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Abstract: One of the most critical quality attributes of orodispersible films (ODFs) is related to the development of dosage forms with tensile properties suitable for the packaging and patient's handling. Aiming to individuate a strategy to reinforce the tensile properties, the current work reported the feasibility to improve the tensile strength of maltodextrins (MDX) based ODFs by adding an amorphous water insoluble nanofiller, namely polyvinylacetate (PVAc).

The possible interactions between components investigated by DSC and ATR-FTIR spectroscopy revealed that MDX and PVAc were immiscible; even if, the presence of plasticizers permitted the homogeneous dispersion of PVAc in the film until the 10% w/w strength was reached. As a consequence, PVAc nanoparticles resulted an effective reinforcing agent only at the concentrations of 3 and 5 % w/w. In this optimal range, the tensile strength increased of at least 1.5 fold and the elastic modulus increased of at least 4 times.

Highlights

- The tensile properties of maltodextrin orodispersible films were improved by loading polyvinyl acetate (PVAc) nanoparticles at the concentrations of 3 and 5 % w/w.
- Maltodextrin and PVAc are immiscible.
- The presence of plasticizers, namely glycerin and sorbitan monooleate, favored the homogenous dispersion of PVAc nanoparticles to the **strength** of 10% w/w.

Nanofiller for the mechanical reinforcement of maltodextrins orodispersible films

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24 **Abstract**

25 One of the most critical quality attributes of orodispersible films (ODFs) is related to the development of
26 dosage forms with tensile properties suitable for the packaging and patient's handling. Aiming to
27 **individuate** a strategy to reinforce the tensile properties, the current work reported the feasibility to
28 improve the tensile strength of maltodextrins (MDX) based ODFs by adding an amorphous water insoluble
29 nanofiller, namely polyvinylacetate (PVAc).

30 The possible interactions between components investigated by DSC and ATR-FTIR spectroscopy revealed
31 that MDX and PVAc were immiscible; even if, the presence of plasticizers permitted the homogeneous
32 dispersion of PVAc in the film until the 10% w/w **strength** was reached. As a consequence, PVAc
33 nanoparticles **resulted an** effective reinforcing agent only at the concentrations of 3 and 5 % w/w. In this
34 optimal range, the tensile strength increased **of** at least 1.5 fold and the elastic modulus increased **of** at
35 least 4 times.

36

37 **Keywords:**

38 Orodispersible films; maltodextrins; nanofiller; mechanical reinforcement; tensile properties; casting
39 solution.

40 1. Introduction

41

42 Orodispersible films (ODFs) are fast-dissolving dosage forms intended to deliver drugs both locally and
43 systemically after administration in the oral cavity. ODFs have gained an increased interest in  academic and
44 industrial world because they have a huge clinical potential to legitimate expectations of special
45 subpopulations, such as pediatric, elderly and dysphagic patients, who have difficulty in swallowing
46 conventional tablets and capsules (Borges, Silva, Coelho, & Simões, 2015; Dixit & Puthli, 2009). ODFs made
47 of maltodextrins (MDX) with a dextrose equivalent in the 6-12 range, have been successfully proposed due
48 to suitable filmogenic properties, high water solubility and low cost (Cilurzo, Cupone, Minghetti, Selmin, &
49 Montanari, 2008). From a technological point of view, ODFs are generally processed by casting an aqueous
50 solution of MDX properly plasticized by hydrogen-bonding plasticizing agent (e.g., glycerol and natural
51 aminoacids), the active compound of interest and some other functional ingredients (e.g., taste-masking
52 agents and flavours) (Cilurzo et al., 2011; Selmin, Franceschini, Cupone, Minghetti, & Cilurzo, 2015). After
53 solvent evaporation, the film is further processed, namely cut into a desirable size and shape and packed.
54 The preparation of films for oromucosal drug delivery is challenging compared to other dosage forms.
55 Indeed, depending on the physical-chemical characteristics of the ingredients loaded and their amount in
56 the formulation, modifications of the mechanical properties can occur resulting in films too flexible or
57 brittle to guarantee the integrity. As an example, the incorporation of an oily drug slightly increased
58 plasticity acting as a plasticizer (Cilurzo et al., 2010). However, as also stated by European Pharmacopoeia
59 (EDQM) the mechanical strength of ODFs is of paramount importance since they should resist handling
60 without being damaged or broken. Currently, there is a growing tendency to develop polymeric film
61 reinforced with nanoparticles to avoid this drawback. For instance, starch-based matrix materials loaded
62 with nanosized fillers, e.g. clay minerals (Chen & Evans, 2005; De Carvalho, Curvelo, & Agnelli, 2001;
63 Wilhelm, Sierakowski, Souza, & Wypych, 2003) or polymers (Le Bolay, Lamure, Gallego Leis, & Suhani, 2012;
64 Le Bolay & Molina-Boisseau, 2014), exhibited drastic modification in their properties, such as improved
65 tensile strength and elongation and decreased the vapour diffusion (Park, H. W., Lee, Park, C. Y., Cho, & Ha,
66 2003). Generally speaking, it was demonstrated that the extent of the reinforcement depends on such
67 factors as (i) the elastic properties of its constitutive phases, (ii) the volume fraction of filler (iii) the
68 morphology (i.e., shape, aspect ratio, and distribution of the filler into the polymeric matrix) and (iv) the
69 interactions between fillers. However, the low interfacial compatibility between polymer and filler can
70 result in the poor dispersion of nanofiller in the polymeric matrix. Consequently, the formation of
71 nanoparticles agglomerates as a consequence of their high specific surface area or hydrogen bonding
72 ability, may result in lower final properties (Pettersson & Oksman, 2006). Homogeneous dispersion of
73 nanoparticles can be achieved preparing nanocomposites in solution by solvent casting methods or by
74 blending and compatibilization with components, which display complementary properties and can

75 contribute to improve the nanofiller dispersion through effective interaction at the interface (Fortunati,
76 Puglia, Kenny, Haque, & Pracella, 2013).

77 In view of the above considerations, the current work reports a comprehensive investigation on the
78 behavior of plasticized MDX based ODFs loaded with different amounts of an amorphous water insoluble
79 nanofiller, namely Kollicoat® SR 30 D (PVAc). The tensile properties of the nanocomposite films were
80 studied taking in account the possible interactions between the polymers at molecular level.

81

82 **2. Materials and Methods**

83

84 *2.1. Materials*

85 Maltodextrins having a DE equal to 6 (Glucidex® IT6, MDX) was obtained from Roquette Frères (F).
86 Kollicoat® SR 30 D (PVAc) was kindly gifted by BASF (G). It is a 27 % w/w colloidal aqueous polymer
87 dispersion of polyvinylacetate with a particle size of about 170 nm stabilized by sodium lauryl sulfate (0.3 %
88 w/w) and povidone (2.7 % w/w) (Kolter, Dashevsky, Irfan, & Bodmeier, 2013). Sorbitan monooleate (Span®
89 80, S80) was purchased by Uniqema (UK) and glycerol (GLY) was obtained by Carlo Erba Reagenti (I). All
90 materials were used without any further manipulation.

91

92 *2.2. Preparation of binary mixtures*

93 Aqueous feeds were prepared dissolving MDX and PVAc in the same ratio as in the placebo films as
94 following: 99/1, 97/3, 95/5, 90/10 and 80/20 % w/w. Binary mixtures were prepared by using a spray-dryer
95 (Formate 4 M8 – ProCepT, B). Process parameters were set as follows: standard nozzle inner diameter: 0.6
96 mm; air speed 0.30 m³/min; air in temperature: 130 °C; dosing speed: 7 mL/min.

97

98 *2.3. Differential scanning calorimetry*

99 In order to characterize the thermal behavior of the binary mixture made of MDX and PVAc in comparison
100 with the raw materials, DSC was performed by using a DSC 1 Star^e System (Mettler Toledo, CH). Samples of
101 about 8 mg accurately weighted were sealed in pin holed aluminum pans and heated from 0 to 150 °C at
102 the cooling and heating rate of 5 K/min. The DSC cell and RCS were purged with dry nitrogen at 80 and 120
103 mL/min, respectively. The glass transition temperature (T_g) was taken as the inflection point of the specific
104 heat increment at the glass–rubber transition. Three samples were tested for each formulation and data
105 expressed as their average \pm standard deviation.

106

107 *2.4. ATR-FTIR measurements*

108 The absorption of the spray-dried binary blends were collected in the spectral range of 4000–650 cm⁻¹
109 using a Spectrum™One spectrometer (PerkinElmer, I) equipped with an attenuated total reflectance (ATR)

110 accessory. The optical resolution of the spectrometer was 4 cm⁻¹ and 256 spectra were summed prior to
111 the analysis. A duplicate spectrum was collected for each sample after 4 min in order to check that the
112 measurements were not affected by sorption of water vapor during the experiments. No measurable
113 differences were observed between the two spectra (data not shown).

114 The absorption spectra were firstly smoothed using the Origin® 9.1 (OriginLab Corporation, USA) and the
115 position of the maximum of the broad peaks was determined by analysis of the second derivative.

116

117 2.5. Film preparation

118 Films were prepared by the solvent casting technique as previously described (Cilurzo et al., 2008). Briefly,
119 MDX were dissolved in water obtained from a Milli-Q® Integral 3 (Merck Millipore Corporation, G) filter
120 system at 80 °C, according to the composition reported in **Table 1**. The dispersion was cooled down to 40
121 °C, all the other ingredients were added and stirred until homogeneity. After a rest period to remove
122 bubbles, the dispersion was cast over a silicone release liner by a laboratory-coating unit Mathis LTE-S(M)
123 (CH). Operative conditions: coating rate 1 m/min; drying temperature 60 °C; drying time between 15-20
124 min; air circulation speed 1,800 rpm. These conditions were set to obtain films having a thickness of about
125 100 µm.

126 Films were cut into suitable shape and size as required for testing, packed in individual **air-tight seal packs**
127 immediately after preparation and stored at 25°C until use.

128

129 2.6. Film characterization

130

131 Thickness

132 Film thickness was measured by using a MI 1000 (ChemInstruments, USA). The accuracy of the instrument
133 was 2.5 µm ± 0.5%. Before samples cutting, film was placed between the anvil and the presser foot of the
134 micrometer. Film thickness was measured in ten different positions and the result was expressed as the
135 mean value.

136

137 Moisture content

138 The moisture content (*MC*) was determined gravimetrically after keeping films samples of 6 cm² surface at
139 the temperature of 130 °C over a 2 hour period. The results were expressed as the mean of three
140 determinations according to the following equation:

141

$$142 \quad MC = \frac{W_0 - W_f}{W_0} \times 100$$

143

144 where W_0 and W_f are the initial and final weight, respectively.

145

146 *Disintegration test*

147 Disintegration test of films was carried out on 3x2 cm sample according to specifications of tablets and
148 capsules reported in European Pharmacopoeia 7th Ed. 2013.

149

150 *2.7. Tensile properties*

151 Tensile testing was conducted using a texture analyzer AG/MC1 (Acquati, I), equipped with a 5 N load cell.
152 The film was cut into 100×15.0 mm strips and equilibrated at 25 °C for 1 week. Tensile tests were
153 performed according to ASTM International Test Method for Thin Plastic Sheeting (D 882-02). Each test
154 strip was longitudinal by placed in the tensile grips on the texture analyzer. Initial grip separation was 40
155 mm and crosshead speed was 25 mm/min. The test was considered concluded at the film break. Tensile
156 strength, elongation at break, elastic modulus and work were computed to evaluate the tensile properties
157 of the films.

158 *Tensile strength* (TS) was calculated by dividing the maximum load by the original cross-sectional area of
159 the specimen and it was expressed in force per unit area (MPa).

160 *Percent elongation at break* (E%) was calculated by dividing the extension at the moment of rupture of the
161 specimen by the initial gage length of the specimen and multiplying by 100 according to the following
162 equation:

$$163 \quad E\% = \frac{L - L_0}{L_0} \times 100$$

164 where L_0 is the initial gage length of the specimen and L is the length at the moment of rupture.

165 *Elastic modulus* or *Young's modulus* (Y) was calculated as the slope of the linear portion of the stress– strain
166 curve. The result was expressed in force per unit area (MPa).

167 *Tensile energy to break* (TEB) was defined by the area under the stress–strain curve. The value is in units of
168 energy.

169 An average of five measurements was taken for each type of specimen.

170

171

172 **3. Results and discussion**

173

174 *3.1. Characterization of nanocomposite films*

175 Films loaded by PVAc in the composition range reported in **Table 1** were homogeneously opaque in
176 appearance, while the film F0 appeared transparent. The residual moisture content was about 10 % (w/w)
177 in all formulations. All films were handled, cut, and packed without failure. Nanocomposite films

178 disintegrated in less than 1 minutes evidencing that the addition of PVAc in different ratios did not affect
179 the dissolution of MDX.

180

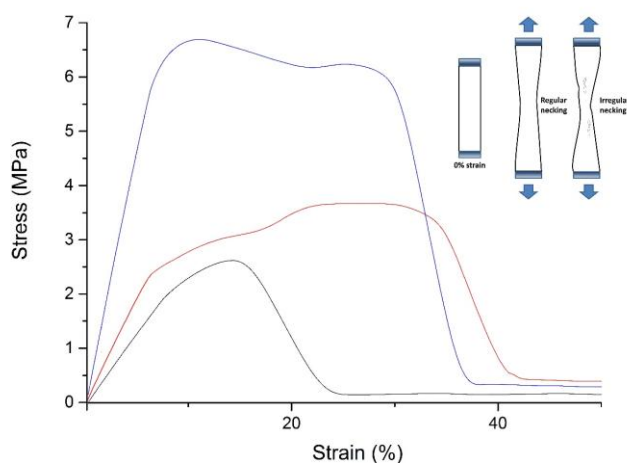
181 **Table 1** – Effect of concentration of PVAc on tensile strength (TS), percentage elongation at break (E%),
182 tensile energy to break (TEB) and elastic modulus (Y) in MDX nanocomposite films. The results are
183 expressed as the mean of five measurements \pm standard deviation.

Form. code	Composition (% w/w)				TS (MPa)	E% (%)	TEB (J)	Y (MPa)
	MDX	GLY	S80	PVAc				
F0	79.0	18.0	3.0	--	1.8 \pm 0.8	23.5 \pm 4.8	0.010 \pm 0.003	4.2 \pm 2.9
F1	78.2	17.8	3.0	1.0	1.9 \pm 0.4	26.5 \pm 3.9	0.007 \pm 0.001	8.9 \pm 2.5
F3	76.5	17.5	3.0	3.0	7.3 \pm 1.1	34.4 \pm 8.0	0.026 \pm 0.007	27.2 \pm 3.5
F5	74.9	17.1	3.0	5.0	3.7 \pm 0.7	58.8 \pm 12.5	0.027 \pm 0.009	11.1 \pm 3.6
F10	70.8	16.2	3.0	10.0	2.8 \pm 0.4	39.0 \pm 5.7	0.012 \pm 0.002	9.3 \pm 1.4
F20	62.7	14.3	3.0	20.0	2.5 \pm 0.4	68.3 \pm 4.2	0.015 \pm 0.002	4.3 \pm 2.1

184

185 The interfacial adhesion between the nanoparticles and the polymer was assessed by studying the stress–
186 strain patterns of films. **Figure 1** shows the stress-strain curve of films representative for all studied
187 formulations.

188



189

190 **Figure 1.** The stress-strain curve of the nanocomposite films representative of all formulations: F0 (black
191 line), F3 (blue line) and F20 (red line). In the insert a schematic representation of films appearance during
192 the tensile test. At nanofiller contents of 3 and 5% w/w, necking occurred after the maximum stress value.

193 With a further addition of PVAc nanoparticles, the necking onset was modified because PVAc nanoparticles
194 segregated preventing the possibility to carry homogeneously the load.

195

196 All films exhibited a linear region at low strain, independently of the composition which is associated to the
197 reversible formation. The higher the slope, the stiffer the material.

198 Increasing the strain, the behavior shifted from elastic to plastic, the curve lost linearity and the
199 deformation became irreversible, until the maximum force (tensile strength, TS) was reached. Then, the
200 force progressively decrease until the breaking of the film. This behavior was typical of films F0 and F1,
201 meanwhile increasing the nanofiller content in the range 3-5% w/w (films F3 and F5), some remarkable
202 changes can be detected after the TS of the stress-strain curve. Indeed, not only the stress required to
203 deform MDX films significantly increased, but also the maximum value was followed by a pronounced
204 plateau at a lower value which was attributed to necking. With a further addition of PVAc nanoparticles
205 (films F10 and F20), the film behavior appeared to be further modified. Indeed, the film homogeneously
206 stretched at low strain before the second yield point. Afterwards, an increase in stress was observed
207 followed by a plateau.

208 This modification in the onset of necking might be attributed to differences in interfacial strength and
209 degree of dispersion due to the very high specific surface area provided by nanoparticle (Wetzel, Hauptert,
210 & Zhang, 2003). Hence, it can be assumed that well distributed PVAc nanoparticles with an adequate
211 interfacial bonding between filler and matrix allows the effective transfer of stress through a shear
212 mechanism from the matrix to the particles that can efficiently carry the load and enhance the strength of
213 the composite (films F3 and F5). At highest filler content (films F10 and F20), PVAc nanoparticles segregated
214 within the matrix and the film consisted **on** a system which does not carry homogeneously the load.

215 These trends are in agreement with the quantitative evaluation of the stress-strain curves (**Table 1**). The
216 control film F0 exhibited a low tensile strength of about 1.8 MPa. Low nanofiller contents resulted in the
217 maximum increase of the film stiffness suggesting that PVAc acts as an effective reinforcing agent at the
218 concentration of 3 and 5 % w/w. Similar results were observed for the toughness, expressed as the tensile
219 energy to break, and the tensile strength. With a further addition of nanofiller, both the mechanical
220 properties progressively decreased, suggesting a weak load transfer from the matrix to nanofiller and/or
221 the failure of the nanofiller included in matrix. As mentioned above, this **features may due** to the volume
222 fraction of PVAc which is related to the formation of aggregates (Bréchet et al., 2001; Liu, Zhong, Chang, Li,
223 & Wu, 2010; Wang & Zhang, 2008).

224 It is worthy to notice the progressive reduction of the elastic modulus at the highest nanofiller
225 concentration is concomitant **to** a significant increase of the elongations at break with respect to the other
226 formulations ($p < 0.05$, **Table 1**). A possible reason lies in the fact that the formation of aggregates made

227 some hydroxyl groups along MDX chains available again for the interaction with water molecules, resulting
228 in an enhanced plasticizing effect. This effect is very similar to that obtained by loading of 10% quercetin
229 nanocrystals which reduced the elastic modulus and increased the elongation at break of the loaded films
230 (Lai et al., 2015).

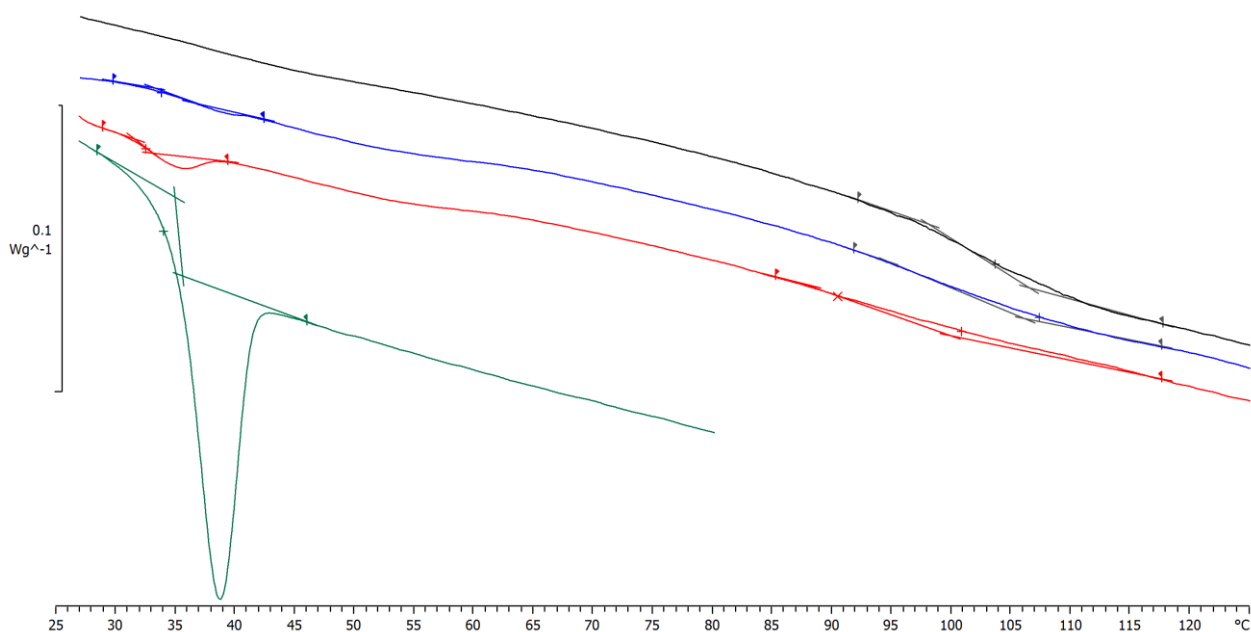
231

232 3.2. Molecular characterization

233 To better understand the mechanism causing the mechanical reinforcing in films, mixtures of spray-dried
234 MDX and PVAc in different ratios were characterized at molecular level using a combination of DSC and
235 ATR-FTIR spectroscopy.

236 DSC data revealed that during the first heating ramp the PVAc glass transition (T_g) was measured in the
237 temperature range 25-41 °C, with an inflection at about 34 °C. This value overlapped the event in the
238 subsequent reheating ramp. Another endothermic peak appears at the temperature higher than 180 °C,
239 probably associated to the thermal decomposition of sodium lauryl sulfate, which is used to stabilize the
240 commercial PVAc used in this study. DSC measurement of the pure MDX showed the T_g at 102.6 ± 2.0 °C.
241 Both distinct events were observed in all composition range, without significant shift in T_g comparing to the
242 single components, suggesting that no chemical reaction or extensive physical interaction between the
243 polymeric components occurred (Figure 2). In other word, a very poor miscibility between PVAc and MDX
244 can be assumed.

245



246

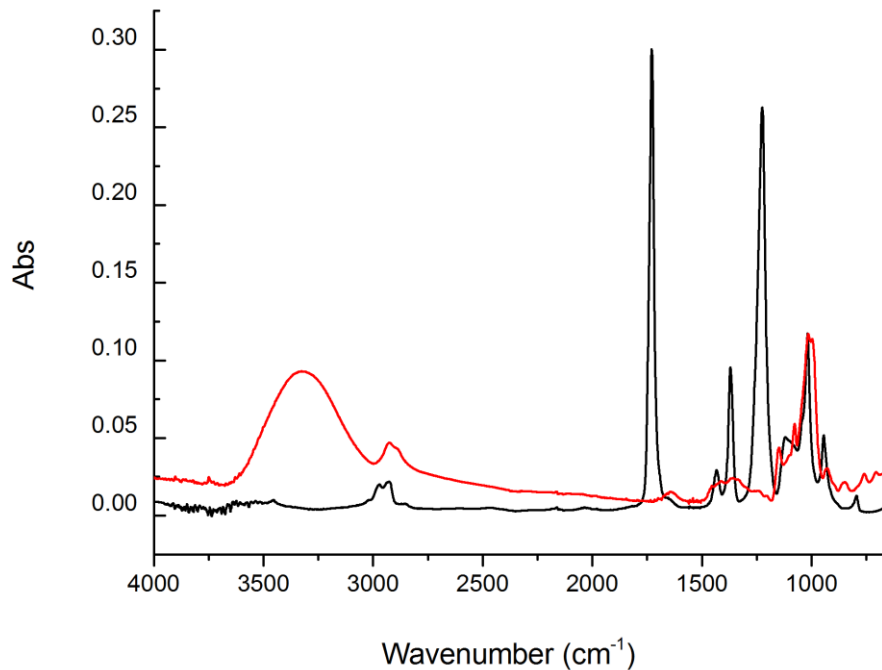
247

248 **Figure 2.** Thermal data of dried PVAc (green line) MDX (black line) and the binary mixtures containing 5%
249 (blue line) and 20% PVAc (red line).

250

251 The mixtures were also subjected to an ATR-FTIR analysis in order to provide some insights into the
252 conformational features. The spectra of dried MDX and PVAc were taken as reference samples to identify
253 the diagnostic bands (**Figure 3**).

254



255

256 **Figure 3.** ATR-FTIR spectra of spray-dried MDX (red line) and PVAc dried in oven until constant weight
257 (black line).

258

259 In particular, the characteristic absorption peaks of PVAc appeared at about 1729 and 1225 cm⁻¹
260 corresponding to the C=O and C-O stretching vibrations, respectively (Wei et al., 2009). The band at 2929
261 cm⁻¹ was attributed to the C-H stretching vibrations band of methyl group.

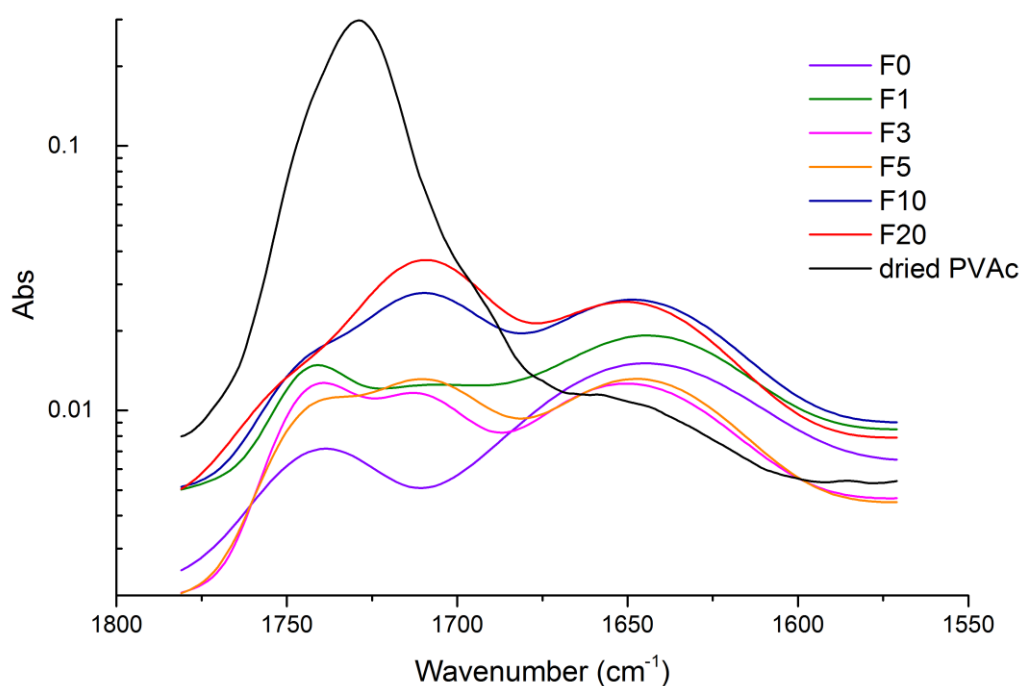
262 As far as MDX is concerned, the principal contributions for carbohydrate polymers are generally in the
263 1300-860 cm⁻¹ region. These vibrations were attributed to C-C and C-O stretching, as well as to COH
264 bending (De Giacomo, Cesàro, & Quaroni, 2008). In particular, the band recorded at about 1022 cm⁻¹ or
265 lower wavenumbers confirmed that MDX used in the current work is completely amorphous (Van Soest,
266 Tournois, De Wit, & Vliegthart, 1995). MDX presents also a broad peak centered at 3309 cm⁻¹ which
267 originated from the fundamental stretching vibration of the -OH group. Hydroxyl groups rarely exist in
268 isolation because they are usually involved in intra- and inter-molecular hydrogen bonding with other
269 hydroxyl groups. The broad features of this peak are due to the wide range of hydrogen bond lengths and
270 orientations, which exists in the amorphous carbohydrate system. Therefore, its position reflects the

271 average hydrogen bond strength in a system (Bellamy, 1975). Finally, the short band in the range of 2908-
272 2930 cm^{-1} arise mostly from the C-H deformation vibrations stretching.

273 In the entire composition range of the MDX/PVAc blends, the main bands of both components were
274 detectable and the minor variations in wavelength and/or peak intensity were attributed to a simple
275 overlapping of the contributes of the two compounds. These data confirming the lack of interaction
276 between MDX and PVAc verified by DSC .

277 Moving to ODFs in which glycerin and Span® 80 were also added, some relevant modifications in the 1800-
278 1550 cm^{-1} region were noted (**Figure 4**). Firstly, the C=O stretching vibration of PVAc shifted from 1729 cm^{-1}
279 to about 1710 cm^{-1} independently of film composition; secondly, this band became predominant in the
280 formulation containing at least the 10 % w/w PVAc. Finally, the intensity of the C=O stretching vibration
281 band was not linearly related to the nanofiller content.

282



283

284 **Figure 4.** Focus on the 1800-1550 cm^{-1} region of the ATR-FTIR spectra of ODFs containing different
285 concentrations of PVAc and the dried PVAc.

286

287 This significant modification in the 1800-1550 cm^{-1} region was concomitant to a slight modification of the
288 band deriving by stretching vibration of hydroxyls at about 3300 cm^{-1} . In this case, the comparison of the
289 spectra recorded on the films and the corresponding binary blends evidenced a reduction in wavenumber
290 and in FWHM (**Table 2**).

291 All together, these results suggested that MDX in reinforced ODFs is involved in a different type of
 292 hydrogen bonds probably mediated by the presence of the plasticizers. The extent of interaction growth
 293 according to the PVAc content. As noticed by the shift and depression of the C=O stretching band of PVAc,
 294 this interaction may be responsible of the nanoparticles aggregation hypothesized on the tensile data. In
 295 other words, at nanofiller content more than 10% the plasticizers resulted insufficient to guarantee a
 296 homogeneous dispersion of PVAc that segregated causing the inhomogeneous transfer of the load within
 297 the film and, therefore, affecting the film tensile properties.

298

299 **Table 2** – Full width at half maximum (FWHM) and wavenumber of –OH stretching vibration in binary
 300 blends and ODFs.

PVAc content (%)	ODFs		Binary blends	
	Wavenumber (cm ⁻¹)	FWHM	Wavenumber (cm ⁻¹)	FWHM
0	3309	332	3335	349
1	3319	348	3334	355
3	3318	342	3336	345
5	3324	343	3335	350
10	3325	346	3335	349
20	3325	339	3336	344

301

302 In conclusion, the reinforcement of ODFs by using a nanofiller was successfully obtained by incorporating
 303 the commercially available dispersion of PVAc in the 3 to 5 % range.

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305

306

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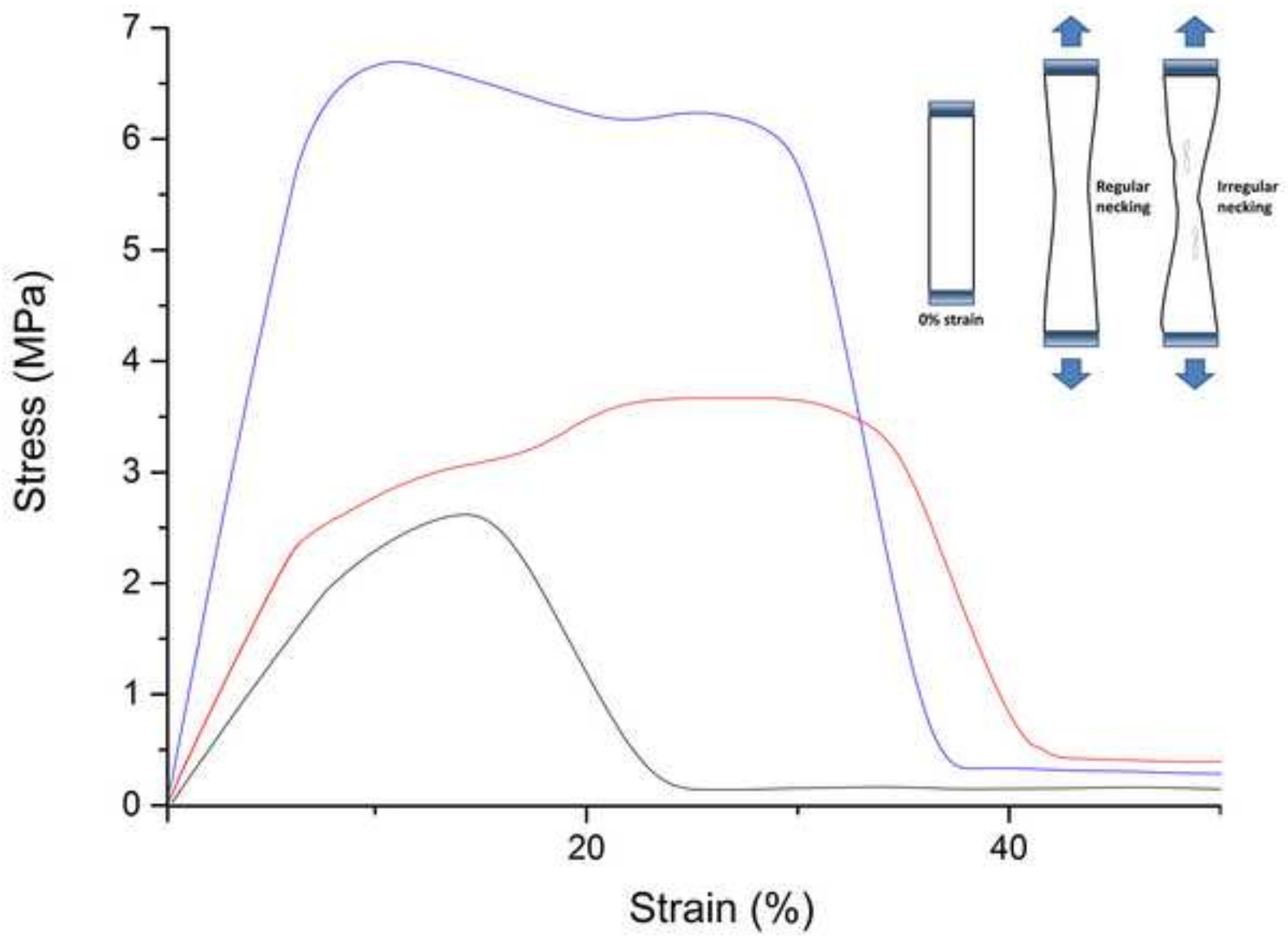


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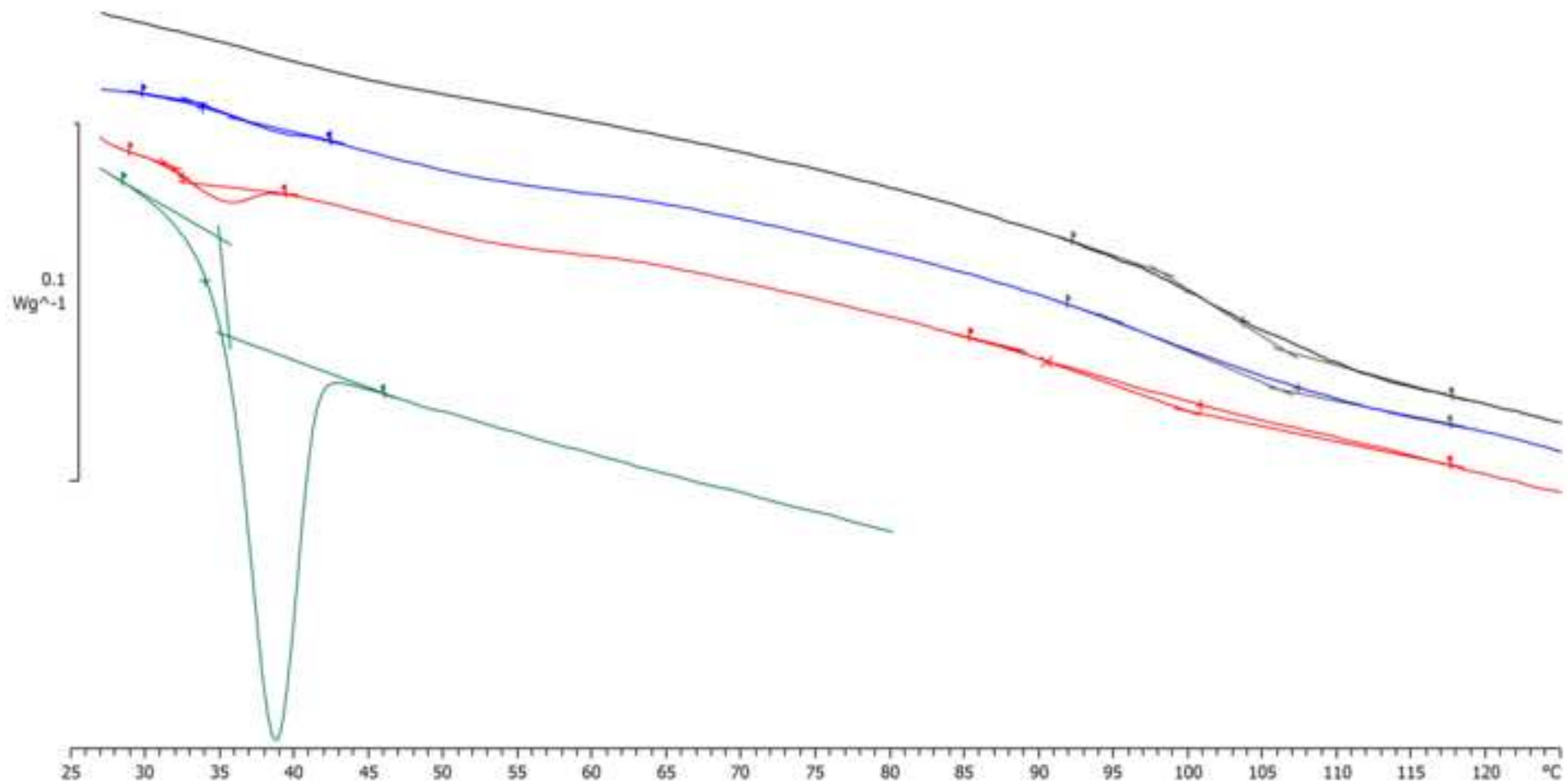


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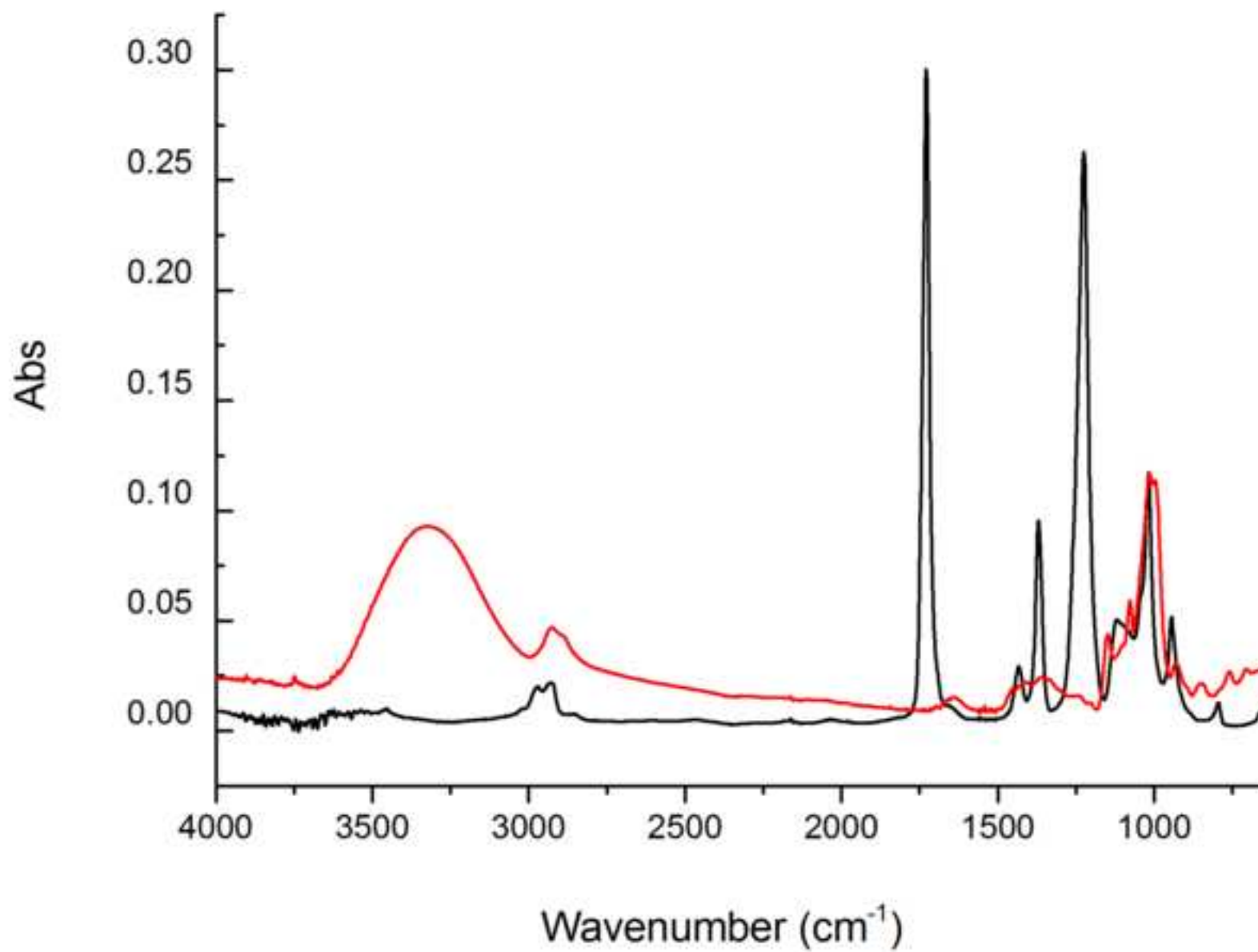


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