

# **Biotechnologies for marine oil spill cleanup: indissoluble ties with microorganisms**

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1 **Keywords**

2 Oil bioremediation; Slow release particles; Oil spill snorkel; Hydrostatic pressure; Deep  
3 sea

4  
5 **Abstract**

6 The ubiquitous exploitation of petroleum hydrocarbons (HCs) has been accompanied by  
7 accidental spills and chronic pollution in marine ecosystems, including the deep ocean.  
8 Physico-chemical technologies are available for oil spill cleanup, but HCs must  
9 ultimately be mineralized by microorganisms. How environmental factors drive the  
10 assembly and activity of hydrocarbon-degrading microbial communities remains  
11 unknown, limiting our capacity to integrate microorganism-based cleanup strategies with  
12 current physico-chemical remediation technologies. Here, we summarize recent findings  
13 about microbial physiology, metabolism and ecology and describe how microbes can be  
14 exploited to create improved biotechnological solutions to cleanup marine surface and  
15 deep waters, sediments and beaches.

16

17

## 18 **Oil Spills in the Oceans and the Role Played by Microorganisms**

19 The first large marine oil spill occurred in 1907 with the sinking of the *Thomas W.*  
20 *Lawson*, which released 7,400 tons of paraffin oil off the coast of the United Kingdom.  
21 Since then, estimates indicate that more than seven million tons of oil have been  
22 released into the environment from over 140 large spills [1], with the Deepwater Horizon  
23 (DWH) disaster releasing more than 700,000 tons of crude oil in the Gulf of Mexico [2].  
24 While accidents like DWH are widely publicized, more than 90% of oil pollution comes  
25 from non-accidental natural and anthropogenic sources [3], including run-off from land-  
26 based sources and routine ship operations, such as deballasting and tank washing [4].

27

28 Oil spills can have extensive environmental and economic impact. For instance,  
29 estimates suggest that more than 250,000 seabirds were killed during the Exxon-Valdez  
30 oil spill in 1989 and that the DWH disaster cost more than \$61 billion [1]. Crude oil that  
31 is released into the environment undergoes chemical, physical and biological  
32 modifications known as weathering. The fate of weathered crude oil hydrocarbons  
33 (HCs) depends on microorganisms that have developed strategies to increase their  
34 bioavailability and degradation pathways that allow them to use HCs as carbon and  
35 energy sources. Hazen et al. [5] observed that the half-lives of n-alkanes that leaked  
36 into the deep sea during the DWH oil spill were shorter than expected (1-6 days). Even  
37 so, slicks of oil heavily enriched with polycyclic aromatic hydrocarbons (PAHs) reached  
38 shores and were detected for at least a year [6], requiring cleanup interventions.

39

## 40 **Current Response Techniques to Oil Spills in Marine Environments**

41 Current emergency responses include mechanical containment and recovery of spilled  
42 oil, addition of chemical dispersants, and physical cleanup of shorelines. When an oil  
43 spill occurs in a marine environment, the magnitude of the spill is assessed to determine  
44 the most suitable response action. Figure 1 highlights the different cleanup approaches  
45 currently used for waters, shorelines and sediments. The primary combat strategy is  
46 mechanically containing and recovering oil by using different types of booms, barriers  
47 and skimmers, as well as the application of natural or synthetic sorbent materials.  
48 Containment booms are typically used in surface waters as barriers to control the oil  
49 from spreading and impacting shorelines or other marine resources like aquaculture  
50 facilities. With the aid of different types of booms (fences, curtains or inflatable booms),  
51 oil is concentrated into thicker layers, facilitating subsequent removal with skimmers.  
52 Different types of skimmers (weir, oleophilic and suction skimmers) have been  
53 developed that are suitable for specific oil types and in the presence of ice or debris.  
54 Sorbents are either natural or synthetic-like polymeric materials used for very small  
55 spills to adsorb the spilled oil either on the surface or internally through swelling. They  
56 are also used to remove final traces of oil after collection with skimmers, or in areas that  
57 cannot be reached by skimmers.

58

59 Dispersants (Box 1) are mixtures of surfactants and solvents that reduce spilled oil into  
60 small droplets (smaller than 100  $\mu\text{m}$  in size) by the action of waves and currents; they  
61 are mainly used in open and deep waters [7,8] (Figure 1). Oil droplets are spread from  
62 the seawater surface into the **water column**, thus preventing the oil slick from reaching  
63 the shoreline. Oil droplets have a higher surface-to-volume ratio than oil slicks have,

64 which allows more of the oil's surface to be exposed to microorganisms. Dispersants  
65 are effective when applied immediately following a spill, before the lightest HC  
66 components volatilize, and their performance depends on water salinity, temperature  
67 and wave action. Considering the toxicity of the dispersants' components [7,8], including  
68 to microorganisms [9], the next technological goal in oil spill remediation is replacing  
69 chemical surfactants with non-toxic **biosurfactants** (see Glossary).

70

71 A series of physical methods have been developed for different types of oil  
72 contamination. When oil slicks reach sandy beaches or rocky coastlines, they are  
73 manually collected or sprayed with water to return them back to the water, where they  
74 are adsorbed by sorbent materials (Figure 1). Another technique for oil removal is *in situ*  
75 burning (ISB), the controlled burning of oil at the spill's location to reduce the amount of  
76 oil on the surface. ISB, which should be coupled to toxicity assessments of burned  
77 residues [10], has been used on several occasions including the DWH accident, where  
78 the amount of burned oil was higher than the amount of mechanically collected oil  
79 despite estimate uncertainties [11]. In deep-sea spills, capping the head of the leaking  
80 oil well is the first action to be taken and is complemented by the other approaches  
81 (Figure 1).

82

### 83 **Bioremediation and natural attenuation**

84 Accurate knowledge of the metabolic potential of microorganisms and of the  
85 environmental factors that shape their interactions, viability and degradation activity is  
86 required to optimize **bioremediation** intervention strategies. HC-degrading marine

87 microorganisms can be divided into specialists and generalists according to their ability  
88 to grow respectively on a narrow range of HCs or on a wider set of carbon sources [12].  
89 They include strains of *Bacteria*, *Archaea* and *Fungi* (Table 1, Table S1).  
90 Hydrocarbonoclastic-bacteria are the most studied in terms of their physiology, ecology  
91 and biotechnology because of their ubiquity and prevalence.

92  
93 **Natural attenuation** is determined by the complex metabolic networks built up by  
94 microorganisms and their interactions with other organisms such as algae [12–14]. In  
95 seawater, one of the main factors limiting the rate of HC degradation is low HC solubility  
96 [13]. By producing biosurfactants and modifying cell membrane hydrophobicity,  
97 microorganisms increase and modulate HC bioavailability according to environmental  
98 conditions, such as in the case of *Alcanivorax* spp. [15]. Oxygen availability determines  
99 the pathway of HC breakdown. In the water column, under aerobic conditions, microbial  
100 enzymes activate HC molecules by incorporating oxygen atoms, generating  
101 corresponding alcohols, which are further oxidized to carboxylic acids that are degraded  
102 through  $\beta$ -oxidation [16]. The most studied aerobic enzymes are alkane hydroxylases,  
103 encoded by the *alkB*, *p450* and *almA* genes [16,17]. Under anaerobic conditions typical  
104 in sediments, HC catabolism is activated by the addition of fumarate to the secondary  
105 carbon and catalysed by alkyl-succinate synthases [16].

106  
107 The entire complexity of the metabolic pathways involved in HC catabolism under *in situ*  
108 conditions is not yet understood, especially in relation to the environmental conditions,  
109 although the increasing availability of **meta-omics** data is clarifying the picture [18–22].

110 Marine microbial communities responding to HC inputs are subjected to ecological  
111 successions generally starting with aliphatic HC-degraders (*e.g.*, *Alcanivorax* spp.) and  
112 the subsequent enrichment of other microorganisms able to metabolize aromatic HCs  
113 (*e.g.*, *Cycloclasticus* spp.) as observed in seawater [23], sediments [24] and beaches  
114 [21]. The response of the autochthonous microbial community after an oil spill depends  
115 upon the community's initial functional diversity. This response is also modulated by the  
116 local environmental conditions.

117 A metagenomic study of chronically polluted sediments across a transect from the  
118 Mediterranean Sea to the Red Sea reconstructed the occurring metabolic pathways,  
119 demonstrating that biodegradation capabilities expand according to the increasing  
120 yearly average sediment temperature across the transect despite a decrease in  
121 bacterial richness [18]. Although some common and specialized HC-degrading bacteria  
122 were not detected by 16S rRNA gene sequencing, the metabolic networks accounting  
123 for both aerobic and anaerobic HC degradation pathways indicated that the chronically  
124 polluted marine sediments harbour a higher catabolic diversification than do freshly  
125 polluted samples [18]. This suggests that microbial communities exposed to chronic  
126 pollution possess a wide repertoire of catabolic abilities that allow them to respond more  
127 promptly to oil spills [18]. HC cleanup may occur through transient enrichment of  
128 specialized microorganisms that degrade the different HC classes; their taxa succession  
129 patterns have been described [23]. The combination of degradation capabilities and  
130 pathways of different microbial community members is required for PAH degradation as  
131 indicated by genomes assembled from metagenomic datasets following stable isotope  
132 probing of the surface and deep plume waters of the DWH spill [14]. This suggests that

133 key degradative pathways are distributed across the different members of the DWH  
134 community that cooperatively and co-ordinately react to initiate PAHs degradation [14].

135

### 136 **Biotechnologies for water and coastal pollution**

137 Bioremediation technologies compatible with natural biogeochemical cycles comprise  
138 biosurfactant amendments, **biostimulation** and **bioaugmentation**. Biosurfactants  
139 favour oil solubilisation and oil droplet formation in water (Figure 2), making HCs  
140 available to non-biosurfactant-producing microorganisms. Bacteria, yeast and fungi  
141 mainly produce anionic or neutral biosurfactants. Renowned producers include  
142 *Acinetobacter*, *Bacillus* and *Pseudomonas*, with *Alcanivorax* predominating in oil-  
143 contaminated marine surface waters worldwide, owing to its capacity to produce large  
144 amounts of glycolipid biosurfactants [25]. Compared to dispersants, biosurfactants have  
145 low to no toxicity, high biodegradability, surface and emulsification activities and high  
146 stability under extreme temperature, pH and salinity conditions [26,27]. Biosurfactants  
147 positively impact HC biodegradation [28,29] and have been applied to oil recovery in  
148 reservoirs, to oil transportation in pipelines and to production of emulsified fuels [30].

149

150 Biostimulation is among the most effective approaches to enhance the degrading  
151 activity of indigenous bacterial communities since it improves the C/N/P ratio that is  
152 unbalanced after oil spills [27,31–33]. Nutrient delivery represents the main limitation of  
153 biostimulation due to rapid leaching along the water column [34]. Nutrient **micro-**  
154 **encapsulation** within slow-release particles (SRPs) eventually combined with  
155 biosurfactants and oil-degrading microorganisms may improve biostimulation efficiency

156 (Figure 2). SRPs are produced with low-toxicity polymers, *e.g.*, polyurethane-polyurea  
157 copolymers, alginates and chitosans [35,36]. A recent strategy based on microinjecting,  
158 cryo-crosslinking and coating with ethyl cellulose films allowed the synthesis of alginate-  
159 beads with increased load capacity and slow release of N/P fertilizers [36]. Even though  
160 the addition of nutrients can promote the growth of heterotrophs, thus creating  
161 competition with HC-degraders [37], increased total petroleum HC removal was  
162 reported by using alginate-encapsulated diesel-degrading bacteria as compared to free-  
163 cells [38]. Recent metagenomic data indicated that biostimulation with different nitrogen  
164 sources favoured different HC catabolic pathways without substantially affecting the  
165 taxonomic structure of the bacterial community [39].

166

167 Indigenous microbial populations may not have the full metabolic ability to process the  
168 complex mixture of spilled HCs or specialized HC-degraders might be underrepresented  
169 in the existing community. Bioaugmentation with site-allochthonous microorganisms  
170 might enhance HC degradation [40] although the long-term presence of such non-  
171 indigenous microorganisms has been questioned [41]. Autochthonous bioaugmentation  
172 (ABA) uses indigenous microorganisms enriched from specific habitats of the  
173 contaminated site (*e.g.*, surface or deep-sea water) (Figure 2). This approach has been  
174 successfully implemented in a study aimed at simulate an oil spill event, showing that  
175 the use of pre-adapted HC degrading bacteria in combination to biostimulation provided  
176 the best results in term of HC removal [42]. A further alternative is bioaugmentation with  
177 **mobile genetic elements** [43], which aims at horizontal transfer of remediation genes

178 from an exogenous inoculant to indigenous microorganisms by using catabolic plasmids  
179 or transposons.

180

### 181 **Biotechnological approaches to remediate oil-contaminated marine sediments**

182 Marine sediments represent an important sink for petroleum HCs after accidental spills.  
183 A number of different chemical, physical, and microbiological processes contribute to  
184 the transport of petroleum from a positively buoyant state in the water column down to  
185 the seafloor, such as weathering, adsorption onto settling particulate matter (including  
186 so-called marine oil snow [44], Figure 2), and the addition of chemical dispersants. In  
187 the case of the DWH spill, 1.8-14% of the oil reached the seafloor as estimated using  
188 hopanes as a biomarker tracer [45] or 0.5-9% as estimated using radiocarbon  
189 distributions [46]. Upon sedimentation, oil penetrates the upper sediment layers  
190 (typically 1-30 cm, depending on site conditions). In the sediments it may persist due to  
191 the prevailing anoxic conditions (Figure 3A) that drastically limit the occurrence of  
192 oxidative biodegradation processes, and the low bioavailability resulting from the strong  
193 sorption onto hydrophobic sedimentary materials.

194

195 *In situ* bioremediation is typically regarded as one of the most effective and sustainable  
196 strategies to cleanup contaminated sediments. Different approaches have been  
197 proposed to stimulate naturally occurring microbial communities that degrade petroleum  
198 HCs in marine sediments [47,48]. These typically involve the subsurface addition of  
199 degradation rate-limiting nutrients, electron acceptors and (bio)surfactants. The  
200 successful stimulation of the indigenous microbial community dwelling in oil polluted

201 sediment was achieved by adding nitrogen and phosphorous, ultimately resulting in a  
202 higher HC removal compared to the non-biostimulated control [31]. Similarly, the activity  
203 of indigenous oil-degraders can be boosted by the addition of biosurfactants [42]. The  
204 interplay between the bioavailability of electron acceptors (*e.g.*, oxygen, nitrate, sulfate,  
205  $\text{Fe}^{3+}/\text{Mn}^{4+}$ ) and HCs is probably the most critical factor affecting the efficacy of sediment  
206 bioremediation systems. Under aerobic conditions, petroleum HC biodegradation occurs  
207 rapidly and therefore different engineered approaches have been proposed to deliver  
208 oxygen to the sediments. Among them, a modular slurry system, which performs *in situ*  
209 aeration of the contaminated sediments, while minimizing the risk of spreading the  
210 contamination away from the treatment zone, has recently been developed (Figure 3B)  
211 [49]. Although the system was highly effective in stimulating the metabolism of aerobic  
212 HC-degrading bacteria and in reducing sediment toxicity, its application turned out to be  
213 highly labour- and energy-intensive. Other methods include supplying oxygen-releasing  
214 compounds (*e.g.*, calcium peroxide-based chemicals) to the contaminated sediment.  
215 However, rapid abiotic oxygen consumption by reactions with reduced chemical species  
216 (*e.g.*,  $\text{Fe}^{2+}$ ,  $\text{S}^{2-}$ ) and the difficulties in controlling the rate of oxygen release over time are  
217 some limitations of these approaches.

218

219 Recently, an innovative bioelectrochemical system termed an “oil-spill snorkel” was  
220 proposed to accelerate oil HC biodegradation in marine sediments [50]. The system  
221 consists of a conductive graphite rod (the “snorkel”) positioned for electrochemically  
222 connecting two spatially segregated redox zones: the anoxic contaminated sediment  
223 and the oxic ( $\text{O}_2$ -containing) overlying water (Figure 3C). The portion of the snorkel

224 positioned in the anoxic sediment serves as an electron acceptor (*i.e.*, an anode),  
225 sinking electrons deriving directly from the microbially catalyzed anaerobic oxidation of  
226 HCs and from the chemical and/or biochemical oxidation of reduced species (*i.e.*,  $S^{2-}$ ,  
227  $Fe^{2+}$ ) occurring in the bulk of the sediment. Upon transfer to the buried portion of the  
228 snorkel, the electrons move to the upper portion (*i.e.*, the cathode), driven by the  
229 existing redox gradient, where they combine with oxygen and protons to form water as a  
230 by-product. Besides serving as a virtually inexhaustible respiratory electron acceptor in  
231 the anaerobic oxidation of petroleum HCs, the snorkel was suggested to indirectly  
232 stimulate HC biodegradation by sulfate-reducing bacteria via the scavenging of toxic  
233 sulfide diffusing from the bulk of the sediment [51]. This finding has major practical  
234 implications as it suggests that the radius of influence of the oil-spill snorkel may extend  
235 far from where the rod is positioned. Although the feasibility of the oil-spill snorkel has  
236 been demonstrated only at the laboratory scale, the technology has a very low energetic  
237 and environmental footprint (*e.g.*, no energy input or maintenance required) and could  
238 be ideally applied for long-term remediation of contaminated sediments in remote open  
239 sea areas.

240

#### 241 **Oil spills in the deep sea**

242 The DWH spill (Gulf of Mexico, April 2010) was the first oil spill originating in the deep  
243 sea (1544 m below the surface level, bsl). Hydrostatic pressure (HP) and temperature  
244 gradients, alongside the large injection of dispersants, fractionated petroleum HCs  
245 along the water column, with the slow buoyant migration of light or dispersed HCs  
246 forming multiple plumes at 800-1300 m bsl (about 8-13 MPa and 5°C) [52,53]. Bacteria

247 detected in the oil plume were subjected to **microbial succession**, with HC  
248 composition and quantity proposed to account for such a change [5,54,55]. While the  
249 main physiological drivers for the microbial community shift remain uncertain [56], there  
250 is a consensus about the initial enrichment of *Oceanospirillales* and pseudomonads  
251 (May 2010) followed by a shift in dominance to *Colwellia*, *Cycloclasticus*,  
252 *Pseudoalteromonas* and methylotrophs (until August 2010). Ammonium and dissolved  
253 inorganic nitrogen (DIN) were unrelated to plume samples [5,54]. Genes related to the  
254 transport of iron or nitrogen- and phosphorous-based compounds [57] or their  
255 metabolism (*e.g.*, nitrate and nitrite reduction, [58]) were highly transcribed, but their  
256 correlation with plume samples or microbial succession is uncertain [5,20,54,57,58].  
257 The gene expression for sulphur cycling and sulphite reduction increased, with sulphate  
258 reduction the postulated activity, despite O<sub>2</sub> levels that remained relatively high [5,58].

259

260 Microbial succession was more difficult to assess in the deep-sea sediments because of  
261 the patchy oil distribution. Marine snow pulses resulting in oil deposition after June 2010  
262 fuelled O<sub>2</sub> respiration on the surface of the seafloor (enriching *Roseobacter*,  
263 *Verrucomicrobiaceae* and *Bacteroidetes* until October 2010). Anoxic microniches were  
264 favoured and *Deltaproteobacteria* and other anaerobes developed [59]. DIN, ammonium  
265 and total petroleum HCs were correlated in surface sediments and denitrification-related  
266 genes were highly transcribed [60]. *Deltaproteobacteria* were also enriched in  
267 subseafloor samples [61]. One year after the DWH spill, surface sediments around the  
268 wellhead area were enriched with *Actinobacteria*, *Firmicutes*, *Chloroflexi* and  
269 methylotrophic bacteria [62]. However, a time-dependent survey suggested that the

270 microbial community of the sediments had returned to pre-spill conditions, with a  
271 response to petroleum only individuated at a finer taxonomic level (as with the obligate  
272 polycyclic aromatic HC-degrader *Cycloclasticus* [59]). Continued sedimentation  
273 imposed anaerobic conditions on the remaining HCs, with total PAHs in 3-cm deep  
274 sediments slowly approaching pre-spill levels [63].

275

276 Understanding cooperation, competition and succession of microorganisms after an oil  
277 spill in the deep sea is key to predicting the success of the oil cleanup. Considering the  
278 difficulty of *in situ* studies at extreme depths, testing physiological response of isolates  
279 to the specific environmental conditions, such as temperature or HP, is a necessary  
280 complementary effort to explain changes in bacterial community composition and  
281 activity. However, the impact of HP on the HC degradation physiology after DWH was  
282 neglected and recent studies show that HP can affect the metabolism of oil-degrading  
283 bacteria. It has been recently proposed that the impaired metabolic response of  
284 *Alcanivorax* spp., which are ubiquitous and typically the first microorganisms identified  
285 after surface oil spills, to HP explains the lack of their detection in the DWH deep oil  
286 plume [64–66]. The DWH deep-sea plume [67] and sediment [61] revealed a low  
287 *Alcanivorax* abundance, which was unrelated to hydrocarbon concentrations [61],  
288 contrary to the abundance of other *Oceanospirillales* [5]. Consequently, the contribution  
289 of *Alcanivorax* to oil degradation in the plume was considered negligible [61].  
290 Meanwhile, results from oiled beach sands [68], oil mounds collected on surface  
291 waters [62] and plume samples [14,67] showed a significant *Alcanivorax* abundance  
292 after cultivation under atmospheric pressure. *Alcanivorax* isolates were also obtained in

293 enrichment cultures from HC-free, deep-sea water samples collected from depths up to  
294 5000 m below sea level [69].

295 We call these observations the '*Alcanivorax paradox*', i.e., the lack of response of  
296 ubiquitous HC-degrading bacteria to HCs in the deep sea. All of the isolation studies on  
297 the deep sea neglected to consider HP as a contributing variable despite it being a  
298 unique feature of deep-water environments. Recent data showed that a HP of 10 MPa  
299 (comparable to that in the DWH plume) inhibits growth of three *Alcanivorax* species (*A.*  
300 *borkumensis*, *A. dieselolei* and *A. jadensis*) on dodecane as a sole carbon source and  
301 that even a HP of 5 MPa causes a significant reduction in cell replication [64–66]. Under  
302 HP transcription, multimeric protein complexes such as ATPase and ribosomes are  
303 increased as compared to those under surface-water-resembling conditions. The  
304 respiratory chain shifts from cytochrome oxidases to reductases, alongside an  
305 enhanced expression of genes of the Na<sup>+</sup>-translocating NADH reductase complex  
306 (RNF-NQR). Translation, energy generation and electron transport are typical targets of  
307 microbial piezoadaptation (Box 2). Under HP, the *A. borkumensis* SK2 strain  
308 synthesizes the osmolyte ectoine [65,66], which has been suggested to be a pressure-  
309 responsive compound (i.e., piezolyte). Synthesis of ectoine is energy intensive and  
310 does not provide apparent advantages to *Alcanivorax* growth under HP [65,66].

311

### 312 **Concluding remarks and future perspectives**

313 Due to increasing demand for energy worldwide, the exploration of novel oil fields has  
314 increased. These efforts increase the risk of exposure of marine life and the marine  
315 environment to HCs [8]. Marine microorganisms are the ultimate HC degraders and play

316 a key role in cleanup events [5]. However, they can be exploited more efficiently if the  
317 black box of their response to multifactorial environmental and pollution parameters is  
318 opened and fundamental questions on their metabolism, physiology and ecology are  
319 answered (see Outstanding Questions). Meta-omics data are largely contributing to  
320 elucidate how environmental factors drive the assembly and function of HC-degrading  
321 communities. More players than those currently known are revealed and they cooperate  
322 to add their 'incomplete' metabolic pathways to HC degradation [14]. Metabolic  
323 complementation and syntrophy could be as actual strategies for effective oil HC  
324 cleanup in marine ecosystems. This suggests that biostimulation and bioaugmentation  
325 approaches, including autochthonous bioaugmentation, should be carefully rethought  
326 [31]. Furthermore, the recent advancements in understanding metabolism, physiology  
327 and adaptation of HC-degrading microorganisms are contributing to explain their  
328 successes and failures in various polluted environments. For instance, the impaired  
329 response of *Alcanivorax* to HP and its associations to different polluted compartments  
330 following the DWH disaster clarifies its absence in the deep-sea oil plume [65].  
331 Similarly, recent developments in microbial bioelectrochemistry are contributing to the  
332 design of novel solutions for HC cleanup through the channelling of electron flows, such  
333 as that exploited by the "oil-spill snorkel," to compensate for the absence of oxygen in  
334 contaminated sediments [50]. For certain, novel discoveries in microbial ecology and  
335 physiology associated with HC pollution in marine ecosystems will enlighten and lead to  
336 more efficient and sustainable microbial biotechnology approaches to ridding the  
337 oceans of oil.  
338

339 **Figure Legends**

340

341 **Figure 1. Current (Non-Microbial) Technologies for Emergency Responses to Oil**

342 **Spills in Marine Environments.** On the sea surface, the oil spots accidentally released

343 by oil tankers or offshore platforms can be chemically dispersed or physically contained

344 by plastic booms and partially recovered with skimmers or they can undergo *in situ*

345 burning. These processes may enhance accumulation of solid residues (e.g., tar balls)

346 that can sink. The volatile fractions move to the atmosphere where the HCs can be

347 transported far away and fall back to the surface again. HCs can reach the coast and

348 contaminate beaches where they can be removed with oil-sorbent materials. On rocky

349 beaches, combinations of high-pressure water spray and application of sorbent

350 materials are used. A further possibility is the removal of the contaminated sand and *ex-*

351 *situ* treatment in specialized centres. HC contamination can occur at the seafloor due to

352 losses from oil wells and shipwrecks or during drilling operations. Oil droplets can then

353 move across the water column towards the seafloor, spread horizontally to form HC

354 plumes in the water column or reach the sea surface and eventually the coastal zone.

355 Dispersants are used to decrease the size of oil droplets and increase the surface-to-

356 volume ratio. Oil from sunken ships can be removed through suction from intake

357 platforms positioned on the sea surface, while wellhead capping is necessary to stop oil

358 blowout from deep-sea wells. Natural attenuation and dispersion phenomena in the

359 water column and volatilization in the atmosphere additionally contribute to HC removal.

360

361 **Figure 2. Biotechnological Approaches to the Remediation of Hydrocarbon (HC)-**  
362 **Polluted Marine Environments.** The activity of HC-degrading microorganisms can be  
363 enhanced by nutrients and biosurfactants, provided alone or in combination. Besides  
364 the HC-degraders, microbes playing key roles such as biosurfactant producers are  
365 pivotal to set up a successful intervention. Bioaugmentation (BA) of HC-degrading  
366 microbes selected under laboratory conditions can be used to enhance HC degradation  
367 rates. A strategy to overcome the reported lack of adaptability of allochthonous  
368 microbes consists of autochthonous bioaugmentation (ABA) using isolates obtained  
369 from the matrix to be treated. Biostimulation is also used to enhance the activity of the  
370 native and/or the augmented microbial communities and biosurfactants can be added to  
371 increase HC bioavailability. By exploiting suitable bio-carriers, nutrients, biosurfactants  
372 and degrading microbes can be mixed to formulate slow-release particles (SRP) that  
373 allow continuous and homogeneous release of the active ingredients in the target  
374 environments. SRP applied to the sediments may contain alternative oxidants (e.g.,  
375  $\text{NO}_3^-$ ). Microalgae are involved in the water column cleanup following an oil spill.  
376 Supporting their growth would enhance marine snow formation, which results in HC  
377 precipitation to the seafloor. This might limit the oil-impacted seafloor area before deep-  
378 sea currents drive the oil plume far from the HC-spill origin and provide nutrients to  
379 microbial HC-degrading populations on the seafloor. All biotechnological solutions must  
380 take into account the environmental conditions of the system, including temperature and  
381 hydrostatic pressure.

382

383 **Figure 3. Natural and Engineered Biodegradation Processes in Marine Sediments**  
384 **Contaminated by Petroleum Hydrocarbons.** (A) Steady-state stratification of  
385 hydrocarbon-fuelled respiratory metabolisms based on the availability of terminal  
386 electron acceptors. In the most superficial sediment layer (ranging from a few  
387 millimetres to several centimetres depending on site characteristics), microorganisms  
388 respire the oxygen that diffuses from the overlaying seawater. Below the oxic zone,  
389 nitrate,  $Mn^{4+}$  and  $Fe^{3+}$ , if present, are used for anaerobic respiration. In the lower  
390 sedimentary layer, sulfate, which is not a limiting factor in marine environments,  
391 becomes the dominant respiratory electron acceptor. (B) Schematic representation of  
392 the “oil-spill snorkel”, a conductive graphite rod half-buried in the contaminated  
393 sediment that creates an electrochemical connection between the anoxic sediment and  
394 the oxic overlaying seawater. In principle, the “snorkel” may accelerate HC oxidation by  
395 both serving as a direct electron acceptor in the respiratory metabolism of electro-active  
396 bacteria growing at its surface and by stimulating, in the bulk of the sediment, the  
397 metabolism of sulphate-reducing bacteria via sulphide scavenging. (C) Schematic  
398 representation of a modular slurry system (MSS) designed for stimulating aerobic  
399 biodegradation processes in otherwise anoxic sediments. The system allows *in situ*  
400 aeration of the sediment, temperature control, and also possibly the delivery of nutrients  
401 and/or other biostimulating agents to increase, for instance, the bioavailability of  
402 sediment-bound contaminants.

403

404

405

**Table1.** Main cultivated marine hydrocarbon-degrading bacteria and their phylogenetic, physiological and ecological features.

A/AN	Genome	Microorganism	Phylogeny	Target substrate	Habitat/Ecology	Physiology	Refs
A	Available	<i>Alcanivorax borkumensis</i>	$\gamma$ -proteobacteria, <i>Alcanivoracaceae</i>	<i>n</i> -alkanes	Seawater, sediment, beach sand, coastal salt marsh	Biosurfactant producer, OHCB	[70]
A	Available	<i>Alcanivorax dieselolei</i>	$\gamma$ -proteobacteria, <i>Alcanivoracaceae</i>	<i>n</i> -alkanes	Seawater, sediment	Resistance to mild pressure increase, OHCB	[69]
A	Available	<i>Marinobacter hydrocarbonoclasticus</i> :	$\gamma$ -proteobacteria, <i>Alteromonadaceae</i>	<i>n</i> -alkanes, PAHs	Seawater, sediment	Biofilm producer; oil surface colonizers	[71]
A	Available	<i>Cycloclasticus pugetii</i>	$\gamma$ -proteobacteria, <i>Piscirickettsiaceae</i>	PAHs	Sediment	Highly efficient transport systems for the capture of nutrients and oligo-elements	[72]
A	Available	<i>Oleispira antarctica</i>	$\gamma$ -proteobacteria, <i>Oceanospirillaceae</i>	<i>n</i> -alkanes	Seawater	cold-adapted OHCB	[73]

A: aerobic; AN: anaerobic; OHCB: obligate hydrocarbonoclastic bacteria; PAH: polycyclic aromatic hydrocarbons.

## 1 **Text Boxes**

### 2 **BOX 1. The use and impacts of dispersants following oil spills.**

3 There are consolidated approaches for dispersant administration on oil-polluted surface  
4 waters. In general, a dispersant administration protocol takes into account the trade-off  
5 between the efficiency of dispersion into fine droplets and the increased toxicity of  
6 water-accommodated oil fractions to marine life. It has been recently highlighted that  
7 dispersants may alter and steer the diversity and activity of microbial communities with  
8 potential, not yet understood effects on oil HC degradation [7]. There are no  
9 formulations specifically designed for deep-sea releases. It is believed that in this case  
10 no solvents are needed as the light HC oil components are all there, and mixing is  
11 favoured by natural gas that is released from the crude oil as it ascends through the  
12 water column. During the DWH oil spill, about seven million litres of the Corexit  
13 dispersant were injected into the surface of the Gulf of Mexico and deep-sea waters  
14 [74]. Key Corexit components such as the anionic surfactant dioctyl sodium  
15 sulfosuccinate were not biodegraded *in situ* [75], while addition of Corexit in *ex situ* tests  
16 inhibited HC biodegradation and shifted microbial communities towards potential  
17 dispersant degraders (e.g., *Colwellia*; [7]).

18

### 19 **BOX 2. Effects of hydrostatic pressure on microbial cells.**

20 Hydrostatic pressure (HP) is an intrinsic feature of the deep sea. HP linearly increases  
21 with depth (about 0.1 MPa every 10 m of seawater column) and affects cell structures  
22 and functions, with a genome-based response proposed for piezophilic adaptation. Le  
23 Châtelier's principle predicts that processes entailing volume reduction are favoured

24 under HP. HP can affect chemical reactions (both for volume change at equilibrium and  
25 for activation) and macromolecule structures, particularly the weak non-covalent bonds  
26 and protein complexes with several subunits. This is the case for ribosome assembly or  
27 for nucleic acid/protein complexes with volume increases [76–78], where effects are  
28 generally observed at sub-lethal pressures. Another recognized HP effect is on cell  
29 membranes, whose relative abundance in mono- (and to a lower extent poly-)  
30 unsaturated fatty acids, necessary to maintain membrane fluidity and cell homeostasis,  
31 can be affected. A high degree of fatty acid unsaturation does not favour fatty acid acyl  
32 group packing due to the physical encumbrance of the lateral chains, with positive  
33 effects on the cell membrane's homeoviscous properties and curvature elastic stress  
34 [79]. Increased unsaturation of fatty acids on the membrane may result from another  
35 typical deep-sea condition, *i.e.*, low temperature, whose effects are additive to HP [80].  
36 HP affects membrane transport systems and transmembrane enzymes, such as  
37 ATPases and cytochromes, due to direct effects on enzyme folding or the lipid  
38 environment on the membrane. Adaptation to HP potentially involves energy generation  
39 in the cell, likely to counteract HP-related stressing effects. The model piezophile  
40 *Photobacterium profundum* SS9 possesses two complete operons for the F<sub>0</sub>F<sub>1</sub> ATPase  
41 and multiple cytochrome sets, supporting the pivotal importance of electron and proton  
42 transport under high HP [81]. Similarly, *Shewanella piezophila* possesses two  
43 respiratory chains, active either under low or high HP [82]. The capacity to offset HP  
44 impact on cell turgor pressure has been little studied. This would involve the intracellular  
45 accumulation of piezolytes, as occurring for osmolytes and salinity. While their  
46 mechanism of action remains unclear, HP increase is consistent with the accumulation

47 of N-trimethylamine oxide (TMAO) [83],  $\beta$ -hydroxybutyrate [84] and ectoine [65]. In  
48 *Alcanivorax borkumensis*, intracellular ectoine accumulation and gene upregulation  
49 under HP were correlated with decreased cell damage and higher cell number, but  
50 culture activity was not increased. Experiments on the synergistic effects of osmotic and  
51 HP increase suggest that ectoine water-reclamation capacity might also explain its  
52 function at high HP [66].

53

54

## 55 **Glossary Box**

56 **Bioaugmentation:** The addition of active microorganisms with specialized metabolic  
57 capacities to enhance a metabolic process, such as HC degradation in a polluted site.

58 **Bioremediation:** The exploitation of the catabolic abilities of microorganisms that use  
59 pollutants (such as HCs) as a source for carbon and energy for their metabolism and  
60 growth. Two general approaches can be defined in which the polluted material is left on  
61 site (*in-situ* bioremediation) or is removed and treated away from the polluted site (*ex-*  
62 *situ* bioremediation). It can be performed by indigenous microbial communities or by  
63 allochthonous microorganisms added to the polluted system through bioaugmentation.

64 **Biostimulation:** The modification of the environmental conditions of a polluted matrix to  
65 favour the metabolism of microorganisms capable of bioremediation, generally  
66 consisting of the addition of nutrients, but also of oxygen or other electron acceptors  
67 and eventually electron donors (for instance in the case of reductive dechlorination), or  
68 of the addition of substances enhancing HC bioavailability (*e.g.*, biosurfactants).

69 **Biosurfactants:** Amphipatic surface-active macromolecules produced by  
70 microorganisms to increase the availability of HCs. They can be grouped as low-  
71 molecular-weight molecules with good solubilization properties (lipopeptides,  
72 phospholipids, but most commonly glycolipids *e.g.*, rhamnolipids, trehalose lipids,  
73 sophorolipids) or high molecular weight compounds with coating properties preventing  
74 oil-droplet coalescence (*e.g.*, proteins, polysaccharides, lipids and their complexes).

75 **Meta-omics:** Molecular analyses based on the comprehensive study of enzymes and  
76 proteins (metaproteomics), RNA and DNA sequences (metatranscriptomics and  
77 metagenomics) or metabolites (metabolomics) in a given environmental sample.

78 **Micro-encapsulation:** The incorporation of different chemical components (such as  
79 nutrients, enzymes, microorganisms, etc.) within a coating to form small capsules in the  
80 size range of micrometers to a few millimetres, to protect the content from early  
81 degradation and favour its localized and time-dependent release.

82 **Microbial succession:** The temporal change in dominant species within a microbial  
83 community in response to the changing environmental conditions, such as the changes  
84 in the type and concentrations of carbon sources.

85 **Mobile genetic elements:** Genetic elements such as plasmids or transposons that  
86 contain genetic modules allowing their movement within the genome of an organism or  
87 between genomes of different organisms. They may carry catabolic genes that enable  
88 the cells that express them to degrade specific compounds.

89 **Natural attenuation:** The process of HC biodegradation following an oil spill, mediated  
90 by the indigenous microbial community. This process relies upon the presence of  
91 microorganisms with HC-catabolic capacities in the ecosystem prior to the spill. The  
92 availability of HC substrates enriches the microorganisms that are capable of exploiting  
93 HCs for growth. Natural attenuation is made possible by consortia of microorganisms  
94 with complementary functions. These microorganisms include HC degraders,  
95 biosurfactant producers, and nutrient providers involved in N and P cycles or  
96 siderophore producers.

97 **Water column:** the term conceptually represents all the different layers of water present  
98 in a water body, such as the oceans, moving from the surface to the sediment on the  
99 bottom. Along the water column, different layers can be identified and their boundaries  
100 are defined by specific water depth.

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## Competing interests

The authors declare no competing financial interests.

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## **Biotechnologies for marine oil spill cleanup: indissoluble ties with microorganisms**

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**Trends Box (Max 900 characters w/spaces. Actual: 898)**

## **Trends Box**

The cleanup of oil spills in marine environments ultimately relies on microbial metabolism of hydrocarbons (HC), which complements the current chemico-physical techniques used in emergency response.

Consolidated biotechnologies include microbial communities biostimulation, biosurfactant supplementation and bioaugmentation HC-degrading microbial cells.

The effectiveness of biotechnologies is limited by our understanding of the microbial ecology of polluted marine systems. We lack knowledge on how environmental factors, such as hydrostatic pressure, temperature and dispersant toxicity, affect microbial successions.

The recent availability of meta-omics data and the improved understanding of microbial metabolism are leading to novel biotechnologies for marine oil spill cleanup, such as slow-release particles for efficient biostimulation and bioelectrochemical approaches for sediment cleanup.

## **Biotechnologies for marine oil spill cleanup: indissoluble ties with microorganisms**

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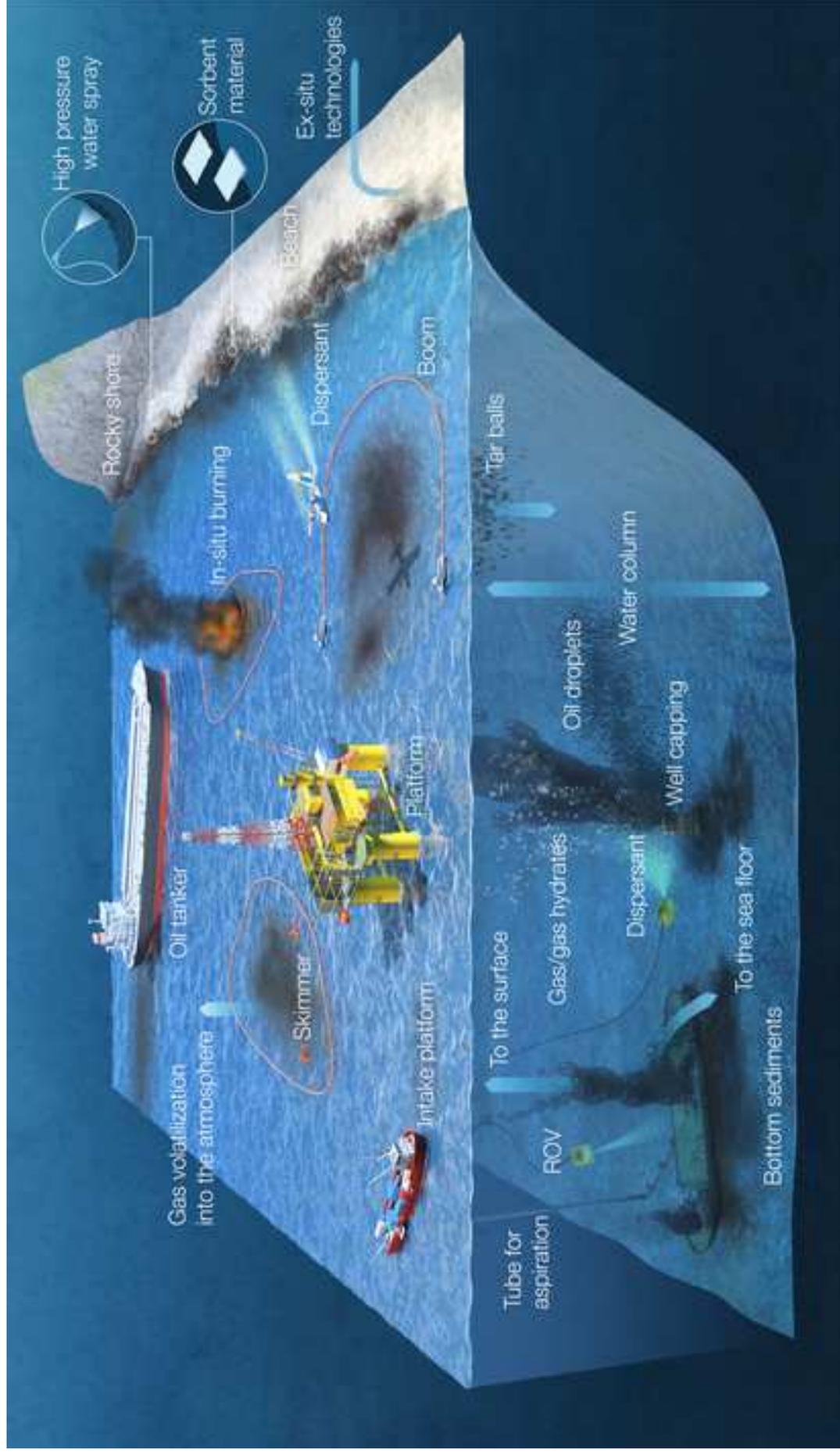
\*Correspondence: [daniele.daffonchio@kaust.edu.sa](mailto:daniele.daffonchio@kaust.edu.sa) (D. Daffonchio). Web site: <https://www.kaust.edu.sa/en/study/faculty/daniele-daffonchio>

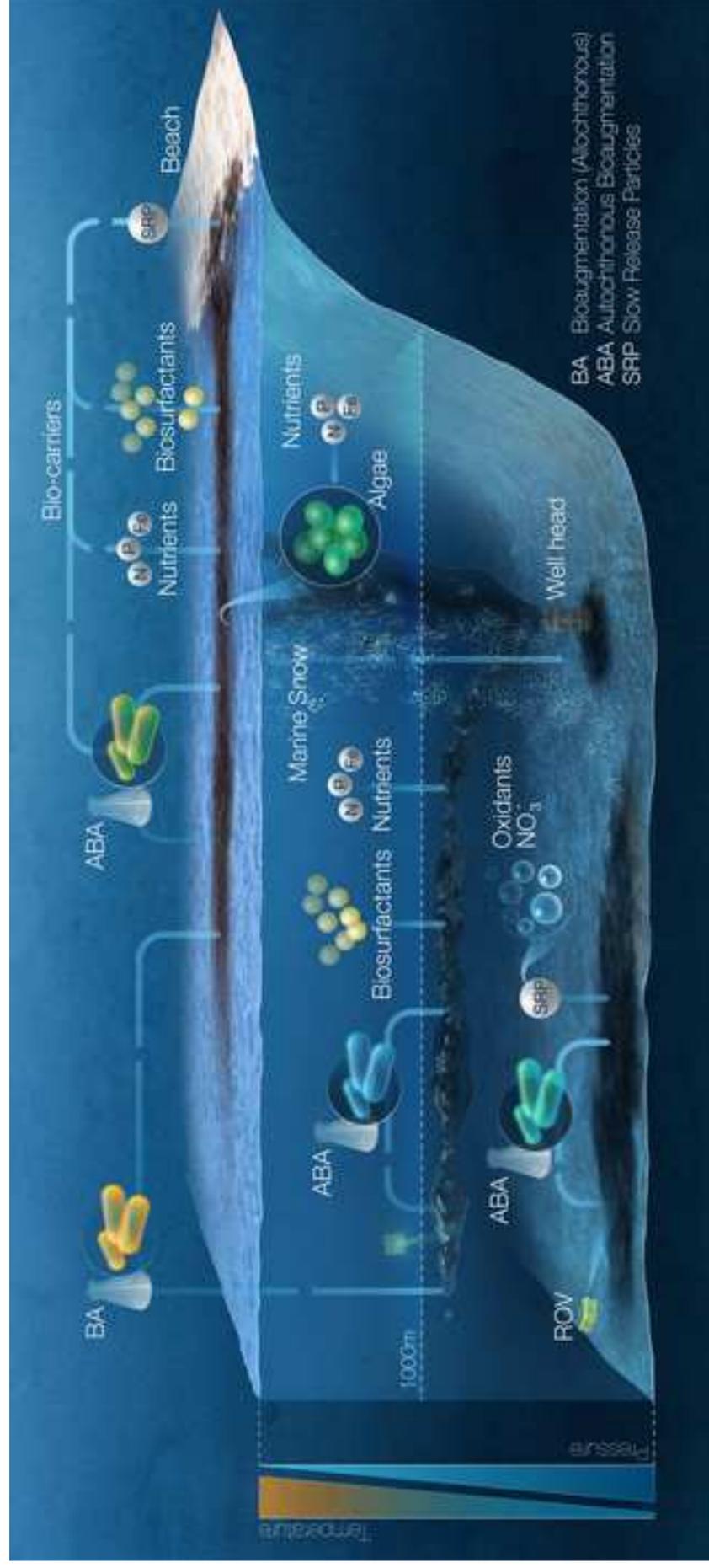
**Outstanding Question Box (Max 2000 characters w/spaces. Actual: 1664)**

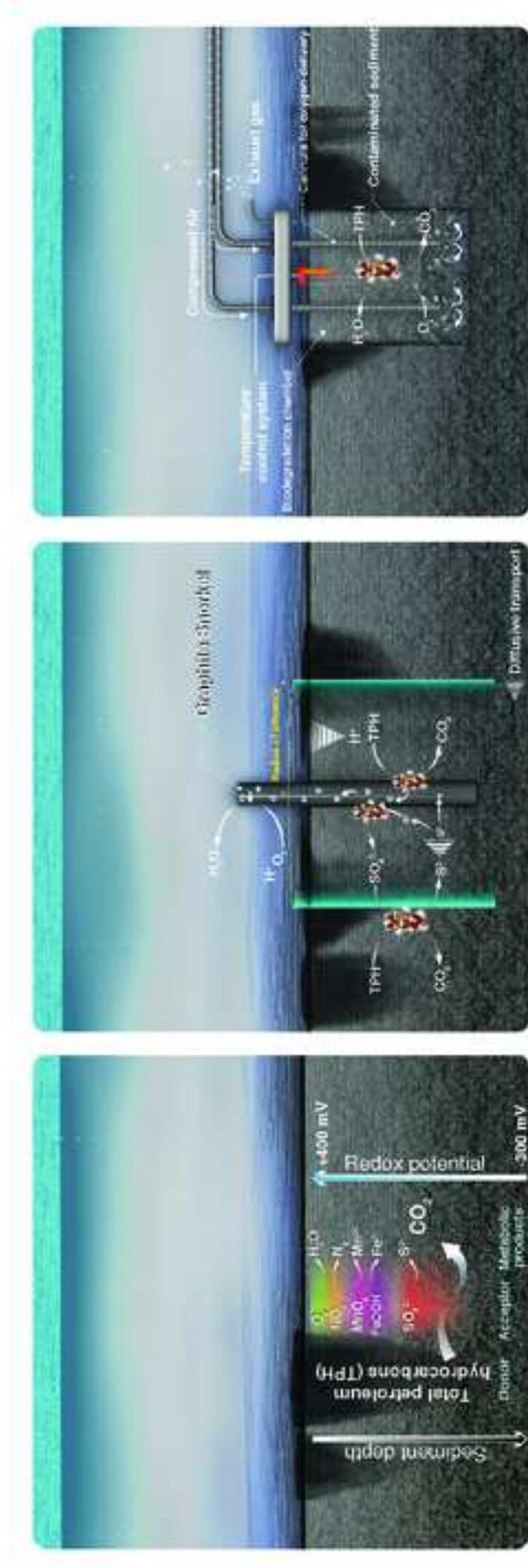
## 1 **Outstanding Question Box**

2 The continuous increasing understanding of microbial physiology, metabolism and  
3 ecology is driving the development of novel, microbially driven biotechnological  
4 applications to the oil hydrocarbon (HC) cleanup in marine environments. Among those  
5 described in this review, the application of slow-release particles (SRPs), the “oil-spill  
6 snorkel” and modular slurry systems are potentially promising biotechnologies for the  
7 remediation of polluted water and sediments based on the activity of microorganisms or  
8 their products. However, a series of questions and unresolved issues regarding both  
9 basic and applied science remain. These questions are:

- 10 1) How do we improve the control of component release by SRPs?
- 11 2) How do we select novel, improved and versatile biosurfactants or modify existing  
12 ones for applications as components of biodispersants in a cost-effective way?
- 13 3) What is the fine structure and dynamics of the full functional and metabolic  
14 cooperation between microbial components of HC-degrading microbial consortia?
- 15 4) How do neglected factors such as hydrostatic pressure affect the response of  
16 autochthonous natural microorganisms in polluted marine compartments?
- 17 5) How do combinations of environmental stressors act on HC-degraders?
- 18 6) Is there a biogeographically driven distribution of marine microbial HC degraders?
- 19 7) What is the prevalence and functional importance of unculturable HC degraders that  
20 are emerging from recent meta-omics reconstructions?
- 21 8) What is the balance and diversification of electron flow in the “oil-spill snorkel”?
- 22 9) Is it advantageous to combine different biotechnologies, for instance SRPs and the  
23 “oil-spill snorkel”?







1 **Biotechnologies for marine oil spill cleanup: indissoluble ties**  
2 **with microorganisms**

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24 **Supplementary Table 1 and Supplementary References**

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**Supplementary Table 1.** A comprehensive list of main cultivated marine hydrocarbon-degrading microorganisms and their phylogenetic, physiological and ecological features.

A/AN	Genome	Microorganism	Phylogeny	Target substrate	Habitat/Ecology	Physiology	Refs
A	Y	<i>Alcanivorax borkumensis</i>	$\gamma$ -proteobacteria, Alcanivoracaceae	n-alkanes	Seawater, sediment, beach sand, coastal salt marsh	Biosurfactant producer, OHCB	[1-4];
A	Y	<i>Alcanivorax dieselolei</i>	$\gamma$ -proteobacteria, Alcanivoracaceae	n-alkanes	Seawater, sediment	Resistance to mild pressure increase, OHCB	[5-7]
A	Y	<i>Marinobacter hydrocarbonoclasticus</i> :	$\gamma$ -proteobacteria, Alteromonadaceae	n-alkanes, PAHs	Seawater, sediment	Biofilm producer; oil surface colonizers	[8-10]
A	Y	<i>Cycloclasticus pugei</i>	$\gamma$ -proteobacteria, Piscirickettsiaceae	PAHs	Sediment	Highly efficient transport systems for the capture of nutrients and oligo-elements	[11, 12]
A	Y	<i>Oleispira antarctica</i>	$\gamma$ -proteobacteria, Oceanospirillaceae	n-alkanes	Seawater	Cold-adapted OHCB	[13]
A	N	<i>Oleibacter marinus</i>	$\gamma$ -proteobacteria, Oceanospirillaceae	n-alkanes	Seawater	Adapted to tropical marine environments	[14-15]
A	N	<i>Oleiphilus messinensis</i>	$\gamma$ -proteobacteria, Oleiphilaceae	n-alkanes	Seawater, sediment	Biofilm producer on oil droplets, OHCB	[16]
A/AN	Y	<i>Pseudomonas pachastrellae</i>	$\gamma$ -proteobacteria, Pseudomonadaceae	n-alkanes, PAHs	Sediment, beach sand	Bioemulsification activity	[17-19]
A/AN	Y	<i>Pseudomonas stutzeri</i>	$\gamma$ -proteobacteria, Pseudomonadaceae	n-alkanes, PAHs, BTEX	Seawater, marsh and marine sediments, beach sand	Biofilm producer	[19-21]
A	N	<i>Halomonas halodurans</i> ; <i>Halomonas organivorans</i>	$\gamma$ -proteobacteria, Halomonadaceae	n-alkanes	Seawater, sediment	Key role in N metabolism to sustain degrading consortia	[22,23]

A/AN	Genome	Microorganism	Phylogeny	Target substrate	Habitat/Ecology	Physiology	Refs
A	Y	<i>Thalassolituus oleivorans</i>	$\gamma$ -proteobacteria, Oceanospirillaceae	<i>n</i> -alkanes	Surface seawaters, sediments, coastal and estuarine areas	OHCB	[24]
A	Y	<i>Alteromonas naphthalenivorans</i>	$\gamma$ -proteobacteria, Alteromonadaceae	PAHs	Seawater, tidal flat sediment	<i>r</i> -strategist, fast growth in nitrogen-deficient seawater	[25]
A	Y	<i>Acinetobacter venetianus</i>	$\gamma$ -proteobacteria, Moraxellaceae	<i>n</i> -alkanes	Surface water, sediment.	Biosurfactant producer	[26]
A	Y	<i>Dietzia maris</i>	Actinobacteria, Dietziaceae	<i>n</i> -alkanes, PAHs	Seawater, deep sea hydrothermal field	Biosurfactant producer	[27,28]
A	N	<i>Rhodobacter</i> sp. SS12.29; <i>Rhodococcus</i> sp. ice-oil-488 s	$\alpha$ -proteobacteria, Rhodobacteraceae	PAHs	Seawater	Key role in reducing the accumulation of metabolites resulting from PAH degradation	[29]
A	N	<i>Sphingopixis</i> sp.	$\alpha$ -proteobacteria, Sphingomonadaceae	PAHs	Seawater	Key role in reducing the accumulation of metabolites resulting from PAH degradation	[29]
AN	Y	<i>Desulfatibacillum alkenivorans</i>	$\delta$ -proteobacteria, Desulfobacteraceae	<i>n</i> -alkanes	Sediment	High metabolic versatility for anaerobic alkane utilization	[30]
AN	N	Desulfosarcina-Desulfococcus cluster strains	$\delta$ -proteobacteria Desulfobacteraceae	Short chain <i>n</i> -alkanes	Sediments of marine HC seeps	Propane and butane degraders; sulfate-reducing bacteria	[31,32]
AN	N	<i>Desulfococcus oleovorans</i>	$\delta$ -proteobacteria, Desulfobacteraceae	<i>n</i> -alkanes, aromatic HCs	Sediment	Sulfate-reducing bacteria	[33,34]

A/AN	Genome	Microorganism	Phylogeny	Target substrate	Habitat/Ecology	Physiology	Refs
A	Y	<i>Bacillus pumilus</i>	Bacilli, Bacillaceae	n-alkanes, PAHs	Sediment	Resistance to heavy metals	[20,35]
A	N	<i>Bacillus stratosphericus</i>	Bacilli, Bacillaceae	PAHs, BTEX	Seawater	High metabolic versatility, biosurfactant producer	[36]
AN	Y	<i>Archaeoglobus fulgidus</i>	Euryarchaeota, Archaeoglobaceae	n-alkanes	Shallow marine hydrothermal system	Extremophile	[37]
AN	Y	<i>Thermococcus sibiricus</i>	Euryarchaeota, Thermococcaceae	n-alkanes	Oil reservoir	High metabolic versatility	[38]
AN	Y	<i>Ferroglobus placidus</i>	Euryarchaeota, Archaeoglobaceae	Aromatic HCs	Shallow marine hydrothermal system	Hyperthermophilic	[39,40]
A	N	<i>Dothideomycetes-related taxa</i>	Fungi	PAHs	Beach sediment, tarballs, salt marshes	-	[41,42]

A: aerobic; AN: anaerobic; OHCB: obligate hydrocarbonoclastic bacteria, PAH: polycyclic aromatic hydrocarbons; Y/N: Y, genome available for at least one strain/ N, genome not available (Genome availability checked on NCBI database on 27 November 2016).

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