



Supporting Information

Dynamic Gold(I) Complexes of Hexa-*tert*-butyl-octaphosphane

Toni Grell and Evamarie Hey-Hawkins*

[ejic201901300-sup-0001-SupMat.pdf](#)

1. Table of Contents

1. Table of Contents.....	1
2. Experimental Section	2
General Remarks.....	2
Synthesis of 2,2',3,3',4,4'-Hexa- <i>tert</i> -butyl-1,1'-bicyclotetraphosphane { <i>cyclo</i> -(P ₄ tBu ₃) ₂ } (1).....	2
Synthesis of [AuCl(1)] (2).....	3
Synthesis of [(AuCl) ₂ (1)] (3)	4
Synthesis of [(AuCl) ₃ (1)] (4)	5
Reactions of { <i>cyclo</i> -(P ₄ tBu ₃) ₂ } (1) with [(AuCl) ₃ (1)] (4)	6
Reaction of { <i>cyclo</i> -(P ₄ tBu ₃) ₂ } (1) with [(AuCl) ₃ (1)] (3)	6
3. NMR Measurements and Spectra of the Compounds.....	7
NMR Spectra of 2,2',3,3',4,4'-Hexa- <i>tert</i> -butyl-1,1'-bicyclotetraphosphane { <i>cyclo</i> -(P ₄ tBu ₃) ₂ } (1).....	7
NMR Spectra of [AuCl(1)] (2).....	10
NMR Spectra of [(AuCl) ₂ (1)] (3)	16
NMR Spectra of [(AuCl) ₃ (1)] (4)	17
NMR Spectra of Reaction Mixtures	20
4. Mass spectrometry measurements.....	20
5. X-Ray Diffraction.....	21
Single Crystal Measurements	21
Powder Measurements	24
CSD Search	25
6. Quantum Chemical Calculations	26
General Remarks.....	26
Benchmark Study	26
Rotational Profile of 2,2',3,3',4,4'-Hexa- <i>tert</i> -butyl-1,1'-bicyclotetraphosphane { <i>cyclo</i> -(P ₄ tBu ₃) ₂ } (1)	27
Rotational Profile of [AuCl(1)] (2)	28
Rotational Profile of [(AuCl) ₂ (1)] (3).....	29
Molecular Orbital Energies.....	30
Transition States.....	31
Complex Formation Energies.....	32
Cartesian Coordinates from Geometry Optimizations of Selected Molecules	33
7. References.....	38

2. Experimental Section

General Remarks

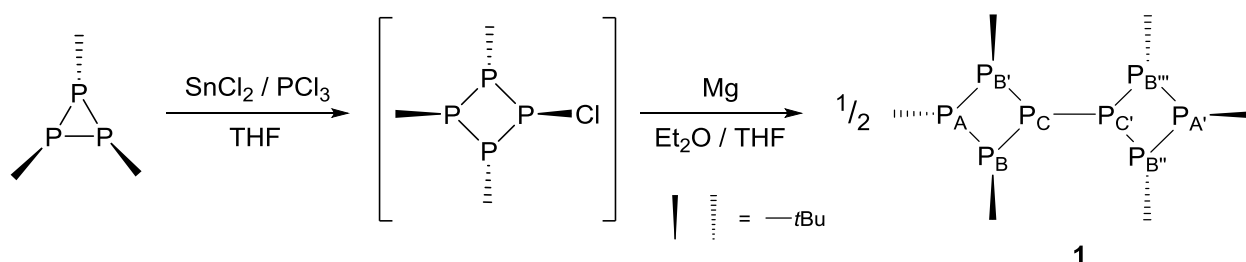
All experiments were carried out under nitrogen using standard Schlenk techniques. Toluene, dichloromethane, diethyl ether, pentanes as well as hexanes (isomeric mixtures) and acetonitrile were dried and degassed with an MB SPS-800 Solvent Purification System (MBRAUN) and kept over molecular sieves (4 Å). THF was distilled from potassium and benzophenone and kept over molecular sieves (4 Å). The compounds *cyclo*-(P₃tBu₃)^[1] and [AuCl(tht)]^[2] were prepared as described in the literature. Magnesium turnings were activated by heating to about 500 °C under vacuum. Cooling baths for temperatures of –83 °C were prepared from ethyl acetate and liquid nitrogen. All other chemicals were purchased and used without further purification.

Elemental Analysis. A VARIO EL (HERAEUS) microanalyzer was used to determine the elemental composition of the compounds.

Infra-red Spectroscopy. Samples were prepared as KBr pellets in a nitrogen-filled glove box and the infrared spectra recorded in a range of 400 – 4000 cm⁻¹. A PERKIN-ELMER (System 2000) FT-IR spectrometer was used for compound **1** and a BRUKER (SENSOR 27) FT-IR spectrometer for compounds **2** to **4**. Abbreviation for labeling the vibrations: w – weak; m – medium; s – strong; δ_s – symmetric deformation vibration; δ_{as} – asymmetric deformation vibration; ν_s – symmetric stretching vibration; ν_{as} – asymmetric stretching vibration; ρ – Rocking vibration.

Luminescence. Luminescence was tested at 254 and 366 nm using a common UV lamp.

Synthesis of 2,2',3,3',4,4'-Hexa-*tert*-butyl-1,1'-bicyclotetraphosphane {*cyclo*-(P₄tBu₃)₂} (**1**)



PCl₃ (7.2 g, 52.7 mmol, 1.0 eq) was added to a freshly prepared solution of SnCl₂ (10.0 g, 52.7 mmol, 1.0 eq) in 140 mL in THF. This solution was immediately added dropwise to a solution of *cyclo*-(P₃tBu₃) (13.9 g, 52.7 mmol, 1.0 eq) in 220 mL THF at –83 °C over the course of about 15 min. Subsequently, the reaction mixture was stirred for one hour at room temperature and kept at 5 °C overnight, resulting in the formation of crystals of SnCl₄(THF)₂. Evaporation of the solvent under reduced pressure gave a yellowish solid which was extracted with 500 mL pentanes twice. Removal of the solvent of the combined phases gave a viscous yellowish oil which was dried under vacuum and then dissolved in a mixture of 450 mL Et₂O and 50 mL THF. This solution was added to magnesium turnings (1.5 g, 62.1 mmol, 1.1 eq) and stirred overnight. The solvents were removed under reduced pressure and the residue extracted with 500 mL hexanes three times. Removal of the combined phases under reduced pressure gave a raw product which was separated by fractionated crystallization as follows:

1. Suspending in 150 mL boiling THF and decantation after 5 min gave a small crystal fraction of pure phosphane **1** after one day which was separated by filtration and washed with 10 mL Et₂O.
2. Keeping the filtrate at –30 °C gave pure octaphosphane **1** as a colorless crystalline solid after six days which was separated by filtration and washed with 20 mL Et₂O.

Yield: 4.3 g (28%, colorless crystalline solid).

MP: 253 °C (Decomp.)

¹H{³¹P} NMR (C₆D₆): δ = 1.33 (s, 18H, (H₃C)₃C-P_A), 1.30 (s, 36H, (H₃C)₃C-P_B) ppm.

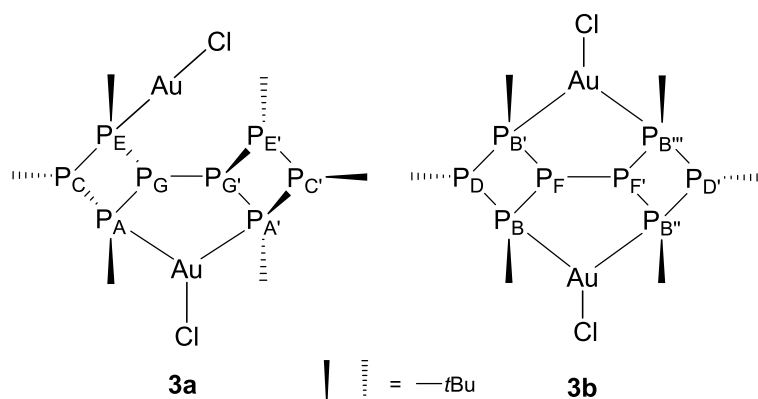
¹³C{¹H, ³¹P} NMR (C₆D₆): δ = 30.6 (s, C-P_A), 30.5 (s, C-P_B), 29.0 (s, (H₃C)₃C-CP_A), 28.8 (s, (H₃C)₃C-CP_B) ppm.

³¹P{¹H} NMR (C₆D₆): δ = –38.3 (m, 2P, P_A), –46.2 (m, 4P, P_B), –103.0 (m, 2P, P_C) ppm.

IR (KBr): $\tilde{\nu}$ = 3182 (w), 2931 (s, ν_{as}(C–H)), 2852 (s, ν_s(C–H)), 2705 (w), 1452 (s, δ_{as}(CH₃)), 1386 (m, δ_s(CH₃)), 1358 (s, δ_s(CH₃)), 1172 (s, ρ(CH₃) + ν_s(C–C)), 1010 (m, ρ(CH₃)), 935 (m, ρ(CH₃) + ν_s(C–C)), 805 (m, ν_s(C–C)), 728 (w), 574 (m), 457 (w) cm⁻¹.

MS (ESI(+), MeCN/THF 1:1): *m/z* = 239.0 [P₄tBu₂H]⁺ (5%), 255.0 [P₄tBu₂OH]⁺ (4%), 271.0 [P₄tBu₂O₂H]⁺ (1%), 287.0 [P₄tBu₂O₃H]⁺ (2%), 303.0 [P₄tBu₂O₄H]⁺ (1%), 327.1 [P₅tBu₃H]⁺ (4%), 343.1 [P₅tBu₃OH]⁺ (4%), 359.1 [P₅tBu₃O₂H]⁺ (3%), 375.1 [P₅tBu₃O₃H]⁺ (4%),

Synthesis of $[(\text{AuCl})_2(\mathbf{1})]$ (**3**)



A freshly prepared solution of $[\text{AuCl}(\text{tth})]$ (80 mg, 250 μmol , 2.0 eq) in 35 mL THF was added dropwise to a solution of phosphane **1** (74 mg, 125 μmol , 1.0 eq) in 20 mL THF over the course of 1.5 hours using a syringe pump, resulting in a pale yellow solution. About 50 mL of the solvent were removed under reduced pressure and a yellow solid precipitated. The solid was filtered off, washed with 2 mL of hexanes and dried in vacuum for one hour. The filtrate was reduced to 1 mL under reduced pressure. Gas diffusion of hexanes over night at 0 °C gave a yellow crystalline solid which was filtered off, washed with 2 mL of hexanes and dried in vacuum for one hour. Both solids are analytically identical and were combined. The solid shows no luminescent behavior and is slightly soluble in THF and toluene. Single crystals were obtained from passing a stream of dry nitrogen over a saturated solution of the complex in THF.

Yield: 111 mg (84%)

MP: 231 °C (Decomp.)

$^1\text{H}\{^{31}\text{P}\}$ NMR (THF- d_6): δ = 1.40 – 1.15 (various s, CH_3) ppm.

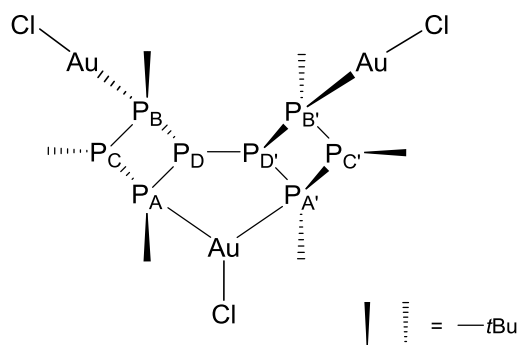
$^{31}\text{P}\{^1\text{H}\}$ NMR (THF- d_6 , 25 °C): δ = 20.5 (m, 2P, P_A), -3.0 (m, P_B), -24.0 (m, 2P, P_C), -29.7 (m, P_D), -48.5 (m, 2P, P_E), -63.3 (m, P_F), -73.3 (m, 2P, P_G) ppm. (The signals P_A , P_C , P_E and P_G correspond to **3b**. The remaining signals possess significantly lower intensity and are assigned to **3a**.)

IR (KBr): $\tilde{\nu}$ = 3445 (s), 2952 (s, $\nu_{\text{as}}(\text{C-H})$), 2892 (s, $\nu_{\text{s}}(\text{C-H})$), 2857 (s, $\nu_{\text{s}}(\text{C-H})$), 1723 (w), 1635 (m), 1459 (s, $\delta_{\text{as}}(\text{CH}_3)$), 1387 (m, $\delta_{\text{s}}(\text{CH}_3)$), 1362 (s, $\delta_{\text{s}}(\text{CH}_3)$), 1262 (m), 1171 (s, $\rho(\text{CH}_3) + \nu_{\text{s}}(\text{C-C})$), 1011 (m, $\rho(\text{CH}_3)$), 988 (m), 951 (w, $\rho(\text{CH}_3) + \nu_{\text{s}}(\text{C-C})$), 803 (m, $\nu_{\text{s}}(\text{C-C})$), 713 (w), 593 (w), 542 (w), 418 (w) cm^{-1} .

MS (ESI(+), MeCN): m/z = 673.0 $[\text{P}_8^t\text{Bu}_4\text{Au}]^+$ (5%), 787.2 $[\text{LAu}]^+$ (33%), 803.2 $[\text{LAu}]^+$ (8%), 1019.1 $[\text{LAu}_2\text{Cl}]^+$ (100%), 1035.1 $[\text{LOAu}_2\text{Cl}]^+$ (2%), 1077.1 $[\text{LAu}_2\text{Cl}_2\text{Na}]^+$ (13%), 1093.1 $[\text{LOAu}_2\text{Cl}_2\text{Na}]^+$ (1%), 1251.0 $[\text{LAu}_3\text{Cl}_2]^+$ (1%). L = $\text{C}_{24}\text{H}_{54}\text{P}_8$.

Elemental Analysis for $\text{C}_{24}\text{H}_{54}\text{Au}_2\text{Cl}_2\text{P}_8$	Found (%)	C: 22.85	H: 4.07
	Calculated (%)	C: 22.39	H: 4.23

Synthesis of [(AuCl)₃(1)] (4)



A freshly prepared solution of [AuCl(tht)] (280 mg, 873 mmol, 3.5 eq) in 30 mL THF was added dropwise to a solution of phosphane **1** (146 mg, 247 mmol, 1.0 eq) in 20 mL THF over the course of 10 minutes and stirred for one hour. About 55 mL of the solvent were removed under reduced pressure, resulting in the precipitation of a yellow solid. The solid was filtered off, washed with 3 mL of hexanes and dried in vacuum for one hour. The filtrate was reduced to 1 mL under reduced pressure. Gas diffusion of hexanes over night at 0 °C gave a yellow solid which was filtered off, washed with 3 mL of hexanes and dried in vacuum for one hour. Both solids are analytically identical and were combined. The solid shows no luminescent behavior and is very poorly soluble in THF and toluene. Single crystals were obtained from a saturated solution of the complex in THF at 0 °C.

Yield: 268 mg (84%)

MP: 183 °C (Decomp.)

¹H{³¹P} NMR (THF-*d*₆): δ = 1.57 (s, 18H, (H₃C)₃C-P_A), 1.49 (s, 18H, (H₃C)₃C-P_C), 1.44 (s, 18H, (H₃C)₃C-P_B) ppm.

¹³C{¹H, ³¹P} NMR (THF-*d*₆): δ = 28.8 (s, (H₃C)₃C-CP_A), 28.5 (s, (H₃C)₃C-CP_C), 26.1 (s, (H₃C)₃C-CP_B) ppm. Quaternary carbon atom could not be detected.

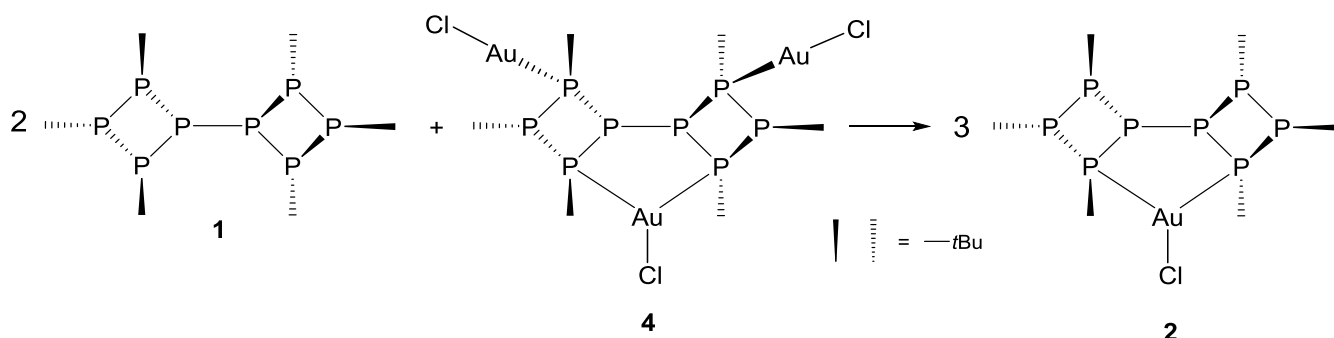
³¹P{¹H} NMR (THF-*d*₆): δ = 12.4 (m, 2P, P_A), 9.6 (m, 2P, P_B) –20.7 (m, 2P, P_C), –45.7 (m, 2P, P_D) ppm.

IR (KBr): $\tilde{\nu}$ = 3442 (w), 2953 (s, $\nu_{\text{as}}(\text{C-H})$), 2891 (s, $\nu_{\text{s}}(\text{C-H})$), 2854 (m, $\nu_{\text{s}}(\text{C-H})$), 1723 (m), 1636 (m), 1544 (w), 1458 (s, $\delta_{\text{as}}(\text{CH}_3)$), 1388 (m, $\delta_{\text{s}}(\text{CH}_3)$), 1361 (s, $\delta_{\text{s}}(\text{CH}_3)$), 1262 (w), 1204 (w), 1165 (s, $\rho(\text{CH}_3) + \nu_{\text{s}}(\text{C-C})$), 1102 (w), 1013 (m, $\rho(\text{CH}_3)$), 937 (w, $\rho(\text{CH}_3) + \nu_{\text{s}}(\text{C-C})$), 869 (w), 802 (m, $\nu_{\text{s}}(\text{C-C})$), 668 (w), 594 (w), 541 (w), 496 (w), 418 (w) cm⁻¹.

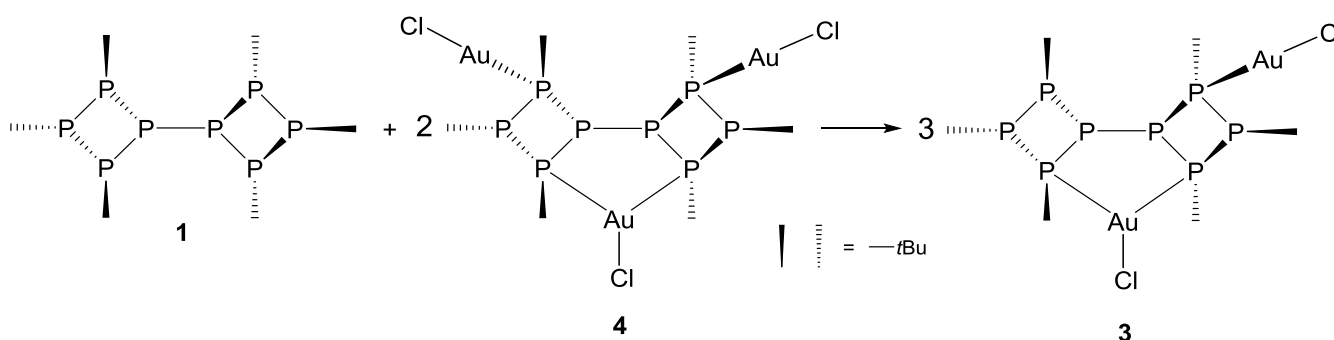
MS (ESI(+), MeCN): *m/z* = 1019.1 [LAu₂Cl]⁺ (4%), 1251.0 [LAu₃Cl₂]⁺ (100%), 1311.0 [LAu₃Cl₃Na]⁺ (5%). L = C₂₄H₅₄P₈.

Elemental Analysis for C ₂₄ H ₅₄ Au ₃ Cl ₃ P ₈	Found (%)	C: 27.42	H: 5.08
	Calculated (%)	C: 27.32	H: 5.16

Reactions of {cyclo-(P*t*Bu₃)₂} (1) with [(AuCl)₃(1)] (4)

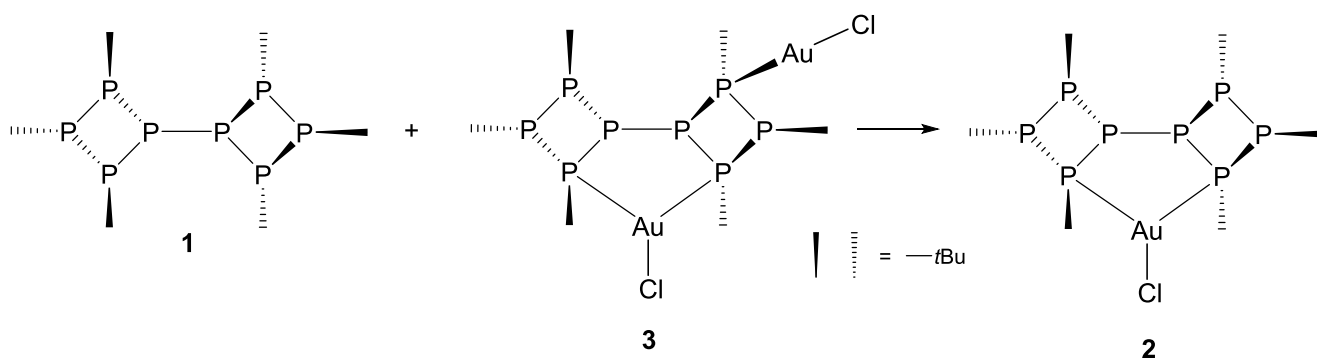


A solution of phosphane 1 (48 mg, 81 mmol, 2.0 eq) in 10 mL THF was added to a suspension of 4 (52 mg, 40 mmol, 1.0 eq) in 20 mL THF. After 5 minutes the suspension became clear and turned pale yellow. About 28 mL of the solvent were removed under reduced pressure and the reaction mixture investigated with ³¹P{¹H} NMR spectroscopy. Removal of the solvent and drying in vacuum for 30 minutes gave a substance which was analytically identical to 2.



A solution of phosphane 1 (15 mg, 25 mmol, 1.0 eq) in 10 mL THF was added to a suspension of 4 (64 mg, 50 mmol, 2.0 eq) in 30 mL THF. After 5 minutes the suspension became clear and turned pale yellow. About 38 mL of the solvent were removed under reduced pressure and the reaction mixture investigated with ³¹P{¹H} NMR spectroscopy. Removal of the solvent and drying in vacuum for 30 minutes gave a substance which was analytically identical to 3.

Reaction of {cyclo-(P*t*Bu₃)₂} (1) with [(AuCl)₃(1)] (3)



15 mL of THF were added to a mixture of phosphane 1 (12 mg, 20 mmol, 1.0 eq) and complex 3 (21 mg, 20 mmol, 1.0 eq). The initial suspension turned clear and bright yellow after 5 minutes. About 13 mL of the solvent were removed under reduced pressure and the reaction mixture investigated with ³¹P{¹H} NMR spectroscopy. Removal of the solvent and drying in vacuum for 30 minutes gave a substance which was analytically identical to 2.

3. NMR Measurements and Spectra of the Compounds

NMR spectra were recorded at 25 °C with a BRUKER AVANCE DRX 400 spectrometer (^1H NMR: 400.13 MHz, ^{13}C NMR: 100.16 MHz, ^{31}P NMR: 161.97 MHz). TMS was used as internal standard for ^1H NMR spectroscopy. ^{13}C and ^{31}P NMR experiments were referenced to the Ξ scale.^[3] Every signal labeling in this publication also extends to the chemically equivalent yet magnetically nonequivalent nuclei *i.e.* P_A for P_A , P_A' as well as P_A'' and so forth. The ^{31}P NMR signals are labeled alphabetically starting from the nucleus, which is most deshielded.

All ^{31}P NMR spectra were measured with a 90° pulse and reduced acquisition (0.4 s) as well as d_1 time (1.0 s). The number of scans necessary to obtain acceptable spectra strongly depended on the solubility of the respective compound. For complex **2**, usually 100 scans sufficed while at least 1000 scans were needed for the other compounds. One inverted-gated coupling ^{31}P NMR experiment for octaphosphane **1** was performed and showed approximately the same integration values as for the standard NMR experiment. Therefore, the integration values for all signals were retrieved from the standard ^{31}P NMR experiments.

NMR Spectra of 2,2',3,3',4,4'-Hexa-*tert*-butyl-1,1'-bicyclotetraphosphane {*cyclo*-(P_4tBu_3)₂ (**1**)

The spectroscopic properties of **1** differ slightly from the literature. Baudler *et al.* reported two multiplets (−45 and −100 ppm) with a relative intensity of 3:1 for the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the phosphane.^[4] In contrast, the spectrum obtained in this work (Figure S2) shows three multiplets at −38.2, −46.0 and −103.0 ppm with an I_{rel} of 1:2:1 and very strong higher order effects. In the proton-coupled ^{31}P NMR spectrum, the signals P_A and P_B (Figure S3) are significantly broadened because of the $^3J_{\text{PH}}$ coupling of the attached *t*Bu groups (12.4 and 13.0 Hz as determined from the two pseudo-doublets observed in the ^1H NMR spectrum). Due to the lack of this coupling, the signal P_C at −103.0 ppm is thus identified as the γ -P atom which is also in accordance with the chemical shift (strong shielding). On the basis of a $^{31}\text{P}\{^1\text{H}\}$ - $^{31}\text{P}\{^1\text{H}\}$ COSY NMR spectrum, signal P_B can be assigned as the phosphorus atoms in β position which also matches its relative intensity. Additionally, the results from the $^1\text{H}\{^{31}\text{P}\}$ NMR (two singlets with $I_{\text{rel}} = 2:1$) and $^{13}\text{C}\{^{31}\text{P}\}$ NMR spectra (two sets of singlets) agree with these assignments. These findings are therefore consistent with the expected AA'B₂B'₂CC' spin system.

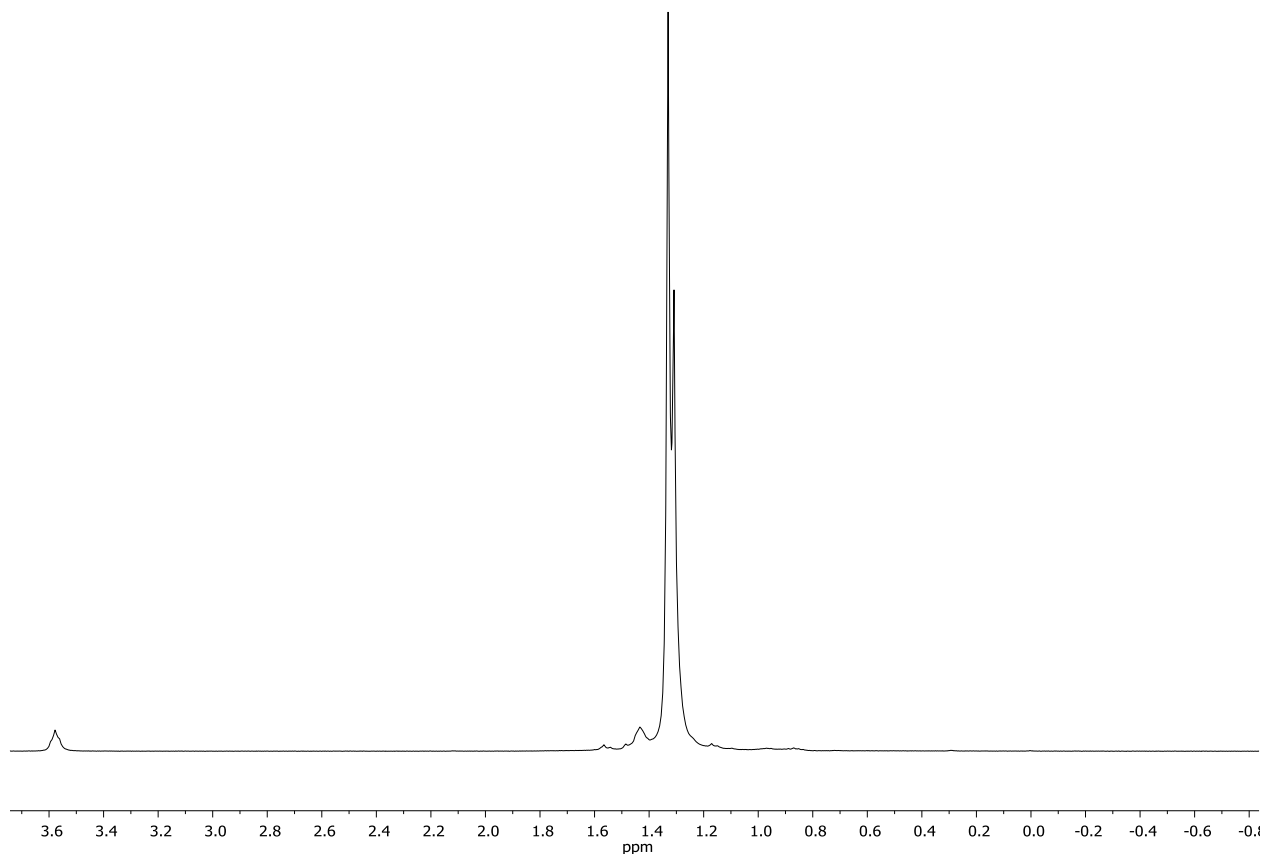


Figure S1. $^1\text{H}\{^{31}\text{P}\}$ NMR spectrum of phosphane **1** in $\text{THF-}d_6$.

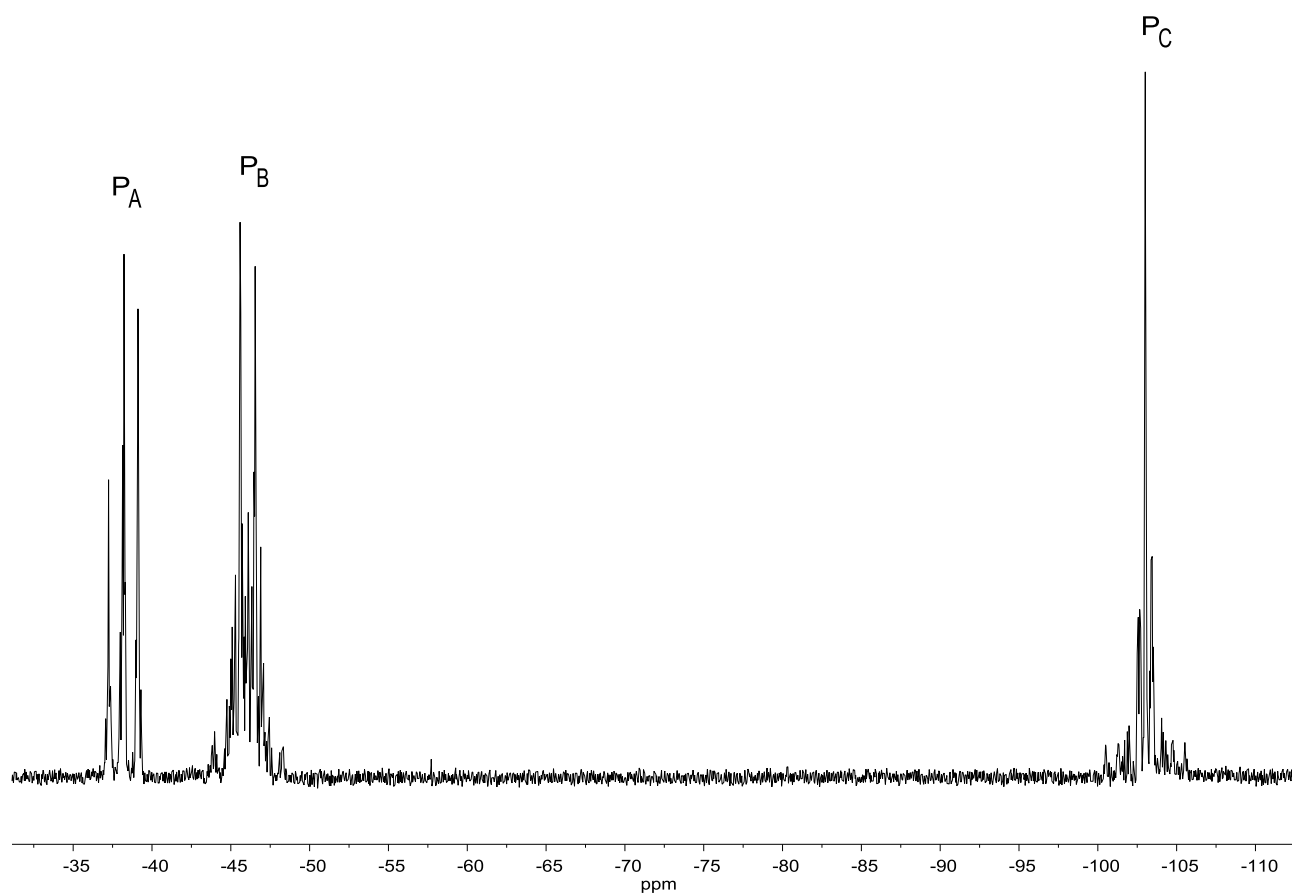


Figure S2. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of phosphane 1 in C_6D_6 .

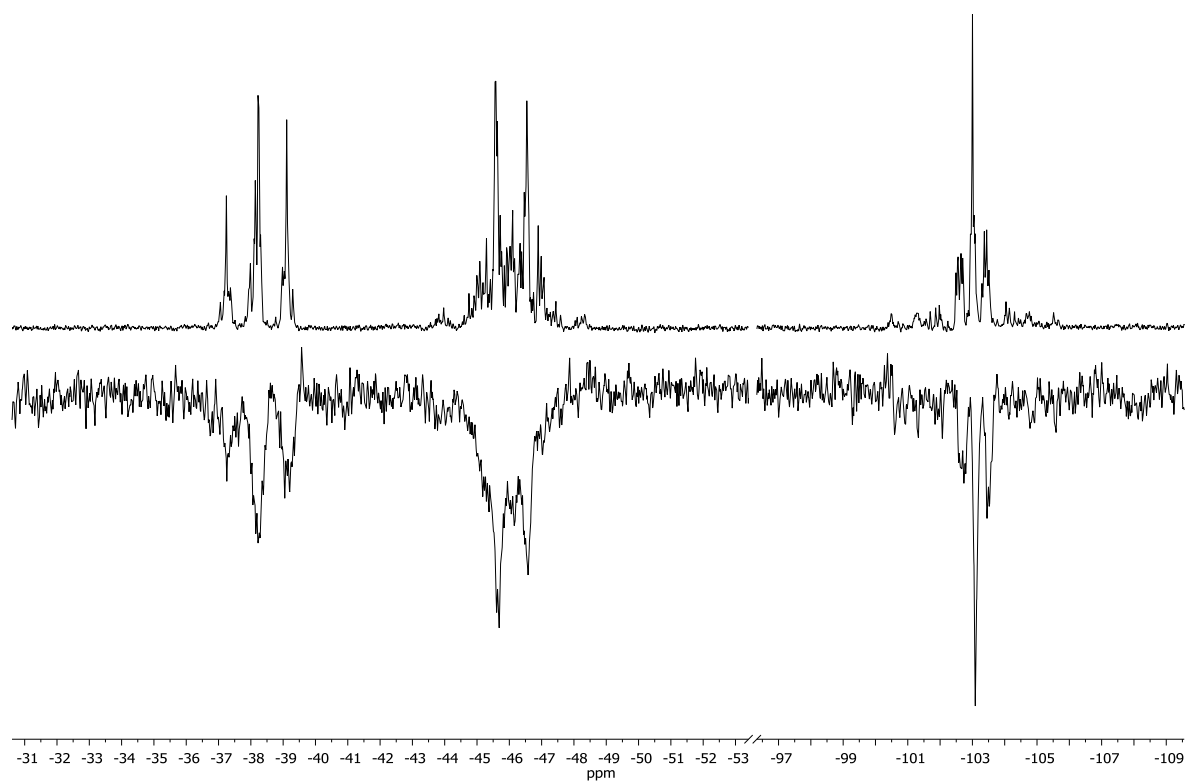


Figure S3. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum (top) and proton coupled ^{31}P NMR spectrum (bottom) of phosphane 1 in C_6D_6 .

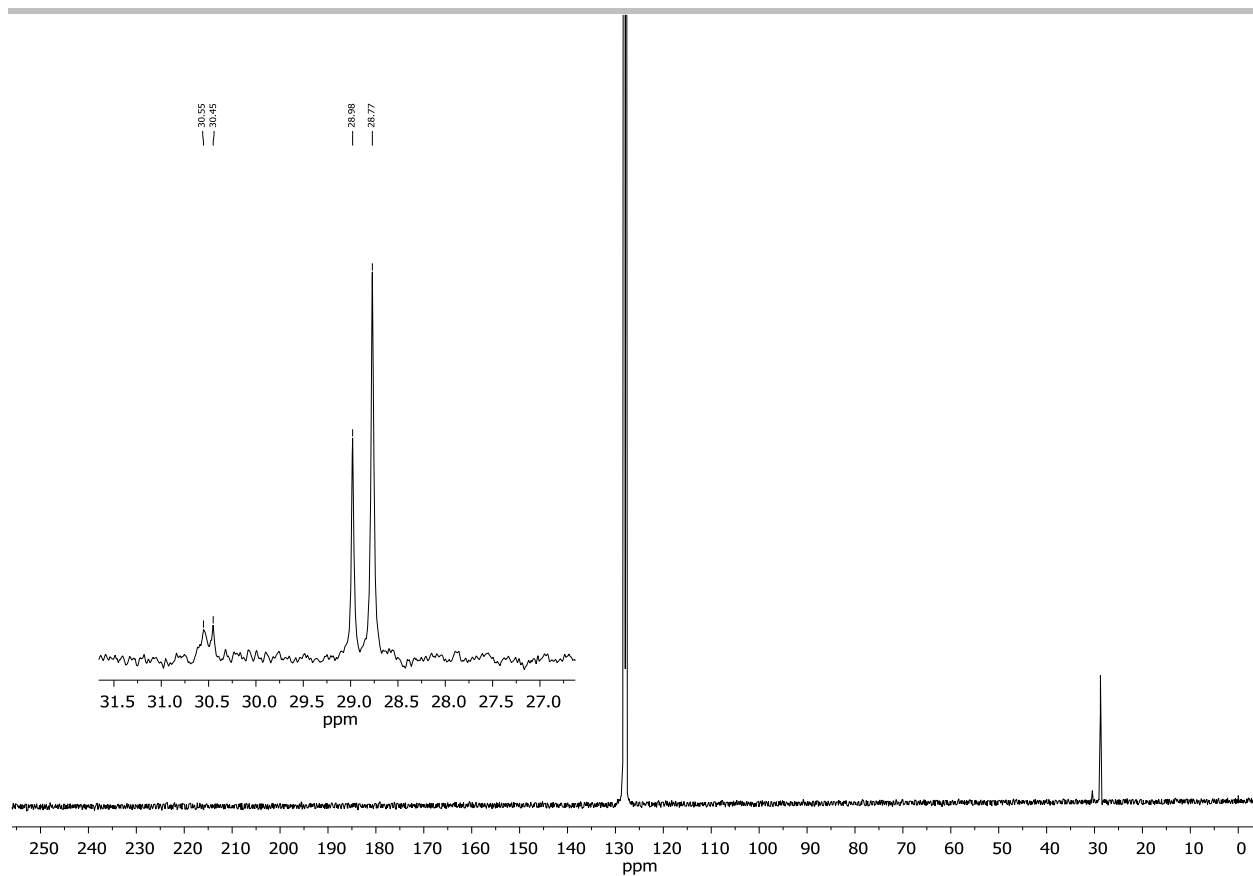


Figure S4. $^{13}\text{C}\{^1\text{H},^{31}\text{P}\}$ NMR spectrum of phosphane 1 in C_6D_6 .

NMR Spectra of [AuCl(1)] (2)

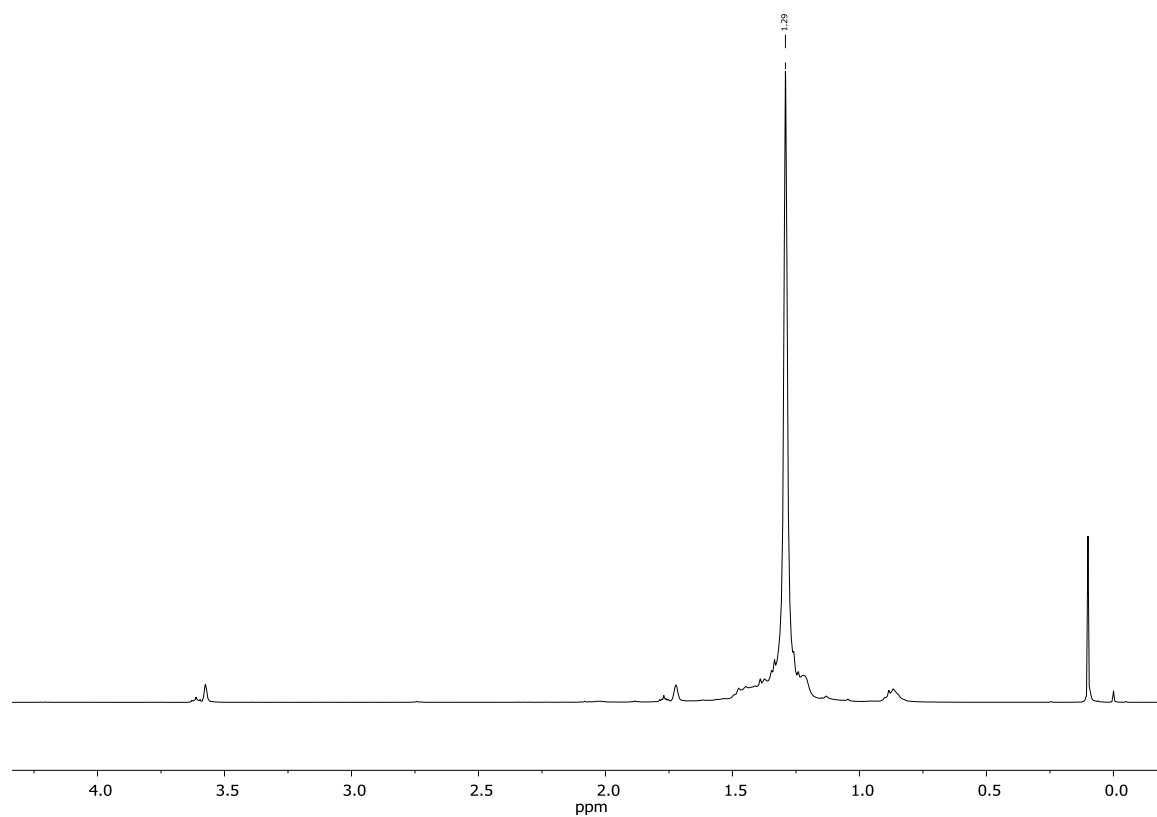


Figure S5. $^1\text{H}\{^{31}\text{P}\}$ NMR spectrum of complex **2** in $\text{THF-}d_6$ at $26\text{ }^\circ\text{C}$.

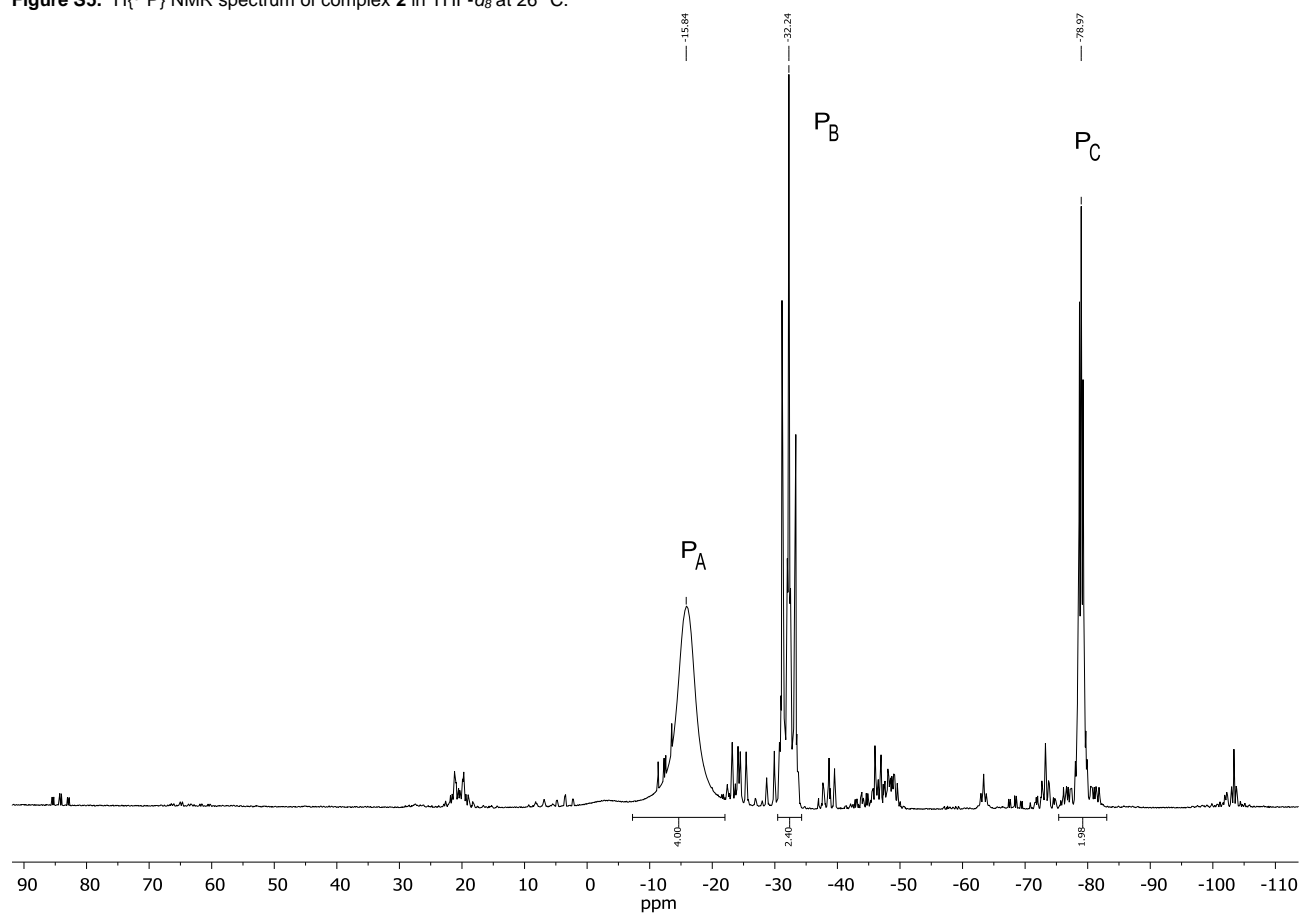


Figure S6. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **2** in $\text{THF-}d_6/\text{C}_6\text{D}_6$ at $26\text{ }^\circ\text{C}$.

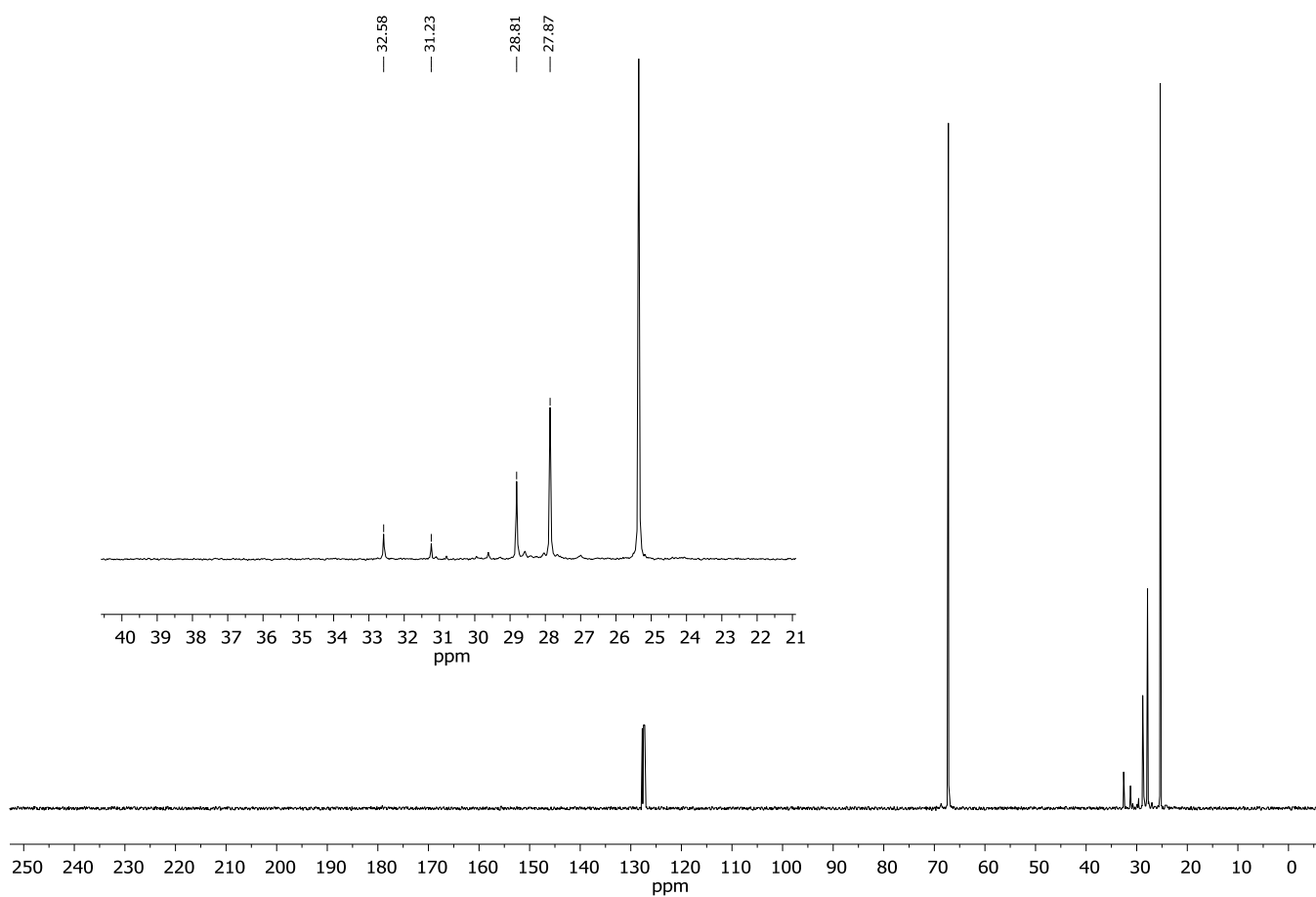


Figure S7. $^{13}\text{C}\{^1\text{H},^{31}\text{P}\}$ NMR spectrum of complex **2** in $\text{THF-}d_8/\text{C}_6\text{D}_6$ at 26 °C.

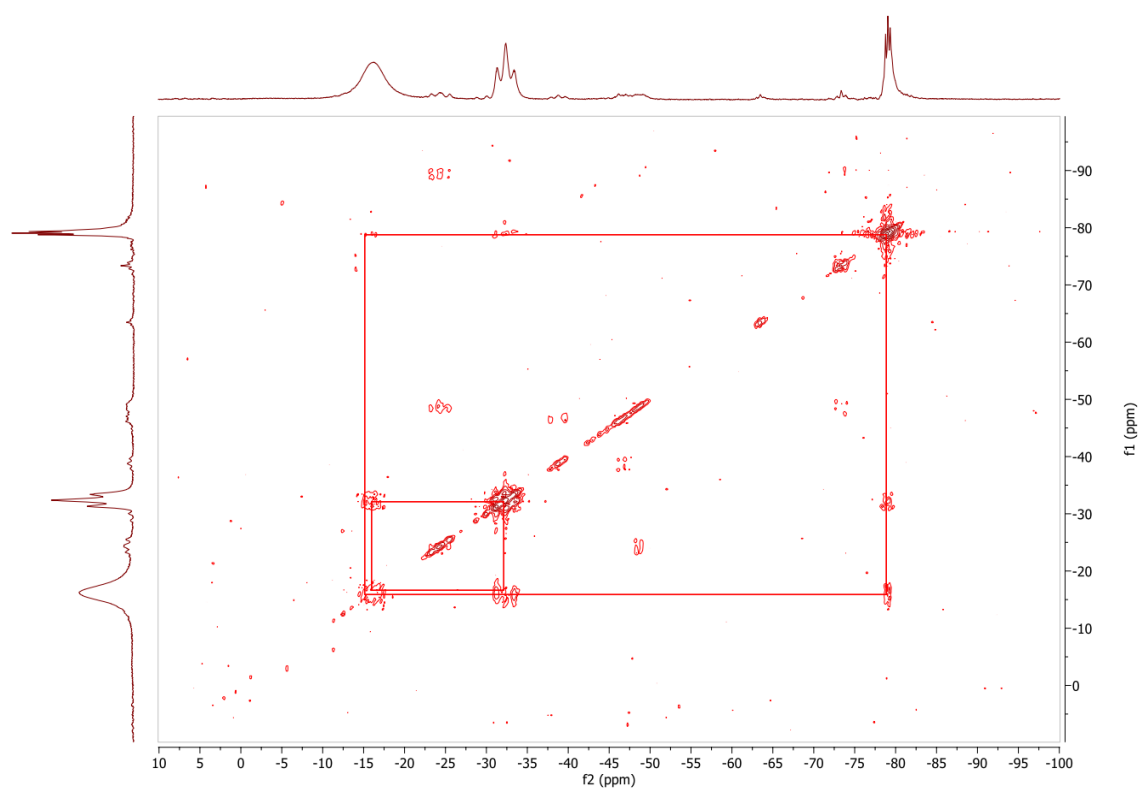


Figure S8. $^{31}\text{P}\{^1\text{H}\}$ - $^{31}\text{P}\{^1\text{H}\}$ COSY NMR spectrum of complex **2** in $\text{THF-}d_8/\text{C}_6\text{D}_6$ at 26 °C.

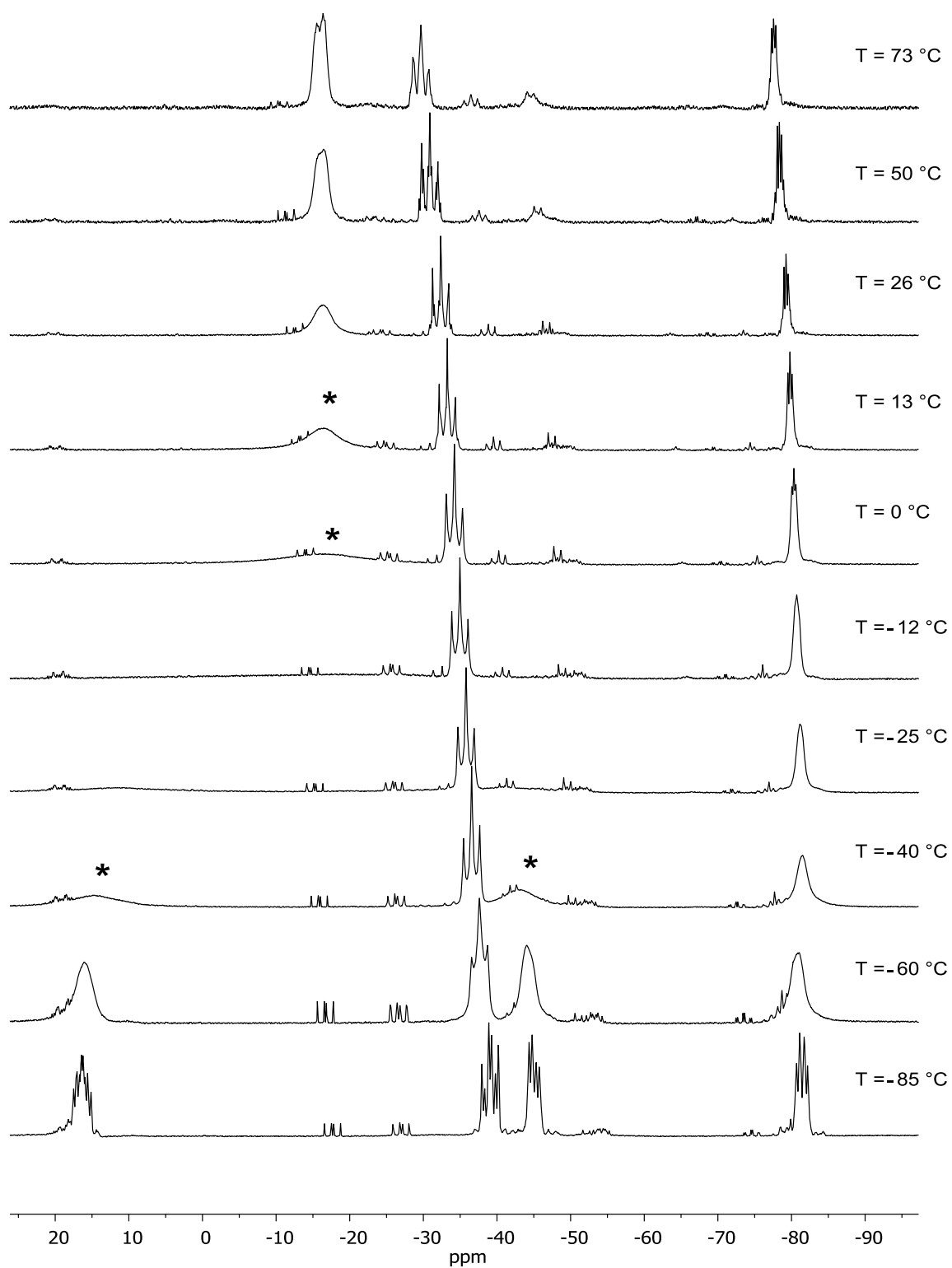


Figure S9. VT $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **2** in $\text{THF-}d_8/\text{C}_6\text{D}_6$. The asterisks mark the decoalescing and emerging signals.

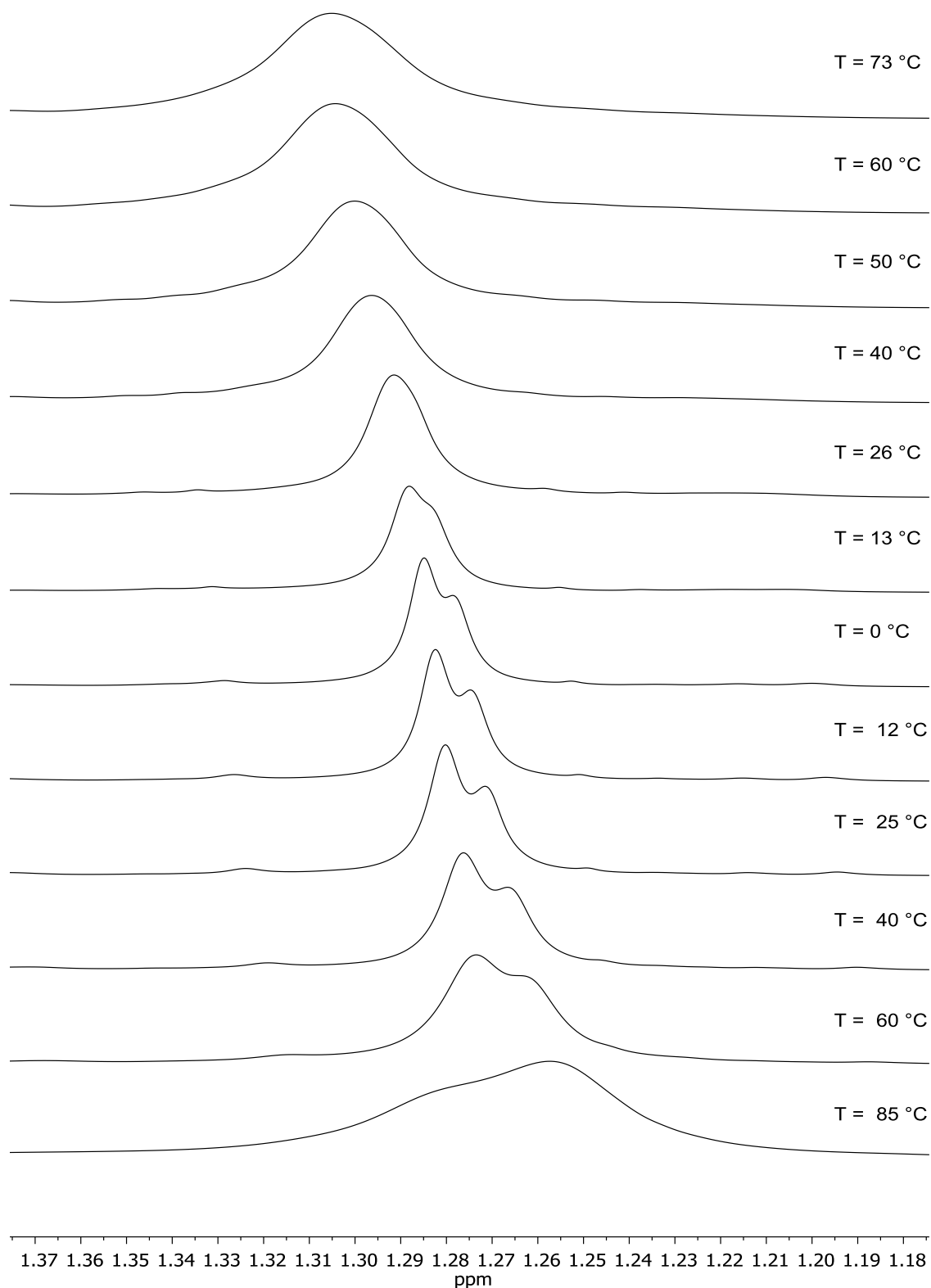


Figure S10. VT $^1\text{H}\{^{31}\text{P}\}$ NMR spectrum of complex **2** in $\text{THF-}d_6/\text{C}_6\text{D}_6$. Due to coincidental overlap of the signals, the decoalescence observed in the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum was not observable here.

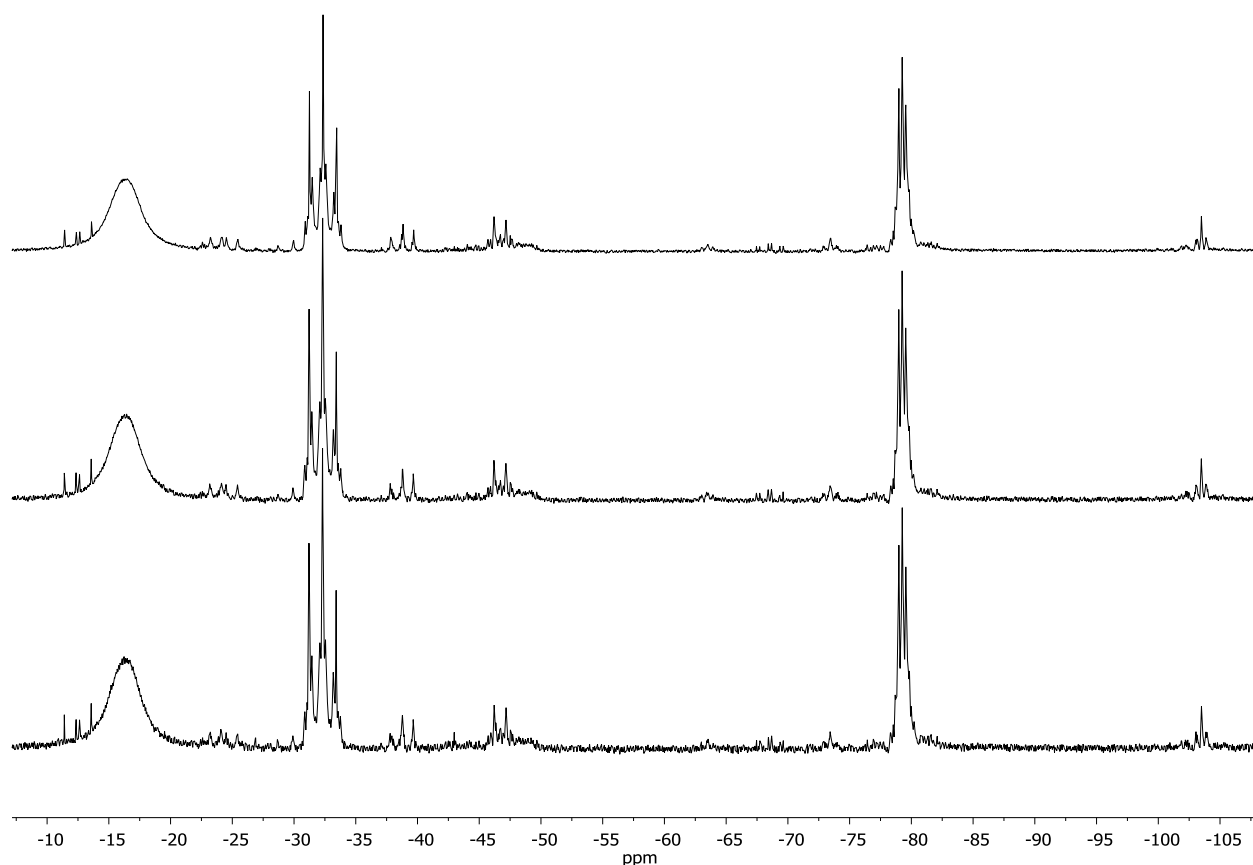


Figure S11. $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of complex **2** in THF/ C_6D_6 at $26\text{ }^\circ\text{C}$ in concentrated solution (top) and diluted with THF- d_6 factor 1.7 (middle) as well as factor 3.3 (bottom). The spectra are virtually identical.

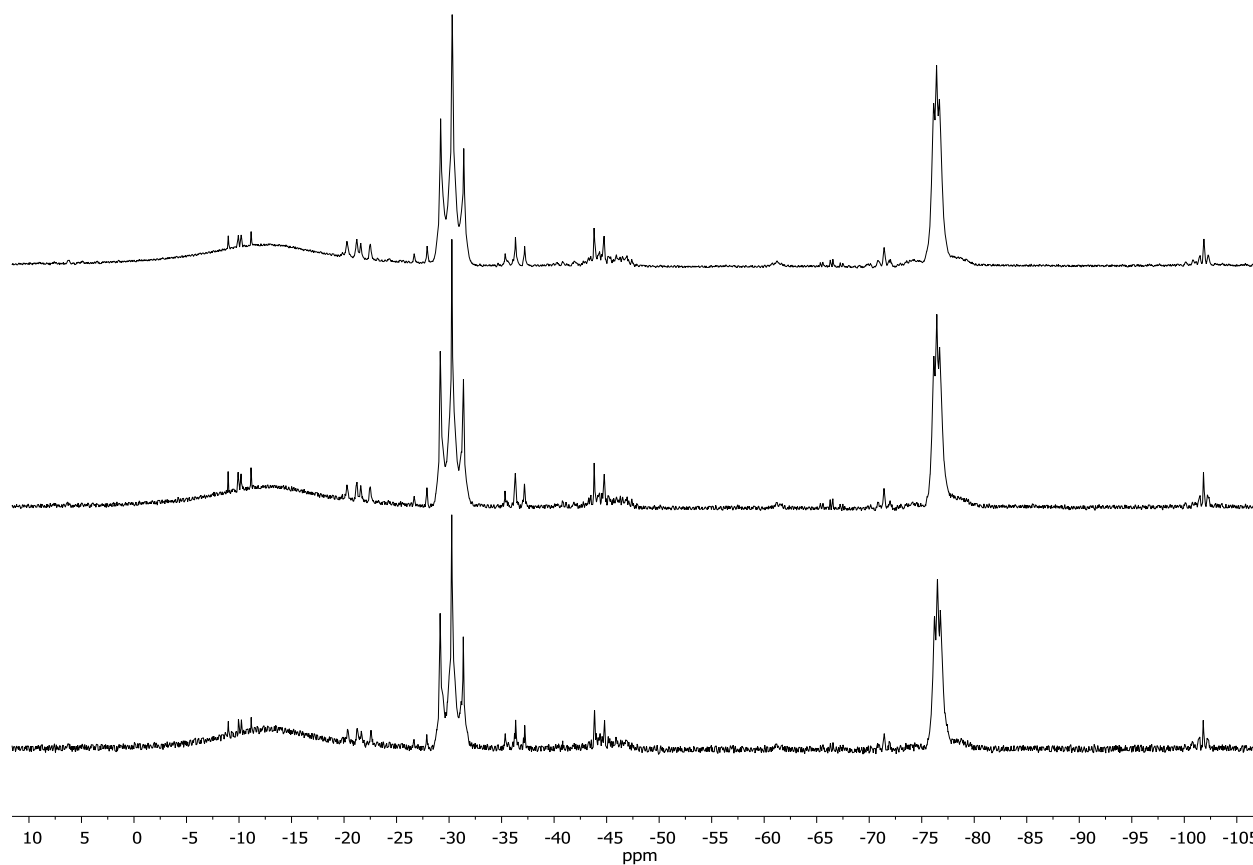


Figure S12. $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of complex **2** in THF/ C_6D_6 at $0\text{ }^\circ\text{C}$ in concentrated solution (top) and diluted with THF- d_6 factor 1.7 (middle) as well as factor 3.3 (bottom). The spectra are virtually identical.

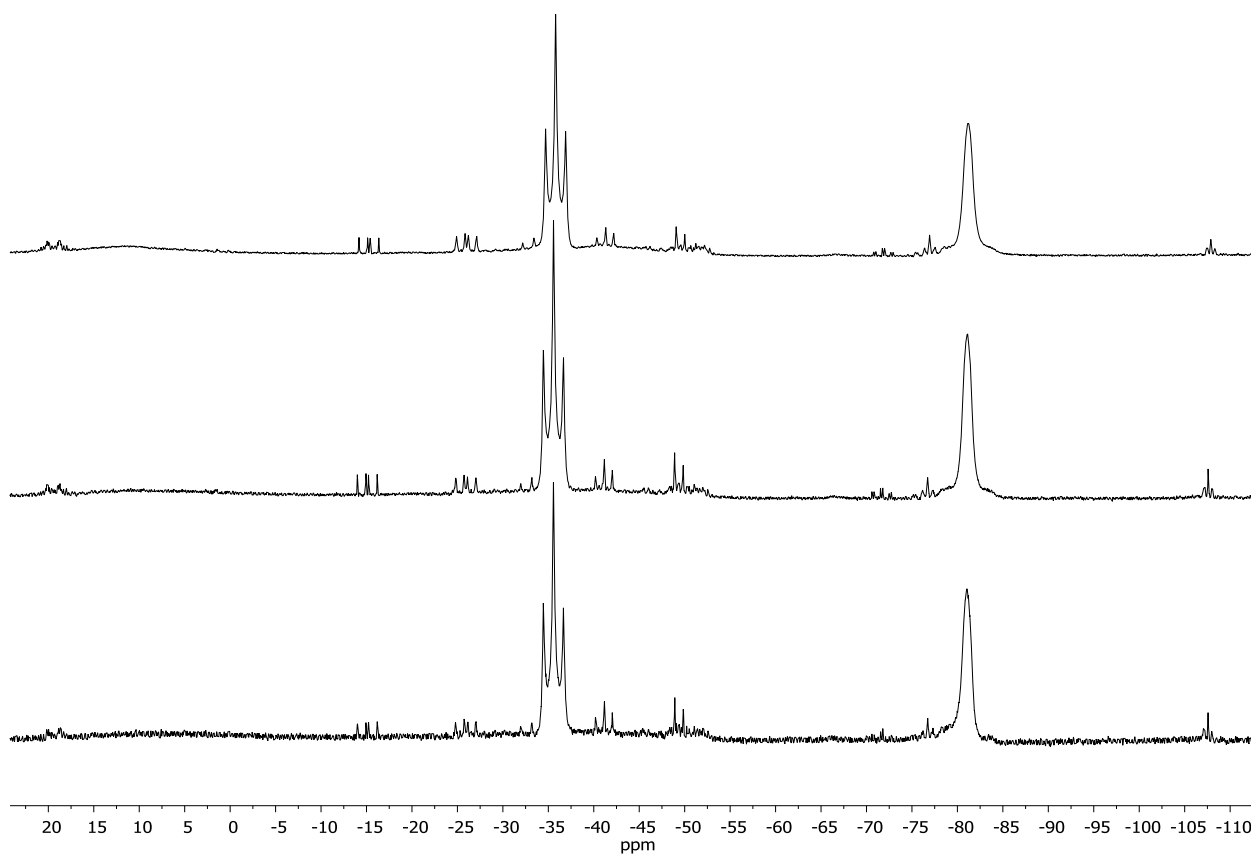


Figure S13. $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of complex **2** in THF/ C_6D_6 at $-25\text{ }^\circ\text{C}$ in concentrated solution (top) and diluted with THF- d_6 factor 1.7 (middle) as well as factor 3.3 (bottom). The spectra are virtually identical.

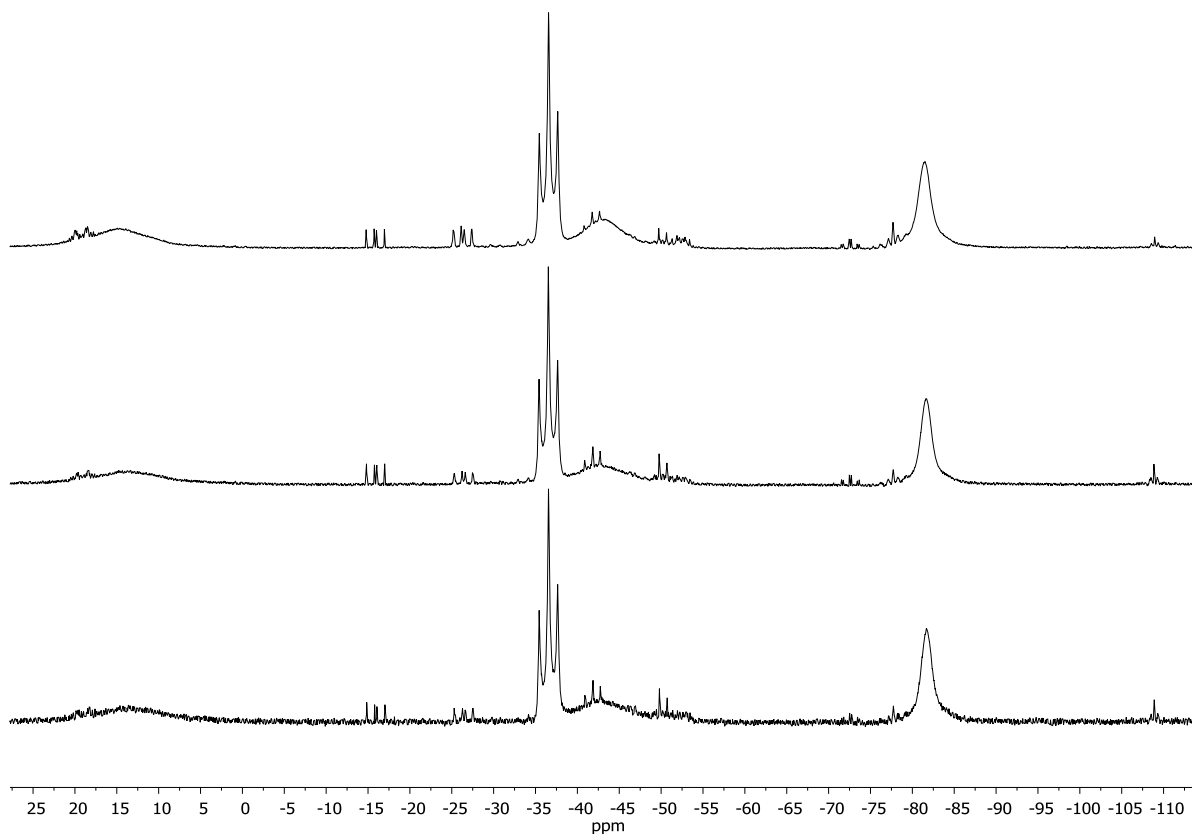


Figure S14. $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of complex **2** in THF/ C_6D_6 at $-40\text{ }^\circ\text{C}$ in concentrated solution (top) and diluted with THF- d_6 factor 1.7 (middle) as well as factor 3.3 (bottom). The spectra are virtually identical.

NMR Spectra of $[(\text{AuCl})_2(1)]$ (**3**)

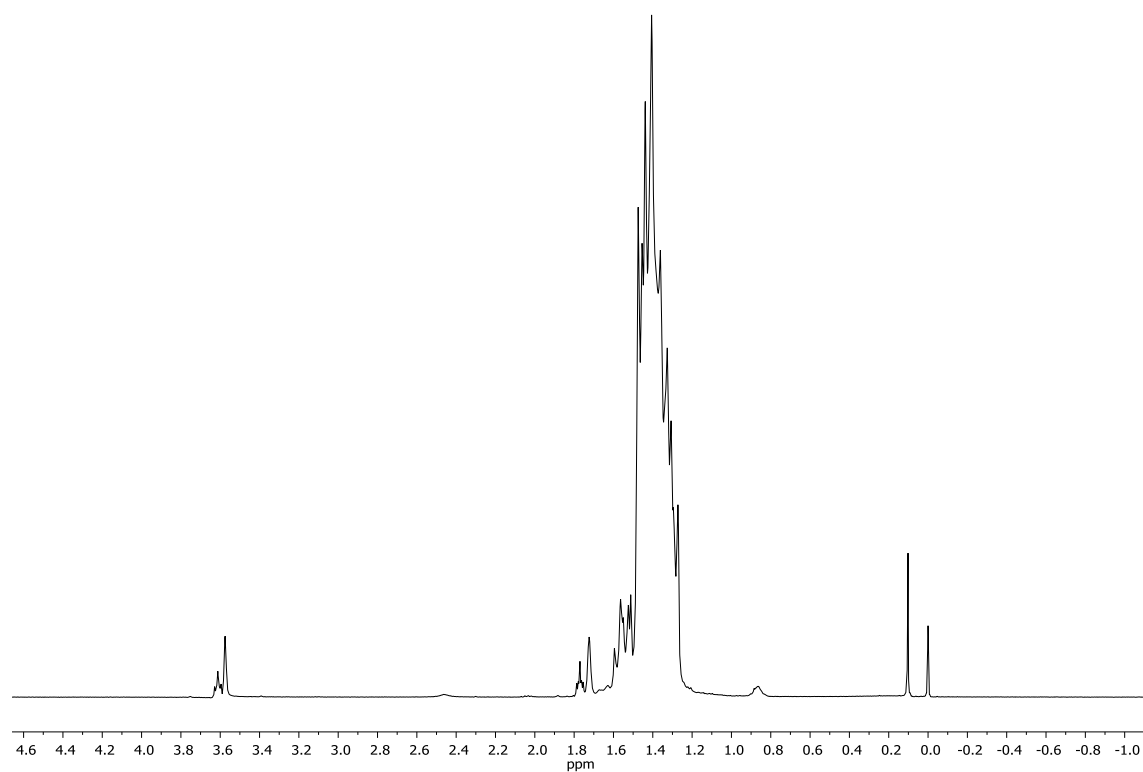


Figure S15. $^1\text{H}\{^{31}\text{P}\}$ NMR spectrum of complex **3** in $\text{THF-}d_8$.

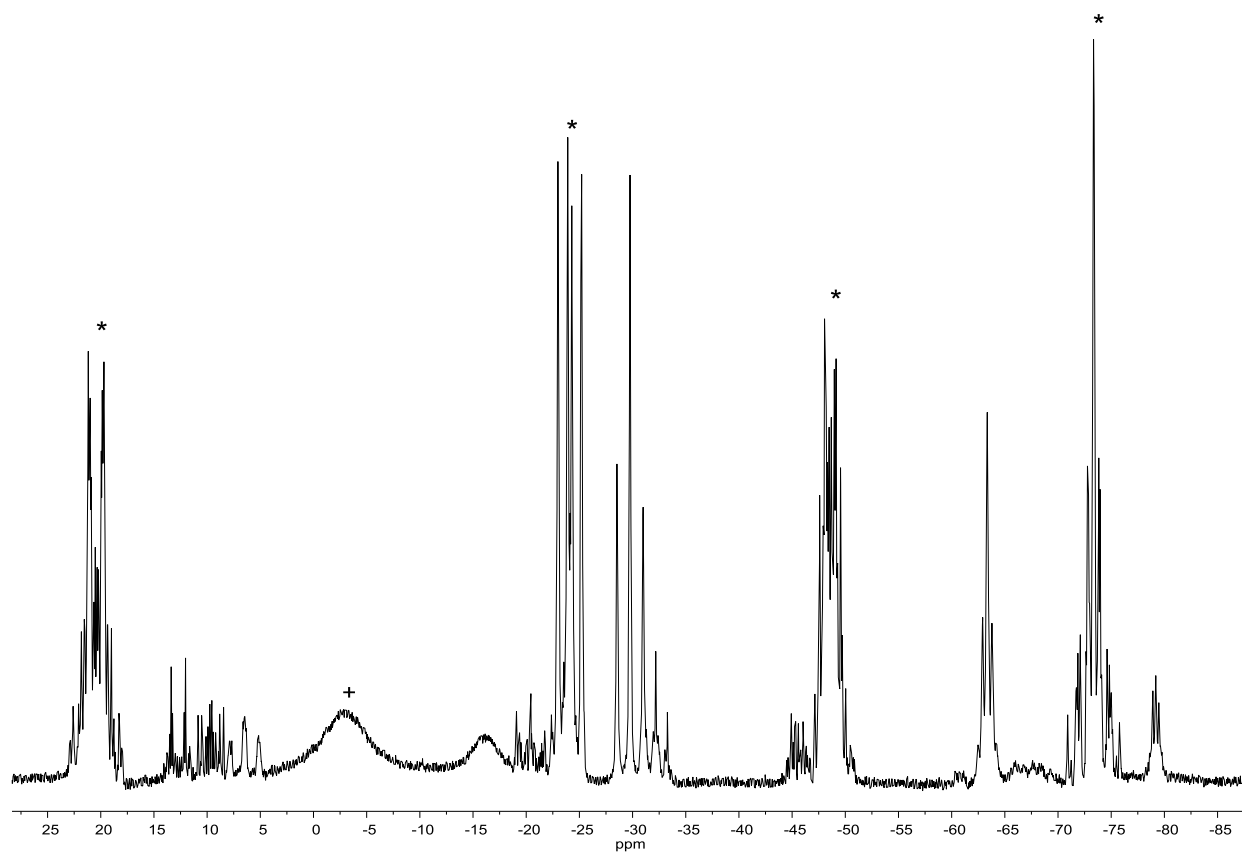


Figure S16. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **3** in $\text{THF-}d_8$. Two sets of signals for the conformers **3a** (*) and **3b** (+) are labeled. Also signals for phosphane **1**, complex **2** and **4** can be detected.

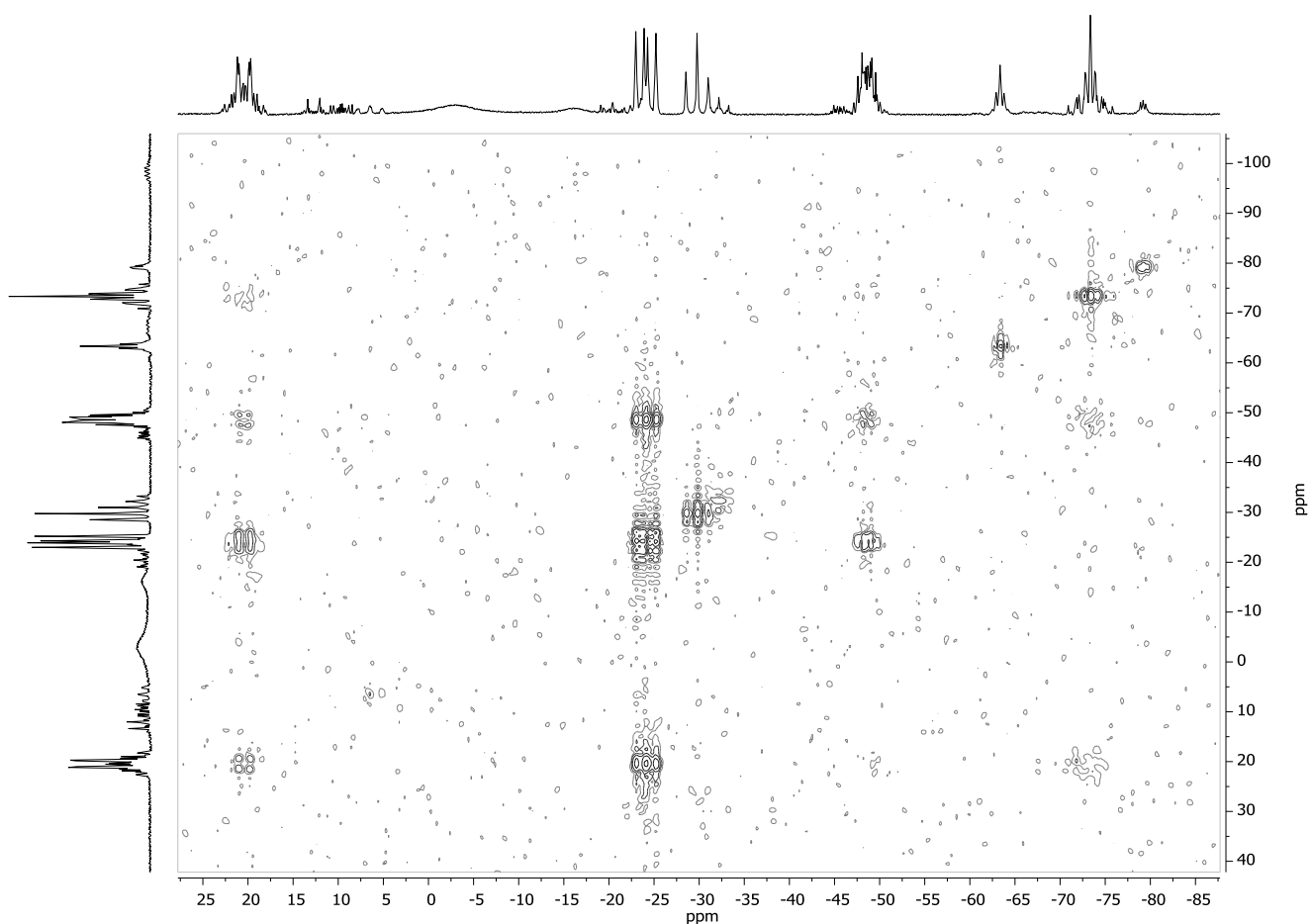


Figure S17. $^{31}\text{P}\{^1\text{H}\}$ - $^{31}\text{P}\{^1\text{H}\}$ COSY NMR spectrum of complex **3** in $\text{THF-}d_8$.

NMR Spectra of $[(\text{AuCl})_3(\mathbf{1})]$ (**4**)

The $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **4** (Figure S19) shows four symmetrical multiplets which are well compatible with an AA'BB'CC'DD' spin system. No information concerning the assignment of the P atoms to the structure could be retrieved from proton-coupled ^{31}P or $^{31}\text{P}\{^1\text{H}\}$ - $^{31}\text{P}\{^1\text{H}\}$ COSY NMR spectroscopy. It is likely that the multiplet at -45.2 ppm corresponds to the γ -P atoms as it is most shielded. Likewise, the two multiplets at lower field ($+12.7$ and $+9.7$ ppm) belong to the coordinating β -P atoms. No change in the spectrum is observed when measuring VT NMR until -80 °C.

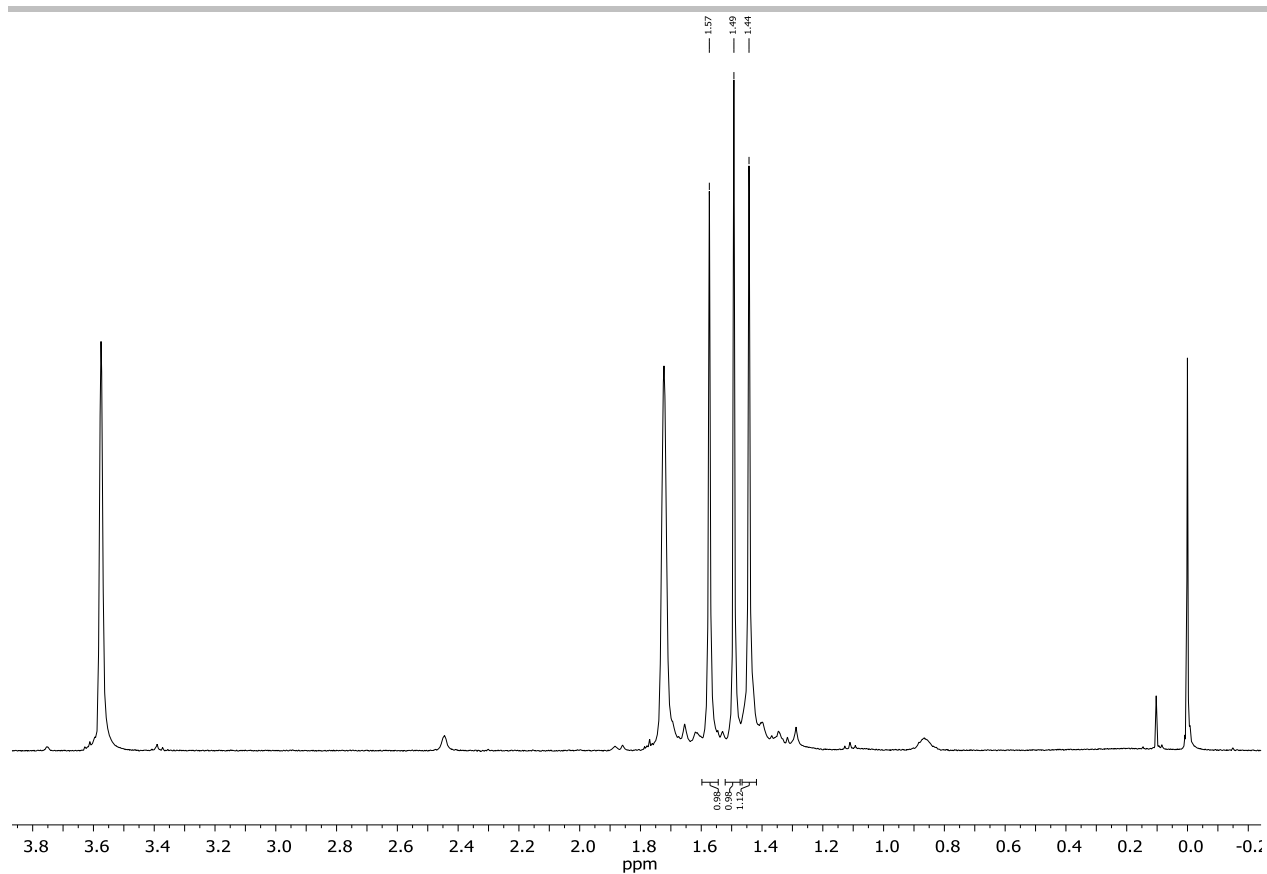


Figure S18. $^1\text{H}\{^{31}\text{P}\}$ NMR spectrum of complex 4 in $\text{THF-}d_8$.

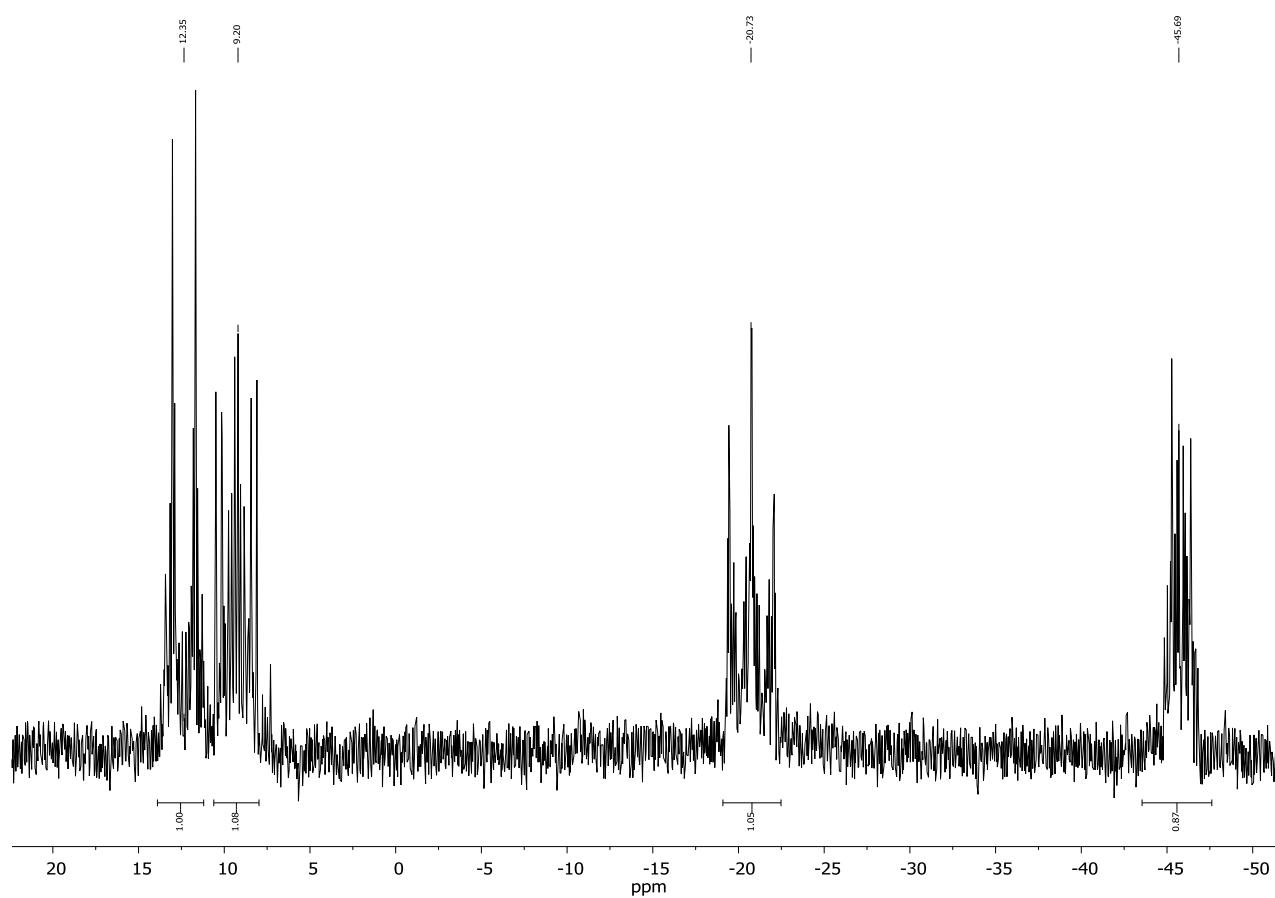


Figure S19. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex 4 in $\text{THF-}d_8$.

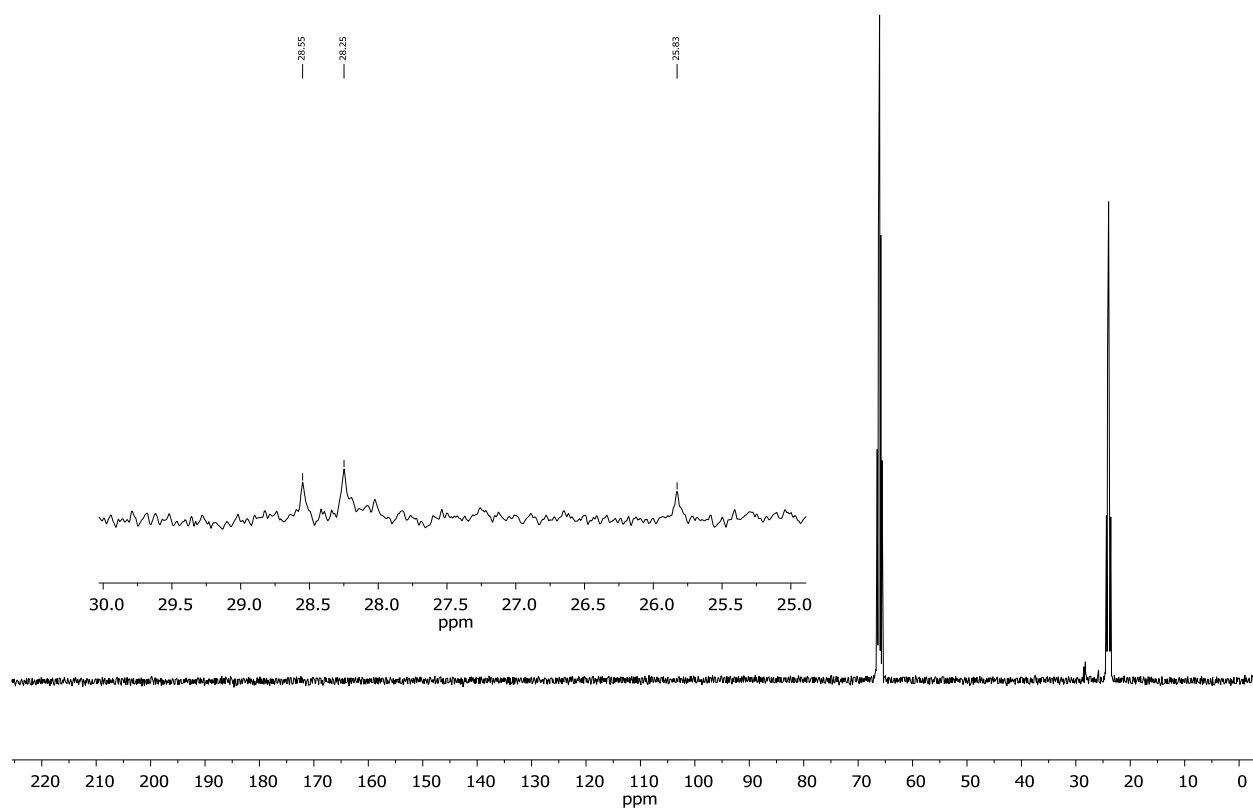


Figure S20. $^{13}\text{C}\{^1\text{H},^{31}\text{P}\}$ NMR spectrum of complex 4 in $\text{THF-}d_8$.

NMR Spectra of Reaction Mixtures

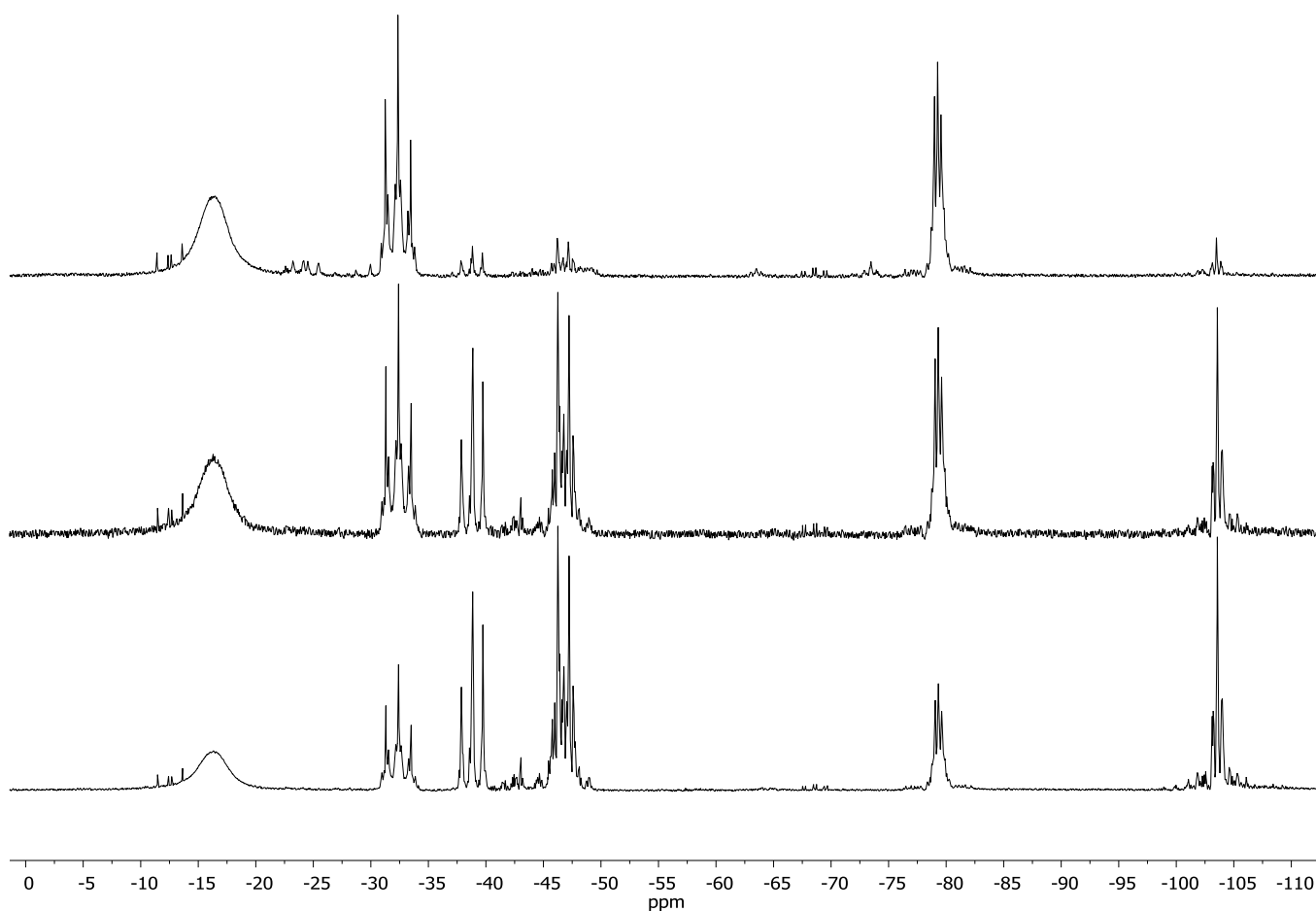


Figure S21. $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of the reaction of phosphane **1** with 1.0 eq. of $[\text{AuCl}(\text{tht})]$ (top), 0.7 eq. of $[\text{AuCl}(\text{tht})]$ (middle) and 0.3 eq. of $[\text{AuCl}(\text{tht})]$ (bottom) in $\text{THF}/\text{C}_6\text{D}_6$.

4. Mass spectrometry measurements

Mass spectrometry measurements were carried out as ESI-MS with a BRUKER DALTONICS FT-ICR-MS spectrometer (Type APEX II, 7 Tesla).

The identity of **1** is supported by mass spectrometry. Two ions attributable to the monoxide of **1** are detected as adducts with H^+ or Na^+ , respectively. They occur together with the higher oxides as a series of six peaks $[\text{M}(\text{P}_8\text{Bu}_6\text{O}_x)]^+$ ($\text{M} = \text{H}, \text{Na}; x = 1-6$) separated by $m/z = 16$. These characteristic series of peaks were also observed for all complexes containing **1** as a ligand. No peaks for $[\text{M}(\text{P}_8\text{Bu}_6\text{O}_x)]^+$ ions with $x = 7$ or 8 are detectable. This is in accordance with the literature, as Baudler *et al.* reported that P_8Bu_6 can be oxidized to the hexoxide but no further.^[5] Obviously, the tendency of the P atoms that carry no $t\text{Bu}$ group to be oxidized is extremely low. Another signal series that can be assigned to $[\text{Na}(\text{P}_8\text{Bu}_6)_2\text{O}_x]^+$ with $x = 2-9$ is found. No peak for the unoxidized phosphane is detectable. This could be due to lower ionization efficiency for the parent phosphane **1** or because it is swiftly oxidized under the conditions of the MS measurement.

5. X-Ray Diffraction

Single Crystal Measurements

Single crystal X-ray diffraction data were collected with a GEMINI CCD diffractometer (RIGAKU). The radiation source was a molybdenum anode (Mo-K α , $\lambda = 0.71073 \text{ \AA}$). The absorption corrections were carried out semiempirically with the SCALE3 ABSPACK module.^[6] All structures were solved by dual space methods with Sir-92.^[7] Structure refinement was done with SHELXL-2015^[8] by using full-matrix least-square routines against F^2 . All hydrogen atoms were calculated on idealized positions. The pictures were generated with the program ORTEP3.^[9] In all pictures, hydrogen atoms were omitted for clarity and thermal ellipsoids are shown with 50% probability. CCDC 1946813 (**1**), CCDC 1946817 (**3**) and CCDC 1946818 (**4**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via <https://summary.ccdc.cam.ac.uk/structure-summary-form> (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: (+44)1223-336-033; or deposit@ccdc.cam.ac.uk). Refinement special details are

- For the crystal structure of **1**: One symmetry-independent THF molecule could be detected in the unit cell but all attempts to refine it failed. Therefore, the SQUEEZE^[10] routine (PLATON) was used. This showed a solvent-accessible void of 334 \AA^3 and 160 electrons per unit cell which fits well with four THF molecules per unit cell resulting in two THF molecules per phosphane molecule. A solvent-free solid-state structure of **1** has already been reported previously.^[11]
- For the crystal structure of **3**: The crystal was non-merohedrally twinned. The model was refined with the twin law $\{-1\ 0\ 0 / 0\ -1\ 0 / 0\ 0\ 1\}$ which can be interpreted as a twofold rotation about $[0\ 0\ 1]$. The twin scale factor was refined to 0.48523. The cell dimensions of a single crystal of **3** was also determined at room temperature (reported in

Table S1) and showed no significant change compared to the measurement temperature (-143 °C).

- For the crystal structure of **4**: The crystal was non-merohedrally twinned. The model was refined with the twin law $\{0\ 0\ 1 / 0\ -1\ 0 / 1\ 0\ 0\}$ which can be interpreted as a twofold rotation about $[1\ 0\ -1]$. The twin scale factor was refined to 0.30586. Three carbon atoms could not be refined anisotropically.

The dihedral angle τ in all molecular structures was extracted from the crystal structures as the mean value of the two dihedral angles between the two bridge phosphorus atoms and two phosphorus atoms connected to them (Figure S22).

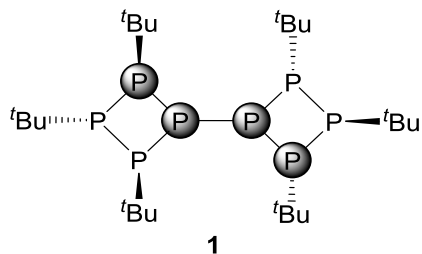
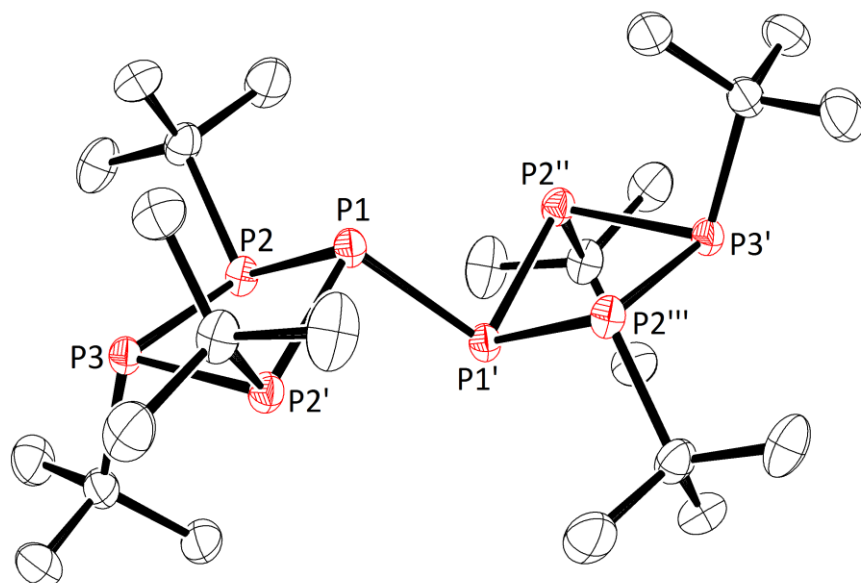


Figure S22. Illustration of the determination of the dihedral angle τ .

Table S1. Summary of crystallographic data.

Molecule	<i>cyclo</i> -(P ₄ Bu ₃) ₂ :2THF(1)	[(AuCl) ₂ (1)] (3)	[(AuCl) ₃ (1)] (4)
CCDC Number	1946813	1946817	1946818
Empirical formula	C ₂₄ H ₅₄ P ₈	C ₂₄ H ₅₄ Au ₂ Cl ₂ P ₈	C ₂₄ H ₅₄ Au ₃ Cl ₃ P ₈
M _w (g·mol ⁻¹)	590.43	1055.26	1287.68
Temperature (K)	220(2)	130(2)	298
Wavelength (Å)	0.71073	0.71073	0.71073
Crystal system	Orthorhombic	Monoclinic	Monoclinic
Space group	<i>Pbam</i>	<i>P2₁/c</i>	<i>P2₁/c</i>
a (Å)	17.6840(4)	9.7470(4)	9.691(12)
b (Å)	11.2190(8)	11.7365(4)	11.843(13)
c (Å)	10.957(4)	33.3202(11)	33.71(4)
α (°)	90	90	89.80(9)
β (°)	90	90.067(5)	89.43(10)
γ (°)	90	90	89.73(10)
Volume (Å ³)	2173.8(8)	3811.7(2)	3869(10)
Z	2	4	4
ρ (calc. in Mg·m ⁻³)	0.902	1.839	2.132
μ (mm ⁻¹)	0.330	8.180	11.481
F(000)	636	2040	2424
Crystal size (mm ³)	0.20 x 0.20 x 0.20	0.05 x 0.05 x 0.05	0.08 x 0.08 x 0.02
θ range for data collection (°)	2.303 – 32.470	2.422 – 31.867	2.335 – 32.435
Index ranges	-26 ≤ h ≤ 22, -6 ≤ k ≤ 16, -15 ≤ l ≤ 15	-7 ≤ h ≤ 13, -15 ≤ k ≤ 17, -48 ≤ l ≤ 46	-29 ≤ h ≤ 29, -17 ≤ k ≤ 17, -29 ≤ l ≤ 29
Reflections collected	19245	15991	16217
Independent reflections	3894 [R(int) = 0.0356]	9163 [R(int) = 0.0463]	16217 [R(int) = 0.1188]
Completeness (%) to 25.242°	99.90	99.90	99.90
Max. and min. transmission	1.00000 and 0.99209	1.00000 and 0.98835	1.00000 and 0.49178
Data / restraints / parameters	3894 / 0 / 83	9163 / 0 / 344	16217 / 0 / 347
Goof on F ²	0.989	1.055	0.963
Final R indices	R1 = 0.0401, [>2σ(I)]	R1 = 0.0683, wR2 = 0.1368	R1 = 0.0629, wR2 = 0.1382
R indices (all data)	R1 = 0.0565, wR2 = 0.0986	R1 = 0.0857, wR2 = 0.1466	R1 = 0.1237, wR2 = 0.1482
Largest diff. peak and hole e ⁻ Å ⁻³	0.277 and -0.169	5.223 and -2.825	4.825 and -3.399


Figure S23. Molecular structure of phosphane 1.

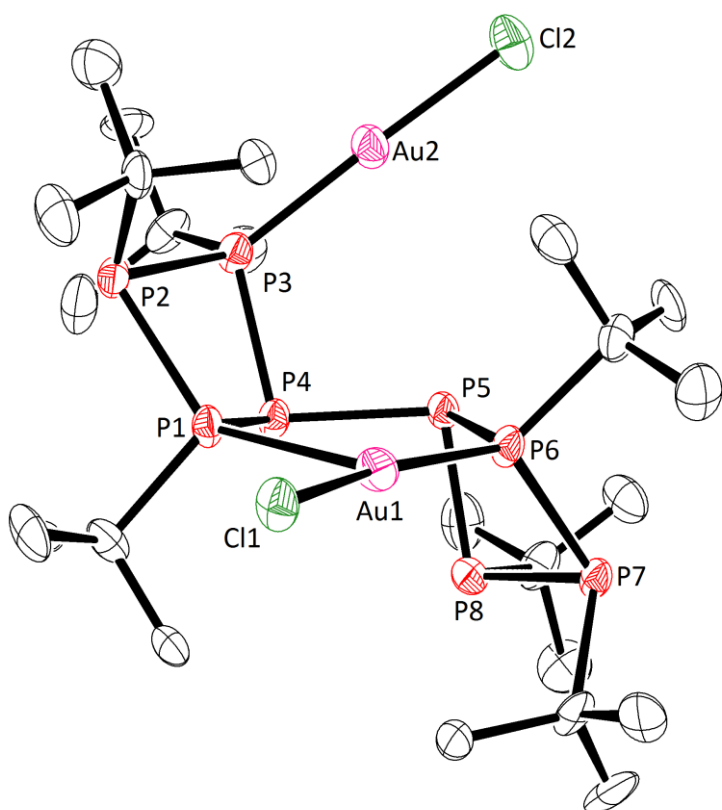


Figure S24. Molecular structure of homobimetallic complex 3.

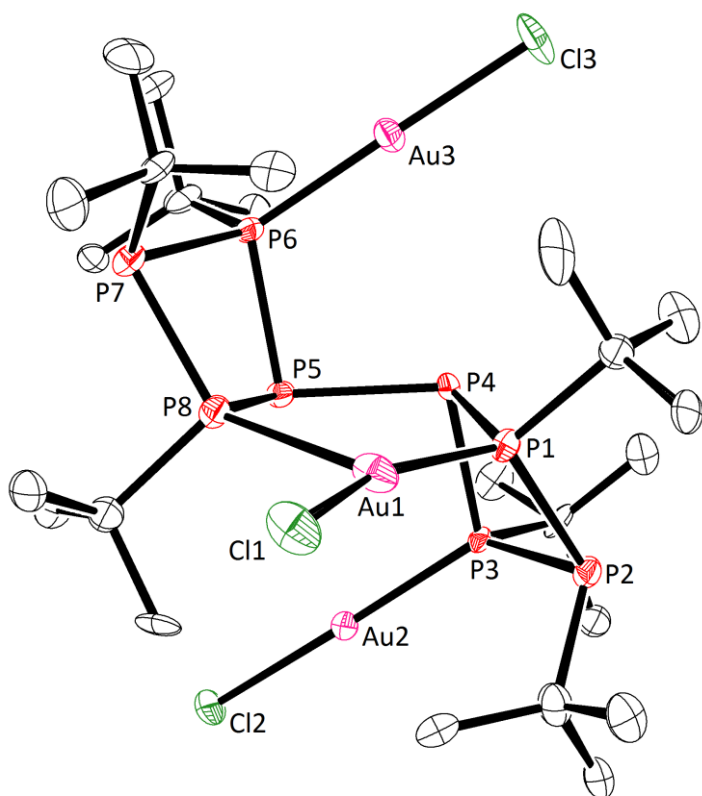


Figure S25. Molecular structure of homotrimetallic complex 4.

Powder Measurements

Powder X-ray diffraction data were collected on a STADI-P diffractometer (STOE) with a silicon solid-state detector Methyn-1K (DECTRIS) at room temperature. The radiation source was a copper anode (Cu-K α , $\lambda = 1.540598 \text{ \AA}$) combined with a germanium single crystal monochromator. Samples were measured in sealed glass capillaries (inner diameter 0.5 mm, HILGENBERG) with Debye-Scherrer geometry. Processing of the raw data was carried out with the diffractometer software WinXPow^[12] (STOE). The theoretical powder pattern from the single crystal structures was calculated using MERCURY.^[13]

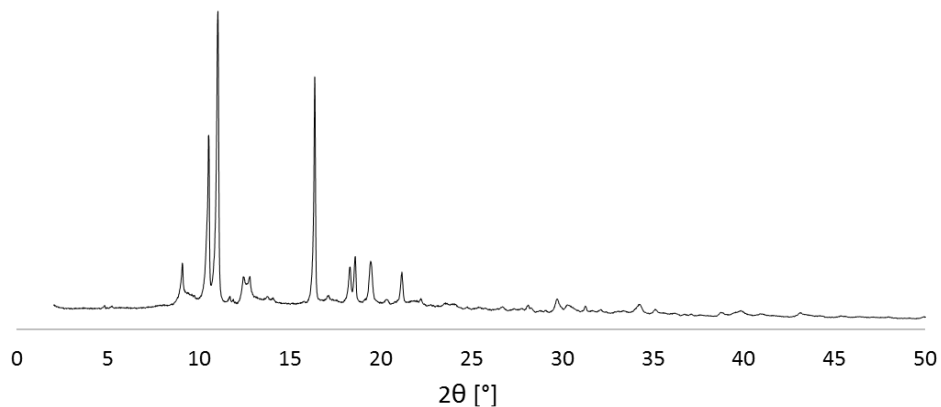


Figure S26. Powder X-ray diffraction pattern of **2**.

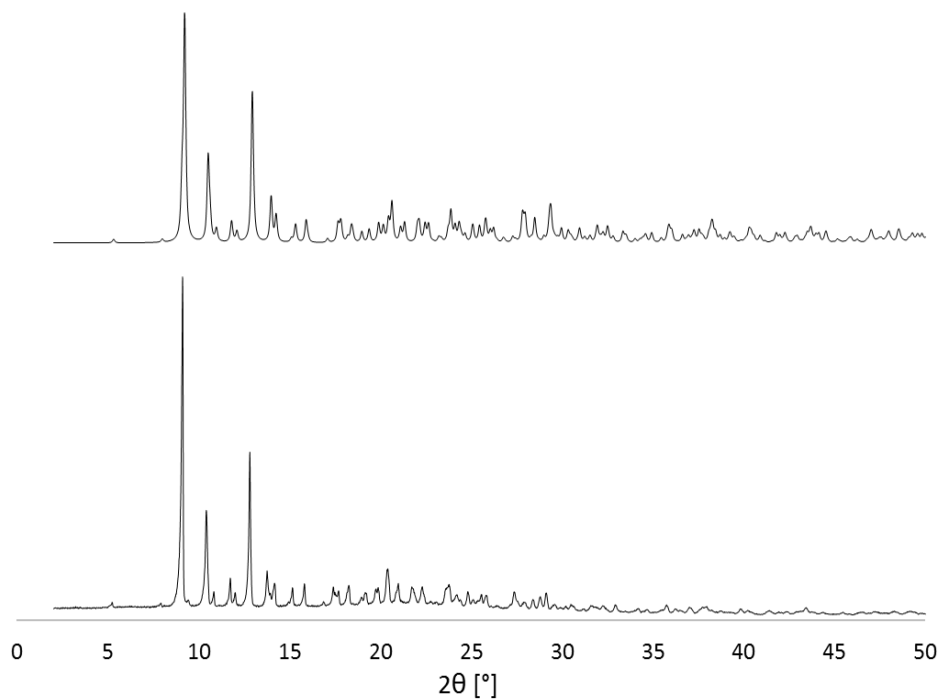


Figure S27. Calculated (top) and experimental (bottom) powder X-ray diffraction pattern of **3**.

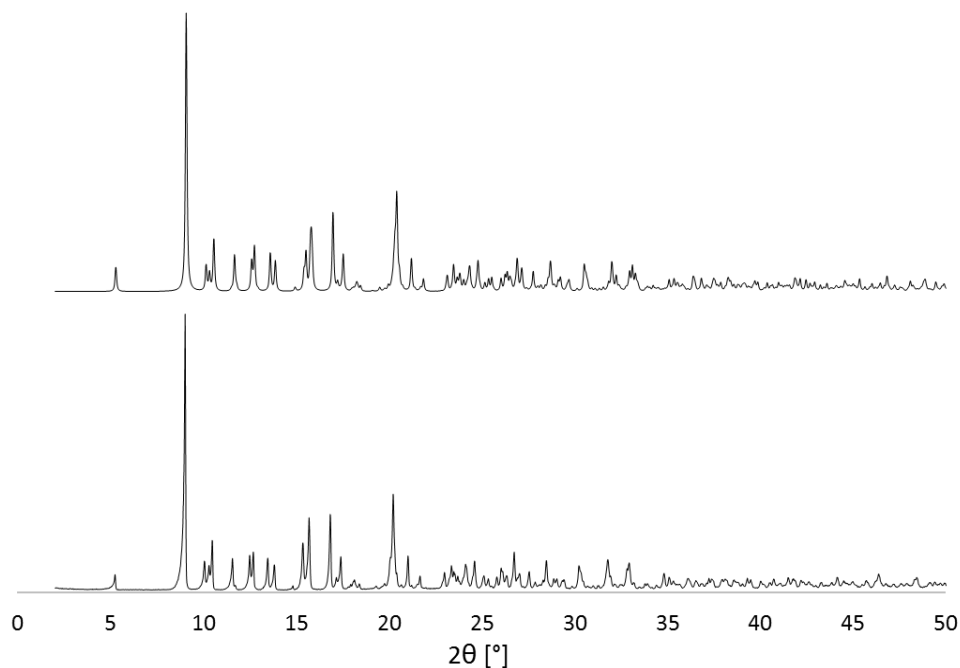


Figure S28. Calculated (top) and experimental (bottom) powder X-ray diffraction pattern of **4**.

CSD Search

The Cambridge structural database (CSD), version 5.39 (February 2018) was used for the crystal structure search. The searches have been performed using the program ConQuest, version 1.21.^[14] The search motif was two phosphorus atoms connected to a gold atom allowing only three bonding atoms for the latter. Figure S29 shows the molecular structures for the search hits.

Table S2. Results of the CSD search. Structural formulas are given in Figure S29. L is the respective co-ligand. (a) This publication; (b) Observations in solution by ³¹P{¹H} NMR spectroscopy; (c) Same observation made at –83 °C; (d) Same observation made at –90 °C; (e) Temperature of the single crystal XRD measurement in °C.

CSD Code	Au–P	Δ(Au–P)	P–Au–L	Δ(P–Au–L)	NMR ^b	Temp. ^e
CEXHAQ ^[15]	2.338(3) – 2.505(2)	0.167	121.9(3) – 156.6(3)	33.7	Equivalent	–173
CEXHIIY ^[15]	2.3681(6) – 2.5311(5)	0.163	130.40(6) – 148.35(6)	18.0	Equivalent	–173
TIJLEF ^[16]	2.333(2) – 2.764(2)	0.431	127.7(2) – 154.5(2)	26.8	Equivalent ^c	–173
24 ^a	2.240(3) – 2.673(3)	0.433	99.7(1) – 162.6(1)	62.87	Equivalent ^d	–143
25 ^a	2.236(4) – 2.629(4)	0.393	105.5(1) – 156.4(2)	50.89	Equivalent ^d	–143

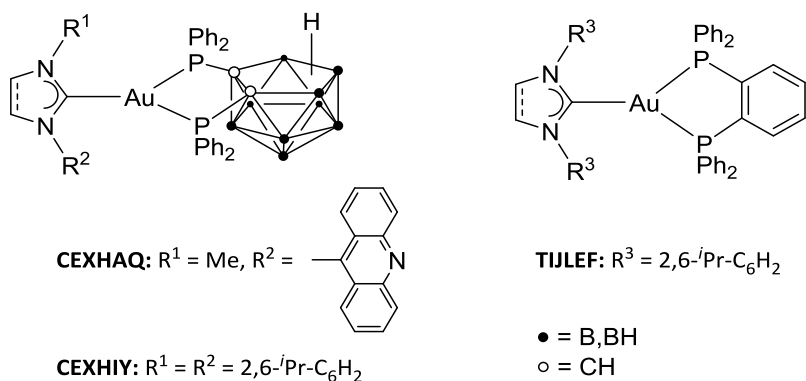


Figure S29. Molecular structures of the hits from the CSD search including their CSD code.

6. Quantum Chemical Calculations

General Remarks

All calculations were carried out with DFT using the program ORCA^[17] (version 3.0.2). For all calculations, the convergence criterion was increased to 10^{-8} Hartree, the atom-pairwise dispersion correction D3^[18] was used and, unless indicated otherwise, a simulation of the solvent environment was performed using the COSMO^[19] model for which the dielectric constant was set to 7.58 (THF). Relativistic effects were considered using effective core potentials^[20]. Furthermore, the RIJCOSX approximation was used to speed up the calculations. In all cases the free Gibbs energy ΔG was calculated using a numerical frequency analysis.

Benchmark Study

The method for optimizing the molecular structures of the gold(I) complexes was benchmarked based on the molecular structure of the trimetallic complex **4** obtained from single crystal XRD measurements. The BP86 functional^[21,22] in combination with the def2-SVP^[23,24] basis set was used due to the best agreement with the experimental data regarding the structural parameters shown in Table S3.

Table S3. Results from the benchmarking study to find the best functional and basis set combination structure for the geometry optimization of the gold(I) complexes. Experimental values taken from the molecular structure of complex **4**.

Method	Au–P	Au...P	Au–Cl	P–P
Experiment	2.236(4)	2.629(4)	2.322(5)	2.226(5)
BP86/def2-SVP	2.281	2.625	2.386	2.237
BP86/def2-TZVP	2.277	2.489	2.392	2.214
M06L/def2-TZVP	2.268	2.952	2.370	2.198
B3LYP/def2-TZVP	2.281	2.698	2.394	2.213
TPPSh/def2-TZVP	2.266	2.589	2.373	2.201
PBE0/def2-TZVP	2.274	2.586	2.373	2.196

Rotational Profile of 2,2',3,3',4,4'-Hexa-*tert*-butyl-1,1'-bicyclotetraphosphane {*cyclo*-(P₄tBu₃)₂} (1)

These calculations were carried out using the B3LYP functional in combination with the def2-TZVP^[23,24] basis set. To obtain the rotational profile of the dihedral angle τ was varied in steps of 5°. The complex was subsequently optimized, restraining the dihedral angle τ . A frequency analysis was carried out to obtain the ΔG values.

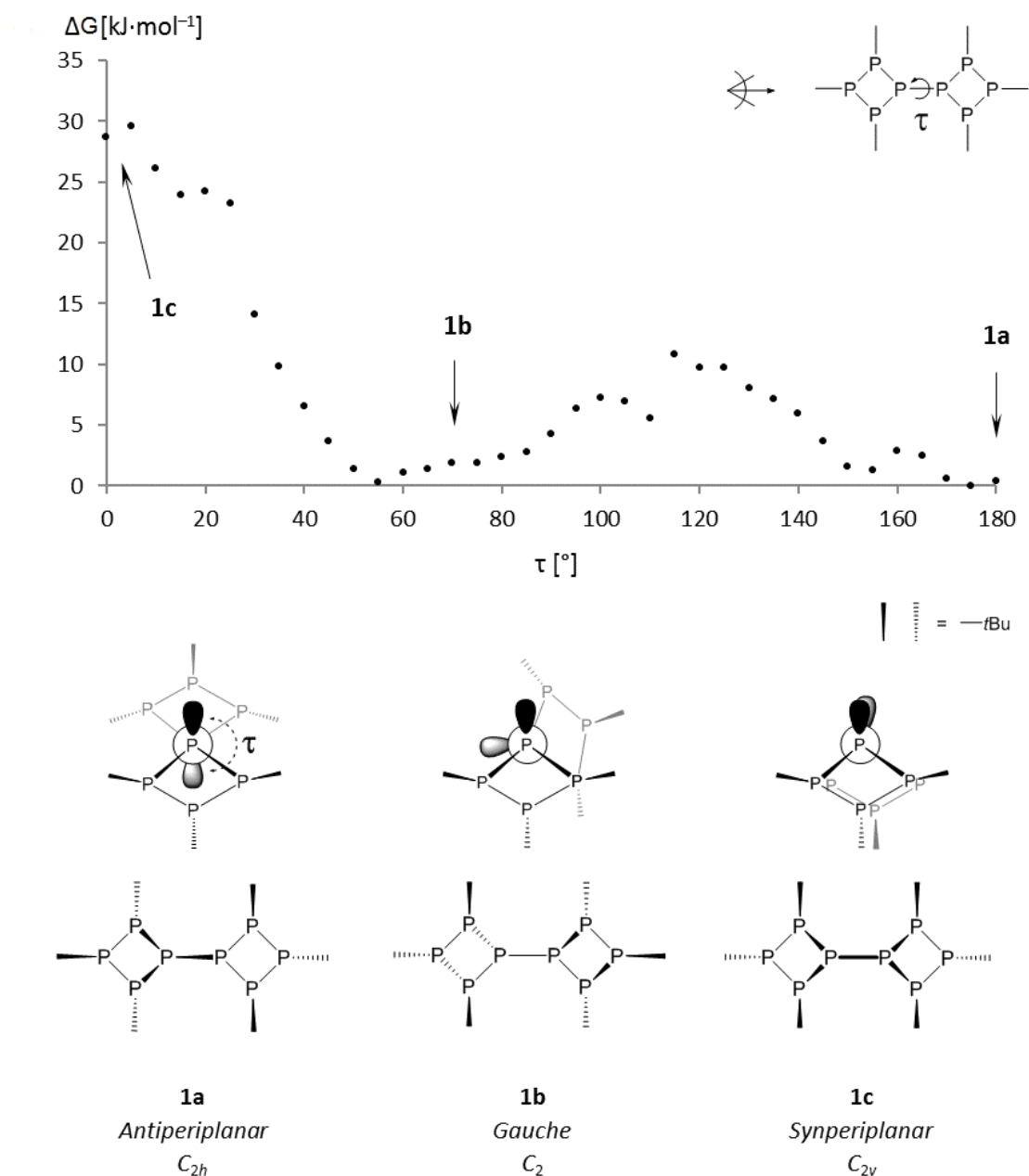


Figure S30. Rotational profile of phosphane 1.

Table S4. Individual energy values [$\text{kJ}\cdot\text{mol}^{-1}$] with dihedral angles τ [°] for the rotational profile of phosphane 1.

τ	ΔG_{rel}	τ	ΔG_{rel}	τ	ΔG_{rel}	τ	ΔG_{rel}	τ	ΔG_{rel}	τ	ΔG_{rel}
0	28.0	35	10.5	70	1.0	105	9.6	140	8.0	175	1.0
5	29.5	40	7.0	75	1.1	110	8.4	145	5.0	180	1.4
10	26.2	45	3.8	80	2.3	115	13.7	150	3.7		
15	23.9	50	1.2	85	3.4	120	12.5	155	3.2		
20	23.8	55	0.0	90	5.2	125	12.2	160	4.4		
25	23.3	60	0.4	95	7.5	130	10.6	165	3.5		
30	14.6	65	0.6	100	9.7	135	9.5	170	1.5		

Rotational Profile of [AuCl(1)] (2)

To obtain the rotational profile of the monometallic gold(I) complex **2**, the angle τ' (Figure S31) was varied in steps of 5°. The complex was subsequently optimized, restraining the dihedral angle which corresponded to the P atom that was attached to the Au–Cl fragment. The individual values are listed in Table S5. In addition to the BP86/def2-SVP, different levels of theory were tested but showed no significant difference.

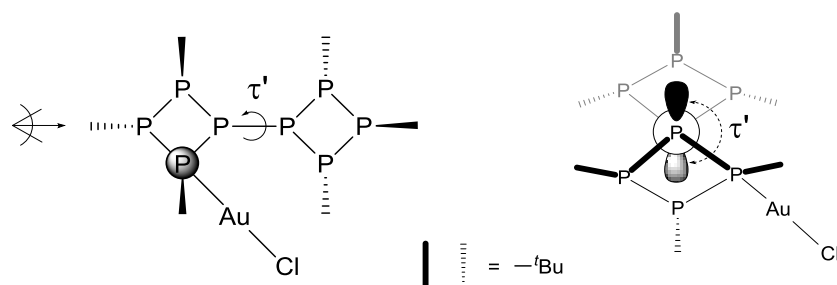


Figure S31. Definition of dihedral angle τ' in complex **2**.

Table S5. Individual energy values (kJ·mol⁻¹) with dihedral angles τ' (°) for the rotational profile of complex **2**.

τ'	ΔG_{rel}	τ'	ΔG_{rel}	τ'	ΔG_{rel}
0	21.8	120	37.8	240	15.7
4	17.0	124	38.3	244	10.7
8	11.2	128	26.1	248	12.4
12	12.0	132	27.3	252	14.2
16	10.8	136	21.7	256	7.6
20	10.1	140	25.9	260	5.2
24	8.1	144	24.9	264	4.7
28	9.8	148	23.6	268	3.2
32	10.4	152	17.4	272	1.5
36	15.6	156	16.1	276	0.0
40	11.4	160	16.6	280	0.9
44	18.2	164	15.9	284	3.3
48	28.5	168	21.5	288	7.7
52	26.3	172	22.8	292	9.9
56	30.2	176	23.8	296	18.3
57	30.9	180	24.8	300	17.8
58	31.6	184	27.9	304	26.9
59	32.3	188	30.8	308	29.3
60	38.5	192	36.0	312	29.8
64	39.6	196	38.5	313	30.5
68	38.8	200	43.2	314	32.0
72	38.5	204	48.0	315	34.1
76	35.2	208	51.5	316	34.0
80	34.8	209	51.8	317	36.5
84	32.5	210	53.0	318	42.4
88	36.0	211	53.6	319	43.5
92	36.2	212	53.7	320	37.6
96	33.2	213	39.3	324	28.2
97	33.4	214	38.5	328	28.2
98	31.3	215	37.9	332	22.3
99	31.8	216	37.2	336	16.6
100	49.1	220	33.5	340	16.4
104	54.0	224	29.3	344	15.9
108	47.6	228	27.7	348	21.5
112	44.2	232	23.6	352	17.3
116	39.1	236	18.8	356	16.6

Rotational Profile of $[(\text{AuCl})_2(\mathbf{1})]$ (**3**)

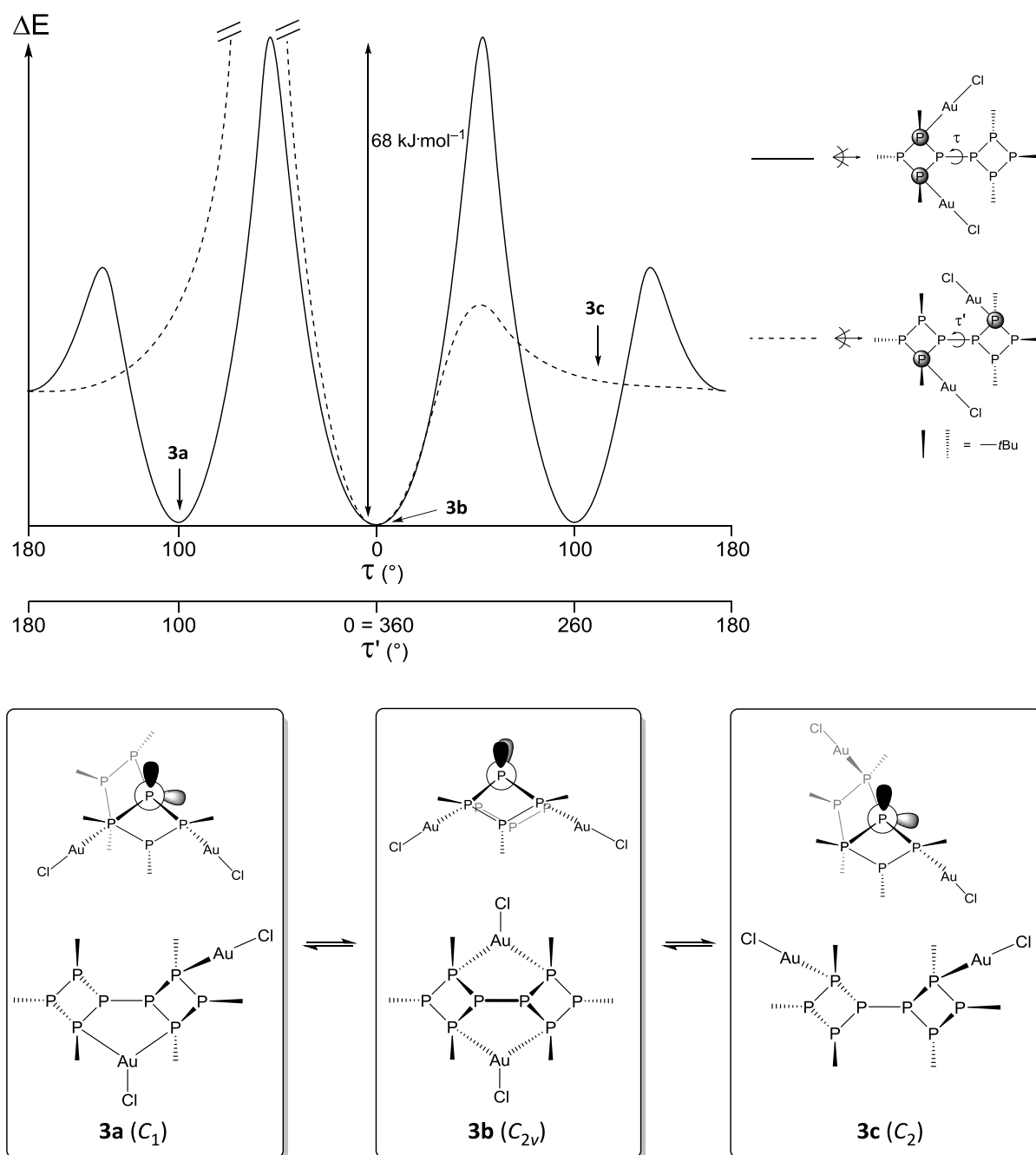


Figure S32. Illustration of the dynamic process in the bimetallic complex **3**. **Top:** Estimated rotational profile; **Bottom:** Involved conformers with corresponding molecular symmetry indicated in brackets.

In the rotational profile of complex **3**, the barriers that arise from the steric repulsion between the bulky $t\text{Bu}$ groups and the Au–Cl fragments and cleavage of the favored distorted T-shaped chelation are $+29.1$ and $+39.1 \text{ kJ mol}^{-1}$ (corresponding activation energies to transform **3a** into **3b** and **3c**, respectively). As bimetallic complex **3** possesses two Au–Cl fragments, rotational barriers of approximately $+68 \text{ kJ mol}^{-1}$ are expected at certain angles τ and τ' for a rotation around the internal P–P bond. In this way, the rotational profile of **3** as shown in Figure S32 can be estimated. It should be noted that two mutually connected rotational profiles are possible. Both cases are discussed in the following:

- If both Au–Cl fragments are coordinated by the same P_4 ring, a symmetrical profile with a range of 180° (represented by τ , solid line in Figure S32) results. Two minima are expected corresponding to a *synperiplanar* conformation **3b** ($\tau \approx 0^\circ$) and a *gauche* conformation **3a** ($\tau \approx 100^\circ$) which should be separated by a barrier of about $+68 \text{ kJ mol}^{-1}$. DFT calculations reveal that both rotamers are almost isoenergetic ($\Delta G(\mathbf{3b}-\mathbf{3a}) = -0.4 \text{ kJ mol}^{-1}$). A low activation barrier is expected for the rotation

which transforms **3a** (C_1 symmetry) into its enantiomer ($\tau \approx 100^\circ$). In this range, an additional rotamer with *antiperiplanar* conformation, featuring two linearly coordinated Au–Cl fragments, is conceivable.

-If each P_4 ring coordinates a gold(I) atom, an asymmetrical profile results which covers 360° (represented by τ' , dashed line in Figure S32). This profile is expected to feature an extremely high rotational barrier due the repulsion of the Au–Cl groups at $\tau = \tau' \approx 100^\circ$. The minimum at $\tau' \approx 0^\circ$ corresponds to a *synperiplanar* rotamer. This should directly be connected to rotamer **3b** by a single Y-shape interconversion rendering this a crossing point between the two systems (with an equally low activation barrier). In addition, at least one minimum which corresponds to an *antiperiplanar* conformer **3c** with two linearly coordinated gold(I) atoms is conceivable.

The assignment to the conformers of **3** to the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum is not entirely clear. The Y-shape coordination flip should lead to three signals with a relative intensity of 2:1:1 for the conformer with *synperiplanar* conformation (**3b**) which would match with the second signal set. The first signal set would accordingly correspond to the *gauche* rotamer **3a**. However, for this conformer either eight signals are expected or three signals with a relative intensity of 2:1:1 considering a rotation around the internal P–P bond leading, similarly to **2**, to a dynamic process which makes all β -P atoms equivalent. Unfortunately, all attempts to elucidate this behavior by further NMR studies were impeded by the low solubility of the complex.

Molecular Orbital Energies

These calculations were carried out using the BP86^[21,22] functional in combination with the def2-SVP^[23,24] basis set.

Table S6. Molecular orbital energies of the molecules 1-4.

<i>cyclo</i> -(P_4tBu_3) ₂ (1)		[AuCl(1)] (2)		[(AuCl) ₂ (1)] (3)		[(AuCl) ₃ (1)] (4)	
Orbital	Energy [eV]	Orbital	Energy [eV]	Orbital	Energy [eV]	Orbital	Energy [eV]
HOMO	-5.098	HOMO	-4.696	HOMO	-4.876	HOMO	-5.109
HOMO-1	-5.219	HOMO-1	-5.497	HOMO-1	-5.659	HOMO-1	-6.108
HOMO-2	-5.244	HOMO-2	-5.593	HOMO-2	-5.996	HOMO-2	-6.124
HOMO-3	-5.416	HOMO-3	-5.887	HOMO-5	-6.196	HOMO-3	-6.272
HOMO-4	-5.581	HOMO-4	-5.949	HOMO-7	-6.468	HOMO-9	-6.616
HOMO-5	-6.021	HOMO-6	-6.185	HOMO-8	-6.599	HOMO-10	-6.728
		HOMO-7	-6.511	HOMO-10	-6.866		

Transition States

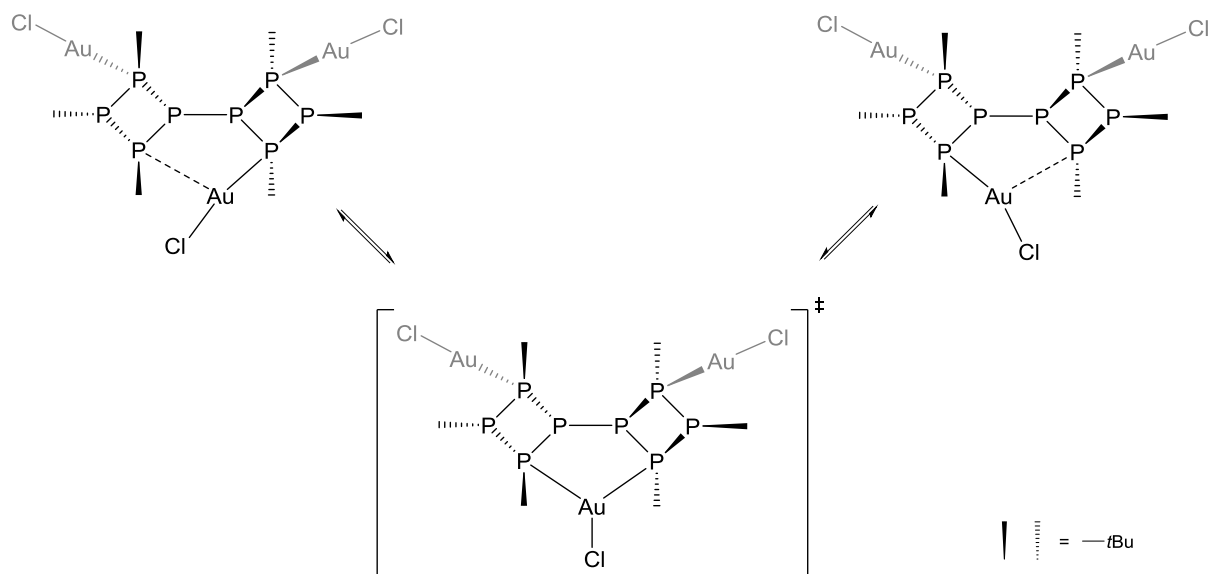


Figure S33. Representation of the Y-shape coordination flip of the chelated gold(I) atom in complexes **2-4**.

These calculations were carried out using the BP86^[21,22] functional in combination with the def2-SVP^[23,24] basis set according to the results of the benchmark study. The transition states for the Y-shape coordination flip were located by a relaxed surface scan of the P–Au–Cl bond angle using the method described above. The transition state structures were verified as such by showing only one negative eigenvalue of the Hesse matrix which corresponds to the described chemical motion. Furthermore, these structures were moved along the respective mode and subsequently optimized which yielded the two corresponding minima.

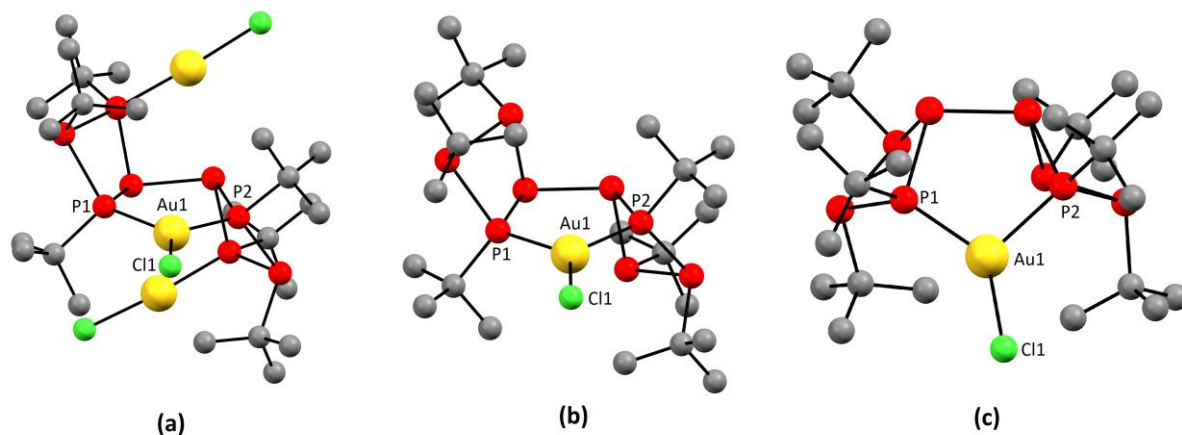


Figure S34. Calculated transition states for the Y-shape coordination flip of gold(I). Homotrimetallic complex **4** with *gauche* conformation (a), monometallic complexes **2a** with *gauche* conformation (b) as well as **2b** with *synperiplanar* conformation (c). Structural parameters given in Table S7.

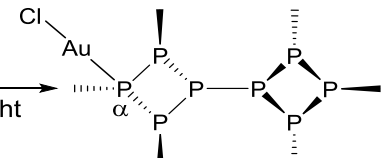
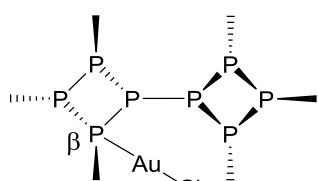
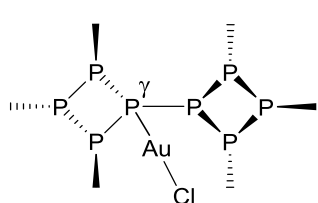
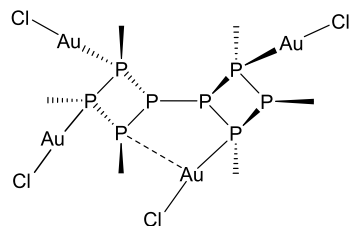
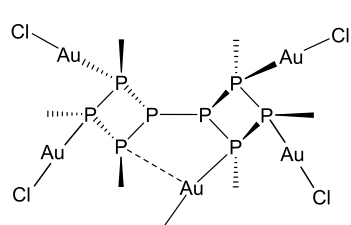
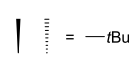
Table S7. Activation energies ΔG [kJ·mol⁻¹] as well as selected bond lengths [Å] and angles [°] of the calculated transition states of complexes **2** and **4**. Letters in brackets correspond to the transition state structures in Figure S34.

Parameter	Complex 4 (a)	Complex 2 (b)	Complex 2 (c)
Conformation	<i>gauche</i>	<i>Gauche</i>	<i>Synperiplanar</i>
G_{act}	+10.0	+13.7	+17.6
Au(1)–P(1)	2.3915	2.3918	2.3732
Au(1)–P(2)	2.3806	2.3918	2.4217
P(1)–Au(1)–P(2)	100.08	99.82	99.05
Cl(1)–Au(1)–P(1)	128.83	129.96	136.2
Cl(1)–Au(1)–P(2)	131.09	130.22	124.33

Complex Formation Energies

These calculations were carried out using the BP86^[21,22] functional in combination with the def2-TZVP^[23,24] basis set.

Table S8. Complex formation energies of hypothetical monodentate complexes as well as complexes with up to five Au–Cl fragments. $\Delta\Delta G$ refers to the complex formation energy with one Au–Cl fragments fewer.

Reaction	ΔG_{react} [kJ·mol ⁻¹]	$\Delta\Delta G$ [kJ·mol ⁻¹]
$1 + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}}$ 	-87.3	
$1 + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}}$ 	-100.4	
$1 + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}}$ 	-82.3	
$\text{P}_8\text{tBu}_6 \quad 1 + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}} [(\text{AuCl})(\text{P}_8\text{tBu}_6)] \quad 2$	-119.8	
$[(\text{AuCl})(\text{P}_8\text{tBu}_6)] \quad 2 + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}} [(\text{AuCl})_2(\text{P}_8\text{tBu}_6)] \quad 3$	-204.3	-84.5
$[(\text{AuCl})_2(\text{P}_8\text{tBu}_6)] \quad 3 + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}} [(\text{AuCl})_3(\text{P}_8\text{tBu}_6)] \quad 4$	-296.6	-92.2
$[(\text{AuCl})_3(\text{P}_8\text{tBu}_6)] \quad 4 + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}}$  $[(\text{AuCl})_4(\text{P}_8\text{tBu}_6)]$	-365.5	-59.9
$[(\text{AuCl})_4(\text{P}_8\text{tBu}_6)] + [\text{AuCl}(\text{tht})] \xrightarrow{-\text{tht}}$  $[(\text{AuCl})_5(\text{P}_8\text{tBu}_6)]$	-406.0	-49.5
		

Cartesian Coordinates from Geometry Optimizations of Selected Molecules

Table S9. Cartesian coordinates of optimized structures for the conformers of complex 2.

Rotamer 2a			Rotamer 2b			Rotamer 2c					
P	0.17401128028648	-0.16947954910993	-0.12325557835051	-0.04757615414377	-0.26276762336931	-0.11390712834349	P	0.12124176635390	0.29063998556102	-0.29781244801520	
P	-0.02837929080140	-0.0526315477206	2.12540817602776	P	-0.20232384670023	0.00506603727223	2.14067332741589	P	0.11658960115917	0.0482837290277	1.95900161316365
P	2.19887370570073	0.11423831897997	2.10277973075011	P	0.202211739181059	-0.063368974036092	2.00036853574809	P	2.9667688062546	0.49970352293338	1.859319389976021
P	0.24352795956685	0.08412383318717	0.08004885647893	P	1.808718115686149	1.02425696978753	0.05695772071929	P	0.2165788127344	1.45024421695717	-0.15724104150251
P	3.28842853193198	-0.00338684423770	-1.39308263102793	P	3.36603755234703	-1.77997815611661	-1.35713670898548	P	3.1986470127344	-0.20059486227292	-1.15484924518926
P	5.24154810113938	1.07623566606937	-0.8156023844308	P	3.86303667097320	-1.89019529098568	-0.6432409338174	P	2.3123100384638	0.32006920925403	-3.1647491032867
P	6.22429798167850	-0.95757165337922	-1.205244110213322	P	3.34687747583155	-2.66559101150198	-2.6967003652155	P	4.4225291728663	0.80077241208637	-3.81712937511081
P	4.28765016738294	-1.27196226427250	-0.36293182391711	P	2.02439870505209	-0.84400371182616	-2.86173312739108	P	4.96242105182064	1.12860817527461	-1.6532847591391
P	3.80895937481777	-2.32476025133374	-1.41222046440150	C	2.62724093377566	-0.06930366588544	-4.50837533032590	C	6.48667209467900	0.00633128735490	-1.38779304357670
C	5.3567082349110	1.81398651913050	-2.82360513525499	C	5.77225516369935	-1.90164717881591	-0.1080753757315	C	2.01372093651444	-1.40116387553146	-3.94755328880804
C	7.778187938544	-0.99025758758376	0.32589159361663	C	2.11015432775345	-0.46069123783649	-2.254664131781334	C	4.21704728464898	2.6036303607584	-4.43659996781911
C	-1.0435387415572	1.13540406487833	-0.76038739432218	C	-1.32454656470209	0.101924200567067	-0.75962396266955	C	-1.17502545406685	1.66971262377524	-0.56129536150561
C	-0.44392636357353	-1.81201708212975	2.68765270851033	C	-0.76201535150333	-1.70154389593412	2.82290950003690	C	-0.00595415842489	-1.85130913413754	2.18206106825839
C	2.17367804450592	-1.49319477551849	3.28959947031893	C	2.79715511895183	1.17013189538023	3.20680065832305	C	2.71059894507006	1.82566534117913	3.128771028636275
Au	3.49648428350428	-1.77321276192456	2.15033114548864	Au	3.02043971558794	-2.1600170055428	1.87960496138602	Au	3.80129262484470	-1.20981541374753	1.82398378108913
Ci	4.5872916194999	-3.76752895145516	2.91925858155516	Ci	3.86455858282826	-4.28427093007742	2.43304726519280	Ci	5.34870357578912	-2.9619134243560	1.87890941348556
C	6.9939875674856	2.30708521540987	-2.82098247431800	C	6.19239946285485	-1.07887315681378	0.61731003695058	C	1.8313127258413	-1.13152985977687	-5.4522256655722
C	5.30704967835188	0.82015168365303	-3.96829184871663	C	6.15000786638236	-3.3830449927455	-0.4438661816345	C	3.16179326448276	-2.3894855854576	-3.7127447843322
C	4.50663640291940	2.99332438266907	-2.92362852774763	C	6.40203385953534	-1.31862857368206	-1.88272749896178	C	0.71388923851116	-1.93457874118815	-3.32153214458903
H	7.70365902771444	1.45905944556104	-2.72068781135637	H	7.29923396164771	-1.0811131727283	0.7115488728002	H	7.26159557103080	-0.7244812325249	-5.90316394011248
H	7.22307540301817	2.82305258280995	-3.78118094649862	H	5.86447446436589	-0.02257018002505	0.52412506886440	H	1.58950952820779	-2.0786975251328	-5.8005102904488
H	7.18673089756468	3.02108039201156	-1.99349187544818	H	5.766858137315850	-1.42632080519360	1.55338443384501	H	1.08005128610340	-0.41315246350497	-5.64425586316406
H	5.48609397039198	1.32056708175304	-4.94525057605163	H	5.69809693709355	-3.8963780625009	0.46529299379785	H	9.22475309346770	-3.36631731058075	-1.8869499277298
H	5.99535214920607	-0.04602827989086	-3.89444396756562	H	5.83745008098873	-3.9797620579762	-1.32409165094730	H	4.11085386651580	-0.21079213531019	-4.1486635542242
H	4.26657267143773	0.4302956800256	-3.97005258862329	H	7.26071137187822	-3.47284464226568	0.65226096079832	H	3.2473403821398	-2.57122490537266	-2.63038384520665
H	4.6904931033641	3.52582125428508	-3.88982587974604	H	7.50835225414376	-1.37183742313548	-1.0103984474318	H	4.0569878254710	-2.91908293636462	-3.76861585108777
H	3.49747905227370	2.641511510085285	-2.88507514349538	H	6.08768953741205	-1.88487910488640	-2.78297046368892	H	8.2030591812780	-0.07037293131082	-2.22570256390892
H	4.69945419484935	3.72486944999066	-2.10263918057708	H	6.11772019427009	-0.25567254514543	-0.2443752922399	H	-0.13750040383880	-1.24393051661018	-3.49036426305980
C	2.53865895346960	0.94081401297050	4.71370082537256	C	2.80329622855984	0.51022849505653	0.59507332037856	C	2.65348004360063	1.14857830169449	4.50811672010259
C	4.19082304714158	1.78200896782556	2.97929624458793	C	4.22921083561920	1.39554155880231	2.69463771981274	C	4.1352525200118	2.29564150489771	2.78353219570004
C	1.84720231727325	2.74162078131798	3.02735950821734	C	1.99841256398840	2.48028280185051	0.21879663596857	C	1.70615989263008	2.98191015936303	3.02998335452462
H	2.87781692547436	1.70038111791812	5.4940651617348	H	3.31363992320678	1.17980920595079	3.51945308394475	H	9.23216270926627	1.88320513847392	5.2956436335948
H	1.47582641117918	1.01034513595771	4.93182777080515	H	3.177430284411965	0.33248142895282	4.96462159124338	H	1.63378390264720	0.1754752956996	4.73130638605874
H	3.13493188774265	0.10795234620818	4.86520736929424	H	3.34122611718415	-0.46220234523395	4.57970155627022	H	3.6268979842538	0.29757248680735	4.57155006652184
H	4.54522139683963	2.62655650624915	3.60675866205016	H	4.7506445426710	-0.12173500078143	3.3570109695070	H	4.43731166211897	3.10747744598652	3.74785920126514
H	4.82432374976913	0.90004242990492	3.19619959287311	H	4.80924094986329	0.45171985460663	2.69368153556571	H	4.86842851581490	1.46860700501625	2.8755131111392
H	3.44213857747150	2.06079677801751	1.91432142081143	H	4.2355386445356	1.81145968786382	1.66453312867392	H	1.9248186483242	1.69433189665197	1.74845942244478
H	2.15464871897757	3.53076310473492	3.79075041750796	H	2.47698293251632	3.19628425020289	3.91996102211995	H	1.96469645287068	3.76119228551137	3.7773576690498
H	1.96728001700653	3.14729076770323	2.04728625818425	H	1.97164356172422	2.94761095309982	2.21352462345543	H	1.72667393335058	3.45341092458131	2.02640190032233
H	0.77252068793855	2.52483681786332	3.23780675678948	H	0.95376443273233	2.315110295152320	3.6507026364499	H	0.67201085306459	2.63934933289598	3.23523336517723
C	0.02940321087015	-1.92008313547597	4.1464308059790	C	-0.12894258285894	-1.84323298185381	4.21688236111544	C	0.59105979516474	-2.16314071596637	3.56493704022568
C	1.98177046512267	-1.87829060644552	2.600922052695062	C	-2.29413612058763	-1.56236385369196	2.93477911853326	C	-1.52416924187709	-2.12251890955061	2.16390567820822
C	0.1804276553277	-2.91698108584830	1.83364613424885	C	-0.40362529501184	-2.89825180683070	1.94187174906502	C	0.68331048513174	-2.67398657010850	1.08177313717545
H	-0.29598998935054	-2.89315746087589	4.57262516327927	H	-0.53390466646396	-2.74735063879039	4.71948809271112	H	0.42457113713398	-3.2338618489586	3.81115959705699
H	1.13699397076286	-1.87952610652876	4.21129557225253	H	0.97208232399553	-1.9667665810361	4.14521600828967	H	1.68780626594077	-1.9860653175751	3.58031393350842
H	-0.39241445045658	-1.11293822097771	4.7813147454049	H	-0.34868035937063	-0.9654620270987	4.862428098498277	H	0.12456016947017	1.05210727500801	4.36521785979463
H	-0.12106478231736	-3.91201590530537	2.22564984998282	H	-0.78997938020498	-3.83253423186715	0.92595480804150	H	0.56619262958475	-3.75689605155936	1.3067090071589
H	-0.13716162400220	-2.84663927176825	0.774811764153877	H	-0.8345183366860	-2.80661010956647	2.02026577511643	H	0.24557679338462	-4.1171868418904	0.09002412623515
H	1.29050103925766	-2.86602905080653	1.86342492699187	H	0.69742291460109	-3.00729794138632	1.83841244783039	H	1.77213116266507	-2.45812617181486	1.02912486634288
H	-2.33394056898078	-2.82624253728669	3.38945170342686	H	-2.71907858050266	-2.48371170342686	3.29451074454516	H	-1.71050880533752	-3.20617023737509	2.32196771740646
H	-2.46602130939651	-1.08564708498956	3.20781138214872	H	-2.58745127940835	-0.69978777851780	3.56705256831794	H	-2.05508803749170	-1.56218297184612	2.96057860504352
H	-2.33277216807778	-1.77896013345941	5.5256327345941	H	-2.76401790343577	-1.43848314524905	1.93712306295556	H	-1.97243870324067	-1.8849910004945	1.18684862410392
C	-2.4586337688895	0.6866261294043	-0.6711838575531	C	-2.68709515035175	0.30857603790808	-0.68812603326976	C	-2.53493908781532	0.1001697029751	-0.29057803111279
C	-1.58360061867870	2.45230131400318	0.0244512412173	C	-1.35113171269070	2.33884794292437	0.10196655067905	C	-0.9835992003763	2.88976393429946	-0.3662532629831
C	-0.66903303911795	1.37411453528783	-2.22735250981883	C	-0.94022159411176	1.27584321763515	-2.22690574230094	C	-1.04054708075440	0.07187763279638	-2.04137387917093
H	-1.79418464675834	3.16152653665770	-0.41487157465046	H	-2.09663154608237	-0.10282534744625	-0.43372763061916	H	-1.7733858677537	3.64384860970322	1.13864862410392
H	-1.32921072321562	2.9331830232376	1.08831519840347	H	-1.63014509356095	2.18001321269590	1.080701112105065	H	-1.05046837455277	2.61325473633636	1.41817265591688
H	-0.06025939763610	2.92370153014291	-0.00932627856545	H	-0.36275094109102	0.84190407481499	-0.00334780440090	H	-0.001		

Table S10. Cartesian coordinates of optimized structures for the conformers for complexes 3 and 4.

Rotamer 3a			Rotamer 3b			Complex 4					
C	-6.33107522894301	3.50007698796155	15.64773067436062	C	6.338992	-0.254311	0.742028	C	-6.42796653075706	3.47592382874835	15.70169983316478
C	-7.36571955499739	2.62467469381122	16.37478233761425	C	6.107669	-1.329154	-0.331867	C	-7.37873764173179	2.37599168037471	16.20558132641314
H	-7.63463762501520	1.72912542232256	15.77719114922601	C	6.597274	-0.837136	-1.699573	H	-7.41581201268211	1.51204371028168	15.51031012897350
H	-8.29337881973665	3.19084585341409	16.59538907353781	P	4.239874	-1.719160	-0.298722	H	-8.40822465286193	2.76114042010262	16.35272751039749
H	-6.93674266613469	2.6986970708451	17.33541505371184	Au	3.423266	-1.683265	2.238508	H	-7.00779714308950	2.00257046354359	17.18284155551935
C	-5.06129823203928	2.70088053833774	15.32433505630834	C	6.628663	-3.578764	3.046117	C	-5.02118988631563	2.91905783770998	15.44646552182490
H	-4.31371474794868	3.20273563475989	14.78923093915114	P	3.341177	0.054306	-1.359313	H	-4.32025197369473	3.71034314925223	15.11431191134459
H	-5.28478153858495	1.80692394305868	14.70861338399203	P	2.257480	-1.536830	-2.515810	H	-5.03544079409916	1.66136700554322	14.69557262743029
H	-4.59764803042370	2.3582752089365	16.27224512451166	C	2.352920	-0.994691	-4.326007	H	-4.61945538894342	2.50197192729533	16.39328150111806
C	-5.99509847169400	4.76246494594249	16.45645128778044	C	1.856244	-2.188024	-5.159418	C	-6.39913468094252	5.66136700554322	16.7554415694564
H	-5.47432956570830	4.47287739264662	17.39283997286771	P	1.844872	1.053843	0.040571	H	-5.91747567745796	4.02824296808070	17.62437896792050
H	-6.91044973063888	5.32335266331173	16.72943990548784	P	2.123446	0.184831	2.092965	H	-7.42150015364190	5.50258321882820	16.9005828544123
H	-5.32321846061475	5.44350137813750	15.89242062116032	C	2.686692	1.629739	3.176636	H	-5.80382522885210	5.50213563167957	16.26892888167881
C	-9.07470501418903	2.09026903431690	12.17169691640829	C	1.768761	2.844223	2.977232	C	-9.03214805032031	2.06523449377938	12.14917255564781
C	-9.72916544213987	1.44816891320725	13.40996747077865	P	-0.088600	0.006186	2.327342	C	-9.71749046507995	1.38031249422799	13.34682481047994
H	-10.77716164242974	1.16046008453078	13.16477189567864	C	-0.423336	-1.788293	2.916898	H	-10.74501999744556	1.0782462147110	13.05453719300597
H	-9.77114295362328	2.16706046770451	14.25488600931413	C	0.140658	-2.886152	2.014674	H	-9.80676299379206	2.07560388448869	14.20706381749457
H	-9.19124025271880	0.53685066241825	13.74330576708405	P	-0.041088	-0.180166	0.074966	H	-9.17380535608414	0.47187121625514	13.6777053468916
C	-8.98994175170751	1.10154365553713	10.99326202844644	C	-1.369455	1.067795	-0.492126	C	-8.88914856267294	1.10656633388321	10.95145177708051
H	-10.01534864514319	0.79618464377731	10.69620995634801	C	-1.191344	1.219313	-2.011268	H	-9.89833495447412	0.78617108704305	10.61751905733177
H	-8.42178649016121	0.18483915109023	11.25388026637861	P	3.941272	-2.963307	-2.163003	H	-8.30912387644520	0.19643969045137	11.20861032973318
H	-8.51606867230839	1.57499361540641	10.10782664879887	C	3.149004	-4.577011	-1.507908	H	-8.40096731923909	1.60997501078479	11.09063391484361
C	-9.83963952512420	3.34448474077072	11.75779931456639	C	2.283704	-4.417245	-0.257478	C	-9.80712461272098	3.31510152372628	11.73958575234132
H	-9.39176717091856	3.82131855537694	10.86204665170569	C	1.415479	0.214366	-4.462067	H	-9.34238293441362	3.82028711556501	10.86784188707741
H	-9.87133778289500	4.08574900170970	11.57489709961129	C	3.793865	-0.621860	-4.702840	H	-9.87862774214942	4.04784625974187	12.57083616404545
H	-10.8873953676249	3.07477914272770	11.50862876378948	C	6.791226	-2.653842	0.053250	H	-10.841101291932297	3.03151543088848	11.45204594502835
-4.80457363941632	2.80131662323667	10.4853234211759	C	2.326843	-5.136769	-2.681648	C	-4.73296889979166	2.82427725357600	10.56768379205212	
C	-5.3579263245864	1.7716802040276	9.48587828263437	C	4.355869	-5.484526	-1.195549	C	-5.23100686761109	1.78431711490779	9.55030596637814
H	-5.96251852375095	2.25680749118584	8.69222266578572	C	-2.710961	3.900852	-0.156447	H	-5.82273861551248	2.8681614449718	8.7392918893321
H	-5.98678007981312	1.0105045098902	9.99182750881007	C	-1.255928	2.431062	0.210503	H	-5.85592891972868	1.00687044036890	10.03650303646009
H	-4.51376849041819	1.23961499054813	8.99923389785679	C	0.176210	-1.874245	4.331021	H	-4.35985180235799	1.27450446048305	9.08868439656009
C	-3.99468201063507	2.12006464836830	11.59680252302976	C	-1.961553	-1.886744	2.959356	C	-3.94218879073213	2.16350981939558	11.70529782749996
H	-3.11539202603752	1.6124144209181	11.14830458013496	C	6.628748	1.125940	4.628514	H	-3.02902157612003	4.0398234126045	11.02878720618206
H	-4.59373775854449	1.35686296735949	12.13260428612055	C	4.127257	1.949868	2.750457	H	-4.53381684822947	1.37501430246745	11.21649178191229
H	-3.61833744355258	2.85992903297149	12.33620305048849	H	7.418688	0.000278	1.971838	C	-3.61900489461422	2.9595402038219	12.46388730909200
C	-3.96663097502051	3.88297085463685	9.77976937573586	H	5.784411	0.678110	0.507610	C	-3.90271730748201	3.92983708027900	9.89015750109097
H	-3.0712852537693	4.31339674305960	9.32171451836288	H	6.022419	-0.606486	1.745180	H	-2.98719378291693	3.8434041063475	9.49877404519022
H	-3.61266813521327	4.6541707205572	10.49601842299945	H	7.885359	-2.490431	0.154919	H	-3.58027328939887	4.70103815791945	10.02120545173151
H	-4.54092210917630	4.39053575155746	8.97797299947778	H	6.404543	-3.051455	1.014783	H	-4.46870967068590	4.43203117373586	9.07911256556059
C	-4.04420866356082	9.32399009251178	12.46063063655732	H	6.643445	-3.424155	-0.371688	C	-4.07089649098387	9.34312983491773	12.26721066077433
C	-2.98887424666734	8.3909462811983	12.02568037041519	H	7.872825	-0.625291	-1.657776	C	-2.99810705298255	8.37893588142771	11.71133603742090
C	-3.27595882059397	7.55975429795606	11.39191301937957	H	6.405412	-1.586637	-2.493674	H	-3.44663076994173	7.5847906531739	11.09770133142821
H	-2.7243958628871	7.95009818665306	12.89724055925486	H	6.104417	1.012745	-1.990271	H	-2.43399685903932	7.89331153439589	12.55356616779980
C	-2.1550582886673	8.96574586000556	11.42415403743204	H	2.973194	1.930897	5.531139	H	-2.27820006349729	8.942270353621936	11.0181229007255
C	-3.52952246143066	10.44734164536310	13.37801317371601	H	1.593962	0.854669	4.922139	C	-3.46689326795442	10.40486374794513	13.20164825511485
H	-3.01547006306212	10.04495147866231	14.27536234559276	H	3.278847	0.239944	4.777159	H	-2.90954619937504	9.93860208594232	14.0397627267865
H	-4.36022703626137	11.10192945015663	13.71623954759963	H	4.507770	2.807679	3.343001	C	-4.25084750663533	11.06611024194861	13.625357573142594
C	-2.80568523753971	11.08474974407730	12.82630220683131	H	4.800828	1.036836	2.920195	H	-2.76115312771172	11.04227067454668	11.04227067454668
C	-4.75600680424811	9.91862563377720	11.23600916493032	H	4.170936	2.235680	1.677932	C	-4.85732265313108	9.98815792884952	11.11706977846677
H	-4.0410503928844	10.51776211960138	10.63836504284659	H	2.109730	3.671673	3.634324	H	-4.16285404577203	10.50875668806015	10.48614134780146
H	-5.59699113473641	10.56988398621728	11.53481004807202	H	1.794622	3.207082	1.929902	H	-5.64469963636452	10.67079354584345	11.49437853343010
H	-5.15661514366628	9.11509595949685	10.57548003616175	H	0.718058	2.607759	3.240353	H	-5.33345379241031	9.22519484971616	11.47712483030990
C	-7.44461021294042	9.45382551622819	15.56303322011517	H	-0.088501	-2.851697	4.780766	C	-7.34134568040028	9.4793007295065	15.28164861497369
C	-7.12168879513008	8.13994045531735	16.27901231470832	H	1.284618	-1.809700	3.400922	C	-7.24173861088586	8.11959917711132	16.05003691564811
H	-6.05147472533217	7.87079528517575	16.18103737539233	H	-0.208323	-1.073113	4.995965	H	-6.23677165848065	6.66668142788239	15.45438945770349
H	-7.72683116405463	7.30344428749467	15.87059146972799	H	-0.041537	-3.879114	2.718109	H	-7.99614758110625	7.40184056278027	16.90010426295510
H	-7.3602662814106	8.22825240818024	17.36082054209215	H	-0.337541	-2.884292	1.015625	H	-7.41487075852832	8.24827253913551	17.36963351522634
C	-5.66623262143266	10.59419748508114	16.1109526787938	H	1.238081	-2.782640	1.879098	C	-6.2570602849764	10.43523984755884	16.10077427602191
H	-6.74756688166567	10.71897564201134	17.20019006994272	H	-2.253746	-2.866723	3.393149	H	-6.40082757198035	10.59861034918815	17.19297384202363
H	-6.78730652275099	11.56116593536660	15.61370681523833	H	-2.415979	-1.088163	3.581372	H	-6.29747135469924	11.42383364890082	16.00210043197187
H	-5.48615562625826	10.3728689076838	15.97687898438126	H	-2.395723	-1.834707	1.939790	H	-5.24105923661838	10.00840698159774	15.96497950455924
C	-8.94054560474948	9.78138064236856	15.71263540852162	H	-2.069411	3.097857	-0.157837	H	-8.75548747194127	10.0577446449014	15.77342266028000
H	-9.57551128098108	8.94008597711810	15.36462388880481	H	-1.352116	2.336619	1.301492	H	-9.53121691145134	10.32689752611751	15.51536955814693
H	-9.22464473981591	10.69839351497342	15.15610157212300	H	-0.287675	2.922981	-0.023179	H	-8.88572171506859	11.04989756193912	15.24951842572611
H	-9.17622111581049	9.94921778914568	16.78566706663674	H	-3.546483						

Table S11. Cartesian coordinates of the optimized transition state structures for the Y-shape coordination flip for complexes **2** and **4**.

Complex 2 (Rotamer 2a)			Complex 2 (Rotamer 2b)			Complex 4					
C	6.358460	2.274347	9.994087	C	9.949406	7.361241	12.694746	C	0.602710	-3.404761	-4.343435
C	5.727252	3.389226	10.850759	C	10.092324	5.994420	12.002677	C	-0.418327	-2.420298	-3.743070
C	6.413599	4.726358	10.566111	C	11.273983	5.204547	12.574062	C	0.045797	-0.979691	-3.943947
P	5.912992	2.767100	12.655164	P	10.571517	6.382712	10.187771	P	-0.609459	-2.994027	-1.828196
P	7.917908	3.570604	13.325634	P	8.599884	7.101716	9.341949	P	1.285793	-2.432515	-0.901163
C	8.836898	2.000360	13.922739	P	8.869530	9.263646	8.700098	C	2.042458	-3.982909	-0.141436
C	7.974714	1.053539	14.766003	P	10.318105	9.279078	6.981063	C	0.957493	-4.799023	0.574091
P	5.067556	4.425203	13.900543	P	11.632169	10.676625	8.134849	P	-1.499567	-1.249800	-0.811830
C	3.640095	3.629066	14.916206	C	13.338429	9.815839	8.112320	C	-3.065529	-1.974228	-0.017085
C	4.028158	2.347086	15.544448	C	14.207498	10.645247	9.075715	C	-2.752004	-3.175547	0.882469
Au	4.833044	6.664762	13.094372	C	8.783978	5.193732	12.153983	Au	-1.712817	1.010776	-1.562530
P	6.703976	7.686654	14.178858	P	10.066901	4.438486	9.127770	P	0.128969	2.002575	-0.426216
P	7.817706	6.207220	15.456146	C	11.721338	3.818013	8.369690	P	1.385292	0.459920	0.678778
P	6.877028	4.217383	15.220565	C	11.356145	3.115459	7.051377	P	0.151720	-1.413096	0.729219
C	4.214677	3.493551	10.592080	P	10.639623	9.841418	9.984324	C	-1.819867	-2.618136	-4.351195
P	6.504843	8.960823	16.008998	C	10.004541	11.437089	10.837700	C	0.040929	3.287035	1.420642
C	4.666270	9.505621	16.011436	C	11.259170	12.077829	11.459234	C	-1.808032	3.668227	1.729990
C	3.659947	8.354005	15.979339	P	9.243964	5.848961	7.601620	C	-2.783014	2.520881	1.476845
C	8.033199	8.484004	13.082505	C	7.635034	5.146596	6.879143	C	1.309987	2.952845	-1.572271
C	3.895853	7.424997	12.029905	C	8.042182	4.196299	5.740240	C	1.385786	2.159432	-2.883146
P	6.512363	6.992438	17.121599	Au	10.692480	7.100234	6.118139	P	0.487624	1.397943	2.510660
C	7.804764	7.276133	18.505567	C	9.030151	10.962472	11.930386	C	1.921358	1.740043	3.689088
C	8.128425	5.864874	19.031368	C	9.307187	12.427135	9.897807	C	2.416979	0.370332	4.191723
C	7.086946	8.114797	19.577920	C	12.757201	4.913742	8.120914	C	1.354102	2.590965	4.840465
P	9.080941	7.983763	18.035647	C	12.554070	2.801655	9.399115	C	3.033696	2.484467	2.933200
C	7.366546	9.706311	12.429388	C	9.503465	10.364877	5.845958	C	0.622084	4.310601	-1.803102
C	9.264663	8.895206	13.899970	C	10.607858	10.633943	4.609098	C	2.694304	3.130300	-0.936966
C	9.321596	1.293465	12.644547	C	8.363723	9.534781	5.038283	C	2.686653	-4.778936	-1.299118
C	10.024185	2.532633	14.746874	C	8.964947	11.679720	6.223606	C	3.110670	-3.480526	0.847387
C	2.463747	3.527685	13.939384	C	13.302125	8.350316	8.547010	C	-3.940326	-2.388893	-1.214875
C	3.319627	4.735345	15.997662	C	13.841907	9.930521	6.662803	C	-3.732247	-0.836328	0.767431
C	4.490302	10.410120	14.779962	C	6.879186	6.364899	6.325132	C	-2.114435	4.841648	0.777791
C	4.509935	10.311786	17.315770	C	6.807203	4.418956	7.945016	C	-1.862173	4.110294	3.204448
H	7.524603	7.183178	11.391046	H	9.098098	7.937148	12.276011	H	0.393759	2.089313	-3.372177
H	8.749689	6.480956	12.496808	H	10.864791	7.976205	12.578105	H	1.771201	1.134967	-2.711896
H	9.207134	7.809656	11.376446	H	9.757248	7.220808	13.780381	H	2.088574	2.660410	-3.580133
H	8.071749	10.173592	11.709725	H	8.573896	5.017484	13.231494	H	1.218441	4.893759	-2.535574
H	7.097694	10.473229	13.183978	H	8.853288	4.208714	11.649858	H	0.564141	4.908700	-0.870176
H	6.447484	9.419637	11.878081	H	7.919799	5.739187	11.722280	H	-0.400067	4.187762	-2.216411
H	9.751362	8.015600	14.368900	H	11.522112	1.991797	9.597936	H	3.207959	2.158489	-0.796717
H	8.998473	9.612445	14.702616	H	12.498893	3.292090	10.364287	H	2.631426	3.655584	0.037194
H	10.006948	9.382962	13.232235	H	13.187311	2.335732	9.016393	H	3.329991	3.741059	-1.611939
H	8.611271	5.243430	18.247305	H	11.120325	5.015387	13.658305	H	2.647552	-0.327860	3.341051
H	7.216686	5.334898	19.376582	H	12.229975	5.756151	12.454916	H	1.668346	-0.124504	4.872008
H	8.833083	9.305333	19.888041	H	11.378072	4.217480	12.076761	H	3.358897	0.528549	4.778663
H	7.744393	8.240327	20.464965	H	7.590679	9.296045	5.797260	H	2.155961	2.784064	5.583786
H	6.145115	7.635092	19.916400	H	8.741937	8.583417	4.609452	H	0.523479	2.070801	5.362274
H	6.844106	9.123343	19.197903	H	7.873544	10.109159	4.224207	H	0.986992	3.575529	4.479942
H	9.781391	8.105314	18.890270	H	8.510206	10.282641	5.408765	H	3.873039	2.682838	3.633932
H	8.858372	8.990559	17.627564	H	9.773766	12.281235	6.684122	H	2.682861	3.461067	2.535396
H	9.602083	7.400672	17.248833	H	8.184953	11.496428	6.989709	H	3.433551	1.878546	2.090580
H	2.624174	8.756320	15.972425	H	13.854458	10.981653	6.306850	H	-3.821208	2.864558	1.666624
H	3.770833	7.688285	16.857531	H	13.209107	9.331378	5.974541	H	-2.589059	1.658279	2.147609
H	3.784211	7.736472	15.063313	H	14.877402	9.533861	6.593439	H	-2.733730	2.161292	0.428081
H	4.579732	8.828373	13.839255	H	10.196244	11.246957	3.779389	H	-2.064000	4.520758	-0.283110
H	5.233338	11.234516	14.761761	H	11.003878	9.689964	4.182237	H	-1.413084	5.691004	0.923911
H	3.477354	10.866213	14.794036	H	11.452516	11.198409	5.055957	H	-3.142027	5.208151	0.967969
H	3.480329	10.724392	17.380409	H	6.636580	7.095204	7.126705	H	-2.882663	4.480211	3.438077
H	5.223638	11.159853	17.365812	H	7.471864	6.884094	5.546985	H	-1.144954	4.928134	3.423012
H	4.668138	9.671180	18.208612	H	5.921902	6.037119	5.867784	H	-1.654111	3.261928	3.890427
H	3.161808	1.938321	16.107521	H	5.871738	4.029745	7.488520	H	-3.703221	-3.617188	1.248164
H	4.874239	2.459808	16.252446	H	7.364490	3.563558	8.376953	H	-2.171357	-2.875976	1.777542
H	4.318308	1.606508	14.771934	H	6.528936	5.101057	8.774157	H	-2.199071	-3.965123	0.334484
H	2.691797	2.779905	13.152770	H	13.680201	4.469720	7.689574	H	-3.485236	-3.221327	-1.790431
H	2.208356	4.488165	13.446475	H	12.379994	5.663388	7.394847	H	-4.126768	-1.536768	-1.900563
H	1.568565	3.172588	14.492781	H	13.028104	5.445979	9.053792	H	-4.921385	-2.744007	-0.835863
H	2.519341	4.349789	16.664031	H	10.560663	2.354024	7.192988	H	-4.659063	-1.212571	1.247699
H	2.965436	5.682276	15.546583	H	11.024596	3.846322	6.285463	H	-4.000293	0.010644	0.104212
H	4.206989	4.966871	16.624890	H	12.251449	2.596677	6.646370	H	-3.075609	-0.460805	1.576873
H	8.576613	0.173374	15.080259	H	7.133137	3.941000	5.210334	H	1.423790	-5.692346	1.039862
H	7.100100	0.685067	14.192575	H	8.698835	4.703098	5.003304	H	0.177639	-5.150541	-0.130854
H	7.599735	1.553906	15.682352	H	8.577280	3.306447	6.128547	H	0.473847	-4.216049	1.383499
H	9.928905	1.966619	12.004392	H	8.163101	10.423221	11.492279	H	3.432134	-4.183153	-1.840821
H	8.465629	0.920146	12.043860	H	9.524583	10.291285	12.660409	H	1.900801	-5.114997	-2.025895
H	9.949700	0.416132	12.910262	H	8.631157	11.837517	12.486863	H	3.164644	-5.684747	-0.892048
H	9.675597	3.078563	15.649442	H	9.988120	13.326643	10.468026	H	2.666808	-2.839747	1.637237
H	10.661594	3.220805	14.154018	H	9.983569	12.759753	9.085121	H	3.907032	-2.906313	0.329878
H	10.657916	1.687212	15.090934	H	8.403732	11.978578	9.437484	H	3.583711	-4.353900	1.343120
H	6.244543	5.021776	9.508366	H	14.323134	7.913670	8.501654	H	0.091978	-0.750847	-5.029179
H	7.505278	4.672993	10.745279	H	12.910921	8.229027	9.576315	H	1.058175	-0.812790	-3.522714
H	5.999729	5.533362	11.208689	H	12.652157	7.752428	7.878095	H	-0.649191	-0.252123	-3.476033
H	6.242629	2.518784	8.916407	H	14.252754	11.713496	8.778721	H	0.679209	-3.231300	-5.437460
H	5.878848	1.290628	10.176795	H	13.821449	10.588145	10.115370	H	0.308692	-4.462864	-4.185363
H	7.445331	2.174165	10.197172	H	15.244587	10.246524	9.083678	H	1.614326	-3.248005	-3.913237
H	3.760201	4.305797	11.196410	H	10.972106	12.968978	12.057769	H	-2.552830	-1.911855	-3.910172
H	3.688542	2.542871	10.818876	H	11.793138	11.373491	12.130496	H	-2.195267	-3.652540	-4.211549
H	4.035594	3.734690	9.522504	H	11.968662	12.417256	10.675526	H	-1.776140	-2.413284	-5.441512
Cl	3.190412	7.621972	11.608221	Cl	11.843296	6.152852	4.22232				

Table S12. Cartesian coordinates of the optimized structures for the monodentate monometallic complexes according to table Table S8.

Coordination on α -P			Coordination on β -P			Coordination on γ -P					
C	2.00042778068444	1.12661072523987	5.04619997975187	P	0.12124176635390	0.29063998556102	-0.29781244801520	C	2.34851937857476	1.19876404175248	4.86879554375585
C	3.22505309989598	1.88146986840759	3.74597554104540	P	0.11658960115917	0.04828372902777	1.95900161316365	C	2.46253689067708	1.93554476351502	3.52215226280180
C	3.79198352774302	2.35957247255969	3.73068937899146	P	2.29667688062546	0.49970352933338	1.85939193899760	C	3.87288039964281	2.51955085660203	3.33042665119986
C	1.37931333483994	3.07027028102772	3.55042471270745	P	2.06156788338641	1.45024421695717	-0.15724104150251	C	1.40510490906020	3.04211637203263	3.42655653212195
P	2.28594550120292	0.58169467493019	2.56093951740178	P	3.19896470127344	-0.20059486227232	-1.15484924518926	P	2.28223292931955	0.57299466474533	2.18642423104152
P	1.93147704134931	1.82252063263381	0.47242753549241	P	2.31231003384638	0.32006920925403	-3.16470491032876	P	1.80248677151042	1.79094337456360	0.34949842792248
P	3.35639704460710	5.40283723147559	-0.73770319839409	P	4.42252912756635	0.80077241208637	-3.81712937511381	P	3.25279450305866	0.74338241936627	-0.95035306730011
P	3.07392741373721	1.86009380515578	-2.55100083429032	P	4.96242105180046	1.12868015372241	-1.65328475910913	P	3.14673308205973	2.05886023404627	-2.75494691353731
C	3.02966358586908	0.60728395661912	-3.99170797604176	C	6.48667209467900	0.00633128735490	-1.38779034035787	C	3.16272544777499	0.88905736678271	-4.26467680025744
C	1.6186765982238	-0.00302002018187	-3.93816176043562	C	2.01307293651444	-0.40116387551446	-3.94755388860334	C	1.77380942261608	0.22465416063640	-4.27390274277707
P	0.46237139908050	-0.00587693253689	2.07119611288286	C	4.21704728464898	2.60363030607554	-0.46359969781911	P	0.06725349086078	1.1359090069877	2.24242296123338
P	1.11842128593635	-1.89554026401681	2.0740486289556	C	-1.17502545406685	1.66971263207521	-0.56129536150561	C	0.02493291846130	-1.78453807752919	2.47536468319051
P	-1.287857377025925	-2.30809897806745	1.68044437701711	C	-0.00595415842489	-1.85130913413754	2.18206106825839	C	-1.4323360996073	-2.19412371260012	2.19859617749602
P	0.16103177786647	6.0658129088621	-0.07891203115664	C	2.7105980450706	1.82566534117913	3.12877240836275	P	0.1063879366319	0.35268172717057	-0.00244480783626
C	-1.20789676453780	1.92331511440493	-1.2920665426967	Au	3.80129262484170	-1.20981541374753	1.8239837108913	C	-1.33281780583710	1.5653021996262	-0.36710169570055
C	-0.98964133292927	2.16577095984056	-0.73900674375868	C	5.348703507578912	-2.96619134243569	1.87599041348556	C	-1.0686066538677	0.04482991289083	-1.80626415600203
C	-2.54824810701405	1.18167850364298	-0.15239371358333	C	1.83193127258413	-1.13152985977687	-5.4522266565572	C	-2.61821377538130	0.7206361703838	-0.28897049300102
P	-1.13049472887874	3.0941608865327	0.68796018991557	C	3.1617962648276	-2.38948558554576	-3.1276747843325	C	-1.40990985091609	2.75228303575889	0.59830728363634
C	1.19672566737967	-2.43931692845007	1.12424365438112	C	0.71386923851116	-0.93457874118815	-3.32153214448903	C	0.97744736436430	-2.54649853949312	1.55079505443009
C	0.38957743520309	-2.33092345309586	3.52386825202236	H	2.7615957103080	-1.27428123723498	-5.90316349041248	C	0.3962699327818	0.3962699327818	3.94936047177766
P	5.2818762592077	2.39171996315904	-2.36024897233079	H	1.58950952820779	-0.27869725251232	-5.98051029480394	P	5.3358688458617	2.48583214340008	-2.35739577035555
C	5.18730214242584	4.30321221299266	-2.26012188675122	H	1.00805198260340	-0.41315246304959	-5.64255863164036	C	5.27731372794965	4.37614266150744	-0.02748306268329
C	4.9266645602943	4.77997817294617	-3.69994576773206	H	2.92475309346772	-3.36631731508075	-4.1886994972798	C	5.0821936945072	5.00646180780057	-0.34335971806853
P	5.18634441328101	1.81602604662899	-0.10282520336484	H	4.1108538685158	-2.01792135315109	-4.488963542242	P	5.12153935006062	1.72366607138859	-0.233215407625228
C	6.53246929147485	0.52947723379176	0.02889147755541	H	3.32473403821398	-2.57122490537266	-2.63038384206665	C	6.42578839084737	0.33249765639764	-0.10199391298542
C	6.28800977627013	-0.1585404903491	0.01469464010009	H	0.45698787254710	-2.91908293636462	-3.76861585108774	C	6.01188965871705	-0.51213546788664	1.1590775792404
C	8.4227048178256	1.33734668929711	0.03646249898976	H	0.82030591812780	-2.07037293131062	-2.22572052090085	C	7.73781757544311	1.09530926980402	0.16928041194062
C	6.56513030880148	-0.53748882275980	-1.07124569796407	H	-0.13750040388890	-1.2439376510618	-3.49036426305935	C	6.57028830439355	-0.54650155935461	-1.3476009095351
C	3.22810588405073	1.44138458118233	-5.26881862534326	C	2.65349804360063	1.14583701694949	4.50816702102599	C	3.33102237438341	1.84246698461883	-5.46449210527989
C	4.0502895733551	-4.49051677364615	-3.89601430098673	C	4.135252200118	2.29564150498771	2.78352195700008	C	4.27219024214505	-0.16572006900502	-4.2659043472213
C	4.10031371038206	0.48042155059647	-1.31543725391693	C	1.70815982863008	2.98191010393603	0.30299835422462	C	1.60484814791951	4.82290294185052	-1.0980875381918
C	5.8079848408004	4.73933266408838	-1.77383238348148	H	9.2321627092627	1.88325013847392	5.29564346335948	C	6.65601252232411	4.73930243929158	-1.46932124690771
H	1.22884201793228	1.80575199714166	5.91569733653075	H	1.633783930264720	0.77157452956996	4.73103638628057	H	5.25219992001329	1.9103457847066	5.70415249048251
H	0.94736215251972	0.77176912499246	5.054161129462266	H	3.36268979842538	0.29757248860735	4.5715006652184	H	1.33640485465464	0.7640859902189	5.0065416045893
H	2.66719045617650	0.25207439115900	5.19403526486293	H	4.43731166211899	3.10747742968673	4.37785920126514	H	0.39405605419456	0.3812489630057	4.9563516910117
H	1.50786790702637	3.7905379469303	4.38662215493424	H	4.86842851581490	1.68680750016125	2.87551311111392	H	1.55471560146713	3.77837100864603	4.2459275648773
H	1.58642431691635	3.66573046911958	2.60202672893255	H	4.19284186483242	2.6943318966517	1.74845942244478	H	1.47618214246892	3.58911058262416	2.46391799559709
H	0.31657658394750	2.74880325763373	3.54744471181855	H	1.96496452870678	3.76119228551137	3.0773756690498	H	0.37952987399334	2.63007958311008	3.5169605018860
H	3.94159885075450	3.12911378852179	4.51760926301805	H	1.72667393335452	3.45341402548131	2.26401930032233	H	4.06699738404902	3.30078122567308	4.90654965034462
H	4.49627712569101	1.62531741894294	3.92598153397440	H	0.67201085306435	2.63934305328598	3.2563935651723	H	4.65650636581132	1.79404079629640	3.42821274958463
H	4.0620609253151	2.81340221176738	2.75307146305612	H	0.59105975916474	-2.16314017596637	3.36293704022568	H	3.98822688810916	1.64865094037571	2.3290249595345
H	0.38437379413145	-3.43934855477600	3.58633966471985	C	-1.52416924187709	-2.12251890959612	2.16395678208222	H	0.31452553929694	-3.10897730923902	4.18581847534377
H	1.38355309369665	-1.97395151994423	3.87170087981593	C	0.68331048513174	-2.67399857010850	1.03717373717545	H	1.44329308206910	-1.17967045731588	4.15483271219600
H	-0.38206118821023	-1.93958538372235	4.21810262895346	H	0.42457113713398	-1.23306618489856	3.81115959705699	H	-0.2712220429307	-1.47327485701568	6.42427352712130
H	1.16793564053662	-3.54917254129061	1.13381461366716	H	1.687807862699407	-0.9663653175751	5.80318933508442	H	0.92410422270224	-3.63797727036328	1.72622122960260
H	1.03518040806349	-2.10068150712994	0.08159130649747	H	1.02456010947071	-1.55191027705801	4.36527859794663	H	0.72121727814250	-2.38985852972269	0.48466975825242
H	2.21185662213922	-1.18959489899473	1.43432890194682	H	0.58619262958475	-3.76898605159336	1.3060790071589	H	0.29292919980268	-2.22476885973884	1.69334120414105
H	-1.36220342405191	-3.41556887894513	1.58553271016119	H	0.24557762938462	-2.47118684118904	0.09002412623515	H	-1.56008117275250	-3.28368802823364	2.37621534427179
H	-2.07335060712122	-1.92080081982558	2.28906950019916	H	1.772131162666507	-2.458112671181486	1.02912465634288	H	-2.14755161285639	-1.65435896110737	2.85445840003777
H	-1.49980568815443	-1.9359905623337	0.58350319217937	H	-1.1050889533752	-3.20617023773509	2.32197881754066	H	-1.71233384797071	-1.99806290327558	1.14221445390137
H	-3.38450505153687	1.87119750990216	-0.3953882872644	H	-2.05508843749170	-1.56218297184612	2.96087860550432	H	-3.49517396219182	1.3366685897146	-0.58251182711238
H	-2.61298317218521	0.31137463346017	-0.83778021213572	H	-1.97248070324067	-1.84845910004945	1.18561269035283	H	-2.57631412382330	-0.15802603506544	-0.96574163831543
H	-2.7041852525213	0.82318990360145	0.89773256021375	C	-2.5349380781532	1.0010619702975	-0.29005780311179	H	-2.79638345473167	0.38606495090274	0.74431404841485
H	-1.92859334622708	0.83501618705964	0.45075951454965	C	-0.9839909203763	2.88937302329496	0.3467252629831	H	-2.27007986046653	3.40307435804131	0.33145488538057
H	-1.28530727187284	2.76699213805349	1.73591678317120	C	-1.04054708075440	2.07187763279638	-2.04137387917093	H	-1.54481903405445	2.41239175565983	1.64509517488735
H	-0.15397416980765	3.62063629597946	0.62614903052700	H	-1.7733858687537	3.64384860970322	0.13864862103927	H	-0.48957115214126	3.37147032973453	0.5522261904213
H	-1.74012582406442	3.1982379571077	-1.9793898911478	H	-1.05046837455277	2.61325473638356	1.41817265591688	H	-1.8713568192547	2.74410375838933	-2.1303278071015
H	0.02052419379570	2.86032793318270	-1.86684819591213	H	-0.00121038630575	3.37437288942618	0.17114406276410	H	-0.10140619749569	2.58816906369070</	

Table S13. Cartesian coordinates of the optimized structures for the hypothetical tetra- and pentametallic complexes according to table Table S8.

Coordination on α -P			Coordination on β -P				
C	2.00042778068444	1.12661072523987	5.04619997975187	P	0.12124176635390	0.29063998556102	-0.29781244801520
C	3.23205330998598	1.88146986840759	3.74597554104540	P	0.11658960115917	0.04828377290277	1.95900161316365
C	3.79198352774302	2.35957247255969	3.73068937899146	P	2.29667688062546	0.49970352293338	1.85931938997602
C	1.37931333483904	3.07027028102772	3.55042471270745	P	2.061567568336641	1.45024421695717	-0.15724104150251
P	2.28594350122092	0.58169467493019	2.34509351740178	P	3.19986470127344	-0.20059486227292	-1.15484924518926
P	1.93147704134931	1.8225063235381	0.47242753549241	P	2.31231003384638	0.32006920925403	-3.16470491032876
P	3.35639704460710	0.54028372314759	-0.73770319083409	P	4.42252917256635	0.80077241208637	-3.81712937511381
P	3.07392741373721	1.86009380515578	-2.55100083429032	P	4.96242105180946	1.12860815372461	-1.65328475910913
C	3.02966358586908	0.60728395661912	-3.9917097604176	C	6.48667209467900	0.00633128735490	-1.38779034035787
C	1.61686765982238	-0.00303020020187	-3.93816176043562	C	2.01372093651444	-1.40116387553146	-3.94755388660334
P	0.14623713990850	-0.00587693253689	2.07119611288286	C	4.21704728464898	2.60363030607584	-4.43659996781911
C	0.11841285936365	-1.89554026401681	2.07404896289556	C	-1.17502545406685	1.66971262377521	-0.56129536150561
C	-1.28785737025925	-2.30809897806745	1.68044377017111	C	-0.00595415842489	-1.85130913413754	2.18206106825839
P	0.16103177786647	0.60658129088621	-0.07891203115664	C	2.71059850450706	1.82566534117913	3.12877240870085
C	-1.20789676453780	1.92331151440943	-0.29020665426967	Cu	3.80129262484170	-1.20981541374753	1.82398370108913
C	-0.98964133292927	2.41657095984056	-1.73590674375868	Cl	5.34870357578912	-2.96619134243560	1.87590941348556
C	-2.54824810701405	1.18167850364298	-0.15239371358333	C	1.83193127258413	-1.13152985977687	-5.4522566665572
C	-1.13049472887874	3.09941608865327	0.68796018991557	C	3.16179626448276	-2.38984558554576	-3.71276477843325
C	1.19672566737967	-2.43931692845007	1.12424365438112	C	0.71386923851116	-1.93457874118815	-3.32153214458903
C	0.38957743520309	-2.33092345309586	3.52368625022236	H	2.76159557103080	-0.72448123723498	-5.90316390411248
P	5.25818762592072	2.39171996315904	-2.36024897233079	H	1.58950952820779	-0.07869725251328	-5.98051029480394
C	5.18730214242584	4.30312212992662	-2.26012188675122	H	1.00805128610340	-0.41315246304959	-5.64425586316406
C	4.92666445602943	4.77997817294617	-3.69997456773206	H	2.92475309346772	-3.36631731508075	-4.18869949772798
P	5.13680441328101	1.81602604662899	-0.18228203336484	H	4.11085358685158	-0.01792135315109	-4.14886535422242
C	6.53246929147485	0.52947723379176	0.02888194755541	H	3.32473403821398	-2.57122490537266	-2.63038384520665
C	6.26800977627013	-0.11585404903419	1.40169464010009	H	0.45698778254710	-2.91908293636462	-3.78661585108777
C	7.84227048178256	1.33734668929711	0.03646248989676	H	0.82030591612780	-2.07037293131082	-2.22575205290085
C	6.56513030880148	-0.53748882275980	-1.07124567976407	H	-0.13750040383880	-1.24393075610618	-3.49036426305930
C	3.22801588405073	1.44138458118233	-5.26981862534326	C	2.65348084360063	1.14857830169449	4.50811670210259
C	4.09502895735551	-0.49051677364615	-3.89601430009873	C	4.13525252200118	2.29564150498771	2.78353219570008
C	4.40031371038206	4.83402155059647	-1.31543725319363	C	1.70615989263008	2.98191015936303	3.02998854524262
C	6.58079848048004	4.73933266408838	-1.77383238348148	H	2.92316270926627	1.88320513847392	5.29564346335948
H	2.12884201739228	1.80575199741166	5.91569733653075	H	1.63378390264720	0.77157452956996	4.73130638028533
H	0.94736215251972	0.77176912499246	5.05416129462266	H	3.36268979842538	0.2975248680735	4.5715006652184
H	2.66719045617650	0.25207439115900	5.19403526486293	H	4.43731166211899	3.10747744598652	3.47785920126514
H	1.50786790702637	3.79035799469303	4.38662215493424	H	4.86842851581490	1.46860750001625	2.87551311111392
H	1.58642331691635	3.60573046911958	2.60202672893255	H	4.19284186483242	2.69433189665197	1.74845942244478
H	0.31657658394750	2.74880325763373	3.54744817181855	H	1.96469645287068	3.76119228551137	3.77735756690498
H	3.94159885075450	3.12911378852139	4.51760926301805	H	1.72667393335542	3.45341092458131	2.02640190302233
H	4.49627712569101	1.52531471894294	3.92598153397440	H	0.67201085306435	2.63934935382598	3.23523935651723
H	4.06206069253151	2.81340221176738	2.75307146305612	C	0.59105975916474	-2.16314071596637	3.56493704022568
H	0.38437379413145	-3.43934855477600	3.58633996471985	C	-1.52416924187709	-2.1225189055961	2.16390567820822
H	1.3835530936965	-1.97935151994423	3.87170807981593	C	0.68331048513174	-2.67399857010850	1.08717373717545
H	-0.38206118821023	-1.93958538372235	4.21810262895346	H	0.42457113713398	-3.23336618489856	3.81115959705699
H	1.16793564053662	-3.54917254129061	1.13381461366716	H	1.68780626569407	-1.98600653175751	3.58031893350842
H	1.03151804806349	-2.10006150712994	0.08159130649747	H	0.12456010947071	-1.55191027750801	4.36522785979463
H	2.21185662213922	-2.11895948989473	1.43432890194682	H	0.56619262958475	-3.75689605155936	1.3060790071589
H	-1.36220342405191	-3.41556887894513	1.58553277101619	H	0.24557679338462	-2.47118684118904	0.09002412623515
H	-2.07335060712122	-1.92080008198258	2.28906950019916	H	1.77213116266507	-2.45812617181486	1.02912485634288
H	-1.49908568815443	-1.9359905623337	0.58350319217937	H	-1.17050885033752	-3.20617023773509	2.3219761754066
H	-3.38450508153687	1.87119750690216	-0.39538822872644	H	-2.05508483749170	-1.56218297184612	2.96057860550432
H	-2.61298317218521	0.31137463346017	-0.83778021213572	H	-1.97243870324087	-1.84845910004945	1.18512690352823
H	-2.70418252252123	0.82318990360145	0.88773256201375	C	-2.53493908761532	1.00106197029751	-0.29005780311179
H	-1.92859334622708	3.83501681570594	0.45075951454965	C	-0.98135909203763	2.88976393429946	0.3467253629831
H	-1.28530727187284	2.76699213805439	1.73591678731720	C	-1.04054708075440	2.07187763279638	-2.04137387917093
H	-0.15397416980765	3.62063269597946	0.62614903052700	H	-1.77338586877537	3.64384860970322	0.13864862410392
H	-1.74012582406442	3.19823795710770	-1.9793898911478	H	-0.05046837455275	2.61325473638356	1.41817265591688
H	0.02052419379570	2.86032793318270	-1.86684819591213	H	-1.004120138630575	3.37437288942618	0.17114406276410
H	-1.09858321306990	1.59285371274176	-2.47047188308070	H	-3.5804006955115	1.71637804756067	-0.50309837891187
H	8.70436489477507	0.66299146181953	0.22893124482349	H	-2.68780713086184	0.10550624652671	-0.92735367516776
H	8.01506643674759	1.83059932656061	-0.94341165272373	H	-6.2871761301860	0.69372884062157	0.77309534952083
H	7.83879082721246	2.11970752103810	0.82364067819266	H	-1.80327992534591	2.83966470166299	-2.29337803633507
H	7.39704472717479	-1.25097740030954	-0.88400876691953	H	-0.03910922694555	2.49980162809351	-2.25524155181491
H	5.62035282008822	-1.11802249981281	-1.09954605204679	H	-1.18383104312898	1.20504183707637	-2.71804633061690
H	6.72209543206542	-0.08196860452372	-2.06973309117017	C	5.64440258471041	3.16140639212711	-4.57368982532279
H	7.05095765685369	-0.87198274602987	1.62570306159560	C	3.53117933193199	2.49141300840383	-5.80957853661116
H	6.27491185290385	0.6380110819479	2.21547042278950	C	3.38539473611789	3.48673837495242	-3.49574854995531
H	5.28332311074046	-0.62988158363592	1.42334539593073	H	3.4180126081044	3.49981319895012	-6.26305185163353
H	6.64353691118108	5.84841619366556	-1.74162808325416	H	4.11668406830675	1.86347818883208	-6.51274450990750
H	6.78250116217951	4.36439615264954	-0.74796030359955	H	2.51516828608193	2.05253073234255	-5.71541240887628
H	7.38561573220555	4.36998705332721	-2.44278249014821	H	5.61220699558065	4.19453613534734	-4.98211964506127
H	4.12059232920569	5.94567813069622	-1.30162139547243	H	6.15325995426719	3.20829291706240	-3.58761022072608
H	3.09015872639265	4.51181957270077	-1.63979035471018	H	6.26513041212526	2.54285127993828	-5.25460662479921
H	4.25414069162210	4.47371750089842	-0.27793115108776	H	1.97243870324087	4.51608791816299	-3.90937430292910
H	4.92213703744629	5.89066047348086	-3.7303879490954	H	2.35721754091251	3.08817443404849	-3.37481151322870
H	5.70484871718235	4.41571715054406	-4.40235118936083	H	3.84667103549991	3.55274159042078	-2.48918232237774
H	3.93777022213432	4.43019365807773	-4.06529732664084	C	6.85069610723574	0.18537945587591	0.09785461961547
H	3.14213256283630	0.79051188392805	-6.16648637849338	C	6.23296111425539	-1.46958017096006	-1.71380162356043
H	2.46861464539262	2.24587373444444	-5.35835207344073	C	7.59228324959602	0.57958824406854	-2.77193091951781
H	4.2348424326499	1.90955283305558	-5.29116038322944	H	7.7263263238259	-0.44704659239711	0.35732485090047
H	4.01390630738501	-1.17476609029588	-4.76863526594232	H	7.09876754574842	1.24095682096062	0.33195919780056
H	5.11705059773034	-0.06096381279690	-3.89075184061474	H	6.01445968355760	-0.12991896625500	0.75809757852759
H	3.96832056868497	-1.0982018469810	-2.9766715320145	H	7.16103647538755	-2.05837717514173	-1.54688663159338
H	1.48767591990587	-0.73992675791332	-4.75987542933097	H	5.44810581047420	-1.90075414240716	-1.06018582567329
H	1.44125205497790	-0.53161090010571	-2.9768470013554	H	5.92942603850826	-1.60051849003996	-2.77183954022912
H	0.83255218407647	0.77375834672027	-0.40608393622269	H	8.54256829062648	0.03050096484809	-2.11768111526217
Au	-1.42954355211388	0.84006390050312	3.46713378281178	H	7.33105998158944	0.47182166313022	

7. References

- [1] M. Baudler, C. Gruner, G. Fürstenberg, B. Kloth, F. Saykowski, U. Özer, *Z. Anorg. Allg. Chem.* **1978**, *446*, 169–176.
- [2] G. R. Bourret, P. J. G. Goulet, R. B. Lennox, *Chem. Mater.* **2011**, *23*, 4954–4959.
- [3] R. K. Harris, E. D. Becker, S. M. Cabral De Menezes, R. J. Goodfellow, P. Granger, *Concepts Magn. Reson.* **2002**, *14*, 326–346.
- [4] M. Baudler, J. Hellmann, P. Bachmann, K.-F. Tebbe, R. Fröhlich, M. Fehér, *Angew. Chem.* **1981**, *93*, 415–417; *Angew. Chem. Int. Ed. Engl.*, **1981**, *20*, 406–408.
- [5] M. Baudler, J. Germeshausen, *Angew. Chem.* **1987**, *99*, 372–373; *Angew. Chem. Int. Ed. Engl.*, **1987**, *26*, 348–349.
- [6] Agilent Technologies Ltd., *CrysAlis-Pro Software package*, **2014**.
- [7] A. Altomare, G. Cascarano, C. Giacovazzo, A. Guagliardi, M. C. Burla, G. Polidori, M. Camalli, *J. Appl. Crystallogr.* **1994**, *27*, 435–436.
- [8] G. M. Sheldrick, *Acta Cryst. A* **2015**, *71*, 3–8.
- [9] L. J. Farrugia, *J. Appl. Crystallogr.* **2012**, *45*, 849–854.
- [10] A. L. Spek, *Acta Cryst. C* **2015**, *71*, 9–18.
- [11] a) M. Fehér, R. Fröhlich, K.-F. Tebbe, *Z. Kristallogr. Kristallgeom. Kristallphys. Kristallchem.* **1982**, 241–253; b) M. Fehér, R. Fröhlich, K.-F. Tebbe, *Z. Kristallogr.* **1981**, 335–337.
- [12] STOE Cie GmbH, *WinXPow*, **2014**.
- [13] C. F. Macrae, P. R. Edgington, P. McCabe, E. Pidcock, G. P. Shields, R. Taylor, M. Towler, J. van de Streek, *J. Appl. Crystallogr.* **2006**, *39*, 453–457.
- [14] I. J. Bruno, J. C. Cole, P. R. Edgington, M. Kessler, C. F. Macrae, P. McCabe, J. Pearson, R. Taylor, *Acta Cryst. A* **2002**, *58*, 389–397.
- [15] R. Visbal, I. Ospino, J. M. López-de-Luzuriaga, A. Laguna, M. C. Gimeno, *J. Am. Chem. Soc.* **2013**, *135*, 4712–4715.
- [16] R. Visbal, J. M. López-de-Luzuriaga, A. Laguna, M. C. Gimeno, *Dalton. Trans.* **2014**, *43*, 328–334.
- [17] F. Neese, *WIREs Comput. Mol. Sci.* **2012**, *2*, 73–78.
- [18] a) S. Grimme, J. Antony, S. Ehrlich, H. Krieg, *J. Chem. Phys.* **2010**, *132*, 154104; b) S. Grimme, S. Ehrlich, L. Goerigk, *J. Comput. Chem.* **2011**, *32*, 1456–1465.
- [19] S. Sinnecker, A. Rajendran, A. Klamt, M. Diedenhofen, F. Neese, *J. Phys. Chem. A* **2006**, *110*, 2235–2245.
- [20] a) P. Schwerdtfeger, M. Dolg, W. H. E. Schwarz, G. A. Bowmaker, P. D. W. Boyd, *J. Chem. Phys.* **1989**, *91*, 1762–1774; b) F. Weigend, R. Ahlrichs, *Phys. Chem. Chem. Phys.* **2005**, *7*, 3297–3305.
- [21] A. D. Becke, *Phys. Rev. A* **1988**, *38*, 3098–3100.
- [22] J. P. Perdew, *Phys. Rev. B* **1986**, *33*, 8822–8824.
- [23] A. Schäfer, H. Horn, R. Ahlrichs, *J. Chem. Phys.* **1992**, *97*, 2571–2577.
- [24] A. Schäfer, C. Huber, R. Ahlrichs, *J. Chem. Phys.* **1994**, *100*, 5829–5835.
- [25] a) J. P. Perdew, M. Ernzerhof, K. Burke, *J. Chem. Phys.* **1996**, *105*, 9982–9985; b) C. Adamo, V. Barone, *J. Chem. Phys.* **1999**, *110*, 6158–6170.