB– $\text{IscU}_2[2\text{Fe}_2\text{S}]$  complex is sufficient.

The effect of chaperone concentration on the initial rate of FeS cluster transfer was initially studied using an excess of ATP. Figure 3A shows that the rate of cluster transfer increases linearly as the amount of HscA and equivalent HscB are increased over the range to where there is one HscA and HscB per  $\text{IscU}$  monomer equivalent. The rates observed correspond to  $\approx 0.1$  mol of holoferredoxin formed ( $\text{mol of HscA}^{-1} \text{ min}^{-1}$ ), an acceleration of approximately 10-fold over the background rate. At each concentration of chaperone tested CD spectra recorded at equilibrium (4–18 h) showed

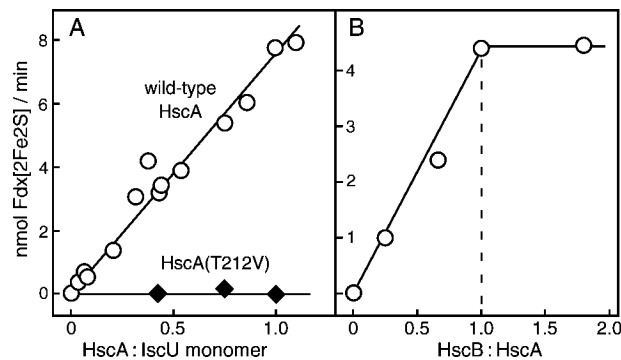


FIGURE 3: Dependence of initial cluster transfer rate from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to apoferredoxin on HscA and HscB ratios. Rates were calculated from the increase in CD ellipticity at 435 nm, and values were corrected for background cluster transfer observed in the absence of HscA and HscB. Conditions are similar to those given in Figure 2 except that the concentrations of HscA and HscB were varied. ATP was added at a final concentration of 2 mM immediately before the addition of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ . Left panel: The concentration of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  was held constant, and the concentration of HscA ( $\circ$ ) or HscA(T212V) ( $\blacklozenge$ ) and equivalent concentrations of HscB were varied. Results are plotted in terms of the concentration of the  $\text{IscU}$  monomer. Right panel: The concentrations of HscA and  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  were held constant, and the concentrations of HscB was varied.

that cluster transfer was complete and quantitative, indicating that the chaperones affect the rate but not the yield of transfer.

Figure 3B shows that the cochaperone HscB is essential for activity. Previous studies showed that HscB and apo-IscU form a ternary complex with HscA(ATP) (17), and the finding that acceleration of FeS cluster transfer increases linearly with the amount of HscB until a concentration equivalent to HscA is reached is consistent with a requirement for formation of a HscA(ATP)–HscB– $\text{IscU}_2[2\text{Fe}_2\text{S}]$  complex for activity. However, it is not known whether the HscB requirement arises from a direct effect on the structure of the ternary T-state complex prior to ATP hydrolysis or reflects synergistic stimulation of HscA ATPase activity by the cochaperone.

**Requirement for ATP Hydrolysis.** To determine whether formation of the HscA(ATP)–HscB– $\text{IscU}_2[2\text{Fe}_2\text{S}]$  T-state complex was sufficient to facilitate cluster transfer or whether ATP hydrolysis and the resulting T  $\Rightarrow$  R conformational transition are required, we employed a mutant of HscA lacking ATPase activity. Earlier studies with the *E. coli* Hsp70 chaperone DnaK (22, 23) showed that mutation of a conserved threonine residue in the nucleotide binding site reduced ATPase activity more than 90%. Thr-212 occupies a similar position in the nucleotide binding site of HscA,<sup>2</sup> and we replaced Thr-212 with valine using site-specific mutagenesis. The HscA(T212V) mutant did not exhibit measurable ATPase activity ( $<0.01$  mol of ATP hydrolyzed  $\text{min}^{-1}$ ), and activity could not be detected in the presence of HscB or  $\text{IscU}$  or mixtures of HscB and  $\text{IscU}$  that stimulate the ATPase activity of wild-type HscA  $>400$ -fold. ATP was able to bind to HscA(T212V) as addition of ATP to the mutant elicited a near-UV difference spectrum indicative of T-state formation as observed with wild-type HscA (24) (see Supporting Information Figure S1). However, in contrast to wild-type HscA the ATP-induced difference spectrum re-

<sup>2</sup> J. R. Cupp-Vickery, P. C. Aoto, D. T. Ta, and L. E. Vickery, unpublished results.



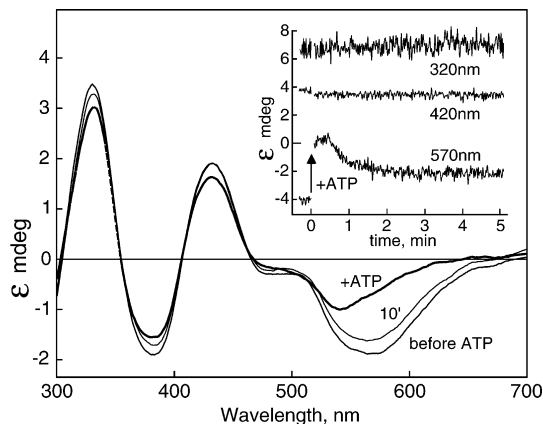


FIGURE 4: Transient effects of HscA, HscB, and ATP on the CD spectrum of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ . Samples contained  $54 \mu\text{M}$   $\text{IscU}_2[2\text{Fe}_2\text{S}]$ ,  $113 \mu\text{M}$  HscA, and  $125 \mu\text{M}$  HscB in TDMK buffer at  $23^\circ\text{C}$ . Spectra were recorded before and immediately or 10 min after the addition of 2 mM ATP. The scan rate was  $100 \text{ nm/min}$ . Inset: Time course of ellipticity changes at selected wavelengths following addition of ATP.

remained stable for  $>30$  min, consistent with loss of ATPase activity. Whereas ATP hydrolysis was defective,  $\text{IscU}$  binding to HscA(T212V) appeared normal. Isothermal titration calorimetry experiments yielded a  $K_d$  of  $1.5 \mu\text{M}$  for  $\text{IscU}$  ( $n = 1.0$ ) in the presence of 1 mM ADP compared to  $1.6 \mu\text{M}$  observed for wild-type HscA (25) (see Supporting Information Figure S2).

Studies on the ability of the HscA(T212V) mutant to accelerate FeS cluster transfer from  $\text{IscU}$  to apoferreredoxin were carried out in a manner similar to the experiments with wild-type HscA. As shown in Figure 3A (◆) HscA(T212V) had no significant effect on the rate of holoferreredoxin formation. Cluster transfer was neither enhanced nor inhibited at HscA(T212V) concentrations up to 1 equiv per  $\text{IscU}$  monomer. These results strongly suggest that formation of the HscA(ATP)–HscB– $\text{IscU}_2[2\text{Fe}_2\text{S}]$  T-state complex is not sufficient to facilitate cluster transfer and that ATP hydrolysis and subsequent conformational changes in HscA are required.

**Chaperone Effects on  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ .** We used CD spectroscopy to investigate whether the interaction of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  with HscA and HscB might have effects on the properties of the  $[2\text{Fe}_2\text{S}]$  cluster. Addition of HscA or HscB alone or each individually in combination with ATP had no detectable effect on the visible region CD of the  $[2\text{Fe}_2\text{S}]$  cluster (data not shown). However, addition of HscA and HscB together with ATP caused a transient change in the CD spectrum in the region from 520 to 620 nm. Figure 4 shows spectra recorded prior to, immediately after, and 10 min after addition of ATP. The spectral scan recorded immediately following ATP addition displays significant loss of the CD trough observed near 570 nm, whereas the scan initiated 10 min after ATP addition shows only a small loss of signal. The time dependence of the change observed suggested that the effects on the cluster are transient, and full effects at shorter wavelengths may not have been observed due to the time required to complete the scan. For this reason, the kinetics of CD changes were followed at wavelengths of 320, 420, and 570 nm corresponding to extrema in the CD spectrum. As shown in the inset to Figure 4, the largest change was associated with the trough near 570 nm. The CD signal was reduced to near zero shortly

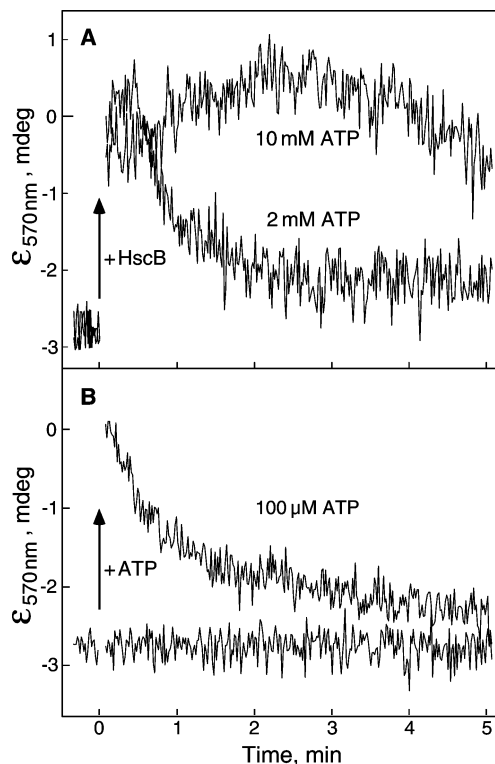


FIGURE 5: Effect of ATP concentration on induced CD changes of  $(\text{IscU})_2[2\text{Fe}_2\text{S}]$ . Panel A: Sample mixtures contained  $54 \mu\text{M}$   $\text{IscU}_2[2\text{Fe}_2\text{S}]$  and  $113 \mu\text{M}$  HscA, and 2 or 10 mM ATP in TDMK buffer at  $20^\circ\text{C}$ . Reactions were initiated by addition of  $125 \mu\text{M}$  HscB, and ellipticity changes were monitored at 570 nm. Panel B: The sample in the upper trace contained  $54 \mu\text{M}$   $\text{IscU}_2[2\text{Fe}_2\text{S}]$  and  $122 \mu\text{M}$  HscA and HscB in TDMK buffer at  $23^\circ\text{C}$ , and the reaction was initiated by addition of  $100 \mu\text{M}$  ATP. The sample in the lower trace lacked HscA and HscB.

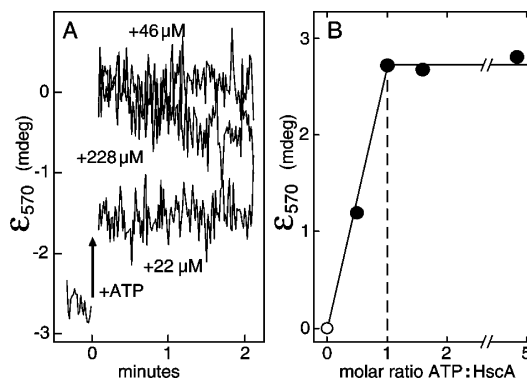


FIGURE 6: Effect of stoichiometric concentrations of HscA, HscB, and ATP on induced CD changes of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$ . Samples contained  $54 \mu\text{M}$   $\text{IscU}_2[2\text{Fe}_2\text{S}]$  and  $48 \mu\text{M}$  HscA and HscB in TDMK buffer at  $20^\circ\text{C}$ . Panel A: Time course of ATP-induced changes. ATP was added to give final concentrations of 22, 46, or  $222 \mu\text{M}$  corresponding to 0.46, 0.96, or 4.6 equiv of HscA, respectively. Panel B: Effects of the ratio of ATP to HscA. The maximum amplitude of  $\Delta\epsilon_{570}$  observed in panel A is plotted as a function of the molar ratio of ATP to HscA. The shaded circle at a molar ratio of 1.6 corresponds to the ellipticity change observed using  $62 \mu\text{M}$  HscA and HscB plus  $100 \mu\text{M}$  ATP in Figure 5B.

following addition of ATP and reappeared slowly over the next several minutes.

To test whether the CD change observed required ATP hydrolysis, we carried out similar CD experiments with the HscA(T212V) mutant. No time-dependent spectral changes were observed using HscA(T212V), indicating that hydroly-

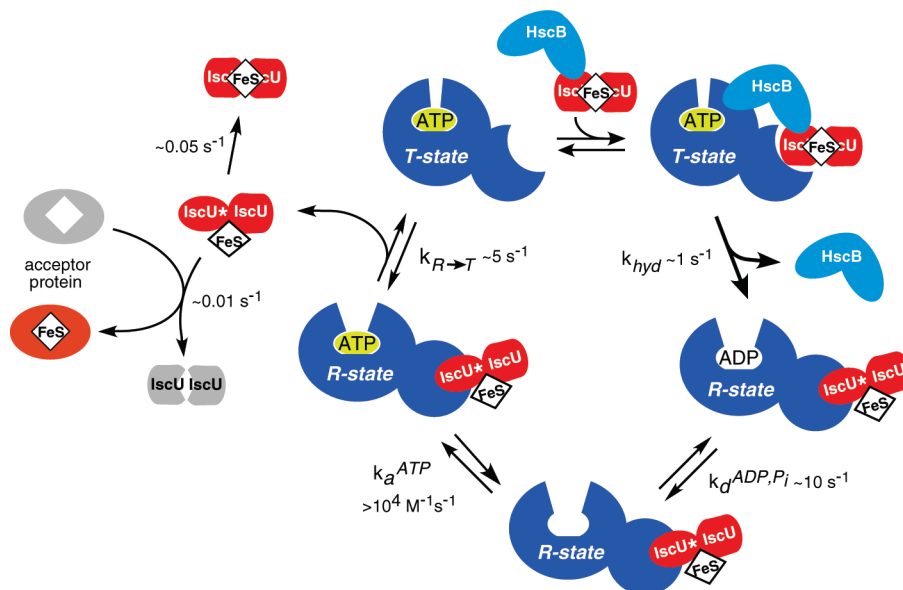


FIGURE 7: Proposed mechanistic scheme for chaperone-catalyzed FeS cluster transfer from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to apoacceptor proteins. The HscA ATPase reaction cycle scheme and rates shown are derived from ref 17. See text for discussion.

sis of ATP is necessary and the formation of an  $\text{HscA}(\text{ATP})\text{-HscB-IscU}_2[2\text{Fe}_2\text{S}]$  complex is not sufficient to elicit the effect (data not shown). The finding that ATP hydrolysis is required suggested that the time-dependent recovery of the CD signal may result from ATP depletion in the system. HscB and IscU synergistically stimulate HscA ATPase activity  $>400$  fold to give rates of  $\sim 1 \text{ s}^{-1}$ , and under the high chaperone concentrations employed 2 mM ATP would be rapidly hydrolyzed. To test this possibility, we compared the rate of recovery of the 570 nm CD signal in the presence of 2 and 10 mM ATP; in this experiment reactions were initiated by addition of HscB. As shown in Figure 5A at the higher ATP level the effect on the CD signal is longer lived, consistent with recovery of the signal corresponding to ATP depletion. To test whether lower concentrations of ATP were sufficient to affect the change, we carried out the measurements under limited turnover conditions. The ATP concentration was reduced to 100  $\mu\text{M}$ , greater than the concentration of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  (54  $\mu\text{M}$ ) but slightly less than the concentration of HscA (122  $\mu\text{M}$ ). As shown in Figure 5B the 570 nm CD signal was completely lost even under these conditions of limited ATP.

The finding that the full effect of the chaperones on the  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  CD signal could be obtained using quasi-single turnover conditions suggested that a single ATPase cycle involving one  $\text{HscA}(\text{ATP})\text{-HscB-IscU}_2[2\text{Fe}_2\text{S}]$  complex might elicit the effect. We tested this possibility by incubating  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  (54  $\mu\text{M}$ ) with a slightly substoichiometric amount of HscA and HscB (48  $\mu\text{M}$ , 0.9 equiv) and 4.6, 0.96, or 0.46 equiv of ATP per HscA. Figure 6 shows that a single equivalent of HscA, HscB, and ATP is sufficient to fully affect the ellipticity change observed at 570 nm. Decreasing the amount of ATP below 1 equiv to 0.46 reduces the effect correspondingly. These results indicate that a single ATPase cycle involving 1 mol of HscA and HscB/mol of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  is sufficient to cause the changes observed in the CD spectrum of the FeS cluster.

## DISCUSSION

The results described herein corroborate earlier findings of Bonomi et al. (15) and Chandramouli and Johnson (16) regarding ATP-dependent chaperone acceleration of  $[2\text{Fe}_2\text{S}]$  cluster transfer from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to apoferreredoxin. The maximal rate observed,  $\cong 0.1 \text{ mol of holoferreredoxin formed (mol of HscA)}^{-1} \text{ min}^{-1}$ , corresponds to a roughly 10-fold enhancement of rates observed in the absence of HscA, HscB, and/or ATP. Furthermore, acceleration observed using less than stoichiometric amounts of HscA and HscB yielded complete formation of holoferreredoxin, consistent with a catalytic role for the chaperones in cluster transfer. The requirement for HscB and ATP suggests that a ternary  $\text{HscA}(\text{ATP})\text{-HscB-IscU}_2[2\text{Fe}_2\text{S}]$  complex plays a critical role in activation of  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  for cluster transfer. As pointed out by Chandramouli and Johnson (16) activation could occur as a result of formation of the ternary complex itself or could occur during subsequent conformational changes coupled to ATP hydrolysis (cf. Figure 7). By employing a mutant form of HscA lacking ATPase activity, we found that ATP hydrolysis is necessary for the acceleration, suggesting that structural changes during the  $\text{T} \rightarrow \text{R}$  conformational transition of HscA are coupled to the activation process.

The mechanism whereby the chaperones enhance  $[2\text{Fe}_2\text{S}]$  cluster transfer is not known. Previous cluster transfer studies conducted in the absence of chaperone indicated that the cluster remains intact during transfer and that transfer likely involves direct protein interactions between  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  and the apoprotein acceptor. Studies by Cowan and co-workers (26) using human Isu and apoferreredoxin showed that free iron in solution does not exchange with the cluster during transfer, indicating that there is not a significant degree of cluster disassembly and reassembly, and kinetic studies by Bonomi et al. (21) on cluster transfer from  $\text{IscU}_2[2\text{Fe}_2\text{S}]$  to apoferreredoxin provided evidence for formation of a donor-acceptor complex during transfer. Analysis of the kinetics of cluster formation and transfer further suggested that a labile FeS-scaffold species formed during initial cluster

assembly on IscU was more efficiently transferred than that of the mature IscU<sub>2</sub>[2Fe2S] complex (21). A similar labile species produced from the mature cluster during the chaperone ATPase cycle might also facilitate transfer, and the transient change observed in the CD spectrum of IscU<sub>2</sub>[2Fe2S] upon incubation with HscA, HscB, and ATP may reflect formation of such an activated complex. The requirements for both HscB and ATP and the lack of activity of the HscA(T212V) mutant in eliciting the CD change are consistent with the observation that acceleration of cluster transfer occurs during the R  $\Rightarrow$  T transition of HscA coincident with ATP hydrolysis. In the scheme presented in Figure 7 we have designated the form of IscU that gives rise to the altered [2Fe2S] cluster CD spectrum as IscU\*. This altered complex, or a species derived from it, may be the form of the scaffold cluster that exhibits an enhanced rate of transfer.

The molecular basis of the effect of HscA and HscB on IscU<sub>2</sub>[2Fe2S] is not understood. Because the chaperones interact with the IscU protein, the effects on the [2Fe2S] cluster are assumed to be indirect, i.e., mediated by structural changes induced in the scaffold protein rather than by direct interaction of the chaperones with the cluster. The chaperones could act by causing changes in IscU conformation that facilitate cluster release or capture by the acceptor protein. HscA binds to a conserved <sup>99</sup>LPPVK<sup>103</sup> sequence (7, 8, 11) positioned close to conserved Cys-106 that may be involved in cluster coordination. Binding of IscU<sub>2</sub>[2Fe2S] to HscA alone does not appear to significantly affect the properties of the cluster or make it more available for transfer, but conformational changes coupled to the HscA T  $\Rightarrow$  R transition during ATP hydrolysis may act to make the cluster more available. The CD changes observed using limiting concentrations of ATP indicate that IscU\* formation requires only a single turnover, and hence IscU\* will be formed during each HscA ATPase cycle. The CD results obtained using limiting concentrations of HscA further indicate that only one HscA is required to affect IscU<sub>2</sub>[2Fe2S], and the scheme therefore shows only one IscU molecule of the dimer as IscU\*. It is possible, however, that effects on one IscU monomer could have effects on the other monomer of the dimer complex as well.

Whether the cochaperone HscB plays a direct role in cluster activation is not clear from the present studies. HscB enhances the binding of IscU to HscA and synergistically stimulates the rate of HscA ATP hydrolysis, but the extent of the interactions between HscB and IscU in the ternary HscA(ATP)–HscB–IscU<sub>2</sub>[2Fe2S] complex and whether these contribute to any changes in IscU structure during the HscA T  $\Rightarrow$  R transition associated with ATP hydrolysis have not been characterized. Recent studies of HscB–IscU interactions have revealed that mutations in the IscU binding region of HscB decrease allosteric cooperativity within the HscA–HscB–IscU complex (29). This finding suggests that HscB–IscU interactions are coupled to the T  $\Rightarrow$  R conformational changes of HscA and these interactions could play an essential role in IscU\* formation.

The finding that HscA and HscB accelerate IscU<sub>2</sub>[2Fe2S] cluster transfer raises the question of whether transfer occurs while the IscU\* is bound to HscA or subsequent to its release. Some insights into this question can be gleaned by analysis of the kinetics of the HscA ATPase cycle, the rate of activated cluster relaxation, and the rate of cluster transfer. The rate-determining step of the HscA ATPase cycle is the hydrolysis of ATP, and  $k_{\text{cat}}$  for this step is  $\sim 1 \text{ s}^{-1}$  (17). The rate of IscU activation, i.e., IscU  $\Rightarrow$  IscU\* (Figure 7), is presumed to occur at a similar rate consistent with the rapid initial CD changes observed in Figures 5 and 6. ATP hydrolysis, nucleotide exchange, and conversion of HscA to the low-affinity T-state state occur rapidly, and this favors the release of IscU\*. Under typical physiological conditions HscA will exist predominantly as the HscA(ATP) T-state complex or as the HscA(ATP)–HscB–IscU<sub>2</sub>[2Fe2S] complex.<sup>3</sup> For this reason it seems most likely that cluster transfer will occur from free IscU\* rather than from IscU\* bound to the chaperone. The rate of cluster relaxation observed in the CD changes of Figures 5 and 6,  $\sim 0.1 \text{ s}^{-1}$ , is also faster than the rate of cluster transfer, and this suggests that multiple ATPase cycles to form IscU\* are likely to occur prior to transfer.

## SUPPORTING INFORMATION AVAILABLE

Figures showing (1) the binding of IscU to HscA(T212V) monitored by isothermal titration calorimetry and (2) the effect of ATP on the absorption spectrum of HscA(T212V). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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<sup>3</sup> Because the rate of cluster relaxation, i.e., IscU\*  $\Rightarrow$  IscU, is significantly slower than ATP hydrolysis, it is possible that IscU\* may rebind to HscA, but the affinities of HscA and HscB for IscU\* are not known.

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