

## Oxidative stability of chia (Salvia hispanica L.) and sesame (Sesamum indicum L.) oil blends

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Complete List of Authors:	Rodriguez, Gilbert; Universidad Nacional del Santa, docente Villanueva, Eudes; Universidad Nacional Agraria La Molina Cortez, Danco; Universidad Nacional del Santa Sanchez, Esther; Universidad Nacional del Santa Aguirre, Elza; universidad nacional del santa Hidalgo, Alyssa; Universita degli Studi di Milano, DeFENS
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0 7 8	3	<b>Running title</b> : Stability of chia and sesame oil blends
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11 12	5	Gilbert Rodríguez <sup>1,*</sup> . Eudes Villanueva <sup>2</sup> . Danco Cortez <sup>1</sup> . Esther Sanchez <sup>1</sup> . Elza Aguirre <sup>1</sup> . Alvssa
13 14	6	Hidalgo <sup>3</sup>
15 16	-	mungo
17 18	/	
19 20	8	<sup>1</sup> Universidad Nacional del Santa, Urb. Bellamar s/n, Chimbote, Peru.
21 22 22	9	<sup>2</sup> Universidad Nacional Agraria La Molina, Av. La Molina, Lima, Peru.
23 24 25	10	<sup>3</sup> Department of Food, Environmental and Nutritional Sciences (DeFENS), University of Milan, Via
26 27	11	Celoria 2, Milan, Italy.
28 29	12	
30 31	13	
32 33	14	* Corresponding author: giropape@yahoo.com (G. Rodriguez)
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19	Abstract
20	Chia and sesame oils are important sources of essential fatty acids; however, their $\omega$ -3: $\omega$ -6 proportions
21	do not comply with nutritional recommendation. A feasible approach to improve the ratio is to blend
22	different oils, but only after understanding physical and chemical changes of the new matrix. Objective
23	of the investigation was to determine the physico-chemical characteristics and the oxidative stability
24	index (OSI), using the Rancimat method, of chia-sesame oil blends. The four $\omega$ -3: $\omega$ -6 blends tested
25	(1:4, 1:6, 1:8 and 1:10) were exposed to temperatures of 110, 120 and 130 °C. The OSI values of the
26	mixtures varied between 6.24-8.08 h, 3.07-4.00 h and 1.62-2.01 h for each temperature, respectively.
27	In addition, their mean activation energy, enthalpy, entropy and $Q_{10}$ were 88.4 kJ/mol, 85.2 kJ/mol, -
28	41.1 J/mol K and 2.0. Finally, a shelf life prediction performed at 25 °C indicated stability times
29	between 80 and 123 days. Therefore, combining chia and sesame oils produced blends with a good
30	balance of essential fatty acids.
31	
32	Keywords: chia, sesame, $\omega$ -3: $\omega$ -6 ratio, oil blends, oxidation stability

#### 33 Introduction

Nutritional features, quality and stability of oils can be modified by hydrogenation, interesterification, fractionation and blending. Hydrogenation leads to trans isomers formation, interesterification and fractionation need special and expensive equipment, therefore blending oils with different compositions and properties represents the simplest method to achieve appropriate oil characteristics (Hashempour-Baltork et al., 2016). An edible oil blend is obtained mixing two or more oils (in a proportion greater than 5%) from different vegetable species (Guiotto et al., 2014). The mixing of different vegetable oils changes the fatty acids profile and may improve the nutritional and functional value. An increase in monounsaturated fatty acids (MUFA) and polyunsaturated fatty acids (PUFA) intake contributes to reduce the risk of coronary heart disease, because  $\omega$ -3 and  $\omega$ -9 fatty acids have anti-inflammatory properties (Ramsden et al., 2013). However, previous research suggests that only ω-3 has a significant effect in the prevention of cardiovascular diseases (Griffin, 2008). Additionally, unsaturated oils with high MUFA and PUFA content are more prone to oxidation; therefore, properly balancing MUFA and PUFA may give oil blends with better nutritional value, high storage stability and even suitable for frying (Adhvaryu et al., 2000; Ramsden et al., 2013).

The  $\omega$ -3 and  $\omega$ -6 fatty acids play fundamental but different roles in the structure of the cells membrane. The different numbers and positions of double bonds in the chain give the fatty acids different physiological properties, making the relationship between  $\omega$ -3: $\omega$ -6 fatty acids in the diet very important. Linoleic acid is metabolized to arachidonic acid and  $\alpha$ -linolenic acid gives eicosapentaenoic acid (EPA) and docosahexaenoic acid (DHA), all of which use the same metabolic pathways and compete for the same elongase and desaturase enzymes, thus  $\omega$ -3: $\omega$ -6 balance in the diet is important (Huerta-Yépez et al., 2016).

The international nutrition and food committees convened by FAO/WHO established that fat in general should not contribute more than 30% of the total calories consumed by adults; in addition, the consumption should be evenly distributed between saturated fatty acids (SFA), MUFA and PUFA (1:1:1 ratio) (FAO, 2012). Simopoulos (2002) recommended a  $\omega$ -3: $\omega$ -6 rate inferior to 1:4 to reduce

the risks of chronic diseases. Gomes et al. (2019) evaluated the association of plasma and erythrocyte  $\omega$ -3: $\omega$ -6 fatty acids with multiple oxidative stress biomarkers in breast cancer patients; they found that  $\omega$ -3: $\omega$ -6 ratio (plasma) was associated with the anti-inflammatory factor.

Chia seeds have 28–32% oil, which presents the highest essential fatty acid content of any known vegetable source and other valuables components such as tocopherols, polyphenols, phytosterols, carotenoids, and phospholipids. The predominant fatty acids in chia oil are  $\alpha$ -linolenic acid and linoleic acid, which comprise about 64% and 20%, respectively. The  $\omega$ -3: $\omega$ -6 ratio of this oil is approximately 3:1, markedly higher than that of most vegetable oils (Julio et al., 2019).

Sesame seed, containing about 50% oil, is one of the oldest and most important oleaginous matrices, with an annual worldwide oil production of 1.63 million tons (FAOSTAT, 2014). The oil extracted from sesame is mainly used in China and Korea as a condiment oil along with fragrant peanut, safflower, perilla and red pepper oils (Ji et al., 2019). Sesame oil has been reported to contain ~80% unsaturated fatty acids (~42.36% linoleic acid) and many bioactive components including tocopherols, phytosterols and lignans (including sesamolin, sesamin, and sesamol) (Dossa et al., 2018). Sesame oil has very low content of  $\omega$ -3 and high content of  $\omega$ -6 fatty acids, thus to have a better combination of essential fatty acids their ratio should be balanced. For example, Hashempour-Baltork et al. (2017; 2018) proposed blending sesame oil with olive and flaxseed oils.

The widely used Rancimat method, an international standard performed under accelerated storage conditions at high temperatures (AOCS, 1998), is reliable, reproducible, does not involve reagents consumption and its measurements can be easily automated (Heidarpour and Farhoosh, 2018).

The main objective of this work was to determine, using the Rancimat method, the oxidative stability of chia and sesame oils blends with different  $\omega$ -3: $\omega$ -6 proportions. Additionally, activation energy (*Ea*), enthalpy ( $\Delta H^{++}$ ), entropy ( $\Delta S^{++}$ ), energy of Gibbs ( $\Delta G^{++}$ ), Q<sub>10</sub> and shelf life of the blends were computed using the oxidative stability index values and extrapolating the results to standard storage temperatures.

85 Materials and methods

86 Materials

*Oils extraction* 

The chia seeds (cv. Negra) were from the Arequipa region, Peru (15°52'S and 72°15'O) and the sesame seeds (cv. Negra) were from the Chanchamayo province, Junín, Peru (11°03'16"S and 75°19'45"W). The seeds had a moisture content of 8.13±0.12% and 10.43±0.07%, respectively. The oils were extracted by cold pressing using an Expeller Thor (Santa Maria, Brasil) with FA57/G Press (Sew-Eurodrive GmbH & Co KG, Bruchsal, Germany) at a screw speed of 35-40 rpm. The oils were stored at 4.00±0.5 °C in dark flasks under nitrogen atmosphere.

95 Preparation of the mixtures

The oil blends were formulated in order to obtain  $\omega$ -3: $\omega$ -6 ratios of 1:4, 1:6, 1:8 and 1:10. The formulations (Table 1, expressed as percentage of each oil type) were calculated considering the content of  $\omega$ -3 (linolenic acid) and  $\omega$ -6 (linoleic acid) fatty acids, obtained by gas chromatography (see below), for each oil type reported in Table 2. The oils amounts (weight:weight) were thoroughly mixed for 30 s using a vortex (Velp Scientifica, Italy).

2 Methods

Physico-chemical characterization

Acidity, peroxide value (PV), expressed as milliequivalents of peroxides per kilogram of oil (mequiv  $O_2/kg$  oil) and *p*-anisidine value (*p*-AV) were determined according to AOCS methods Cd 3d-63, Cd 8-53 and Cd 18-90, respectively (AOCS, 1998). The total oxidation value (TotOx) was calculated from PV and *p*-AV as follows: TotOx = 2 PV + *p*-AV. The density was determined with an automatic digital densimeter (DOM 2911, Rudolph Research Analytical, USA). The iodine value was measured according to AOAC Official Methods 920.158 (AOAC, 2005). The fatty acid composition

of the chia and sesame oils as well as of their blends was determined by gas chromatography according
to method n. 991.39 (AOAC, 2005).

113 Oxidative stability index OSI

The oxidative stability index (OSI) of each oil and of the blends was evaluated by the method AOCS Cd 12b-92 (AOCS, 1998) using a 743 Rancimat equipment (Metrohm Schweiz AG, Zofigen, Switzerland). The assays were carried out using  $3.0 \pm 0.1$  g of oil sample with an air flow of 15 L/h at 90, 100 and 110 °C for chia oil (Villanueva et al., 2017) and at 110, 120 and 130 °C for sesame oil (Villanueva et al., 2013) and blends. The temperatures were chosen according to the nature of the oil and its resistance to oxidation; a very high temperature originates very short analysis times and a very low temperature would require many hours of study (Bodoira et al., 2017; Martínez et al., 2015; Prasad-Timilsena et al., 2016). The oxidative stability index (OSI) was expressed in hours.

#### 23 Thermodynamic analysis

The activation energy (*Ea*) was determined from the slope of the line representing the natural logarithm of the OSI values versus the inverse of the absolute temperature (1/T) (Villanueva et al., 2013).

$$Ln(OSI) = Ln\left(\frac{-Ln(1-\alpha^*)}{Z}\right) + \frac{E_a}{RT}$$

where  $\alpha^*$  represents the degree of transformation of unsaturated molecules for the induction time, R is the universal gas constant (8.314 J/mol K) and Z is the pre-exponential factor of the Arrhenius equation. The enthalpy ( $\Delta H^{++}$ ) and entropy ( $\Delta S^{++}$ ) of activation were obtained by regression of the logarithm of (K/T) vs. (1/T) (Heidarpour and Farhoosh, 2018):

$$Log\left(\frac{K}{T}\right) = Log\left(\frac{k_B}{h}\right) + \left(\frac{\Delta S^{++}}{2.303R}\right) - \left(\frac{\Delta H^{++}}{RT}\right)$$

where K is the inverse of OSI,  $k_B$  is Boltzmann constant (1.380658 x 10<sup>-23</sup> J/K) and h is Planck's constant (6.6260755 x 10<sup>-34</sup> Js).

1 2		
2 3 4	134	The Gibbs free energy $(\Delta G^{++})$ was calculated according to the Gibbs equation:
5 6	135	$\Delta G^{++} = \Delta H^{++}$ - T $\Delta S^{++}$
7 8	136	The prediction of the shelf life was determined by extrapolation of the linear correlation of the
9 10 11	137	logarithm of OSI vs. T for a 25 °C temperature (Heidarpour and Farhoosh, 2018):
12 13	138	$Log(OSI) = \alpha(T) + \beta$
14 15	139	While the $Q_{10}$ number, which indicates the increase in reaction rate due to a 10 °C rise in
16 17 18	140	temperature, was computed as: $\frac{\text{OSI at time T}}{\text{OSI at T} + 10 ^{\circ}\text{C}}$ (Farhoosh, 2017)
19 20 21	141	
22 23	142	Statistical analysis
24 25	143	All analyses were carried out in triplicate and the data were subjected to one-way analysis of variance
26 27 28	144	(ANOVA). When significant differences were found (p≤0.05), Fisher's lowest significant difference
29 30	145	(LSD) at 95% significance level was computed. The analyses were performed using the Statgraphics®
31 32	146	Centurion XVI statistical program (Statpoint Technologies, Inc., Warrenton, Virginia, USA). The
33 34 25	147	average values and the standard deviation were calculated using the Excel program (Microsoft® Office
35 36 37	148	Excel 2016).
38 39	149	
40 41	150	Results and discussion
42 43 44	151	
45 46	152	Fatty acids characterization
47 48	153	Chia oil composition (Table 2) was 12.3% SFA ( $C_{16:0} + C_{18:0}$ ), 7.6% MUFA ( $C_{18:1}$ ) and 82.7% PUFA
49 50 51	154	$(C_{18:2} + C_{18:3})$ . The $\alpha$ -linolenic acid $(C_{18:3}; \omega$ -3) represented 63.5% of all fatty acids, a proportion very
52 53	155	similar to those (63.26%, 65.2% and 61.8%) observed by Bodoira et al. (2017), Guiotto et al. (2014)
54 55	156	and Villanueva et al. (2017), respectively. Sesame oil composition was 12.3% SFA, 39.5% MUFA and
56 57 58 59	157	45.2% PUFA. Linoleic acid ( $C_{18:2}$ ; $\omega$ -6) was the predominant fatty acid (44.8% of total), a result close

to 46.0% reported by Dossa et al. (2018), and within the range (36.9-47.9%) indicated by the Codex
Alimentarius (2011).

The  $\omega$ -3: $\omega$ -6 ratios of chia and sesame oils were 3.3:1 and 1:100.3 (Table 2), respectively, well outside the ranges recommended by FAO (2012). The  $\omega$ -3: $\omega$ -6 fatty acid composition of the chiasesame oil blends (Table 2), determined by gas chromatography, were 1:4.0 for the first formulation, 1:6.0 for the second, 1:8.0 for the third and 1:10.0 for the last.

In 80:20 and 90:10 (w:w) sunflower-chia oil blends, Guiotto et al. (2014) reported  $\omega$ -3: $\omega$ -6 ratios of 1:2.68 and 1:5.31, i.e. within FAO (2012) recommendations. Similarly, Ract et al. (2015), in their quest for oil blends able to heal wounds by enzymatic interesterification, mixed sunflower-canola oils at 85:15 (w:w) and canola-flaxseed at 70:30 (w:w), obtaining  $\omega$ -3: $\omega$ -9 ratios of 0.62:1 and 1.24:1.

The chia oil acidity (0.43%), peroxides (0.67 meq  $O_2/kg$ ) and *p*-AV (0.36; Table 2) were similar to those reported by Guiotto et al. (2014) and Ixtaina et al. (2012); similarly, the density (0.870 g/mL) and iodine index (103.3 g/100 g; Table 2) were within the range of virgin and cold pressed oils (Codex Alimentarius, 2011), indicating the high quality of the oil used; sesame oil has density (0.903 g/mL), acidity (0.61%), iodine (108.1 g/100 g) and peroxide (1.14 meq  $O_2/kg$ ) values similar to those reported by Gul et al. (2011). The blends, as expected, showed values within the range of the chia and sesame oils.

#### 6 Oxidative stability index

The OSI of sesame oil, chia oil and blends (Table 3) showed that the rate of autooxidation doubled for each 10 °C increase in temperature. Therefore, the OSI values increased as the temperature decreases from 130 to 90 °C. The high concentrations of unsaturated fatty acids in the samples tested played a fundamental role in their oxidative stability, because high degrees of unsaturation are directly associated with lower OSI (García-Moreno et al., 2013). However, the lowest OSI did not always correspond to the highest PUFA, confirming that other factors, e.g. the presence of tocopherols and

lignans, mainly sesamin and sesamolin, is related to the superior oxidative stability of this oil and the beneficial physiological effects of sesame (Hashempour-Baltork et al., 2018).

The OSI values of sesame oil (Table 3) were similar to those reported by Villanueva et al. (2013), i.e. 11.37 and 2.42 h at 110 and 130 °C. The OSI values for chia oil were similar to the data (1.49 h for 110 °C) by Villanueva et al. (2017) and the values (1.4 h for 110 °C) by Martínez et al. (2015). Lower OSI values (2.4 h at 100 and 3.2 h at 90 °C), but at a 20 L/h flow, were reported by González et al. (2016) and by Prasad-Timilsena et al., (2016), respectively.

Table 3 also shows the OSI of the oil blends. An increase in OSI due to the addition of sesame oil is observed; this stabilizing function can be attributed to the augment of linoleic acid and the reduction of linolenic acid in the oil blends. In trials of olive oil adulteration with palm oil or sunflower oil, Heidarpour and Farhoosh (2018) observed an improved OSI value as a consequence of the enrichment of the SFA and MUFA from palm oil. In sunflower-chia oil blends (80:20 and 90:10 w:w) with ω-3:00-6 ratios of 1:32.7 and 1:5.3, respectively, at 98.5 °C and a 20 L/h flow, Guiotto et al. (2014) observed OSI values of 7.6 and 9.2 h and attributed the OSI boost to an increase in linoleic acid supplied by sunflower oil. Z.C

### Thermodynamic study

The activation energy indicates the delay of the initial oxidation process due to the cleavage of the fatty acid chain junction that forms primary oxidation products. It has been suggested that Ea is influenced by the degree of oil polyunsaturation, so that a high content of linoleic ( $\omega$ -6) and/or linolenic acids  $(\omega$ -3) should decrease the *Ea* of lipid oxidation while a high oleic acid contents should increase it; on the contrary, an increase in saturated fatty acids content should improve the resistance to the initial thermal break (Adhvarvu et al., 2000). However, there are exceptions such as sacha inchi oil, whose degree of unsaturation and Ea values are both high, mainly because of the antioxidant activity that tocopherols exert in this oil matrix (Rodríguez et al., 2015).

2 3 As expected, Table 4 shows a greater influence of the unsaturated fatty acids on chia oil Ea (82.0 208 4 5 kJ/mol) than on sesame oil Ea (96.2 kJ/mol). The sesame oil Ea is similar to those (97.28, 98.79 and 209 6 7 96.86 kJ/mol) reported by Villanueva et al. (2013) using three different air flows (15, 20 and 25 L/h, 8 210 9 10 211 respectively). The chia oil *Ea* is similar to the value reported by Villanueva et al. (2017; 81.98 kJ/mol) 11 12 but higher than the values observed by Guiotto et al. (2014) and Ixtaina et al. (2012), i.e. 71.95 and 212 13 14 <sub>15</sub> 213 69.50 kJ/mol, respectively. The Ea of the new formulations obeys to the fatty acids nature, therefore it 16 17 214 can be inferred that their increased  $\omega$ -3: $\omega$ -6 ratio will result in an oxidation faster than sesame oil and 18 19 a stability higher than chia oil. The  $\Delta H^{++}$  and  $\Delta S^{++}$  values showed a significant difference (p $\leq 0.05$ ) 215 20 21 <sub>22</sub> 216 among treatments. The positive sign of the activation enthalpy ( $\Delta H^{++}>0$ ) reflected the endothermic 23 24 217 nature of the formation of the activated complex (Farhoosh and Hoseini-Yazdi, 2014). Table 4 shows 25 <sup>26</sup> 218 that sesame oil during the auto-oxidation absorbed more heat than all the mixtures including chia oil, 27 28 an expected result since OSI was highest in sesame oil at 110 °C, as reported in Table 3. Rancimat-219 29 30 31 220 based studies in olive oil have  $\Delta H^{++}$  values between 91.00 and 103.60 kJ/mol (Heidarpour and 32 33 221 Farhoosh, 2018), in canola oil 86.78 kJ/mol, in soy oil 89.20 kJ/mol, in maize oil 84.92 kJ/mol and in 34 35 222 sunflower oil 87.52 kJ/mol (Farhoosh et al., 2008). On the other hand, the activation entropy resulted 36 37 <sub>38</sub> 223 negative ( $\Delta S^{++} < 0$ ) with values from -44. 4 J/mol K in chia oil to -23.7 J/mol K in sesame oil. Negative 39 entropy values suggest that the activated complexes are more ordered than their reactants, and high 40 224 41 <sup>42</sup> 225 negative values imply that fewer species are involved in the activated complex state. In other words, 43 44 45<sup>226</sup> the activated complex will have a lower potential for the oxidation reaction and, therefore, a slower 46 rate. Studies in olive oil show  $\Delta S^{++}$  between -78.8 and -95.4 J/mol K (Heidarpour and Farhoosh, 2018) 47 227 48 49 228 and between -69.1 and -70.9 J/mol K (Farhoosh and Hoseini-Yazdi, 2014). In other vegetable oils they 50 51 229 vary: in canola -112.99 J/mol K, in soybeans -104.35 J/mol K, in maize -112.28 J/mol K and in 52 53 sunflower -107.73 J/mol K (Farhoosh et al., 2008). Finally, the energy of Gibbs was always positive 54 230 55 56 231  $(\Delta G^{++})$ , suggesting that in Rancimat the autooxidation process is not spontaneous at different 57 58 temperatures; sesame oil had a higher  $\Delta G^{++}$  value (83.7 kJ/mol) than the other oil samples, showing 232 59 60 233 lower oxidation reaction rates.

Shelf life

Table 5 shows the linear relationship between temperature and Log(OSI) for each chia-sesame oil

blend, starting from the results of Table 5, where the  $\alpha$  values vary between -0.0293 and -0.0326.

Villanueva et al. (2013) reported similar values ( $\alpha$ : -0.031 to -0.032) using sesame oil under different

air flows (15, 20 and 25 L/h). The Q<sub>10</sub> values ranged between 1.96 and 2.12, as in other vegetable oils

such as soybean (1.99-2.09) (Farhoosh, 2007), olive (2.09-2.48) (Farhoosh and Hoseini-Yazdi, 2014),

maize (2.00-2.01) and canola (2.00-2.01) (Farhoosh et al., 2008). All these Q<sub>10</sub> around 2.0 represent

the temperature acceleration factor, indicating that the reaction speed doubles for every 10 °C of

Table 5 also shows the extrapolation values for the determination of the shelf life of the blends at

25 °C. All the oil samples present a significant difference ( $p \le 0.05$ ), reflecting the strong influence of

the unsaturated fatty acids of chia oil; as expected, the OSI25 of the four chia-sesame oil blends (from

80.4 to 123.2 days) fell between the 25.8 days of chia oil and the 253.3 of sesame oil. The Rancimat

extrapolation method has been successfully applied in the study of different oils, such as sacha inchi

(Rodríguez et al., 2015); interestingly, soybean oil, whose  $\omega$ -3: $\omega$ -6 ratio is around 1:6-7 (Kulkarni et

al., 2017), shows a shelf life between 125 and 149 days (Farhoosh, 2007), similar to that of our 1:10

blend (Table 5). The shelf life was also determined under accelerated conditions, using combined

models such as a regression model in storage at 50 °C and high temperatures (100-130 °C) which got

a shelf life for olive oil between 1.14 and 1.63 years (Farhoosh and Hoseini-Yazdi, 2013). Finally,

Guiotto et al. (2014) for the shelf life of chia oil, sunflower oil and their blends, tested under non-

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## accelerated storage conditions at 20 °C, reported values between 120 and 240 days for a peroxide index

not exceeding 10 meqv O2/kg.

temperature increase (Farhoosh and Hoseini-Yazdi, 2014).

258 Conclusions

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The results of this study show that the OSI of the chia-sesame oil blends, maintaining a  $\omega$ -3: $\omega$ -6 ratio between 1:4 and 1:10, increased with the addition of sesame oil and decreased with that of chia oil. These results are directly related to the composition of the predominant fatty acids in the oils, which are  $\alpha$ -linolenic in chia and linoleic in sesame. The process of accelerated oxidation of chia-sesame oil blends allowed to estimate the shelf life by extrapolation: the OSI obtained were outside the range of the experimental values at 25 °C, giving stability times ranging between 80 and 123 days. The thermodynamic behaviour was studied by calculating activation energy (86.7 - 89.1 kJ/mol), enthalpy (83.4 - 86.1 kJ/mol) and entropy (-44.3 - -39.1 J/mol) of the oxidation reaction.

The oil obtained from the mixture of chia and sesame oils has a better proportion of  $\omega$ -3: $\omega$ -6 fatty acids, with a good oxidative stability that fluctuates between 6 to 8 h at 110 °C and a shelf life at 25 °C between 80 and 123 days.

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Table 1. Formulation of sesame and chia oil blends and their omegas ratios.

% oil (w/w)		Omegas ratio
Chía	Sesame	( <b>ω-3:ω-6</b> )
15.5	84.5	1:4
10.4	89.6	1:6
7.7	92.3	1:8
6.1	93.9	1:10

Table 2. Physico-chemical characteristics of chia oil, sesame oil and their blends. Mean value  $\pm$  SD

(n=3).

D (71	Chia	Sesame -	$\omega$ -3: $\omega$ -6 ratios of chia–sesame oil blends					
Profile			1:4	1:6	1:8	1:10		
Fatty acid %								
C <sub>16:0</sub>	8.31e±0.03	8.32 <sup>d</sup> ±0.01	8.55°±0.01	8.63 <sup>b</sup> ±0.01	8.37 <sup>d</sup> ±0.01	8.73 <sup>a</sup> ±0.02		
C <sub>18:0</sub>	4.02°±0.02	4.02°±0.02	4.04°±0.04	4.20 <sup>b</sup> ±0.02	4.39ª±0.06	4.04°±0.09		
C <sub>18:1</sub>	7.55 <sup>e</sup> ±0.07	39.47ª±0.47	36.65 <sup>d</sup> ±0.32	37.19°±0.25	38.71 <sup>b</sup> ±0.03	38.90 <sup>b</sup> ±0.10		
$C_{18:2}$ ( $\omega$ -6)	19.18°±0.85	44.76 <sup>a</sup> ±0.16	40.33 <sup>b</sup> ±2.85	43.19 <sup>a</sup> ±0.23	43.52ª±0.45	43.87 <sup>a</sup> ±0.64		
C <sub>18:3</sub> (ω-3)	63.49ª±0.17	0.45e±0.05	10.16 <sup>b</sup> ±0.73	7.41°±1.29	5.44 <sup>d</sup> ±0.38	4.41 <sup>d</sup> ±0.33		
ω-3:ω-6	3.3±0.2:1	1:100.3±11.4	1:4.0±0.4	1:6.0±1.1	1:8.0±0.6	1:10.0±0.7		
Density (g/mL)	$0.870^{\circ}\pm0.00$	$0.903^{a}\pm0.01$	$0.873^{\circ}\pm0.01$	$0.877^{c}\pm 0.01$	$0.890^{b}\pm 0.00$	$0.897^{ab}\pm0.01$		
Acidity (%)	$0.43^{d}\pm 0.05$	$0.61^{a}\pm0.02$	$0.51^{\circ}\pm0.04$	$0.53^{bc} \pm 0.02$	$0.55^{bc}\pm 0.01$	$0.58^{ab}\pm 0.01$		
Iodine $(g/100 g)$	103.3 <sup>d</sup> ±0.6	$108.1^{b}\pm1.0$	$105.4^{bcd} \pm 1.3$	$106.4^{bc}\pm 2.6$	112.6ª±1.2	$104.4^{cd} \pm 1.8$		
PV (meq $O_2/kg$ )	0.67°±0.15	1.14ª±0.04	$0.98^{b}\pm0.03$	$1.02^{ab}\pm 0.03$	1.05 <sup>ab</sup> ±0.04	1.10 <sup>ab</sup> ±0.03		
<i>p</i> -AV	$0.36^{d}\pm0.05$	$0.86^{a}\pm0.02$	$0.48^{c}\pm0.05$	0.53°±0.03	$0.73^{b}\pm 0.03$	0.81ª±0.02		
TotOx	$1.69^{d}\pm 0.26$	3.14 <sup>a</sup> ±0.07	$2.44^{c}\pm0.08$	2.57°±0.08	2.83 <sup>b</sup> ±0.10	$3.01^{ab}\pm0.05$		
8 Different letters in the same row indicate significant differences between samples ( $p \le 0.05$ ).								

3 Table 3. Oxidative stability index (OSI; h) of chia oil, sesame oil and their blends. Mean value  $\pm$  SD 389 4 5 390 (n=3). 6 7 8  $\omega$ -3: $\omega$ -6 ratios of chia–sesame oil blends 9 Temperature Chia Sesame 1:4 1:6 1:8 1:10 10  $(^{\circ}C)$ **OSI±SD** 11 **OSI±SD OSI±SD OSI±SD OSI±SD OSI±SD** 12 130 2.33<sup>a</sup>±0.02  $1.62^{e}\pm 0.05$  $1.82^{d} \pm 0.02$  $1.90^{\circ}\pm 0.03$ 2.01<sup>b</sup>±0.01 13 120  $4.74^{a}\pm0.12$  $3.07^{e} \pm 0.05$  $3.52^{d} \pm 0.09$ 3.75°±0.08  $4.00^{b} \pm 0.10$ \_ 14 110 10.45<sup>a</sup>±0.25  $6.24^{e}\pm0.04$  $7.22^{d} \pm 0.13$ 7.65°±0.34  $8.08^{b}\pm 0.08$ 15  $1.49^{f} \pm 0.00$ 16 100 \_  $3.03 \pm 0.05$ \_ -\_ 17 90  $6.16 \pm 0.01$ -18 19 391 Different letters in the same row indicate significant difference between samples ( $p \le 0.05$ ). 20 21 392 22 23 24 393 25 26 394 27 28 395 Table 4. Activation energy (*Ea*), enthalpy  $(\Delta H^{++})$ , entropy  $(\Delta S^{++})$ , energy of Gibbs  $(\Delta G^{++})$  of chia 29 30 396 oil, sesame oil and their blends. 31 32 33 34  $\Delta H^{++}$  (kJ/mol) Oil samples  $\Delta S^{++}$  (J/mol K)  $\Delta G^{++}$  (kJ/mol) *Ea* (kJ/mol) 35 Chia 82.0<sup>d</sup>±0.1  $78.9^{d}\pm0.1$ -44.4<sup>b</sup>±0.1 36 62.3°±0.1 37  $96.2^{a}\pm 2.0$ Sesame  $93.0^{a}\pm 2.0$  $-23.7^{a}\pm 5.0$ 83.7<sup>a</sup>±3.9 38 Blends ( $\omega$ -3: $\omega$ -6 ratios) 39 1:4 83.4<sup>c</sup>±1.7 -44.3<sup>b</sup>±4.5  $66.0^{bc} \pm 3.5$ 86.7<sup>c</sup>±1.7 40 41 88.6<sup>bc</sup>±0.9 85.3<sup>bc</sup>±0.9 1:6  $-40.7^{b}\pm 2.1$  $69.3^{b}\pm1.7$ 42 1:8 89.4<sup>b</sup>±2.0 86.1<sup>b</sup>±2.0 -39.1<sup>b</sup>±4.8  $70.7^{b}\pm3.9$ 43 1:10 89.1<sup>bc</sup>±0.9 85.9<sup>bc</sup>±0.9  $-40.2^{b}\pm 2.3$  $70.1^{b}\pm1.8$ 44 397 Different letters in the same column indicate significant difference between samples ( $p \le 0.05$ ). 45 46 47 398 48 <sup>49</sup> 399 50 51 52 53 54 55 56 57 58 59 60

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Table 5. Shelf life (oxidative stability index at 25  $^{\circ}$ C, OSI<sub>25</sub>) and Q<sub>10</sub> (increase of reaction rate due to 0

a 10 °C temperature rise) of chia oil, sesame oil and their blends.

Oil samples	$Log OSI = \alpha(T) + \beta$			OSI <sub>25</sub>	Q <sub>10</sub>
on samples	$\alpha \pm SD$	$\beta \pm SD$	R <sup>2</sup>	(days)	
Chia	$-0.0308 \pm 0.0000$	$3.5621 \pm 0.0038$	0.999	25.8 <sup>d</sup> ±0.2	$2.03 \pm 0.00$
Sesame	$-0.0326 \pm 0.0007$	4.5949±0.0773	0.999	253.3ª±36.1	2.12±0.12
Blends ( $\omega$ -3: $\omega$ -6 ratios)					
1:4	-0.0293±0.0006	4.0168±0.0667	0.999	80.4 <sup>c</sup> ±9.8	1.96±0.10
1:6	$-0.0300 \pm 0.0003$	4.1515±0.0409	0.999	105.4 <sup>bc</sup> ±8.3	$1.99 \pm 0.08$
1:8	$-0.0302 \pm 0.0007$	4.2076±0.0862	0.999	118.9 <sup>b</sup> ±18.6	2.01±0.05
1:10	$-0.0302 \pm 0.0003$	4.2242±0.0411	0.999	123.2 <sup>b</sup> ±9.5	2.01±0.02
<b>x</b> and $\beta$ : constants; <b>R</b> <sup>2</sup>	: coefficient of de	termination. Diffe	rent letter	rs in the same	column indic
significant difference bet	tween samples ( $p \leq$	0.05).			