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## Whey Protein Concentrate and Egg White Powder as Structuring Agents of Double Emulsions for Food Applications

--Manuscript Draft--

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<b>Abstract:</b>	<p>This work aims at studying the effects of whey protein concentrate (WP) and egg white powder (EW) as structuring agents in double emulsions (W1/O/W2). A D-optimal design was developed considering the following factors: type (WP, EW) and concentration (0, 5, 10 g/100 mL) of protein used to gel the inner water phase (W1), W1 volume percentage (20, 30, and 40%) in primary emulsion (W1/O), and W1/O volume percentage (40, 50, 60%) in W1/O/W2. The 21 samples were investigated by FT-IR spectroscopy, which revealed different protein conformations depending on W1 and W1/O fractions, and a better interaction with oil of WP rather than EW. Highly significant (<math>p &lt; 0.001</math>) multivariate models were computed for yield, rheological properties and creaming stability of W1/O/W2, being W1 and W1/O the most influent factors. Protein type significantly affected W1/O/W2 rheology, revealing a better structuring ability of EW with respect to WP, resulting in higher apparent viscosity and consistency coefficient values. A W1/O/W2 optimized for maximum values of apparent viscosity, yield, and creaming stability was developed, composed of 10 g/100 mL EW in W1, 29% W1, and 60% W1/O, with an oil content of 42.6 mL/100 mL. The optimized emulsion gave results in good agreement with the predicted values, thus confirming the validity of the developed multivariate models for the design of double emulsions with desired features.</p>

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## Whey Protein Concentrate and Egg White Powder as Structuring Agents of Double Emulsions for Food Applications

Maria Eletta Moriano, Cristina Alamprese

### ANSWERS TO REVIEWER#1

*Reviewer #1: The manuscript is well written and has a strong food technology aspect. The results of double emulsion yield and creaming stability show the same value independent of the formulation proposed. The emulsion formulation developed should be applied to encapsulate some bioactive component that shows the performance.*

We would like to thank the Reviewer for the positive evaluation of our work. The possibility to use the formulated double emulsion for encapsulation of bioactive compounds has been now mentioned in the Conclusions section.

### ANSWERS TO REVIEWER#2

*Reviewer #2: The manuscript "Whey Protein Concentrate and Egg White Powder as Structuring Agents of Double Emulsions for Food Applications" have studied the effect of some process variables on multiple emulsions containing whey protein concentrate or egg white powder. The manuscript is well written and I consider that the work was well conducted.*

We would like to thank the Reviewer for the positive evaluation of our work and the valuable suggestions that helped in improving our manuscript.

*I would recommend some changes before publishing this article:*

*Line 21: W1 and W1/O concentrations?*

Factor ranges have been added in the abstract.

*Lines 22-23: Which means "a better structuring ability"? A higher viscosity?*

Yes, we refer to the higher apparent viscosity and consistency coefficient values obtained with EW with respect to WP. The sentence has been clarified.

*Line 66: W1/O/W2 emulsion*

The word "emulsion" has been added in the sentence as suggested.

*Lines 66-68: Cite some products that these multiple emulsions can be applied and what is the desired rheological properties for these products.*

The text has been implemented as suggested.

*Lines 114-115: Were the emulsions destabilized when they were heated due to the decrease of the continuous phase viscosity?*

The primary emulsions were not destabilized during heating because of the induced gelation of the protein contained in  $W_1$ . As reported by Perez-Moral et al. (2014), the increased viscoelasticity of the gelled droplets can possibly be more resistant to coalescence. Gelation procedure was performed according to Surh et al. (2007); the reference has been added.

*Line 116: Which protein was added to the external water phase?*

The external water phase contained the same protein used in the inner water phase. The sentence has been clarified.

*Line 117: Why was NaCl used in the external phase in the same concentration than in the inner phase?*

The same NaCl concentration was used in  $W_2$  and  $W_1$  in order to balance the osmotic pressure and avoid water migration phenomena. The information has been added in the text.

*Line 143: Why was used a very high speed for centrifugation?*

The high speed was necessary in order to induce the separation of  $W_2$  and  $W_1/O$ . We chose the best  $g$  value in preliminary trials, following the indications given by Surh et al. (2007) and Perez-Moral et al. (2014).

*Line 144: Add details about the spectrophotometer.*

Details have been added.

*Line 156: Why was this shear rate chosen?*

The shear rate range was chosen based on preliminary trials, in order to have a good sensitivity of the rheometer and avoid emulsion system disruption. The information has been added in the text.

*Line 163: Creaming stability (CS)*

We used CS because the abbreviation was already explained in line 124 (now line 140).

*Line 211 and Figure 1: Why did NaCl solutions also show the bands related to amide and other peptide linkages?*

NaCl solution only shows the bands related to water, but, as reported in lines 226-227 (now lines 249-250), at  $1,645\text{ cm}^{-1}$  the  $\text{H}_2\text{O}$  bending vibrational band overlaps with amide I band.

*Lines 243-245: Can some differences occur due to the heat treatment that was applied in the multiple emulsions but not in the protein solutions?*

The Reviewer is right, this is possible. The text has been modified accordingly. In any case, the specific examples commented in the following lines refer to emulsions in which the gelation of the inner water phase was not induced.

*Line 259: 2,826 and 2,954?*

Thank you for the observation. The text is right, whereas the bands indicated in Fig. 1 were wrong. Now we have corrected them.

*Lines 259-260: Would not the change in the bands in relation to the pure oil indicate the interaction between oil and protein?*

We agree with the Reviewer. Indeed, the explanation in the following lines refers to oil and protein interactions.

*Line 262: Does EW not contain hydrophilic and hydrophobic regions?*

The Reviewer is right: also EW contains hydrophilic and hydrophobic regions. However, the sentence refers to the emulsifying activity of WP. To the best of our knowledge, EW has not being reported in the literature as a good W/O emulsifier, maybe because it has a hydrophilicity higher than WP, as explained in the following lines.

*Line 270 and Figure 3: What is PC1 and PC2?*

The abbreviations have been explained.

*Lines 283 and 290: W1 and W1/O concentrations?*

There is not a specific concentration to indicate, because the models are valid in the whole range considered for the experimental factors.

*Line 310:  $n = 1.11$  indicates a shear thickening behavior*

The Reviewer is right, but considering the error associated to the measurement we think that the behavior can be compared to that of a Newtonian fluid. Actually, the grand mean for WP samples was  $1.04 \pm 0.05$ . This information has been added and the sentence has been modified accordingly.

*Lines 312-313: It would be interesting to present the microscopy images of emulsions.*

Unfortunately, for this work we had not the possibility to perform the microscopy analysis of all the samples.

*Lines 340-342: Rheological behavior was already described earlier.*

Thank you for the observation. The lines have been deleted.

*Line 374: How can stability be greater than 100%?*

The CS higher than 100 is only a predicted value, with no physical meaning. The sentence has been clarified.

*Conclusions: More details about the main results should be added to the conclusions.*

Some details have been added as suggested, but keeping in mind the guidelines of the Journal, which suggest "Do not repeat in detail data given in Results and Discussion section".

#### **ANSWERS TO REVIEWER#4**

*Reviewer #4: General Comments.*

*The manuscript is well written, the design of experiments, methodology and analysis of results are appropriate, and the originality is acceptable. I think it should be accepted with minor corrections. See detailed comments.*

The authors wish to thank the Reviewer for the positive evaluation of the work and the valuable suggestions given, which helped us to improve the manuscript.

*Line 156. Why were these values of shear rate selected?*

The shear rate range was chosen based on preliminary trials, in order to have a good sensitivity of the rheometer and avoid emulsion system disruption. The information has been added in the text.

*Line 165. Explain briefly about the red oil layer.*

Some details have been added as suggested.

*Line 183. I guess the term  $\beta_{11}$  is missing because it's a categoric factor. This should be mentioned.*

The Reviewer is right and the explanation has been added in the text.

*Line 335. Why "with  $W_1$  decreasing"? I understand it should say "increasing" or nothing.*

We apologize for the mistake. The sentence has been corrected.

*Line 374. I understand CS 105% is a predicted value with no physical meaning. It should be mentioned.*

Thank you for the observation. The explanation has been added in the text.

*Table 3. For values of coded coefficients use scientific notation with 3 significant digits.*

Values have been modified as suggested.



[Click here to view linked References](#)

1           1           **Whey Protein Concentrate and Egg White Powder as Structuring**

2                           **Agents of Double Emulsions for Food Applications**

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## 34 **Introduction**

35 An interesting approach for the development of reduced-fat foods is the application of  
36 water-in-oil-in-water double emulsions ( $W_1/O/W_2$ ). They consist of small droplets of an  
37 inner water phase ( $W_1$ ) entrapped in oil droplets (O) that are, in their turn, dispersed in  
38 another aqueous phase ( $W_2$ ) (Muschiolik 2007). The concentration of  $W_1$  droplets inside  
39 the primary emulsion ( $W_1/O$ ) is affected by the ratio of  $W_1$  to O, while the concentration  
40 of oil droplets in the final emulsion can be controlled by using different  $W_1/O$  to  $W_2$   
41 ratios. The main advantage of  $W_1/O/W_2$  lies in the possibility to obtain a typical oil-in-  
42 water (O/W) emulsion structure, but with a reduced fat content. Actually, rheology of  
43  $W_1/O/W_2$  can be considerably different from that of an O/W emulsion due to the higher  
44 effective volume fraction of the particles, which is approximately represented by the  
45 sum of the volume fraction of O and  $W_1/O$ . As a result, highly viscous products can be  
46 made with lower fat content (McClements 2016). Another advantage of double  
47 emulsions is the possibility to encapsulate a bioactive component within one of the  
48 emulsion phases, in order to improve its release behaviour, oral administration and  
49 digestion (Artiga-Artigas et al. 2019; Lamba et al. 2015; Momeni et al. 2017).

50 One of the main issues connected to real food applications of  $W_1/O/W_2$  is their high  
51 susceptibility to breakdown during storage or when exposed to the common stresses  
52 involved in food preparation (e.g., mechanical stresses, chilling, freezing, heating, etc.)  
53 (McClements 2016). In order to overcome this issue, gelation of  $W_1$  may be a useful  
54 strategy (Oppermann et al. 2015; Perez-Moral et al. 2014), together with the use in  
55  $W_1/O$  of a strong lipophilic emulsifier such as polyglycerol polyricinoleate (PGPR)  
56 (Artiga-Artigas et al. 2019; Muschiolik and Dickinson 2017). The incorporation in  $W_1$   
57 of a polymer able to form a network in the water phase droplets or a gel-like layer at the

1 58 inner water-oil interface (e.g., xanthan, alginate, gelatine, agarose, bovine serum  
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3 59 albumin, sodium caseinate) has been proposed as a way of improving the long-term  
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5 60 stability of  $W_1/O/W_2$ . For instance, Surh et al. (2007) studied the possibility of  
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7 61 improving  $W_1/O/W_2$  stability by thermal gelation of whey proteins contained within  $W_1$ .  
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9 62 They demonstrated that gelling the internal aqueous phase of the primary emulsion does  
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11 63 not facilitate the formation of the double emulsion, but it can improve the long-term  
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13 64 stability of the system; however, the Authors suggest further investigations.

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15 65 The effects of using double emulsions as fat replacers in food products have been  
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17 66 studied so far for dairy products (Leong, Zhou, Kukan, Ashokkumar, & Martin, 2017;  
18  
19 67 Lobato-Calleros, Rodriguez, Sandoval-Castilla, Vernon-Carter, & Alvarez-Ramirez,  
20  
21 68 2006; Lobato-Calleros et al., 2008; Lobato-Calleros, Recillas-Mota, Espinosa-Solares,  
22  
23 69 Alvarez-Ramirez, & Vernon-Carter, 2009; Márquez & Wagner, 2010), meat products  
24  
25 70 (Cofrades, Antoniou, Solas, Herrero, & Jiménez-Colmenero, 2013; Freire, Bou,  
26  
27 71 Cofrades, Solas, & Jiménez-Colmenero, 2016), mayonnaise and dressings (Matsumoto  
28  
29 72 & Kohda, 1980; Takahashi, Aizawa, Tamai, Yoshida, & Takahashi, 1986), but there are  
30  
31 73 still numerous challenges to be addressed before a successful commercial use  
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33 74 (McClements, 2016).

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35 75 Since properties of  $W_1/O/W_2$  are affected by several factors, Design of Experiments  
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37 76 (DoE) techniques can be successfully applied to the systematic characterization of these  
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39 77 systems, in order to acquire deeper knowledge about their structure and be able to  
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41 78 design a  $W_1/O/W_2$  emulsion with the desired features. The aim of this work was thus to  
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43 79 study the effects of two different biopolymers (i.e., whey proteins and egg white), used  
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45 80 as gelation agents, on yield, rheological properties, and creaming stability of  $W_1/O/W_2$ .  
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47 81 A D-optimal design was developed considering as experimental factors the type and  
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1 82 concentration of biopolymer used to gel  $W_1$ , the  $W_1$  volume percentage in the primary  
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3 83 emulsion, and the  $W_1/O$  volume percentage in the final double emulsion. Whey proteins  
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5 84 and egg albumen were selected as structuring agents because they are widely used  
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8 85 natural food-grade ingredients, able to form thermo-irreversible gels upon heating.  
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10 86 While few works deal with the use of whey proteins to gel the inner water phase of  
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12 87 double emulsions (Balcaen et al. 2016; Opperman et al. 2015; Sağlam et al. 2011; Surh  
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14 88 et al. 2007), to the best of our knowledge, no studies have been published so far about  
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16 89 the use of egg white as a gelation agent for multiple emulsions.  
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## 22 23 91 **Materials and Methods**

### 24 25 92 **Materials**

26 93 Whey protein concentrate WPC80 (WP) and spray-dried egg white powder (EW) were  
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28 94 kindly supplied by Milkiland EU (Ostrów Mazowiecka, Poland) and Lactosan-Sanovo  
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30 95 Ingredients Group (Zeven/Aspe, Germany), respectively. According to label data, both  
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32 96 WP and EW contained 80 g/100 g proteins in dry matter. Corn oil (Carrefour, Boulogne-  
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34 97 Billancourt, France) was bought in a local supermarket. PGPR was kindly supplied by  
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36 98 Lasenor (Barcelona, Spain; product commercial name: VEROL PR). NaCl, 1,3,6,8-  
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38 99 pyrenetetrasulfonic acid tetrasodium salt hydrate (PTSA) and Oil red O dye were  
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40 100 purchased from Sigma Aldrich (Saint Louis, MO, USA).  
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### 50 102 **Experimental Design**

51 103 Experiments were planned according to a D-optimal design in order to study  
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53 104 simultaneously the main and interaction effects of four factors: type of protein powder  
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55 105 (WP and EW); protein powder concentration in  $W_1$  (0, 5, and 10 g/100 mL);  $W_1$  volume  
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1 106 percentage (20, 30, and 40%) in the primary emulsion;  $W_1/O$  volume percentage (40,  
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3 107 50, and 60%) in the final  $W_1/O/W_2$ . Factor levels were chosen based on literature study  
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5 108 and preliminary trials. A total of 21 experiments, comprising 3 replicates of the central  
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7 109 point, were performed (Table 1). The order of experiments was fully randomized in  
8  
9 110 order to avoid possible bias related to systematic effects and uncontrolled variations.  
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11 111 Sample identification refers to the protein biopolymer code (WP or EW), followed by  
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13 112 the actual level of the other three experimental factors considered (protein powder  
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15 113 concentration,  $W_1$  percentage,  $W_1/O$  percentage).  
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### 23 115 **Preparation of Double Emulsions**

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25 116 Double emulsions were prepared according to the procedure suggested by Perez-Moral  
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27 117 et al. (2014) slightly modified and adapted to the subsequent system characterizations.  
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29 118 For rheological and Fourier-transform infrared (FT-IR) spectroscopy measurements,  
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31 119 double emulsions (150 mL) were prepared as follows: WP or EW (5 or 10 g/100 mL)  
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33 120 were dissolved in NaCl solution (0.4 g/100 mL) and stirred at room temperature (22 °C)  
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35 121 for 1 h. These solutions were used as  $W_1$  in  $W_1/O$ . When no proteins were required in  
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37 122  $W_1$ , NaCl solution (0.4 g/100 mL) without protein powder addition was used as  $W_1$ .  
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39 123  $W_1/O$  were prepared by adding the established amount of  $W_1$  (20, 30 or 40%) to corn oil  
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41 124 containing 4 g/100 mL PGPR. The phases were mixed using a heavy duty blender  
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43 125 (Waring Laboratory, Torrington, CT, USA) at 18,000 rpm for 30 s and at 20,000 rpm for  
44  
45 126 further 30 s. When  $W_1/O$  contained proteins, they were heated in a water bath at 80 °C  
46  
47 127 for 20 min and cooled down in iced water for 15 min in order to induce gelation (Surh  
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49 128 et al., 2007). External water phase ( $W_2$ ) was prepared by dissolving in NaCl solution  
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51 129 (0.4 g/100 mL) 5 g/100 mL of the same protein powder used in  $W_1$ , and stirring for 1 h  
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1 130 at room temperature. The same NaCl concentration was used in  $W_1$  and  $W_2$  in order to  
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3 131 balance the osmotic potential and avoid water diffusion phenomena (Perez-Moral et al.,  
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5 132 2014).  $W_1/O/W_2$  were prepared by adding the given amount of  $W_1/O$  (40, 50 or 60%) to  
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8 133  $W_2$  and mixing as previously described for  $W_1/O$  preparation.  
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10 134 When intended for yield determination,  $W_1/O/W_2$  (150 mL) were prepared following  
11  
12 135 the same procedure but using sodium phosphate buffer 5 mM with NaCl 100 mM (pH  
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14 136 7) and PTSA (0.2 g/100 mL) as  $W_1$ . The corresponding  $W_1/O/W_2$  prepared without dye  
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16 137 addition were used as blank.  
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19 138 For creaming stability (CS) evaluation,  $W_1/O/W_2$  (50 mL) were prepared as previously  
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21 139 described for rheological characterization, but colouring the oil phase with the Oil red O  
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23 140 dye (0.0015 g/100 mL).  
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### 29 142 **FT-IR Spectroscopy Analysis**

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31 143 FT-IR spectra of  $W_1/O/W_2$ , corn oil, protein solutions (EW and WP both at 5 and 10  
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33 144 g/100 mL in 0.4 g/100 mL NaCl solution) and NaCl solution (0.4 g/100 mL) were  
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35 145 acquired by a Vertex 70 spectrometer (Bruker Optics, Milan, Italy) equipped with a  
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37 146 Germanium multiple reflection ATR cell. Spectral data were collected in duplicate in the  
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39 147 range 4,000-400  $\text{cm}^{-1}$  at 20 °C, with a 4  $\text{cm}^{-1}$  resolution and 32 scans for both  
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41 148 background and samples. Opus software (v. 6, Bruker Optics, Ettlingen, Germany) was  
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43 149 used for instrument control and spectral acquisition. Before elaboration, duplicated  
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45 150 spectra were averaged.  
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### 51 152 **Double Emulsion Yield**

52 153 Yield, expressed as the percentage of emulsified  $W_1$  droplets remaining inside the oil  
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154 droplets, was determined spectrophotometrically using PTSA as a tracer, following a  
155 method adapted from Perez-Moral et al. (2014). W<sub>1</sub>/O/W<sub>2</sub> samples, after storage at 4 °C  
156 overnight, were centrifuged in a benchtop centrifuge (Centrikon T-42K, Neufahrn,  
157 Germany) at 44,800 x g (20,000 rpm) for 20 min at 10 °C. The concentration of dye  
158 appearing in W<sub>2</sub> was determined spectrophotometrically (V-650 spectrophotometer,  
159 **Jasco Europe, Cremella, LC, Italy**) at 374 nm by using a PTSA calibration curve. Yield  
160 was calculated through the following equation adapted from Surh et al. (2007):

$$161 \quad \text{Yield (\%)} = 1 - \left( \frac{C_f}{C_i - C_f} \cdot \frac{1 - \phi_{W_1O}}{\phi_{W_1} \cdot \phi_{W_1O}} \right) \quad (1)$$

162 where:  $C_i$  is the initial PTSA concentration in W<sub>1</sub> (0.2 g/100 mL);  $C_f$  is the final PTSA  
163 concentration in W<sub>1</sub> (g/100 mL);  $\phi_{W_1O}$  is the volume percentage of W<sub>1</sub>/O in W<sub>1</sub>/O/W<sub>2</sub>;  
164  $\phi_{W_1}$  is the volume percentage of W<sub>1</sub> in W<sub>1</sub>/O.

165 Yield measurements were performed in quadruplicate.

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### 167 **Rheological Characterization of Double Emulsions**

168 Flow curves of W<sub>1</sub>/O/W<sub>2</sub> were measured in triplicate at 25 °C using a Physica MCR 300  
169 rheometer (Anton Paar, Graz, Austria) equipped with coaxial cylinders (CC27), in a  
170 150-500 s<sup>-1</sup> range of shear rate. **Preliminary trials were performed to choose a shear rate**  
171 **range able to guarantee a good sensitivity of the rheometer, while avoiding emulsion**  
172 **destabilization.** Apparent viscosity was taken at 310 s<sup>-1</sup>, while consistency coefficient  
173 (K) and flow behaviour index (n) were obtained by fitting the curves with the power law  
174 equation:

$$175 \quad \tau = K \cdot \dot{\gamma}^n \quad (2)$$

176 where  $\tau$  is the shear stress (mPa) and  $\dot{\gamma}$  is the shear rate (s<sup>-1</sup>) (Steffe 1996).

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3 178 **Creaming Stability of Double Emulsions**

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6 179 CS of W<sub>1</sub>/O/W<sub>2</sub> was determined in triplicate according to a method adapted from  
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8 180 Karaca et al. (2011). Immediately after preparation, samples stained with the Oil red O  
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10 181 dye were poured into 10 mL graduated cylinders and stored at 4 °C. After 1 and 24 h,  
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12 182 the height of the visible darker red oil layer creamed and separated from the bottom  
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14 183 turbid layer was registered. CS was calculated through the following equation:  
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$$CS (\%) = 100 - \left( \frac{H_t}{H_e} \cdot 100 \right) \quad (3)$$
  
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22 185 where H<sub>t</sub> is the height of the red oil layer creamed after 1 or 24 h of storage, and H<sub>e</sub> is  
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24 186 the total height of the W<sub>1</sub>/O/W<sub>2</sub>.

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29 188 **Data Analysis**

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32 189 In order to obtain a high quality protein spectrum, reduced FT-IR spectra (considered  
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34 190 ranges: 3,728-2,754 cm<sup>-1</sup> and 2,272-1,000 cm<sup>-1</sup>) were subtracted of the NaCl solution  
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36 191 spectrum and pre-treated with smoothing (moving average, segment size = 5) and  
37  
38 192 second derivative transformation (Savitzky-Golay algorithm, polynomial order = 2,  
39  
40 193 smoothing points = 11) (Kong and Yu, 2007). Afterwards, the obtained spectra were  
41  
42 194 analyzed by means of Principal Component Analysis (PCA) in order to investigate  
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44 195 possible sample patterns and importance of spectral variables (The Unscrambler X  
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46 196 software, v. 10.4.1, CAMO Process A/S, Oslo, Norway).

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49 197 Data collected for DoE were elaborated by means of the Response Surface

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52 198 Methodology, postulating a quadratic model for each of the considered response  
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55 199 variables:  
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1 200  $y = \beta_0 + \beta_1 x_1 + \beta_2 x_2 + \beta_3 x_3 + \beta_4 x_4 + \beta_{12} x_1 x_2 + \beta_{13} x_1 x_3 + \beta_{14} x_1 x_4 + \beta_{23} x_2 x_3 + \beta_{24} x_2 x_4 + \beta_{34} x_3 x_4 + \beta_{22} x_2^2 + \beta_{33} x_3^2 + \beta_{44} x_4^2 + \varepsilon$  (4)

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4 201 where  $y$  is the value of the considered response variable;  $x_1$ ,  $x_2$ ,  $x_3$ , and  $x_4$  are,  
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6 202 respectively, the type and the level of protein powder concentration, the volume  
7  
8 203 percentage of  $W_1$ , and the volume percentage of  $W_1/O$ ;  $\beta_0$  is the model intercept;  $\beta_1$ ,  $\beta_2$ ,  
9  
10 204  $\beta_3$ , and  $\beta_4$  are the linear coefficients;  $\beta_{12}$ ,  $\beta_{13}$ ,  $\beta_{14}$ ,  $\beta_{23}$ ,  $\beta_{24}$ , and  $\beta_{34}$  are the interaction  
11  
12 205 coefficients;  $\beta_{22}$ ,  $\beta_{33}$ , and  $\beta_{44}$  are the quadratic coefficients ( $\beta_{11}$  is missing because of the  
13  
14 206 **category nature of the factor**);  $\varepsilon$  is the random error.  
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18 207 Before model calculation, data were checked for normal distribution and skewness;  
19  
20 208 apparent viscosity and  $K$  data were transformed in the inverse root square and log  
21  
22 209 values, respectively. The significance of each coefficient was determined by one-way  
23  
24 210 analysis of variance (ANOVA). When non-significant terms not counting to support  
25  
26 211 hierarchy were present, a model reduction was considered. A multi-objective  
27  
28 212 optimization of the double emulsion was also carried out, calculating an overall  
29  
30 213 desirability function (Alamprese et al. 2007; Montgomery 2001). Data elaboration was  
31  
32 214 performed by Design Expert 10.0.6 (Stat-Ease Inc., Minneapolis, MN, USA).  
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## 40 216 **Results and Discussion**

### 42 217 **Double Emulsion Theoretical Composition**

43  
44 218 As expected, the lowest volume percentage of  $W_1$  in  $W_1/O$  and the highest volume  
45  
46 219 percentage of  $W_1/O$  in  $W_1/O/W_2$  corresponded to the maximum theoretical oil content  
47  
48 220 (48 mL/100 mL) in  $W_1/O/W_2$  (Table 1). Increasing  $W_1$  levels and decreasing  $W_1/O$   
49  
50 221 levels implied a progressive decrease in fat content, until reaching the minimum value  
51  
52 222 (24 mL/100 mL) in the emulsions containing 40% of both  $W_1$  and  $W_1/O$ . The lowest  
53  
54 223 protein concentration (1.6 g/100 mL) was obviously obtained in samples with no protein  
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1 224 addition in  $W_1$ , for which the only protein contribution came from EW or WP dispersed  
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3 225 in  $W_2$ .

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### 7 8 227 **FT-IR Spectroscopy Analysis**

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10 228 Fig. 1 shows FT-IR spectra of double emulsions, corn oil, dispersions of EW and WP  
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12  
13 229 (both at 5 and 10 g/100 mL) and NaCl solution (0.4 g/100 mL) after the spectral interval  
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15  
16 230 reduction and the baseline offset correction. In order to eliminate the less informative  
17  
18 231 and noisiest regions, as well as the absorption band of  $\text{CO}_2$ , only the following spectral  
19  
20 232 ranges were considered:  $3,728\text{-}2,754\text{ cm}^{-1}$  and  $2,272\text{-}1,000\text{ cm}^{-1}$ .

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22  
23 233 The broad band in the OH stretch region ( $3,000\text{-}3,700\text{ cm}^{-1}$ ) is mostly related to water.

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25 234 In WP-based double emulsions (Fig. 1a), the height of this band is more variable than in

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27  
28 235 EW-based samples, indicating a lower availability of water in some of the WP-based

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30 236 systems (Hesso et al. 2015). Moreover, in WP-based double emulsions a more

31  
32 237 pronounced shoulder around  $3,270\text{-}3,280\text{ cm}^{-1}$  is evident, which can be related to an

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34  
35 238 increase in hydrogen-bonded-associated chains of water with the emulsion proteins

36  
37 239 (Bock and Damodaran 2013).  $W_1/O/W_2$  spectra are characterized also by two bands

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39 240 around  $1,640$  and  $1,550\text{ cm}^{-1}$ , related to amide I and II, respectively, and due to the  $\text{C}=\text{O}$

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41  
42 241 and  $\text{C-N}$  stretching of the peptide linkages (Kong and Yu 2007; Mohammadian et al.

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44  
45 242 2018). The intensity of these absorption bands decreased in  $W_1/O/W_2$  with respect to

46  
47 243 protein dispersions, due to a lower protein and water content of the multiphase systems

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49 244 (Hesso et al. 2015). Actually, at  $1,645\text{ cm}^{-1}$  also the  $\text{H}_2\text{O}$  bending vibrational band

50  
51 245 shows a high absorbance, overlapping with amide I band (Kong and Yu 2007).

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53  
54 246 Frequencies of the amide I band components are correlated to each secondary elements

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56  
57 247 of proteins, while amide II band is much less sensitive to protein conformation changes.

1 248 In particular, the second derivative spectra between 1,600 and 1,700  $\text{cm}^{-1}$  allow the  
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3 249 identification of various secondary structures of the proteins, even if caution has to be  
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5 250 exercised in the IR spectra interpretations (Kong and Yu 2007). After the subtraction of  
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7  
8 251 the NaCl solution spectrum and the second derivative (d2) transformation (Fig. 2),  
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10 252 spectra of  $W_1/O/W_2$  in the considered range (1,600-1,700  $\text{cm}^{-1}$ ) still showed a high  
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12  
13 253 numbers of extensively overlapped bands, thus deconvolution and peak quantification  
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15 254 were not attempted, rather observing qualitative spectral changes.  
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17 255 WP and EW protein solutions showed major peaks around 1,631 and 1,637  $\text{cm}^{-1}$ ,  
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19  
20 256 respectively, which indicate proteins with a high content of  $\beta$ -sheet structure. A second  
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23 257 high peak centered at 1,654  $\text{cm}^{-1}$  was also visible in solutions of 5 g/100 mL WP and 10  
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26 258 g/100 mL EW, indicating a relevant content of  $\alpha$ -helix structure. WP at 10 g/100 mL  
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28 259 showed, instead, a relevant content of random coil structures (peak at 1,647  $\text{cm}^{-1}$ , Fig.  
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30 260 2a). The different amount of water in the measured protein solutions accounts for the  
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33 261 different protein conformations. In  $W_1/O/W_2$  samples, shifts and different intensities of  
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35 262 these peaks can be observed, indicating changes in protein conformation, probably due  
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38 263 to heating eventually applied during emulsion preparation and the protein exposure to  
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40 264 different amounts of water and oil. For instance, sample EW\_00\_20\_60 (dotted line  
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43 265 among double emulsion samples in Fig. 2b) has a low amount of proteins (1.6 g/100  
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45 266 mL), a similar amount of fat (48 mL/100 mL) and water (49.9 mL/100 mL) and it  
46  
47 267 shows a low intensity of the amide I band, mostly constituted by random coil structures  
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49  
50 268 (d2 peak centered at 1,643  $\text{cm}^{-1}$ ). On the contrary, sample EW\_00\_40\_60 (dashed line  
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53 269 among the double emulsion sample in Fig. 2b), composed of the same amount of  
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55 270 proteins (1.6 g/100 mL), but a higher amount of water (61.9 mL/100 mL) rather than fat  
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57 271 (36 mL/100 mL), showed a bimodal distribution of d2 spectra, centered at 1,637 and



1 272 1,652 cm<sup>-1</sup>, indicating similar content of β-sheet and α-helix structures.  
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4 273 The typical peaks of oil (2,800-3,030, 1,745, 1,000-1,490 cm<sup>-1</sup>) resulted more evident in  
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6 274 WP-based emulsions rather than in those containing EW (Fig. 1). This result indicates  
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8 275 different interactions of the two types of proteins with oil. In particular, the appearance  
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10 276 in WP-based W<sub>1</sub>/O/W<sub>2</sub> spectra of the bands related to CH asymmetric and symmetric  
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12 277 stretching of CH<sub>2</sub> (2,926 and 2,854 cm<sup>-1</sup>, respectively) can indicate conformational  
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14 278 changes of the oil alkyl chains in the oil/water interfaces, suggesting the involvement of  
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16 279 CH<sub>2</sub> groups of oil with proteins (Guo et al. 2019). Actually, as reported in the literature,  
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18 280 whey proteins naturally have hydrophobic and hydrophilic regions that make them  
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20 281 surface-active, thus facilitating emulsion formation and stabilization (Cheong et al.  
21  
22 282 2015). Moreover, whey protein solutions have higher values of hydrophobicity than egg  
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24 283 white protein solutions under the same pH conditions (Kuropatwa et al. 2009), thus  
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26 284 supporting the hypothesis of a better interaction of WP with oil. As a further  
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28 285 confirmation of the different interactions of WP and EW proteins with oil, a PCA  
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30 286 applied to the reduced spectra after subtraction of the NaCl solution spectrum and the  
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32 287 second derivative transformation showed that W<sub>1</sub>/O/W<sub>2</sub> samples differentiated along the  
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34 288 first principal component (PC1) depending on the type of protein used (Fig. 3a).  
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36 289 Loading values revealed that the most influent variables in the differentiation of samples  
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38 290 along PC1 corresponded to the absorption peaks typical of oils (2,800-3,030, 1,745 cm<sup>-1</sup>;  
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40 291 Fig. 3b).  
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## 52 293 **Multivariate Models for Double Emulsion Characteristics**

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54 294 Double emulsion yield, rheological parameters and creaming stability are reported in  
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56 295 Table 2, while the corresponding multivariate models are shown in Table 3. All the  
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1 296 calculated models resulted highly significant ( $p < 0.001$ ). Adjusted  $R^2$  values were  
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3 297 higher than 0.89 for rheological variable models and around 0.72-0.75 for those of yield  
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5 298 and stability. Predicted  $R^2$  values resulted low for yield and CS, but the lack of fit was  
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8 299 always not significant except for the model of  $n$  ( $p = 0.01$ ), for which the response  
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10 300 surface did not fit adequately the experimental domain. Adequate precision was always  
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13 301 good, being much higher than 4.  $W_1$  and  $W_1/O$  appeared to be the most influent factors.  
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15 302 All the  $W_1/O/W_2$  samples showed high yields ( $> 95\%$ ), meaning a very good retention  
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17 303 of  $W_1$  inside the oil droplets; no differences were observed between EW- or WP-based  
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20 304 samples (Table 2). This result could be ascribed to the use of PGPR as lipophilic  
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22 305 emulsifier, which ensure a high water encapsulation efficiency (Perez-Moral et al. 2014)  
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24 306 due to a steric stabilization of the interfacial layer (Márquez al. 2010) and a smaller  
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26 307 droplet size in  $W_1/O$  due to the high hydrophobicity (Artiga-Artigas et al. 2019).  $W_1$  and  
27  
28 308  $W_1/O$  showed positive and highly significant ( $p < 0.001$ ) coded coefficients for yield  
29  
30 309 (Table 3), revealing a direct effect on  $W_1$  retaining ability of  $W_1/O/W_2$ . The response  
31  
32 310 surface (Fig. 4a) is characterized by a slight curvature, due to the significance ( $p < 0.05$ )  
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34 311 of the interaction term between  $W_1$  and  $W_1/O$  ( $\beta_{34}$ ). This means that the effect of  $W_1/O$   
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36 312 on yield is more accentuated at low levels of  $W_1$  (20%) rather than at high levels.  
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38 313 Protein concentration and type did not significantly affect yield, contrarily to what  
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40 314 reported by Opperman et al. (2015) and Perez-Moral et al. (2014) who found an  
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42 315 increase in yield with  $W_1$  gelation. The contrasting results may be due to the fact that  
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44 316 the cited studies considered a constant  $W_1/O/W_2$  formulation without investigating  
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46 317 changes associated to the variation of oil and water volume fractions. In the present  
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48 318 study, it is presumable that the effect on yield of oil and water fraction changes was  
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50 319 more marked than the effect of  $W_1$  gelation. Moreover, in this study, proteins were  
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1 320 added also to  $W_2$  in order to further reduce water mobility in the system and this may  
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3 321 have leveled the effect of  $W_1$  gelation.  
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5 322 Viscosity results are very interesting for fat reduction strategies since they affect both  
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7 323 air incorporation ability of the system (Márquez and Wagner 2010) and sensory  
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9 324 perception of emulsion creaminess (van Aken et al. 2011). Apparent viscosity and  $K$   
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11 325 showed a comparable trend, with samples containing 60%  $W_1/O$  resulting in the highest  
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13 326 values (Table 2). Emulsions containing EW were characterized by  $n$  ranging from 0.36  
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15 327 to 0.91, indicating a non-Newtonian pseudoplastic behavior of different intensity. On  
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17 328 the contrary, WP-based samples showed  $n$  values in the range 0.94-1.11, with a grand  
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19 329 mean and standard deviation value of  $1.04 \pm 0.05$ , revealing a quite Newtonian behavior,  
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21 330 as already reported in previous studies about single emulsions (Erçelebi and Ibanoglu  
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23 331 2009). Shear thinning of EW-emulsions may be connected to the presence of aggregated  
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25 332 droplets (Panagopoulou et al. 2017) that are deformed and disrupted as the shear rate  
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27 333 increases (Demetriades et al. 1997). Thus, the differences observed between samples  
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29 334 containing the two different protein powders could be linked to the higher ability of WP  
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31 335 proteins in interacting with other components of the emulsion and creating a more  
32  
33 336 homogenously dispersed phase, in agreement with the FT-IR results. Apparent viscosity  
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35 337 was significantly influenced by all the experimental factors (Table 3), with a tendency to  
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37 338 increase at higher concentrations of protein powder in  $W_1$  ( $p < 0.01$ ), higher level of  
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39 339  $W_1/O$  ( $p < 0.001$ ), and lower percentages of  $W_1$  ( $p < 0.001$ ). The effect of protein  
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41 340 concentration was in agreement with the findings of Oppermann et al. (2016) who  
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43 341 demonstrated that gelation of  $W_1$  in double emulsions allows to obtain viscosity values  
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45 342 higher than those of simple O/W emulsions, despite the lower oil content of  $W_1/O/W_2$ .  
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47 343 In addition, they found that all the fat-related mouth-feel and after-feel attributes  
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1 344 increase in gelled samples, balancing the effects of fat reduction. In the present work,  
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3 345 also the protein powder type resulted significant ( $p < 0.001$ ), demonstrating EW to be a  
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5 346 better structuring agent with respect to WP (Fig. 4c-d). Actually, at equal levels of  $W_1$   
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7 347 and  $W_1/O$ , changing the protein powder from WP to EW quite doubled the apparent  
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9 348 viscosity of double emulsions (Table 2). Similarly, protein type and percentage of  $W_1/O$   
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11 349 had a strong positive effect ( $p < 0.001$ ) on K, with EW-containing samples showing the  
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13 350 highest values (Table 2). On the other hand, increasing  $W_1$  led to a significant decrease  
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15 351 ( $p < 0.001$ ) of K due to the reduction of the oil content (Fig. 4b). Protein concentration  
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17 352 in  $W_1$  was involved in significant ( $p < 0.05$ ) interactions with both  $W_1$  ( $\beta_{23}$ ) and  $W_1/O$   
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19 353 ( $\beta_{24}$ ). In fact, the reduction of K values with  $W_1$  increasing was more marked at high (10  
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21 354 g/100 mL) than at low (0-5 g/100 mL) protein concentrations. Similarly, at high and  
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23 355 intermediate levels of protein powder concentration, the increase of K while increasing  
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25 356  $W_1/O$  was less marked than in absence of proteins. The reduction of K values with the  
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27 357 increase in  $W_1$  was higher at high levels of  $W_1/O$ , due to the significant and negative  
28  
29 358 interaction  $\beta_{34}$  ( $p < 0.05$ ; Fig. 4b).  $W_1$  showed a direct effect ( $p < 0.05$ ) on n, while  
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31 359  $W_1/O$  resulted in an opposite effect ( $p < 0.01$ ): the less the  $W_1/O$  percentage, the higher  
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33 360 the n values.  
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35 361 CS is of paramount importance for double emulsions intended as alternative fats in food  
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37 362 formulation, since they have to be prepared beforehand and stored until use. All the  
38  
39 363 samples showed no creaming phenomena 1 h after production (data not shown), while  
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41 364 after 24 h samples with the lowest  $W_1/O$  level (40%) resulted in the lowest stability (<  
42  
43 365 80%), irrespective the protein type (Table 2). The only exception was sample  
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45 366 WP\_00\_20\_40, for which a CS of  $88 \pm 1\%$  was registered. In particular, the highest  
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47 367 water content (about 72 mL/100 mL) produced a stability lower than 50% in samples  
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1 368 EW\_00\_40\_40, WP\_05\_40\_40, and WP\_00\_40\_40, suggesting the unsuitability of  
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3 369 these emulsions as storable ingredients. CS was significantly ( $p < 0.001$ ) improved as  
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5 370 the percentage of  $W_1/O$  increased, while an inverse significant effect ( $p < 0.01$ ) was  
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8 371 found for  $W_1$  volume fraction (Table 3). These findings are in agreement with previous  
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10 372 studies demonstrating that 20-30% of  $W_1$  in primary emulsions is the optimum phase  
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12 373 volume ratio in terms of stability (Su et al. 2006). As reported above for yield, also for  
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14 374 stability a significant effect of the  $W_1$  gelling was not observed, confirming that the  
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16 375 other factors had a stronger effect on the system.  
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### 22 377 **Optimization of a Double Emulsion Intended as Shortening Replacer**

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25 378 The developed multivariate models can be used to design  $W_1/O/W_2$  with peculiar  
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27 379 characteristics. As an example, a  $W_1/O/W_2$  intended for shortening replacement in  
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29 380 reduced-fat baked goods was optimised by means of desirability function. The  
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31 381 optimization criteria were chosen in order to obtain an emulsion with a good structure  
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33 382 (maximum values of apparent viscosity and yield) and a high stability (maximum value  
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35 383 of CS). The same level of importance was given to all the constraints and a linear  
36  
37 384 desirability function was used (weight = 1). The calculated formulation had a very high  
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39 385 desirability value ( $d = 0.990$ ) and it was composed of 10 g/100 mL EW in  $W_1$ , 29%  $W_1$   
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41 386 in  $W_1/O$ , and 60%  $W_1/O$  in  $W_1/O/W_2$ , thus reaching a content of oil and protein of 42.6  
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43 387 mL/100 mL and 3.5 g/100 mL, respectively. The optimized emulsion was produced and  
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45 388 characterized in duplicate, giving results in good agreement with the predicted values  
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47 389 (mean  $\pm$  standard error): yield,  $98.71 \pm 0.04\%$  vs.  $99.35 \pm 0.30\%$ ; apparent viscosity,  
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49 390  $70.4 \pm 4.7$  vs.  $63.1 \pm 0.1$  mPa s; CS,  $100 \pm 1\%$  vs.  $105 \pm 5\%$  (the predicted value is higher  
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57 391 than 100% just as a result of model extrapolation). The good results obtained confirmed  
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1 392 the validity of the developed multivariate models for the design of  $W_1/O/W_2$  with given  
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3 393 features.

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## 7 8 395 **Conclusions**

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11 396 WP and EW resulted to be good structuring biopolymers for double emulsions intended  
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13 397 for food applications and possible encapsulation of bioactive compounds. In particular,  
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15 398 despite a better oil interaction of WP assessed by FT-IR spectroscopy, the use of EW  
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17 399 proved to be more effective in structuring  $W_1/O/W_2$  emulsions, resulting in higher  
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21 400 apparent viscosity and consistency coefficient values.

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23 401 Highly significant multivariate models were computed for yield, rheological properties,  
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25 402 and CS of  $W_1/O/W_2$  emulsions as a function of protein type and concentration,  $W_1$  and  
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27 403  $W_1/O$  volume percentages.  $W_1$  and  $W_1/O$  were the most influent factors, allowing the  
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29 404 tuning of the system characteristics. To the best of our knowledge, no papers in the  
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32  
33 405 literature deal with the study of  $W_1/O/W_2$  properties in such a systematic and  
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35 406 comprehensive approach, including all the factors considered in this research. Thus, the  
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38 407 obtained multivariate models represent a useful starting point for the development of  
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40 408  $W_1/O/W_2$  systems designed for targeted purposes, as demonstrated also by the good  
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43 409 results obtained in the optimization example.

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48  
49  
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54  
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56  
57  
58 415 respect to the work described in this manuscript.

1 416 **Availability of data and material:** not applicable  
2  
3 417 **Code availability:** not applicable  
4  
5  
6 418 **Authors' contributions. Maria E. Moriano:** Methodology, Formal analysis,  
7  
8 419 Investigation, Data curation, Writing-original draft preparation. **Cristina Alamprese:**  
9  
10 420 Conceptualization, Methodology, Formal analysis, Writing-review and editing,  
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12  
13 421 Visualization, Supervision.  
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18 423 **References**  
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1 540 **Figure captions**  
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4 541 **Fig. 1** FT-IR spectra after interval reduction and baseline offset correction of: (a) corn  
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6 542 oil (dotted grey line), dispersions of whey protein concentrate (WP) at 5 g/100 mL  
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8 543 (black dashed line) and 10 g/100 mL (black line), NaCl solution at 0.4 g/100 mL (black  
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10 544 dotted line), and WP-based double emulsions (grey lines); (b) corn oil (dotted grey  
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12 545 line), dispersions of egg white powder (EW) at 5 g/100 mL (black dashed line) and 10  
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14 546 g/100 mL (black line), NaCl solution at 0.4 g/100 mL (black dotted line), and EW-based  
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16 547 double emulsions (grey lines)  
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23 549 **Fig. 2** FT-IR spectra in the 1,600-1,700  $\text{cm}^{-1}$  range, after subtraction of the NaCl  
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25 550 solution (0.4 g/100 mL) spectrum and second derivative transformation: (a) dispersions  
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27 551 of whey protein concentrate (WP) at 5 g/100 mL (black dashed line) and 10 g/100 mL  
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29 552 (black solid line), and WP-based double emulsions (grey lines); (b) dispersions of egg  
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31 553 white powder (EW) at 5 g/100 mL (black dashed line) and 10 g/100 mL (black solid  
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33 554 line), and EW-based double emulsions (grey lines, black dotted line, black dash-dot  
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35 555 line). In Fig. 2b samples EW\_00\_20\_60 (black dotted line) and EW\_00\_40\_60 (black  
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37 556 dash-dot line) are highlighted  
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45 558 **Fig. 3** Results of Principal Component Analysis carried out on FT-IR reduced spectra  
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47 559 of double emulsions after subtraction of the NaCl solution (0.4 g/100 mL) spectrum  
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49 560 and second derivative transformation: (a) **first (PC1) vs. second (PC2) principal**  
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51 561 **component** score plot of double emulsion samples containing whey proteins (squares)  
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53 562 or egg white powder (circles); (b) PC1 (black line) and PC2 (grey line) loading plots  
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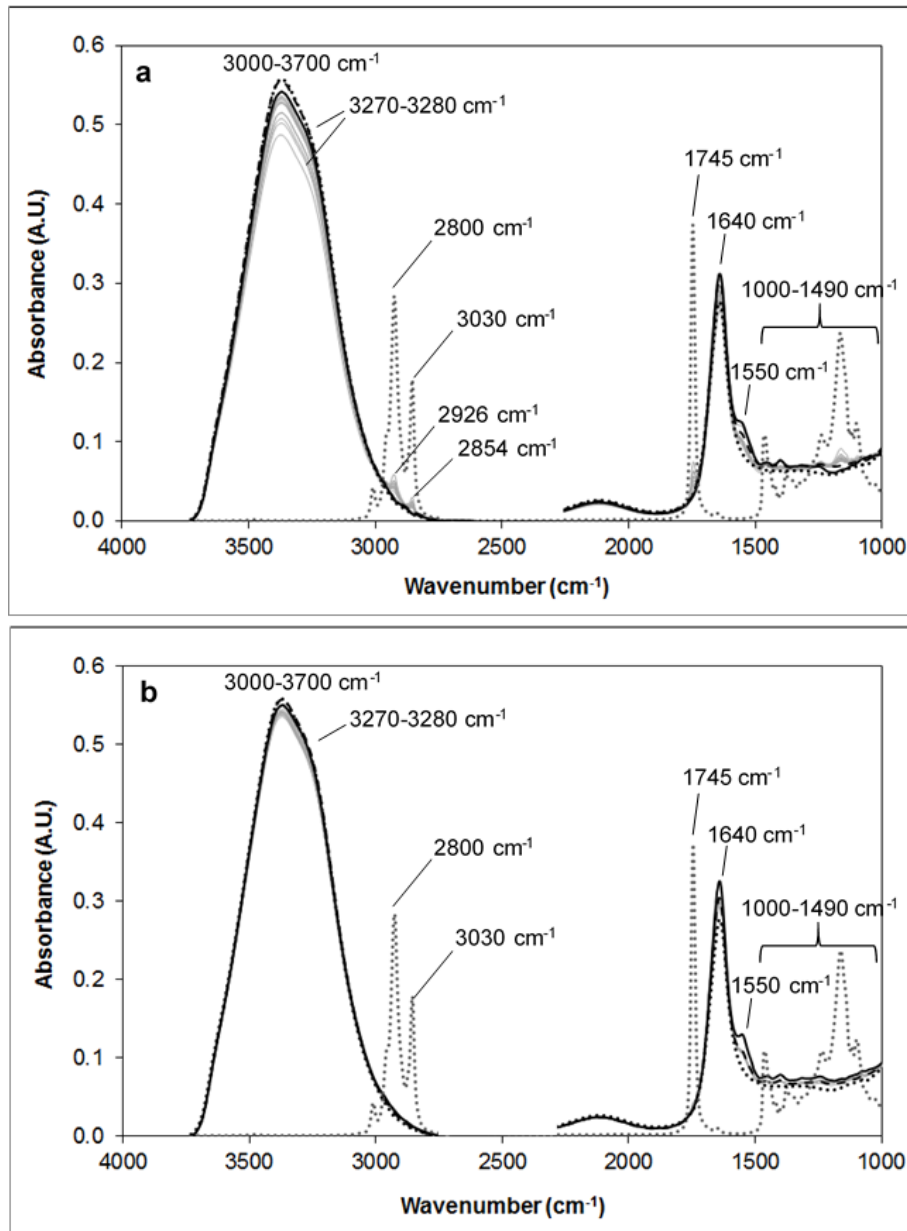
1 564 **Fig. 4** Response surfaces of double emulsion characteristics: (a) yield (constant  
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3 565 factors: egg white protein; 5 g/100 mL protein in inner water phase); (b) consistency  
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5 566 coefficient K (constant factors: egg white powder; 5 g/100 mL protein powder in  
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7 567 inner water phase); (c) apparent viscosity (constant factors: whey protein concentrate;  
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9 568 5 g/100 mL protein powder in inner water phase); (d) apparent viscosity (constant  
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11 569 factors: egg white powder; 5 g/100 mL protein powder in inner water phase).  $W_1$  (%),  
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13 570 inner water phase volume percentage;  $W_1/O$  (%), primary emulsion volume  
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15 571 percentage  
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23 **Figure captions (color version for online only)**

24  
25 574 **Fig. 1** FT-IR spectra after interval reduction and baseline offset correction of: (a) corn  
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27 575 oil (yellow line), dispersions of whey protein concentrate (WP) at 5 g/100 mL (black  
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29 576 dashed line) and 10 g/100 mL (black line), NaCl solution at 0.4 g/100 mL (black dotted  
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31 577 line), and WP-based double emulsions (blue lines); (b) corn oil (yellow line),  
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33 578 dispersions of egg white powder (EW) at 5 g/100 mL (black dashed line) and 10 g/100  
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35 579 mL (black line), NaCl solution at 0.4 g/100 mL (black dotted line), and EW-based  
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37 580 double emulsions (red lines)  
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45 582 **Fig. 2** FT-IR spectra in the 1,600-1,700  $\text{cm}^{-1}$  range, after subtraction of the NaCl  
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47 583 solution (0.4 g/100 mL) spectrum and second derivative transformation: (a) dispersions  
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49 584 of whey protein concentrate (WP) at 5 g/100 mL (black dashed line) and 10 g/100 mL  
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51 585 (black line), and WP-based double emulsions (blue lines); (b) dispersions of egg white  
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53 586 powder (EW) at 5 g/100 mL (black dashed line) and 10 g/100 mL (black line), and EW-  
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55 587 based double emulsions (red and green lines). In Fig. 2b samples EW\_00\_20\_60 (green  
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1 588 dotted line) and EW\_00\_40\_60 (green dashed line) are highlighted  
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6 590 **Fig. 3** Results of Principal Component Analysis carried out on FT-IR reduced spectra  
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8 591 of double emulsions after subtraction of the NaCl solution (0.4 g/100 mL) spectrum  
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10 592 and second derivative transformation: (a) first (PC1) vs. second (PC2) principal  
11 593 component score plot of double emulsion samples containing whey proteins (blue  
12 594 squares) or egg white powder (red circles); (b) PC1 (black line) and PC2 (grey line)  
13 595 loading plots  
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18 597 **Fig. 4** Response surfaces of double emulsion characteristics: (a) yield (constant  
19 598 factors: egg white protein; 5 g/100 mL protein in inner water phase); (b) consistency  
20 599 coefficient K (constant factors: egg white powder; 5 g/100 mL protein powder in  
21 600 inner water phase); (c) apparent viscosity (constant factors: whey protein concentrate;  
22 601 5 g/100 mL protein powder in inner water phase); (d) apparent viscosity (constant  
23 602 factors: egg white powder; 5 g/100 mL protein powder in inner water phase).  $W_1$  (%),  
24 603 inner water phase volume percentage;  $W_1/O$  (%), primary emulsion volume  
25 604 percentage  
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**Fig. 1**



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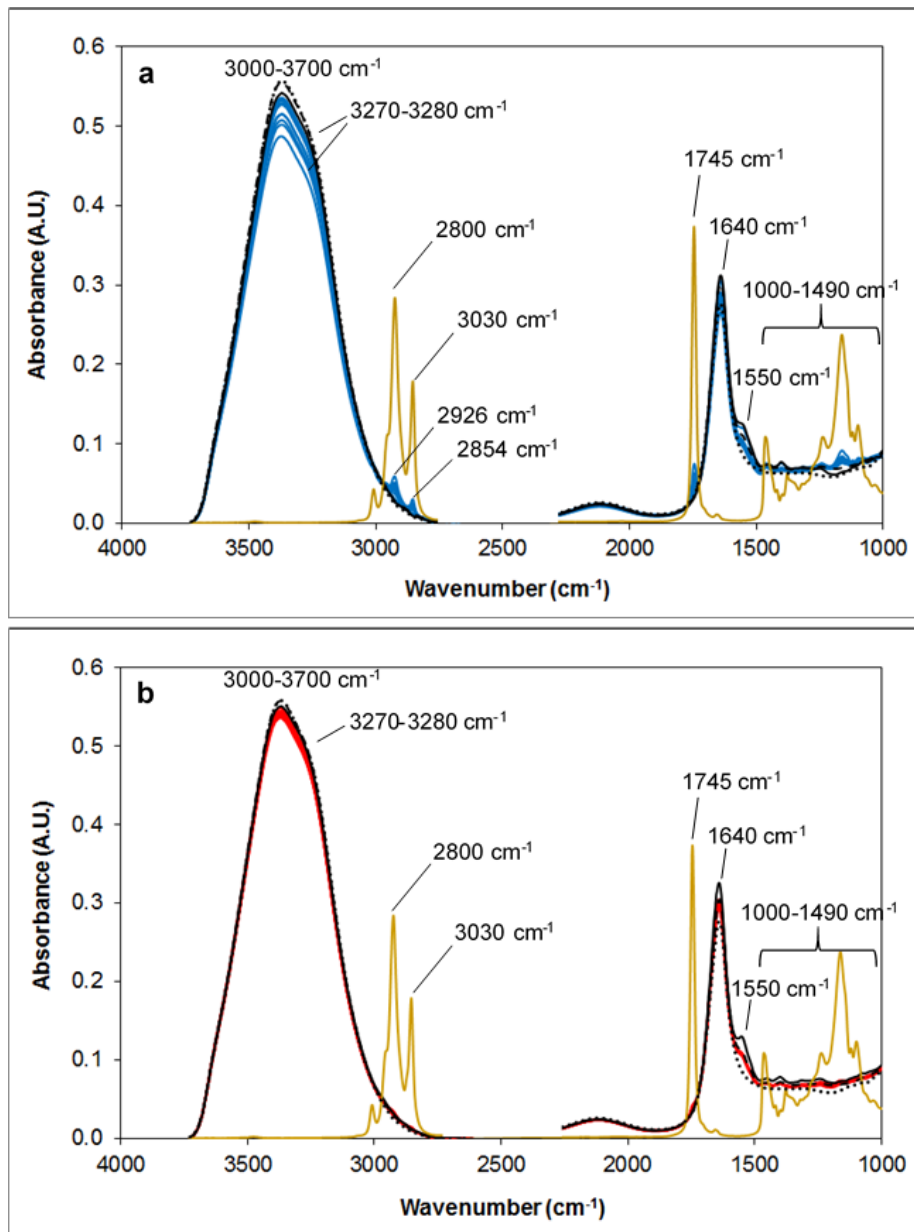


Fig. 2

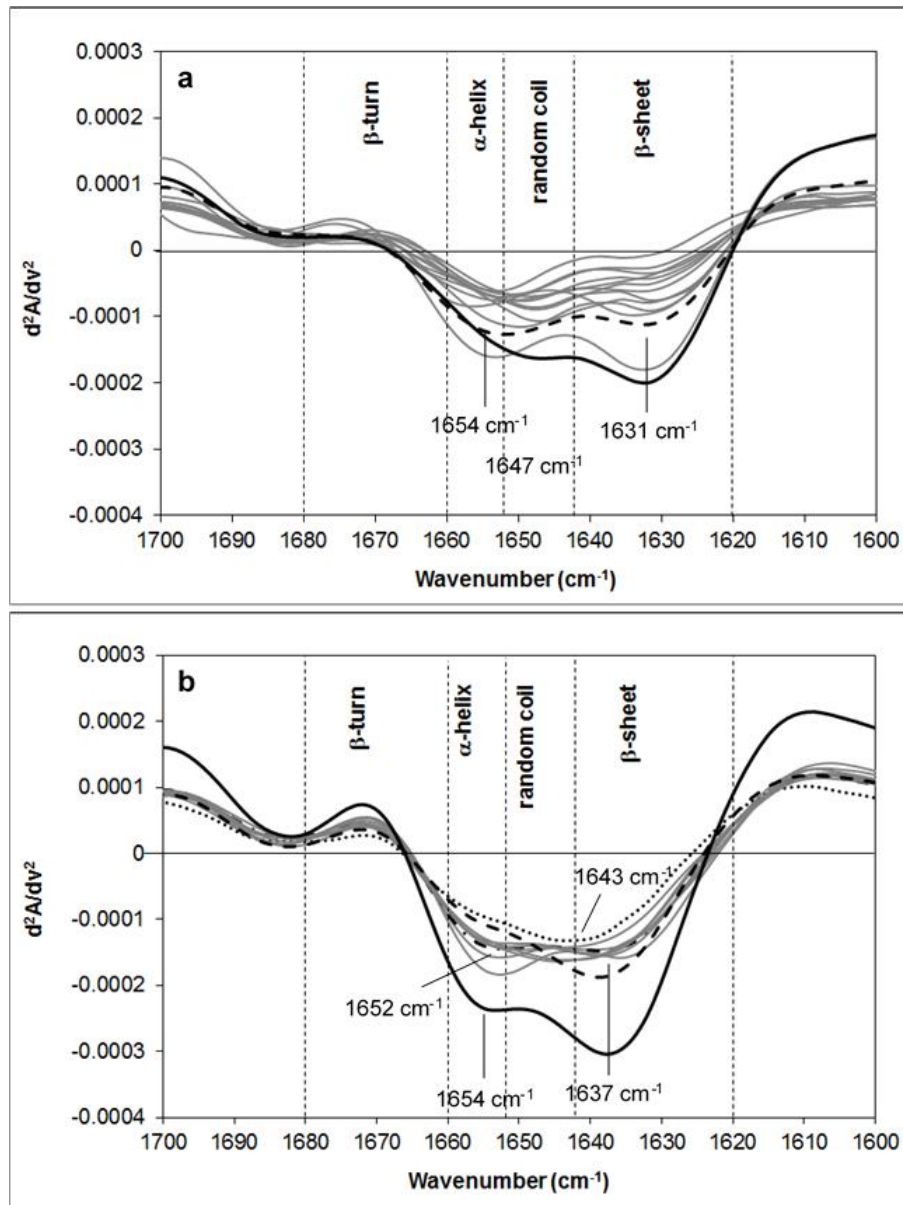


Fig. 2 Color version for online only

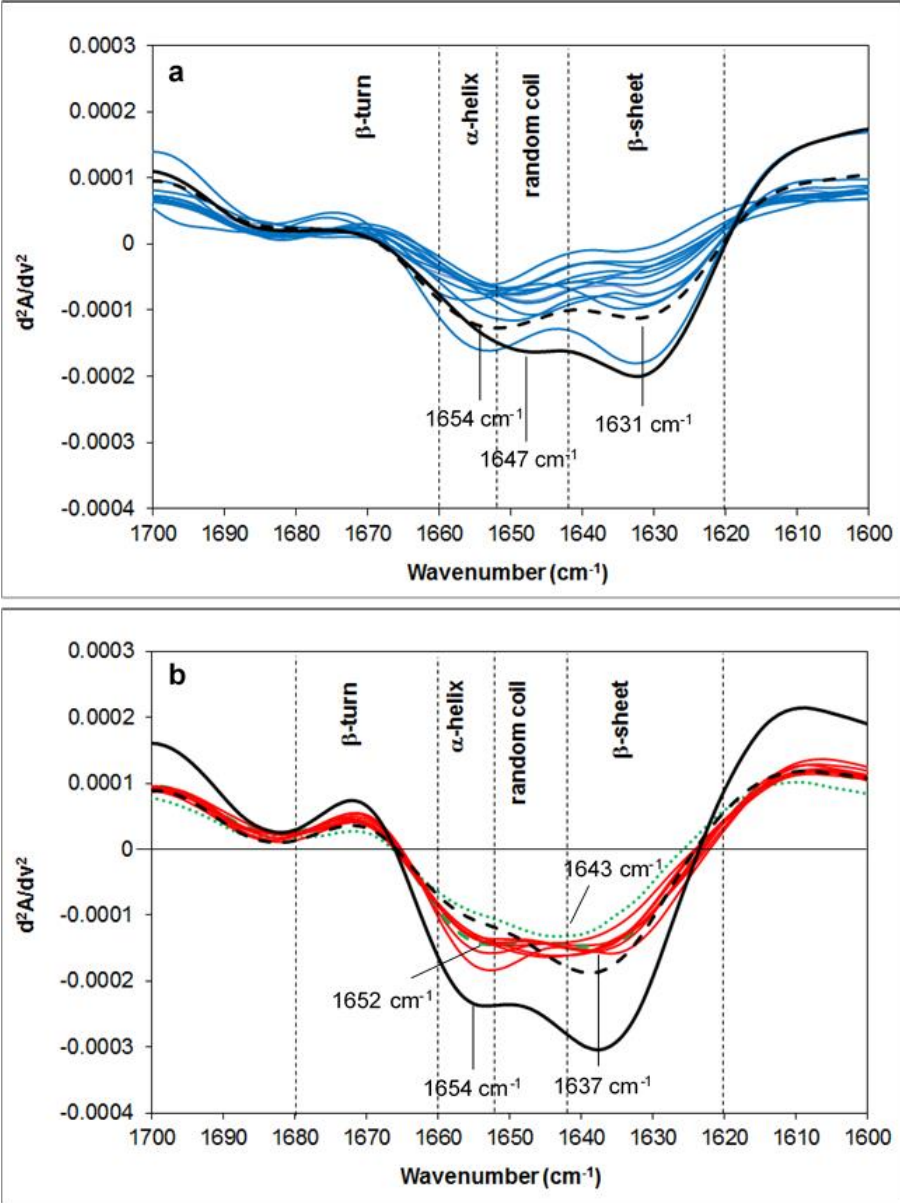




Fig. 3

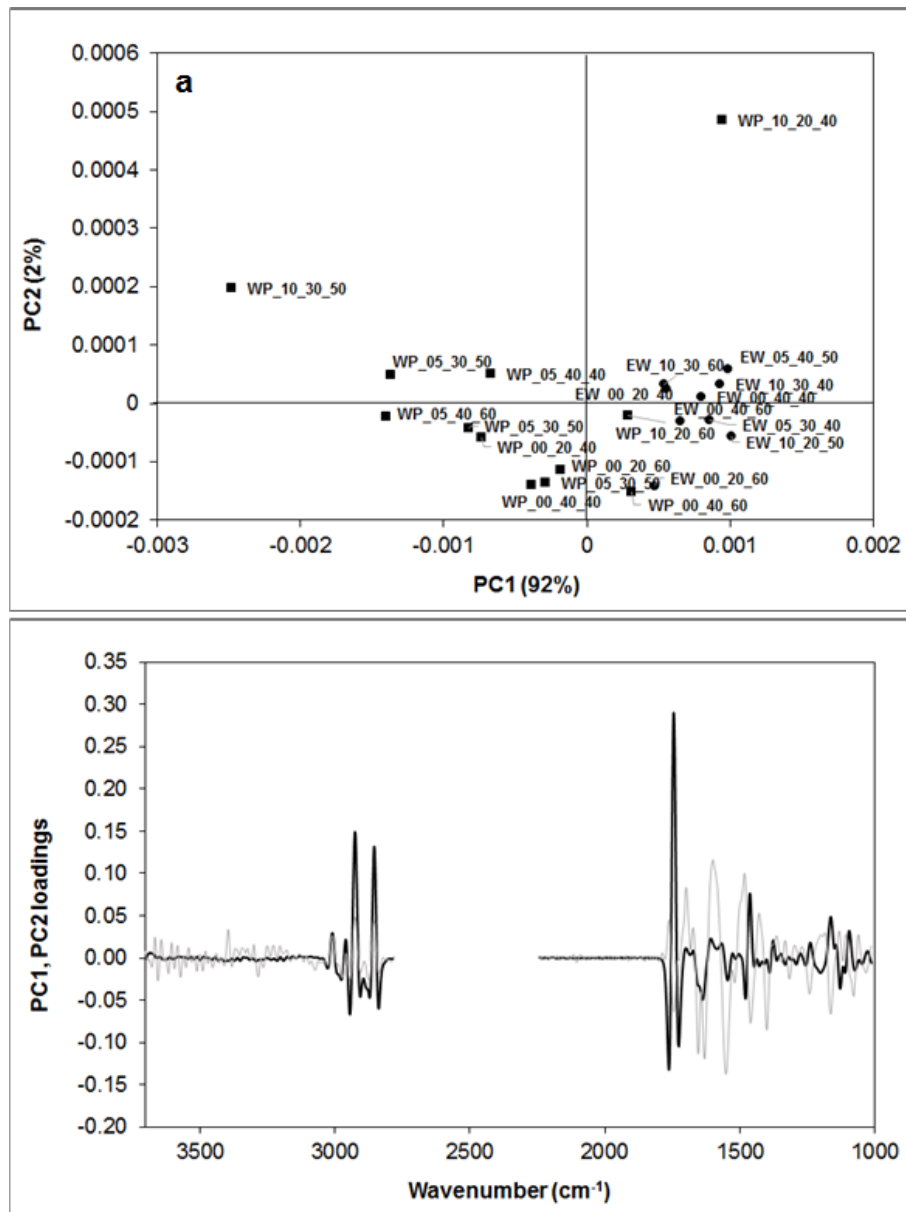
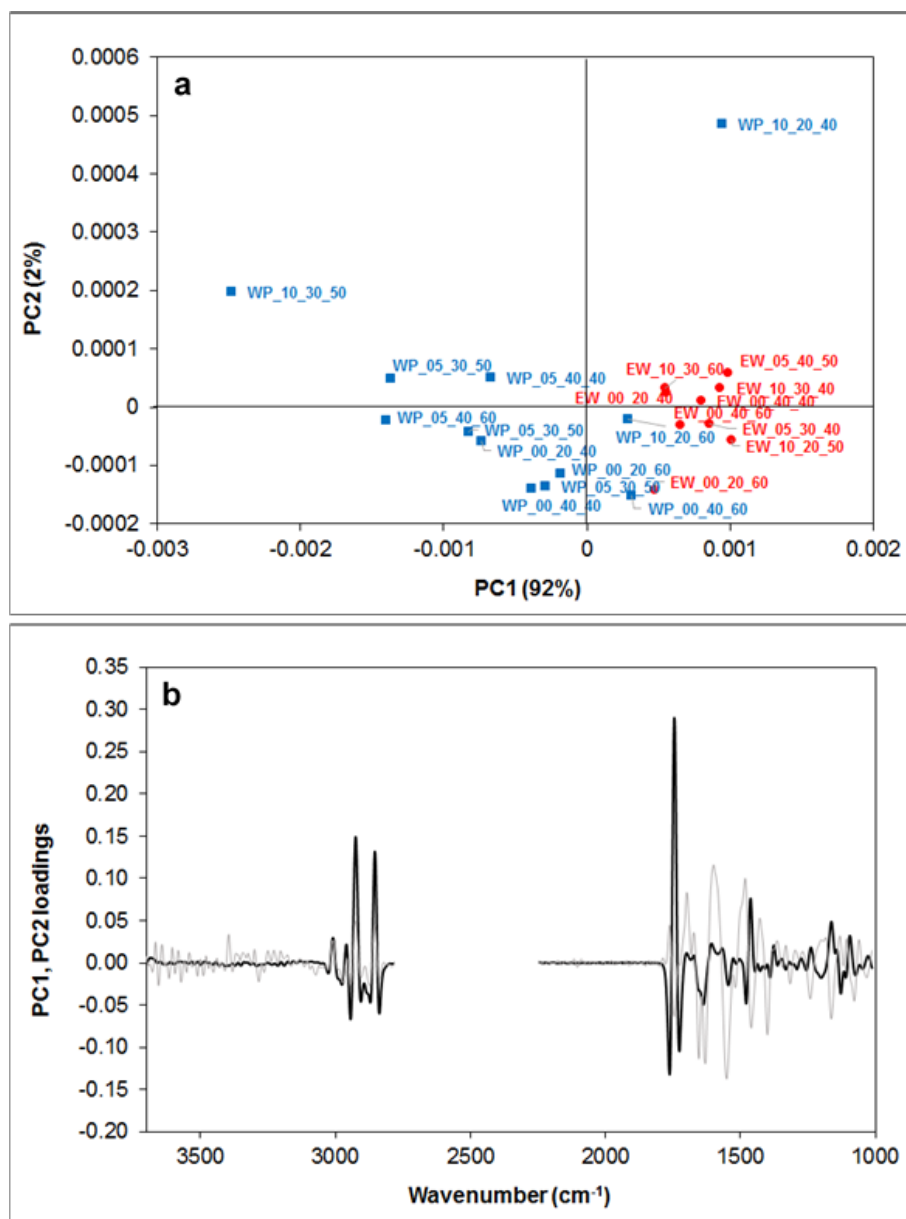
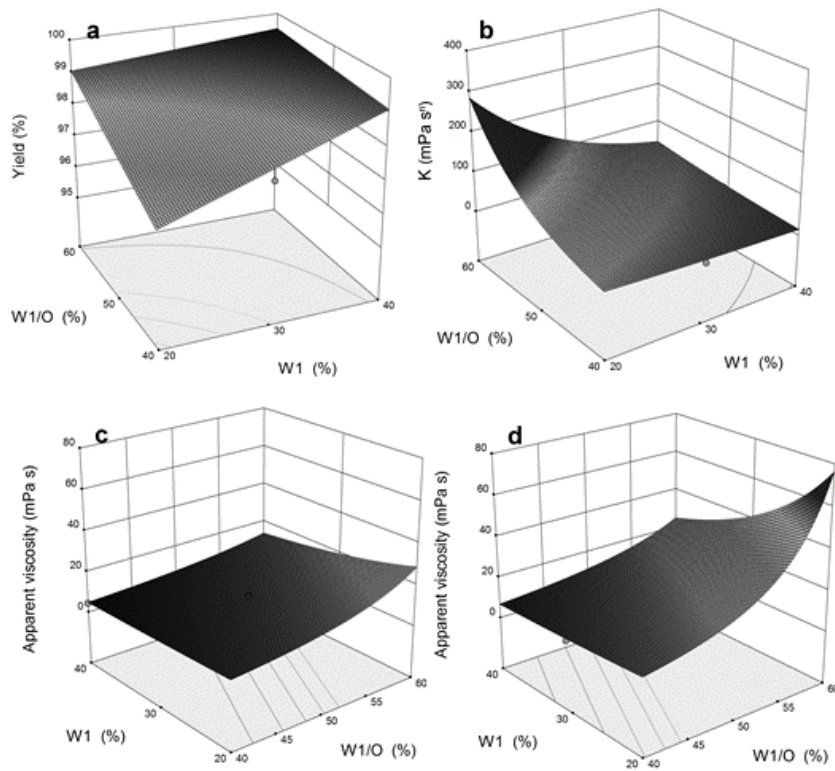
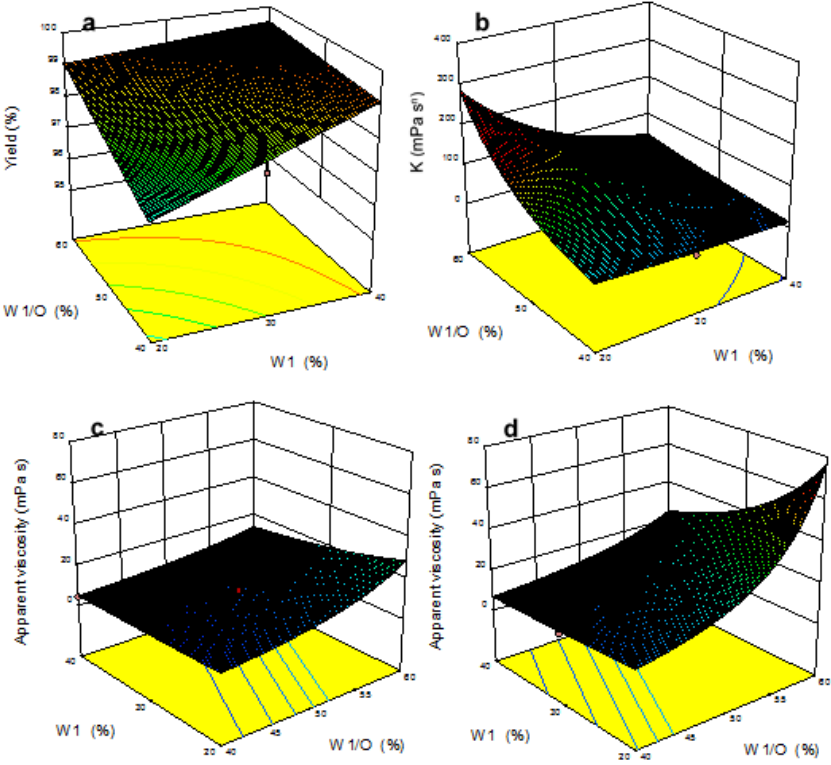


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**Fig. 4.**

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**Table 1.** D-optimal experimental design matrix developed to investigate the effects on the properties of double emulsions of four experimental factors: protein powder type (whey protein concentrate, WP; egg white powder, EW) and concentration in the internal water phase ( $W_1$ ), volume percentage of  $W_1$  in the primary emulsion ( $W_1/O$ ), and volume percentage of  $W_1/O$  in the final double emulsion ( $W_1/O/W_2$ ). Sample identification, factor levels, and theoretical composition of experimental emulsions in run order.

Sample	Experimental factors				Theoretical composition		
	Protein powder type	Protein powder concentration in $W_1$ (g/100 mL)	$W_1$ in $W_1/O$ (%)	$W_1/O$ in $W_1/O/W_2$ (%)	Water (mL/100 mL)	Proteins (g/100 mL)	Oil (mL/100 mL)
EW_00_40_40	EW	0	40	40	72.9	2.3	24
EW_10_30_40	EW	10	30	40	67.7	3.3	28
WP_05_40_40	WP	5	40	40	72.1	3.0	24
WP_10_20_60	WP	10	20	60	48.7	2.5	48
WP_00_20_60	WP	0	20	60	49.9	1.6	48
WP_00_40_60	WP	0	40	60	61.9	1.6	36
EW_05_40_50	EW	5	40	50	66.4	2.7	30
WP_05_30_50	WP	5	30	50	61.6	2.5	35
WP_05_30_50	WP	5	30	50	61.6	2.5	35
WP_10_20_40	WP	10	20	40	64.1	3.0	32
EW_00_20_40	EW	0	20	40	64.9	2.3	32
WP_10_30_50	WP	10	30	50	60.9	3.1	35
EW_10_30_60	EW	10	30	60	54.1	3.0	42
EW_00_40_60	EW	0	40	60	61.9	1.6	36
EW_10_20_50	EW	10	20	50	56.4	2.7	40
WP_05_30_50	WP	5	30	50	61.6	2.5	35
WP_05_40_60	WP	5	40	60	60.7	2.5	36
EW_05_30_40	EW	5	30	40	68.3	2.8	28
WP_00_20_40	WP	0	20	40	64.9	2.3	32
WP_00_40_40	WP	0	40	40	72.9	2.3	24
EW_00_20_60	EW	0	20	60	49.9	1.6	48

**Table 2.** Characteristics of double emulsions (mean  $\pm$  standard deviation values<sup>a</sup>) formulated according to the D-Optimal experimental design described in Table 1.

Sample	Yield (%)	Apparent viscosity (mPa s)	K (mPa s <sup>n</sup> )	n	CS (%)
EW_00_40_40	99.23 $\pm$ 0.03	6.8 $\pm$ 0.1	16.4 $\pm$ 1.1	0.847 $\pm$ 0.008	50 $\pm$ 1
EW_10_30_40	98.84 $\pm$ 0.06	13.7 $\pm$ 0.4	43.8 $\pm$ 5.1	0.809 $\pm$ 0.016	73 $\pm$ 1
WP_05_40_40	98.39 $\pm$ 0.13	4.3 $\pm$ 0.1	2.4 $\pm$ 0.1	1.105 $\pm$ 0.003	40 $\pm$ 1
WP_10_20_60	98.53 $\pm$ 0.05	32.7 $\pm$ 0.3	47.2 $\pm$ 0.4	0.935 $\pm$ 0.001	80 $\pm$ 1
WP_00_20_60	99.31 $\pm$ 0.06	28.7 $\pm$ 0.5	39.7 $\pm$ 1.8	0.943 $\pm$ 0.005	100 $\pm$ 1
WP_00_40_60	99.45 $\pm$ 0.06	13.0 $\pm$ 0.3	13.1 $\pm$ 0.5	0.999 $\pm$ 0.002	80 $\pm$ 1
EW_05_40_50	98.69 $\pm$ 0.05	9.9 $\pm$ 0.1	17.5 $\pm$ 0.3	0.900 $\pm$ 0.001	80 $\pm$ 1
WP_05_30_50	98.88 $\pm$ 0.05	9.6 $\pm$ 0.3	7.5 $\pm$ 0.2	1.045 $\pm$ 0.001	100 $\pm$ 1
WP_05_30_50	99.16 $\pm$ 0.18	11.1 $\pm$ 0.2	9.2 $\pm$ 0.2	1.032 $\pm$ 0.001	89 $\pm$ 1
WP_10_20_40	96.60 $\pm$ 0.11	9.1 $\pm$ 0.1	7.0 $\pm$ 0.2	1.046 $\pm$ 0.001	75 $\pm$ 1
EW_00_20_40	97.23 $\pm$ 0.04	8.6 $\pm$ 0.3	19.6 $\pm$ 2.7	0.857 $\pm$ 0.018	60 $\pm$ 1
WP_10_30_50	98.74 $\pm$ 0.10	11.0 $\pm$ 0.4	8.3 $\pm$ 0.4	1.050 $\pm$ 0.002	89 $\pm$ 1
EW_10_30_60	99.30 $\pm$ 0.02	63.1 $\pm$ 3.7	3089 $\pm$ 440	0.357 $\pm$ 0.061	100 $\pm$ 1
EW_00_40_60	99.45 $\pm$ 0.02	18.1 $\pm$ 0.3	85.4 $\pm$ 0.8	0.731 $\pm$ 0.001	90 $\pm$ 1
EW_10_20_50	95.47 $\pm$ 0.08	22.2 $\pm$ 0.9	142 $\pm$ 13	0.677 $\pm$ 0.008	90 $\pm$ 1
WP_05_30_50	98.79 $\pm$ 0.03	11.4 $\pm$ 1.2	8.6 $\pm$ 0.6	1.039 $\pm$ 0.005	90 $\pm$ 1
WP_05_40_60	99.21 $\pm$ 0.02	9.7 $\pm$ 0.2	7.4 $\pm$ 0.2	1.049 $\pm$ 0.001	100 $\pm$ 1
EW_05_30_40	97.62 $\pm$ 0.02	7.5 $\pm$ 0.4	13.0 $\pm$ 2.5	0.905 $\pm$ 0.025	63 $\pm$ 4
WP_00_20_40	95.83 $\pm$ 0.23	5.2 $\pm$ 0.3	3.3 $\pm$ 0.2	1.082 $\pm$ 0.002	88 $\pm$ 1
WP_00_40_40	99.01 $\pm$ 0.06	3.8 $\pm$ 0.1	2.5 $\pm$ 0.1	1.103 $\pm$ 0.001	47 $\pm$ 1
EW_00_20_60	98.61 $\pm$ 0.05	43.1 $\pm$ 0.2	545 $\pm$ 2	0.559 $\pm$ 0.001	100 $\pm$ 1

<sup>a</sup>Number of replicates (n): yield, n=4; apparent viscosity, K, n, and CS, n=3.  
K, coefficient of consistency; n, flow behavior index; CS, creaming stability.

**Table 3.** Multivariate model coefficient values (in terms of coded factors) and results of one-way analysis of variance for the D-optimal design developed to study double emulsions structured by whey protein concentrate and egg white powder.

Response variable	$\beta_0$	$\beta_1$	$\beta_2$	$\beta_3$	$\beta_4$	$\beta_{23}$	$\beta_{24}$	$\beta_{34}$	$R^2$	Adj. $R^2$	Pred. $R^2$	Adequate precision	LOF
Yield (%)	9.86E+01	6.94E-02 <sup>ns</sup>	7.87E-02 <sup>ns</sup>	6.79E-01 <sup>***</sup>	6.51E-01 <sup>***</sup>			-4.42E-01 <sup>*</sup>	0.791 <sup>***</sup>	0.723	0.561	9.873	0.114 <sup>ns</sup>
1/ $\sqrt{\text{Apparent viscosity}}$ (1/ $\sqrt{\text{mPa s}}$ )	2.89E-01	-3.43E-02 <sup>***</sup>	-2.62E-02 <sup>**</sup>	4.09E-02 <sup>***</sup>	-9.04E-02 <sup>***</sup>				0.934 <sup>***</sup>	0.920	0.882	24.97	0.201 <sup>ns</sup>
Log K [Log(mPa s <sup>n</sup> )]	1.29E+01	3.92E-01 <sup>***</sup>	5.14E-04 <sup>ns</sup>	-2.86E-01 <sup>***</sup>	3.60E-01 <sup>***</sup>	-1.32E-01 <sup>*</sup>	-1.08E-01 <sup>*</sup>	-1.11E-01 <sup>*</sup>	0.967 <sup>***</sup>	0.946	0.871	23.55	0.127 <sup>ns</sup>
n	9.11E-01	-1.24E-01 <sup>***</sup>	-7.20E-003 <sup>ns</sup>	3.36E-02 <sup>*</sup>	-5.92E-02 <sup>**</sup>				0.915 <sup>***</sup>	0.891	0.836	18.24	0.010 <sup>**</sup>
CS (%)	8.31E+01	-2.04E+00 <sup>ns</sup>	4.92E+00 <sup>ns</sup>	-9.29E+00 <sup>**</sup>	18.1E+00 <sup>***</sup>				0.806 <sup>***</sup>	0.751	0.652	12.46	0.298 <sup>ns</sup>

$\beta_0$ , model intercept;  $\beta_1$ ,  $\beta_2$ ,  $\beta_3$ ,  $\beta_4$ , linear coefficients for protein powder type, protein powder concentration, volume percentage of inner water phase, and volume percentage of primary emulsion, respectively;  $\beta_{23}$ ,  $\beta_{24}$ ,  $\beta_{34}$ , interaction coefficients.  $R^2$ , coefficient of determination; Adj.  $R^2$ , adjusted  $R^2$ ; Pred.  $R^2$ , predicted  $R^2$ ; LOF, lack of fit (p-value). Significance levels: <sup>ns</sup>, not significant; <sup>\*</sup>  $p \leq 0.05$ ; <sup>\*\*</sup>  $p \leq 0.01$ ; <sup>\*\*\*</sup>  $p \leq 0.001$

# Graphical abstract

