

Ferritin nanocages: a biological platform for drug delivery, imaging and theranostics in cancer

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KEYWORDS: protein-based nanocages, ferritin, drug delivery, imaging, cancer.

Abstract

Nowadays cancer represents a prominent challenge in clinics. Main achievements in cancer management would be the development of highly accurate and specific diagnostic tools for early detection of cancer onset, and the generation of smart drug delivery systems for targeted chemotherapy release in cancer cells. In this context, protein-based nanocages hold a tremendous potential as devices for theranostics purposes. In particular, ferritin has emerged as an excellent and promising protein-based nanocage thanks to its unique architecture, surface properties and high biocompatibility. By exploiting natural recognition of the Transferrin Receptor 1, which is overexpressed on tumor cells, ferritin nanocages may ensure a proper drug delivery and release. Moreover, researchers have applied surface functionalities on ferritin cages for further providing active tumor targeting. Encapsulation strategies of non metal-containing drugs within ferritin cages have been explored and successfully performed with encouraging results. Various preclinical studies have demonstrated that

nanoformulation within ferritin nanocages significantly improved targeted therapy and accurate imaging of cancer cells. Aims of this review are to describe structure and functions of ferritin nanocages, and to provide an overview about the nanotechnological approaches implemented for applying them to cancer diagnosis and treatment.

1. Introduction

Cancer is a leading cause of death worldwide, accounting for 8.2 million deaths in 2012 [1]. A main goal of current cancer research is early recognition of cancer cells and selective treatment, before the cancer can progress to an unfavourable stage. Based on this rationale, the aim of several modern therapeutic approaches for cancer is to selectively remove the tumor before it evolves into its superior stages, and the eventual onset of metastases. For this scenario to be successful there is a growing need to develop: 1) novel and more sensitive diagnostic tools to improve screening accuracy and the detection rate of malignancies, staging and follow up settings, and to achieve less extensive surgery or medical therapy; 2) new drug formulations with enhanced efficacy toward cancer cells and reduced toxicity toward healthy cells [2].

Nanotechnology holds great clinical potential in reaching these major goals and a number of nanodevices have been designed to specifically target cancer cells. These nanodevices can be loaded with different kinds of chemotherapeutics and/or contrast agents, and provide unmatched promise for development of theranostic devices for both cancer diagnosis and treatment. Indeed, nanoparticles have demonstrated their potential to solve crucial issues involving bioavailability, tissue penetration and circulation time [3]. Several nanoparticles constructed from metals, semi-conductors or polymers, have been tested in preclinical studies [4], but only a few of them have been clinically approved due to their toxicity, immunogenicity and sequestration by the reticulo-endothelial system [5]. Therefore, much effort has been directed into the design and synthesis of chemically engineered nanostructures that are biologically compatible.

Naturally occurring nanoparticles have existed for millions of years, and they include a wide variety of nanostructures, which may be formed by inorganic (i.e. magnetosomes) or organic materials. These nanoparticles can be intracellular or extracellular and may have different biological roles [6]. Despite these differences, natural nanoparticles are charming from the biomedical point of view due to their uniform structure, low toxicity, and ability to evade the immune system. Moreover, they are easily degraded after fulfilling their function [6,7]. These features place them ahead of inorganic or synthetic materials in clinical translation [7]. Biological nanoparticles are assembled from molecules or atoms synthesized in a biological system with sizes ranging from 1 to 100 nm. They include magnetosomes, lipoproteins, viruses, exosomes and ferritins [6].

Magnetosomes are specialized organelles with narrow size (50-70 nm in diameter), uniform morphology and low toxicity; they have evolved from magnetotactic bacteria and are composed of a lipid bilayer surrounding magnetic iron-containing minerals [8-10]. Lipoproteins are self-assembling structures that contain phospholipids, non-esterified cholesterol and specialized proteins, forming a spherical or discoidal shell (from 7 to >80 nm), which surrounds a core of non-polar lipids, triacylglycerols and esterified cholesterols [11, 12]. Exosomes are nanometer-sized structures secreted by cells and consist of an external lipid bilayer, which offers a non-immunogenic and non-toxic mechanism for the delivery of targeted proteins or nucleic acids, protecting them from degradation and macrophage endocytosis [13-16]. Viruses and virus-like particles may also be considered to be biological nanoparticles. They include a wide variety of sizes and morphologies with defined geometries, uniform shape and a robust protein shell. They are suitable for chemical- and bio-conjugation. Although their capsids are stable over a wide range of pH and temperatures, they are difficult to produce in bulk, and this represents a severe limitation for their use in nanotechnology [17].

Among all classes of biological nanoparticles here we focus on ferritins, which are iron storage and transport proteins found in most living organisms [18]. Recombinant ferritin provides a central cavity, which can be efficiently loaded with transition metals, drugs, fluorescent molecules or contrast agents

[19, 20]. Since it has a uniform cage, ferritin allows the precise control of the amount of encapsulated molecules, which is a critical feature in defining drug dosage. Moreover, the protein shell of ferritins can be easily modified either chemically or genetically to introduce different functionalities [21, 22]. All of these attributes highlight ferritin and its derivatives as powerful systems with potential application in nanomedicine. Indeed, they have been investigated as a novel type of nano-platform for imaging and drug delivery in cancer [19, 23]. The aim of this review is to describe recent findings about ferritin nanocages' structure and function, and their biotechnological and biomedical applications for MRI, optical imaging and drug delivery in cancer. An overview of strategies used for ferritin surface modification to obtain formulations suitable for different clinical uses will be also provided.

2. Ferritin nanocages: structure and physiological role

Ferritin is probably the most studied protein after hemoglobin. It has been investigated since 1937, when Laufberger described its purification with Cadmium salts [24], and over the last decade it has attracted the interest of many nanotechnologists due to its properties [25]. It is a ubiquitous protein found in eubacteria, archea, plants and animals, but not in yeast [25]. Ferritin exists in extracellular and intracellular compartments, such as the cytosol, nucleus and mitochondria and its main roles are iron storage and homeostasis [26]. Therefore, it is one of the most highly conserved molecules.

Ferritin is a large protein of 450 kDa composed of 24 subunits that self-assemble into a spherical cage-like structure with inner and outer dimensions of 8 and 12 nm, respectively. Eukaryotes have two ferritin genes encoding the heavy (H; 21 kDa) and the light (L; 19 kDa) chains. The H-chain is responsible for the oxidation of Fe(II) to Fe(III) and includes the ferroxidase catalytic site, while the L-chain plays a role in iron nucleation [27]. H and L chains co-assemble into a 24-mer heteropolymer, where the H-chain to L-chain ratio varies according to a tissue specific distribution [25]. Ferritin protects cells from the damage caused by the Fenton reaction during oxidative stress, by sequestering Fe^{2+} , a source of toxic reactive oxygen species, and converting it into harmless Fe^{3+} , which is stored as ferrihydrite crystals. The ferritin protein cage can accommodate up to 4500 iron atoms [25]. This protective action is crucial in the nucleus, where the DNA has to be particularly preserved from iron-induced oxidative damage

[28]. Ferritin translocates into the nucleus through an ATP-dependent mechanism [28, 29], and O-glycosylation of the H-chain seems to be involved in its nuclear translocation, but a Nuclear Localization Sequence (NLS) is not required [29]. However, it was not clear whether the integrity of ferritin was maintained during translocation. Recently, the Knez group has shed light on this issue by demonstrating that ferritin nanoparticles are intact during translocation [30].

The ferritin quaternary structure has eight hydrophilic channels that seem to mediate iron transit in and out of the protein cage. There are also six hydrophobic channels, which do not seem to be involved in iron exchange although they could mediate the transit of protons [31]. In spite of the structural rigidity of ferritins in the physiological environment, the protein cages can be reversibly disassembled when the pH becomes extremely acidic (pH 2-3) or basic (pH 10-12) [27, 32]. When the pH returns to neutrality ferritin monomers are able to self-assemble in a shape memory fashion. Moreover, the ferritin cage is resistant to denaturants, including heating to high temperatures (>80 °C) [25].

Ferritin features have been extensively studied and implemented by researchers, thus allowing the use of ferritin nanocages as biological nanoparticles for nanomedical applications.

3. Surface modification of ferritin nanoparticles and cellular interactions

Extracellular ferritin interacts with cells through the receptor of transferrin 1 (TfR1), which was identified in 2010 by Seaman's group [33]. They demonstrated that TfR1 specifically binds H-ferritin, while L-ferritin shows low interaction. After binding on the cell surface, the H-ferritin-TfR1 complex is internalized, and can be found in early and recycling endosomes [33, 27], demonstrating that TfR1 coordinates the processing and the use of iron by binding both Transferrin and H-ferritin [34]. H-ferritin is also able to interact with the T-cell Immunoglobulin and Mucin domain-2 (TIM-2) [35, 36], which is overexpressed in oligodendrocytes and B-cells [37, 38], and is internalized in endosomes upon binding [39]. Otherwise, L-ferritin binds to the Scavenger Receptor Class A, Member 5 (SCARA-5) [40], which is found in macrophages and the retina. Both H and L ferritin nanoparticles have been produced via DNA-recombinant technology. H-ferritin, which naturally interacts with the

TfR1, are the most extensively studied [33]. By exploiting the TfR1 overexpression in many types of tumor cells, Fan and coworkers have developed a ferritin-based system for specifically visualizing tumors among normal cells and healthy tissues [41]. That study demonstrated that the intrinsic targeting of ferritin towards cancer cells could be used to design naturally targeted nanoparticles for theranostic purposes.

Despite the natural targeting of ferritin towards cancer cells, several research groups have modified the surface of ferritin nanocages by inserting a number of target motifs, such as antibodies, peptides and antibody fragments, in order to drive nanoparticles towards specific cells by selective recognition (Figure 1). Both lysines and cysteines on the ferritin surface have been exploited for chemical conjugation using different heterobifunctional cross-linkers with N-hydroxysuccinimide (NHS) ester and maleimide groups [42-44]. Lys and Cys residues have also been employed to chemically attach dyes, quencher molecules or polyethylene glycol (PEG) molecules [22, 45]. Most conjugation strategies have been designed and realized by genetically engineering ferritin monomers. Targeting moieties, such as the RGD peptide [46, 47], the anti-melanocyte stimulating hormone peptide [48, 45] or the extracellular domain of myelin oligodendrocyte glycoprotein (MOG) [49] have been expressed as N-terminal fusion proteins, allowing 24 targeting domains to be exposed on the ferritin cage surface, with uniform and precise orientation. In the last year, the surface conjugation of ferritin nanoparticles with a composite of four different peptides was reported. That was undertaken by N-terminal cloning four peptides onto the ferritin sequence: an enzymatically cleaved peptide (ECP) to release siRNA, a cationic peptide (CAP) to capture siRNA, a cell penetrating peptide (CPP) and a tumor cell targeting (CTP) moiety [50]. N-terminal fusion proteins have also been produced to display viral hemagglutinin on the ferritin surface, with the resulting neutralization of H1N1 viruses [51]. Alternatively, the production of ferritin nanoparticles with peptides fused at the C-terminal has been reported for the development of dendritic cell-based vaccines [52]. Beyond these strategies, other approaches for the conjugation of targeting moieties on ferritin surface have also been explored, including ferritin nanocage biotinylation in order to insert targeting functionalities [53], or the genetic

engineering of ferritin for generating a fusion protein, which exploits N-terminal protein G for antibody immobilization [54].

4. Nanotechnological applications of ferritin

Ferritin has attracted much interest due to its potential use as reaction chamber for the production of metal nanoparticles, or as a template for semi-conductor production [55]. Ferritin nanocages allow chemical synthesis to occur in restricted cavities with homogenous shape and atomic composition [56]. Ferritin nanocages have been extensively exploited for the biomineralization of metal oxides, such as iron [57], manganese [58], cobalt [59], chromium and nickel oxides [60]. Among the different protocols, the simplest and most suitable method of mass-production is the one-pot synthesis strategy developed by the Yamashita group [56], which developed metal-loaded ferritin nanocages with potential for nanoelectronic application [56].

Nanoparticles obtained from mineralization of semiconductors are very attractive from the nanotechnological point of view, since their fluorescence properties are closely related to nanoparticle size and shape [56]. In this context, ferritin nanocage cavities have been used for the synthesis of semi-conductor nanoparticles. However, the main drawback of this process is that ion aggregation is induced by high concentrations of Cd^{2+} or Zn^{2+} and Se^{2-} during the chemical reaction [56, 61], and because of this only a few research groups are engaged in its development [56, 61]. The synthesis of CdSe and ZnSe using ferritins as a reaction chamber has been successfully obtained by Yamashita and coworkers, who have designed a new chemical approach called the Slow Chemical Reaction System [62, 63]. This method avoids ion aggregation since Cd^{2+} or Zn^{2+} are attracted inside the apoferritin chamber by inner negative residues, thus preventing the aggregation process; therefore, mineralization inside an apoferritin cavity slows down the nucleation of CdSe or ZnSe [62-65].

5. Potential of ferritin nanoparticles in cancer

While ferritin nanoparticles have been widely investigated in the context of nanotechnology, their main application concerns the field of nanomedicine. Indeed, apart from their application in the

development of vaccines [51, 52], several research groups are engaged in the study of ferritin as imaging and drug delivery systems for the diagnosis and treatment of tumors, as summarized in Table 1.

5.1 Ferritin nanoparticles as a drug delivery platform

Their high stability, biocompatibility, ability to disassemble and reassemble in a shape memory fashion and disposition for surface modification make ferritin nanoparticles an ideal platform for drug delivery [55]. Indeed, in physiological conditions ferritin has a stable 24-mer cage architecture, while in highly acidic and basic conditions its quaternary structure disassembles, and then reassembles when the pH of the solution is brought to neutrality [66]. This feature has been used to encapsulate molecules in solution within ferritin cavities, simply by disassembling and reassembling the 24-mer architecture (Figure 2). Drugs with a natural tendency to bind metals, such as cisplatin and desferoxamine B [67], have been easily entrapped in the ferritin shell. Cisplatin encapsulation was first reported in 2007 by Gao and coworkers [68], who also studied the cellular uptake of these nanoparticles and several applications in tumor treatment [69]. The Huang group has used cisplatin-loaded apoferritin to study, with a proteomic approach, the apoptotic process induced by nanoparticles in gastric cancer cells [70]. Moreover, a drug delivery device targeted to melanomas has been developed using cisplatin-loaded ferritin, chemically functionalized with an antibody against the melanoma antigen CSPG4 [42]. These studies have provided strong evidence of an enhanced efficacy of antitumor therapy when it is encapsulated in ferritin-based nanoparticles, particularly in terms of targeting cancer cells such as melanoma, a type of cancer totally refractory to chemotherapy in later stages [42].

However, most of the currently used chemotherapies are not based on metals such as cisplatin, and the incorporation of non-metal-containing drugs within ferritin is complicated by their limited interaction with the protein cage. To overcome this drawback, different approaches have been evaluated, mainly focused on forming complexes of drugs with transition metals or the addition of charged accessory

molecules [3, 71]. Doxorubicin (DOX) encapsulation in ferritin nanocages represents the most extensively investigated system for delivery of anticancer drugs [47]. DOX is a widely used cytotoxic drug for several types of solid tumors and has an excellent anticancer activity; however, its use is dose-limiting since it is associated with several major toxicities, when administered at high doses [47]. DOX pre-complexed with Cu(II) has been loaded inside ferritin nanocages and evaluated *in vitro* and *in vivo* in U87MG glioblastoma tumor models. Eighty percent of DOX was gradually released from RGD-functionalized apoferritin within 10 hours of incubation at 37 °C in phosphate buffered saline (PBS), while the copper remained bound to the internal surfaces of the nanocages. This targeted nanoformulation has increased drug tumor uptake and accumulation, tumor growth inhibition, circulation half-life, and has reduced the cardiotoxicity induced by free DOX [47]. Other groups have set up DOX loading strategies that do not involve complexes with transition metals. In these cases, good encapsulation efficiency has also been obtained [72, 73, 27], and excellent results in terms of tumor growth inhibition and reduced toxicity have been reported, both *in vitro* [27] and *in vivo* [73]. *In vivo* results clearly demonstrated that ferritin nanocages improve DOX bioavailability, tumor accumulation and clearance, and suggested that ferritin-mediated active targeting provides a major contribution (Figure 4) [73]. The mechanisms and kinetics of drug release from ferritin shell has not been completely elucidated, although reported data with DOX suggest that encapsulation is stable in serum and that a pH-triggered release takes place *in vitro* [75]. Moreover, the Knez group demonstrated that ferritin translocates into the nucleus mediating the nuclear release of DOX, while our group has hypothesized that DOX loaded ferritin acts as a Trojan horse because it has been translocated into the nucleus following DNA damage caused by the partial release in the cytoplasm of encapsulated DOX, by a self-triggered translocation mechanism which releases the drug directly into the nuclear compartment [27, 30].

In this context, the use of unfunctionalized ferritin seems to be advantageous for many reasons: 1) higher purification yield [72, 27]; 2) high tumor recognition [72] specific tumor recognition mediated by TfR1; 3) ferritin specific physiological behavior is not prevented, allowing the nuclear

translocation for the release of DOX to be exploited [27]. Furthermore, ferritin nanoparticles have been used to stabilize lipophilic drugs, such as curcumin [74]. However, to date few studies have been published on this, probably because low drug hydro solubility strongly affects the encapsulation reaction.

Ferritin-based nanocages could be implemented for the delivery of radioisotopes increasing loading efficiency and improving pharmacokinetics. Hainfeld has developed ferritin cages with a payload of about 800 ^{235}U atoms/nanoparticle, able to kill surrounding tumor cells [75]. The same strategy has been suggested as being applicable with other isotopes currently used in clinics, such as $^{90}\text{yttrium}$ and $^{177}\text{lutetium}$ [3].

Finally, ferritin nanocages have been employed for siRNA or miRNA delivery. These short non-coding RNA molecules, have been strongly related to either cancer progression or resistance, suggesting that miRNA/siRNA-based therapy could be associated with standard treatment. Lee and coworkers have recently proposed a genetic variant of H-ferritin designed to mediate targeted delivery and internalization of siRNA immobilized on the nanoparticle surfaces through charge-charge interaction [50]. *In vitro* results demonstrated the efficacy of the as-designed nanoparticle in targeting tumor cells and the nanoparticle-mediated intracellular delivery of an anti-red fluorescent protein (RFP) siRNA, which induced RFP suppression [50]. This result paves the way for future application of ferritin nanoparticles in gene silencing.

5.2 Ferritin nanoparticles for in vivo imaging

The plasticity of ferritin as a mineralization chamber for heavy atoms or complexes, and the possibility of modifying ferritin nanocages by protein engineering or by chemical reaction to insert probe molecules, are features that make ferritin an ideal device for imaging [53]. In particular, ferritins have found great application in magnetic resonance imaging (MRI) and optical [73] imaging (Figure 3). Indeed, MRI is a powerful diagnostic technique with wide involvement in tumor imaging, thanks to its high sensitivity and accuracy [55]. However, the currently used gadolinium-based contrast

agents indistinctly enhance all highly vascularized tissues, and thus lack specificity for cancer cells resulting in a high rate of false positives. On the other hand, occult cancer micro-deposits may be not detected on MRI due to an insufficient spatial resolution, and oligometastatic disease could be misdiagnosed [76, 77]. Both endogenous and exogenous ferritins have demonstrated great utility in this area by overcoming these limitations. Ferritin nanoparticles loaded with magnetic species have been developed as MRI contrast agents for tumor imaging. Cao and coworkers have obtained a nanoparticle with transverse relaxivity r_2 of $224 \text{ mM}^{-1} \text{ s}^{-1}$, by synthesizing a single crystal of magnetite inside the ferritin cage [78]. Injection of such nanoparticles in MDA-231 tumor-bearing mice caused a drop of the T2 signal due to TfR1-dependent tumor accumulation [78]. Moreover, iron nanoparticles have been produced into ferritin cages, and optimizing strategies have been developed by Uchida *et al.* by simply adjusting the extent of Fe loading [79]. The increased amount of iron in the reaction mixture determines an increase in the core size of iron nanoparticles encapsulated into ferritin, which is directly related to T2 contrast power [79, 80]. Other types of MRI-detectable nanoparticles have been produced using the ferritin cavity as a reaction chamber. Five nm Gd nanoparticles have been produced inside ferritins, obtaining nanoparticles with longitudinal and transverse relaxivity from 10 to 70 times higher than commercially available Gd-chelates [81]. Mn-coupled ferritin-based MRI contrast agent, obtained by reduction of β -MnOOH, has been reported by Kalman and coworkers. This nanoparticle contains up to 300-400 Mn(II) ions and an r_1 relaxivity ranging between $4000\text{-}7000 \text{ s}^{-1}$ [82]. Aime *et al.* exploited a pH-mediated ferritin disassembly to load about 8-10 Gd-HPDO3A/nanoparticle, obtaining a r_1 relaxivity of $80 \text{ mM}^{-1} \text{ s}^{-1}$ [83]. Upon modification with C3d, a peptide that specifically recognizes vessel endothelium, Gd-nanoparticles showed a 30% signal increase in tumor in a mouse model [53].

Ferritin nanoparticles can also be conjugated with dye molecules and find application as optical imaging probes in imaging-guided surgery for cancer [84]. Li *et al.* employed a ferritin H-chain genetically fused with epidermal growth factor (EGF), and subsequently labeled Cys residues with Alexa Fluor 750, for targeting EGFR-overexpressing breast cancer cells in mice [85]. Besides, Lin *et*

al. developed a Cy5.5-labelled ferritin by reacting the dye with the protein's primary amine. Then, using the pH-disassembly procedure, the Cy5.5-labelled ferritin monomers were mixed with RGD-bearing ferritin monomers. The resulting hybrid ferritin, which displays on the surface both the Cy5.5 dye and the RGD peptide, has been injected into U87MG tumor-bearing mice demonstrating good tumor accumulation [22]. In another study, Lin and coworkers mixed monomers of ferritin conjugated with the Cy5.5-GPLGVRG peptide or with the black hole quencher-3 (BHQ-3), obtaining a hybrid nanoparticle that emits fluorescence only in a metalloproteinase rich environment, such as that found in tumors [43]. This nanoparticle was further implemented through the encapsulation of ^{64}Cu , a radioisotope commonly used in positron emission tomography (PET), thus obtaining a ferritin platform for multimodal imaging (Figure 5) [22]. Combining optical and nanotargeted strategies could improve and refine intraoperative imaging, to properly excise a tumor lesion with adequate margins, and provide targeted diagnostic tools for cancer follow up.

5.3 Ferritin nanoparticles in photothermal therapy

Thanks to its high biocompatibility, low immunogenicity and good pharmacokinetics profile ferritin is useful in photothermal therapy (PTT) [86]. PTT employs photothermal agents, which absorb the irradiation energy of an optical laser and convert it into heat for killing cancer cells. Indeed, most of the photothermal agents show high immunogenicity, non-biodegradability, long term toxicity and poor pharmacokinetics, which strongly affect their clinical application [87, 88]. Haung *et al.* have reported the development of a near infra-red (NIR)-loaded ferritin, which displayed strong absorbance in the NIR region for photoacoustic/fluorescence multimodal imaging-guided PTT [89, 90]. The as-designed ferritin nanovector exhibited *in vivo* high PTT efficacy and low systemic toxicity in comparison to the free dye [90]. Another kind of ferritin nanoparticle has been developed by the Ceci group as a platform for hyperthermia. They functionalized the ferritin cage with the melanoma targeting peptide and used it as a mineralization chamber for the synthesis of cobalt-doped ferrite nanoparticles. The as-synthesized nanoparticles were uniform in size (6-7 nm) and morphology (spherical). Moreover, they

displayed an *in vitro* hyperthermic efficacy inversely related to the amount of Co used for the doping [45]. Altogether, the reported data suggest that ferritin nanocages could be a promising vector for photothermal therapy, although further studies have to be performed to better investigate ferritin's potential in this field.

6. Conclusions and future perspectives

So far, nanotechnology has emerged as promising frontier in cancer treatment. Super-paramagnetic iron oxide nanoparticles (SPIONs) are being tested in clinical trials as magnetic contrast agents for MRI and for intraoperative sentinel node identification during breast cancer surgery [91, 92]. Liposomes have been already established in clinical practice for improving bioavailability and reducing toxicity profile of some excellent cytotoxic drugs associated with severe cardiotoxicity, such as anthracyclines [93]. Nanoformulated albumin-bound paclitaxel has been notoriously introduced in clinical management of various types of cancer with encouraging results in terms of anticancer efficacy [94]. However, these basic nanoparticles lack a specific method of cancer targeting, and despite their benefit in reducing side effects of cytotoxic drugs in clinical trials, no substantial improvement in long-term outcomes has been documented [95]. In this context, ferritin nanocages appear to be powerful and fascinating tools for both imaging and therapeutic purposes, in particular considering that ferritin is a protein normally present in physiological systems.

Different strategies of loading ferritin with drugs, molecules and metals have been successfully explored; nevertheless, some aspects particularly regarding their kinetic release from the nanovector still need to be investigated in detail. Much work has already been done to increase the types of molecules encapsulated, particularly in the case of non-metal-containing drugs, which include the majority of chemotherapeutics. However, a few questions remain: (1) the fate of ferritin and its derivatives after systemic injection and (2) the possibility to suppress ferritin uptake in off-target organs and/or sequestration by the reticulo-endothelial system, and thereby (3) the hope to extend a drug's half-life by modifying the ferritin surface with stealth moieties. Some research groups have

taken advantage of the ability of H-ferritin to specifically recognize TfR1 [33], a receptor expressed on the surface of many types of cancer cells [41]. Other groups have chemically or genetically inserted targeting motifs onto the ferritin surface, to improve tumor homing [42-54]. This strategy has been demonstrated to be very promising, since targeted therapies are probably the key for cancer treatment [96]. Specific targeting of cancer cell receptors enhances anticancer activity by selective inhibition of the underlying cellular pathways, and mediates a selective release of cytotoxic drugs. However, despite the fact that surface modification of ferritin imparts target specificity, it is not clear whether or not these modifications alter the nanoparticle's immunogenicity. Another aspect which should not be underestimated is the possibility of combination therapy. Encapsulation of various drugs into the same nanocage would provide contemporary multiple anticancer activities towards various cancer survival pathways. Finally, the intrinsic capability of ferritin nanocages to undergo nuclear translocation can be exploited to design a novel class of smart nanodrugs, which could act as "Trojan horses", by releasing chemotherapy into the nucleus of cancer cells, thus strongly improving their cytotoxic activity. At present, unfunctionalized DOX-loaded ferritin represents the most extensively investigated of the ferritin-based nanoparticles, and several *in vitro* and *in vivo* studies have demonstrated it to be successful. However, such nanoparticles are currently not used in clinical practice, and further research is required before clinical translation. Ferritin has also found an intriguing application in the field of *in vivo* imaging. Particularly, ferritin-based contrast agents for MRI have been developed, providing an imaging platform with increased tumor accumulation in comparison to the contrast agents currently used in clinical practice. However, toxicity, biodistribution and clearance of these nanocompounds needs to be more clearly understood before their clinical translation. Overall, ferritin nanocages are promising nano-platforms for imaging and therapy of cancer. Some issues still need to be thoroughly investigated, but many significant results have already been obtained. Physiological features and the bioengineering versatility of ferritin-based nanocages make their translation from bench to bedside a reasonable possibility. Moreover, the combination of imaging and therapeutic

functionality into ferritin nanocages seems to be the most promising and fascinating frontier in cancer theranostics.

Acknowledgements

The research was supported by Fondazione Regionale per la Ricerca Biomedica (NANODRUG platform project). FRRB supported S.M, L.F. and M.T. We thank Chiara SRL and E. Fezzuoglio for figures and graphical abstract.

FIGURES AND TABLES

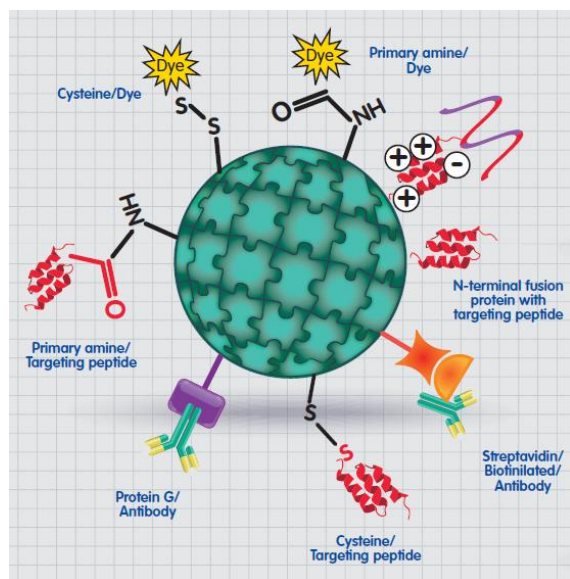


Figure 1. Schematic representation of strategies adopted to provide surface functionalization of ferritin with targeting peptides, antibodies, siRNA and fluorescent dyes.

Table 1. Main application prospects of ferritin nanocages in cancer diagnosis and treatment.

Loaded with...	Functionalized with...	Application	Ref. number
Cisplatin	-	Chemotherapy	68, 69, 70
Cisplatin	antibody against the melanoma antigen CSPG4	Chemotherapy	42
Doxorubicin	RGD peptide	Chemotherapy	47
Doxorubicin	-	Chemotherapy	27, 72, 73, 75, 30
Curcumin	-	Therapy	74
^{235}U	-	Radionuclide Therapy	75
-	anti-red fluorescent protein (RFP) siRNA	RNA interference Therapy	50
Magnetite crystal		MR Imaging	78, 79, 80
Gadolinium		MR Imaging	81, 83

Mn(II)		MR Imaging	82
Gadolinium	C3d	MR Imaging	53
-	EGF Alexa Fluor 750	Fluorescence Imaging	85
-	RGD peptide Cy5.5	Fluorescence Imaging	22
-	Cy5.5 GPLGVRG peptide black hole quencher-3 (BHQ-3)	Fluorescence Imaging	
⁶⁴ Cu	RGD peptide Cy5.5	Positron Emission Tomography	22
NIR	-	Photothermal Therapy	90, 91
Cobalt-doped ferrite nanoparticles	Melanoma targeting peptide (MSH)	Magnetic Fluid Hyperthermia	45

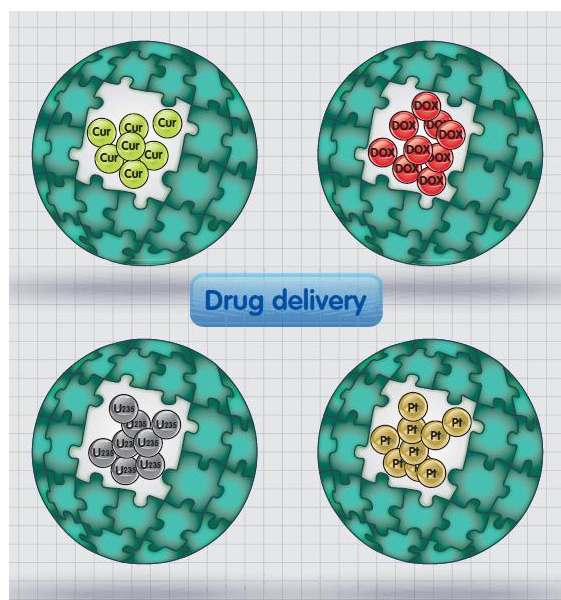


Figure 2. Schematic representation of ferritin developed for drug delivery to cancer.

Figure 3. Schematic representation of ferritin developed for cancer imaging.

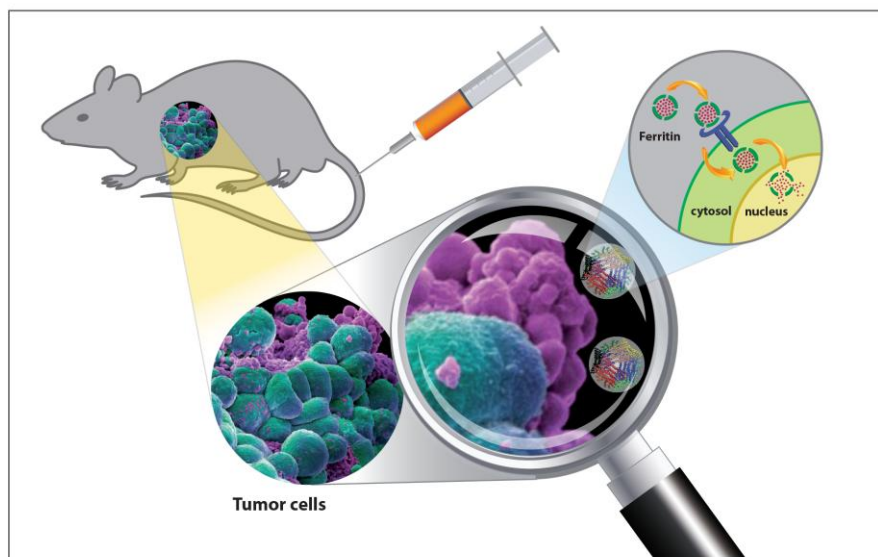


Figure 4. Schematic representation of *in vivo* drug delivery of ferritin nanoparticles.

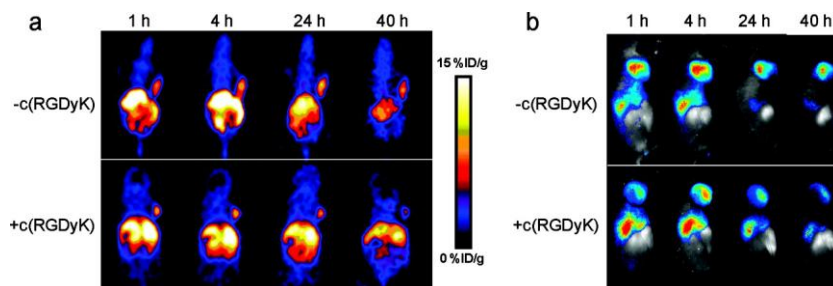


Figure 5. *In vivo* imaging with ferritin nanoparticles. *In vivo* images of (a) positron emission tomography (PET) and (b) near-infrared fluorescence images after administration of ferritin nanocages. 30 min before ferritin probe administration, mice were injected with a blocking dose of c(RGDyK) to demonstrate target specificity. (Reprinted with permission from Ref 22. Copyright 2011 American Chemical Society).

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