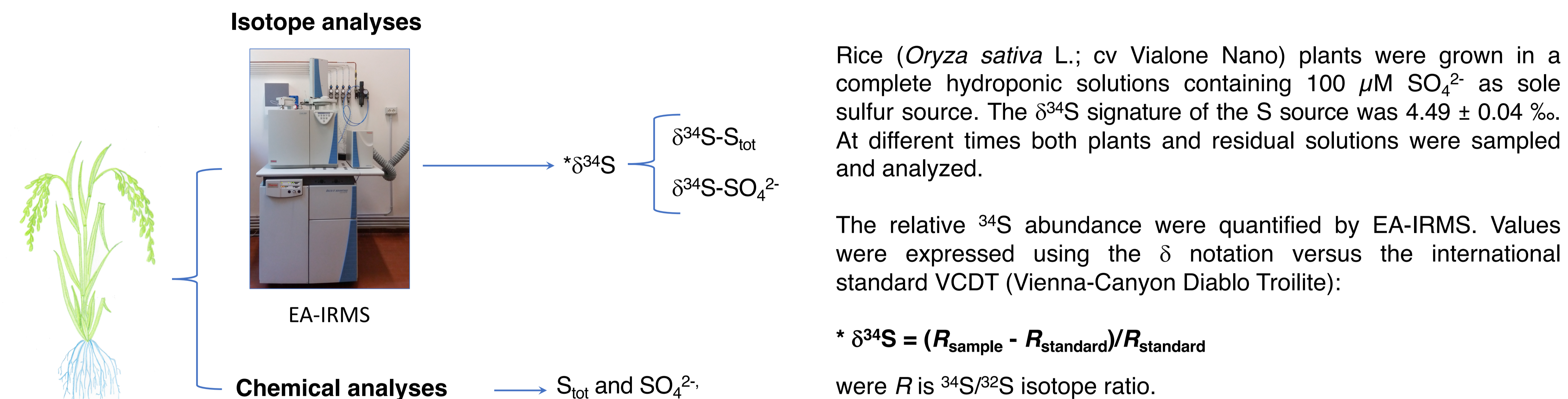


Viviana Cavallaro, Mariachiara Caschetto, Moez Maghrebi, Gian Attilio Sacchi, Fabio Francesco Nocito
 Dipartimento di Scienze Agrarie e Ambientali – Produzione, Territorio, Agroenergia
 Università degli Studi di Milano – ITALY

Background

Four stable isotopes of sulfur exist (^{32}S , ^{33}S , ^{34}S , ^{36}S) whose natural isotopic percentage abundances are 0.94499, 0.0075, 0.0425 and 0.0001 atom fraction, respectively. The most abundant isotopes – ^{32}S and ^{34}S – are now commonly measured using elemental analyzers coupled with isotope ratio mass spectrometers (EA-IRMS). Such an approach is based on the complete transformation of total S to SO_2 , which is subsequently analyzed by the mass spectrometer with regards to masses 64 ($^{32}\text{S}^{16}\text{O}_2$) and 66 ($^{34}\text{S}^{16}\text{O}_2$ or $^{32}\text{S}^{16}\text{O}^{18}\text{O}$) atomic mass units. S stable isotopes have been used to trace the movements of the related compounds in plants, in testing S flux models, and in identifying and determining the impact of natural and anthropogenic S sources on the environment. However, the isotope technique applied for S metabolism investigations, as well as for sulfate transport and allocation within the plants, is limited by our current knowledge of the potential $^{32}\text{S}/^{34}\text{S}$ isotope discrimination that may occur during both S metabolism and sulfate transport.

Experimental design and methods



Results and discussion

The $\delta^{34}\text{S}$ signature of the total biomass produced by a plant generally reflects that of the available sulfate in the soil solution, indicating the fractionation against ^{34}S during sulfate acquisition negligible. However, our preliminary results (Fig. 1) revealed some disparity amongst $\delta^{34}\text{S}$ value of the S source and the values of total S measured in rice total biomass, suggesting that substantial $^{32}\text{S}/^{34}\text{S}$ isotope effects occurred during sulfate uptake. Moreover, the $\delta^{34}\text{S}\text{-SO}_4^{2-}$ values measured in root and shoot were significantly different, indicating that other fractionation events may occur during sulfate translocation and/or assimilation.

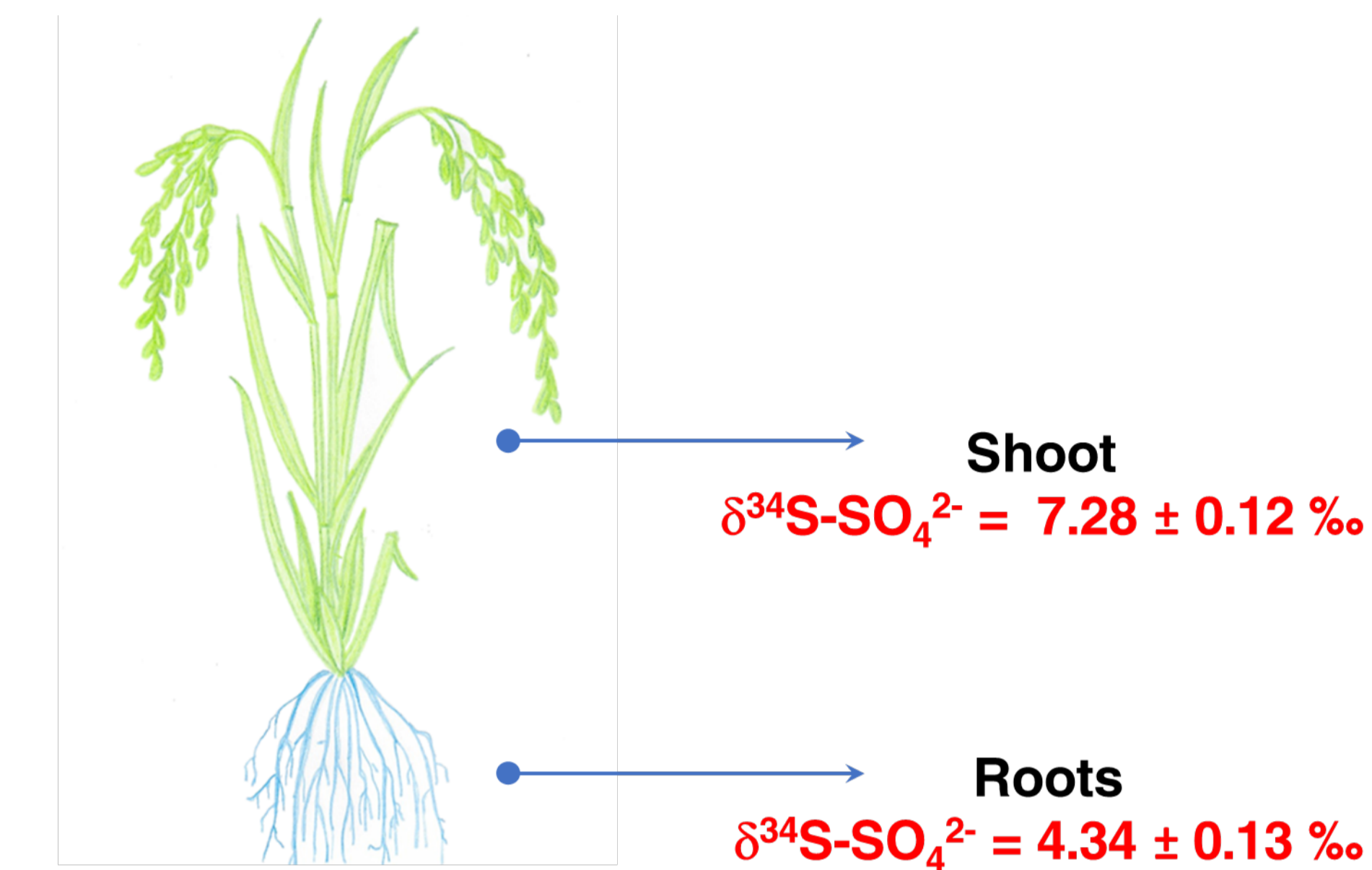
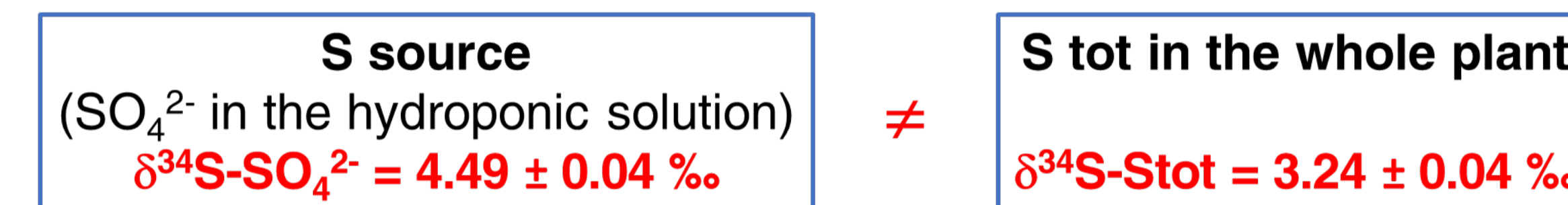


Figure 1. Natural isotope composition ($\delta^{34}\text{S}$) of total sulfur (Stot), measured in the whole plants, and sulfate, measured in roots and shoot of rice plants grown in hydroponic solution.



We approached the topic of $^{32}\text{S}/^{34}\text{S}$ fractionation in rice by investigating the hypothesis that the $\delta^{34}\text{S}\text{-tot}$ of the whole plant could be determined by the activities of the sulfate transporters involved in sulfate uptake (i.e. **sulfate transporters discriminate between ^{32}S and ^{34}S**). In particular, we considered a closed system in which a limited amount of substrate (i.e., the sulfate ions in the nutrient solution) is continuously removed from the solution - by the activity of the high-affinity sulfate transporters of the roots - and converted in a final product (i.e., the Stot of the plant).

Substrate (external SO_4^{2-}) \rightarrow Product (rice Stot)



In such a system, if fractionation against $^{34}\text{S}\text{-SO}_4^{2-}$ occurs, the $\delta^{34}\text{S}\text{-Stot}$ of the plant (product) is expected to be less than the $\delta^{34}\text{S}\text{-SO}_4^{2-}$ of the external solution (substrate). Moreover, fractionation will also cause the $\delta^{34}\text{S}\text{-SO}_4^{2-}$ of the remaining substrate to increase over the time, and thus a corresponding increase in the $\delta^{34}\text{S}\text{-Stot}$ of the product is expected.

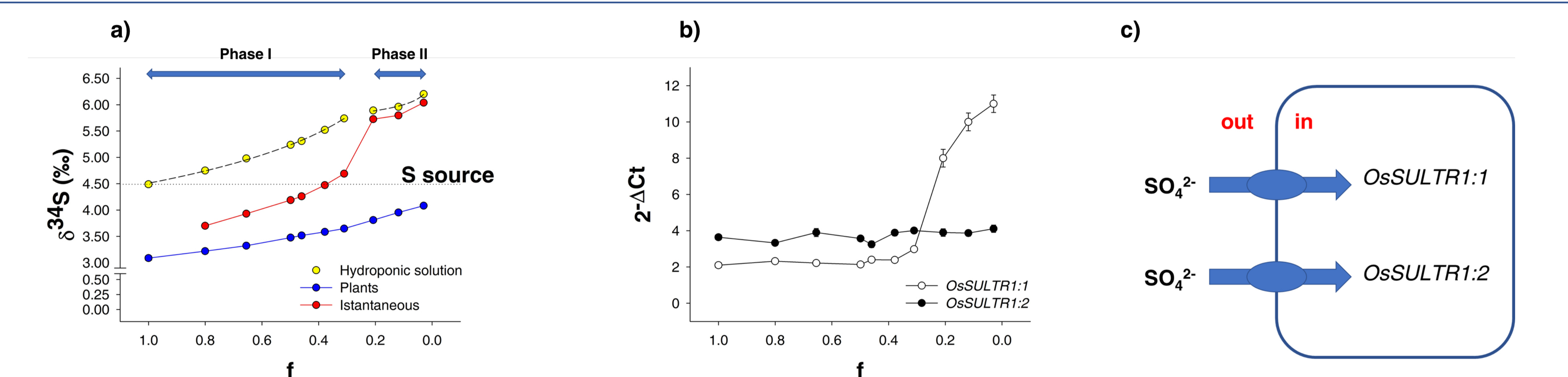


Figure 2. a) S isotope dynamics in the closed system (hydroponic solution + plants). b) Dynamics of *OsSULTR1:1* and *OsSULTR1:2* transcript in the roots. c) Main sulfate transporter genes involved in root sulfate uptake.

The $\delta^{34}\text{S}\text{-SO}_4^{2-}$ value measured for the sulfate ions in the external medium increased over the time with respect to initial $\delta^{34}\text{S}\text{-SO}_4^{2-}$ of the source (Fig. 2a). It is worth to note that – because of mass balance – the $\delta^{34}\text{S}\text{-Stot}$ of the rice plants tends to $\delta^{34}\text{S}\text{-SO}_4^{2-}$ of the original source as the sulfate concentration in the external medium approaches zero. Data analysis also reveals that a dual phase Rayleigh fractionation occurred during root sulfate absorption. In the first phase of the process ($f > 0.3$) a significant isotope fractionation against $^{34}\text{S}\text{-SO}_4^{2-}$ occurred ($\epsilon = 1.02 \text{ ‰}$), whilst in the final part of the process ($f < 0.3$) a less evident isotope effect ($\epsilon = 0.16 \text{ ‰}$) was associate to sulfate uptake (please, see the difference between $\delta^{34}\text{S}\text{-SO}_4^{2-}$ in hydroponic solution and instantaneous products). Transcriptional analysis clearly shows that the transition from phase I to phase II was associated to changes in the ratio between the transcript levels of *OsSULTR1:1* and *OsSULTR1:2*, the two main genes involved in root sulfate uptake (Fig. 2b,c).