

# Constrained-DFT for diabatic PES mapping

## The case of an H atom scattering from a Ge(111) surface

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### 1. Aim

During molecules-surface scattering, energy can be dissipated through the formation of electron-hole pairs. A curious case is the measured energy dissipation of H atoms scattering from a Ge(111) surface.

In this context, constrained DFT (cDFT) offers the possibility to calculate diabatic potential energy surfaces (PESs) in which the H atom has a fixed charge.

### 2. Constrained DFT

ABINIT allows to perform cDFT calculations using an algorithm where the constrained is introduced through Lagrangian multipliers.

[1]

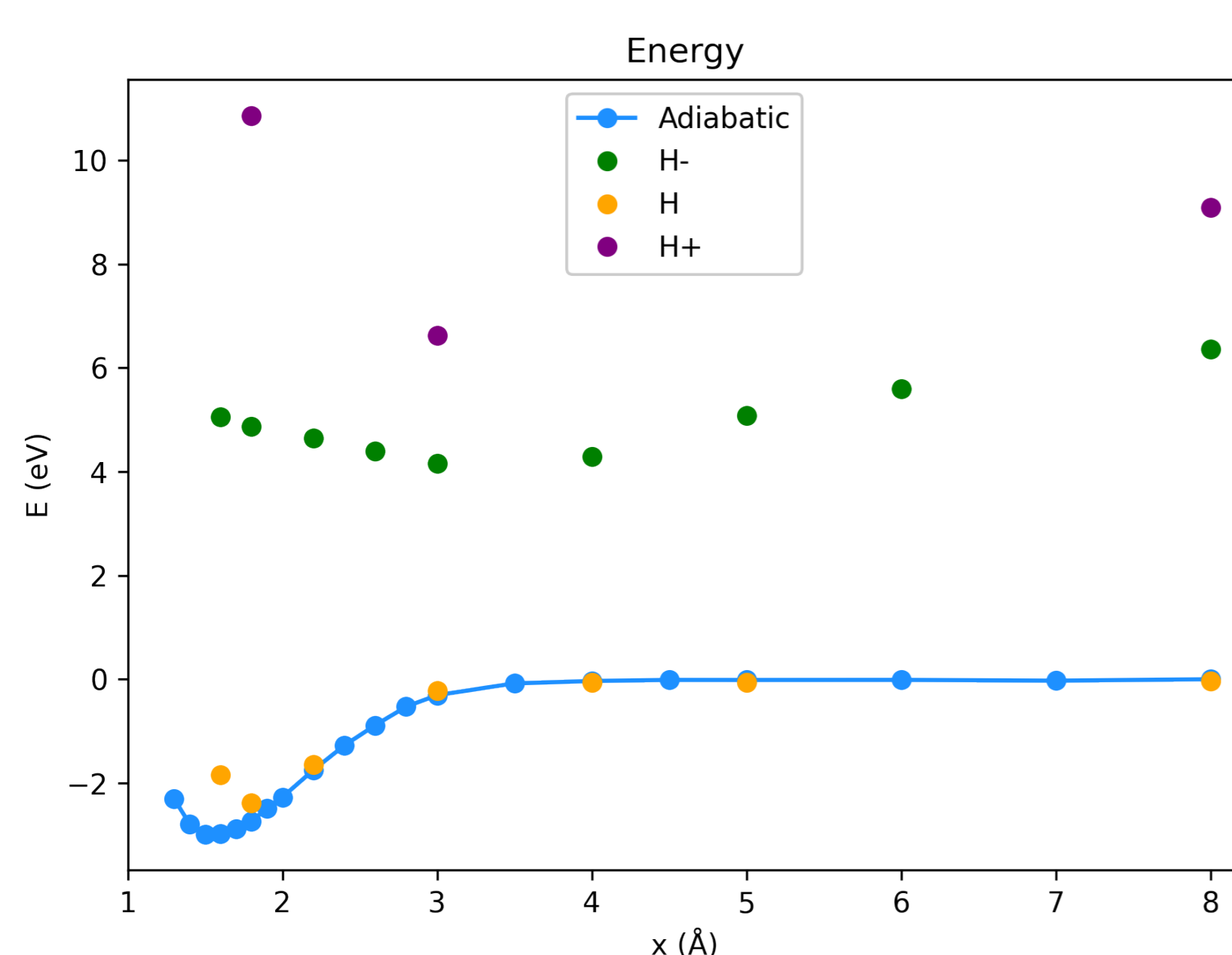
$$E_+^V[u, \Lambda] = E^V[u] + \Lambda(\rho_A^V[u] - N_A)$$

$$N_A \equiv \int w_A(\mathbf{r})\rho(\mathbf{r})d\mathbf{r}$$

A fragment A (represented as a sphere in the code) is considered, with an imposed charge  $N_A$ .  $w_A$  is a weight function, it's equal 1 inside the fragment and smoothly vanishes outside.

[1] X. Gonze *et al.* J. Chem. Theory Comput. 2022 18 (10), 6099-6110

### 3. Diabatic PESs



By combining cDFT with charge analysis (paragraph 4), diabatic PESs associated with H in neutral (orange), anionic (green) and cationic (purple) states are obtained.

These values are displayed together with the adiabatic PES (blue), obtained without imposing any charge constraints.

**Notes:** PESs analysis was performed on a model system, a Ge(111) (2x2) surface cell, which is smaller than the well known Ge(111)c(2x8) reconstructed cell. Although these results can not be directly used to model the dynamics, they provide a qualitative understanding of the scattering event.

### 4. Quantum theory of atoms in molecules

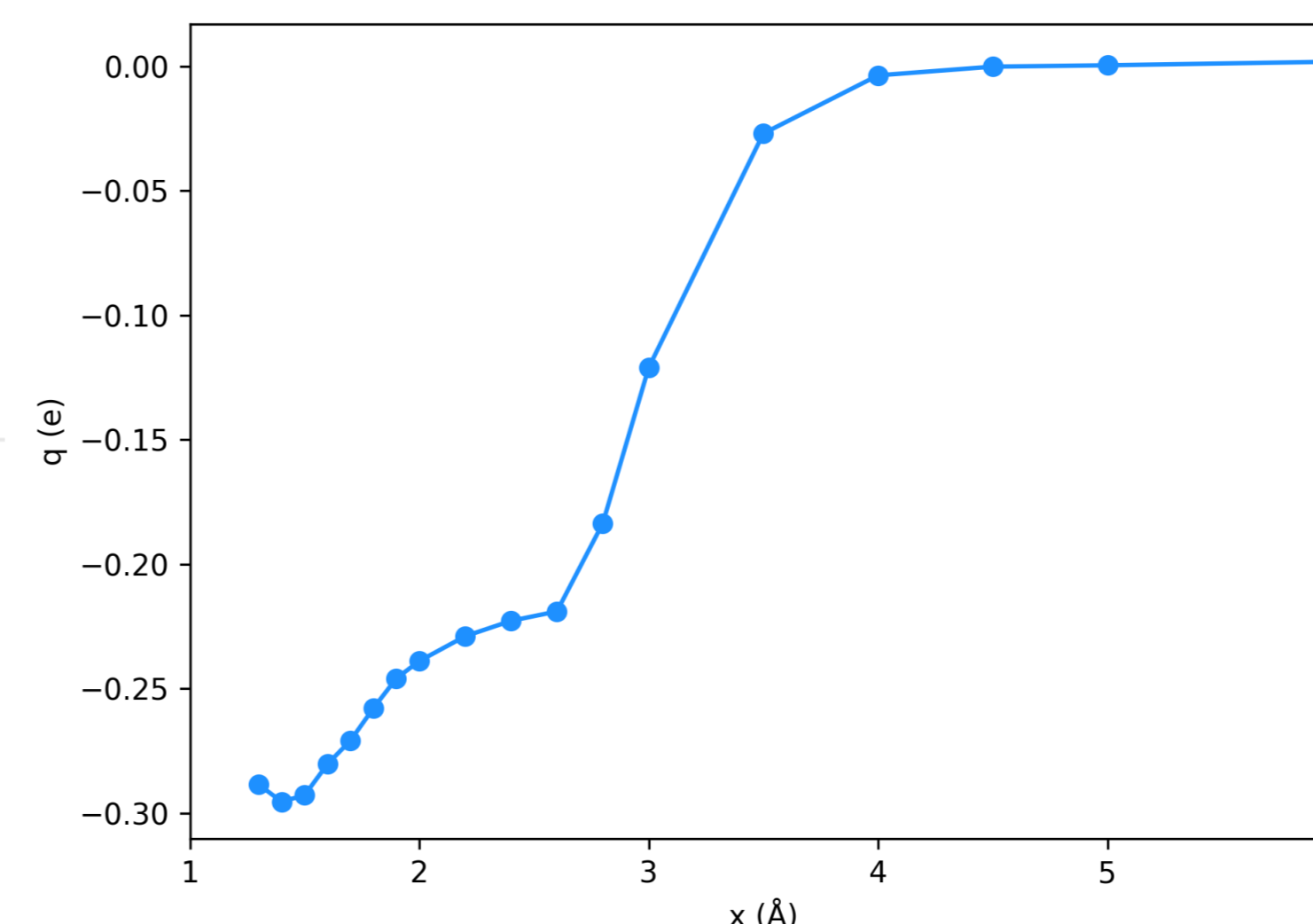
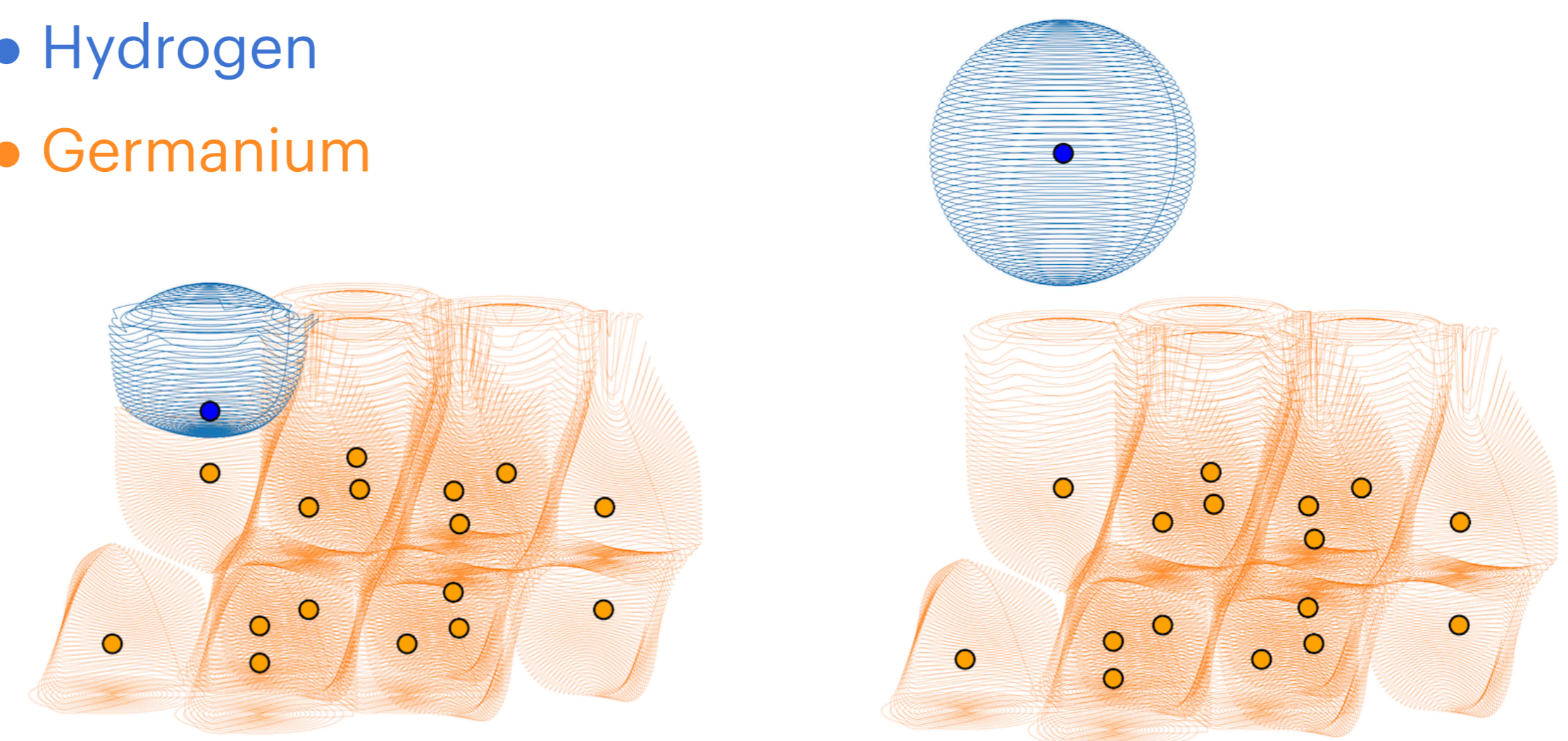
The theory was formulated by Bader and provides a quantum mechanical justification of the concept of atoms. Atomic basins are regions delimited by zero-flux surfaces, where the directional derivative of the charge density along the surface normal  $\mathbf{n}(\mathbf{r})$  is zero.

$$\nabla\rho(\mathbf{r}) \cdot \mathbf{n}(\mathbf{r}) = 0$$

The atomic charge is obtained by integrating the electron density over the atomic basin  $\Omega_H$ :

$$q_H = Z_H - \int_{\Omega_H} \rho(\mathbf{r})d\mathbf{r}$$

- Hydrogen
- Germanium



The Bader charge was calculated for a H atom at different distances from a Ge(111) surface in the adiabatic setup. The hydrogen partially gains electrons as it approaches the surface.

### 5. Application in quantum-classical dynamics [2]

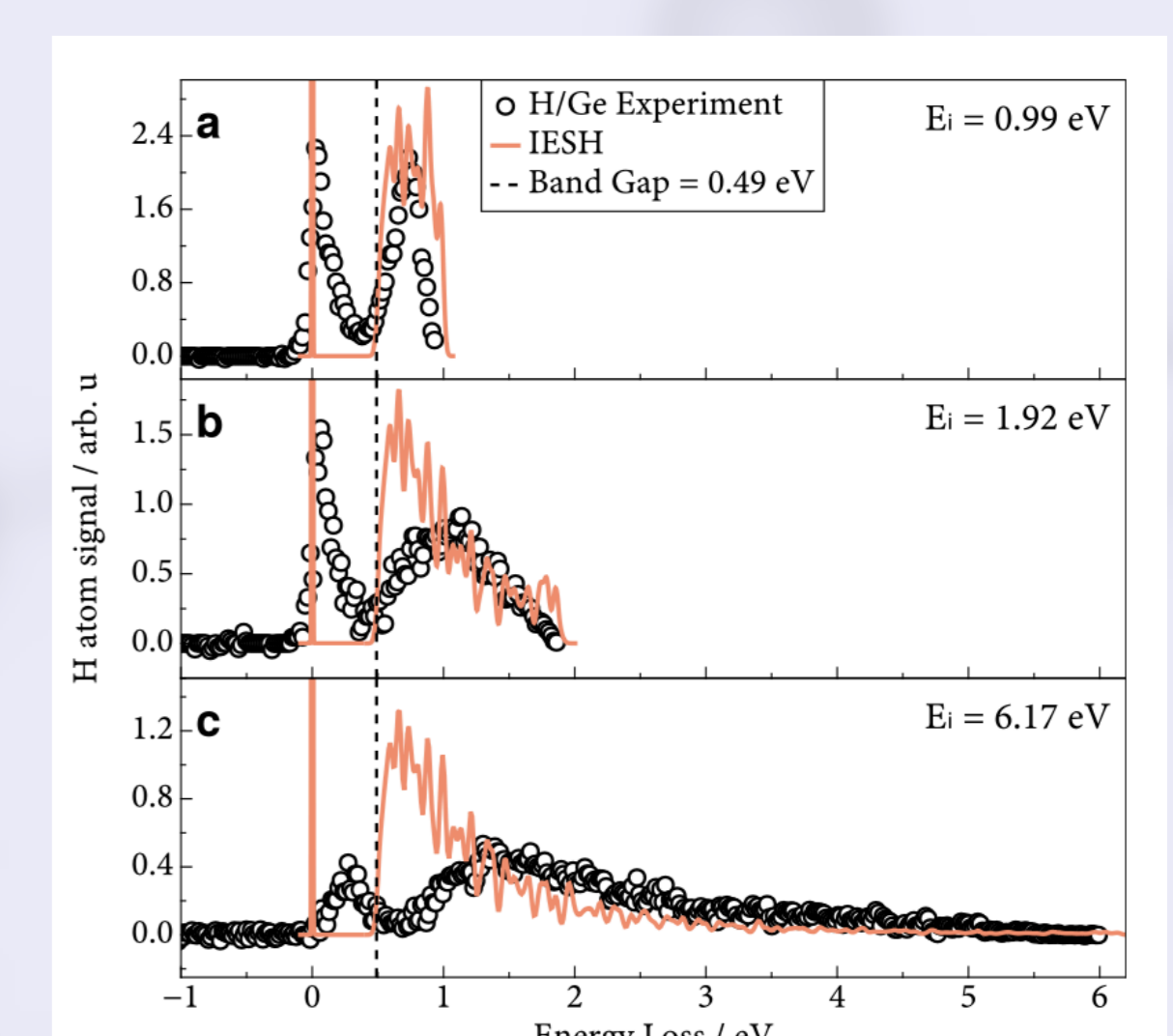
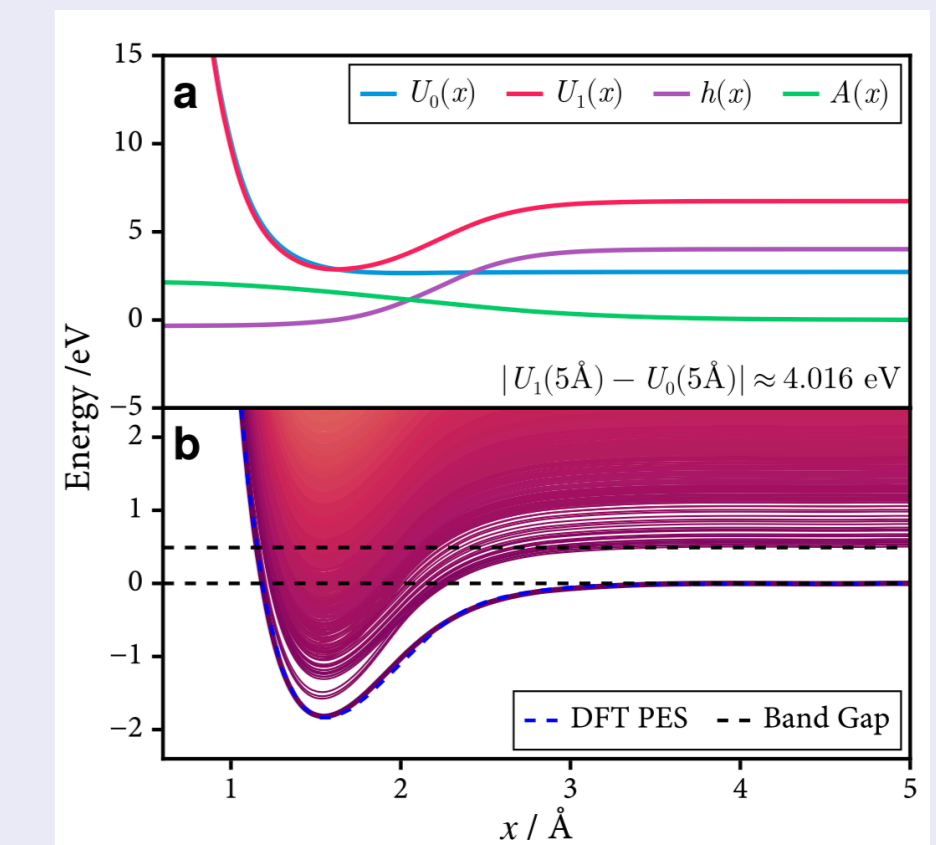
Electronic Hamiltonian in second quantization:

$$\hat{H}_{el}(x) = h(x)\hat{d}^\dagger\hat{d} + \sum_{k=1}^N \epsilon_k \hat{c}_k^\dagger \hat{c}_k + \sum_{k=1}^N V_k(x)(\hat{d}^\dagger \hat{c}_k + \hat{c}_k^\dagger \hat{d})$$

$$h(x) = U_1(x) - U_0(x)$$

↑ H- anionic    ↑ H neutral

Independent Electron Surface Hopping (IESH) calculates the hopping probability between electronic states and propagates the dynamics accordingly.



[2] L. Xuexun *et al.* J. Chem. Phys. 2026 164, 024707