

Anisotropic compressional behavior of ettringite

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A Portland cement is a complex multi-component system and, to predict its elastic properties, an exhaustive database of the thermodynamic parameters of the main constituents is needed. Ettringite (ideally $\text{Ca}_6\text{Al}_2(\text{SO}_4)_3(\text{OH})_{12}\cdot 27\text{H}_2\text{O}$, with $a=b \sim 11.21$ and $c \sim 21.43$ Å, Sp. Gr. $P31c$) is one of the most important crystalline phases in Portland cements: its crystallization, in the early hydration stages, governs the set rate of the highly reactive “C3A” phase ($\text{Ca}_3\text{Al}_2\text{O}_6$), whereas in aged cements the formation of ettringite is commonly associated with degradation processes (Taylor et al. 2001). The crystal structure of ettringite is significantly complex with a H-bonding net which connects $[\text{Ca}_3[\text{Al}(\text{OH})_6]\cdot 12\text{H}_2\text{O}]$ -columns (in which $\text{Al}(\text{OH})_6$ -octahedra are alternated with triplets of $\text{Ca}(\text{OH})_4(\text{OH}_2)_4$ -polyhedra) to sulphate groups (Gatta et al. 2019). Despite the previous studies at high pressure on this material (e.g., Cuesta et al. 2017, Clark et al. 2008), the linear bulk moduli (K_a and K_c) and a description of the deformation mechanisms at the atomic scale are still missing. In this light, we have investigated the compressional behavior of ettringite up to 4.2 GPa by means of *in-situ* single-crystal synchrotron X-ray diffraction, using a diamond-anvil cell (DAC) and the mix methanol:ethanol (4:1) as *P*-transmitting fluid.

Ettringite shows a marked anisotropic compressional pattern (K_a 21(1) GPa, K_c 47(1) GPa). This anisotropic elastic scheme dramatically changes at $P > 3$ GPa; K_{V0} drops from 26.6(5) to 10.4(8) GPa, which mainly affects the structure on the *ab* plane (K_a drops from 21(1) to 7.3(8) GPa whereas K_c decreases only moderately). Structure refinements reveal that the elastic softening reflects the collapse of the H-bonding network, due an average decrease of the $\text{O}_{\text{donor}}\cdots\text{O}_{\text{acceptor}}$ distances (up to 0.20 Å in some cases), which mainly affect the interaction between the sulphate groups and the $\text{Ca}(\text{OH})_4(\text{OH}_2)_4$ -polyhedra.

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