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Pyroxenite Layers in the Northern Apennines' Upper Mantle (Italy) – Generation by Pyroxenite Melting and Melt-infiltration

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Complete List of Authors:	Borghini, Giulio; University of Milano, Dipartimento di Scienze della Terra Ardito Desio; Università degli Studi Di Genova, Dipartimento di Scienze della Terra, dell'Ambiente e della Vita; Columbia University, Lamont Doherty Earth Observatory Rampone, Elisabetta; Università degli Studi Di Genova, Dipartimento di Scienze della Terra, dell'Ambiente e della Vita Zanetti, Alberto; Consiglio Nazionale delle Ricerche (CNR), Istituto di Geoscienze e Georisorse Class, Cornelia; Columbia University, Lamont Doherty Earth Observatory Cipriani, Anna; University of Modena and Reggio Emilia, Dipartimento di Scienze Chimiche e Geologiche; Columbia University, Lamont Doherty Earth Observatory Hofmann, Albrecht; Max Plank Institute of Chemistry, ; Columbia University, Lamont Doherty Earth Observatory Goldstein, Steven; Columbia University, Department of Earth and Environmental Sciences; Columbia University, Lamont Doherty Earth Observatory
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9 6 **GIULIO BORGHINI^{1,2,3}, ELISABETTA RAMPONE², ALBERTO ZANETTI⁴,**
10 7 **CORNELIA CLASS³, ANNA CIPRIANI^{3,5}, ALBRECHT W. HOFMANN^{3,6}, STEVEN**
11 8 **L. GOLDSTEIN^{3,7}**
12 9

13 10
14 11
15 12 ¹DIPARTIMENTO DI SCIENZE DELLA TERRA, UNIVERSITÀ DI MILANO, VIA BOTTICELLI 23, 20133
16 13 MILANO, ITALY

17 14 ²DIPARTIMENTO DI SCIENZE DELLA TERRA, DELL'AMBIENTE E DELLA VITA, UNIVERSITÀ DI
18 15 GENOVA, CORSO EUROPA 26, 16132 GENOVA, ITALY

19 16 ³LAMONT-DOHERTY EARTH OBSERVATORY OF COLUMBIA UNIVERSITY, 61 ROUTE 9W,
20 17 PALISADES, NY 10964, USA

21 18 ⁴CNR-ISTITUTO DI GEOSCIENZE E GEORISORSE, SEZIONE DI PAVIA, VIA FERRATA 1, I-27100
22 19 PAVIA, ITALY

23 20 ⁵DIPARTIMENTO DI SCIENZE CHIMICHE E GEOLOGICHE, UNIVERSITÀ DEGLI STUDI DI MODENA E
24 21 REGGIO EMILIA, LARGO S. EUFEMIA 19, 41100 MODENA, ITALY

25 22 ⁶MAX PLANCK INSTITUTE FOR CHEMISTRY, P.O. BOX 3060, 55020 MAINZ, GERMANY

26 23 ⁷DEPARTMENT OF EARTH AND ENVIRONMENTAL SCIENCES OF COLUMBIA UNIVERSITY,
27 24 LAMONT-DOHERTY EARTH OBSERVATORY, 61 ROUTE 9W, PALISADES, NY 10964, USA
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35 29 **Corresponding Author:**

36 30
37 31 **Giulio Borghini**

38 32 Dipartimento di Scienze della Terra "Ardito Desio"
39 33 Università degli Studi di Milano
40 34 Via Botticelli 23
41 35 20133 Milano (Italy)

42 36 Tel. 0039 02 50315600

43 37 Fax. 0039 02 50314597

44 38 Email: giulio.borghini@unimi.it
45 39
46 40
47 41
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52 43 **Keywords:**

53 44 Pyroxenites, mantle heterogeneity, melt-rock reaction, mantle geochemistry, trace element,
54 45 isotopes, External Ligurides ophiolites, mafic layers, ultramafic
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Abstract

Pyroxenite layers embedded within peridotite represent wide-spread lithological mantle heterogeneities and are potential components in the mantle source of many oceanic basalts. Pyroxenites can be generated by several magmatic and metamorphic processes. However, in most natural samples (especially in ultramafic massifs), their primary characteristics are partially or completely erased by later petrologic evolution (e.g. metamorphism, metasomatism or partial melting). Here we investigate a suite of pyroxenites from the External Liguride Jurassic ophiolites (Northern Apennines, Italy). They are spinel-bearing websterites and clinopyroxenites, partially recrystallized under plagioclase-facies conditions, and occur as cm-scale layers parallel to the tectonite foliation of their host peridotites. Pyroxenites have bulk Mg-numbers from 74 to 88 and display rather constant LREE depletion over the MREE ($La_N/Sm_N = 0.15-0.35$), but variable MREE-HREE fractionation, with some of them having markedly positive HREE slopes ($Sm_N/Yb_N=0.30-0.96$). The HREE enrichment coupled with high Zr and Sc contents in clinopyroxene porphyroclasts from spinel-bearing domains provides strong evidence that garnet was present in the precursor mineral associations. Mass balance calculations suggest that the pyroxenites originally contained up to about 40 vol% of garnet, indicating they originated by segregation of melts at rather high pressure ($P > 1.5$ GPa). Parental melts of pyroxenites have reacted to some extent with the host peridotite during mantle infiltration. Lack of olivine in the primary mineral assemblage and orthopyroxene-rich rims along the contact with wall-rock peridotites indicate that pyroxenites have crystallized from silica-rich melts. These, likely, had REE patterns and Sr-Nd isotope compositions similar to enriched MORB. We propose that the pyroxenites originated from melts derived from a hybrid eclogite-bearing peridotite source, and then reacted with the host peridotite to form “secondary pyroxenites”. Their existence has been invoked in current models of basalts petrogenesis. During later decompression, they experienced an intermediate recrystallization at spinel-facies conditions, at 1.2-1.5 GPa and minimum temperature of 950-1000°C, and partial re-equilibration at low-pressure plagioclase-facies. The latter is dated by internal Sm-Nd isochrons at 178 (± 8) Ma and is associated with Mesozoic exhumation, during extension of the Tethys lithosphere.

1 INTRODUCTION

2 Geochemical studies on oceanic basalts have documented that their upper mantle
3 sources are chemically and isotopically heterogeneous (e.g. Hofmann, 2003). The nature,
4 distribution and scale of such mantle heterogeneities are of fundamental importance for the
5 understanding of the dynamics and long-term evolution of the upper mantle. Direct
6 observations on mantle rocks, such as mantle xenoliths, abyssal peridotites and tectonically
7 exhumed mantle sectors (Alpine-type peridotite massifs), have indicated that
8 pyroxenite/mafic layers, commonly embedded within peridotite, represent wide-spread
9 lithological mantle heterogeneities (e.g. Bodinier & Godard, 2003; Downes, 2007). Whether
10 or not pyroxenite components occur in the upper mantle has been hotly debated during the
11 last decades, mostly because of their potential involvement in mantle partial melting and
12 generation of basalt (e.g. Hirschmann and Stolper 1996; Stracke et al., 1999; Salters & Dick,
13 2002; Kogiso et al., 2004a,b; Sobolev et al., 2005; Lambart et al., 2013).

14 Experimental studies (e.g. Yaxley & Green, 1998; Hirschmann et al., 2003;
15 Pertermann & Hirschmann, 2003; Kogiso et al., 2004a; Sobolev et al., 2007; Lambart et al.,
16 2009, 2012) have shown that the partial melting of mafic components, occurring at mantle
17 depths as pyroxenites and eclogites, can produce highly variable melt compositions and,
18 together with melts from depleted peridotite, reproduce the wide compositional range
19 documented in oceanic basalts. On this basis, geochemical models have highlighted the role
20 of a mixed pyroxenite-peridotite mantle source in the genesis of mid-ocean ridge (MORB)
21 and ocean island (OIB) basalts (e.g. Allegre & Turcotte, 1986; Hirschmann & Stolper, 1996;
22 Kogiso et al., 2004b; Sobolev et al., 2005; Stracke & Bourdon, 2009; Herzberg, 2011; Mallik
23 & Dasgupta, 2012; Lambart et al., 2012).

24 The nature of mantle pyroxenites, however, remains a controversial subject, mostly
25 because their origin has been related to a variety of magmatic and metamorphic processes,
26 including: i) recycling of subducted oceanic crust stirred by mantle convection and
27 incorporated into the lithosphere (e.g. Allègre & Turcotte, 1986; Kornprobst et al., 1990;
28 Morishita & Arai, 2001; Morishita et al., 2003, Yu et al., 2010), ii) moderate- to high-pressure
29 crystal segregation from magmas derived by melting of asthenospheric sources (Bodinier et
30 al., 1987a,b; Vannucci et al., 1993; Rivalenti et al., 1995; Kempton et al., 1997; Takazawa et
31 al., 1999; Mukasa & Shervais, 1999; Keshav et al., 2007; Dantas et al., 2007; Warren et al.,
32 2009; Gysi et al., 2011) or subducted oceanic crust (Davies et al., 1993; Pearson et al., 1993;
33 Becker, 1996; Pearson & Nowell, 2004), iii) refertilization of 'infertile' or 'depleted' upper
34 mantle via melt-peridotite reactions (Garrido & Bodinier, 1999; Bodinier et al., 2008; Dantas

1 et al., 2009, van Acken et al., 2010). An additional difficulty in characterizing the role of
2 pyroxenites in melting is associated with their reactivity within the mantle, causing them to be
3 easily modified by magmatic and metamorphic processes subsequent to their emplacement at
4 depth (e.g. Takazawa et al., 1999; Morishita & Arai, 2001; Morishita et al., 2003), interaction
5 with transient melts (e.g. Garrido & Bodinier, 1999; Chazot et al., 2005; Kaeser et al., 2009),
6 and decompressional partial melting (e.g. Pearson et al., 1993; Blichert-Toft et al., 1999;
7 Montanini et al., 2012).

8 Orogenic and ophiolitic massifs offer the possibility to observe structural, spatial and
9 geochemical relations between pyroxenites and host peridotites at different stages of their
10 evolution (Bodinier & Godard, 2003). They are also important as the only locations with the
11 possibility to gain insights on the distribution and extent of small-scale heterogeneities in the
12 upper mantle through direct field observations. In this paper, we present petrographic,
13 chemical and Sm-Nd isotope data on spinel pyroxenite samples from three, hundred-meter-
14 sized ultramafic bodies from the western sector of the External Liguride Jurassic ophiolites
15 (Northern Apennines, Italy; Fig. 1). Our microstructural and geochemical combined approach
16 allows us to infer primary characteristics of pyroxenites, conditions and timing of their
17 emplacement, geochemical affinities of their parental melts, and the later petrologic evolution.
18 Our results impact the fundamental question of how pyroxenite components originate in the
19 upper mantle, and provide further evidence of a “secondary-type” pyroxenite (also called
20 “2stage” pyroxenite, Sobolev et al., 2005, 2007; Lambart et al., 2013) emplacement at rather
21 deep mantle levels.

22 23 **GEOLOGICAL BACKGROUND**

24 Alpine-Apennine ophiolites represent lithospheric remnants of the Piemont-Ligurian
25 ocean basin, a branch of the Mesozoic Tethys separating the European and Adriatic passive
26 margins (Rampone & Hofmann, 2012, and references therein). The Middle Jurassic opening
27 of the basin led to uplift and denudation of the subcontinental lithospheric mantle, the
28 dominant lithotype exposed in Alpine-Apennine ophiolite sequences. Most of these mantle
29 bodies have been affected by thermochemical erosion and refertilization processes, related to
30 asthenospheric upwelling during rifting and oceanization (e.g. Rampone et al., 1997, 2004,
31 2008; Müntener et al., 2004; Piccardo et al., 2007). However, some of the mantle sequences
32 from the Northern Apennines, namely the ultramafic massifs from the External Liguride Unit,
33 have escaped these melt-rock interaction processes and preserved their old subcontinental
34 signature (Beccaluva et al., 1984; Rampone et al., 1995; Montanini et al., 2006; Borghini et

1 al., 2011). The External Liguride (EL) ophiolites consist of large slide-blocks (up to km-scale)
2 of ultramafic rocks, and minor MORB-type gabbros and basalts, within Cretaceous
3 sedimentary melanges that were obducted during the closure of the Ligurian Tethys oceanic
4 basin (Rampone et al., 1993, 1995; Tribuzio et al., 2004; Montanini et al., 2008). These
5 mantle sequences consist mostly of fertile spinel-plagioclase lherzolites with disseminated Ti-
6 rich amphibole and widespread spinel pyroxenite layers (Beccaluva et al., 1984). Lherzolites
7 locally retain extremely “depleted” Nd-Sr isotopic compositions (Rampone et al., 1995)
8 coupled with ancient Os model ages (1.6 Ga, Snow et al., 2000), which have been interpreted
9 to reflect long residence times in the subcontinental mantle. The External Liguride ophiolitic
10 mantle sequences have therefore been inferred to represent remnants of a fossil ocean-
11 continent transition, where exhumed subcontinental mantle was associated with embryonic
12 oceanic crust and rocks of continental origin (Marroni et al., 1998; Tribuzio et al., 2004;
13 Montanini et al., 2008), an analogue to the modern Iberian passive margin (e.g. Whitmarsh et
14 al., 2001).

15 Garnet pyroxenite layers from the northern sector of the External Liguride Unit have
16 been recently investigated by Montanini et al. (2006, 2012) and consist of thin websterites and
17 meter-scale thick garnet clinopyroxenites recording equilibration in the subcontinental
18 lithosphere at $T = 1100^{\circ}\text{C}$ and $P = 2.8 \text{ GPa}$. Sm-Nd and Lu-Hf geochronology on garnet-
19 plagioclase pairs yielded $186 \pm 2 \text{ Ma}$ and $220 \pm 13 \text{ Ma}$ ages, interpreted as cooling ages that
20 reflect the rifting-related exhumation of these mantle sectors (Montanini et al., 2006, 2012).
21 Pyroxenites investigated in this study are from the southern sector of the Western External
22 Liguride Unit (Suvero, La Spezia; Fig. 1). Borghini et al. (2011) recently showed that the
23 peridotites from this mantle sector experienced a rather cold spinel- to plagioclase-facies
24 decompressional tectonic evolution ($T=870\text{-}930^{\circ}\text{C}$), thus confirming that this pyroxenite-
25 peridotite association escaped melt-rock interaction and re-fertilization processes widely
26 documented in most Alpine-Apennine ophiolitic peridotites (e.g. Müntener & Hermann,
27 1996; Rampone et al., 1997, 2004, 2008; Rampone & Borghini, 2008; Piccardo et al., 2004,
28 2007).

30 **FIELD OCCURRENCE AND PETROGRAPHY**

31 A summary of the location and description of samples is presented in Table 1.
32 Pyroxenites occur as layers within the host mantle peridotites, mostly ranging in thickness
33 from several millimetres to about 15 centimetres (Fig. 2). Samples GV12 and GV14 occur as
34 lenses about 30 cm thick. Layers are parallel to the foliation plane of the host peridotites,

1 showing sharp contacts, and can be weakly boudinaged (Fig. 2d). In outcrop, pyroxenite
2 layers are recognised by their pinkish or dark-grey colour (Fig. 2a,b,c) and by the occurrence
3 of coarse black spinel grains. Pyroxenite abundance is variable, and in some outcrops can be
4 as high as 50% (Fig. 2a,b). In some places, pyroxenite layers alternate with harzburgite
5 portions within the peridotite, thus forming a compositional layering parallel to the tectonite
6 foliation (Fig. 2a,d).

7 Most of the pyroxenites are characterized by a porphyroclastic texture defined by
8 coarse-grained (up to 10-12 mm) porphyroclasts of pyroxenes and spinels, stretched along the
9 tectonite foliation. Porphyroclastic minerals are partially substituted by a plagioclase-bearing
10 granoblastic assemblage (Fig. 3a,b,c,d), indicating a retrograde evolution from spinel- to
11 plagioclase-facies. The low-P metamorphic recrystallization is recorded by the development
12 of i) millimeter and sub-millimeter bands of plagioclase and olivine (\pm amphibole) crystallized
13 around coarse porphyroclasts of greenish spinels and pyroxenes (Fig. 3a,b), ii) orthopyroxene
14 plus plagioclase exsolutions in large clinopyroxene porphyroclasts (Fig. 3b), iii) fine-grained
15 (\sim 200–300 μ m) granoblastic aggregates made of plagioclase, olivine, newly formed
16 pyroxenes and occasionally dark-brown spinel and brownish amphibole, partially replacing
17 the spinel-facies pyroxene porphyroclasts (Fig. 3c,d). Amphibole is heterogeneously
18 distributed in the pyroxenite samples and is always associated with the plagioclase-bearing
19 neoblastic assemblage (Fig. 3d). In the most altered samples (GV14 and GV17), granoblastic
20 olivine and plagioclase are almost completely substituted by low-T alteration products
21 (chlorite and saussurite, respectively).

22 Most of the pyroxenite samples are characterized by two alternating mineralogical
23 domains: i) spinel-rich portions, where green spinel relics are rimmed by plagioclase and
24 olivine (\pm amphibole), ii) pyroxene-bearing domains in which the large pyroxene
25 porphyroclasts are partially substituted by neoblastic aggregates of new pyroxenes and
26 plagioclase. The latter mineralogical feature is clearly observable in the thickest pyroxenite
27 layers, whereas the two domains are usually stretched and boudinaged in the thinner bands
28 (discussed further below). Most of the pyroxenite bands show irregular orthopyroxene-rich
29 borders along the contact with the hosting peridotite (Fig. 3e). This is the result of interaction
30 between the melt from which the pyroxenite crystallized and the host mantle peridotite during
31 the pyroxenite emplacement (Borghini et al., 2013).

32 Olivine and amphibole are totally absent in pyroxenite GV12; this sample is texturally
33 distinct and characterized by few small relics of clinopyroxene porphyroclasts that are
34 strongly substituted by a fine-grained equigranular metamorphic texture consisting of

1 clinopyroxene, dark-green spinel and plagioclase (completely replaced by low-T alteration
2 minerals, Fig. 3f). This granoblastic assemblage often forms clusters and seams stretched
3 along the foliation (Fig. 3f).

4 5 **ANALYTICAL METHODS**

6 Whole-rock major and trace element analyses of pyroxenites were made by lithium
7 metaborate/tetraborate fusion ICP-OES (for major elements) and ICP/MS (for trace elements)
8 at the Actlabs Laboratories (Western Ontario, Canada; <http://www.actlabs.com>).

9 Mineral major element compositions were analyzed using: (1) a Philips SEM 515
10 equipped with an X-ray dispersive analyzer (accelerating potential 15 kV, beam current 20
11 nA), at the Dipartimento di Scienze della Terra, dell'Ambiente e della Vita, University of
12 Genova, and (2) a JEOL JXA 8200 Superprobe equipped with five wavelength-dispersive
13 (WDS) spectrometers, an energy dispersive (EDS) spectrometer, and a cathodoluminescence
14 detector (accelerating potential 15 kV, beam current 15 nA), at the Dipartimento di Scienze
15 della Terra "Ardito Desio", University of Milano. *In situ* trace element mineral analyses were
16 carried out by laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) at
17 IGG-CNR in Pavia. A detailed description of analytical procedures is given in Miller et al.
18 (2007).

19 Sm and Nd isotopic compositions were measured on clinopyroxene and plagioclase
20 separates and on whole rock powders from pyroxenite samples. Clinopyroxenes and
21 plagioclases (90-150 μm sized) were separated by grinding, sieving, electromagnetic
22 separation and handpicking under a microscope. Weights of unleached mineral concentrates
23 were 25-82 mg for clinopyroxenes and 54-132 mg for plagioclases. Minerals were leached in
24 at least three steps using a solution of 5% HF + 6.2M HCl. Nd isotopic compositions, together
25 with elemental abundances of Sm and Nd by isotope dilution, were measured by thermal
26 ionization mass spectrometry (TIMS) on a VG Sector 54-30 at the Lamont-Doherty Earth
27 Observatory of Columbia University (Palisades, NY, USA). The $^{143}\text{Nd}/^{144}\text{Nd}$ ratios were
28 normalized to $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$, and reported relative to JNdi $^{143}\text{Nd}/^{144}\text{Nd} = 0.512115$
29 (Tanaka et al., 2000). Samples were analyzed during the course of 6 analytical sessions.
30 Individual sample data were corrected to the standard values measured during the individual
31 time intervals. Analyses of the JNdi-1 standard during the 6 sessions yielded $^{143}\text{Nd}/^{144}\text{Nd}$
32 ratios of 0.512080 ± 9 (2σ , mean of $n=7$ analyses, 34 ppm external reproducibility),
33 0.512074 ± 10 ($n=10$, 18 ppm), 0.512075 ± 12 ($n=13$, 23 ppm), 0.512088 ± 14 ($n=5$, 27 ppm),

1 0.512096±13 (n=26, 25 ppm), 0.512095±10 (n=5, 20 ppm). Total Sr and Nd blanks did not
2 exceed 150 pg.

3 4 **RESULTS**

5 **Bulk-rock chemistry**

6 The bulk-rock major and trace element data of the pyroxenite samples (Table 2)
7 display a large total range of Mg-numbers [molar $\text{Mg}/(\text{Mg}+\text{Fe}^{2+}_{\text{tot}})*100$] (74-88 mol%),
8 although most values are concentrated between 83-85. Conversely, pyroxenites cover a large
9 range of Al_2O_3 and CaO contents (10.6-16.5 wt% and 6.8-19.5 wt%, respectively; Fig. 4),
10 mostly reflecting their modal variation between clinopyroxenite and websterite assemblages.
11 In Figure 4, selected major and trace elements are plotted against Mg-number, together with
12 the compositional fields of garnet pyroxenites and spinel-bearing websterites from orogenic
13 and ophiolitic massifs (details and references in Fig. 4 caption).

14 Major element compositions of most samples lie in the fields of overlap between
15 garnet-pyroxenites and spinel-bearing websterites from orogenic massifs at intermediate Mg-
16 numbers (Fig. 4). However, they diverge from literature data by having generally higher CaO
17 and lower Na_2O contents. Samples GV14 and GV17 have unusual, very low SiO_2 abundances
18 (Table 2). Petrographic observations indicate that this is likely due to a high extent of low-
19 temperature hydrothermal alteration, mostly dominated by chloritization of pyroxenes and
20 olivine, as observed for many oceanic and ophiolitic mafic rocks (e.g. Alt et al., 1998).
21 Pyroxenite lens GV12 plots within the compositional field of garnet pyroxenites, but with
22 slightly higher CaO content (Fig. 4). Garnet-pyroxenites from other EL ultramafic massifs,
23 recently studied by Montanini et al. (2012), show lower Mg-numbers and higher Na_2O , Al_2O_3
24 contents relative to our samples, but similar SiO_2 and CaO compositional variations (Fig. 4).
25 In terms of Cr, Ni, V, Sc abundances, the pyroxenites also lie in the compositional field of
26 overlap between garnet pyroxenites and websterites in the literature, except for samples GV12
27 and GV14, which plot within the compositional field of garnet-pyroxenites, at lower Mg
28 values (Fig. 4). Our pyroxenites have low Cr, Ni (560-1530 ppm, 210-860 ppm, respectively)
29 and moderate Sc, V (31-51 ppm, 163-280 ppm, respectively) concentrations. EL garnet-
30 pyroxenites studied by Montanini et al. (2012) show similar values, at lower Mg-numbers
31 (Fig. 4).

32 The CI-normalized REE and other refractory elements (Fig. 5) of most of the
33 pyroxenites (BG3, BG4, BG5, BG13, BG14, BG22, GV10, GV17, MC3, MC5) display rather
34 constant LREE depletion over the MREE ($\text{La}_N/\text{Sm}_N = 0.15\text{-}0.35$), but variable MREE-HREE

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3 1 fractionation. In fact, the HREE slopes range from flat (GV17) to markedly positive (BG13)
4 (Sm_N/Yb_N=0.30-0.96), with absolute concentrations varying from 3 to 20 x CI. In particular,
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6 3 samples BG13, GV10, MC3 and MC5 display HREE enriched patterns as observed in some
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8 5 garnet pyroxenites (e.g. Pearson et al., 1993; Gysi et al., 2011; Montanini et al., 2012). Spinel-
9 6
10 7 rich clinopyroxenite GV12 has an overall lower REE content, showing a LREE-depleted
11 8
12 9 pattern combined with slightly fractionated MREE to HREE (Sm_N/Yb_N=0.77). The
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14 11 pyroxenite GV14 shows a convex-upward pattern with HREE depletion over MREE
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16 13 (Sm_N/Yb_N=1.66). Finally, pyroxenite GV8 displays a nearly flat pattern at 3-4 x CI. Zr and Hf
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18 15 show a rather flat trend between Nd and Sm, with some small positive Zr anomalies. In
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20 17 contrast, Sr is highly enriched compared to the LREE. These “spikes” probably derived from
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22 19 later fluid-driven contamination as evidenced by very thin whitish veins orthogonal to the
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24 21 layer direction not completely eliminated from the samples during powder preparation for
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26 23 bulk-rock analysis. This is further supported by the absence of positive Sr anomalies in any of
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28 25 the porphyroclasts. Our samples have MREE-HREE abundances within the compositional
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30 27 field defined by the External Liguride garnet-pyroxenites studied by Montanini et al. (2012)
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32 29 (Fig. 5). However, they have higher LREE content, with the exception of sample GV12,
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34 31 which is consistent with literature data.
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33 Mineral chemistry

34 20 Pyroxenes and spinel have Mg-numbers varying according to the Mg-number of their
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36 22 bulk-rock, and within each sample they show systematic major element variations according
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38 24 to the different mineral generations (porphyroblast vs. neoblast). Clinopyroxene
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40 26 porphyroclasts have Mg-numbers varying in agreement with their bulk-rock Mg-number,
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42 28 from 0.77 in the olivine-free clinopyroxenite GV12, to 0.88 in the high Mg-number
43 29
44 30 websterites MC3 and MC5 (Table 3 and Fig. 6a). They also have rather high Al₂O₃ contents
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46 32 (6.21-10.12 wt%) and low Cr₂O₃ and Na₂O abundances (≤0.36 wt% and ≤1.0 wt%
47 33
48 34 respectively; Table 3). A chemical profile on a representative pyroxenite layer shows that
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50 36 clinopyroxene porphyroclasts have Mg-numbers increasing from the center of the pyroxenite
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52 38 layer towards its margins (Suppl. Fig.1, and Suppl. Table 1), where Mg-numbers approach
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54 40 those of clinopyroxene of the host peridotite. As shown in Figure 6a, clinopyroxene neoblasts
55 41
56 42 in each sample exhibit systematically lower Al₂O₃ concentrations (3.16-6.62 wt%) and higher
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58 44 Mg-numbers relative to corresponding clinopyroxene porphyroclasts. In the orthopyroxene
59 45
60 46 porphyroclasts, the Mg-number varies from 0.82 to 0.88, and Al₂O₃ and CaO contents range
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62 48 between 4.33-7.56 wt% and 0.52-1.19 wt%, respectively (Table 4). Similar to clinopyroxenes,
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1 orthopyroxene neoblasts display lower Al₂O₃ concentrations (1.17-3.43 wt%) and increased
2 Mg-numbers (Fig. 6b); CaO abundance does not exceed 0.79 wt% (Table 4).

3 Coarse, millimeter-sized green spinels are characterized by very low Cr-numbers
4 [molar Cr/(Cr+Al)] (average Cr-number between 0.01 and 0.03), Mg-numbers between 0.61-
5 0.78, and TiO₂ lower than 0.24 wt% (Table 5). Small-sized, dark-brown spinels, associated
6 with fine-grained plagioclase-bearing aggregates, show higher Cr-numbers (0.10-0.29) and
7 TiO₂ contents (0.21-0.89 wt%), coupled to lower Mg-numbers (Fig. 7a,b).

8 Olivines have rather homogeneous compositions within each sample. Overall, olivine
9 Mg-numbers vary between 0.80-0.87, in agreement with the bulk-rock Mg-number (Table 6).
10 Plagioclase has anorthite contents [An = Ca/(Ca+Na)*100] of An₅₅₋₆₈ in the cores of
11 neoblasts; a few analyses on rims of plagioclase neoblasts gave significantly higher anorthite
12 values (An₇₄₋₈₃; Table 7). Amphibole compositions range between kaersutite and Ti-rich
13 pargasite (according to Leake classification 1997) with low K₂O (≤0.3 wt%) and Cr₂O₃
14 abundance not exceeding 1.37 wt% (Table 8).

15 *In situ* LA-ICP-MS trace element analyses have been performed mostly on
16 clinopyroxene porphyroclasts and a few representative clinopyroxene, plagioclase and
17 amphibole neoblasts (Table 9). Consistent with the bulk-rock REE variability described
18 above, clinopyroxene porphyroclasts display heterogeneous trace element composition in the
19 different samples (Fig. 8). LREE depletion (La_N/Sm_N=0.11-0.24) of clinopyroxene within a
20 single sample is comparatively uniform, similar to the bulk-rock REE composition. However,
21 the MREE to HREE are variably fractionated. Samples with the greatest bulk-rock MREE-
22 HREE fractionation (BG13, GV10, MC3, MC5, Table 2) locally show marked HREE
23 enrichment over the MREE in their clinopyroxenes (Sm_N/Yb_N=0.25-0.94; Yb_N=17.3-87.6)
24 (Fig. 8a). Clinopyroxene porphyroclasts from pyroxenites with moderate bulk-rock HREE
25 enrichment (BG4, BG5, BG14, BG22, GV17, Table 2) are similarly characterized by rather
26 variable MREE-HREE fractionation, but with less pronounced HREE enrichment
27 (Sm_N/Yb_N=0.60-1.20; Yb_N=10.2-42; Fig. 8b). Clinopyroxenes from the other pyroxenites
28 show more homogeneous REE patterns (GV8, GV12, GV14, Fig. 8c), similar to their
29 respective bulk-rock REE abundances (Fig. 5).

30 As described above, the thickest pyroxenite layer (sample GV10, Fig. 9) is
31 characterized by two mineralogical domains, namely a spinel-rich and a pyroxene-rich
32 domain. A trace element chemical profile within this single sample shows grain-to-grain (thin
33 section scale, <5 cm) REE variability in clinopyroxene porphyroclasts, related to their
34 microstructural location, i.e. in the spinel-rich versus pyroxene-rich domains (Suppl. Table 2).

1 We found that clinopyroxene porphyroclasts with the highest MREE-HREE fractionation are
2 systematically located in the spinel-rich zone, and this HREE enrichment is coupled with
3 higher Zr and Sc abundance relative to those in clinopyroxenes from the pyroxene-rich
4 domains (Fig. 9). As discussed below, this peculiar chemical feature was likely inherited by
5 metamorphic breakdown of precursor garnet in the primary assemblage.

6 Figure 8d shows the REE spectra of neoblasts from the plagioclase-bearing mineral
7 assemblage of a representative sample (BG13). Compared to clinopyroxene porphyroclasts,
8 clinopyroxene neoblasts always exhibit overall higher REE absolute concentrations and
9 development of negative Eu anomalies. Amphiboles have REE patterns similar in shape to
10 those of clinopyroxene neoblasts, but with higher absolute concentrations. Plagioclases have
11 low REE abundances and LREE enrichment ($La_N/Sm_N=3.55-5.41$), as well as marked positive
12 Eu anomalies.

14 Geothermometry

15 Conventional geothermometers, mostly based on Ca-Mg exchange between the two
16 pyroxenes, have been applied to both spinel-facies porphyroclasts and plagioclase-bearing
17 neoblasts (Table 10), in order to constrain the thermal evolution during the tectonic
18 exhumation. For the spinel-facies pyroxenes, the two-pyroxene thermometers provide similar
19 temperature ranges, between 830-940°C (Taylor, 1998) and 860-980°C ($T_{Opx-Cpx}$; Brey &
20 Kohler, 1990), which are in good agreement with temperature estimated by the single-
21 clinopyroxene thermometer of Nimis & Taylor (2000), between 820-960°C (Table 10).
22 Systematically higher temperatures are obtained using the Ca-in-Opx thermometer of Brey &
23 Kohler (1990), ranging between about 930 and 1100°C. As discussed by Nimis & Grutter
24 (2010), the Ca-in-Opx thermometer is affected by a positive bias at temperatures lower than
25 1000°C; however, we observe (Table 10) that this overestimation of calculated T persists even
26 after the empirical correction proposed by Nimis & Grutter (2010). Temperatures estimated
27 for spinel-facies pyroxenes partially overlap the temperature range documented by Borghini et
28 al. (2011) in the associated peridotites (940-1000°C), though extending to lower values. This
29 is probably due to the higher extent of recrystallization shown by the pyroxenites relative to
30 the host peridotites. The occurrence of thin (2-10 μm) exsolution in both clinopyroxene and
31 orthopyroxene porphyroclasts of the pyroxenites suggests that the spinel-bearing
32 porphyroclasts are likely affected by low-P re-equilibration. Accordingly, temperatures
33 obtained for the spinel-facies stage must be considered as minimum T estimates.

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3 1 Systematically lower temperature estimates have been calculated for the plagioclase-
4 bearing neoblasts with four different thermometers: 830-890°C and 790-860°C with the Brey
5 & Kohler (1990) thermometers (Ca-in Opx and Opx-Cpx, respectively), 770-860°C with the
6 Taylor (1998) thermometer and 780-900°C with Nimis & Taylor (2000) thermometer. As
7 observed for the spinel-facies minerals, Borghini et al. (2011) found temperatures up to 950°C
8 in the plagioclase-facies recrystallization in the associated peridotites, about 50°C higher than
9 those of the pyroxenites (Table 10).
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16 **Sr-Nd isotope compositions and Sm-Nd geochronology**

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18 Sr and Nd isotopic compositions of clinopyroxenes from pyroxenites investigated in
19 this work have been reported by Borghini et al. (2013), together with spatially associated host
20 rocks and distal peridotites, i.e. peridotite sampled in pyroxenite-free outcrops. In Figure 10a,
21 the present-day Sr and Nd isotope ratios of pyroxenites are plotted along with isotopic data
22 previously documented in orogenic and ophiolitic massifs (references reported in caption of
23 Fig. 10), and the compositional field of global MORB (Class & Lehnert, 2012: PetDB Expert
24 MORB, Mid-Ocean Ridge Basalt, Compilation. EarthChem Library.
25 <http://dx.doi.org/10.1594/IEDA/100060>) and OIB (GEOROC database: [http://georoc.mpch-
26 mainz.gwdg.de/georoc/](http://georoc.mpch-mainz.gwdg.de/georoc/)).
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33 Our pyroxenites have MORB-type Sr and Nd isotopic compositions, and when
34 compared to the data of pyroxenites from orogenic and ophiolitic massifs, they plot among
35 the samples with the lowest $^{87}\text{Sr}/^{86}\text{Sr}$ and highest $^{143}\text{Nd}/^{144}\text{Nd}$ ratios (Fig. 10a). Initial ϵNd
36 values computed at the inferred age of pyroxenite emplacement (433 ± 51 Ma; Borghini et al.,
37 2013) range between +3 and +8, consistent with enriched MORB compositions.
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41 Here, we present new Sm-Nd isotopic data on plagioclase separates and whole-rock
42 powders of selected pyroxenite samples that, together with clinopyroxene data from Borghini
43 et al. (2013), are used to construct internal isochrons. The Sm-Nd isotopic compositions of
44 whole-rocks and leached clinopyroxene and plagioclase mineral separates of selected
45 pyroxenites are listed in Table 11, together with elemental abundances of Nd and Sm
46 determined by isotope dilution. The latter are largely within 10% of the bulk-rock analyses
47 (Table 2), and most of Sm and Nd concentrations in clinopyroxenes and plagioclases obtained
48 by ID are within the variability shown by corresponding *in situ* LA-ICP-MS data (Table 9).
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53 Sm-Nd internal isochrons have been defined by whole rock, plagioclase and
54 clinopyroxene for pyroxenites BG13, BG14 and GV8, and by plagioclase and clinopyroxene
55 for sample BG5 (Fig. 11). Age calculations have been performed using Isoplot (Ludwig,
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1 2003). The strong Sm-Nd fractionation between plagioclase and clinopyroxene in the four
2 pyroxenite samples provides for well-constrained ages of 177 ± 13 , 178 ± 11 , 174 ± 25 and
3 183 ± 14 Ma. According to petrographic features, these ages date the event of partial
4 metamorphic recrystallization at low-pressure plagioclase-facies conditions, as temperature
5 estimates for plagioclase-facies neoblasts (Table 10) are largely higher than the closure
6 temperature for the Sm-Nd system (Mezger et al., 1992). Considering that the ages are
7 indistinguishable within uncertainty, and given the field and geochemical evidence of the
8 decompressional evolution, documented also in the associated host peridotites (Borghini et
9 al., 2011), we conclude that the EL pyroxenite-peridotite mantle experienced subsolidus
10 recrystallization in a single event. The age of this event is given by the weighted average of
11 the individual internal isochrons, at 178 ± 8 Ma.

12 13 **DISCUSSION**

14 **Reconstruction of the precursor mineral assemblage**

15 The pyroxenites investigated in this study contain an older spinel-bearing mineral
16 assemblage partially replaced by subsequent plagioclase-bearing neoblastic aggregates (Fig.
17 3). However, some microstructural and chemical features shown by several pyroxenite
18 samples suggest that variable modal amounts of garnet were present in the precursor mineral
19 associations. The most important evidence is that clinopyroxene porphyroclasts from spinel-
20 rich domains systematically show HREE enrichment coupled to high Zr and Sc contents (Fig.
21 9). These compositional features can be explained by the replacement of previously existing
22 garnet, which preferentially partitions HREE over the MREE. Moreover, the Zr and Sc Kd
23 values of garnet/melt are much higher than those of clinopyroxene/melt (e.g. Hauri et al.,
24 1994; Johnson, 1998, Ionov et al., 2002). Conversely, experimentally determined Kds
25 (spinel/melt) for HFSE are about an order of magnitude lower than those of both garnet and
26 clinopyroxene (Horn et al., 1994). Similar trace element characteristics of precursor garnets
27 have been previously documented by Vannucci et al. (1993) in pyroxenes of
28 spinel+plagioclase-bearing pyroxenites from Zabargad (Red Sea). They also observed that
29 this garnet imprint has been partially erased by subsequent metamorphic recrystallization at
30 lower pressure and is only evident in a few relicts. Low pressure overprinting could also
31 explain the variability of HREE-MREE fractionation of clinopyroxene porphyroclasts in
32 spinel-rich domains of our pyroxenites (Fig. 9). Zr and Sc enrichments seem to be better
33 preserved, as shown by the kink in the chemical profile of Figure 9, even though Zr and Sc
34 contents could have been higher in the early garnet and later partially smoothed by chemical

1 diffusion during subsequent spinel- and plagioclase-facies partial re-equilibration. It is
2 remarkable that Sc contents observed in clinopyroxene porphyroclasts from spinel-rich and
3 pyroxene-rich domains match very well with the range of Sc variations in garnet and its
4 equilibrium clinopyroxene, respectively, documented by Müntener et al. (2010) in garnet
5 pyroxenite from Totalp and Malenco (Fig. 9).

6 The CMAS projection in the garnet-pyroxene plane of pyroxenite bulk-rock
7 compositions is usually considered as evidence of primary crystallization of a mineral
8 assemblage dominated by garnet and pyroxene (e.g. Garrido & Bodinier, 1999; Takazawa et
9 al., 1999). Accordingly, in the CaTs-Fo-Qz triangle of Figure 12, most pyroxenites plot close
10 to the enstatite-garnet-CaTs line.

11 HREE enrichment over the MREE shown by some pyroxenite bulk-rock REE spectra
12 (samples BG13, GV10, MC3, MC5; Fig. 5) is comparable to that observed in garnet-bearing
13 pyroxenites from orogenic ultramafic massifs (e.g. Beni Bousera, Pearson et al., 1991, 1993;
14 Kumar et al., 1996; Gysi et al., 2011; Ronda, Garrido & Bodinier, 1999; Pyrenees, Bodinier et
15 al., 1987a), and has been interpreted as indicative of high-pressure magmatic segregation of
16 garnet (e.g. Kornprobst et al., 1969, 1970; Loubet & Allegre 1982). Similar bulk HREE-
17 MREE fractionation indicating precursor garnet has been documented in low-pressure
18 recrystallized pyroxenites from Zabargad and Horoman (Vannucci et al., 1995; Takazawa et
19 al., 1999).

20 In order to test the hypothesis of a precursor garnet-bearing assemblage in some of our
21 pyroxenites, we performed a mass balance calculation to derive an estimate of the original
22 modal compositions, in two separate steps. First, we calculated the modal abundances of a
23 potential garnet-bearing assemblage by using whole-rock major element data of the studied
24 pyroxenites (Table 2) and mineral compositions of garnet pyroxenites having similar bulk-
25 rock chemistry from the literature (see details below). Mass balance calculations, performed
26 by weighted least-squares minimization, included seven major element oxides (SiO₂, TiO₂,
27 Al₂O₃, Cr₂O₃, FeO, MgO, CaO and Na₂O), and were computed by using the software
28 Wolfram Mathematica 8. The quality of the fit was evaluated based on the sum of the squares
29 of the residuals (R₂ in Table 12); we considered as realistic only the computed garnet-bearing
30 assemblages with errors well below 1.

31 In the second step, we derived the hypothetical Sm_N/Yb_N ratios of bulk-rocks from the
32 Sm_N/Yb_N ratios of minerals used to calculate the primary garnet-bearing assemblage.

33 Results of the first step of the mass balance calculation are reported in Table 12,
34 together with the spinel-facies modes computed using the whole rock major element

1 abundances and the compositions of spinel-facies porphyroclasts in each pyroxenite. We
2 estimated garnet-bearing modes by adopting the mineral compositions of selected garnet
3 pyroxenite samples from the Beni Bousera (Gysi et al., 2011) and Pyrenees (Bodinier et al.,
4 1987a) ultramafic massifs, with bulk-rock compositions similar to our pyroxenites (sample
5 B7, B9 and B15 with Mg-numbers ranging from 78-85, Gysi et al., 2011; sample 70-379 Mg-
6 number = 86, Bodinier et al., 1987a; [Figure 4](#), [Suppl. Table 3](#) and [Suppl. Fig. 2](#)). Beni
7 Bousera pyroxenites cover a broader range in both whole-rock and mineral chemistry, thus
8 providing a larger choice of compositions that can better simulate our samples, particularly in
9 terms of major element contents of garnet and pyroxenes (Gysi et al., 2011). Garnet
10 websterites from the Pyrenees have minerals with Mg-numbers comparable with our samples
11 MC3 and MC5 ([Fig. 4](#) and [Suppl. Table 3](#)).

12 In addition to pyroxene, garnet and spinel compositions from literature garnet
13 pyroxenites, we used, in the mass balance calculation, the average composition of olivine
14 measured in each sample, as it strongly depends on bulk-rock Mg-number and it is not
15 expected to change composition in the different mineral assemblages within a single sample.

16 The mass balance shows that all the bulk pyroxenite compositions characterized by
17 HREE enrichment over the MREE are consistent with primary garnet-bearing assemblages,
18 with modal garnet varying between 20-40 vol% (samples BG13, GV8, GV10, GV12, MC3,
19 MC5, [Table 12](#)). Moreover, our calculations show that lower garnet modal abundances
20 (generally <17 vol%) could have been present in the precursor assemblage also for other
21 pyroxenites with nearly flat MREE-HREE spectra (samples BG4, BG5, BG14, [Table 12](#)).
22 This is, for example, documented in some garnet pyroxenites from Beni Bousera,
23 characterized by bulk-rock REE patterns lacking MREE-HREE fractionation (Gysi et al.,
24 2011).

25 The reliability of the computed garnet-bearing mineral modes has then been checked
26 by comparing calculated bulk $(\text{Sm}/\text{Yb})_N$ ratios with measured values ([Fig. 13](#)). Except for
27 sample BG14, good agreement is generally observed between the two values. Pyroxenites
28 with Mg-numbers between 74 and 84 are well modelled by major and trace element data of
29 minerals from Beni Bousera garnet pyroxenites (sample B7, B9 and B15; Gysi et al., 2011),
30 whereas pyroxenites MC3 and MC5, with slightly higher Mg-numbers (86-87), yield a better
31 fit using data from Pyrenean garnet websterites (Bodinier et al., 1987a). For samples GV14
32 and GV17 with very low SiO_2 bulk-rock abundances, we could not compute realistic garnet-
33 bearing assemblages, and their spinel-bearing modal compositions have also the highest errors
34 (R2, [Table 12](#)). In terms of the estimated REE patterns ([Suppl. Fig. 2](#)), our mass balance

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3 1 modelling is able to simulate the overall MREE-HREE fractionation, but does not reproduce
4 2 the kink between Sm and Gd. It must be noted that modelling is very sensitive to the MREE-
5 3 to-HREE compositions of garnet and pyroxenes used in the calculation. However, selective
6 4 Gd_N-Yb_N enrichment in whole-rock chemistry has been documented also in garnet pyroxenite
7 5 from the literature (e.g. Gysi et al., 2011; Marchesi et al., 2013).

8 6 We are well aware that the above mass balance calculations do not constitute absolute
9 7 proof of the occurrence of garnet in the precursor assemblage. They do, however, demonstrate
10 8 that this is a highly plausible explanation for the otherwise rather difficult-to-explain
11 9 abundance patterns of the HREE. Garnet has important implications for the chemistry of
12 10 melts formed in its presence, and for the depth of melting, and we will explore some of the
13 11 consequences of this in the remainder of this paper.

12 13 **Origin of pyroxenites by deep melt infiltration**

14 14 Garnet pyroxenites have been described in orogenic and ophiolitic ultramafic massifs
15 15 (Downes, 2007, and references therein; Montanini et al., 2006, 2012; Gysi et al., 2011,
16 16 Marchesi et al., 2013), and in mantle xenoliths (e.g. Jacob, 2004; Bizimis et al., 2005;
17 17 Gonzaga et al., 2009), and their origin has been related to a range of processes. Garnet is the
18 18 typical high-pressure metamorphic Al-rich mineral in both mantle peridotites and pyroxenites
19 19 (Herzberg, 1978; Gasparik, 1984). In addition, a garnet-bearing assemblage can result from
20 20 high-pressure crystallization of a mafic magma or by complete high-pressure metamorphic
21 21 recrystallization of mafic rocks with primary low-pressure mineral assemblages (i.e.,
22 22 “eclogites”).

23 23 Garnet pyroxenites of both origins can be present in a single ultramafic massif, as
24 24 documented in Lherz (Bodinier et al., 1987a,b), Ronda (Garrido & Bodinier, 1999) and Beni
25 25 Boussera (Kornprobst et al., 1990). In some garnet pyroxenites, whole-rock trace element
26 26 compositions with positive Eu and Sr anomalies, relatively low HREE and MREE, and high
27 27 LREE/HREE, reflect high modal abundances of primary plagioclase, and thus they have been
28 28 considered of crustal origin (e.g. Kornprobst et al., 1990; Garrido & Bodinier, 1999;
29 29 Morishita et al., 2003, Marchesi et al., 2013). In some garnet pyroxenites from Beni Boussera,
30 30 a low-P crustal imprinting is also supported by peculiar Nd-Hf-O isotopic compositions
31 31 (Pearson et al., 1993; Pearson & Nowell, 2004). Based on geochemical and isotopic data, a
32 32 similar conclusion has also been inferred for a few garnet pyroxenites from other sectors of
33 33 the External Liguride ophiolites by Montanini et al. (2006, 2012). When compared to our
34 34 pyroxenites, the latter have significantly lower bulk Mg-numbers and higher Al₂O₃ and Na₂O

1 contents (Fig. 4). Montanini et al. (2012) inferred that these garnet pyroxenites represent
2 either melt residues of precursor eclogites, or high-pressure segregation products of eclogite-
3 derived melts. Pyroxenites investigated in our study are not consistent with these origins.
4 First, they lack the trace element bulk-rock compositions indicative of plagioclase-rich
5 gabbros characterized by positive Sr and Eu anomalies, rather high LREE and very low
6 HREE, thus, ruling out an origin as recrystallization products from low-pressure magmatic
7 protoliths (Fig. 5). Second, almost flat or weakly LREE-depleted bulk REE spectra
8 ($La_N/Sm_N = 0.15-0.66$; Table 2) do not support an origin as melting residues of precursor
9 mafic rocks (Fig. 5).

10 Instead, the inferred occurrence of primary garnet in most samples (Table 12)
11 indicates that the pyroxenites likely originated by segregation of melts at rather high pressure,
12 as garnet is stable at $P > 1.5$ GPa in pyroxenitic assemblages (e.g. Kogiso et al., 2004, Downes
13 2007). Garnet-bearing assemblages have been obtained by crystallization of a hydrous
14 primitive magma in experiments at 1.2 GPa and variable T (Müntener et al., 2001). However,
15 the bulk compositions of these experimental magmatic products do not match the
16 compositions of our pyroxenites (Fig. 4), and the lack of evidence of primary amphibole in
17 the pyroxenites does not support the origin from a hydrous melt.

19 **Geochemical affinity of the parental melts**

20 In this section we discuss the nature of the parental melts to the pyroxenites, but note
21 two major limitations: i) the primary garnet-facies mineral assemblage has been completely
22 obliterated by metamorphic recrystallization events, most likely related to the tectonic
23 exhumation of the pyroxenite-peridotite mantle sequence (Borghini et al., 2011), ii) magmatic
24 interaction and/or subsolidus diffusion between pyroxenite and host peridotite have likely
25 partially modified the original chemical signatures of the melts (e.g. the Mg-number). This
26 precludes the possibility to compute equilibrium melts by using the present trace element
27 mineral composition. Therefore, the nature of pyroxenite parental melts can only be inferred
28 by combining available microstructural and chemical observations.

29 The bulk-rock Mg-numbers vary from 74-88, and are inversely correlated with the
30 thickness of layers and lenses, and positively correlated with the bulk Ni contents (Fig. 14).
31 An inverse relationship between dyke thickness and bulk Mg-number has been documented in
32 pyroxenitic layers from the Lherz ultramafic massif and explained as the result of local
33 chemical re-equilibration with the adjacent peridotite (Bodinier et al., 1987b; Fabries et al.,
34 2001). Bodinier et al. (2008) have investigated in detail this process by studying some

1 pyroxenites from Ronda (Spain). These authors performed numerical modeling demonstrating
2 that interaction between the host peridotite and variably evolved melts can generate
3 pyroxenites with Mg-numbers intermediate between those of the peridotite and melts.

4 The chemical profile along the GV10 pyroxenite (9 cm-long, Fig. 9 and Suppl. Fig. 1),
5 where data were acquired only in cores of large clinopyroxene porphyroclasts, shows a
6 significant zonation across the pyroxenite layer, with Mg-numbers of clinopyroxenes at the
7 margins of the layer close to those of clinopyroxenes in the surrounding peridotite (a complete
8 data set of this profile is provided in Suppl. Tables 1 and 2). Similar chemical gradients have
9 been described across mafic-peridotite associations from orogenic massifs (e.g. Pearson et al.,
10 1993; Takazawa et al., 1999; Mazzucchelli et al., 2010). They indicate that most of these
11 pyroxenites, having a thickness of only a few centimetres, do not reflect equilibrium with the
12 infiltrating parental melts but rather show a modified Mg-number due to the interaction with
13 the surrounding peridotite. Accordingly, the composition of clinopyroxene porphyroclasts
14 from the core of our thickest pyroxenite layer, GV10 and lens GV12, are likely to be the least
15 affected by interaction with the peridotite. We thus used their compositions to calculate the
16 Mg-number of the equilibrium parental melt, with the equation $Mg\#_{melt} = (K^{cpx}_{Fe-Mg} \cdot Mg\#_{cpx}) / (1 - Mg\#_{cpx} + Mg\#_{cpx} \cdot K^{cpx}_{Fe-Mg})$ proposed by Wood and Blundy (1997).

17 We assume that primary garnet-bearing clinopyroxenes had Mg-numbers similar to
18 those of spinel-facies clinopyroxenes, because Mg-number is not expected to be significantly
19 modified by subsequent lower-pressure recrystallization (e.g. Robinson & Wood, 1998).
20 Computed Mg-numbers of the melt vary between 52-56 for layer GV10 and 44-50 for lens
21 GV12. Melts with such values, significantly lower than the Mg-number of melts in
22 equilibrium with mantle minerals, must have been very reactive with the surrounding
23 peridotite. This is consistent with the occurrence of orthopyroxene-rich rims at the
24 pyroxenite-peridotite boundary, as well as the orthopyroxene blebs and orthopyroxene modal
25 enrichment in the wall rocks documented by Borghini et al. (2013).

26 Orthopyroxene enrichment at the pyroxenite-peridotite boundaries has been
27 documented in other ultramafic massifs (e.g. Bodinier et al., 1987a; Varfalvy et al., 1996;
28 Takazawa et al., 1999), and interpreted as the result of either solid-state reaction (e.g. Kogiso
29 et al., 2004b; Herzberg et al., 2011) or reaction between high-silica activity melts, coming
30 from the pyroxenites, and the surrounding peridotite (e.g. Varfalvy et al., 1996; Yaxley &
31 Green, 1998; Sobolev et al., 2005; Lambart et al., 2012). In the studied pyroxenites, Borghini
32 et al. (2013) have documented that clinopyroxenes in the first few (<5) centimetres of the host
33 peridotite, at the contact with the pyroxenite layer (i.e. our so-called “wall-rock” peridotite),
34

1 have been chemically modified. In particular, the cm-scale progressive REE enrichment
2 observed in clinopyroxene porphyroclasts of wall-rock peridotite points to reactive
3 percolation of the pyroxenite-derived melt into the adjacent mantle. This is also consistent
4 with the complete isotopic resetting observed in the first centimeters of the wall-rock
5 peridotite (Borghini et al., 2013).

6 Lack of evidence for precursor garnet in the host peridotites suggests that this melt-
7 rock reaction process occurred under spinel-facies mantle conditions, at pressures below
8 garnet-lherzolite stability ($P < 2.8$ GPa, e.g. O'Neill et al., 1981; Robinson & Wood, 1998),
9 but likely above 1.5 GPa, to account for the garnet-bearing primary modes reconstructed for
10 the pyroxenites. Spinel-facies clinopyroxene porphyroclasts in the wall-rock peridotites, thus,
11 retain a trace element signature acquired by the interaction with the pyroxenite-derived melts
12 (Borghini et al., 2013), and their REE composition can be used to obtain information on the
13 geochemical signature of the pyroxenite parental melts.

14 Computed melts in equilibrium with wall-rock clinopyroxene porphyroclasts in the
15 three profiles investigated by Borghini et al. (2013) are shown in Figure 15. We note here that
16 a large set of trace element data has been filtered in order to select only the clinopyroxene
17 porphyroclasts not affected by the subsequent plagioclase-bearing recrystallization, for
18 example by checking Sr-Zr elemental variations (Rampone et al., 1993). However, the effect
19 of a partial subsolidus plagioclase-facies re-equilibration would tend to slightly increase the
20 REE absolute contents in clinopyroxene, without causing significant change of the
21 LREE/HREE fractionation (Rampone et al., 1993; Hellebrand et al., 2005). Equilibrium melts
22 are characterized by enrichment of LREE over HREE ($La_N/Yb_N = 4-7.6$) and moderate
23 MREE-HREE fractionation ($Dy_N/Yb_N = 1.2-1.4$). The resulting melts have LREE
24 significantly higher than a normal MORB and, rather, they resemble the REE compositions of
25 enriched MORBs (E-type MORB) sampled in oceanic settings (Hemond et al., 1993, 2006;
26 Hoernle et al., 2011; Waters et al., 2011). This is consistent with the initial epsilonNd values
27 of clinopyroxenes in the pyroxenites, ranging from +3.29 to +8.25 (Borghini et al., 2013).

28 In summary, based on our microstructural and chemical observations, we conclude
29 that the studied pyroxenites have originated from the segregation of low-Mg-number E-
30 MORB melts. Moreover, very low alkali abundances of the bulk-rock pyroxenite
31 compositions (Fig. 4 and Table 2), suggest a tholeiitic affinity of the parental melts.

32 Borghini et al. (2013) showed that interaction of EL pyroxenites with wall-rock
33 peridotites reset the Nd isotopic compositions of the wall-rocks. They demonstrated that Sm-
34 Nd isotopic investigations on clinopyroxene from pyroxenite-wall rock peridotite pairs

1 provide linear correlations yielding ages of 424-452 Ma, which are in good agreement with
2 the age resulting from the errorchron defined by all pyroxenites (433 ± 51 Ma), thus indicating
3 the most likely timing of melt infiltration and pyroxenite emplacement. The ages indicate that
4 this event significantly predated the Mesozoic evolution of this mantle sector (discussed
5 below), and may have been related to Late Ordovician-Silurian lithosphere extension that led
6 to the opening of the PaleoTethys oceanic basin (e.g. Stampfli & Borel 2002; von Raumer &
7 Stampfli, 2008; Murphy et al., 2008).

9 **Inferences on the nature of mantle source**

10 The computed Mg-numbers of the parental melt (44-56) of the pyroxenites are too low
11 to represent a primitive basaltic melt from a peridotitic mantle source (e.g. Walter, 1998).
12 This indicates that pyroxenites originated either from i) a peridotite-derived melt that
13 experienced significant chemical evolution, or ii) a melt derived by melting of a low-Mg-
14 number hybrid source. The first hypothesis has been inferred for garnet pyroxenites from
15 xenoliths (e.g. Frey et al., 1980; Keshav et al., 2007) and orogenic massifs (e.g. Bodinier et
16 al., 1987a,b), and requires processes of liquid-crystal separation. A commonly invoked
17 mechanism of high-pressure crystal segregation and melt differentiation is the dynamic flow
18 crystallization in conduits described by Irving (1980). This model provides a possible
19 alternative to the cumulus processes requiring the settling of crystals by gravity in magma
20 chambers, a process difficult to be envisaged at pressures above 1.5 GPa within a plastically-
21 deforming peridotite. However, how hydraulic fractures can form at such depth is unknown,
22 because the deep mantle is not expected to be brittle enough to fracture (e.g. Kelemen et al.,
23 1995, 1997). Moreover, it is noteworthy that the lowest Mg-number computed for our
24 pyroxenite parental melts (around 44-50) would require more than 60% of fractional
25 crystallization of a primitive melt with a Mg-number of 70 (Villiger et al., 2004). This would
26 imply that a large amount of magmatic products should have been created in the mantle,
27 before generating the evolved melt parental to the observed pyroxenites. Although there is no
28 evidence of such large fractionation processes, we cannot exclude that this process played a
29 role in the origin of our pyroxenites.

30 An alternative hypothesis calls for the involvement of a low-MgO mafic source
31 component, as proposed for some pyroxenites from orogenic and ophiolitic massifs (e.g.
32 Garrido & Bodinier, 1999; van Acken et al., 2010; Montanini et al., 2012; Marchesi et al.,
33 2013; Montanini & Tribuzio, 2015). This means that low-Mg-number melts parental to
34 pyroxenites can be the result of partial melting of a precursor mafic/pyroxenitic component or

1 a mixed pyroxenite-peridotite source. Several experimental studies have been dedicated to the
2 melting behaviour of mafic lithologies at variably high pressure, and they have shown that a
3 wide range of melt compositions can be produced, depending on the choice of starting bulk
4 composition (e.g. Kogiso et al., 2004; Lambart et al., 2013).

5 In a review study on the composition of melts produced by pyroxenite melting at high
6 pressure (>2 GPa), Kogiso et al. (2004) suggested that silica-deficient (SD) pyroxenites can
7 be involved in the genesis of alkalic OIB lavas, whereas mixing of silica-excess (SE)
8 pyroxenite melts with MgO-rich peridotitic melts would account for the genesis of tholeiitic
9 OIB. Lack of olivine in the primary garnet-bearing assemblage of our pyroxenites (Table 12),
10 together with the occurrence of orthopyroxene-rich rims along the contact with wall-rock
11 peridotites, indicate that the observed pyroxenites have crystallized from melts with high
12 silica activity, suggesting the involvement of a silica-excess pyroxenite component (e.g.
13 Sobolev et al., 2005, 2007). Conversely, olivine-rich websterite and wehrlite are expected to
14 form by reaction between melts produced by melting of SD pyroxenites and a peridotite
15 (Lambart et al., 2012). Furthermore, the compositions of the pyroxenites in the proximity of
16 the thermal divide in the pseudo-ternary CaTs-Fo-Qz diagram (see Fig. 12), suggest that they
17 could have formed by partial melting of mantle sources with mixed eclogite-peridotite
18 lithologies (Yaxley and Green, 1998; Kogiso et al., 1998; Takahashi & Nakajima, 2002;
19 Mallik and Dasgupta, 2012; Fig. 12). Mallik and Dasgupta (2012) have recently demonstrated
20 that saturated tholeiitic melts, with Mg-numbers as low as about 56, can form through
21 reaction between eclogite-derived partial melts and a fertile peridotite. Remarkably, the
22 compositions of such melts cover the range defined by our pyroxenites in the CMAS system
23 (Fig. 12).

24 We propose that the EL pyroxenites originated from melts produced by a hybrid
25 eclogite-peridotite mantle source, further modified through reaction with the host peridotite
26 after their infiltration. The E-MORB signature inferred for the parental melts is consistent
27 with this origin, given that similar E-MORB melts from modern mid-ocean-ridge settings are
28 often interpreted to be the products of partial melting of a mantle peridotite source enriched
29 by a pyroxenitic/mafic component (e.g. Hémond et al., 2006; Waters et al., 2011; Hoernle et
30 al., 2011).

31 The major limitation of this hypothesis is that these low-Mg-number melts are highly
32 reactive in the mantle; indeed, if their migration occurs by pervasive porous flow, then the
33 compositions of these melts are expected to be rapidly buffered by reaction with the host
34 peridotite (e.g. Daines & Kohlstedt, 1994). However, the occurrence of residues of precursor

1 eclogites within the EL mantle sector, in the form of garnet pyroxenites (Montanini et al.,
2 2012), may support the scenario of a pristine eclogite-bearing mantle affected by partial
3 melting. This could also indicate that such melts, after their production, migrated only over
4 short distances, thus, preventing significant re-equilibration with the surrounding peridotite.
5 The migration of these melts could have been enhanced by stress-driven melt segregation
6 along shear zones, a mechanism able to produce high-permeability pathways within partially
7 molten ductile rocks (e.g. Holtzman et al., 2003; Havlin et al., 2013). This is consistent with
8 some field and geochemical evidence (Borghini unpublished), suggesting that the pyroxenite
9 emplacement occurred within an already plastically deformed mantle. Part of the evolved
10 signature of parental melts to the pyroxenites, such as the low Mg-numbers, could have been
11 acquired by crystal segregation and melt evolution along these high-porosity magma conduits.

12 In summary, a possible scenario is that low-Mg-number melts produced by moderate
13 degrees of SE pyroxenite/eclogite melting were mixed with low-degree peridotitic melts,
14 which infiltrated and reacted with the overlying mantle to form the Suvero suite of
15 pyroxenites layers.

16 This scenario is in agreement with recent models of basalt petrogenesis that invoke the
17 generation of secondary pyroxenites in the upper mantle by interaction between peridotites
18 and silica-rich melts derived by deeper melting of mafic components prior to major basalt
19 production with larger degrees of melting (Sobolev et al., 2005, 2007; Lambart et al., 2012).
20 According to our petrological reconstruction, the EL pyroxenites represent an excellent
21 natural example of such secondary pyroxenites. We argue that partial melting of mafic
22 components within a rising heterogeneous mantle, and the infiltration of the resulting melts,
23 reacting with the host mantle peridotites, are efficient processes to generate additional
24 heterogeneities in the upper mantle (Garrido & Bodinier, 1999; Borghini et al., 2013;
25 Marchesi et al., 2013; Montanini & Tribuzio, 2015).

26 27 **Metamorphic evolution and timing of exhumation**

28 The mass balance modelling in this work accounts for the fractionation of HREE over
29 the MREE shown by the bulk composition of many of the pyroxenites, thus supporting the
30 hypothesis that they originally contained a primary garnet-bearing mineral assemblage, which
31 is not preserved. The oldest mineral assemblage observed in the pyroxenites consists of coarse
32 porphyroclasts of green spinel, clinopyroxene \pm orthopyroxene, partially replaced by
33 plagioclase-bearing neoblasts. The occurrence of plagioclase+olivine coronas around spinel
34 porphyroclasts suggests that plagioclase did not originate directly from the breakdown of the

1 garnet-bearing assemblage. Rather, textural observations indicate that the plagioclase-bearing
2 neoblastic assemblage replaced a previous spinel-bearing association, thus suggesting that
3 plagioclase appeared at lower pressure than the spinel+pyroxenes assemblage. This is in
4 agreement with experimental results on a low-Na garnet-spinel websterite (pyroxenite R394
5 from Irving, 1974) and computed subsolidus phase relations for a low-Na, high-Mg websterite
6 from Ronda ultramafic massif (southern Spain, Hidas et al., 2013). A similar mineralogical
7 evolution has been documented in Mg-rich clinopyroxenites and websterites from other
8 External Liguride Units, and related to progressive mantle exhumation, from garnet- to spinel-
9 to plagioclase-facies conditions (Montanini et al., 2006).

10 Thermometric estimates on cores of coarse spinel-facies pyroxenes indicate
11 temperatures around 950-1000°C, which must be considered as minimum temperatures, given
12 the occurrence of exsolution textures in both pyroxenes. According to experimental results on
13 a garnet-spinel websterite, the spinel-bearing assemblage observed in the pyroxenites should
14 record a recrystallization stage within the P-T range of 1.2-1.5 GPa and $T < 1200^{\circ}\text{C}$, as this is
15 expected to be the solidus temperature at 1.5 GPa (Irving 1974).

16 Continued pressure decrease led to crystallization of a plagioclase-bearing neoblastic
17 assemblage at the expense of spinel-facies porphyroclasts. This process is accompanied by
18 systematic changes in mineral compositions, including lower Al content in pyroxenes (Fig. 6)
19 and higher Cr-number and TiO_2 content at lower Mg-number in the small relicts of spinel
20 (Fig. 7), similar to what is observed in plagioclase peridotites of metamorphic origin (e.g.
21 Gasparik, 1984; Rampone et al., 1993; Montanini et al., 2006; Borghini et al., 2010, 2011).
22 Borghini et al. (2011) previously traced the exhumation of mantle peridotite hosting the
23 pyroxenites from 0.7 to 0.3 GPa by applying thermometric estimates on neoblastic pyroxenes
24 and experimentally-derived An-isopleths of plagioclase. Therefore, we infer that associated
25 pyroxenites reasonably followed the same P-T path as the host peridotites. Average
26 compositions of plagioclase in pyroxenites (core: An = 0.58-0.64; rim: An = 0.76-0.82; Table
27 6) are very close to those documented in the associated peridotites (core: An = 0.56-0.59; rim:
28 An = 0.74-0.79; Borghini et al., 2011). This may indicate that the plagioclase composition
29 does not depend on the bulk-rock chemistry even in pyroxenite bulks, which are significantly
30 different from peridotite lithologies (Borghini et al., 2010), however, this needs to be
31 confirmed by further experimental data. As a consequence, according to the An in plagioclase
32 geobarometer proposed by Borghini et al. (2011), the similarity of An contents would
33 constitute robust evidence that plagioclase crystallized contemporaneously in both
34 pyroxenites and peridotites during their decompressional evolution.

1 Internal Sm-Nd isochrons, based on clinopyroxene-plagioclase-whole rocks analyses
2 performed on four samples, give an average age of 178 (± 8) Ma (Fig. 11) for the plagioclase-
3 facies recrystallization event, which dates the exhumation of this mantle sector at relatively
4 shallow depth. This age is slightly younger than the ages obtained by internal Sm-Nd
5 isochrons documented in garnet pyroxenites investigated by Montanini et al. (2006). They
6 proposed a Middle to Late Triassic age for the near-isothermal spinel- to plagioclase-facies
7 decompression recorded by the External Liguride mantle section. Sm-Nd ages provided by
8 this study indicate that exhumation of this EL mantle sector occurred in the Early Jurassic,
9 which is coeval with the first syn-rift MOR-type intrusions in the subcontinental mantle of the
10 Piedmont-Ligurian oceanic basin (Tribuzio et al., 2004, Rampone et al., 2014).

11 12 **CONCLUDING REMARKS**

13 Pyroxenites form centimetre-thick layers and lenses in peridotite from the External
14 Liguride ultramafic massif (Northern Apennines, Italy); they range from websterites to
15 clinopyroxenites, and are characterized by a porphyroclastic spinel-facies assemblage
16 partially replaced by a low-pressure plagioclase-bearing neoblastic association.

17 Microstructural and chemical features shown by several pyroxenite samples suggest
18 that variable modal amounts of garnet were present in the primary mineral association, thus
19 indicating that pyroxenites likely originated by segregation of melts at rather high pressure
20 (above 1.5 GPa).

21 Chemical and Sr-Nd isotopic compositions indicate that pyroxenites crystallized from
22 a E-MORB-type melt, similar to those documented in modern mid-ocean-ridge settings. Low
23 Mg-numbers together with the lack of olivine in most of the primary mineral assemblages
24 suggest that their parental melts were produced by partial melting of a hybrid pyroxenite-
25 bearing mantle, which then reacted with the host peridotite during high-pressure infiltration.
26 The EL pyroxenites studied in this work are an example of a “secondary pyroxenite”
27 component in the upper mantle, which has been proposed in recent models of basalt
28 petrogenesis at oceanic intra-plate and mid-ocean ridge settings.

29 Finally, later decompressional metamorphic evolution from spinel- to plagioclase-
30 facies followed the same P-T path defined for the host peridotites (Borghini et al., 2011).
31 Low-P plagioclase-facies re-equilibration is dated by internal (clinopyroxene-plagioclase-
32 whole rocks) Sm-Nd isochrons at 178 \pm 8 Ma, indicating that the exhumation of this mantle
33 sector can be attributed to the Mesozoic continental rifting that led to the opening of the
34 Jurassic Tethys.

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18 FIGURE CAPTIONS

19 **Figure 1.** (a) Tectonic sketch map of the Northern Apennines. Abbreviations for the tectonic
 20 units: IL, Internal Liguride Units; EL, External Liguride Units. Also reported are SV,
 21 crystalline massif of Savona, and GV, Voltri Group, belonging to the Ligurian Alps. Garnet
 22 pyroxenites investigated by Montanini et al. (2006, 2012) were sampled in outcrops located
 23 further north than the ultramafic bodies here studied, outside this map. (b) Simplified
 24 geological map showing ultramafic bodies hosting the investigated pyroxenites.

25 **Figure 2.** Centimeters-thick pyroxenite layers embedded in mantle peridotite from (a) M.te
 26 Castellaro and (b, c) Suvero ultramafic bodies (External Liguride Units, Northern Apennines,
 27 Italy). (d) A representative macroscopic sample of a pyroxenite layer, showing thin
 28 orthopyroxenite rims at the boundary with the host peridotite. White dotted lines delimit the
 29 pyroxenite layers in (b) and (d).

30 **Figure 3.** Photomicrographs of representative microstructures of EL spinel-plagioclase
 31 pyroxenites. Porphyroclasts of spinel and clinopyroxene partially replaced by fine-grained
 32 plagioclase-bearing assemblage, at parallel (a) and crossed (b) nichols. (c) Coarse
 33 orthopyroxene porphyroclast intensely substituted by the plagioclase-facies neoblastic
 34 assemblage in a spinel-bearing websterite. (d) Coarse mm-sized green spinel porphyroclast

1 surrounded by a rim of neoblastic assemblage mostly made by plagioclase, olivine and
2 brownish amphibole. (e) Contact between a pyroxenite layer (on the right of the white dotted
3 line) and orthopyroxene-rich rim (on the left); presence of large orthopyroxene porphyroclast
4 (characterized by undulatory extinction) indicates that opx-rich rim was a mineralogical
5 feature that predated the low-pressure metamorphic recrystallization. (f) Texture in pyroxenite
6 lens GV12 showing a small clinopyroxene porphyroclast surrounded by a fined-grained band
7 of clinopyroxene and dark-green spinel.

8 **Figure 4.** Variation diagrams of SiO₂, Al₂O₃, Na₂O, CaO (wt%) and Cr, Ni, Sc, V (ppm) vs
9 Mg-number for the EL pyroxenite (circles). The compositional fields reported for comparison
10 are: garnet pyroxenites (continue line) and spinel websterite (dashed line) in ultramafic
11 orogenic massifs from Beni Boussera (Morocco, Pearson et al., 1993; Kumar et al., 1996;
12 Gysi et al., 2011), Ronda (Spain, Suen & Frey, 1987; Garrido & Bodinier, 1999; Bodinier et
13 al., 2008), Horoman (Japan, Takazawa et al., 1999; Morishita & Arai, 2001), Pyrenees
14 (France, Bodinier et al., 1987a,b), and Balmuccia (Italy, Sinigoï et al., 1983; Voshage et al.,
15 1988; Mukasa & Shervais, 1999). Grey field refers to the bulk compositions of garnet
16 pyroxenite of the External Liguride Units investigated by Montanini et al. (2006). Dotted line
17 defines the range of bulk solid compositions calculated using mineral phases in melting
18 experiments on an hydrous arc melt (Müntener et al., 2001). Bulk compositions of
19 pyroxenites adopted in mass balance calculations for estimating the garnet-bearing primary
20 assemblages are also reported (see [Suppl. Table 3](#)): stars are samples from Beni Boussera
21 (grey B7, black B9; white B15; Gysi et al., 2011), the square is sample 70379 selected from
22 the Pyrenean garnet websterites (Bodinier et al., 1987a).

23 **Figure 5.** Chondrite-normalized rare earth element (REE) abundances in whole rock
24 pyroxenites. Samples BG3, BG4, BG5 and BG14 are shown with the same symbol because
25 they have very similar bulk REE compositions. Normalizing values are from Anders &
26 Grevesse (1989). Also reported for comparison are the compositional fields of External
27 Liguride garnet pyroxenites (Montanini et al., 2012), and Alpine-Apennine (A-A) gabbros
28 (Borghini et al., 2007; Rampone et al., 1998; Rampone et al., 2014).

29 **Figure 6.** Variation of Mg# vs Al (a.p.f.u.) in clinopyroxene (a) and orthopyroxene (b) from
30 pyroxenites. Filled symbols refer to spinel-facies porphyroclasts, open symbols refer to
31 plagioclase-facies neoblasts (see the legend in figure).

32 **Figure 7.** Mg# vs Cr# (a), and Cr# vs Ti x 1000 (a.p.f.u.) (b) in spinel from pyroxenites. Filled
33 symbols refer to core of large spinel-facies porphyroclasts, open symbols refer to small relicts
34 of plagioclase-facies spinel intensely replaced by the plagioclase-facies assemblage.

1 **Figure 8.** Chondrite-normalized REE patterns of clinopyroxene porphyroclasts from (a)
2 HREE-enriched pyroxenites, (b) pyroxenites with rather flat MREE-HREE spectra, (c)
3 samples GV8, GV12 and GV14 (details in text). (d) Representative chondrite-normalized
4 REE patterns of clinopyroxene spinel-facies porphyroclast, and neoblasts of clinopyroxene,
5 plagioclase and amphibole in pyroxenite BG13. Normalizing values are from Anders &
6 Grevesse (1989).

7 **Figure 9.** Sc, Zr and REE abundance in clinopyroxene porphyroclasts along profile A-B
8 analyzed in pyroxenite layer GV10, showing alternating px-rich and sp-rich zones ([Suppl.](#)
9 [Table 2](#)). Data collected by in-situ laser ablation. Normalizing values for REE are from
10 Anders & Grevesse (1989). Grey fields in the Sc diagram define the Sc content variations in
11 garnet and its equilibrium clinopyroxene documented in garnet peridotites by Müntener et al.
12 (2010).

13 **Figure 10.** Present-day $^{87}\text{Sr}/^{86}\text{Sr}$ versus $^{143}\text{Nd}/^{144}\text{Nd}$ measured on clinopyroxene separates
14 from studied pyroxenites, compared to isotope data of pyroxenites from literature: Cr-Di
15 websterites and Sp pyroxenites from Balmuccia (Ivrea Zone, Italy, Voshage et al., 1988;
16 Mukasa & Shervais, 1999); pyroxenite, websterites and gnt-pyroxenites from Beni Boussera
17 (BB, North Morocco, Pearson et al., 1993; Kumar et al., 1996); gnt-pyroxenites from (LA,
18 Lowe Austria, Becker et al., 1996); websterites from Ronda (South Spain, Marchesi et al.,
19 2012). The field of mid-ocean-ridge basalts (MORB) is from Class & Lehnert (2012); the
20 field of ocean island basalts (OIB) is from GEOROC database ([http://georoc.mpch-](http://georoc.mpch-mainz.gwdg.de/georoc/)
21 [mainz.gwdg.de/georoc/](http://georoc.mpch-mainz.gwdg.de/georoc/)).

22 **Figure 11.** (a) Present-day $^{143}\text{Nd}/^{144}\text{Nd}$ versus $^{147}\text{Sm}/^{144}\text{Nd}$ in clinopyroxenes from
23 pyroxenites that define an errochron (dashed line) giving an age of 433 ± 51 Ma. Empty
24 symbols refer to spatially controlled pyroxenite – wall-rock profiles, and relative isochrones
25 (continuous lines), reported in Borghini et al. (2013). Data of EL garnet pyroxenites from
26 Montanini et al. (2012) are reported for comparison. (b) Sm-Nd internal isochrons defined by
27 whole rock (wr), plagioclase (plag) and clinopyroxene (cpx) from four pyroxenite samples
28 (BG13, BG14, GV8 and BG5, this latter only based on clinopyroxene and plagioclase
29 analyses). Ages and initial Nd isotopic ratios for the internal isochrons are: 178 ± 11 Ma, initial
30 $^{143}\text{Nd}/^{144}\text{Nd} = 0.512961 (\pm 24)$, MSWD = 0.22 for BG13; 177 ± 13 Ma, initial $^{143}\text{Nd}/^{144}\text{Nd} =$
31 $0.512837 (\pm 20)$, MSWD = 0.69 for BG14; 174 ± 25 Ma, initial $^{143}\text{Nd}/^{144}\text{Nd} = 0.512727 (\pm 34)$,
32 MSWD = 0.0024 for GV8; 183 ± 14 Ma, initial $^{143}\text{Nd}/^{144}\text{Nd} = 0.512827 (\pm 17)$ for BG5.

33 **Figure 12.** Whole-rock normative compositions of studied pyroxenites projected from
34 Diopside within the (CMAS) triangle forsterite (Fo) - calcium tschermak (CaTs) - quartz (Qz)

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(O'Hara, 1968). The compositional fields of EL garnet pyroxenites (EL gnt-pyrox, Montanini et al., 2012), Type-1 pyroxenites from Horoman (Takazawa et al., 1999), and Beni Boussera garnet pyroxenites (BB-gnt-pyrox, Gysi et al., 2011), are also shown for comparison. In the smaller CMAS triangle to the right, pyroxenite compositions are plotted together with the compositions of experimental partial melts from hybrid mafic-peridotite sources (Yaxley & Green, 1998; Kogiso et al., 1998; Takahashi & Nakajima, 2002), and the compositional field of melts in the eclogite-peridotite reaction experiments by Mallik & Dasgupta (2012)(tick-marks on the black arrow indicate the amount in wt% of eclogitic melt added to peridotite).

Figure 13. Bulk-rock Sm_N/Yb_N resulting from mass balance calculations compared to the Sm_N/Yb_N values measured in each pyroxenite sample (detailed explanation in the text). Open circles are the values calculated by using mineral compositions from Beni Boussera garnet pyroxenites (Gysi et al., 2011), open squares are obtained adopting in the calculation mineral compositions from Pyrenees garnet pyroxenites (Bodinier et al., 1987a).

Figure 14. Variation of bulk-rock Mg-number versus thickness (cm) of the studied pyroxenite layers and lens. Close to the symbols are also shown the Ni contents (ppm) of each pyroxenite sample.

Figure 15. (a) Chondrite-normalized REE patterns of clinopyroxene porphyroclasts in wall-rock peridotites from three profiles investigated by Borghini et al. (2013). (b) Computed melts in equilibrium with wall-rock clinopyroxenes shown in (a), using partition coefficients from Ionov et al. (2002). Also reported are the composition of average N-MORB (Hofmann, 1988), and average E-MORB from Mid-Atlantic Ridge 10-24°N (Hémond et al., 2006). The compositional fields refer to: 1) E-MORBs from South Atlantic Mid-Ocean-Ridge 5-11°S (A3 Seamount; Hoernle et al., 2011), 2) tholeiitic lavas (both olivine- and quartz-normative) from Iceland plume (Icelandic tholeiites; Hémond et al., 1993), 3) East Pacific Rise 9-10°N (Waters et al., 2011). Normalizing values are from Anders & Grevesse (1989).

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3 1 Pyroxenite Layers in the Northern Apennines' Upper Mantle (Italy) –
4 2 Generation by Pyroxenite Melting and Melt-infiltration
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9 6 **GIULIO BORGHINI^{1,2,3}, ELISABETTA RAMPONE², ALBERTO ZANETTI⁴,**
10 7 **CORNELIA CLASS³, ANNA CIPRIANI^{3,5}, ALBRECHT W. HOFMANN^{3,6}, STEVEN**
11 8 **L. GOLDSTEIN^{3,7}**
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14
15
16 12 ¹DIPARTIMENTO DI SCIENZE DELLA TERRA, UNIVERSITÀ DI MILANO, VIA BOTTICELLI 23, 20133
17 13 MILANO, ITALY

18 14 ²DIPARTIMENTO DI SCIENZE DELLA TERRA, DELL'AMBIENTE E DELLA VITA, UNIVERSITÀ DI
19 15 GENOVA, CORSO EUROPA 26, 16132 GENOVA, ITALY

20 16 ³LAMONT-DOHERTY EARTH OBSERVATORY OF COLUMBIA UNIVERSITY, 61 ROUTE 9W,
21 17 PALISADES, NY 10964, USA

22 18 ⁴CNR-ISTITUTO DI GEOSCIENZE E GEORISORSE, SEZIONE DI PAVIA, VIA FERRATA 1, I-27100
23 19 PAVIA, ITALY

24 20 ⁵DIPARTIMENTO DI SCIENZE CHIMICHE E GEOLOGICHE, UNIVERSITÀ DEGLI STUDI DI MODENA E
25 21 REGGIO EMILIA, LARGO S. EUFEMIA 19, 41100 MODENA, ITALY

26 22 ⁶MAX PLANCK INSTITUTE FOR CHEMISTRY, P.O. BOX 3060, 55020 MAINZ, GERMANY

27 23 ⁷DEPARTMENT OF EARTH AND ENVIRONMENTAL SCIENCES OF COLUMBIA UNIVERSITY,
28 24 LAMONT-DOHERTY EARTH OBSERVATORY, 61 ROUTE 9W, PALISADES, NY 10964, USA
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35 29 **Corresponding Author:**

37 30 **Giulio Borghini**

38 31 Dipartimento di Scienze della Terra "Ardito Desio"
39 32 Università degli Studi di Milano
40 33 Via Botticelli 23
41 34 20133 Milano (Italy)

44 36 Tel. 0039 02 50315600

45 37 Fax. 0039 02 50314597

46 38 Email: giulio.borghini@unimi.it
47 39
48 40
49 41
50 42
51 43

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Abstract

Pyroxenite layers embedded within peridotite represent wide-spread lithological mantle heterogeneities and are potential components in the mantle source of many oceanic basalts. Pyroxenites can be generated by several magmatic and metamorphic processes. However, in most natural samples (especially in ultramafic massifs), their primary characteristics are partially or completely erased by later petrologic evolution (e.g. metamorphism, metasomatism or partial melting). Here we investigate a suite of pyroxenites from the External Liguride Jurassic ophiolites (Northern Apennines, Italy). They are spinel-bearing websterites and clinopyroxenites, partially recrystallized under plagioclase-facies conditions, and occur as cm-scale layers parallel to the tectonite foliation of their host peridotites. Pyroxenites have bulk Mg-numbers from 74 to 88 and display rather constant LREE depletion over the MREE ($La_N/Sm_N = 0.15-0.35$), but variable MREE-HREE fractionation, with some of them having markedly positive HREE slopes ($Sm_N/Yb_N=0.30-0.96$). The HREE enrichment coupled with high Zr and Sc contents in clinopyroxene porphyroclasts from spinel-bearing domains provides strong evidence that garnet was present in the precursor mineral associations. Mass balance calculations suggest that the pyroxenites originally contained up to about 40 vol% of garnet, indicating they originated by segregation of melts at rather high pressure ($P > 1.5$ GPa). Parental melts of pyroxenites have reacted to some extent with the host peridotite during mantle infiltration. Lack of olivine in the primary mineral assemblage and orthopyroxene-rich rims along the contact with wall-rock peridotites indicate that pyroxenites have crystallized from silica-rich melts. These, likely, had REE patterns and Sr-Nd isotope compositions similar to enriched MORB. We propose that the pyroxenites originated from melts derived from a hybrid eclogite-bearing peridotite source, and then reacted with the host peridotite to form “secondary pyroxenites”. Their existence has been invoked in current models of basalts petrogenesis. During later decompression, they experienced an intermediate recrystallization at spinel-facies conditions, at 1.2-1.5 GPa and minimum temperature of 950-1000°C, and partial re-equilibration at low-pressure plagioclase-facies. The latter is dated by internal Sm-Nd isochrons at 178 (± 8) Ma and is associated with Mesozoic exhumation, during extension of the Tethys lithosphere.

1 INTRODUCTION

2 Geochemical studies on oceanic basalts have documented that their upper mantle
3 sources are chemically and isotopically heterogeneous (e.g. Hofmann, 2003). The nature,
4 distribution and scale of such mantle heterogeneities are of fundamental importance for the
5 understanding of the dynamics and long-term evolution of the upper mantle. Direct
6 observations on mantle rocks, such as mantle xenoliths, abyssal peridotites and tectonically
7 exhumed mantle sectors (Alpine-type peridotite massifs), have indicated that
8 pyroxenite/mafic layers, commonly embedded within peridotite, represent wide-spread
9 lithological mantle heterogeneities (e.g. Bodinier & Godard, 2003; Downes, 2007). Whether
10 or not pyroxenite components occur in the upper mantle has been hotly debated during the
11 last decades, mostly because of their potential involvement in mantle partial melting and
12 generation of basalt (e.g. Hirschmann and Stolper 1996; Stracke et al., 1999; Salters & Dick,
13 2002; Kogiso et al., 2004a,b; Sobolev et al., 2005; Lambart et al., 2013).

14 Experimental studies (e.g. Yaxley & Green, 1998; Hirschmann et al., 2003;
15 Pertermann & Hirschmann, 2003; Kogiso et al., 2004a; Sobolev et al., 2007; Lambart et al.,
16 2009, 2012) have shown that the partial melting of mafic components, occurring at mantle
17 depths as pyroxenites and eclogites, can produce highly variable melt compositions and,
18 together with melts from depleted peridotite, reproduce the wide compositional range
19 documented in oceanic basalts. On this basis, geochemical models have highlighted the role
20 of a mixed pyroxenite-peridotite mantle source in the genesis of mid-ocean ridge (MORB)
21 and ocean island (OIB) basalts (e.g. Allegre & Turcotte, 1986; Hirschmann & Stolper, 1996;
22 Kogiso et al., 2004b; Sobolev et al., 2005; Stracke & Bourdon, 2009; Herzberg, 2011; Mallik
23 & Dasgupta, 2012; Lambart et al., 2012).

24 The nature of mantle pyroxenites, however, remains a controversial subject, mostly
25 because their origin has been related to a variety of magmatic and metamorphic processes,
26 including: i) recycling of subducted oceanic crust stirred by mantle convection and
27 incorporated into the lithosphere (e.g. Allègre & Turcotte, 1986; Kornprobst et al., 1990;
28 Morishita & Arai, 2001; Morishita et al., 2003, Yu et al., 2010), ii) moderate- to high-pressure
29 crystal segregation from magmas derived by melting of asthenospheric sources (Bodinier et
30 al., 1987a,b; Vannucci et al., 1993; Rivalenti et al., 1995; Kempton et al., 1997; Takazawa et
31 al., 1999; Mukasa & Shervais, 1999; Keshav et al., 2007; Dantas et al., 2007; Warren et al.,
32 2009; Gysi et al., 2011) or subducted oceanic crust (Davies et al., 1993; Pearson et al., 1993;
33 Becker, 1996; Pearson & Nowell, 2004), iii) refertilization of 'infertile' or 'depleted' upper
34 mantle via melt-peridotite reactions (Garrido & Bodinier, 1999; Bodinier et al., 2008; Dantas

1 et al., 2009, van Acken et al., 2010). An additional difficulty in characterizing the role of
2 pyroxenites in melting is associated with their reactivity within the mantle, causing them to be
3 easily modified by magmatic and metamorphic processes subsequent to their emplacement at
4 depth (e.g. Takazawa et al., 1999; Morishita & Arai, 2001; Morishita et al., 2003), interaction
5 with transient melts (e.g. Garrido & Bodinier, 1999; Chazot et al., 2005; Kaeser et al., 2009),
6 and decompressional partial melting (e.g. Pearson et al., 1993; Blichert-Toft et al., 1999;
7 Montanini et al., 2012).

8 Orogenic and ophiolitic massifs offer the possibility to observe structural, spatial and
9 geochemical relations between pyroxenites and host peridotites at different stages of their
10 evolution (Bodinier & Godard, 2003). They are also important as the only locations with the
11 possibility to gain insights on the distribution and extent of small-scale heterogeneities in the
12 upper mantle through direct field observations. In this paper, we present petrographic,
13 chemical and Sm-Nd isotope data on spinel pyroxenite samples from three, hundred-meter-
14 sized ultramafic bodies from the western sector of the External Liguride Jurassic ophiolites
15 (Northern Apennines, Italy; Fig. 1). Our microstructural and geochemical combined approach
16 allows us to infer primary characteristics of pyroxenites, conditions and timing of their
17 emplacement, geochemical affinities of their parental melts, and the later petrologic evolution.
18 Our results impact the fundamental question of how pyroxenite components originate in the
19 upper mantle, and provide further evidence of a “secondary-type” pyroxenite (also called
20 “2stage” pyroxenite, Sobolev et al., 2005, 2007; Lambart et al., 2013) emplacement at rather
21 deep mantle levels.

22 23 **GEOLOGICAL BACKGROUND**

24 Alpine-Apennine ophiolites represent lithospheric remnants of the Piemont-Ligurian
25 ocean basin, a branch of the Mesozoic Tethys separating the European and Adriatic passive
26 margins (Rampone & Hofmann, 2012, and references therein). The Middle Jurassic opening
27 of the basin led to uplift and denudation of the subcontinental lithospheric mantle, the
28 dominant lithotype exposed in Alpine-Apennine ophiolite sequences. Most of these mantle
29 bodies have been affected by thermochemical erosion and refertilization processes, related to
30 asthenospheric upwelling during rifting and oceanization (e.g. Rampone et al., 1997, 2004,
31 2008; Müntener et al., 2004; Piccardo et al., 2007). However, some of the mantle sequences
32 from the Northern Apennines, namely the ultramafic massifs from the External Liguride Unit,
33 have escaped these melt-rock interaction processes and preserved their old subcontinental
34 signature (Beccaluva et al., 1984; Rampone et al., 1995; Montanini et al., 2006; Borghini et

1 al., 2011). The External Liguride (EL) ophiolites consist of large slide-blocks (up to km-scale)
2 of ultramafic rocks, and minor MORB-type gabbros and basalts, within Cretaceous
3 sedimentary melanges that were obducted during the closure of the Ligurian Tethys oceanic
4 basin (Rampone et al., 1993, 1995; Tribuzio et al., 2004; Montanini et al., 2008). These
5 mantle sequences consist mostly of fertile spinel-plagioclase lherzolites with disseminated Ti-
6 rich amphibole and widespread spinel pyroxenite layers (Beccaluva et al., 1984). Lherzolites
7 locally retain extremely “depleted” Nd-Sr isotopic compositions (Rampone et al., 1995)
8 coupled with ancient Os model ages (1.6 Ga, Snow et al., 2000), which have been interpreted
9 to reflect long residence times in the subcontinental mantle. The External Liguride ophiolitic
10 mantle sequences have therefore been inferred to represent remnants of a fossil ocean-
11 continent transition, where exhumed subcontinental mantle was associated with embryonic
12 oceanic crust and rocks of continental origin (Marroni et al., 1998; Tribuzio et al., 2004;
13 Montanini et al., 2008), an analogue to the modern Iberian passive margin (e.g. Whitmarsh et
14 al., 2001).

15 Garnet pyroxenite layers from the northern sector of the External Liguride Unit have
16 been recently investigated by Montanini et al. (2006, 2012) and consist of thin websterites and
17 meter-scale thick garnet clinopyroxenites recording equilibration in the subcontinental
18 lithosphere at $T = 1100^{\circ}\text{C}$ and $P = 2.8 \text{ GPa}$. Sm-Nd and Lu-Hf geochronology on garnet-
19 plagioclase pairs yielded $186 \pm 2 \text{ Ma}$ and $220 \pm 13 \text{ Ma}$ ages, interpreted as cooling ages that
20 reflect the rifting-related exhumation of these mantle sectors (Montanini et al., 2006, 2012).
21 Pyroxenites investigated in this study are from the southern sector of the Western External
22 Liguride Unit (Suvero, La Spezia; Fig. 1). Borghini et al. (2011) recently showed that the
23 peridotites from this mantle sector experienced a rather cold spinel- to plagioclase-facies
24 decompressional tectonic evolution ($T=870\text{-}930^{\circ}\text{C}$), thus confirming that this pyroxenite-
25 peridotite association escaped melt-rock interaction and re-fertilization processes widely
26 documented in most Alpine-Apennine ophiolitic peridotites (e.g. Müntener & Hermann,
27 1996; Rampone et al., 1997, 2004, 2008; Rampone & Borghini, 2008; Piccardo et al., 2004,
28 2007).

30 **FIELD OCCURRENCE AND PETROGRAPHY**

31 A summary of the location and description of samples is presented in Table 1.
32 Pyroxenites occur as layers within the host mantle peridotites, mostly ranging in thickness
33 from several millimetres to about 15 centimetres (Fig. 2). Samples GV12 and GV14 occur as
34 lenses about 30 cm thick. Layers are parallel to the foliation plane of the host peridotites,

1 showing sharp contacts, and can be weakly boudinaged (Fig. 2d). In outcrop, pyroxenite
2 layers are recognised by their pinkish or dark-grey colour (Fig. 2a,b,c) and by the occurrence
3 of coarse black spinel grains. Pyroxenite abundance is variable, and in some outcrops can be
4 as high as 50% (Fig. 2a,b). In some places, pyroxenite layers alternate with harzburgite
5 portions within the peridotite, thus forming a compositional layering parallel to the tectonite
6 foliation (Fig. 2a,d).

7 Most of the pyroxenites are characterized by a porphyroclastic texture defined by
8 coarse-grained (up to 10-12 mm) porphyroclasts of pyroxenes and spinels, stretched along the
9 tectonite foliation. Porphyroclastic minerals are partially substituted by a plagioclase-bearing
10 granoblastic assemblage (Fig. 3a,b,c,d), indicating a retrograde evolution from spinel- to
11 plagioclase-facies. The low-P metamorphic recrystallization is recorded by the development
12 of i) millimeter and sub-millimeter bands of plagioclase and olivine (\pm amphibole) crystallized
13 around coarse porphyroclasts of greenish spinels and pyroxenes (Fig. 3a,b), ii) orthopyroxene
14 plus plagioclase exsolutions in large clinopyroxene porphyroclasts (Fig. 3b), iii) fine-grained
15 (\sim 200–300 μ m) granoblastic aggregates made of plagioclase, olivine, newly formed
16 pyroxenes and occasionally dark-brown spinel and brownish amphibole, partially replacing
17 the spinel-facies pyroxene porphyroclasts (Fig. 3c,d). Amphibole is heterogeneously
18 distributed in the pyroxenite samples and is always associated with the plagioclase-bearing
19 neoblastic assemblage (Fig. 3d). In the most altered samples (GV14 and GV17), granoblastic
20 olivine and plagioclase are almost completely substituted by low-T alteration products
21 (chlorite and saussurite, respectively).

22 Most of the pyroxenite samples are characterized by two alternating mineralogical
23 domains: i) spinel-rich portions, where green spinel relics are rimmed by plagioclase and
24 olivine (\pm amphibole), ii) pyroxene-bearing domains in which the large pyroxene
25 porphyroclasts are partially substituted by neoblastic aggregates of new pyroxenes and
26 plagioclase. The latter mineralogical feature is clearly observable in the thickest pyroxenite
27 layers, whereas the two domains are usually stretched and boudinaged in the thinner bands
28 (discussed further below). Most of the pyroxenite bands show irregular orthopyroxene-rich
29 borders along the contact with the hosting peridotite (Fig. 3e). This is the result of interaction
30 between the melt from which the pyroxenite crystallized and the host mantle peridotite during
31 the pyroxenite emplacement (Borghini et al., 2013).

32 Olivine and amphibole are totally absent in pyroxenite GV12; this sample is texturally
33 distinct and characterized by few small relics of clinopyroxene porphyroclasts that are
34 strongly substituted by a fine-grained equigranular metamorphic texture consisting of

1 clinopyroxene, dark-green spinel and plagioclase (completely replaced by low-T alteration
2 minerals, Fig. 3f). This granoblastic assemblage often forms clusters and seams stretched
3 along the foliation (Fig. 3f).

4 5 **ANALYTICAL METHODS**

6 Whole-rock major and trace element analyses of pyroxenites were made by lithium
7 metaborate/tetraborate fusion ICP-OES (for major elements) and ICP/MS (for trace elements)
8 at the Actlabs Laboratories (Western Ontario, Canada; <http://www.actlabs.com>).

9 Mineral major element compositions were analyzed using: (1) a Philips SEM 515
10 equipped with an X-ray dispersive analyzer (accelerating potential 15 kV, beam current 20
11 nA), at the Dipartimento di Scienze della Terra, dell'Ambiente e della Vita, University of
12 Genova, and (2) a JEOL JXA 8200 Superprobe equipped with five wavelength-dispersive
13 (WDS) spectrometers, an energy dispersive (EDS) spectrometer, and a cathodoluminescence
14 detector (accelerating potential 15 kV, beam current 15 nA), at the Dipartimento di Scienze
15 della Terra "Ardito Desio", University of Milano. *In situ* trace element mineral analyses were
16 carried out by laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) at
17 IGG-CNR in Pavia. A detailed description of analytical procedures is given in Miller et al.
18 (2007).

19 Sm and Nd isotopic compositions were measured on clinopyroxene and plagioclase
20 separates and on whole rock powders from pyroxenite samples. Clinopyroxenes and
21 plagioclases (90-150 μm sized) were separated by grinding, sieving, electromagnetic
22 separation and handpicking under a microscope. Weights of unleached mineral concentrates
23 were 25-82 mg for clinopyroxenes and 54-132 mg for plagioclases. Minerals were leached in
24 at least three steps using a solution of 5% HF + 6.2M HCl. Nd isotopic compositions, together
25 with elemental abundances of Sm and Nd by isotope dilution, were measured by thermal
26 ionization mass spectrometry (TIMS) on a VG Sector 54-30 at the Lamont-Doherty Earth
27 Observatory of Columbia University (Palisades, NY, USA). The $^{143}\text{Nd}/^{144}\text{Nd}$ ratios were
28 normalized to $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$, and reported relative to JNdi $^{143}\text{Nd}/^{144}\text{Nd} = 0.512115$
29 (Tanaka et al., 2000). Samples were analyzed during the course of 6 analytical sessions.
30 Individual sample data were corrected to the standard values measured during the individual
31 time intervals. Analyses of the JNdi-1 standard during the 6 sessions yielded $^{143}\text{Nd}/^{144}\text{Nd}$
32 ratios of 0.512080 ± 9 (2σ , mean of $n=7$ analyses, 34 ppm external reproducibility),
33 0.512074 ± 10 ($n=10$, 18 ppm), 0.512075 ± 12 ($n=13$, 23 ppm), 0.512088 ± 14 ($n=5$, 27 ppm),

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3 1 0.512096±13 (n=26, 25 ppm), 0.512095±10 (n=5, 20 ppm). Total Sr and Nd blanks did not
4 exceed 150 pg.
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8 **RESULTS**

9 **Bulk-rock chemistry**

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11 The bulk-rock major and trace element data of the pyroxenite samples (Table 2)
12 display a large total range of Mg-numbers [molar $\text{Mg}/(\text{Mg}+\text{Fe}^{2+}_{\text{tot}})*100$] (74-88 mol%),
13 although most values are concentrated between 83-85. Conversely, pyroxenites cover a large
14 range of Al_2O_3 and CaO contents (10.6-16.5 wt% and 6.8-19.5 wt%, respectively; Fig. 4),
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18 mostly reflecting their modal variation between clinopyroxenite and websterite assemblages.
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20 In Figure 4, selected major and trace elements are plotted against Mg-number, together with
21 the compositional fields of garnet pyroxenites and spinel-bearing websterites from orogenic
22 and ophiolitic massifs (details and references in Fig. 4 caption).
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25 Major element compositions of most samples lie in the fields of overlap between
26 garnet-pyroxenites and spinel-bearing websterites from orogenic massifs at intermediate Mg-
27 numbers (Fig. 4). However, they diverge from literature data by having generally higher CaO
28 and lower Na_2O contents. Samples GV14 and GV17 have unusual, very low SiO_2 abundances
29 (Table 2). Petrographic observations indicate that this is likely due to a high extent of low-
30 temperature hydrothermal alteration, mostly dominated by chloritization of pyroxenes and
31 olivine, as observed for many oceanic and ophiolitic mafic rocks (e.g. Alt et al., 1998).
32
33 Pyroxenite lens GV12 plots within the compositional field of garnet pyroxenites, but with
34 slightly higher CaO content (Fig. 4). Garnet-pyroxenites from other EL ultramafic massifs,
35 recently studied by Montanini et al. (2012), show lower Mg-numbers and higher Na_2O , Al_2O_3
36 contents relative to our samples, but similar SiO_2 and CaO compositional variations (Fig. 4).
37
38 In terms of Cr, Ni, V, Sc abundances, the pyroxenites also lie in the compositional field of
39 overlap between garnet pyroxenites and websterites in the literature, except for samples GV12
40 and GV14, which plot within the compositional field of garnet-pyroxenites, at lower Mg
41 values (Fig. 4). Our pyroxenites have low Cr, Ni (560-1530 ppm, 210-860 ppm, respectively)
42 and moderate Sc, V (31-51 ppm, 163-280 ppm, respectively) concentrations. EL garnet-
43 pyroxenites studied by Montanini et al. (2012) show similar values, at lower Mg-numbers
44 (Fig. 4).
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47 The CI-normalized REE and other refractory elements (Fig. 5) of most of the
48 pyroxenites (BG3, BG4, BG5, BG13, BG14, BG22, GV10, GV17, MC3, MC5) display rather
49 constant LREE depletion over the MREE ($\text{La}_N/\text{Sm}_N = 0.15-0.35$), but variable MREE-HREE
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3 1 fractionation. In fact, the HREE slopes range from flat (GV17) to markedly positive (BG13)
4 (Sm_N/Yb_N=0.30-0.96), with absolute concentrations varying from 3 to 20 x CI. In particular,
5 2 samples BG13, GV10, MC3 and MC5 display HREE enriched patterns as observed in some
6 3 garnet pyroxenites (e.g. Pearson et al., 1993; Gysi et al., 2011; Montanini et al., 2012). Spinel-
7 4 rich clinopyroxenite GV12 has an overall lower REE content, showing a LREE-depleted
8 5 pattern combined with slightly fractionated MREE to HREE (Sm_N/Yb_N=0.77). The
9 6 pyroxenite GV14 shows a convex-upward pattern with HREE depletion over MREE
10 7 (Sm_N/Yb_N=1.66). Finally, pyroxenite GV8 displays a nearly flat pattern at 3-4 x CI. Zr and Hf
11 8 show a rather flat trend between Nd and Sm, with some small positive Zr anomalies. In
12 9 contrast, Sr is highly enriched compared to the LREE. These “spikes” probably derived from
13 10 later fluid-driven contamination as evidenced by very thin whitish veins orthogonal to the
14 11 layer direction not completely eliminated from the samples during powder preparation for
15 12 bulk-rock analysis. This is further supported by the absence of positive Sr anomalies in any of
16 13 the porphyroclasts. Our samples have MREE-HREE abundances within the compositional
17 14 field defined by the External Liguride garnet-pyroxenites studied by Montanini et al. (2012)
18 15 (Fig. 5). However, they have higher LREE content, with the exception of sample GV12,
19 16 which is consistent with literature data.
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32 19 **Mineral chemistry**

33 20 Pyroxenes and spinel have Mg-numbers varying according to the Mg-number of their
34 21 bulk-rock, and within each sample they show systematic major element variations according
35 22 to the different mineral generations (porphyroblast vs. neoblast). Clinopyroxene
36 23 porphyroclasts have Mg-numbers varying in agreement with their bulk-rock Mg-number,
37 24 from 0.77 in the olivine-free clinopyroxenite GV12, to 0.88 in the high Mg-number
38 25 websterites MC3 and MC5 (Table 3 and Fig. 6a). They also have rather high Al₂O₃ contents
39 26 (6.21-10.12 wt%) and low Cr₂O₃ and Na₂O abundances (≤ 0.36 wt% and ≤ 1.0 wt%
40 27 respectively; Table 3). A chemical profile on a representative pyroxenite layer shows that
41 28 clinopyroxene porphyroclasts have Mg-numbers increasing from the center of the pyroxenite
42 29 layer towards its margins (Suppl. Fig.1, and Suppl. Table 1), where Mg-numbers approach
43 30 those of clinopyroxene of the host peridotite. As shown in Figure 6a, clinopyroxene neoblasts
44 31 in each sample exhibit systematically lower Al₂O₃ concentrations (3.16-6.62 wt%) and higher
45 32 Mg-numbers relative to corresponding clinopyroxene porphyroclasts. In the orthopyroxene
46 33 porphyroclasts, the Mg-number varies from 0.82 to 0.88, and Al₂O₃ and CaO contents range
47 34 between 4.33-7.56 wt% and 0.52-1.19 wt%, respectively (Table 4). Similar to clinopyroxenes,
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1 orthopyroxene neoblasts display lower Al₂O₃ concentrations (1.17-3.43 wt%) and increased
2 Mg-numbers (Fig. 6b); CaO abundance does not exceed 0.79 wt% (Table 4).

3 Coarse, millimeter-sized green spinels are characterized by very low Cr-numbers
4 [molar Cr/(Cr+Al)] (average Cr-number between 0.01 and 0.03), Mg-numbers between 0.61-
5 0.78, and TiO₂ lower than 0.24 wt% (Table 5). Small-sized, dark-brown spinels, associated
6 with fine-grained plagioclase-bearing aggregates, show higher Cr-numbers (0.10-0.29) and
7 TiO₂ contents (0.21-0.89 wt%), coupled to lower Mg-numbers (Fig. 7a,b).

8 Olivines have rather homogeneous compositions within each sample. Overall, olivine
9 Mg-numbers vary between 0.80-0.87, in agreement with the bulk-rock Mg-number (Table 6).
10 Plagioclase has anorthite contents [$An = Ca/(Ca+Na)*100$] of An₅₅₋₆₈ in the cores of
11 neoblasts; a few analyses on rims of plagioclase neoblasts gave significantly higher anorthite
12 values (An₇₄₋₈₃; Table 7). Amphibole compositions range between kaersutite and Ti-rich
13 pargasite (according to Leake classification 1997) with low K₂O (≤ 0.3 wt%) and Cr₂O₃
14 abundance not exceeding 1.37 wt% (Table 8).

15 *In situ* LA-ICP-MS trace element analyses have been performed mostly on
16 clinopyroxene porphyroclasts and a few representative clinopyroxene, plagioclase and
17 amphibole neoblasts (Table 9). Consistent with the bulk-rock REE variability described
18 above, clinopyroxene porphyroclasts display heterogeneous trace element composition in the
19 different samples (Fig. 8). LREE depletion ($La_N/Sm_N=0.11-0.24$) of clinopyroxene within a
20 single sample is comparatively uniform, similar to the bulk-rock REE composition. However,
21 the MREE to HREE are variably fractionated. Samples with the greatest bulk-rock MREE-
22 HREE fractionation (BG13, GV10, MC3, MC5, Table 2) locally show marked HREE
23 enrichment over the MREE in their clinopyroxenes ($Sm_N/Yb_N=0.25-0.94$; $Yb_N=17.3-87.6$)
24 (Fig. 8a). Clinopyroxene porphyroclasts from pyroxenites with moderate bulk-rock HREE
25 enrichment (BG4, BG5, BG14, BG22, GV17, Table 2) are similarly characterized by rather
26 variable MREE-HREE fractionation, but with less pronounced HREE enrichment
27 ($Sm_N/Yb_N=0.60-1.20$; $Yb_N=10.2-42$; Fig. 8b). Clinopyroxenes from the other pyroxenites
28 show more homogeneous REE patterns (GV8, GV12, GV14, Fig. 8c), similar to their
29 respective bulk-rock REE abundances (Fig. 5).

30 As described above, the thickest pyroxenite layer (sample GV10, Fig. 9) is
31 characterized by two mineralogical domains, namely a spinel-rich and a pyroxene-rich
32 domain. A trace element chemical profile within this single sample shows grain-to-grain (thin
33 section scale, <5 cm) REE variability in clinopyroxene porphyroclasts, related to their
34 microstructural location, i.e. in the spinel-rich versus pyroxene-rich domains (Suppl. Table 2).

1 We found that clinopyroxene porphyroclasts with the highest MREE-HREE fractionation are
2 systematically located in the spinel-rich zone, and this HREE enrichment is coupled with
3 higher Zr and Sc abundance relative to those in clinopyroxenes from the pyroxene-rich
4 domains (Fig. 9). As discussed below, this peculiar chemical feature was likely inherited by
5 metamorphic breakdown of precursor garnet in the primary assemblage.

6 Figure 8d shows the REE spectra of neoblasts from the plagioclase-bearing mineral
7 assemblage of a representative sample (BG13). Compared to clinopyroxene porphyroclasts,
8 clinopyroxene neoblasts always exhibit overall higher REE absolute concentrations and
9 development of negative Eu anomalies. Amphiboles have REE patterns similar in shape to
10 those of clinopyroxene neoblasts, but with higher absolute concentrations. Plagioclases have
11 low REE abundances and LREE enrichment ($La_N/Sm_N=3.55-5.41$), as well as marked positive
12 Eu anomalies.

14 Geothermometry

15 Conventional geothermometers, mostly based on Ca-Mg exchange between the two
16 pyroxenes, have been applied to both spinel-facies porphyroclasts and plagioclase-bearing
17 neoblasts (Table 10), in order to constrain the thermal evolution during the tectonic
18 exhumation. For the spinel-facies pyroxenes, the two-pyroxene thermometers provide similar
19 temperature ranges, between 830-940°C (Taylor, 1998) and 860-980°C ($T_{Opx-Cpx}$; Brey &
20 Kohler, 1990), which are in good agreement with temperature estimated by the single-
21 clinopyroxene thermometer of Nimis & Taylor (2000), between 820-960°C (Table 10).
22 Systematically higher temperatures are obtained using the Ca-in-Opx thermometer of Brey &
23 Kohler (1990), ranging between about 930 and 1100°C. As discussed by Nimis & Grutter
24 (2010), the Ca-in-Opx thermometer is affected by a positive bias at temperatures lower than
25 1000°C; however, we observe (Table 10) that this overestimation of calculated T persists even
26 after the empirical correction proposed by Nimis & Grutter (2010). Temperatures estimated
27 for spinel-facies pyroxenes partially overlap the temperature range documented by Borghini et
28 al. (2011) in the associated peridotites (940-1000°C), though extending to lower values. This
29 is probably due to the higher extent of recrystallization shown by the pyroxenites relative to
30 the host peridotites. The occurrence of thin (2-10 μm) exsolution in both clinopyroxene and
31 orthopyroxene porphyroclasts of the pyroxenites suggests that the spinel-bearing
32 porphyroclasts are likely affected by low-P re-equilibration. Accordingly, temperatures
33 obtained for the spinel-facies stage must be considered as minimum T estimates.

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3 1 Systematically lower temperature estimates have been calculated for the plagioclase-
4 bearing neoblasts with four different thermometers: 830-890°C and 790-860°C with the Brey
5 & Kohler (1990) thermometers (Ca-in Opx and Opx-Cpx, respectively), 770-860°C with the
6 Taylor (1998) thermometer and 780-900°C with Nimis & Taylor (2000) thermometer. As
7 observed for the spinel-facies minerals, Borghini et al. (2011) found temperatures up to 950°C
8 in the plagioclase-facies recrystallization in the associated peridotites, about 50°C higher than
9 those of the pyroxenites (Table 10).
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16 **Sr-Nd isotope compositions and Sm-Nd geochronology**

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18 Sr and Nd isotopic compositions of clinopyroxenes from pyroxenites investigated in
19 this work have been reported by Borghini et al. (2013), together with spatially associated host
20 rocks and distal peridotites, i.e. peridotite sampled in pyroxenite-free outcrops. In Figure 10a,
21 the present-day Sr and Nd isotope ratios of pyroxenites are plotted along with isotopic data
22 previously documented in orogenic and ophiolitic massifs (references reported in caption of
23 Fig. 10), and the compositional field of global MORB (Class & Lehnert, 2012: PetDB Expert
24 MORB, Mid-Ocean Ridge Basalt, Compilation. EarthChem Library.
25 <http://dx.doi.org/10.1594/IEDA/100060>) and OIB (GEOROC database: [http://georoc.mpch-](http://georoc.mpch-mainz.gwdg.de/georoc/)
26 [mainz.gwdg.de/georoc/](http://georoc.mpch-mainz.gwdg.de/georoc/)).
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33 Our pyroxenites have MORB-type Sr and Nd isotopic compositions, and when
34 compared to the data of pyroxenites from orogenic and ophiolitic massifs, they plot among
35 the samples with the lowest $^{87}\text{Sr}/^{86}\text{Sr}$ and highest $^{143}\text{Nd}/^{144}\text{Nd}$ ratios (Fig. 10a). Initial ϵNd
36 values computed at the inferred age of pyroxenite emplacement (433 ± 51 Ma; Borghini et al.,
37 2013) range between +3 and +8, consistent with enriched MORB compositions.
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41 Here, we present new Sm-Nd isotopic data on plagioclase separates and whole-rock
42 powders of selected pyroxenite samples that, together with clinopyroxene data from Borghini
43 et al. (2013), are used to construct internal isochrons. The Sm-Nd isotopic compositions of
44 whole-rocks and leached clinopyroxene and plagioclase mineral separates of selected
45 pyroxenites are listed in Table 11, together with elemental abundances of Nd and Sm
46 determined by isotope dilution. The latter are largely within 10% of the bulk-rock analyses
47 (Table 2), and most of Sm and Nd concentrations in clinopyroxenes and plagioclases obtained
48 by ID are within the variability shown by corresponding *in situ* LA-ICP-MS data (Table 9).
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53 Sm-Nd internal isochrons have been defined by whole rock, plagioclase and
54 clinopyroxene for pyroxenites BG13, BG14 and GV8, and by plagioclase and clinopyroxene
55 for sample BG5 (Fig. 11). Age calculations have been performed using Isoplot (Ludwig,
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1 2003). The strong Sm-Nd fractionation between plagioclase and clinopyroxene in the four
2 pyroxenite samples provides for well-constrained ages of 177 ± 13 , 178 ± 11 , 174 ± 25 and
3 183 ± 14 Ma. According to petrographic features, these ages date the event of partial
4 metamorphic recrystallization at low-pressure plagioclase-facies conditions, as temperature
5 estimates for plagioclase-facies neoblasts (Table 10) are largely higher than the closure
6 temperature for the Sm-Nd system (Mezger et al., 1992). Considering that the ages are
7 indistinguishable within uncertainty, and given the field and geochemical evidence of the
8 decompressional evolution, documented also in the associated host peridotites (Borghini et
9 al., 2011), we conclude that the EL pyroxenite-peridotite mantle experienced subsolidus
10 recrystallization in a single event. The age of this event is given by the weighted average of
11 the individual internal isochrons, at 178 ± 8 Ma.

12 13 **DISCUSSION**

14 **Reconstruction of the precursor mineral assemblage**

15 The pyroxenites investigated in this study contain an older spinel-bearing mineral
16 assemblage partially replaced by subsequent plagioclase-bearing neoblastic aggregates (Fig.
17 3). However, some microstructural and chemical features shown by several pyroxenite
18 samples suggest that variable modal amounts of garnet were present in the precursor mineral
19 associations. The most important evidence is that clinopyroxene porphyroclasts from spinel-
20 rich domains systematically show HREE enrichment coupled to high Zr and Sc contents (Fig.
21 9). These compositional features can be explained by the replacement of previously existing
22 garnet, which preferentially partitions HREE over the MREE. Moreover, the Zr and Sc Kd
23 values of garnet/melt are much higher than those of clinopyroxene/melt (e.g. Hauri et al.,
24 1994; Johnson, 1998, Ionov et al., 2002). Conversely, experimentally determined Kds
25 (spinel/melt) for HFSE are about an order of magnitude lower than those of both garnet and
26 clinopyroxene (Horn et al., 1994). Similar trace element characteristics of precursor garnets
27 have been previously documented by Vannucci et al. (1993) in pyroxenes of
28 spinel+plagioclase-bearing pyroxenites from Zabargad (Red Sea). They also observed that
29 this garnet imprint has been partially erased by subsequent metamorphic recrystallization at
30 lower pressure and is only evident in a few relicts. Low pressure overprinting could also
31 explain the variability of HREE-MREE fractionation of clinopyroxene porphyroclasts in
32 spinel-rich domains of our pyroxenites (Fig. 9). Zr and Sc enrichments seem to be better
33 preserved, as shown by the kink in the chemical profile of Figure 9, even though Zr and Sc
34 contents could have been higher in the early garnet and later partially smoothed by chemical

1 diffusion during subsequent spinel- and plagioclase-facies partial re-equilibration. It is
2 remarkable that Sc contents observed in clinopyroxene porphyroclasts from spinel-rich and
3 pyroxene-rich domains match very well with the range of Sc variations in garnet and its
4 equilibrium clinopyroxene, respectively, documented by Müntener et al. (2010) in garnet
5 pyroxenite from Totalp and Malenco (Fig. 9).

6 The CMAS projection in the garnet-pyroxene plane of pyroxenite bulk-rock
7 compositions is usually considered as evidence of primary crystallization of a mineral
8 assemblage dominated by garnet and pyroxene (e.g. Garrido & Bodinier, 1999; Takazawa et
9 al., 1999). Accordingly, in the CaTs-Fo-Qz triangle of Figure 12, most pyroxenites plot close
10 to the enstatite-garnet-CaTs line.

11 HREE enrichment over the MREE shown by some pyroxenite bulk-rock REE spectra
12 (samples BG13, GV10, MC3, MC5; Fig. 5) is comparable to that observed in garnet-bearing
13 pyroxenites from orogenic ultramafic massifs (e.g. Beni Bousera, Pearson et al., 1991, 1993;
14 Kumar et al., 1996; Gysi et al., 2011; Ronda, Garrido & Bodinier, 1999; Pyrenees, Bodinier et
15 al., 1987a), and has been interpreted as indicative of high-pressure magmatic segregation of
16 garnet (e.g. Kornprobst et al., 1969, 1970; Loubet & Allegre 1982). Similar bulk HREE-
17 MREE fractionation indicating precursor garnet has been documented in low-pressure
18 recrystallized pyroxenites from Zabargad and Horoman (Vannucci et al., 1995; Takazawa et
19 al., 1999).

20 In order to test the hypothesis of a precursor garnet-bearing assemblage in some of our
21 pyroxenites, we performed a mass balance calculation to derive an estimate of the original
22 modal compositions, in two separate steps. First, we calculated the modal abundances of a
23 potential garnet-bearing assemblage by using whole-rock major element data of the studied
24 pyroxenites (Table 2) and mineral compositions of garnet pyroxenites having similar bulk-
25 rock chemistry from the literature (see details below). Mass balance calculations, performed
26 by weighted least-squares minimization, included seven major element oxides (SiO_2 , TiO_2 ,
27 Al_2O_3 , Cr_2O_3 , FeO , MgO , CaO and Na_2O), and were computed by using the software
28 Wolfram Mathematica 8. The quality of the fit was evaluated based on the sum of the squares
29 of the residuals (R^2 in Table 12); we considered as realistic only the computed garnet-bearing
30 assemblages with errors well below 1.

31 In the second step, we derived the hypothetical Sm_N/Yb_N ratios of bulk-rocks from the
32 Sm_N/Yb_N ratios of minerals used to calculate the primary garnet-bearing assemblage.

33 Results of the first step of the mass balance calculation are reported in Table 12,
34 together with the spinel-facies modes computed using the whole rock major element

1 abundances and the compositions of spinel-facies porphyroclasts in each pyroxenite. We
2 estimated garnet-bearing modes by adopting the mineral compositions of selected garnet
3 pyroxenite samples from the Beni Bousera (Gysi et al., 2011) and Pyrenees (Bodinier et al.,
4 1987a) ultramafic massifs, with bulk-rock compositions similar to our pyroxenites (sample
5 B7, B9 and B15 with Mg-numbers ranging from 78-85, Gysi et al., 2011; sample 70-379 Mg-
6 number = 86, Bodinier et al., 1987a; [Figure 4](#), [Suppl. Table 3](#) and [Suppl. Fig. 2](#)). Beni
7 Bousera pyroxenites cover a broader range in both whole-rock and mineral chemistry, thus
8 providing a larger choice of compositions that can better simulate our samples, particularly in
9 terms of major element contents of garnet and pyroxenes (Gysi et al., 2011). Garnet
10 websterites from the Pyrenees have minerals with Mg-numbers comparable with our samples
11 MC3 and MC5 ([Fig. 4](#) and [Suppl. Table 3](#)).

12 In addition to pyroxene, garnet and spinel compositions from literature garnet
13 pyroxenites, we used, in the mass balance calculation, the average composition of olivine
14 measured in each sample, as it strongly depends on bulk-rock Mg-number and it is not
15 expected to change composition in the different mineral assemblages within a single sample.

16 The mass balance shows that all the bulk pyroxenite compositions characterized by
17 HREE enrichment over the MREE are consistent with primary garnet-bearing assemblages,
18 with modal garnet varying between 20-40 vol% (samples BG13, GV8, GV10, GV12, MC3,
19 MC5, [Table 12](#)). Moreover, our calculations show that lower garnet modal abundances
20 (generally <17 vol%) could have been present in the precursor assemblage also for other
21 pyroxenites with nearly flat MREE-HREE spectra (samples BG4, BG5, BG14, [Table 12](#)).
22 This is, for example, documented in some garnet pyroxenites from Beni Bousera,
23 characterized by bulk-rock REE patterns lacking MREE-HREE fractionation (Gysi et al.,
24 2011).

25 The reliability of the computed garnet-bearing mineral modes has then been checked
26 by comparing calculated bulk $(\text{Sm}/\text{Yb})_N$ ratios with measured values ([Fig. 13](#)). Except for
27 sample BG14, good agreement is generally observed between the two values. Pyroxenites
28 with Mg-numbers between 74 and 84 are well modelled by major and trace element data of
29 minerals from Beni Bousera garnet pyroxenites (sample B7, B9 and B15; Gysi et al., 2011),
30 whereas pyroxenites MC3 and MC5, with slightly higher Mg-numbers (86-87), yield a better
31 fit using data from Pyrenean garnet websterites (Bodinier et al., 1987a). For samples GV14
32 and GV17 with very low SiO_2 bulk-rock abundances, we could not compute realistic garnet-
33 bearing assemblages, and their spinel-bearing modal compositions have also the highest errors
34 (R2, [Table 12](#)). In terms of the estimated REE patterns ([Suppl. Fig. 2](#)), our mass balance

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3 1 modelling is able to simulate the overall MREE-HREE fractionation, but does not reproduce
4 2 the kink between Sm and Gd. It must be noted that modelling is very sensitive to the MREE-
5 3 to-HREE compositions of garnet and pyroxenes used in the calculation. However, selective
6 4 Gd_N-Yb_N enrichment in whole-rock chemistry has been documented also in garnet pyroxenite
7 5 from the literature (e.g. Gysi et al., 2011; Marchesi et al., 2013).

8 6 We are well aware that the above mass balance calculations do not constitute absolute
9 7 proof of the occurrence of garnet in the precursor assemblage. They do, however, demonstrate
10 8 that this is a highly plausible explanation for the otherwise rather difficult-to-explain
11 9 abundance patterns of the HREE. Garnet has important implications for the chemistry of
12 10 melts formed in its presence, and for the depth of melting, and we will explore some of the
13 11 consequences of this in the remainder of this paper.

12 13 **Origin of pyroxenites by deep melt infiltration**

14 14 Garnet pyroxenites have been described in orogenic and ophiolitic ultramafic massifs
15 15 (Downes, 2007, and references therein; Montanini et al., 2006, 2012; Gysi et al., 2011,
16 16 Marchesi et al., 2013), and in mantle xenoliths (e.g. Jacob, 2004; Bizimis et al., 2005;
17 17 Gonzaga et al., 2009), and their origin has been related to a range of processes. Garnet is the
18 18 typical high-pressure metamorphic Al-rich mineral in both mantle peridotites and pyroxenites
19 19 (Herzberg, 1978; Gasparik, 1984). In addition, a garnet-bearing assemblage can result from
20 20 high-pressure crystallization of a mafic magma or by complete high-pressure metamorphic
21 21 recrystallization of mafic rocks with primary low-pressure mineral assemblages (i.e.,
22 22 “eclogites”).

23 23 Garnet pyroxenites of both origins can be present in a single ultramafic massif, as
24 24 documented in Lherz (Bodinier et al., 1987a,b), Ronda (Garrido & Bodinier, 1999) and Beni
25 25 Boussera (Kornprobst et al., 1990). In some garnet pyroxenites, whole-rock trace element
26 26 compositions with positive Eu and Sr anomalies, relatively low HREE and MREE, and high
27 27 LREE/HREE, reflect high modal abundances of primary plagioclase, and thus they have been
28 28 considered of crustal origin (e.g. Kornprobst et al., 1990; Garrido & Bodinier, 1999;
29 29 Morishita et al., 2003, Marchesi et al., 2013). In some garnet pyroxenites from Beni Boussera,
30 30 a low-P crustal imprinting is also supported by peculiar Nd-Hf-O isotopic compositions
31 31 (Pearson et al., 1993; Pearson & Nowell, 2004). Based on geochemical and isotopic data, a
32 32 similar conclusion has also been inferred for a few garnet pyroxenites from other sectors of
33 33 the External Liguride ophiolites by Montanini et al. (2006, 2012). When compared to our
34 34 pyroxenites, the latter have significantly lower bulk Mg-numbers and higher Al₂O₃ and Na₂O

1 contents (Fig. 4). Montanini et al. (2012) inferred that these garnet pyroxenites represent
2 either melt residues of precursor eclogites, or high-pressure segregation products of eclogite-
3 derived melts. Pyroxenites investigated in our study are not consistent with these origins.
4 First, they lack the trace element bulk-rock compositions indicative of plagioclase-rich
5 gabbros characterized by positive Sr and Eu anomalies, rather high LREE and very low
6 HREE, thus, ruling out an origin as recrystallization products from low-pressure magmatic
7 protoliths (Fig. 5). Second, almost flat or weakly LREE-depleted bulk REE spectra
8 ($La_N/Sm_N = 0.15-0.66$; Table 2) do not support an origin as melting residues of precursor
9 mafic rocks (Fig. 5).

10 Instead, the inferred occurrence of primary garnet in most samples (Table 12)
11 indicates that the pyroxenites likely originated by segregation of melts at rather high pressure,
12 as garnet is stable at $P > 1.5$ GPa in pyroxenitic assemblages (e.g. Kogiso et al., 2004, Downes
13 2007). Garnet-bearing assemblages have been obtained by crystallization of a hydrous
14 primitive magma in experiments at 1.2 GPa and variable T (Müntener et al., 2001). However,
15 the bulk compositions of these experimental magmatic products do not match the
16 compositions of our pyroxenites (Fig. 4), and the lack of evidence of primary amphibole in
17 the pyroxenites does not support the origin from a hydrous melt.

19 **Geochemical affinity of the parental melts**

20 In this section we discuss the nature of the parental melts to the pyroxenites, but note
21 two major limitations: i) the primary garnet-facies mineral assemblage has been completely
22 obliterated by metamorphic recrystallization events, most likely related to the tectonic
23 exhumation of the pyroxenite-peridotite mantle sequence (Borghini et al., 2011), ii) magmatic
24 interaction and/or subsolidus diffusion between pyroxenite and host peridotite have likely
25 partially modified the original chemical signatures of the melts (e.g. the Mg-number). This
26 precludes the possibility to compute equilibrium melts by using the present trace element
27 mineral composition. Therefore, the nature of pyroxenite parental melts can only be inferred
28 by combining available microstructural and chemical observations.

29 The bulk-rock Mg-numbers vary from 74-88, and are inversely correlated with the
30 thickness of layers and lenses, and positively correlated with the bulk Ni contents (Fig. 14).
31 An inverse relationship between dyke thickness and bulk Mg-number has been documented in
32 pyroxenitic layers from the Lherz ultramafic massif and explained as the result of local
33 chemical re-equilibration with the adjacent peridotite (Bodinier et al., 1987b; Fabries et al.,
34 2001). Bodinier et al. (2008) have investigated in detail this process by studying some

1 pyroxenites from Ronda (Spain). These authors performed numerical modeling demonstrating
2 that interaction between the host peridotite and variably evolved melts can generate
3 pyroxenites with Mg-numbers intermediate between those of the peridotite and melts.

4 The chemical profile along the GV10 pyroxenite (9 cm-long, Fig. 9 and Suppl. Fig. 1),
5 where data were acquired only in cores of large clinopyroxene porphyroclasts, shows a
6 significant zonation across the pyroxenite layer, with Mg-numbers of clinopyroxenes at the
7 margins of the layer close to those of clinopyroxenes in the surrounding peridotite (a complete
8 data set of this profile is provided in Suppl. Tables 1 and 2). Similar chemical gradients have
9 been described across mafic-peridotite associations from orogenic massifs (e.g. Pearson et al.,
10 1993; Takazawa et al., 1999; Mazzucchelli et al., 2010). They indicate that most of these
11 pyroxenites, having a thickness of only a few centimetres, do not reflect equilibrium with the
12 infiltrating parental melts but rather show a modified Mg-number due to the interaction with
13 the surrounding peridotite. Accordingly, the composition of clinopyroxene porphyroclasts
14 from the core of our thickest pyroxenite layer, GV10 and lens GV12, are likely to be the least
15 affected by interaction with the peridotite. We thus used their compositions to calculate the
16 Mg-number of the equilibrium parental melt, with the equation $Mg\#_{melt} = (K^{cpx}_{Fe-Mg} * Mg\#_{cpx}) / (1 - Mg\#_{cpx} + Mg\#_{cpx} * K^{cpx}_{Fe-Mg})$ proposed by Wood and Blundy (1997).

17 We assume that primary garnet-bearing clinopyroxenes had Mg-numbers similar to
18 those of spinel-facies clinopyroxenes, because Mg-number is not expected to be significantly
19 modified by subsequent lower-pressure recrystallization (e.g. Robinson & Wood, 1998).
20 Computed Mg-numbers of the melt vary between 52-56 for layer GV10 and 44-50 for lens
21 GV12. Melts with such values, significantly lower than the Mg-number of melts in
22 equilibrium with mantle minerals, must have been very reactive with the surrounding
23 peridotite. This is consistent with the occurrence of orthopyroxene-rich rims at the
24 pyroxenite-peridotite boundary, as well as the orthopyroxene blebs and orthopyroxene modal
25 enrichment in the wall rocks documented by Borghini et al. (2013).

26 Orthopyroxene enrichment at the pyroxenite-peridotite boundaries has been
27 documented in other ultramafic massifs (e.g. Bodinier et al., 1987a; Varfalvy et al., 1996;
28 Takazawa et al., 1999), and interpreted as the result of either solid-state reaction (e.g. Kogiso
29 et al., 2004b; Herzberg et al., 2011) or reaction between high-silica activity melts, coming
30 from the pyroxenites, and the surrounding peridotite (e.g. Varfalvy et al., 1996; Yaxley &
31 Green, 1998; Sobolev et al., 2005; Lambart et al., 2012). In the studied pyroxenites, Borghini
32 et al. (2013) have documented that clinopyroxenes in the first few (<5) centimetres of the host
33 peridotite, at the contact with the pyroxenite layer (i.e. our so-called “wall-rock” peridotite),
34

1 have been chemically modified. In particular, the cm-scale progressive REE enrichment
2 observed in clinopyroxene porphyroclasts of wall-rock peridotite points to reactive
3 percolation of the pyroxenite-derived melt into the adjacent mantle. This is also consistent
4 with the complete isotopic resetting observed in the first centimeters of the wall-rock
5 peridotite (Borghini et al., 2013).

6 Lack of evidence for precursor garnet in the host peridotites suggests that this melt-
7 rock reaction process occurred under spinel-facies mantle conditions, at pressures below
8 garnet-lherzolite stability ($P < 2.8$ GPa, e.g. O'Neill et al., 1981; Robinson & Wood, 1998),
9 but likely above 1.5 GPa, to account for the garnet-bearing primary modes reconstructed for
10 the pyroxenites. Spinel-facies clinopyroxene porphyroclasts in the wall-rock peridotites, thus,
11 retain a trace element signature acquired by the interaction with the pyroxenite-derived melts
12 (Borghini et al., 2013), and their REE composition can be used to obtain information on the
13 geochemical signature of the pyroxenite parental melts.

14 Computed melts in equilibrium with wall-rock clinopyroxene porphyroclasts in the
15 three profiles investigated by Borghini et al. (2013) are shown in Figure 15. We note here that
16 a large set of trace element data has been filtered in order to select only the clinopyroxene
17 porphyroclasts not affected by the subsequent plagioclase-bearing recrystallization, for
18 example by checking Sr-Zr elemental variations (Rampone et al., 1993). However, the effect
19 of a partial subsolidus plagioclase-facies re-equilibration would tend to slightly increase the
20 REE absolute contents in clinopyroxene, without causing significant change of the
21 LREE/HREE fractionation (Rampone et al., 1993; Hellebrand et al., 2005). Equilibrium melts
22 are characterized by enrichment of LREE over HREE ($La_N/Yb_N = 4-7.6$) and moderate
23 MREE-HREE fractionation ($Dy_N/Yb_N = 1.2-1.4$). The resulting melts have LREE
24 significantly higher than a normal MORB and, rather, they resemble the REE compositions of
25 enriched MORBs (E-type MORB) sampled in oceanic settings (Hemond et al., 1993, 2006;
26 Hoernle et al., 2011; Waters et al., 2011). This is consistent with the initial epsilonNd values
27 of clinopyroxenes in the pyroxenites, ranging from +3.29 to +8.25 (Borghini et al., 2013).

28 In summary, based on our microstructural and chemical observations, we conclude
29 that the studied pyroxenites have originated from the segregation of low-Mg-number E-
30 MORB melts. Moreover, very low alkali abundances of the bulk-rock pyroxenite
31 compositions (Fig. 4 and Table 2), suggest a tholeiitic affinity of the parental melts.

32 Borghini et al. (2013) showed that interaction of EL pyroxenites with wall-rock
33 peridotites reset the Nd isotopic compositions of the wall-rocks. They demonstrated that Sm-
34 Nd isotopic investigations on clinopyroxene from pyroxenite-wall rock peridotite pairs

1 provide linear correlations yielding ages of 424-452 Ma, which are in good agreement with
2 the age resulting from the errorchron defined by all pyroxenites (433 ± 51 Ma), thus indicating
3 the most likely timing of melt infiltration and pyroxenite emplacement. The ages indicate that
4 this event significantly predated the Mesozoic evolution of this mantle sector (discussed
5 below), and may have been related to Late Ordovician-Silurian lithosphere extension that led
6 to the opening of the PaleoTethys oceanic basin (e.g. Stampfli & Borel 2002; von Raumer &
7 Stampfli, 2008; Murphy et al., 2008).

9 **Inferences on the nature of mantle source**

10 The computed Mg-numbers of the parental melt (44-56) of the pyroxenites are too low
11 to represent a primitive basaltic melt from a peridotitic mantle source (e.g. Walter, 1998).
12 This indicates that pyroxenites originated either from i) a peridotite-derived melt that
13 experienced significant chemical evolution, or ii) a melt derived by melting of a low-Mg-
14 number hybrid source. The first hypothesis has been inferred for garnet pyroxenites from
15 xenoliths (e.g. Frey et al., 1980; Keshav et al., 2007) and orogenic massifs (e.g. Bodinier et
16 al., 1987a,b), and requires processes of liquid-crystal separation. A commonly invoked
17 mechanism of high-pressure crystal segregation and melt differentiation is the dynamic flow
18 crystallization in conduits described by Irving (1980). This model provides a possible
19 alternative to the cumulus processes requiring the settling of crystals by gravity in magma
20 chambers, a process difficult to be envisaged at pressures above 1.5 GPa within a plastically-
21 deforming peridotite. However, how hydraulic fractures can form at such depth is unknown,
22 because the deep mantle is not expected to be brittle enough to fracture (e.g. Kelemen et al.,
23 1995, 1997). Moreover, it is noteworthy that the lowest Mg-number computed for our
24 pyroxenite parental melts (around 44-50) would require more than 60% of fractional
25 crystallization of a primitive melt with a Mg-number of 70 (Villiger et al., 2004). This would
26 imply that a large amount of magmatic products should have been created in the mantle,
27 before generating the evolved melt parental to the observed pyroxenites. Although there is no
28 evidence of such large fractionation processes, we cannot exclude that this process played a
29 role in the origin of our pyroxenites.

30 An alternative hypothesis calls for the involvement of a low-MgO mafic source
31 component, as proposed for some pyroxenites from orogenic and ophiolitic massifs (e.g.
32 Garrido & Bodinier, 1999; van Acken et al., 2010; Montanini et al., 2012; Marchesi et al.,
33 2013; **Montanini & Tribuzio, 2015**). This means that low-Mg-number melts parental to
34 pyroxenites can be the result of partial melting of a precursor mafic/pyroxenitic component or

1 a mixed pyroxenite-peridotite source. Several experimental studies have been dedicated to the
2 melting behaviour of mafic lithologies at variably high pressure, and they have shown that a
3 wide range of melt compositions can be produced, depending on the choice of starting bulk
4 composition (e.g. Kogiso et al., 2004; Lambart et al., 2013).

5 In a review study on the composition of melts produced by pyroxenite melting at high
6 pressure (>2 GPa), Kogiso et al. (2004) suggested that silica-deficient (SD) pyroxenites can
7 be involved in the genesis of alkalic OIB lavas, whereas mixing of silica-excess (SE)
8 pyroxenite melts with MgO-rich peridotitic melts would account for the genesis of tholeiitic
9 OIB. Lack of olivine in the primary garnet-bearing assemblage of our pyroxenites (Table 12),
10 together with the occurrence of orthopyroxene-rich rims along the contact with wall-rock
11 peridotites, indicate that the observed pyroxenites have crystallized from melts with high
12 silica activity, suggesting the involvement of a silica-excess pyroxenite component (e.g.
13 Sobolev et al., 2005, 2007). Conversely, olivine-rich websterite and wehrlite are expected to
14 form by reaction between melts produced by melting of SD pyroxenites and a peridotite
15 (Lambart et al., 2012). Furthermore, the compositions of the pyroxenites in the proximity of
16 the thermal divide in the pseudo-ternary CaTs-Fo-Qz diagram (see Fig. 12), suggest that they
17 could have formed by partial melting of mantle sources with mixed eclogite-peridotite
18 lithologies (Yaxley and Green, 1998; Kogiso et al., 1998; Takahashi & Nakajima, 2002;
19 Mallik and Dasgupta, 2012; Fig. 12). Mallik and Dasgupta (2012) have recently demonstrated
20 that saturated tholeiitic melts, with Mg-numbers as low as about 56, can form through
21 reaction between eclogite-derived partial melts and a fertile peridotite. Remarkably, the
22 compositions of such melts cover the range defined by our pyroxenites in the CMAS system
23 (Fig. 12).

24 We propose that the EL pyroxenites originated from melts produced by a hybrid
25 eclogite-peridotite mantle source, further modified through reaction with the host peridotite
26 after their infiltration. The E-MORB signature inferred for the parental melts is consistent
27 with this origin, given that similar E-MORB melts from modern mid-ocean-ridge settings are
28 often interpreted to be the products of partial melting of a mantle peridotite source enriched
29 by a pyroxenitic/mafic component (e.g. Hémond et al., 2006; Waters et al., 2011; Hoernle et
30 al., 2011).

31 The major limitation of this hypothesis is that these low-Mg-number melts are highly
32 reactive in the mantle; indeed, if their migration occurs by pervasive porous flow, then the
33 compositions of these melts are expected to be rapidly buffered by reaction with the host
34 peridotite (e.g. Daines & Kohlstedt, 1994). However, the occurrence of residues of precursor

1 eclogites within the EL mantle sector, in the form of garnet pyroxenites (Montanini et al.,
2 2012), may support the scenario of a pristine eclogite-bearing mantle affected by partial
3 melting. This could also indicate that such melts, after their production, migrated only over
4 short distances, thus, preventing significant re-equilibration with the surrounding peridotite.
5 The migration of these melts could have been enhanced by stress-driven melt segregation
6 along shear zones, a mechanism able to produce high-permeability pathways within partially
7 molten ductile rocks (e.g. Holtzman et al., 2003; Havlin et al., 2013). This is consistent with
8 some field and geochemical evidence (Borghini unpublished), suggesting that the pyroxenite
9 emplacement occurred within an already plastically deformed mantle. Part of the evolved
10 signature of parental melts to the pyroxenites, such as the low Mg-numbers, could have been
11 acquired by crystal segregation and melt evolution along these high-porosity magma conduits.

12 In summary, a possible scenario is that low-Mg-number melts produced by moderate
13 degrees of SE pyroxenite/eclogite melting were mixed with low-degree peridotitic melts,
14 which infiltrated and reacted with the overlying mantle to form the Suvero suite of
15 pyroxenites layers.

16 This scenario is in agreement with recent models of basalt petrogenesis that invoke the
17 generation of secondary pyroxenites in the upper mantle by interaction between peridotites
18 and silica-rich melts derived by deeper melting of mafic components prior to major basalt
19 production with larger degrees of melting (Sobolev et al., 2005, 2007; Lambart et al., 2012).
20 According to our petrological reconstruction, the EL pyroxenites represent an excellent
21 natural example of such secondary pyroxenites. We argue that partial melting of mafic
22 components within a rising heterogeneous mantle, and the infiltration of the resulting melts,
23 reacting with the host mantle peridotites, are efficient processes to generate additional
24 heterogeneities in the upper mantle (Garrido & Bodinier, 1999; Borghini et al., 2013;
25 Marchesi et al., 2013; Montanini & Tribuzio, 2015).

26 27 **Metamorphic evolution and timing of exhumation**

28 The mass balance modelling in this work accounts for the fractionation of HREE over
29 the MREE shown by the bulk composition of many of the pyroxenites, thus supporting the
30 hypothesis that they originally contained a primary garnet-bearing mineral assemblage, which
31 is not preserved. The oldest mineral assemblage observed in the pyroxenites consists of coarse
32 porphyroclasts of green spinel, clinopyroxene \pm orthopyroxene, partially replaced by
33 plagioclase-bearing neoblasts. The occurrence of plagioclase+olivine coronas around spinel
34 porphyroclasts suggests that plagioclase did not originate directly from the breakdown of the

1 garnet-bearing assemblage. Rather, textural observations indicate that the plagioclase-bearing
2 neoblastic assemblage replaced a previous spinel-bearing association, thus suggesting that
3 plagioclase appeared at lower pressure than the spinel+pyroxenes assemblage. This is in
4 agreement with experimental results on a low-Na garnet-spinel websterite (pyroxenite R394
5 from Irving, 1974) and computed subsolidus phase relations for a low-Na, high-Mg websterite
6 from Ronda ultramafic massif (southern Spain, Hidas et al., 2013). A similar mineralogical
7 evolution has been documented in Mg-rich clinopyroxenites and websterites from other
8 External Liguride Units, and related to progressive mantle exhumation, from garnet- to spinel-
9 to plagioclase-facies conditions (Montanini et al., 2006).

10 Thermometric estimates on cores of coarse spinel-facies pyroxenes indicate
11 temperatures around 950-1000°C, which must be considered as minimum temperatures, given
12 the occurrence of exsolution textures in both pyroxenes. According to experimental results on
13 a garnet-spinel websterite, the spinel-bearing assemblage observed in the pyroxenites should
14 record a recrystallization stage within the P-T range of 1.2-1.5 GPa and $T < 1200^{\circ}\text{C}$, as this is
15 expected to be the solidus temperature at 1.5 GPa (Irving 1974).

16 Continued pressure decrease led to crystallization of a plagioclase-bearing neoblastic
17 assemblage at the expense of spinel-facies porphyroclasts. This process is accompanied by
18 systematic changes in mineral compositions, including lower Al content in pyroxenes (Fig. 6)
19 and higher Cr-number and TiO_2 content at lower Mg-number in the small relicts of spinel
20 (Fig. 7), similar to what is observed in plagioclase peridotites of metamorphic origin (e.g.
21 Gasparik, 1984; Rampone et al., 1993; Montanini et al., 2006; Borghini et al., 2010, 2011).
22 Borghini et al. (2011) previously traced the exhumation of mantle peridotite hosting the
23 pyroxenites from 0.7 to 0.3 GPa by applying thermometric estimates on neoblastic pyroxenes
24 and experimentally-derived An-isopleths of plagioclase. Therefore, we infer that associated
25 pyroxenites reasonably followed the same P-T path as the host peridotites. Average
26 compositions of plagioclase in pyroxenites (core: An = 0.58-0.64; rim: An = 0.76-0.82; Table
27 6) are very close to those documented in the associated peridotites (core: An = 0.56-0.59; rim:
28 An = 0.74-0.79; Borghini et al., 2011). This may indicate that the plagioclase composition
29 does not depend on the bulk-rock chemistry even in pyroxenite bulks, which are significantly
30 different from peridotite lithologies (Borghini et al., 2010), however, this needs to be
31 confirmed by further experimental data. As a consequence, according to the An in plagioclase
32 geobarometer proposed by Borghini et al. (2011), the similarity of An contents would
33 constitute robust evidence that plagioclase crystallized contemporaneously in both
34 pyroxenites and peridotites during their decompressional evolution.

1 Internal Sm-Nd isochrons, based on clinopyroxene-plagioclase-whole rocks analyses
2 performed on four samples, give an average age of 178 (± 8) Ma (Fig. 11) for the plagioclase-
3 facies recrystallization event, which dates the exhumation of this mantle sector at relatively
4 shallow depth. This age is slightly younger than the ages obtained by internal Sm-Nd
5 isochrons documented in garnet pyroxenites investigated by Montanini et al. (2006). They
6 proposed a Middle to Late Triassic age for the near-isothermal spinel- to plagioclase-facies
7 decompression recorded by the External Liguride mantle section. Sm-Nd ages provided by
8 this study indicate that exhumation of this EL mantle sector occurred in the Early Jurassic,
9 which is coeval with the first syn-rift MOR-type intrusions in the subcontinental mantle of the
10 Piedmont-Ligurian oceanic basin (Tribuzio et al., 2004, Rampone et al., 2014).

11 12 **CONCLUDING REMARKS**

13 Pyroxenites form centimetre-thick layers and lenses in peridotite from the External
14 Liguride ultramafic massif (Northern Apennines, Italy); they range from websterites to
15 clinopyroxenites, and are characterized by a porphyroclastic spinel-facies assemblage
16 partially replaced by a low-pressure plagioclase-bearing neoblastic association.

17 Microstructural and chemical features shown by several pyroxenite samples suggest
18 that variable modal amounts of garnet were present in the primary mineral association, thus
19 indicating that pyroxenites likely originated by segregation of melts at rather high pressure
20 (above 1.5 GPa).

21 Chemical and Sr-Nd isotopic compositions indicate that pyroxenites crystallized from
22 a E-MORB-type melt, similar to those documented in modern mid-ocean-ridge settings. Low
23 Mg-numbers together with the lack of olivine in most of the primary mineral assemblages
24 suggest that their parental melts were produced by partial melting of a hybrid pyroxenite-
25 bearing mantle, which then reacted with the host peridotite during high-pressure infiltration.
26 The EL pyroxenites studied in this work are an example of a “secondary pyroxenite”
27 component in the upper mantle, which has been proposed in recent models of basalt
28 petrogenesis at oceanic intra-plate and mid-ocean ridge settings.

29 Finally, later decompressional metamorphic evolution from spinel- to plagioclase-
30 facies followed the same P-T path defined for the host peridotites (Borghini et al., 2011).
31 Low-P plagioclase-facies re-equilibration is dated by internal (clinopyroxene-plagioclase-
32 whole rocks) Sm-Nd isochrons at 178 \pm 8 Ma, indicating that the exhumation of this mantle
33 sector can be attributed to the Mesozoic continental rifting that led to the opening of the
34 Jurassic Tethys.

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18 FIGURE CAPTIONS

19 **Figure 1.** (a) Tectonic sketch map of the Northern Apennines. Abbreviations for the tectonic
20 units: IL, Internal Liguride Units; EL, External Liguride Units. Also reported are SV,
21 crystalline massif of Savona, and GV, Voltri Group, belonging to the Ligurian Alps. Garnet
22 pyroxenites investigated by Montanini et al. (2006, 2012) were sampled in outcrops located
23 further north than the ultramafic bodies here studied, outside this map. (b) Simplified
24 geological map showing ultramafic bodies hosting the investigated pyroxenites.

25 **Figure 2.** Centimeters-thick pyroxenite layers embedded in mantle peridotite from (a) M.te
26 Castellaro and (b, c) Suvero ultramafic bodies (External Liguride Units, Northern Apennines,
27 Italy). (d) A representative macroscopic sample of a pyroxenite layer, showing thin
28 orthopyroxenite rims at the boundary with the host peridotite. White dotted lines delimit the
29 pyroxenite layers in (b) and (d).

30 **Figure 3.** Photomicrographs of representative microstructures of EL spinel-plagioclase
31 pyroxenites. Porphyroclasts of spinel and clinopyroxene partially replaced by fine-grained
32 plagioclase-bearing assemblage, at parallel (a) and crossed (b) nichols. (c) Coarse
33 orthopyroxene porphyroclast intensely substituted by the plagioclase-facies neoblastic
34 assemblage in a spinel-bearing websterite. (d) Coarse mm-sized green spinel porphyroclast

1 surrounded by a rim of neoblastic assemblage mostly made by plagioclase, olivine and
2 brownish amphibole. (e) Contact between a pyroxenite layer (on the right of the white dotted
3 line) and orthopyroxene-rich rim (on the left); presence of large orthopyroxene porphyroclast
4 (characterized by undulatory extinction) indicates that opx-rich rim was a mineralogical
5 feature that predated the low-pressure metamorphic recrystallization. (f) Texture in pyroxenite
6 lens GV12 showing a small clinopyroxene porphyroclast surrounded by a fined-grained band
7 of clinopyroxene and dark-green spinel.

8 **Figure 4.** Variation diagrams of SiO₂, Al₂O₃, Na₂O, CaO (wt%) and Cr, Ni, Sc, V (ppm) vs
9 Mg-number for the EL pyroxenite (circles). The compositional fields reported for comparison
10 are: garnet pyroxenites (continue line) and spinel websterite (dashed line) in ultramafic
11 orogenic massifs from Beni Boussera (Morocco, Pearson et al., 1993; Kumar et al., 1996;
12 Gysi et al., 2011), Ronda (Spain, Suen & Frey, 1987; Garrido & Bodinier, 1999; Bodinier et
13 al., 2008), Horoman (Japan, Takazawa et al., 1999; Morishita & Arai, 2001), Pyrenees
14 (France, Bodinier et al., 1987a,b), and Balmuccia (Italy, Sinigoï et al., 1983; Voshage et al.,
15 1988; Mukasa & Shervais, 1999). Grey field refers to the bulk compositions of garnet
16 pyroxenite of the External Liguride Units investigated by Montanini et al. (2006). Dotted line
17 defines the range of bulk solid compositions calculated using mineral phases in melting
18 experiments on an hydrous arc melt (Müntener et al., 2001). Bulk compositions of
19 pyroxenites adopted in mass balance calculations for estimating the garnet-bearing primary
20 assemblages are also reported (see [Suppl. Table 3](#)): stars are samples from Beni Boussera
21 (grey B7, black B9; white B15; Gysi et al., 2011), the square is sample 70379 selected from
22 the Pyrenean garnet websterites (Bodinier et al., 1987a).

23 **Figure 5.** Chondrite-normalized rare earth element (REE) abundances in whole rock
24 pyroxenites. Samples BG3, BG4, BG5 and BG14 are shown with the same symbol because
25 they have very similar bulk REE compositions. Normalizing values are from Anders &
26 Grevesse (1989). Also reported for comparison are the compositional fields of External
27 Liguride garnet pyroxenites (Montanini et al., 2012), and Alpine-Apennine (A-A) gabbros
28 (Borghini et al., 2007; Rampone et al., 1998; Rampone et al., 2014).

29 **Figure 6.** Variation of Mg# vs Al (a.p.f.u.) in clinopyroxene (a) and orthopyroxene (b) from
30 pyroxenites. Filled symbols refer to spinel-facies porphyroclasts, open symbols refer to
31 plagioclase-facies neoblasts (see the legend in figure).

32 **Figure 7.** Mg# vs Cr# (a), and Cr# vs Ti x 1000 (a.p.f.u.) (b) in spinel from pyroxenites. Filled
33 symbols refer to core of large spinel-facies porphyroclasts, open symbols refer to small relicts
34 of plagioclase-facies spinel intensely replaced by the plagioclase-facies assemblage.

1 **Figure 8.** Chondrite-normalized REE patterns of clinopyroxene porphyroclasts from (a)
 2 HREE-enriched pyroxenites, (b) pyroxenites with rather flat MREE-HREE spectra, (c)
 3 samples GV8, GV12 and GV14 (details in text). (d) Representative chondrite-normalized
 4 REE patterns of clinopyroxene spinel-facies porphyroclast, and neoblasts of clinopyroxene,
 5 plagioclase and amphibole in pyroxenite BG13. Normalizing values are from Anders &
 6 Grevesse (1989).

7 **Figure 9.** Sc, Zr and REE abundance in clinopyroxene porphyroclasts along profile A-B
 8 analyzed in pyroxenite layer GV10, showing alternating px-rich and sp-rich zones (Suppl.
 9 Table 2). Data collected by in-situ laser ablation. Normalizing values for REE are from
 10 Anders & Grevesse (1989). Grey fields in the Sc diagram define the Sc content variations in
 11 garnet and its equilibrium clinopyroxene documented in garnet peridotites by Müntener et al.
 12 (2010).

13 **Figure 10.** Present-day $^{87}\text{Sr}/^{86}\text{Sr}$ versus $^{143}\text{Nd}/^{144}\text{Nd}$ measured on clinopyroxene separates
 14 from studied pyroxenites, compared to isotope data of pyroxenites from literature: Cr-Di
 15 websterites and Sp pyroxenites from Balmuccia (Ivrea Zone, Italy, Voshage et al., 1988;
 16 Mukasa & Shervais, 1999); pyroxenite, websterites and gnt-pyroxenites from Beni Boussera
 17 (BB, North Morocco, Pearson et al., 1993; Kumar et al., 1996); gnt-pyroxenites from (LA,
 18 Lowe Austria, Becker et al., 1996); websterites from Ronda (South Spain, Marchesi et al.,
 19 2012). The field of mid-ocean-ridge basalts (MORB) is from Class & Lehnert (2012); the
 20 field of ocean island basalts (OIB) is from GEOROC database ([http://georoc.mpch-
 21 mainz.gwdg.de/georoc/](http://georoc.mpch-mainz.gwdg.de/georoc/)).

22 **Figure 11.** (a) Present-day $^{143}\text{Nd}/^{144}\text{Nd}$ versus $^{147}\text{Sm}/^{144}\text{Nd}$ in clinopyroxenes from
 23 pyroxenites that define an errochron (dashed line) giving an age of 433 ± 51 Ma. Empty
 24 symbols refer to spatially controlled pyroxenite – wall-rock profiles, and relative isochrones
 25 (continuous lines), reported in Borghini et al. (2013). Data of EL garnet pyroxenites from
 26 Montanini et al. (2012) are reported for comparison. (b) Sm-Nd internal isochrons defined by
 27 whole rock (wr), plagioclase (plag) and clinopyroxene (cpx) from four pyroxenite samples
 28 (BG13, BG14, GV8 and BG5, this latter only based on clinopyroxene and plagioclase
 29 analyses). Ages and initial Nd isotopic ratios for the internal isochrons are: 178 ± 11 Ma, initial
 30 $^{143}\text{Nd}/^{144}\text{Nd} = 0.512961 (\pm 24)$, MSWD = 0.22 for BG13; 177 ± 13 Ma, initial $^{143}\text{Nd}/^{144}\text{Nd} =$
 31 $0.512837 (\pm 20)$, MSWD = 0.69 for BG14; 174 ± 25 Ma, initial $^{143}\text{Nd}/^{144}\text{Nd} = 0.512727 (\pm 34)$,
 32 MSWD = 0.0024 for GV8; 183 ± 14 Ma, initial $^{143}\text{Nd}/^{144}\text{Nd} = 0.512827 (\pm 17)$ for BG5.

33 **Figure 12.** Whole-rock normative compositions of studied pyroxenites projected from
 34 Diopside within the (CMAS) triangle forsterite (Fo) - calcium tschermak (CaTs) - quartz (Qz)

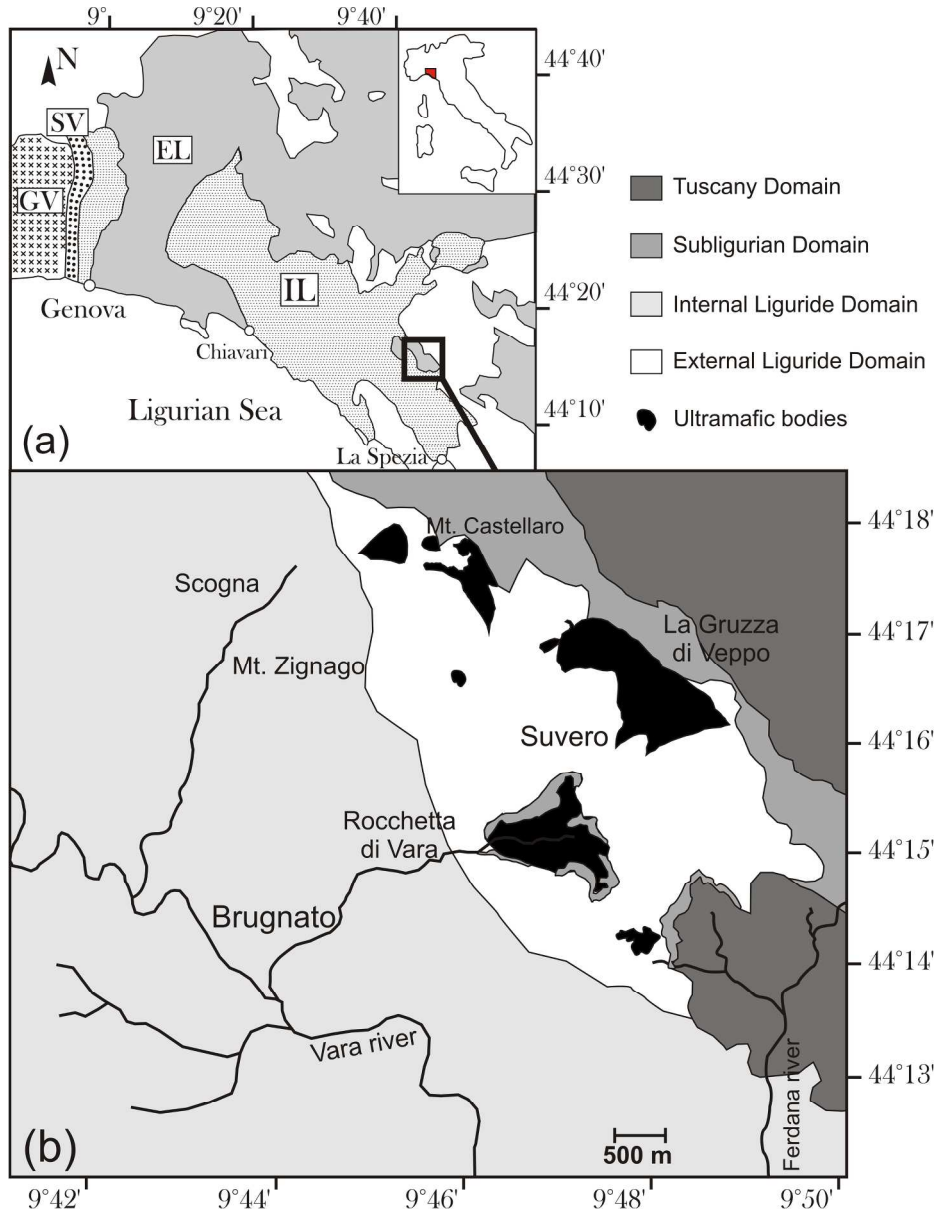
1 (O'Hara, 1968). The compositional fields of EL garnet pyroxenites (EL gnt-pyrox, Montanini
2 et al., 2012), Type-1 pyroxenites from Horoman (Takazawa et al., 1999), and Beni Boussera
3 garnet pyroxenites (BB-gnt-pyrox, Gysi et al., 2011), are also shown for comparison. In the
4 smaller CMAS triangle to the right, pyroxenite compositions are plotted together with the
5 compositions of experimental partial melts from hybrid mafic-peridotite sources (Yaxley &
6 Green, 1998; Kogiso et al., 1998; Takahashi & Nakajima, 2002), and the compositional field
7 of melts in the eclogite-peridotite reaction experiments by Mallik & Dasgupta (2012)(tick-
8 marks on the black arrow indicate the amount in wt% of eclogitic melt added to peridotite).

9 **Figure 13.** Bulk-rock Sm_N/Yb_N resulting from mass balance calculations compared to the
10 Sm_N/Yb_N values measured in each pyroxenite sample (detailed explanation in the text). **Open**
11 **circles are the values calculated by using mineral compositions from Beni Boussera garnet**
12 **pyroxenites (Gysi et al., 2011), open squares are obtained adopting in the calculation mineral**
13 **compositions from Pyrenees garnet pyroxenites (Bodinier et al., 1987a).**

14 **Figure 14.** Variation of bulk-rock Mg-number versus thickness (cm) of the studied pyroxenite
15 layers and lens. Close to the symbols are also shown the Ni contents (ppm) of each pyroxenite
16 sample.

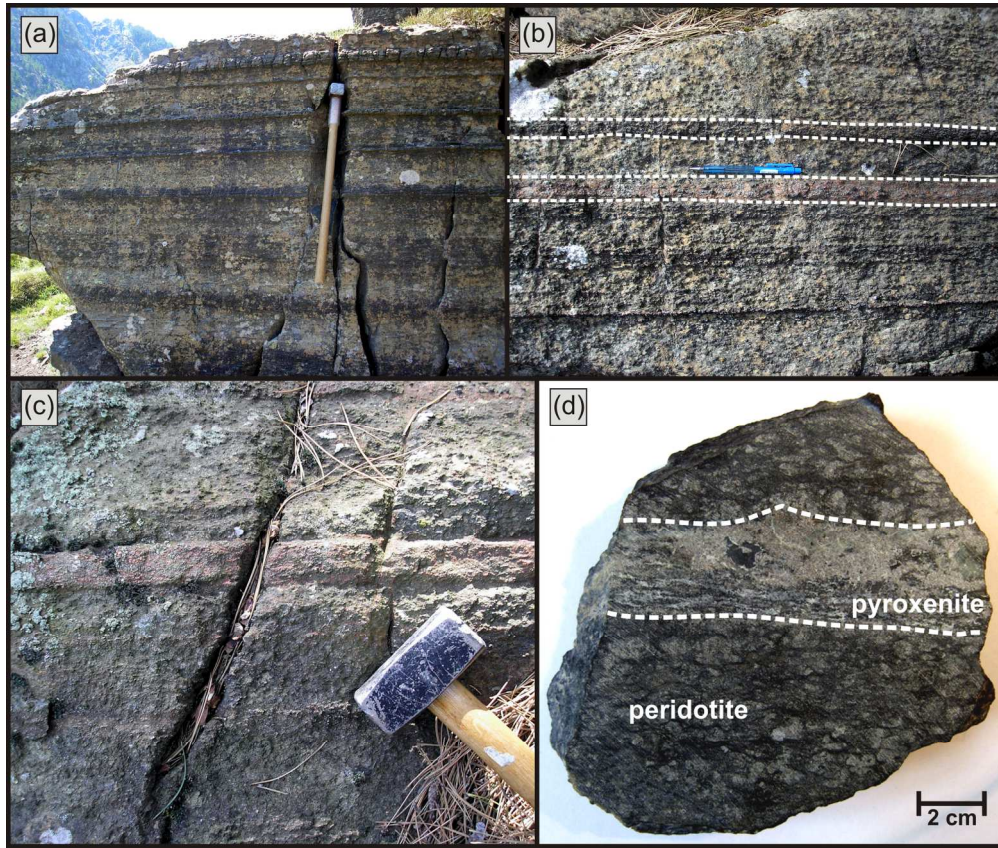
17 **Figure 15.** (a) Chondrite-normalized REE patterns of clinopyroxene porphyroclasts in wall-
18 rock peridotites from three profiles investigated by Borghini et al. (2013). (b) Computed melts
19 in equilibrium with wall-rock clinopyroxenes shown in (a), using partition coefficients from
20 Ionov et al. (2002). Also reported are the composition of average N-MORB (Hofmann, 1988),
21 and average E-MORB from Mid-Atlantic Ridge 10-24°N (Hémond et al., 2006). The
22 compositional fields refer to: 1) E-MORBs from South Atlantic Mid-Ocean-Ridge 5-11°S
23 (A3 Seamount; Hoernle et al., 2011), 2) tholeiitic lavas (both olivine- and quartz-normative)
24 from Iceland plume (Icelandic tholeiites; Hémond et al., 1993), 3) East Pacific Rise 9-10°N
25 (Waters et al., 2011). Normalizing values are from Anders & Grevesse (1989).

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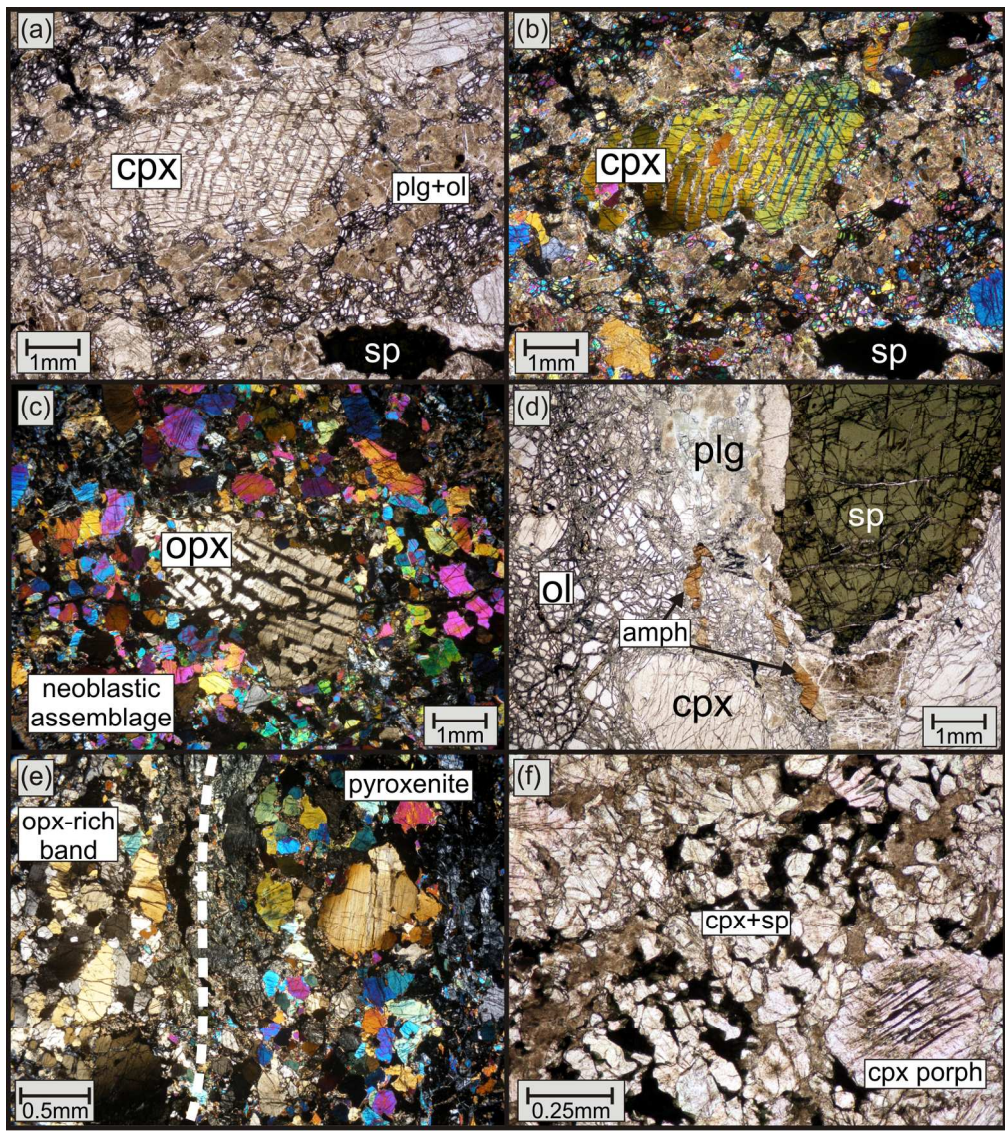
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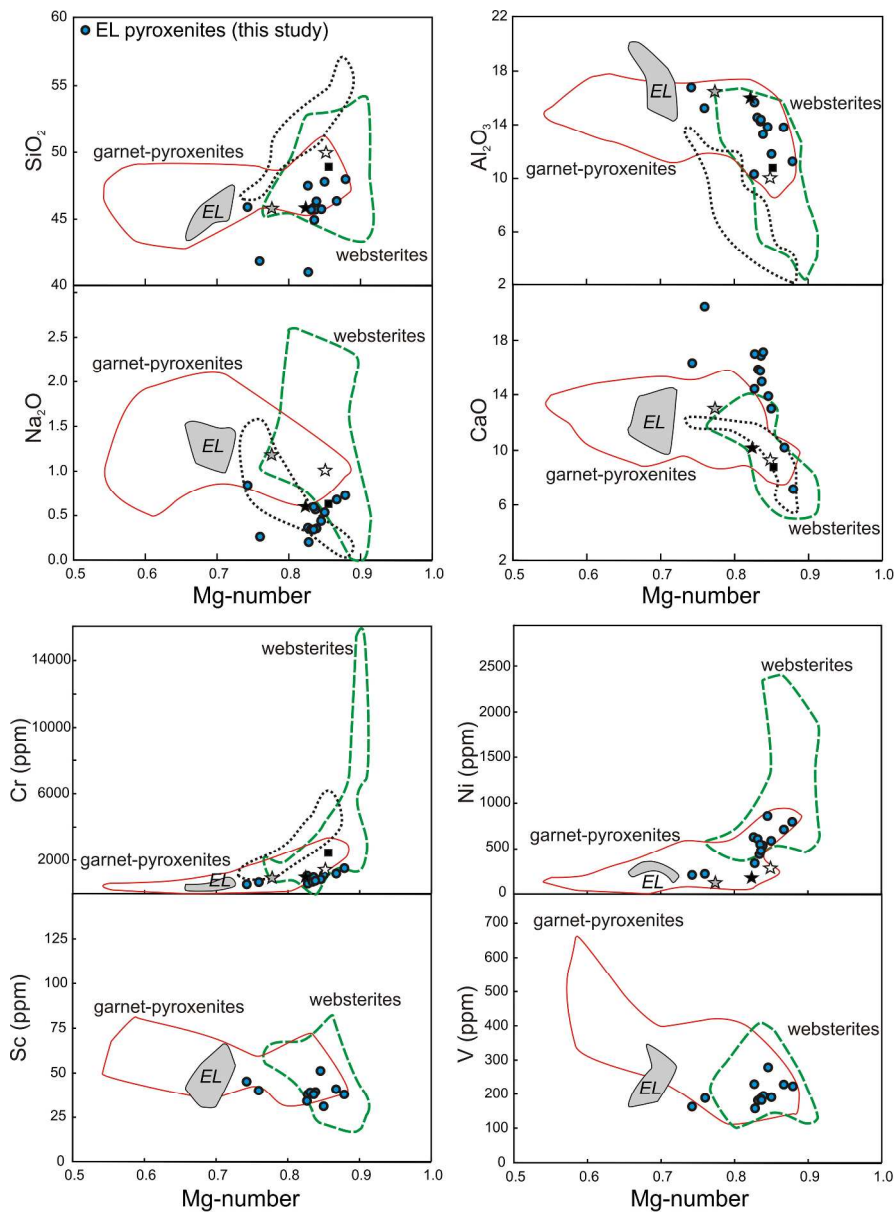
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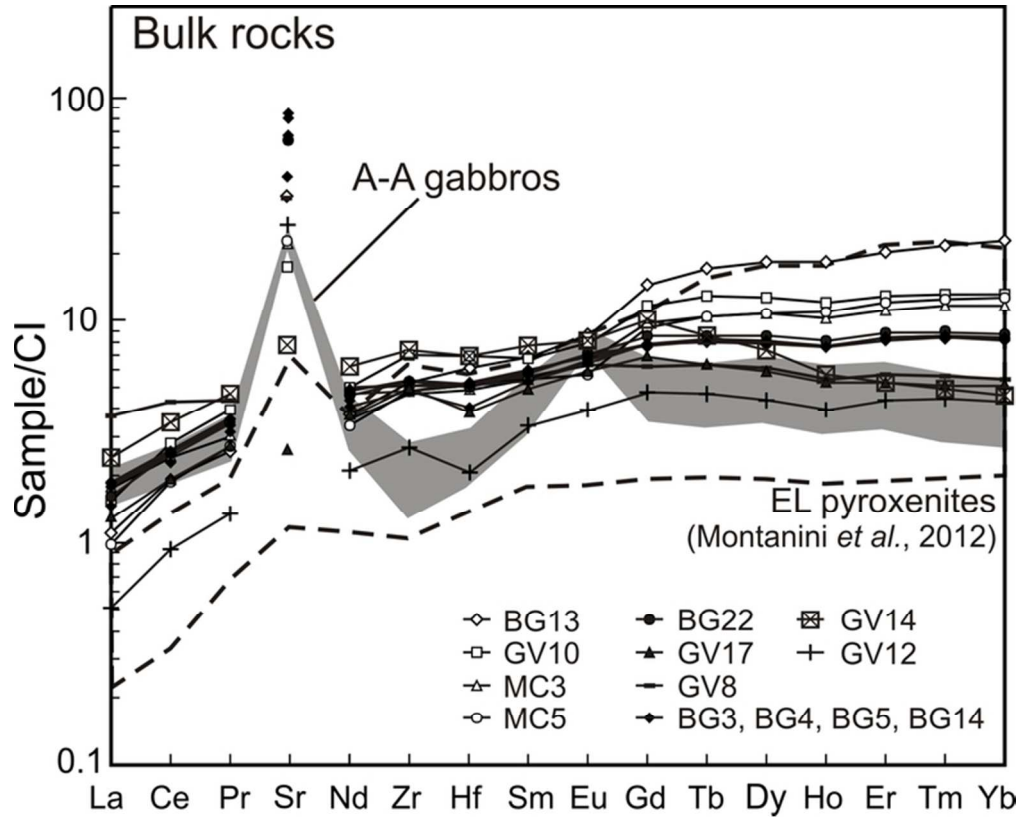


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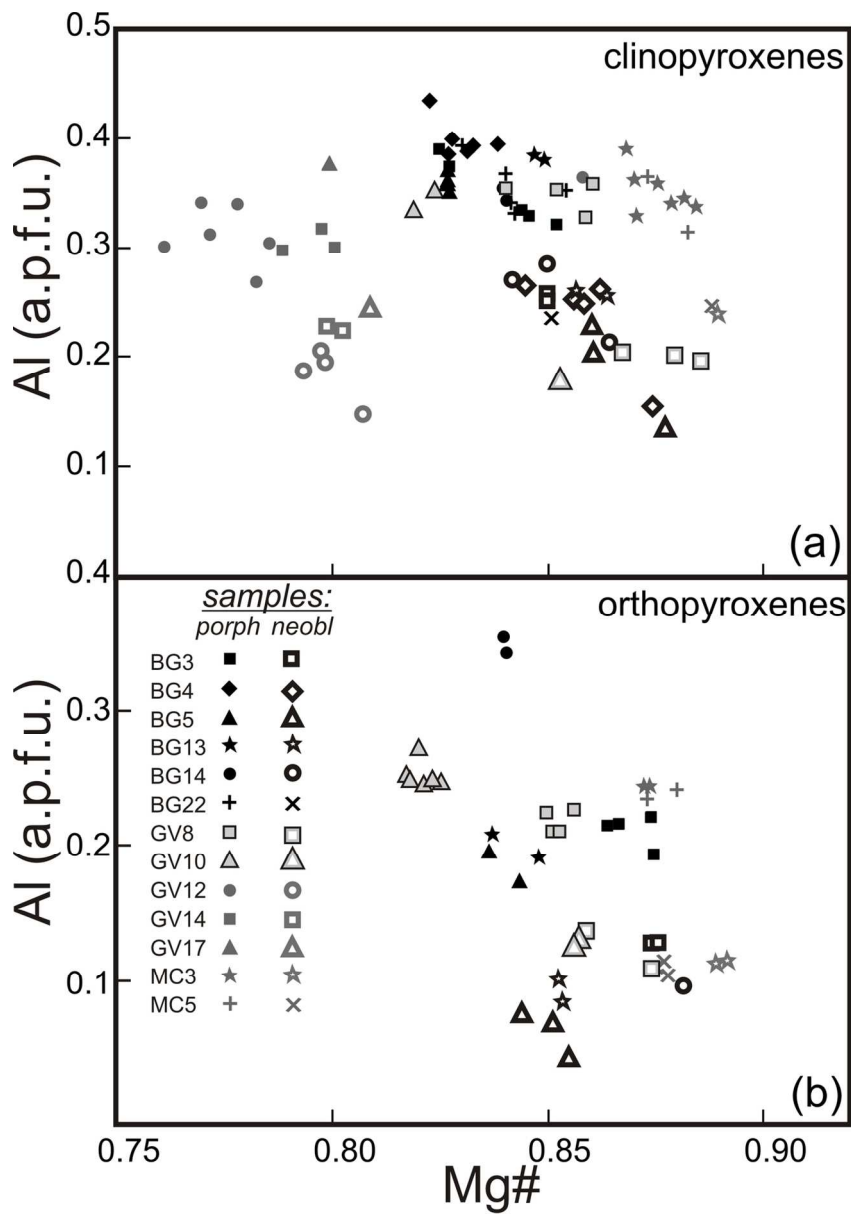
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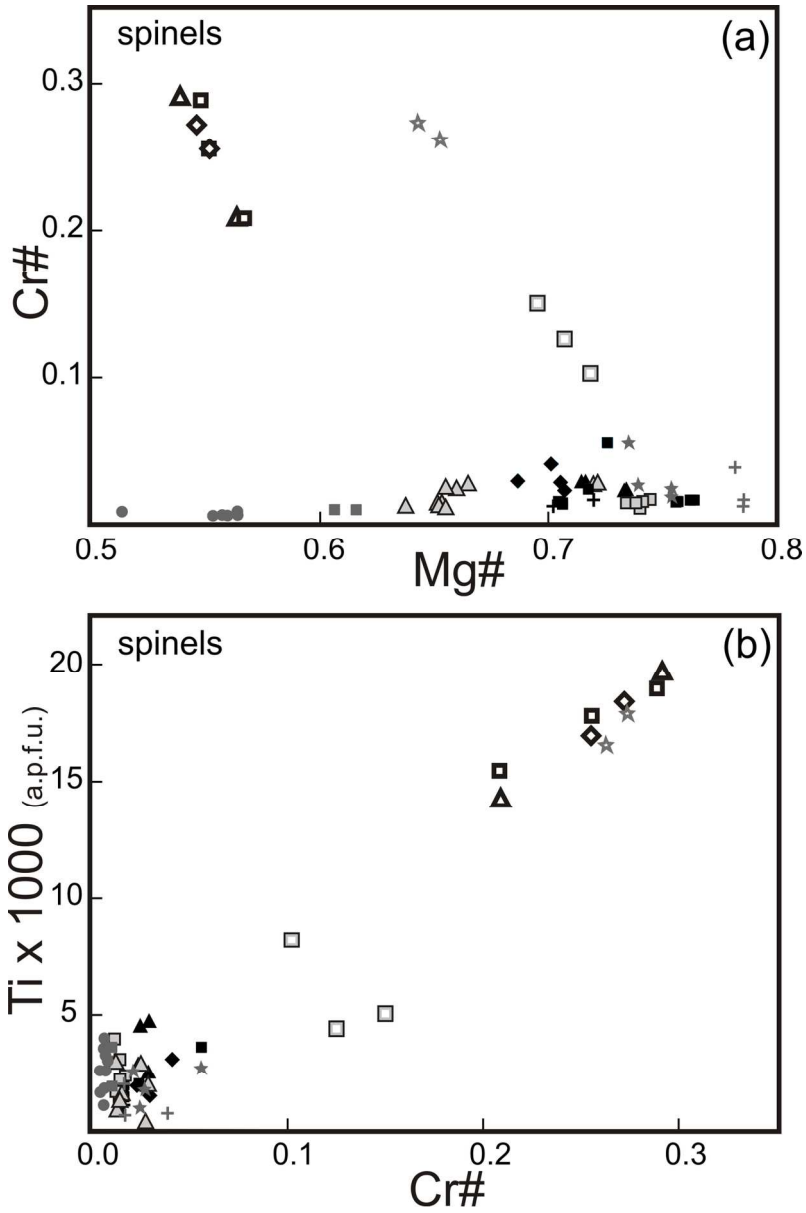
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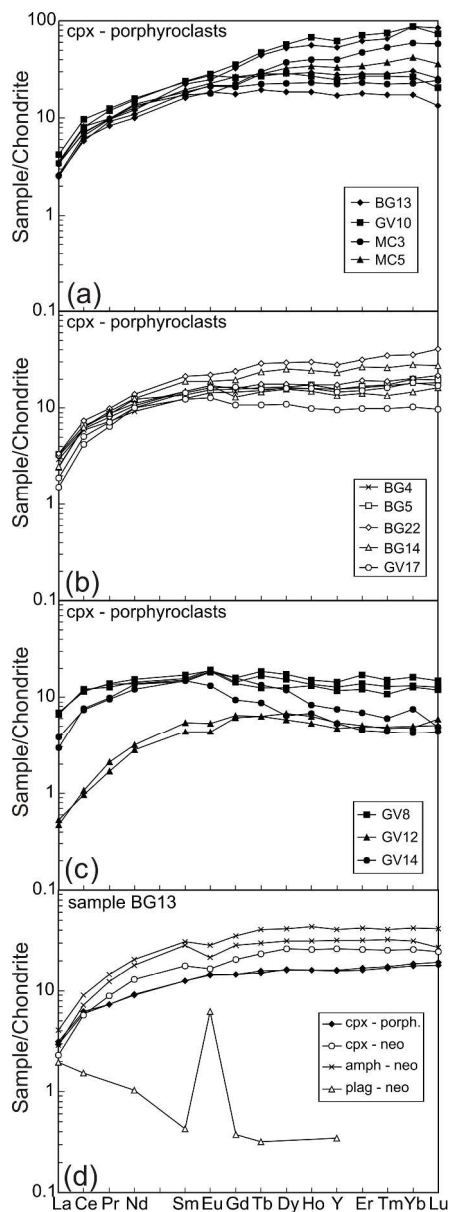


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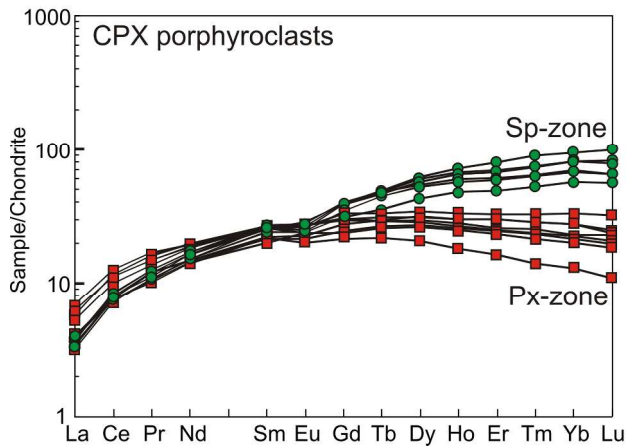
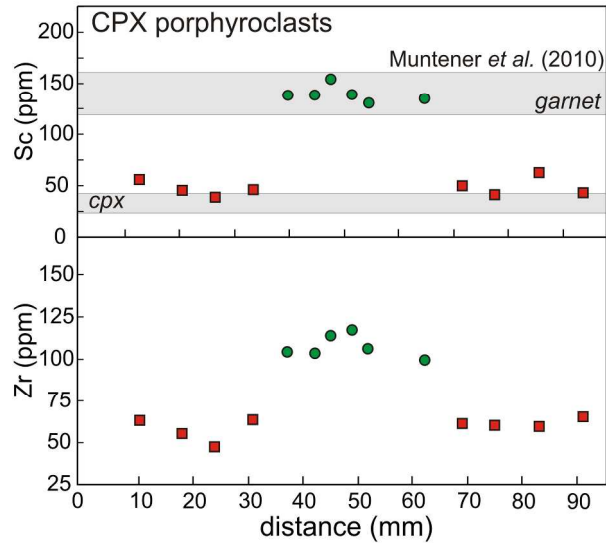
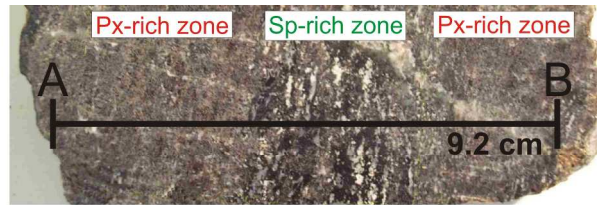
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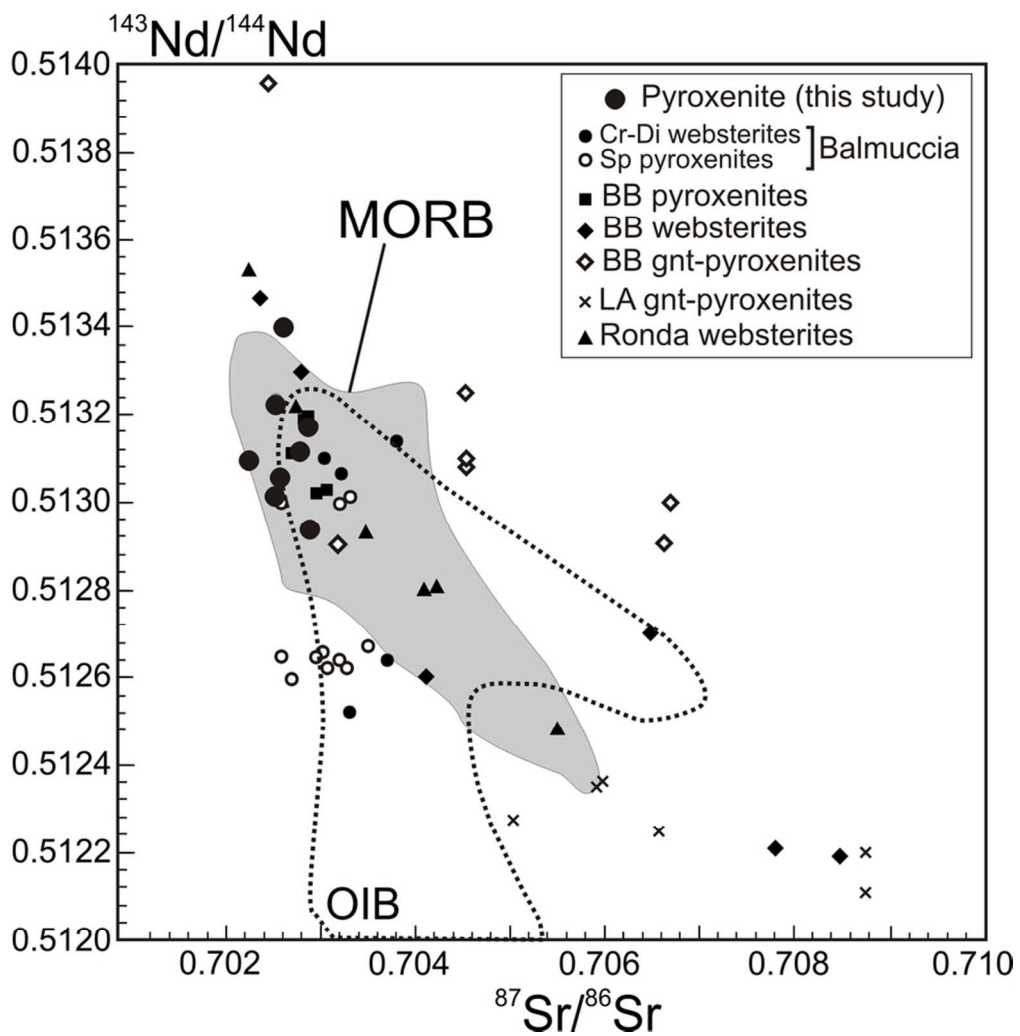
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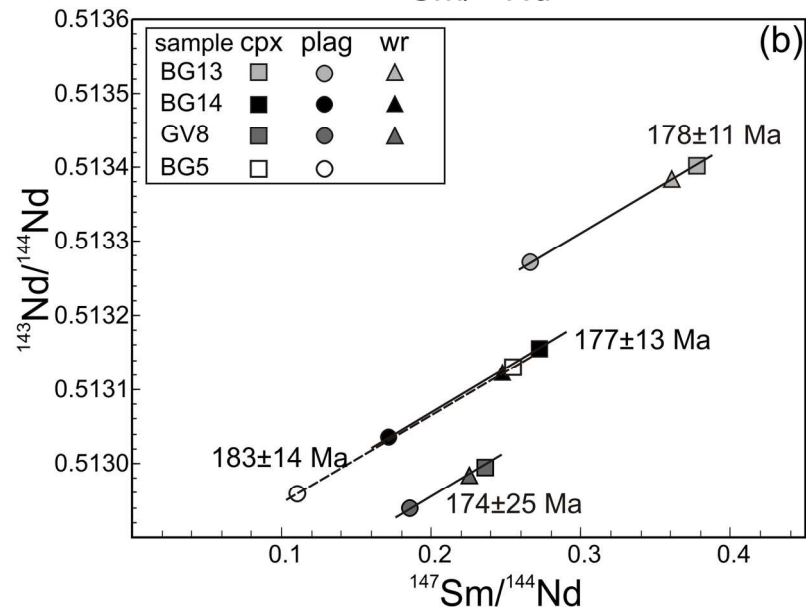
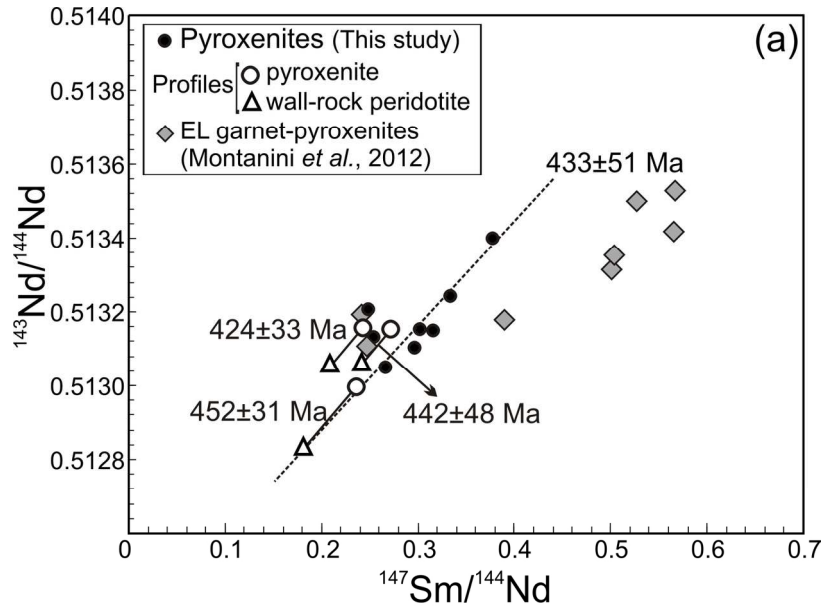
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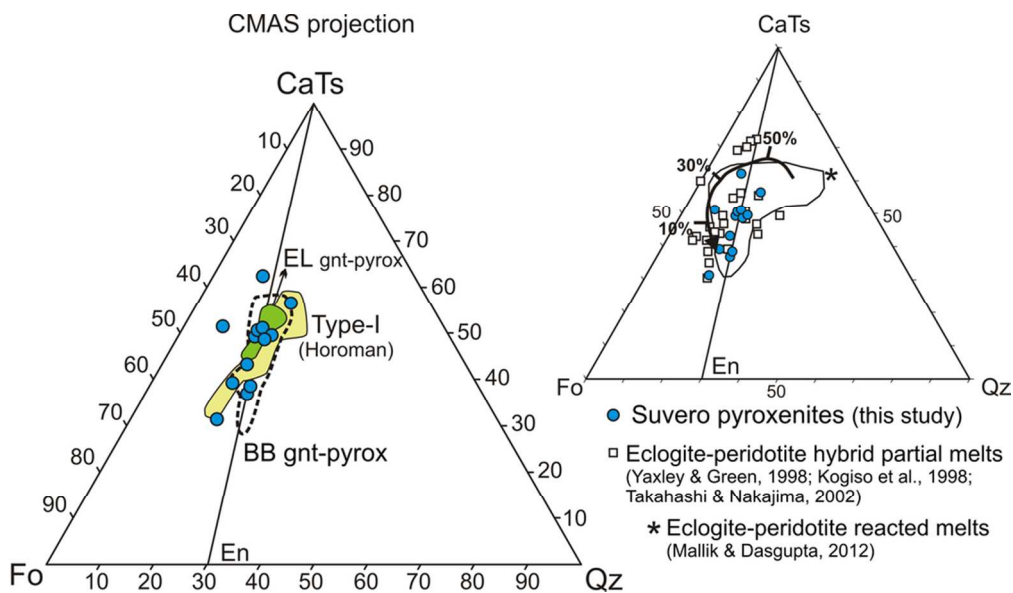
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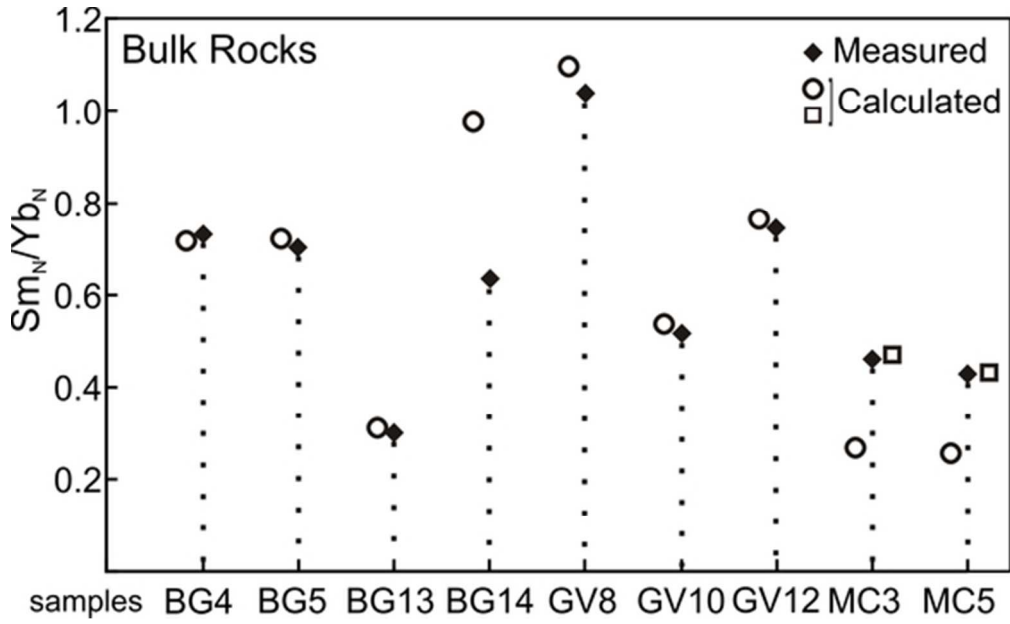
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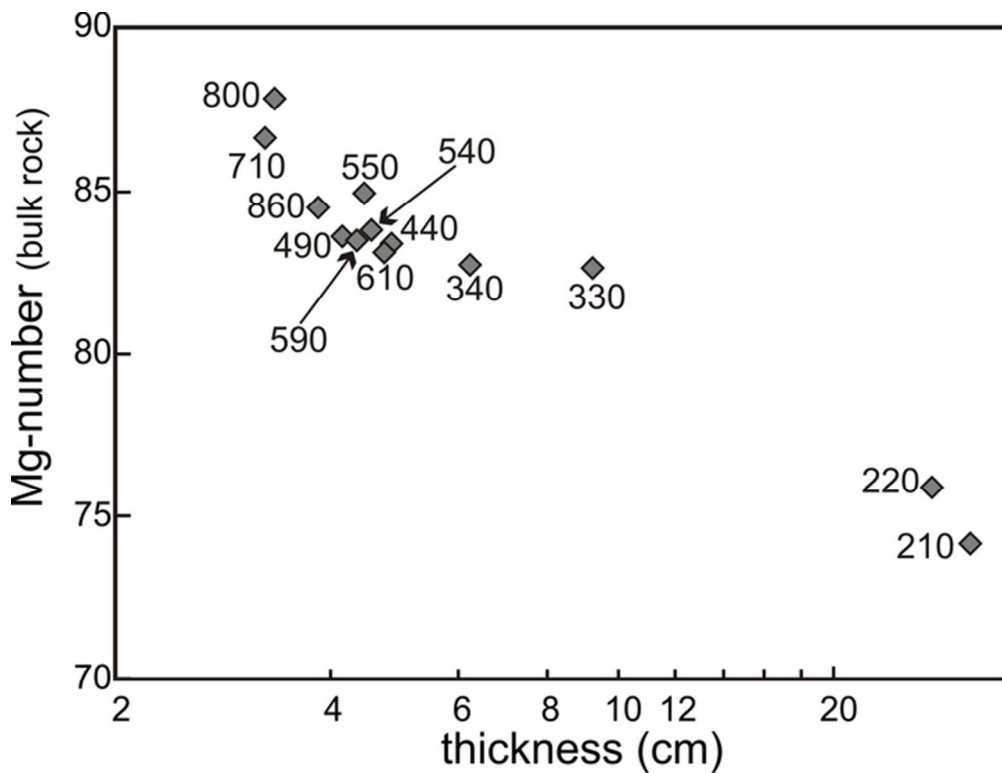
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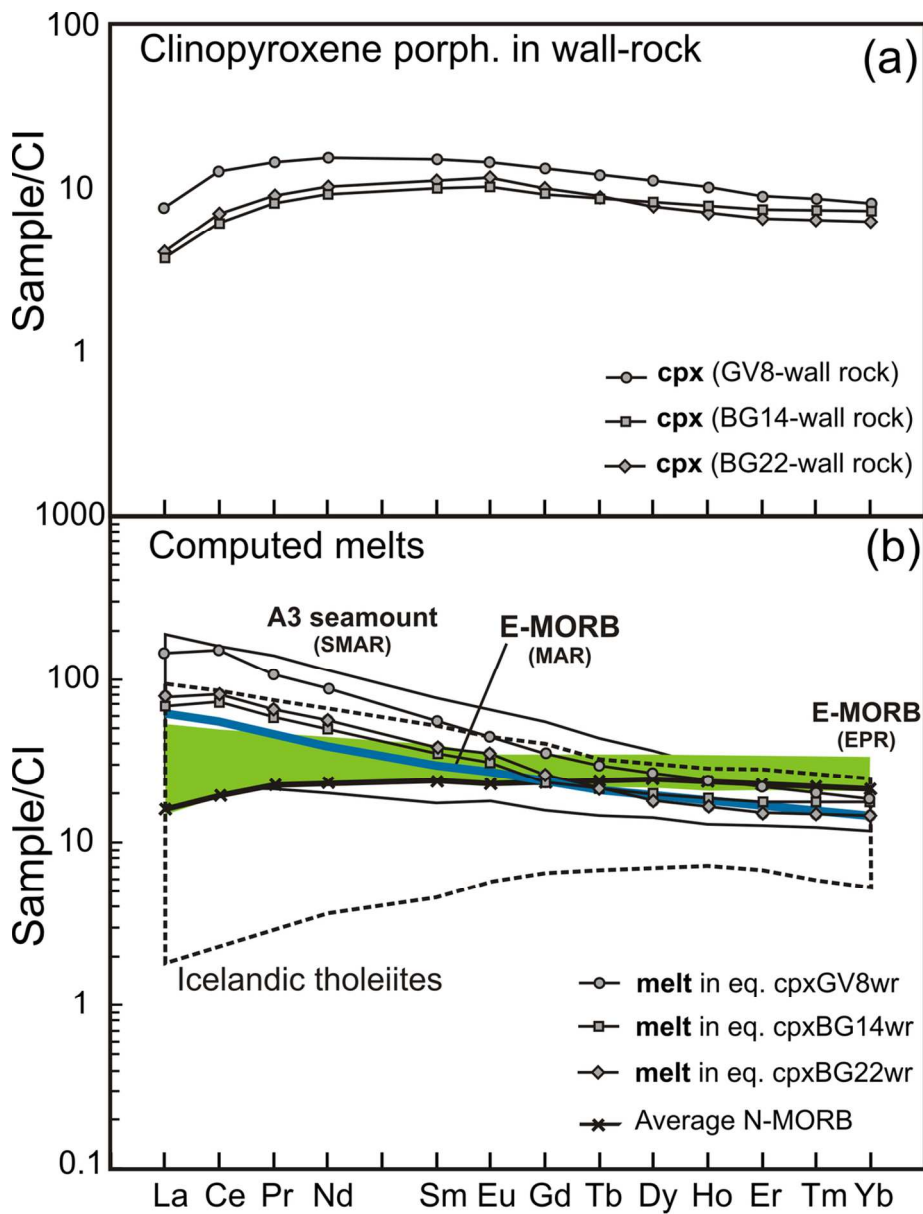
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Table 1: Localization, characteristics and texture of pyroxenites

Sample	latitude	longitude	Geometry	Rock-type	Texture
BG3P	44°16'44.68"	9°47'49.96"	4.3 cm layer	cpx-rich websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
BG4"	44°16'44.36"	9°47'49.08"	4.1 cm layer	clinopyroxenite	porphyroclastic sp+cpx; plag-bearing neobalstic assemblage
BG5	44°16'44.57"	9°47'48.64"	4.8 cm layer	cpx-rich websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
BG13	44°16'45.93"	9°47'48.93"	3.8 cm layer	cpx-rich websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
BG14	44°16'44.15"	9°47'50.15"	4.7 cm layer	cpx-rich websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
BG22	44°16'44.69"	9°47'49.57"	4.5 cm layer	cpx-rich websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
GV8	44°16'15.65"	9°48'27.76"	4.4 cm layer	websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
GV10	44°16'15.86"	9°48'26.82"	9.2 cm layer	websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
GV12	44°16'15.95"	9°48'27.14"	31 cm lens	clinopyroxenite	clusters and seams of dark green sp + cpx + plg on cpx porph.
GV14	44°16'15.95"	9°48'27.14"	27 cm lens	clinopyroxenite	porphyroclastic sp+cpx; plag-bearing neobalstic assemblage
GV17	44°16'28.53"	9°48'17.59"	6.2 cm layer	clinopyroxenite	porphyroclastic sp+cpx; plag-bearing neobalstic assemblage
MC3	44°17'38.25"	9°45'42.58"	3.2 cm layer	websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage
MC5	44°17'37.54"	9°45'48.54"	3.3 cm layer	websterite	porphyroclastic sp+pxs; plag-bearing neobalstic assemblage

Table 2: Bulk-rock major and trace element compositions

Sample	BG3P	BG4 ^m	BG5	BG13	BG14*	BG22*	GV8*	GV10	GV12	GV14	GV17	MC3	MC5
SiO ₂	43.05	43.96	44.04	43.84	43.62	44.38	46.46	45.99	45.05	39.93	37.76	43.19	45.19
TiO ₂	0.35	0.35	0.36	0.42	0.34	0.35	0.45	0.44	0.25	0.57	0.34	0.44	0.39
Al ₂ O ₃	13.67	13.70	13.83	13.28	13.86	12.75	11.50	9.94	16.46	14.59	14.44	12.87	10.61
Fe ₂ O ₃	6.25	6.40	6.19	6.55	6.35	5.98	6.71	7.77	8.11	8.14	7.05	6.18	6.52
MnO	0.14	0.14	0.14	0.19	0.14	0.14	0.14	0.16	0.14	0.15	0.16	0.17	0.17
MgO	15.99	16.52	15.73	18.06	15.80	15.65	19.17	18.53	11.78	12.98	17.04	20.29	23.91
CaO	16.16	14.37	15.14	13.33	15.13	16.41	12.64	13.96	16.03	19.52	15.68	9.52	6.77
Na ₂ O	0.33	0.55	0.57	0.42	0.33	0.34	0.52	0.48	0.83	0.25	0.19	0.63	0.68
K ₂ O	0.23	0.62	0.53	0.20	0.28	0.18	0.07	0.10	0.14	0.00	0.00	0.31	0.36
P ₂ O ₅	0.02	0.02	0.02	0.02	0.03	0.02	0.04	0.00	0.00	0.02	0.00	0.02	0.00
LOI	4.73	3.75	3.62	4.48	4.98	4.66	3.13	2.82	1.62	4.83	6.43	5.98	5.41
Tot	100.92	100.38	100.17	100.79	100.86	100.86	100.83	100.35	100.41	100.98	99.09	99.59	100.02
Mg-no	83.51	83.64	83.42	84.52	83.13	83.82	84.98	82.53	74.06	76.11	82.72	86.67	87.90
Cr	980	810	740	890	920	820	1140	1090	560	720	610	1230	1530
Ni	550	490	440	860	610	540	590	630	210	220	340	710	800
Sr	340	486	648	331	602	491	274	132	196	57	19	166	173
Y	11.4	12.2	12.5	27.7	11.8	12.5	8.2	18.5	6.2	8.4	7.8	15.5	17
Zr	18	18	20	20	18	20	19	26	10	28	18	18	19
La	0.44	0.42	0.43	0.26	0.34	0.38	0.87	0.36	0.12	0.56	0.3	0.42	0.23
Ce	1.53	1.53	1.55	1.16	1.39	1.53	2.6	1.67	0.56	2.09	1.15	1.48	1.13
Pr	0.3	0.32	0.32	0.23	0.28	0.31	0.39	0.35	0.12	0.42	0.24	0.27	0.24
Nd	1.98	2.21	2.17	1.69	1.84	2.1	2.1	2.27	0.95	2.79	1.64	1.71	1.51
Sm	0.78	0.89	0.85	1.01	0.78	0.86	0.83	1	0.49	1.13	0.72	0.8	0.8
Eu	0.371	0.385	0.392	0.487	0.357	0.394	0.352	0.443	0.222	0.453	0.33	0.363	0.321
Gd	1.55	1.53	1.5	2.84	1.54	1.66	1.22	2.28	0.93	1.97	1.34	1.9	1.83
Tb	0.3	0.29	0.29	0.62	0.29	0.31	0.23	0.47	0.17	0.31	0.23	0.38	0.38
Dy	1.95	1.89	1.92	4.44	1.93	2.05	1.48	3.08	1.07	1.77	1.44	2.59	2.59
Ho	0.42	0.42	0.42	1.02	0.42	0.45	0.3	0.67	0.22	0.32	0.29	0.57	0.61
Er	1.28	1.32	1.3	3.21	1.32	1.4	0.91	2.04	0.69	0.84	0.83	1.77	1.93
Tm	0.201	0.207	0.199	0.522	0.205	0.213	0.137	0.316	0.108	0.118	0.124	0.282	0.301
Yb	1.32	1.34	1.33	3.68	1.35	1.41	0.88	2.12	0.7	0.75	0.83	1.91	2.05
Lu	0.208	0.197	0.211	0.665	0.208	0.223	0.141	0.355	0.103	0.113	0.128	0.301	0.331
Hf	0.5	0.5	0.5	0.6	0.4	0.5	0.5	0.7	0.2	0.7	0.4	0.5	0.5
Sc	38	38	38	51	39	39	31	34	45	40	38	41	38
V	187	183	186	280	182	195	191	231	163	190	159	230	225

(*) data are from Borghini et al. (2013). Major elements are in wt%; trace elements in ppm. Mg-no is the Mg-number.

Table 3: Average major element compositions of clinopyroxene in pyroxenites

sample	BG3		BG4		BG5		BG13		BG14		BG22		GV8	
	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl
<i>p.to an.</i>	5	3	6	5	7	3	5	2	3	3	5	1	7	3
<i>N. an.</i>	5	3	6	5	7	3	5	2	3	3	5	1	7	3
SiO ₂	49.88 (43)	51.16 (15)	49.01 (17)	50.79 (69)	49.22 (11)	51.42 (11)	48.95 (5)	50.58 (64)	49.53 (2)	50.69 (94)	49.63 (69)	51.07	49.58 (43)	51.40 (14)
TiO ₂	0.84 (13)	0.99 (9)	0.69 (7)	1.09 (19)	0.67 (3)	1.07 (45)	1.02 (2)	0.95 (3)	0.92 (9)	0.97 (5)	0.77 (10)	0.71	1.20 (7)	1.30 (18)
Al ₂ O ₃	8.21 (70)	5.96 (8)	9.29 (42)	5.5 (108)	8.32 (18)	4.41 (112)	8.84 (7)	5.98 (14)	8.18 (15)	5.95 (83)	8.33 (50)	5.45	8.13 (29)	4.67 (11)
Cr ₂ O ₃	0.13 (2)	0.17 (10)	0.20 (7)	0.34 (10)	0.15 (4)	0.24 (6)	0.17 (3)	0.18 (9)	0.10 (5)	0.12 (5)	0.15 (10)	0.19	0.20 (3)	0.19 (13)
FeO	4.80 (30)	4.65 (1)	4.86 (17)	4.42 (26)	4.93 (3)	4.22 (18)	4.16 (7)	4.16 (28)	4.68 (3)	4.55 (24)	4.65 (28)	4.56	4.32 (24)	3.78 (33)
MgO	14.01 (48)	14.76 (2)	13.29 (20)	15.13 (53)	13.19 (6)	15.35 (75)	13.01 (6)	14.32 (36)	13.76 (15)	14.72 (60)	13.87 (28)	14.60	14.01 (44)	15.18 (18)
MnO	0.17 (3)	0.18 (1)	0.21 (3)	0.20 (6)	0.00	0.00	0.24 (1)	0.22 (1)	0.22 (1)	0.20 (2)	0.20 (5)	0.19	0.12 (7)	0.13 (4)
CaO	22.08 (43)	22.22 (9)	21.95 (39)	21.84 (56)	21.88 (11)	22.36 (23)	21.86 (13)	22.04 (42)	22.24 (14)	22.30 (18)	21.89 (29)	22.28	22.01 (34)	22.19 (36)
Na ₂ O	0.77 (6)	0.73 (1)	0.69 (19)	0.65 (8)	0.79 (3)	0.65 (4)	0.86 (4)	0.63 (2)	0.83 (3)	0.67 (4)	0.81 (5)	0.64	0.70 (7)	0.60 (6)
Total	100.89	100.83	100.19	99.95	99.15	99.72	99.17	99.15	100.46	100.17	100.24	99.70	100.25	99.44
Mg#	0.84	0.85	0.83	0.86	0.83	0.87	0.85	0.86	0.84	0.85	0.84	0.85	0.85	0.88

sample	GV10		GV12		GV14		GV17		MC3		MC5	
	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl
<i>p.to an.</i>	11	1	6	4	5	2	1	1	7	1	5	1
<i>N. an.</i>	11	1	6	4	5	2	1	1	7	1	5	1
SiO ₂	49.20 (15)	51.32	50.37 (38)	51.59 (25)	49.73 (26)	50.63 (3)	48.58	50.06	49.45 (43)	50.26	49.31 (52)	50.31
TiO ₂	1.13 (3)	0.86	0.43 (7)	0.39 (5)	1.13 (19)	0.81 (2)	0.70	0.75	1.17 (18)	1.58	1.03 (10)	1.65
Al ₂ O ₃	8.02 (30)	4.15	7.18 (63)	4.23 (58)	7.06 (25)	5.18 (6)	8.75	5.65	8.25 (44)	5.62	8.08 (67)	5.70
Cr ₂ O ₃	0.30 (7)	0.18	0.04 (2)	0.05 (3)	0.08 (4)	0.05 (2)	0.14	0.19	0.22 (10)	0.23	0.13 (7)	0.21
FeO	5.44 (24)	4.81	7.00 (65)	6.58 (25)	6.14 (22)	6.28 (5)	6.80	6.32	3.59 (15)	3.36	3.72 (32)	3.38
MgO	14.02 (29)	15.64	13.47 (79)	14.66 (20)	13.38 (18)	14.14 (36)	15.18	14.99	14.16 (34)	15.11	14.12 (47)	15.08
MnO	0.22 (15)	0.18	0.16 (5)	0.19 (1)	0.22 (2)	0.19 (3)	0.23	0.22	0.13 (2)	0.12	0.17 (2)	0.12
CaO	21.78 (35)	22.38	20.27 (96)	21.29 (35)	21.55 (20)	21.39 (33)	19.32	21.40	22.04 (38)	22.15	22.31 (2)	22.13
Na ₂ O	0.59 (45)	0.43	0.70 (7)	0.54 (4)	0.67 (6)	0.70 (5)	0.24	0.37	0.86 (11)	0.72	0.80 (3)	0.77
Total	100.70	99.95	99.63	99.51	99.95	99.37	99.94	99.95	99.88	99.15	99.68	99.35
Mg#	0.82	0.85	0.77	0.80	0.80	0.80	0.80	0.81	0.88	0.89	0.87	0.89

N. an., number of analyses. Values in parentheses represent 2 σ standard deviation of the last unit cited. Abbreviations: porph = porphyroclast, neobl = neoblast.

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Table 4: Average major element compositions of orthopyroxene in pyroxenites

sample	BG3	BG3	BG5	BG5	BG13	BG13	BG14	BG14	GV8	GV8	GV10	GV10	MC3	MC3	MC5	MC5
<i>p.to an.</i>	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl	porph	neobl
<i>N. an.</i>	4	2	2	3	2	2	2	2	4	2	10	3	3	2	2	3
SiO ₂	54.57 (43)	55.28 (11)	54.51 (24)	55.81 (33)	54.37 (28)	55.22 (13)	54.56 (8)	55.44 (3)	53.52 (17)	55.08 (40)	53.19 (29)	55.34 (12)	54.12 (10)	55.90 (2)	53.05 (21)	55.23 (11)
TiO ₂	0.22 (2)	0.28 (4)	0.23 (1)	0.17 (2)	0.23 (3)	0.36 (1)	0.16 (1)	0.16 (1)	0.23 (3)	0.29 (4)	0.24 (6)	0.21 (3)	0.17 (2)	0.28 (1)	0.21 (3)	0.29 (2)
Al ₂ O ₃	5.26 (29)	3.21 (1)	4.61 (40)	1.64 (42)	4.95 (25)	2.36 (30)	4.85 (41)	2.50 (8)	5.34 (23)	3.10 (47)	6.55 (55)	3.23 (9)	6.04 (3)	2.90 (6)	5.84 (14)	2.77 (11)
Cr ₂ O ₃	0.24 (6)	0.25 (2)	0.12 (1)	0.09 (3)	0.12 (1)	0.11 (9)	0.35 (3)	0.32 (1)	0.10 (2)	0.26 (7)	0.16 (7)	0.13 (2)	0.34 (3)	0.22 (1)	0.08 (2)	0.12 (1)
FeO	8.38 (38)	8.26 (1)	10.33 (30)	9.94 (20)	10.05 (27)	9.51 (3)	7.89 (13)	7.84 (6)	9.26 (15)	8.70 (67)	10.41 (99)	9.30 (7)	8.10 (4)	7.08 (10)	7.94 (30)	8.07 (5)
MgO	31.32 (21)	32.10 (6)	30.36 (23)	31.64 (76)	30.11 (93)	30.93 (8)	32.22 (7)	32.58 (2)	29.96 (24)	31.66 (45)	29.25 (69)	31.26 (9)	31.09 (7)	32.32 (5)	31.55 (18)	32.25 (29)
MnO	0.17 (2)	0.19 (1)	0.17 (5)	0.09 (9)	0.17 (10)	0.41 (1)	0.18 (4)	0.20 (1)	0.21 (5)	0.23 (1)	0.21 (6)	0.17 (2)	0.16 (6)	0.20 (0)	0.22 (7)	0.27 (2)
CaO	0.72 (5)	0.42 (2)	0.71 (23)	0.59 (18)	0.72 (21)	0.52 (6)	0.64 (17)	0.49 (5)	0.94 (18)	0.51 (3)	0.69 (11)	0.36 (2)	0.72 (4)	0.49 (4)	0.50 (12)	0.50 (1)
Na ₂ O	0.02 (1)	0.02 (1)	0.15 (10)	0.01 (1)	0.13 (9)	0.04 (2)	0.02 (2)	0.01 (1)	0.04 (3)	0.02 (2)	0.03 (2)	0.00	0.01 (0)	0.03 (1)	0.05 (1)	0.02 (0)
Total.	100.90	100.01	101.20	99.99	100.84	99.47	100.88	99.55	99.61	99.83	100.72	99.99	100.75	99.47	99.43	99.51
Mg#	0.87	0.87	0.84	0.85	0.84	0.85	0.88	0.88	0.85	0.87	0.83	0.86	0.87	0.89	0.88	0.88

N. an., number of analyses. Values in parentheses represent 2 σ standard deviation of the last unit cited. Abbreviations as in Table 3.

Table 5: Representative major element compositions of spinel in pyroxenites

sample	BG3	BG3	BG4	BG4	BG5	BG5	BG22	GV8	GV8	GV10	GV12	GV14	MC3	MC3	MC5
<i>p.to an.</i>	porph	neobl	porph	neobl	porph	neobl	porph	porph	neobl	porph	porph	porph	porph	neobl	porph
<i>N. an.</i>	9	3	4	2	4	2	4	10	5	10	8	2	4	3	4
TiO ₂	0.09 (4)	0.81 (7)	0.11 (4)	0.82 (4)	0.19 (6)	0.78 (16)	0.08 (4)	0.10 (5)	0.29 (10)	0.10 (5)	0.12 (4)	0.14 (6)	0.11 (4)	0.81 (4)	0.06 (4)
Al ₂ O ₃	62.99 (143)	39.28 (327)	61.87 (74)	38.90 (91)	59.85 (74)	38.85 (478)	63.52 (144)	61.56 (16)	52.00 (224)	61.84 (105)	61.26 (66)	57.85 (60)	63.92 (158)	38.35 (28)	62.60 (111)
Cr ₂ O ₃	2.01 (125)	19.54 (263)	2.93 (70)	20.78 (78)	2.47 (28)	19.20 (362)	1.40 (19)	1.35 (15)	11.16 (194)	1.91 (68)	0.62 (12)	0.91 (4)	3.17 (154)	21.01 (64)	2.26 (124)
FeO	15.37 (155)	26.70 (61)	16.62 (53)	26.44 (4)	17.19 (40)	27.00 (117)	16.72 (2)	15.07 (34)	16.23 (68)	19.51 (161)	22.22 (65)	23.39 (34)	12.89 (21)	23.94 (4)	12.36 (22)
MgO	19.28 (78)	13.21 (48)	18.27 (28)	13.06 (14)	18.70 (28)	13.09 (76)	18.64 (6)	18.79 (12)	17.32 (52)	17.26 (84)	13.85 (55)	15.16 (18)	19.57 (48)	15.56 (14)	20.30 (19)
MnO	0.00	0.12 (2)	0.00	0.12 (1)	0.00	0.00	0.15 (2)	0.12 (2)	0.11 (3)	0.17 (5)	0.15 (4)	0.19 (1)	0.12 (2)	0.12 (1)	0.13 (2)
NiO	0.00	0.00	0.00	0.00	0.00	0.00	0.33 (5)	0.39 (4)	0.33 (3)	0.44 (13)	0.18 (5)	0.19 (2)	0.34 (4)	0.00	0.33 (8)
Total	99.74	99.66	99.80	100.11	98.39	98.92	100.87	97.38	97.44	101.23	98.40	97.84	100.12	99.78	98.04
Mg#	0.73	0.56	0.70	0.55	0.72	0.55	0.71	0.74	0.71	0.67	0.55	0.61	0.75	0.65	0.78
Cr#	0.02	0.25	0.03	0.26	0.03	0.25	0.01	0.01	0.13	0.02	0.01	0.01	0.03	0.27	0.02
Tix1000	1.80	17.38	2.08	17.63	3.68	16.96	1.58	2.00	5.85	1.86	2.46	2.76	2.06	17.23	1.23

N. an., number of analyses. Values in parentheses represent 2σ standard deviation of the last unit cited. Abbreviations as in Table 3.

Table 6: Representative major element compositions of olivine in pyroxenites

sample	BG3	BG4	BG5	BG13	BG14	BG22	GV8	GV10	MC3	MC5
N. an.	6	4	5	7	4	6	6	4	4	5
SiO ₂	40.17 (6)	39.94 (28)	40.04 (37)	40.09 (43)	40.45 (8)	40.21 (3)	40.22 (15)	39.80 (14)	40.74 (40)	40.53 (10)
FeO	14.60 (11)	14.92 (95)	15.31 (85)	14.72 (29)	14.78 (4)	14.75 (9)	12.69 (5)	18.13 (16)	12.47 (29)	12.22 (12)
MgO	44.72 (8)	44.69 (69)	44.61 (64)	44.46 (48)	45.46 (15)	45.37 (16)	46.64 (13)	42.60 (40)	46.19 (96)	47.17 (2)
MnO	0.43 (1)	0.44 (5)	0.34 (8)	0.47 (5)	0.27 (2)	0.26 (6)	0.22 (1)	0.40 (3)	0.23 (3)	0.22 (3)
NiO	0.26 (1)	0.28 (0)	0.35 (12)	0.27 (10)	0.12 (2)	0.12 (1)	0.32 (1)	0.25 (5)	0.26 (1)	0.25 (2)
CaO	0.03 (1)	0.04 (1)	0.03 (0)	0.01 (1)	0.02 (1)	0.02 (1)	0.02 (1)	0.03 (1)	0.01 (1)	0.01 (0)
Total	100.21	100.28	100.69	100.04	101.12	100.74	100.11	101.22	99.91	100.41
Fo	0.85	0.84	0.84	0.84	0.85	0.85	0.87	0.81	0.87	0.87

N. an., number of analyses. Values in parentheses represent 2 σ standard deviation of the last unit cited.

Table 7: Average major element compositions of plagioclase in pyroxenites

sample site	BG4		BG5		BG13	GV8		GV10	GV12		MC3	MC5
	core	rim	core	rim	core	core	rim	core	core	rim	core	core
N. an.	5	3	7	5	3	4	3	5	6	5	5	4
SiO ₂	52.21 (97)	47.80 (31)	52.27 (51)	47.70 (5)	52.78 (57)	53.61 (60)	49.85 (6)	53.59 (63)	52.60 (60)	47.84 (3)	53.52 (31)	53.61 (40)
TiO ₂	0.02 (2)	0.02 (0)	0.02 (2)	0.03 (2)	0.04 (1)	0.06 (6)	0.00	0.01 (1)	0.01 (2)	0.04 (1)	0.02 (2)	0.04 (5)
Al ₂ O ₃	30.70 (61)	33.24 (17)	30.16 (20)	33.45 (11)	30.61 (45)	29.20 (43)	31.89 (4)	30.12 (41)	29.70 (46)	33.00 (13)	29.90 (31)	28.91 (22)
FeO	0.21 (10)	0.39 (8)	0.21 (6)	0.33 (15)	0.18 (2)	0.12 (1)	0.40 (4)	0.14 (3)	0.33 (15)	0.33 (10)	0.19 (4)	0.17 (7)
MgO	0.00	0.12 (2)	0.03 (3)	0.10 (7)	0.01 (1)	0.01 (1)	0.00	0.02 (1)	0.01 (1)	0.01 (1)	0.01 (1)	0.01 (1)
CaO	13.12 (16)	16.28 (21)	13.05 (16)	16.59 (15)	12.43 (37)	12.33 (17)	15.51 (8)	12.00 (41)	13.36 (36)	16.65 (7)	12.19 (66)	11.87 (60)
Na ₂ O	4.08 (12)	2.27 (10)	4.14 (13)	2.09 (7)	4.47 (20)	4.52 (14)	2.71 (11)	4.77 (23)	3.96 (20)	1.97 (5)	4.69 (44)	4.78 (37)
K ₂ O	0.02 (1)	0.01 (1)	0.00	0.00	0.05 (1)	0.01 (1)	0.01 (1)	0.02 (1)	0.03 (1)	0.02 (1)	0.02 (1)	0.02 (1)
Total	100.36	100.10	99.89	100.31	100.58	99.86	100.38	100.66	100.01	99.85	100.54	99.41
An	0.64	0.80	0.64	0.81	0.61	0.60	0.76	0.58	0.65	0.82	0.59	0.58

N. an., number of analyses. Values in parentheses represent 2 σ standard deviation of the last unit cited.

Table 8: Representative major element compositions of amphibole in pyroxenites

sample	BG4		BG5		BG13		BG14	BG22	GV8		GV10		MC3		MC5	
	Ks	Ti-prg	Ti-prg	Ti-prg	Ks	Ks	Ks	Ti-prg	Ks	Ti-prg	Ks	Ti-prg	Ks	Ks	Ks	Ks
SiO ₂	42.07	42.94	42.61	42.36	41.47	40.56	41.96	42.25	41.62	42.33	41.12	41.49	42.22	41.76	41.51	41.98
TiO ₂	4.66	4.37	4.43	4.56	4.90	5.01	4.84	3.89	5.25	4.33	5.00	3.74	5.41	5.60	5.44	5.31
Al ₂ O ₃	12.45	12.3	12.84	12.42	14.23	13.79	13.37	13.78	12.09	11.77	12.84	14.58	13.47	13.75	12.34	12.36
Cr ₂ O ₃	0.53	0.68	1.1	0.57	0.67	0.74	0.56	0.53	0.21	0.22	0.28	0.28	0.54	0.57	0.27	0.62
FeO	6.9	6.68	6.93	6.73	5.92	6.10	6.47	7.01	6.29	6.11	8.01	7.51	5.50	5.25	5.25	5.10
MgO	15.66	15.78	15.06	15.5	15.47	15.10	15.06	15.31	15.22	15.84	14.75	14.75	15.75	15.39	15.50	15.61
MnO	0.22	0.25	0.07	0.19	0.15	0.24	0.11	0.15	0.08	0.08	0.20	0.08	0.12	0.09	0.10	0.08
CaO	11.78	11.76	12.18	11.48	12.02	11.89	11.82	11.94	11.69	11.88	11.92	11.81	11.89	11.96	11.75	11.55
Na ₂ O	3.88	3.44	2.91	3.68	2.86	3.28	3.42	3.27	3.24	3.13	3.37	3.42	3.54	3.47	3.39	3.61
K ₂ O	0.14	0.15	0.03	0.13	0.22	0.29	0.17	0.21	0.07	0.07	0.05	0.03	0.07	0.07	0.04	0.02
Total	98.29	98.35	98.16	97.62	97.91	97.00	97.77	98.34	95.76	95.76	97.54	97.69	98.51	97.92	95.59	96.25
Mg#	0.80	0.81	0.79	0.80	0.82	0.82	0.81	0.80	0.81	0.82	0.77	0.78	0.84	0.84	0.84	0.85

Ks = Kaersutite, Ti-prg = Titanian pargasite.

Table 9: Representative trace element compositions (ppm) of minerals in pyroxenites

sample mineral type	BG13							GV10		MC3		MC5	GV17	
	cpx porph	cpx porph	plag neobl	cpx neobl	amph neobl	amph neobl	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph
Sc	38.2	65.3	0.827	127	244	192	59.5	42.4	162	58.4	98.8	71.7	86.5	36.8
V	516	535	0.803	557	1017	751	506	430	522	333	322	419	275	280
Sr	4.75	6.95	196	5.06	66.5	20.8	3.84	5.68	7.51	3.20	3.65	7.64	3.39	3.79
Y	26.5	82.5	0.538	41.1	63.7	49.6	43.4	38.5	97.4	34.9	62.3	51.6	22.8	14.9
Zr	25.9	44.9	0.054	54.8	75.9	58.7	33.1	51.4	77.0	38.2	39.8	44.7	37.2	25.9
Nb	0.154	0.094		0.106	9.24	4.98	0.058	0.095	0.027	0.128	0.1705	0.347	0.178	0.390
La	0.585	0.620	0.456	0.544	0.962	0.677	0.601	0.996	0.805	0.807	0.795	0.843	0.360	0.446
Ce	3.68	3.99	0.917	3.48	5.50	4.37	3.54	5.82	4.85	4.25	4.31	4.91	2.55	3.09
Pr	0.737	0.862	0.078	0.796	1.31	1.11	0.822	1.12	1.06	0.874	0.888	0.886	0.577	0.637
Nd	4.56	5.52	0.467	5.85	9.30	8.09	4.99	7.25	6.94	6.31	6.03	5.76	4.58	4.56
Sm	2.38	3.27		2.59	4.52	4.15	2.64	3.44	3.51	2.71	2.53	2.89	2.11	1.81
Eu	1.03	1.40	0.349	0.922	1.59	1.20	1.20	1.54	1.59	1.19	1.01	1.28	0.900	0.720
Gd	3.48	6.43	0.074	4.03	6.94	5.60	4.28	5.12	6.91	4.11	4.39	5.18	3.22	2.10
Tb	0.714	1.61	0.012	0.849	1.47	1.08	1.04	0.975	1.71	0.816	1.09	1.04	0.567	0.390
Dy	4.50	12.7	0.121	6.41	10.12	7.60	6.93	6.96	13.8	5.54	8.99	7.93	3.88	2.64
Ho	1.04	3.10	0.029	1.45	2.40	1.75	1.63	1.50	3.79	1.29	2.22	1.91	0.890	0.550
Er	2.87	9.94		4.08	6.64	5.06	4.54	4.26	11.4	3.68	7.58	5.43	2.42	1.56
Tm	0.417	1.59		0.617	0.980	0.785	0.685	0.650	1.81	0.542	1.28	0.899	0.390	0.238
Yb	2.81	14.2		4.16	6.83	5.08	4.90	4.34	14.1	3.70	9.65	6.83	2.98	1.66
Lu	0.327	2.04		0.596	1.007	0.657	0.626	0.501	1.80	0.594	1.41	0.880	0.414	0.237
Hf	0.523	1.02		1.24	2.68	2.24	1.01	1.54	2.25	0.827	0.910	0.927	0.621	0.662

sample mineral type	BG4		BG5	BG14		BG22		GV8			GV12		GV14	
	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph	cpx porph
Sc	61.7	62.2	39.8	84.9	57.3	66.5	72.4	88	62	50.5	74	94	44	33
V	288	274	280	379	344	276	286	339	324	273	183	230	289	257
Sr	11.4	13.1	3.56	7.16	5.67	7.25	4.22	18.69	10.63	14.1	4.1	3.83	4.14	5.94
Y	24.6	24.9	24.3	36.1	21.1	27.0	43.5	28.53	22.53	20.0	8.44	23.31	11.71	8.35
Zr	29.4	27.6	30.9	47.3	30.6	33.6	49.8	54.6	44.9	33.2	7.90	18.2	43.3	28.3
Nb	0.495	0.638	0.500	0.286	0.435	0.595	0.628	0.105	0.270	0.275	0.061	0.287	0.186	0.251
La	0.692	0.738	0.782	0.579	0.587	0.761	0.788	1.34	1.62	1.56	0.124	0.274	0.698	0.896
Ce	3.58	3.74	3.99	3.86	3.85	3.68	4.45	7.74	6.98	7.28	0.577	1.31	4.62	4.47
Pr	0.649	0.656	0.775	0.821	0.780	0.722	0.874	1.375	1.233	1.14	0.150	0.273	0.878	0.852
Nd	4.15	4.08	5.55	5.70	5.08	4.75	6.25	7.12	6.91	6.55	1.28	2.11	6.20	5.44
Sm	1.85	1.86	2.02	2.76	2.20	2.18	3.13	3.10	2.51	2.31	0.647	1.32	2.15	2.17
Eu	0.803	0.813	0.863	1.066	0.949	0.947	1.23	1.20	1.06	1.07	0.243	0.371	1.01	0.742
Gd	2.86	2.85	3.06	3.87	2.56	3.11	4.70	4.55	3.11	2.82	1.20	2.68	3.13	1.86
Tb	0.550	0.570	0.591	0.848	0.529	0.638	1.05	0.772	0.672	0.607	0.230	0.590	0.489	0.321
Dy	3.96	3.90	3.97	6.17	3.82	4.29	7.12	5.28	4.20	3.73	1.66	4.06	2.90	1.57
Ho	0.890	0.890	0.965	1.36	0.823	0.968	1.65	1.11	0.848	0.755	0.353	0.899	0.463	0.378
Er	2.57	2.68	2.64	4.22	2.24	3.04	4.99	3.37	2.69	2.22	0.810	2.56	1.10	0.717
Tm	0.409	0.420	0.422	0.631	0.327	0.455	0.843	0.463	0.366	0.312	0.116	0.420	0.146	0.072
Yb	2.89	3.01	3.22	4.52	2.36	3.31	5.72	4.02	2.65	2.16	0.782	2.99	1.23	0.685
Lu	0.441	0.465	0.475	0.663	0.393	0.528	0.990	0.403	0.360	0.308	0.145	0.466	0.120	0.106
Hf	0.685	0.617	0.439	0.799	0.867	0.691	0.730	1.65	1.18	1.05	0.23	0.266	1.21	0.988

Abbreviations: cpx = clinopyroxene, plag = plagioclase, amph = amphibole.

Table 10: Temperature estimates for pyroxene porphyroblasts and neoblasts within pyroxenites

sample		T(°C)Ca-in-opx	T(°C)Ca-in-opx	T(°C)2-pyrox	T(°C)2-pyrox	sample	T(°C)epx	
		B&K(1990)	Corrected*	B&K(1990)	Taylor(1998)		porph	neobl
BG3	<i>porph</i>	986 ± 21	971	978 ± 33	939 ± 31	BG3	927 ± 28	840 ± 32
	<i>neobl</i>	856 ± 27	810	895 ± 36	826 ± 35			
BG5	<i>porph</i>	1018 ± 19	1008	856 ± 38	839 ± 36	BG5	828 ± 37	856 ± 41
	<i>neobl</i>	868 ± 23	826	869 ± 42	833 ± 40			
BG13	<i>porph</i>	1018 ± 32	1008	863 ± 29	841 ± 29	BG13	819 ± 27	775 ± 38
	<i>neobl</i>	891 ± 22	855	817 ± 44	768 ± 42			
BG14	<i>porph</i>	986 ± 17	971	881 ± 36	830 ± 33	BG14	830 ± 32	848 ± 36
	<i>neobl</i>	880 ± 14	841	894 ± 41	828 ± 38		BG22	936 ± 31
GV8	<i>porph</i>	1107 ± 28	1107	965 ± 21	942 ± 20	GV8	924 ± 21	803 ± 31
	<i>neobl</i>	887 ± 23	850	846 ± 38	797 ± 36			
GV10	<i>porph</i>	1018 ± 21	1008	933 ± 35	913 ± 33	GV10	934 ± 34	882 ± 31
	<i>neobl</i>	830 ± 16	776	900 ± 31	862 ± 30		GV12	953 ± 33
MC3	<i>porph</i>	986 ± 12	971	913 ± 25	873 ± 24	MC3	907 ± 21	859 ± 17
	<i>neobl</i>	880 ± 25	841	878 ± 33	794 ± 32		GV14	1136 ± 19
MC5	<i>porph</i>	932 ± 27	906	867 ± 24	845 ± 25	MC5	905 ± 23	814 ± 32
	<i>neobl</i>	891 ± 29	855	871 ± 26	791 ± 25			

(*) correction of average T according to Nimis & Grutter (2010)

Table 11: Sm-Nd isotope data determined on bulk-rock, clinopyroxene and plagioclase from pyroxenites

sample	bulk rocks				clinopyroxenes				plagioclases			
	Sm (ppm)	Nd (ppm)	$^{147}\text{Sm}/^{144}\text{Nd}^\wedge$	$^{143}\text{Nd}/^{144}\text{Nd}$ ($\pm 2\sigma \times 10^6$) [†]	Sm (ppm)	Nd (ppm)	$^{147}\text{Sm}/^{144}\text{Nd}^\wedge$	$^{143}\text{Nd}/^{144}\text{Nd}$ ($\pm 2\sigma \times 10^6$) [†]	Sm (ppm)	Nd (ppm)	$^{147}\text{Sm}/^{144}\text{Nd}^\wedge$	$^{143}\text{Nd}/^{144}\text{Nd}$ ($\pm 2\sigma \times 10^6$) [†]
BG13	0.946	1.586	0.3608	0.513383 \pm 06	3.054	4.888	0.3778	0.513401 \pm 07	0.086	0.195	0.2663	0.513272 \pm 06
BG14	0.765	1.867	0.2478	0.513123 \pm 06	1.992	4.426	0.2721	0.513155 \pm 07	0.138	0.488	0.1714	0.513037 \pm 06
GV8	0.757	2.029	0.2255	0.512984 \pm 06	2.596	6.657	0.2358	0.512995 \pm 09	0.250	0.812	0.1860	0.512939 \pm 05
BG5					1.585	3.766	0.2545	0.513131 \pm 09	0.038	0.202	0.1106	0.512959 \pm 09

Note: All data by isotope dilution.

[†] Internal precision.

[^] Estimated error is 0.3%.

Table 12: Phase abundances of model spinel- and garnet-facies assemblages by mass balance

sample	BG3	BG4	BG5	BG13	BG14	BG22	GV8	GV10	GV12	GV14	GV17	MC3	MC5
<i>spinel-facies assemblage</i> (*):													
CPX	77	80	74	64	75	80	59	63	87	85	74	47	31
OPX	12	0	14	24	13	10	32	30	0	0	0	42	61
OL	0	11	0	0	0	0	0	0	0	0	9	0	0
SP	11	9	12	12	12	10	9	7	13	15	17	11	8
R2	0.15	0.21	0.27	0.18	0.23	0.22	0.21	0.16	0.24	0.33	0.41	0.16	0.18
<i>garnet-facies assemblage</i> (§):													
CPX	---	77	81	55	78	---	62	61	69	---	---	45	30
GNT	---	17	12	40	11	---	23	18	31	---	---	40	29
OPX	---	0	0	5	0	---	15	21	---	---	---	5	31
OL	---	0	0	0	0	---	0	0	---	---	---	10	10
SP	---	6	7	0	11	---	0	0	---	---	---	0	0
R2	---	0.57	0.61	0.35	0.45	---	0.45	0.55	0.56	---	---	0.35	0.48

(*) modal abundances obtained by mass balance calculation using bulk rock analyses and average mineral compositions measured on spinel-facies porphyroclasts in each sample.

(§) modal abundances obtained by mass balance calculation using bulk rock analyses and mineral compositions from garnet-bearing assemblages documented in selected orogenic garnet pyroxenites (see text and supplementary files for details).