

# **Oxidative stability of chia (Salvia hispanica L.) and sesame (Sesamum indicum L.) oil blends**









### **Abstract**

 Chia and sesame oils are important sources of essential fatty acids; however, their ω-3:ω-6 proportions do not comply with nutritional recommendation. A feasible approach to improve the ratio is to blend different oils, but only after understanding physical and chemical changes of the new matrix. Objective of the investigation was to determine the physico-chemical characteristics and the oxidative stability index (OSI), using the Rancimat method, of chia-sesame oil blends. The four ω-3:ω-6 blends tested 25 (1:4, 1:6, 1:8 and 1:10) were exposed to temperatures of 110, 120 and 130 °C. The OSI values of the mixtures varied between 6.24-8.08 h, 3.07-4.00 h and 1.62-2.01 h for each temperature, respectively. 27 In addition, their mean activation energy, enthalpy, entropy and  $Q_{10}$  were 88.4 kJ/mol, 85.2 kJ/mol, -28 41.1 J/mol K and 2.0. Finally, a shelf life prediction performed at 25 °C indicated stability times between 80 and 123 days. Therefore, combining chia and sesame oils produced blends with a good balance of essential fatty acids.

Ry. **Keywords**: chia, sesame, ω-3:ω-6 ratio, oil blends, oxidation stability

 $\mathbf{1}$  $\overline{2}$ 

# **Introduction**

unsaturated fatty acids (MUFA) and polyd<br>the risk of coronary heart disease, becaus<br>s (Ramsden et al., 2013). However, previously<br>in the prevention of cardiovascular disease<br>IUFA and PUFA content are more prone to<br>a may g Nutritional features, quality and stability of oils can be modified by hydrogenation, interesterification, fractionation and blending. Hydrogenation leads to trans isomers formation, interesterification and fractionation need special and expensive equipment, therefore blending oils with different compositions and properties represents the simplest method to achieve appropriate oil characteristics (Hashempour-Baltork et al., 2016). An edible oil blend is obtained mixing two or more oils (in a proportion greater than 5%) from different vegetable species (Guiotto et al., 2014). The mixing of different vegetable oils changes the fatty acids profile and may improve the nutritional and functional value. An increase in monounsaturated fatty acids (MUFA) and polyunsaturated fatty acids (PUFA) intake contributes to reduce the risk of coronary heart disease, because ω-3 and ω-9 fatty acids have anti-inflammatory properties (Ramsden et al., 2013). However, previous research suggests that only ω-3 has a significant effect in the prevention of cardiovascular diseases (Griffin, 2008). Additionally, unsaturated oils with high MUFA and PUFA content are more prone to oxidation; therefore, properly balancing MUFA and PUFA may give oil blends with better nutritional value, high storage stability and even suitable for frying (Adhvaryu et al., 2000; Ramsden et al., 2013).

 The ω-3 and ω-6 fatty acids play fundamental but different roles in the structure of the cells membrane. The different numbers and positions of double bonds in the chain give the fatty acids different physiological properties, making the relationship between ω-3: ω-6 fatty acids in the diet very 51 important. Linoleic acid is metabolized to arachidonic acid and  $\alpha$ -linolenic acid gives eicosapentaenoic acid (EPA) and docosahexaenoic acid (DHA), all of which use the same metabolic pathways and compete for the same elongase and desaturase enzymes, thus ω-3: ω-6 balance in the diet is important (Huerta-Yépez et al., 2016).

 The international nutrition and food committees convened by FAO/WHO established that fat in general should not contribute more than 30% of the total calories consumed by adults; in addition, the consumption should be evenly distributed between saturated fatty acids (SFA), MUFA and PUFA (1:1:1 ratio) (FAO, 2012). Simopoulos (2002) recommended a ω-3:ω-6 rate inferior to 1:4 to reduce

 the risks of chronic diseases. Gomes et al. (2019) evaluated the association of plasma and erythrocyte 60  $\omega$ -3: $\omega$ -6 fatty acids with multiple oxidative stress biomarkers in breast cancer patients; they found that 61  $\omega$ -3: $\omega$ -6 ratio (plasma) was associated with the anti-inflammatory factor.

 Chia seeds have 28–32% oil, which presents the highest essential fatty acid content of any known vegetable source and other valuables components such as tocopherols, polyphenols, phytosterols, carotenoids, and phospholipids. The predominant fatty acids in chia oil are α-linolenic acid and linoleic 65 acid, which comprise about 64% and 20%, respectively. The  $\omega$ -3: $\omega$ -6 ratio of this oil is approximately 3:1, markedly higher than that of most vegetable oils (Julio et al., 2019).

Example 18 and 50% oil, is one of the oldest a<br>orldwide oil production of 1.63 million to<br>nly used in China and Korea as a condimen<br>ppper oils (Ji et al., 2019). Sesame oil has<br>36% linoleic acid) and many bioactive com<br>cl Sesame seed, containing about 50% oil, is one of the oldest and most important oleaginous matrices, with an annual worldwide oil production of 1.63 million tons (FAOSTAT, 2014). The oil extracted from sesame is mainly used in China and Korea as a condiment oil along with fragrant peanut, safflower, perilla and red pepper oils (Ji et al., 2019). Sesame oil has been reported to contain ~80% unsaturated fatty acids (~42.36% linoleic acid) and many bioactive components including tocopherols, phytosterols and lignans (including sesamolin, sesamin, and sesamol) (Dossa et al., 2018). Sesame oil has very low content of ω-3 and high content of ω-6 fatty acids, thus to have a better combination of essential fatty acids their ratio should be balanced. For example, Hashempour-Baltork et al. (2017; 2018) proposed blending sesame oil with olive and flaxseed oils.

 The widely used Rancimat method, an international standard performed under accelerated storage conditions at high temperatures (AOCS, 1998), is reliable, reproducible, does not involve reagents consumption and its measurements can be easily automated (Heidarpour and Farhoosh, 2018).

 The main objective of this work was to determine, using the Rancimat method, the oxidative stability of chia and sesame oils blends with different ω-3:ω-6 proportions. Additionally, activation 81 energy (*Ea*), enthalpy ( $\Delta H^{++}$ ), entropy ( $\Delta S^{++}$ ), energy of Gibbs ( $\Delta G^{++}$ ),, Q<sub>10</sub> and shelf life of the blends were computed using the oxidative stability index values and extrapolating the results to standard storage temperatures.

  **Materials and methods**

**Materials**

*Oils extraction* 

 The chia seeds (cv. Negra) were from the Arequipa region, Peru (15º52'S and 72º15'O) and the sesame 89 seeds (cv. Negra) were from the Chanchamayo province, Junín, Peru (11°03'16"S and 75°19'45"W). The seeds had a moisture content of 8.13±0.12% and 10.43±0.07%, respectively. The oils were extracted by cold pressing using an Expeller Thor (Santa Maria, Brasil) with FA57/G Press (Sew- Eurodrive GmbH & Co KG, Bruchsal, Germany) at a screw speed of 35-40 rpm. The oils were stored 93 at  $4.00\pm0.5$  °C in dark flasks under nitrogen atmosphere.

## *Preparation of the mixtures*

For example atmosphere.<br>
Independent atmosphere.<br>
Indeed in order to obtain  $\omega$ -3: $\omega$ -6 ratios cressed as percentage of each oil type) w<br>
I) and  $\omega$ -6 (linoleic acid) fatty acids, obtains<br>
orted in Table 2. The oils am The oil blends were formulated in order to obtain ω-3: ω-6 ratios of 1:4, 1:6, 1:8 and 1:10. The formulations (Table 1, expressed as percentage of each oil type) were calculated considering the content of ω-3 (linolenic acid) and ω-6 (linoleic acid) fatty acids, obtained by gas chromatography (see below), for each oil type reported in Table 2. The oils amounts (weight:weight) were thoroughly mixed for 30 s using a vortex (Velp Scientifica, Italy).

#### **Methods**

*Physico-chemical characterization*

Acidity, peroxide value (PV), expressed as milliequivalents of peroxides per kilogram of oil (mequiv O <sup>2</sup>/kg oil) and *p*-anisidine value (*p*-AV) were determined according to AOCS methods Cd 3d- 63, Cd 8-53 and Cd 18-90, respectively (AOCS, 1998). The total oxidation value (TotOx) was 107 calculated from PV and  $p$ -AV as follows: TotOx = 2 PV +  $p$ -AV. The density was determined with an automatic digital densimeter (DOM 2911, Rudolph Research Analytical, USA). The iodine value was measured according to AOAC Official Methods 920.158 (AOAC, 2005). The fatty acid composition

110 of the chia and sesame oils as well as of their blends was determined by gas chromatography according 111 to method n. 991.39 (AOAC, 2005).

**Oxidative stability index OSI** 10 113

I blends. The temperatures were chosen ac<br>
in; a very high temperature originates very<br>
re many hours of study (Bodoira et al., 201<sup>o</sup><br>
oxidative stability index (OSI) was express<br>
vas determined from the slope of the lin 114 The oxidative stability index (OSI) of each oil and of the blends was evaluated by the method AOCS 115 Cd 12b-92 (AOCS, 1998) using a 743 Rancimat equipment (Metrohm Schweiz AG, Zofigen, Switzerland). The assays were carried out using  $3.0 \pm 0.1$  g of oil sample with an air flow of 15 L/h at 117 90, 100 and 110 °C for chia oil (Villanueva et al., 2017) and at 110, 120 and 130 °C for sesame oil (Villanueva et al., 2013) and blends. The temperatures were chosen according to the nature of the oil and its resistance to oxidation; a very high temperature originates very short analysis times and a very low temperature would require many hours of study (Bodoira et al., 2017; Martínez et al., 2015; Prasad-121 Timilsena et al., 2016). The oxidative stability index (OSI) was expressed in hours. 15 115 17 116 22 118 24 119 26 120

122 31 32

48

112

16

23

25

#### 123 *Thermodynamic analysis* 33 123

124 The activation energy (*Ea*) was determined from the slope of the line representing the natural logarithm of the OSI values versus the inverse of the absolute temperature  $(1/T)$  (Villanueva et al., 2013). 38 125

126 
$$
\operatorname{Ln}(OSI) = \operatorname{Ln}\left(\frac{-\operatorname{Ln}(1-\alpha^*)}{Z}\right) + \frac{E_a}{RT}
$$

127 where  $\alpha^*$  represents the degree of transformation of unsaturated molecules for the induction time, R is the universal gas constant  $(8.314 \text{ J/mol K})$  and Z is the pre-exponential factor of the Arrhenius equation. 129 The enthalpy (*ΔH++*) and entropy (*ΔS ++*) of activation were obtained by regression of the logarithm of 130 (K/T) vs. (1/T) (Heidarpour and Farhoosh, 2018): 47 128 49 129

131 
$$
Log\left(\frac{K}{T}\right) = Log\left(\frac{k_B}{h}\right) + \left(\frac{\Delta S^{+ +}}{2.303R}\right) - \left(\frac{\Delta H^{+ +}}{RT}\right)
$$

where K is the inverse of OSI,  $k_B$  is Boltzmann constant (1.380658 x 10<sup>-23</sup> J/K) and h is Planck's constant  $(6.6260755 \times 10^{-34} \text{ Js})$ . 58 132 59 60 133



158 to 46.0% reported by Dossa et al. (2018), and within the range (36.9-47.9%) indicated by the Codex 159 Alimentarius (2011).

160 The ω-3:ω-6 ratios of chia and sesame oils were 3.3:1 and 1:100.3 (Table 2), respectively, well outside the ranges recommended by FAO (2012). The  $\omega$ -3: $\omega$ -6 fatty acid composition of the chia-162 sesame oil blends (Table 2), determined by gas chromatography, were 1:4.0 for the first formulation, 1:6.0 for the second, 1:8.0 for the third and  $1:10.0$  for the last.

In 80:20 and 90:10 (w:w) sunflower-chia oil blends, Guiotto et al. (2014) reported  $\omega$ -3: $\omega$ -6 ratios 165 of 1:2.68 and 1:5.31, i.e. within FAO (2012) recommendations. Similarly, Ract et al. (2015), in their quest for oil blends able to heal wounds by enzymatic interesterification, mixed sunflower-canola oils at 85:15 (w:w) and canola-flaxseed at 70:30 (w:w), obtaining  $\omega$ -3: $\omega$ -9 ratios of 0.62:1 and 1.24:1.

eal wounds by enzymatic interesterificatio<br>axseed at 70:30 (w:w), obtaining  $\omega$ -3: $\omega$ -9<br>3%), peroxides (0.67 meq O<sub>2</sub>/kg) and p-A<br>et al. (2014) and Ixtaina et al. (2012); sim<br>00 g; Table 2) were within the range of virg The chia oil acidity  $(0.43\%)$ , peroxides  $(0.67 \text{ meg } O<sub>2</sub>/kg)$  and p-AV  $(0.36; \text{ Table } 2)$  were similar 169 to those reported by Guiotto et al. (2014) and Ixtaina et al. (2012); similarly, the density (0.870 g/mL) and iodine index (103.3  $g/100$  g; Table 2) were within the range of virgin and cold pressed oils (Codex Alimentarius, 2011), indicating the high quality of the oil used; sesame oil has density (0.903 g/mL), 172 acidity (0.61%), iodine (108.1 g/100 g) and peroxide (1.14 meg  $O_2$ /kg) values similar to those reported by Gul et al. (2011). The blends, as expected, showed values within the range of the chia and sesame oils.

**Oxidative stability index** 

The OSI of sesame oil, chia oil and blends (Table 3) showed that the rate of autooxidation doubled for 178 each 10 °C increase in temperature. Therefore, the OSI values increased as the temperature decreases from 130 to 90  $\degree$ C. The high concentrations of unsaturated fatty acids in the samples tested played a fundamental role in their oxidative stability, because high degrees of unsaturation are directly associated with lower OSI (García-Moreno et al., 2013). However, the lowest OSI did not always correspond to the highest PUFA, confirming that other factors, e.g. the presence of tocopherols and 47 177  $\frac{1}{52}$  179 54 180 56  $\frac{36}{59}$  182

 lignans, mainly sesamin and sesamolin, is related to the superior oxidative stability of this oil and the beneficial physiological effects of sesame (Hashempour-Baltork et al., 2018).

 The OSI values of sesame oil (Table 3) were similar to those reported by Villanueva et al. (2013), i.e. 11.37 and 2.42 h at 110 and 130 °C. The OSI values for chia oil were similar to the data (1.49 h for 187 110 °C) by Villanueva et al. (2017) and the values (1.4 h for 110 °C) by Martínez et al. (2015). Lower OSI values (2.4 h at 100 and 3.2 h at 90 °C), but at a 20 L/h flow, were reported by González et al.  $(2016)$  and by Prasad-Timilsena et al.,  $(2016)$ , respectively.

Function can be attributed to the augment of<br>plends. In trials of olive oil adulteration v<br>018) observed an improved OSI value as a<br>n palm oil. In sunflower-chia oil blends (<br>1:5.3, respectively, at 98.5  $^{\circ}$ C and a 20 Table 3 also shows the OSI of the oil blends. An increase in OSI due to the addition of sesame oil is observed; this stabilizing function can be attributed to the augment of linoleic acid and the reduction of linolenic acid in the oil blends. In trials of olive oil adulteration with palm oil or sunflower oil, Heidarpour and Farhoosh (2018) observed an improved OSI value as a consequence of the enrichment 194 of the SFA and MUFA from palm oil. In sunflower-chia oil blends  $(80:20$  and  $90:10$  w:w) with  $\omega$ -31 195 3: ω-6 ratios of 1:32.7 and 1:5.3, respectively, at 98.5 °C and a 20 L/h flow, Guiotto et al. (2014) observed OSI values of 7.6 and 9.2 h and attributed the OSI boost to an increase in linoleic acid supplied by sunflower oil.

# **Thermodynamic study**

The activation energy indicates the delay of the initial oxidation process due to the cleavage of the fatty acid chain junction that forms primary oxidation products. It has been suggested that *Ea* is influenced by the degree of oil polyunsaturation, so that a high content of linoleic ( ω-6) and/or linolenic acids ( ω-3) should decrease the *Ea* of lipid oxidation while a high oleic acid contents should increase it; on the contrary, an increase in saturated fatty acids content should improve the resistance to the initial thermal break (Adhvaryu et al., 2000). However, there are exceptions such as sacha inchi oil, whose degree of unsaturation and *Ea* values are both high, mainly because of the antioxidant activity that tocopherols exert in this oil matrix (Rodríguez et al., 2015).

tive sign of the activation enthalpy  $(\Delta H^{+})$ <br>e activated complex (Farhoosh and Hosein<br>to-oxidation absorbed more heat than all t<br>I was highest in sesame oil at 110 °C, as<br>have  $\Delta H^{++}$  values between 91.00 and 10<br>iil 86. As expected, Table 4 shows a greater influence of the unsaturated fatty acids on chia oil *Ea* (82.0 kJ/mol) than on sesame oil *Ea* (96.2 kJ/mol). The sesame oil *Ea* is similar to those (97.28, 98.79 and 96.86 kJ/mol) reported by Villanueva et al. (2013) using three different air flows (15, 20 and 25 L/h, respectively). The chia oil *Ea* is similar to the value reported by Villanueva et al. (2017; 81.98 kJ/mol) but higher than the values observed by Guiotto et al. (2014) and Ixtaina et al. (2012), i.e. 71.95 and 69.50 kJ/mol, respectively. The *Ea* of the new formulations obeys to the fatty acids nature, therefore it can be inferred that their increased  $\omega$ -3: $\omega$ -6 ratio will result in an oxidation faster than sesame oil and 215 a stability higher than chia oil. The  $\Delta H^{++}$  and  $\Delta S^{++}$  values showed a significant difference (p≤0.05) among treatments. The positive sign of the activation enthalpy  $(AH^{++}> 0)$  reflected the endothermic nature of the formation of the activated complex (Farhoosh and Hoseini-Yazdi, 2014). Table 4 shows that sesame oil during the auto-oxidation absorbed more heat than all the mixtures including chia oil, an expected result since OSI was highest in sesame oil at 110 °C, as reported in Table 3. Rancimatbased studies in olive oil have  $\Delta H^{++}$  values between 91.00 and 103.60 kJ/mol (Heidarpour and Farhoosh, 2018), in canola oil 86.78 kJ/mol, in soy oil 89.20 kJ/mol, in maize oil 84.92 kJ/mol and in sunflower oil 87.52 kJ/mol (Farhoosh et al., 2008). On the other hand, the activation entropy resulted 223 negative  $(\Delta S^{++} \le 0)$  with values from -44. 4 J/mol K in chia oil to -23.7 J/mol K in sesame oil. Negative entropy values suggest that the activated complexes are more ordered than their reactants, and high negative values imply that fewer species are involved in the activated complex state. In other words, the activated complex will have a lower potential for the oxidation reaction and, therefore, a slower 227 rate. Studies in olive oil show ΔS<sup>++</sup> between -78.8 and -95.4 J/mol K (Heidarpour and Farhoosh, 2018) and between -69.1 and -70.9 J/mol K (Farhoosh and Hoseini-Yazdi, 2014). In other vegetable oils they vary: in canola -112.99 J/mol K, in soybeans -104.35 J/mol K, in maize -112.28 J/mol K and in sunflower -107.73 J/mol K (Farhoosh et al., 2008). Finally, the energy of Gibbs was always positive  $(\Delta G^{+})$ , suggesting that in Rancimat the autooxidation process is not spontaneous at different temperatures; sesame oil had a higher *ΔG++* value (83.7 kJ/mol) than the other oil samples, showing lower oxidation reaction rates. 10 211 15 213 17 214 22 216 24 217 26 218 31 220 33 221 38 223 40 224 42 225 47 227 54 230 



**Shelf life** Table 5 shows the linear relationship between temperature and Log(OSI) for each chia-sesame oil 237 blend, starting from the results of Table 5, where the  $\alpha$  values vary between -0.0293 and -0.0326. 238 Villanueva et al. (2013) reported similar values ( $\alpha$ : -0.031 to -0.032) using sesame oil under different air flows (15, 20 and 25 L/h). The  $Q_{10}$  values ranged between 1.96 and 2.12, as in other vegetable oils such as soybean (1.99-2.09) (Farhoosh, 2007), olive (2.09-2.48) (Farhoosh and Hoseini-Yazdi, 2014), maize (2.00-2.01) and canola (2.00-2.01) (Farhoosh et al., 2008). All these  $Q_{10}$  around 2.0 represent the temperature acceleration factor, indicating that the reaction speed doubles for every 10  $\degree$ C of temperature increase (Farhoosh and Hoseini-Yazdi, 2014).

a factor, indicating that the reaction speeds of and Hoseini-Yazdi, 2014).<br>
Extrapolation values for the determination<br>
resent a significant difference (p $\leq$ 0.05), ref chia oil; as expected, the OSI<sub>25</sub> of the fot<br>
veen Table 5 also shows the extrapolation values for the determination of the shelf life of the blends at 25 °C. All the oil samples present a significant difference ( $p\leq 0.05$ ), reflecting the strong influence of the unsaturated fatty acids of chia oil; as expected, the OSI<sub>25</sub> of the four chia-sesame oil blends (from 80.4 to 123.2 days) fell between the 25.8 days of chia oil and the 253.3 of sesame oil. The Rancimat extrapolation method has been successfully applied in the study of different oils, such as sacha inchi 249 (Rodríguez et al., 2015); interestingly, soybean oil, whose  $\omega$ -3: $\omega$ -6 ratio is around 1:6-7 (Kulkarni et al., 2017), shows a shelf life between 125 and 149 days (Farhoosh, 2007), similar to that of our  $1:10$ blend (Table 5). The shelf life was also determined under accelerated conditions, using combined models such as a regression model in storage at 50  $^{\circ}$ C and high temperatures (100-130  $^{\circ}$ C) which got a shelf life for olive oil between 1.14 and 1.63 years (Farhoosh and Hoseini-Yazdi, 2013). Finally, Guiotto et al. (2014) for the shelf life of chia oil, sunflower oil and their blends, tested under nonaccelerated storage conditions at 20  $^{\circ}$ C, reported values between 120 and 240 days for a peroxide index 256 not exceeding 10 meqv  $O_2/kg$ .

**Conclusions**

 The results of this study show that the OSI of the chia-sesame oil blends, maintaining a ω-3:ω-6 ratio between 1:4 and 1:10, increased with the addition of sesame oil and decreased with that of chia oil. These results are directly related to the composition of the predominant fatty acids in the oils, which are α-linolenic in chia and linoleic in sesame. The process of accelerated oxidation of chia-sesame oil blends allowed to estimate the shelf life by extrapolation: the OSI obtained were outside the range of the experimental values at 25  $\degree$ C, giving stability times ranging between 80 and 123 days. The thermodynamic behaviour was studied by calculating activation energy (86.7 - 89.1 kJ/mol), enthalpy (83.4 - 86.1 kJ/mol) and entropy (-44.3 - -39.1 J/mol) of the oxidation reaction.

The oil obtained from the mixture of chia and sesame oils has a better proportion of  $\omega$ -3: $\omega$ -6 fatty acids, with a good oxidative stability that fluctuates between 6 to 8 h at 110  $^{\circ}$ C and a shelf life at 25  $^{\circ}$ C between 80 and 123 days.

#### **References**

- For Peer Review Adhvaryu, A., Erhan, S., Liu, Z., & Perez, J. (2000). Oxidation kinetic studies of oils derived from unmodified and genetically modified vegetables using pressurized differential scanning calorimetry and nuclear magnetic resonance spectroscopy. *Thermochimica Acta*, 364:87–97. https://doi.org/10.1016/S0040-6031(00)00626-2
- AOAC (2005). Official methods of analysis of the Association of Analytical Chemists International,  $(18th$  ed.) Gathersburg, MD U.S.A.: AOAC International.
- AOCS (1998). Official methods and recommended practices of the AOCS, 5th edn. AOCS Press, Champaign.
- Bodoira, R. M., Penci, C. M., Ribotta, P. D., & Martinez, M. L. (2017). Chia (*Salvia hispanica* L.) oil stability: study of the effect of natural antioxidants. *LWT – Food Science and Technology*, 75:107-113. https://doi.org/10.1016/j.lwt.2016.08.031 54 281

# Codex Alimentarius (2011). Preliminary draft revision of the codex standard for specified vegetable oils. Penang, Malaysia.











Table 1. Formulation of sesame and chia oil blends and their omegas ratios.



Table 2. Physico-chemical characteristics of chia oil, sesame oil and their blends. Mean value  $\pm$  SD

 $(n=3)$ .



388 Different letters in the same row indicate significant differences between samples ( $p \le 0.05$ ).

Ea), enthalpy  $(AH^{++})$ , entropy  $(AS^{++})$ , en<br>
dds.<br>
Ea (kJ/mol)  $AH^{++}$  (kJ/mol)  $AS^{++}$ <br>
82.0<sup>4</sup>±0.1 78.9<sup>4</sup>±0.1 -44<br>
96.2<sup>a</sup>±2.0 93.0<sup>a</sup>±2.0 -22<br>
86.7<sup>c</sup>±1.7 83.4<sup>c</sup>±1.7 -44<br>
88.6<sup>b</sup>e±0.9 85.3<sup>be</sup>±0.9 -41<br>
89.4<sup>b</sup>±2.0 86.1<sup></sup> 389 Table 3. Oxidative stability index (OSI; h) of chia oil, sesame oil and their blends. Mean value  $\pm$  SD (n=3). Different letters in the same row indicate significant difference between samples ( $p \le 0.05$ ).  $\frac{2}{24}$  393 394 Table 4. Activation energy (*Ea*), enthalpy (*ΔH++*), entropy (*ΔS++*), energy of Gibbs (*ΔG++*) of chia oil, sesame oil and their blends. Different letters in the same column indicate significant difference between samples ( $p \le 0.05$ ). 398 49 399 Temperature Chia Sesame  $\frac{\omega-3:\omega-6 \text{ ratios of chia}-\text{sesame oil blends}}{1:4}$   $\frac{1:6}{1:8}$  1:8  $\frac{1:10}{1:10}$  $\frac{1.4}{1.0}$  OSI±SD OSI±SD OSI±SD OSI±SD OSI±SD OSI±SD OSI±SD 130 -  $2.33^{\circ} \pm 0.02$   $1.62^{\circ} \pm 0.05$   $1.82^{\circ} \pm 0.02$   $1.90^{\circ} \pm 0.03$   $2.01^{\circ} \pm 0.01$ 120 -  $4.74a \pm 0.12$   $3.07e \pm 0.05$   $3.52d \pm 0.09$   $3.75e \pm 0.08$   $4.00b \pm 0.10$ 110  $1.49f_{\pm}0.00 \quad 10.45a_{\pm}0.25 \quad 6.24e_{\pm}0.04 \quad 7.22d_{\pm}0.13 \quad 7.65e_{\pm}0.34 \quad 8.08b_{\pm}0.08$   $3.03 \pm 0.05$  - - - - - - - - - 6.16±0.01 - - - - - Oil samples *Ea* (kJ/mol)  $\Delta H^{++}$  (kJ/mol)  $\Delta S^{++}$  (J/mol K)  $\Delta G^{++}$  (kJ/mol) Chia 82.0<sup>d</sup> $\pm$ 0.1 78.9<sup>d</sup> $\pm$ 0.1 -44.4<sup>b</sup> $\pm$ 0.1 62.3<sup>c</sup> $\pm$ 0.1 Sesame  $96.2^{a} \pm 2.0$   $93.0^{a} \pm 2.0$   $-23.7^{a} \pm 5.0$   $83.7^{a} \pm 3.9$ Blends (ω-3:ω-6 ratios) 1:4  $86.7^{\circ}\pm1.7$   $83.4^{\circ}\pm1.7$   $-44.3^{\circ}\pm4.5$   $66.0^{\circ}\pm3.5$ 1:6 88.6<sup>bc</sup> $\pm$ 0.9 85.3<sup>bc</sup> $\pm$ 0.9 -40.7<sup>b</sup> $\pm$ 2.1 69.3<sup>b</sup> $\pm$ 1.7 1:8 89.4b $\pm$ 2.0 86.1b $\pm$ 2.0 -39.1b $\pm$ 4.8 70.7b $\pm$ 3.9 1:10 89.1<sup>bc</sup> $\pm$ 0.9 85.9<sup>bc</sup> $\pm$ 0.9 -40.2<sup>b</sup> $\pm$ 2.3 70.1<sup>b</sup> $\pm$ 1.8 19 391 28 395 



00 Table 5. Shelf life (oxidative stability index at 25  $^{\circ}C$ , OSI<sub>25</sub>) and Q<sub>10</sub> (increase of reaction rate due to

10<sup>o</sup>C temperature rise) of chia oil, sesame oil and their blends.

