

1 **Evaluation of ammonia and odours emission from animal slurry and digestate**
2 **storage in the Po Valley (Italy).**

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15
16 **Abstract**

17 Ammonia and odours emission from one lagoon (Lagoon 1: pig slurry) and three open tanks (Tank
18 2: cow slurry; Tank 3: digestate from pig slurry and energy crops; Tank 4: digestate from pig and cow
19 slurries plus energy crop) used for slurry storage were seasonally sampled for two years in 2015 –
20 2017 in four livestock farms that differed for animal breeding (pig or cow) and manure management
21 (anaerobic digestion).

22 On average, of the two years of observations, the ammonia emission rate (AER) was higher for the
23 Tank 3 (AER of 30.68 ± 28.1 g N-NH₃ m⁻² d⁻¹) than for Lagoon 1 and Tank 2 and 4, i.e. 9.29 ± 14.89
24 gN-NH₃ m⁻² d⁻¹, 9.38 ± 13.75 g N-NH₃ m⁻² d⁻¹, 15.74 ± 21.91 g N-NH₃ m⁻² d⁻¹, respectively. PLS
25 regression analysis ($R^2 = 0.544$; $R^2_{Adj.} = 0.484$) indicated that temperature was the main predictor of

26 ammonia emitted, followed by the concentration in the slurry of total ammonia and the relative
27 percentage of volatile solids (VS).

28 On the other hand PLS analysis ($R^2 = 0.529$, $R^2_{adj.} = 0.417$) indicated that odour emissions from
29 animal slurry storages depended similarly upon total solids and VS (both referred to fresh weight)
30 slurry contents, TAN/TKN ratio and degrees of biological stability (measured by anaerobic biogas
31 potential – ABP), resulting in the Specific Odours Emission Rates (SOER) of $12,124 \pm 7,914$ and
32 $35,207 \pm 41,706$ OUE $m^{-2} h^{-1}$, $65,430 \pm 45,360$ and $43,971 \pm 53,350$ OUE $m^{-2} h^{-1}$, for Lagoon 1 and
33 Tanks 2, 3 and 4.

34

35 Key Word: Ammonia emission; Animal slurries storage; Full scale measurement; Odour emission;

36

37 **1. Introduction**

38 Livestock production in North America, Europe and Australia is rising because of the increase in
39 demand for livestock commodities (McGilloway, 2005). Intensive farming has been proven to be
40 economically effective but many adverse effects, with particular reference to the handling of livestock
41 wastes, have become evident. In particular, the increase in the quantity of manure produced by the
42 animals has triggered concerns for environmental sustainability in relation to air, surface, ground
43 water and land pollution (Edeogu et al., 2001; McCrory and Hobbs, 2001).

44 Livestock manure is the most important source of atmospheric NH_3 in Europe. It is estimated that
45 agricultural activities contribute to 80–95% of ammonia emissions across Europe (Hayes et al., 2004).

46 Atmospheric NH_3 readily reacts with atmospheric acids to form ammonium (NH_4^+) which is an
47 important constituent of aerosols causing respiratory diseases and with atmospheric precipitation,
48 acidification and eutrophication of soil and surface waters (Ndegwa et al., 2008). Ammonia is the
49 main precursor for the formation of secondary inorganic particles ($PM_{2.5}$ and PM_{10}) in the atmosphere,
50 which are considered among the most important atmospheric pollutants as regards their potential
51 impact on human health (Carnevale et al., 2010; Erisman and Schaap, 2004).

52 Environmental regulations are becoming more rigorous concerning ammonia emissions into the
53 atmosphere. In particular, the European Union has recently adopted the new EU directive 2016/2284
54 concerning the reduction of national emissions of certain atmospheric pollutants, including ammonia,
55 i.e. member states must reduce their annual ammonia emissions progressively from 2020 to 2030.
56 Italy in particular, should bring its annual ammonia emissions below the permitted threshold of 354.5
57 GgNH₃ y⁻¹, which is equivalent to a reduction of 10% of the current value (393 GgNH₃ for the year
58 2015). The North of Italy (Po valley) has to make the major effort in ammonia reduction since it
59 contributes greatly to total Italian emissions because this area represents the industrial and agricultural
60 “locomotive” of Italy. In addition the particular geographical conformation of the Po Valley, closed
61 between high mountain ranges on three sides out of four, causes the accumulation of pollutants in the
62 air, especially NO_x, O₃, PM₁₀ and PM_{2.5}, which in the last decades have always exceeded the limits
63 indicated in the Directive 2008/50/EC.

64 To counteract this negative phenomenon, the Lombardy Region (North Italy) is implementing a series
65 of measures (DGR 6675/2017; PRIA 2018) aiming for the improvement of the air quality in the area,
66 including ammonia emission reduction in agriculture. National statistics reveal that the agricultural
67 sector in northern Italy contributes to 94% of the total ammonia emitted annually in Italy (year 2017)
68 (ISPRA, 2019). In particular ammonia losses seem to be due, above all, to animal breeding,
69 manure/slurry storage facilities and manure/slurry application to soils (Sommer, 1997). All these are
70 not only responsible for ammonia emissions but they are responsible, also, for annoyance to the
71 inhabitants due to odour production.

72 Odour is one of the major environmental concerns for the livestock industry. The main sources of
73 odours include building ventilation, manure storage and land spreading. Manure storage represents
74 the major source of complaints and lawsuits. Three major mechanisms control the odour emission
75 from a manure storage facility: i. the chemical-physical characteristics of the state of the manure (e.g.
76 total solid and volatile solids contents, pH, ammonia content, biological stability etc.), ii. the bio-
77 process which the manure has undergone (e.g. aerobic vs. anaerobic digestion) and iii. the state of the

78 air above the manure surface, which controls the process of gases transport (Q. Liu et al., 1995). Often
79 the odours produced are a direct consequence of animal manure decomposed anaerobically to form
80 unstable intermediate by-products, resulting in a complex mixture of over 168 volatile compounds of
81 which 30 are odorous (Hayes et al., 2004). These compounds created from natural biological reactions
82 include organic acids, aldehydes, alcohols, fixed gases, carbonyls, esters, amines, sulphides,
83 mercaptans, aromatics and nitrogen heterocycles (Hayes et al., 2004). Complaints about the nuisance
84 caused by odours from livestock farms have led to the need for alternative manure management and
85 odour control strategies (Edeogu et al., 2001).

86 Manure/storage ammonia and odour emissions represent an important issue above all in those
87 countries in which storage tank coverage is not obligatory. Unfortunately, there are not many
88 available data with reference to these two types of emissions with particular reference to manure
89 storage. Lab scale data are available but there are very few data from full scale plant studies, although
90 these data are useful to understand the real contribution of manure storage to the total emissions and
91 to propose appropriate counter-measures to reduce emissions.

92 The goal of this study is to analyse ammonia emissions and odours from uncovered storage
93 tanks/lagoon at full scale to fill in the lack of knowledge, with particular interest to the Po Valley.
94 This work has been financed by the Lombardy Region in search of data to better address practical
95 measures and correct politics for emission reduction.

96

97 **2. Material and methods**

98 *2.1 Experimental design*

99 One lagoon (Lagoon 1) and three open storage tanks (Tank 2, 3 and 4) used for manure storage were
100 seasonally sampled for two years in 2015 – 2017 (Table S1) at four livestock farms located in
101 Lombardy Region (Italy). The farms were chosen for the differences of breeding species (pig or cow)
102 and manure management (untreated manure, anaerobically digested manure post-treated by

103 solid/liquid separation, and anaerobically digested manure post-treated by solid/liquid separation and
104 ammonia stripping) (Table 1).

105 **Table 1.** Technical data of farms and storage tanks/lagoon.

FARM	BREEDING CONSISTENCY AND TYPOLOGY	CATTLE FEEDING	MANURE TREATMENT	STORAGE TANK DIMENSION	RETENTION TIME IN STORAGE LAGOON/TANK
1 (Lagoon 1)	700 pigs	Corn mash, bran, soy flour, whey, supplements.	Untreated	Lagoon: Volume: 2400 m ³ Surface 800 m ²	4 months
2 (Tank 2)	350 cows	Soy flour, swiss chard, maize flour, molasses, stable hay, medical hay, ryegrass silage, maize silage, calcium carbonate, magnesium oxide.	Untreated	Tank: Volume: 315 m ³ Surface: 63 m ²	4.5 months
3 (Tank 3)	12,000 pigs	Maize, barley, soybean meal, wheat bran, animal fat, calcium carbonate, monocalcium phosphate, sodium chloride, wheat, sodium bicarbonate.	CSTR mesophilic anaerobic digestors (999 kWe and 42 days HRT) with feed mixture composed by: pig slurry, pollen, corn silage, silage barley, hard wheat flour. Subsequent removal of the solid with helical separator	Tank: Volume: 1980 m ³ Surface: 850 m ²	3 months
4 (Tank 4)	7,400 pigs + 200 cows	Maize, barley, soy, bran, calcium carbonate, monocalcium phosphate, sodium chloride.	Mechanical separation of thick and fine solid fractions. Biological nitrification and denitrification treatment. Anaerobic digestion with biogas recovery.	Tank: Volume: 13750 m ³ Surface: 2750m ²	6 months

106

107 *2.2 Manure sampling and characterization*

108 During the measurement campaigns of the ammonia and odour emissions from the storage tanks
109 and lagoon, representative samples from each storage tank were taken by using a 500 ml jar with a
110 telescopic bar. At the time of sampling, the air temperature was measured immediately above the
111 surface of the tank or lagoon (about 10 cm above the surface). Samples were chemically
112 characterized and their biological stability was determined. In particular, total solids (TS) and
113 volatile solids (VS) were determined following standard procedures (APHA, 1998). Total N-
114 Kjeldahl (TKN) and total ammonia nitrogen (TAN) were analysed on fresh samples according to
115 the analytical method established for wastewater sludge (APHA, 1998). pHs were determined
116 according to standard procedures (US Department of Agriculture – US Composting Council, 2002).
117 Total P and K contents were determined by inductively coupled plasma mass spectrometry (Varian,
118 Fort Collins, USA). Standard samples (National Institute of Standards and Technology,
119 Gaithersburg, MD, USA) and blanks were run with all samples to ensure precision in the analyses.
120 P and K detection was preceded by acid digestion (EPA, 1998) of the fertilized samples. Biological
121 stability related to long term degradability was performed by measuring the anaerobic
122 biogasification potential test (ABP) according to Orzi et al. (2015) using 0.62 g of dried matter
123 sample, 37.5 ml of inoculum, and 22 ml of deionized water in 100-ml serum bottles. A control blank
124 was prepared with 60 ml of inoculum. Inoculum was incubated at 37 ± 1 °C for 15 days before
125 being used in ABP assays. The bottles were stored at 37 ± 1 °C for 60 days. The biogas production
126 was determined periodically and expressed as NI kg TS^{-1} . All analyses were performed in triplicate
127 and data referred to fresh weight (fw) or TS (Orzi et al., 2015).

128

129 *2.3 Measurement of odour and ammonia emission from manure tanks*

130 From each manure storage facility, the odour and ammonia emissions were collected by positioning
131 over the emitting surface a wind tunnel system (area of 0.1225 m^2). A known neutral air flow (air
132 flow rate of $0.38 \text{ m}^3 \text{ h}^{-1}$) was introduced into the device, simulating the action of wind on the liquid

133 or solid surface (Brattoli et al., 2011). For odour collection and storage, Nalophan bags were
134 connected to the output of the wind tunnel. Air in bags was analysed through dynamic olfactometry
135 within 30 hours of sampling. Inside the wind tunnel, during the fluxing period, ammonia
136 concentration was measured through the exposure of acid coated passive samplers (Tang et al., 2001).
137 All odour measurements and ammonia determinations were performed in triplicate for each storage
138 tank and lagoon.

139

140 *2.4 Ammonia determination*

141 The exposed filters of acid coated passive samplers were leached with deionized water (3 mL) and
142 then analysed by spectrometric detection (FIAstar 5000 system, FOSS, Denmark) through a gas semi
143 permeable membrane (ISO 11732, 1997), in order to measure the concentration of NH₄-N. The
144 ammonia emission rate (AER) (NH₃ μg m² h⁻¹) was calculated following the equation:

$$145 \text{ AER} = \text{NH}_3 \text{Q/S}$$

146 Considering the incoming air rate to the air flux (Q) and the surface covered by the wind tunnel (S),
147 see paragraph 2.3.

148

149 *2.5 Dynamic olfactometry*

150 Olfactometric analyses were carried out in conformity with the standardized EN method n. 13725
151 (CEN, 2003). An Olfaktomat-n 6 olfactometer (PRA-Odournet B.V., Amsterdam, NL), based on the
152 forced choice method, was used as a dilution device. The results of the Dynamic Olfactometry were
153 expressed as odour concentration value OU (OU_E m⁻³). The specific odour emission rate SOER (OU_E
154 m⁻² h⁻¹) was calculated following the equation:

$$155 \text{ SOER} = \text{OUQ/S}$$

156 considering the incoming air rate to the air flux (Q) and the surface covered by the wind tunnel (S),
157 see paragraph 2.3. In particular, air velocity was multiplied by the exponent 0.5 for the liquid
158 condition, according to Bliss et al., (1995)(Bliss et al., 1995).

159

160 *2.6 Statistical analysis*

161 Statistical analyses were conducted using IBM SPSS 25, (IBM Corp. Released 2017. IBM SPSS
162 Statistics for Windows, Version 25.0. Armonk, NY: IBM Corp). To overcome the normality and
163 homogeneity of variance problems the ANOVA analysis and the multiple comparison of means was
164 done with a bootstrap-based procedure (Acutis et al., 2012; Xu et al., 2013). Multiple comparison
165 was done using the Tukey test.

166 The main predictors of the ammonia emission, of the Lagoon 1 and Tank 2,3, and 4 and the effects
167 corresponding different origin of the slurries contained were determined thanks to partial least square
168 regression (PLS) (Ferreira et al., 2016). PLS is a powerful technique that generalizes and combines
169 features from principal component analysis and multiple regression when the number of predictors
170 is close or bigger than the observations (Abdi, 2003).

171

172

173 **3. Results and discussion**

174 *3.1 Characterization of the manures*

175 The analytical characterization of the manures (Table 2) was important as these parameters can affect
176 both ammonia and odours emission from tanks/lagoon. On average, the TS was higher for cow
177 manure (Tank 2) (TS of 4.04 ± 2 % fw) because of the presence of straw and residual lignocellulose
178 fractions that were also responsible for the formation of surface crust in the storage tank during the
179 whole period of this study. The TS content was however lower for Tanks (digestates) 3 and 4 (TS of
180 1.88 ± 0.5 % fw and of 2.55 ± 0.6 % fw, respectively), because of both the biological process
181 degrading TS, and the subsequent solid/liquid separation which digestates underwent as post
182 treatment. Lowest values (TS of 0.99 ± 0.2 % fw) were found for the pig manure stored in the lagoon
183 (Lagoon 1), in agreement with the literature (Orzi et al., 2018; Sørensen and Amato, 2002).

184 **Table 2.** Chemical characterization and biological stability of the manures stored in the tank and lagoon.

Tank/ Lagoon	Sample	pH	TS (% fw)	VS (%)	TKN (g kg ⁻¹ TS)	TAN (g NH ₄ ⁺ kg ⁻¹ TS)	TAN/TKN (%)	P (g kg ⁻¹ TS)	K (g kg ⁻¹ TS)	ABP (Nm ³ mg ⁻¹ TS)
1	sp I ^a	7.48±0.03c ^b	0.64±0.01b	46.6±0.6b	250±2c	219±1f	88	28.38±1.3d	208±7d	150±7c
1	su I	7.32±0.02b	1.67±0.03e	60.0±0.3e	92.2±2.0a	62.1±1.3a	67	83.4±0.2e	47.2±0.9a	111±1b
1	au I	7.28±0.01b	1.4±0.0d	37.5±1.4a	86.7±0.4a	75.1±1.6b	87	17.33±2.4b	197±2d	21±3a
1	wi I	7.09±0.03a	1.61±0.02e	59.4±1.3e	84.6±6.1a	78.1±3.4b	92	21.65±1.4c	55±0a	230±10e
1	sp II	7.36±0.05bc	0.74±0.02c	52.1±2.7c	214±9.2b	184±5e	86	8.77±2a	193±13d	331±4g
1	su II	7.42±0.01c	0.68±0.02bc	55.7±0.6cd	223±13b	156±5d	70	19.58±2.9bc	132±6b	265±12f
1	au II	7.41±0.04c	0.48±0.03a	48.6±1.1b	260±4c	139±16c	53	20.74±0.6c	166±3c	205±5d
1	wi II	7.05±0.08a	0.68±0.03bc	56.1±0.3d	208±10b	176±17de	85	16.35±0.4b	132±8b	280±3f
1	Mean	7.3±0.2	1±0	52±7.6	177±76	136±58	78.5±13.6	27.1±23.4	141±62	199±101
2	sp I	7.03±0.02a	2.75±0.04b	72.1±0.6a	65.9±0.1c	34.2±0.7cd	52	13.3±1c	58.8±1.2d	119±2b
2	su I	7.12±0.02b	3.38±0.08c	80.1±2.1c	75.3±3.1d	35.1±0.1d	47	10±1b	33.3±0.6ab	154±14c
2	au I	7.23±0.01c	3.34±0.02c	74.2±1.6b	61.2±0.4c	32.7±0.4c	53	13.1±1c	55.6±0.9c	105±1a
2	wi I	7.31±0.02d	3.37±0.06c	72.9±2.2ab	66.3±0.4c	32.9±1.0c	50	11.6±1b	63.4±0.6e	108±1a
2	sp II	7.11±0.01b	1.84±0.01a	70.3±1.3a	66.7±0.2c	36.7±1.4d	55	13.5±0.2c	77.3±1.9g	261±3c
2	su II	7.24±0.01c	7.50±0.40e	81.6±1.8c	36.2±1.3a	14.6±1.2a	40	7.9±0.8a	31.5±1.7ab	358±12e
2	au II	7.38±0.02d	3.28±0.08c	70.2±0.9a	72.1±3.2d	26.5±2.7b	37	12.9±1bc	70.5±0.4f	151±10c
2	wi II	7.91±0.02e	6.87±0.11d	82.7±2.0c	42.2±1.4b	35.2±1.3d	83	6.9±0.6a	35.6±0.9b	318±4d
2	Mean	7.3±0.3	4±2	75.5±5.1	61±14	30.9±7.3	52±14	11.2±2.6	53.2±16.6	197±101
3	sp I	7.83±0.08c	2.46±0.22d	54.4±1.1bc	123±1a	100±2a	81	18.2±0.7b	149±3a	104±10c
3	su I	7.03±0.02a	1.24±0.04a	47.2±1.3a	253±4e	195±8e	77	22.6±1c	154±4a	45±1.5a
3	au I	7.65±0.03b	1.59±0.06b	46.9±2.5a	184±2c	151±6d	82	20.7±0.6c	176±4a	51±1.3b
3	wi I	7.88±0.02c	2.41±0.03d	56.8±2.2c	168±3b	135±4b	80	13.17±0a	233±6c	97±10c
3	sp II	8.05±0.03d	1.60±0.06b	52.7±0.4b	190±2d	160±1d	84	16.9±0.7b	208±4b	163±12d
3	su II	8.17±0.01e	1.96±0.11c	56.5±1.9c	198±7d	139±6c	70	15.6±0.1b	215±0bc	205±12e
3	Mean	7.8±0.4	1.9±0.5	52.4±4.4	186±42	146±31	79±5	17.9±3.4	189±35	111±623
4	wi I	8.82±0.03c	2.05±0.08a	43.1±2.5cd	71.7±4.2c	42.7±3.1cd	60	10.5±0.4c	140±1a	68±7b
4	sp II	8.32±0.01a	2.14±0.02a	44.9±0.5d	80.2±2.8d	37.8±1.2e	92	9.8±0.1bc	181±2b	245±17e
4	su II	8.34±0.03a	2.50±0.08b	42.4±0.3c	67.7±3.4b	49.2±4.3d	73	7.7±0.1a	222±2b	172±16c
4	au II	8.63±0.02b	3.63±0.04c	32.7±1.4a	46.3±1.9a	13.1±1.4a	28	9.7±0.1b	136±1a	43±4a
4	wi II	9.04±0.05d	2.42±0.04b	39.9±0.6b	97±3e	35.2±4.5b	36	10±1c	139±1a	210±5d
4	Mean	8.6±0.3	2.5±0.6	40.6±4.8	72.6±18.6	35.6±13.7	57.8±26.3	9.5±1.1	164±39	148±88

185 ^awi = winter; sp = spring; su = summer; au = autumn; I = 1st year; II = 2nd year

186 ^bValues followed by the same letter, in the same tanks, are not statistically different (ANOVA bootstrap and Tukey test, p<0.05)

187 The organic content, expressed as volatile solids (VS) (on average) was high for cow manure (VS of
188 75.51 ± 5.1 % TS) (Tank 2) because of the presence of straw and carbon residues coming from the
189 polygastric (ruminants) diet. The other tanks/lagoon showed lower VS contents which were similar
190 between them. Tanks 3 and 4 (digestates), indeed, showed reduced volatile solids contents (VS of
191 52.42 ± 4.4 % TS and of 40.6 ± 4.8 % TS, respectively) because of both the biological process
192 (anaerobic digestion) and the subsequent solid/liquid separation. Finally, Lagoon 1 (pig manure)
193 showed the lowest VS value (52 ± 7.6 % TS) in line with data reported for manure stored in lagoons
194 (Safley and Westerman, 1992).

195 pH represents an interesting data source as it directly affects ammonia volatilization, with reported
196 high ammonia losses for pHs in the range 7-10 (Ndegwa et al., 2008). pH values (as average) for the
197 four tanks/lagoon studied in this work (Table 3) fell within this range. Above all Tanks (digestates) 3
198 and 4 showed alkaline pHs, i.e. pH of 7.77 ± 0.4 and 8.63 ± 0.3 , respectively.

199 The concentrations of N (NTK and $\text{NH}_4\text{-N}$) contained in the biomasses analysed (average TKN of
200 177 ± 76 g kg^{-1} TS, 60.7 ± 14 g kg^{-1} TS, 186 ± 42 g kg^{-1} TS and 72.6 ± 18.6 g kg^{-1} TS, for the Lagoon 1,
201 and Tanks 2, 3 and 4, respectively) were always in line with those reported in the past (Akhiar et al.,
202 2017; Orzi et al., 2018; Riva et al., 2016; Tambone et al., 2019) for similar matrices. Looking at the
203 data in detail (Table 2) it will be noted that the NTK and TAN concentrations of each lagoon/tanks
204 showed high variability, with a tendency to their decrease over time because of ammonia
205 volatilization. In addition, measurements made showed that the highest NTK concentrations
206 corresponded to the periods immediately following the addition of new material to the tank/lagoon.

207 **Table 3.** Gran mean of emission data and salient values for the ammonia emission (AER) from the manures.

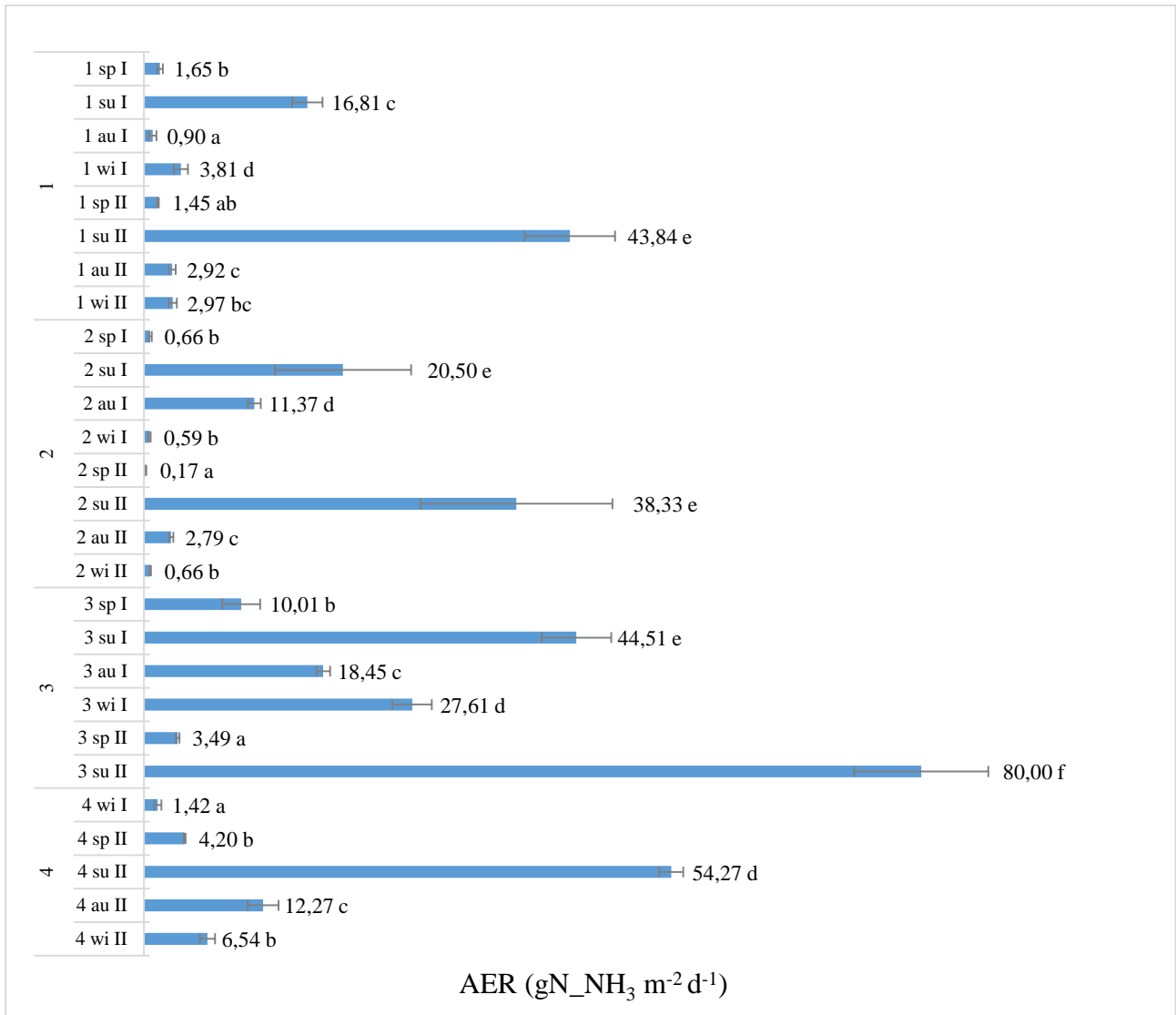
Tank/ Lagoon	AER (gN-NH₃ m⁻² d⁻¹)	T (°C)	pH	TS (% fw)	VS (%)	TKN (g kg⁻¹ fw)	TAN (g NH₄⁺ kg⁻¹ fw)	TAN/TKN (% fw)
1	9.29±14.89 ^a	18±10	7.30±0.15	0.98±0.9	0.52±0.29	91±68	1.13±0.24	78.51±13.61
2	9.38±13.75 ^a	18±10	7.29±0.27	4.04±2.01	3.12±1.77	61±14	1.05±0.22	47.73±6.33
3	30.68±28.1 ^b	20±10	7.97±0.52	2.67±1.81	1.74±1.67	186±43	2.64±0.73	79.14±4.94
4	15.74±21.91 ^a	15±9	8.63±0.81	2.34±0.60	1.04±0.17	73±18	1.00±0.42	57.82±26.25

208 ^aValues followed by the same letter (AER) are not statistically different (ANOVA bootstrap and Tukey test, p<0.05).

209 *3.2 Ammonia emission rate measured from storage tanks*

210 The ammonia emission (AER) was measured once per season over the course of two years (in total 8
211 measurements per storage lagoon/tank) for all the manure storage facilities monitored (Table 3)
212 (Figure 1). The data obtained (TAN emissions) were in line with those reported in the past (De Bode,
213 2005; Misselbrook et al., 2016; Sommer, 1997).

214 Looking at the data in detail (Figure 1) it can be seen that for all tanks and the lagoon monitored, the
215 emission of ammonia ($\text{g N-NH}_3 \text{ m}^{-2} \text{ d}^{-1}$) varied with the season, as confirmed by ANOVA analysis
216 (ANOVA bootstrap and Tukey test; $p < 0.05$). The season that registered the greatest ammonia
217 emission was the hot season, i.e. summer. Indeed, the PLS regression analysis ($R^2 = 0.544$; $R^2_{\text{Adj.}} =$
218 0.484) applied to the data (Table S2) confirmed that temperature (T) was the main predictor of TAN
219 emitted from tanks (importance of the predictor of 0.45), followed by the TAN concentration in the
220 sample (TAN referred to fw) (importance of predictor of 0.25) and by the percentage of VS in the
221 sample (importance of predictor of 0.12).



222

223 **Figure 1.** Ammonia Emission Rate (AER) from slurry/digestate storage tanks/lagoon (^awi = winter;
 224 sp = spring; su = summer; au = autumn; I = 1st year; II = 2nd year). Values followed by the same
 225 letter, in the same tanks, are not statistically different (ANOVA bootstrap and Tukey test, p<0.05).

226 The driving force for NH_3 volatilization has been reported to depend upon the difference in the partial
227 pressure of NH_3 between the liquid phase and the atmosphere that depends, mainly, on the NH_4^+
228 concentration in the slurry, pH and temperature (McCrorry and Hobbs, 2001), confirming PLS results
229 (T and TAN). The pH, that is considered of primary importance in determining the quantity of
230 ammonia emitted by a biomass (Webb et al., 2005), was only the fifth predictor, in order of
231 importance (importance of predictor 0.05), affecting TAN emission. This was probably due to the
232 fact that pH remained relatively stable during the year for all tanks (Table 2).

233 By comparing the TAN emissions during the two years (Grand mean) (Table 3), it can be seen that
234 the Tank 3 emitted a markedly higher amount of NH_3 than the other three storage tanks/lagoon (i.e.
235 AER of $30.68 \pm 28.1 \text{ g N-NH}_3 \text{ m}^{-2} \text{ d}^{-1}$ emitted from Tank 3, to be compared with AER of $15.74 \pm$
236 $21.91 \text{ g N-NH}_3 \text{ m}^{-2} \text{ d}^{-1}$, of $9.38 \pm 13.75 \text{ g N-NH}_3 \text{ m}^{-2} \text{ d}^{-1}$ and of $9.29 \pm 14.89 \text{ gN-NH}_3 \text{ m}^{-2} \text{ d}^{-1}$ emitted
237 from Tank 4 and 2, and Lagoon 1, respectively). The ANOVA analysis (ANOVA bootstrap, $p < 0.05$,
238 Tukey test) confirmed the difference between the amount of ammonia emitted, on average, from the
239 Tank 3 and the other storage facilities, while no significant differences were found between the
240 amounts of ammonia released by the other three tanks/lagoon (Lagoon 1 and, Tanks 2 and 4), despite
241 the different origin of the slurries which they contained.

242 From Table 3 it can be seen that Tank 3 showed both higher average temperature ($20 \pm 10 \text{ }^\circ\text{C}$) and,
243 NTK and TAN (NTK of $2.64 \pm 0.73 \text{ g kg}^{-1} \text{ fw}$ and TAN of $186 \pm 43 \text{ g kg}^{-1} \text{ fw}$, respectively) than
244 those measured for the other three tanks/lagoon, confirming the importance of these parameters in
245 regulating ammonia emissions as indicated, also, by the PLS analysis.

246

247 *3.3. A tentative in quantifying total ammonia emission from storage tank/lagoon.*

248 Taking into consideration ammonia emitted from storage tanks/lagoon ($\text{g N-NH}_3 \text{ m}^{-2} \text{ d}^{-1}$) and total
249 surface area, we calculated the total annual amount of ammonia emitted from the three tanks and
250 lagoon monitored (Table 4). These data represent a rough, but in our opinion useful, estimate of
251 average annual emission. In fact although the data suffer from high standard deviations, these were

252 due to the variability of ammonia emission due to the variability of both atmospheric conditions and
253 slurry composition that occurred during the year, giving approximate but real data.

254 Results reported in Table 4 reflected specific emissions previously discussed. On the other hand, to
255 rank the contribution of manure/slurry storage to total ammonia emissions, it was interesting to
256 compare these data with those coming from manure/slurry spreading onto the soil, since in this case
257 ammonia emissions are much more studied and well known (Chantigny et al., 2009; Nicholson et al.,
258 2017; Riva et al., 2016). We considered what would be the effect of spreading onto soil of
259 slurries/digestates stored in tanks or in the lagoon studied during the year, assuming the specific
260 ammonia emission rate typical for each slurry (e.g. cow, and pig slurry, and liquid fractions of
261 digestate), as reported in the literature (Table 4).

262 **Table 4.** Estimated comparison of ammonia emissions per year in the two different cases hypothesised: slurry/digestate storage Vs. slurry/digestate
 263 spreading onto soil (N dosed = 150 – 180 kgN ha⁻¹).

Tank/ Lagoon	AER (gN-NH₃ m⁻² d⁻¹)	NH₃ emission per year from the tank/lagoon (tN-NH₃ y⁻¹)	NH₃ emission per year in case of manure spreading (tN-NH₃ y⁻¹)	^aReference
1	9.29±14.89a ^b	3.14	2.86	Chantigny <i>et al.</i> 2009
2	9.38±13.75a	0.22	0.32	Nicholson <i>et al.</i> 2017
3	30.68±28.1b	13.80	9.83	Riva <i>et al.</i> 2016
4	15.74±21.91a	21.50	12.94	Riva <i>et al.</i> 2016

264 ^aLiterature reference reporting the ammonia emission rate considered to calculate total ammonia emission during manure spreading.

265 ^bValues followed by the same letter, in the same tanks, are not statistically different (ANOVA bootstrap and Tukey test, p<0.05).

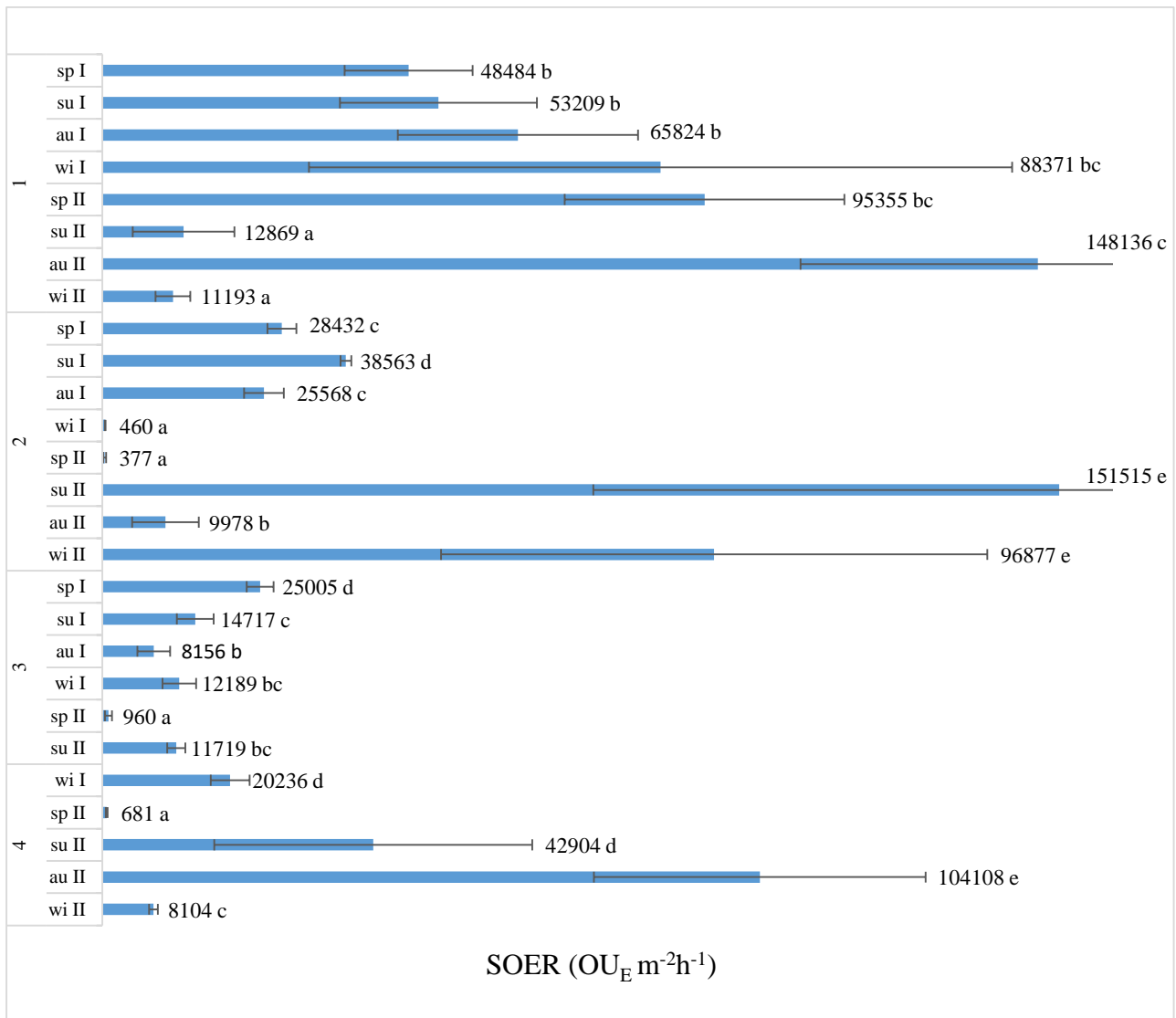
266 Results indicated (Table 4) that there were no differences for the untreated biomasses (not digested)
267 between the ammonia released into the atmosphere during untreated slurries storage and soil
268 spreading, i.e. 3.14 vs 2.86 tN-NH₃ y⁻¹ for Lagoon 1 and 0.22 vs 0.32 tN-NH₃ y⁻¹ for Tank 2, in
269 agreement with previous Italian data (ISPRA, 2019). On the other hand, anaerobic digestion and the
270 subsequent solid/liquid separation (Tanks 3 and 4) seemed to double the ammonia emissions from
271 tanks compared to those emitted in the case of spreading of digestate in open fields (13.8 vs 9.83 tN-
272 NH₃ y⁻¹ for Tank 2 and 21.5 vs 12.94 tN-NH₃ y⁻¹ for Tank 3), which makes these biomasses
273 particularly polluting if kept for a long time in uncovered tanks.

274

275 *3.4 Specific odour emission rates measured from storage tanks*

276 Odours from livestock slurries are due to a complex mixture of volatile compounds arising from
277 anaerobic degradation of plant fibers and protein (McCrorry and Hobbs, 2001). Some of these
278 compounds are sulphur compounds, organic acids, phenolic compounds and indoles. Among these,
279 some may even be hazardous to human health, such as (di)hydrogen sulphide (H₂S), which is lethal
280 at high concentrations (Sommer and Feilberg, 2013).

281 The odorous emissions (SOER) were monitored on a seasonal basis, as well as reported for
282 ammonia sampling, for all four tanks under analysis (Figure 2).



283
 284 **Figure 2.** Odour emission (SOER) from slurry/digestate storage tanks/lagoon (^awi = winter; sp =
 285 spring; su = summer; au = autumn; I = 1st year; II = 2nd year). Values followed by the same letter, in
 286 the same tanks, are not statistically different (ANOVA bootstrap and Tukey test, p<0.05).

287 As shown in the graph in Figure 2, the values of odorous emissions were very variable over time
288 and the data showed, as expected, a high standard deviation. This was probably due to the high
289 number of factors that influenced this type of emissions, i.e. environmental factors (temperature and
290 air movements), storage (shape of the tank/lagoon, agitation of the matrix, formation of crusts),
291 treatment (solid/liquid separation, anaerobic digestion) and biomass composition (chemical and
292 biological composition, feeding of the animals) (Edeogu et al., 2001; Hayes et al., 2004; Q. Liu et
293 al., 1995; Sommer and Feilberg, 2013).

294 The high variability of the levels of odour emissions made it impossible to determine a period of the
295 year during which biomasses are particularly subject to odours emission. However, observing the
296 grand means for each of the storage tanks/lagoon (Table 5), it was possible to note that the Lagoon
297 1 and Tank 2 showed average odour emissions (SOER of $65,430 \pm 45,360$ and of $43,971 \pm 53,350$
298 OUE $\text{m}^{-2} \text{h}^{-1}$, respectively) higher than those measured for Tanks 3 and 4 (SOER of $12,124 \pm 7,914$
299 and of $35,207 \pm 41,706$ OUE $\text{m}^{-2} \text{h}^{-1}$, respectively).

300 **Table 5.** Gran mean of emission data and salient values for the odour emission (SOER) from the manures.

Tank /Lagoon	SOER (OUE m⁻²h⁻¹)	TS (% fw^a)	VS (%)	TAN/TKN (% raw)	ABP (Nm³ mg⁻¹ TS)	TAN (gNH₄⁺ kg⁻¹ fw)	T (°C)	pH	TKN (g kg⁻¹ fw)
1	65,430±45,360	0.98±0.9	0.52±0.29	78.51±13.61	199±101	1.13±0.24	18±10	7.30±0.15	91±68
2	43,971±53,350	4.04±2.01	3.12±1.77	47.73±6.33	197±101	1.05±0.22	18±10	7.29±0.27	61±14
3	12,124±7,914	2.67±1.81	1.74±1.67	79.14±4.94	122±74	2.64±0.73	20±10	7.97±0.52	186±43
4	35,207±41,706	2.34±0.61	1.04±0.17	57.82±26.25	148±88	1.00±0.42	15±9	8.63±0.81	73±18

301 ^afw = fresh weight

302 The variables most implicated in determining the levels of odorous emissions from the biomasses
303 were isolated by PLS analysis ($R^2 = 0.529$, $R^2_{adj.} = 0.417$) which showed that TS, VS (referred to
304 fresh weight), TAN/TKN and ABP all affected similarly (importance of the predictor in the range
305 0.18-0.21) (Table S3) the odorous emissions (Table 5). TS and VS affected odour emissions as they
306 represent quantitative data of the total amount of substrate available for microorganisms that
307 degrade organic matter under anaerobic conditions to produce odours (Orzi et al., 2010). The
308 TAN/TKN ratio affected total odour because a high ratio indicates the high presence of N under
309 ammonia forms instead of the organic forms, since ammonia is a gas characterized by a pungent
310 odour that greatly contributes to odours emitted by a matrix (Sommer and Feilberg, 2013). The
311 anaerobic biogasification potential (ABP), contrarily to the TS and VS, represents qualitative data
312 of the organic matter (or VS) contained in a slurry. This variable is a direct measurement of the
313 biological stability of the organic matter and so of its degradability, from which depends the
314 potential of a biomass to produce volatile organic compounds under anaerobic conditions,
315 contributing to biomass odours (Orzi et al., 2010). In particular, the Lagoon 1 and Tank 2 which
316 contained the most odorous matrices (Table 5), showed higher ABP values (ABP of 199 ± 101 and
317 of $197 \pm 101 \text{ Nm}_3 \text{ Mg}^{-1}\text{TS}$, respectively) than those reported for tanks 3 and 4 (ABP of 122 ± 74
318 and of $148 \pm 88 \text{ Nm}_3 \text{ Mg}^{-1} \text{ TS}$, respectively) that contained digested slurries. This observation
319 agreed with what was previously observed by Orzi and colleagues who identified the ABP index as
320 one of the main indicators of the potential of odours emission from an organic matrix (Orzi et al.,
321 2010). Moreover, these results confirmed the ability of anaerobic digestion to reduce the potential
322 odour emissions of a biomass (Orzi et al., 2015).

323 TAN, pH and T, similarly, also affected odour emissions (Table 5) although much less than the
324 previous parameters discussed, as, also, indicated by the “importance of the predictor” detected
325 performing PLS, which were much lower than the others (Table 5). These parameters probably
326 affected total odours emission because they directly (TAN referred to the fresh weight) and indirectly
327 (T and pH) influenced the presence of free ammonia.

328 In any case despite the strong variability of the detected odorous emissions (Figure 2), the intensity
329 of the odours (SOER) was generally very high and in the range of data previously reported for similar
330 organic matrices (Orzi et al., 2018, 2015), underlying the potential impact of these matrices when
331 stored and subsequently used for direct soil applications.

332

333 **4. Conclusion**

334 In this paper data on ammonia and odours emission have been provided at full scale. Results obtained
335 indicated that the slurry type affects ammonia emission (pig vs. cow slurry) as well as
336 biological/mechanical treatment (anaerobic digestion plus S/L separation vs. untreated slurry). In
337 particular, the liquid fraction of digestate, because of its high ammonia content, showed much higher
338 ammonia emissions than untreated slurries. This fact led, also, to the contribution of storage to
339 ammonia emissions being much higher than those due, potentially, to the subsequent spreading of
340 slurry, suggesting that the covering of tanks/lagoons containing digestate should be a priority in
341 reducing ammonia emissions, as well as digestate injection (Ndegwa et al., 2008; Orzi et al., 2018;
342 Riva et al., 2016; Webb et al., 2005).

343 On the other hand, digested slurries, because of the biological process to which they have been
344 subjected led to the acquirement of a high degree of biological stability, reducing greatly the odours
345 emitted.

346

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352 **Notes**

353 The authors declare no competing financial interest

354

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