

Synthesis and Photophysical Properties of Isocoumarin-based D- π -A systems

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Supporting Information

Photophysical Characterization:

Onsager cavity radius

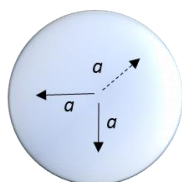
$$a = \sqrt[3]{\frac{3M}{4\pi\delta N_A}}$$

M = molecular weight

N_A = Avogadro's number

δ = compound density

Lippert-Mataga equation



Sphere

$$V = \frac{4}{3}\pi a^3$$

$$\bar{\nu}_a - \bar{\nu}_f = \frac{2}{hc} \left(\frac{\epsilon - 1}{2\epsilon + 1} - \frac{n^2 - 1}{2n^2 + 1} \right) \frac{(\mu^* - \mu)^2}{a^3} + const$$

$\bar{\nu}_a$ and $\bar{\nu}_f$ = wavenumbers in cm^{-1}

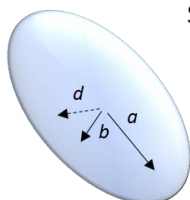
h = Planck's constant

c = speed of light

μ^* and μ = dipole moment of the molecule in the excited and ground state

a = Onsager cavity radius

Modified Lippert-Mataga equation



Spheroid prolate

$$V = \frac{4}{3}\pi abd$$

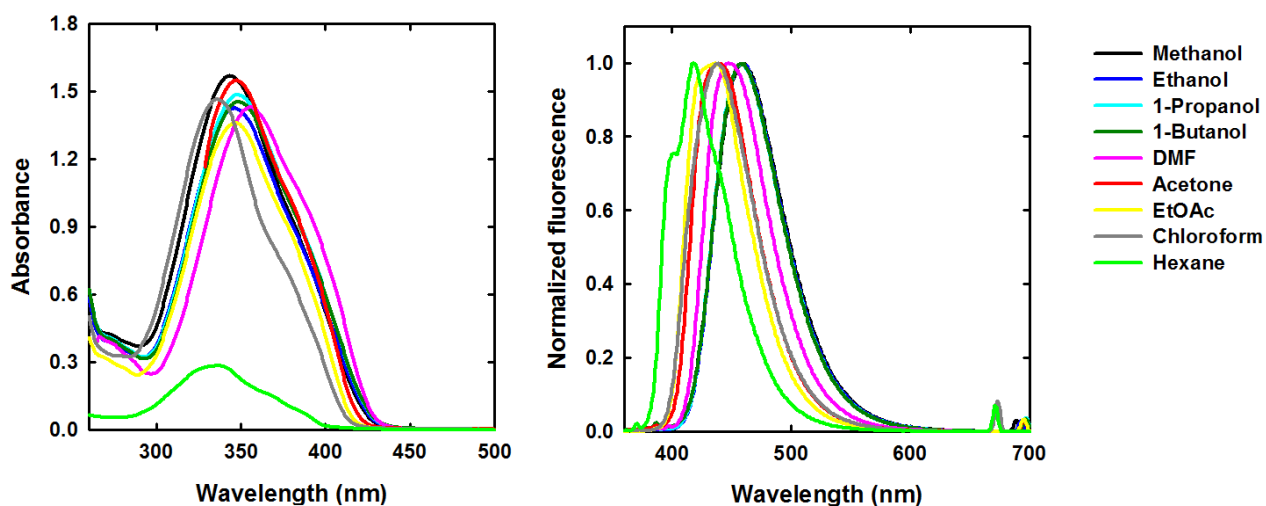
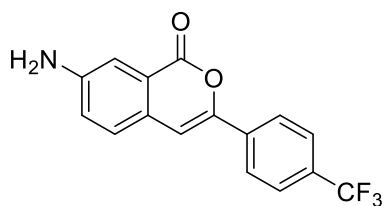
$$a > b = d$$

$$\bar{\nu}_a - \bar{\nu}_f = \frac{3}{hc} \left(\frac{\epsilon - 1}{2\epsilon + 1} - \frac{n^2 - 1}{2n^2 + 1} \right) \frac{(\mu^* - \mu)^2}{abd} + const$$

References:

1. Vequi-Suplicy C. C., Coutinho, K.; Lamyet, M. T. *Biophys. Rev.* **2014**, 6, 63–74.
2. Scholte T.G., *Physica* **1949**, 15, 437–449.

1a

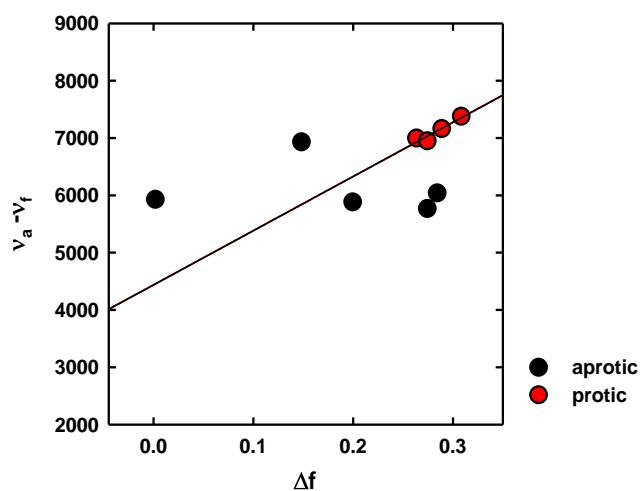


Calculated $\epsilon_{346\text{nm}}$ in ethanol = $21764 \text{ M}^{-1} \text{ cm}^{-1}$

Measurements carried out at 20°C , $l=1 \text{ cm}$

$[1a] = 65.52 \text{ }\mu\text{M}$

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	344	461
Ethanol	346	460
1-Propanol	348	459
1-Butanol	348	460
1-Octanol	Not soluble	Not soluble
DMF	356	448
Acetone	347	439
Ethyl Acetate	347	436
Chloroform	336	438
Hexane	335	418



Molecular volume = 357.02 \AA^3
 $\mu = 3.68$ Debye (Discovery Studio)

Sphere - protic

Onsager cavity radius = 4.4 \AA

$\mu^* - \mu = 8.94 \pm 0.04$ Debye

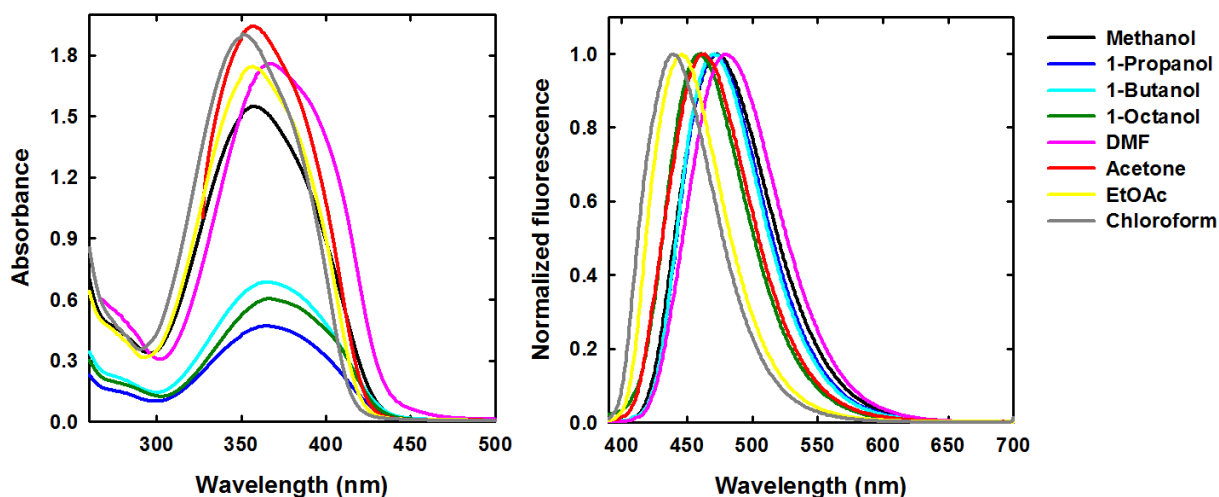
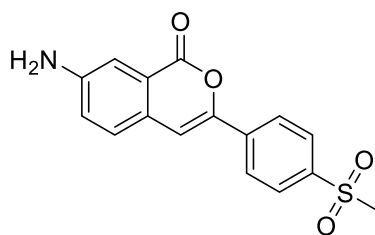
Spheroid prolate - protic

$a = 5.7 \text{ \AA}$

$b = d = 1.84 \text{ \AA}$

$\mu^* - \mu = 3.48 \pm 0.42$ Debye

1b

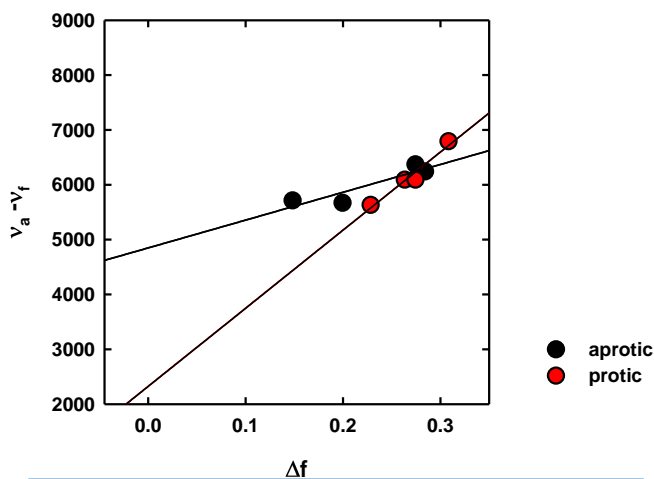


Calculated $\epsilon_{358\text{nm}}$ in methanol = $24406 \text{ M}^{-1} \text{ cm}^{-1}$

Measurements carried out at 20°C , $l=1 \text{ cm}$

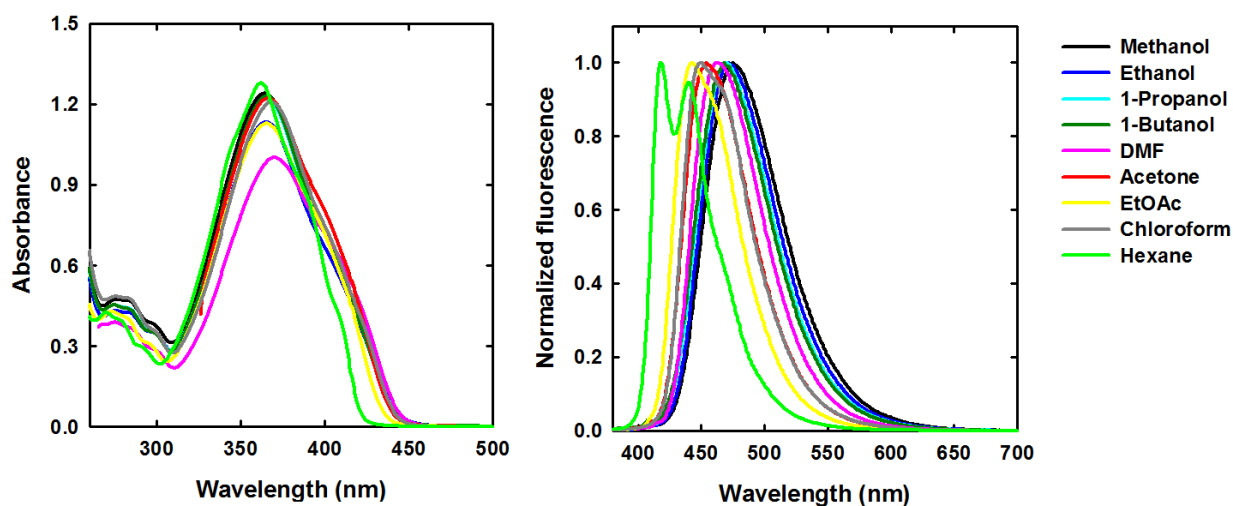
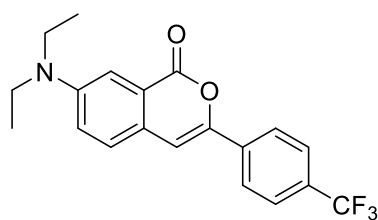
$[1b] = 63.42 \mu\text{M}$

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	358	473
Ethanol	Not soluble	Not soluble
1-Propanol	366	471
1-Butanol	366	471
1-Octanol	366	461
DMF	367	479
Acetone	358	461
Ethyl Acetate	356	446
Chloroform	351	439
Hexane	Not soluble	Not soluble



Molecular volume = 370.81 \AA^3
 $\mu = 2.18$ Debye (Discovery Studio)
Sphere
 Onsager cavity radius = 4.46 \AA
aprotic: $\mu^* - \mu = 6.69 \pm 1.09$ Debye
protic: $\mu^* - \mu = 11.20 \pm 0.82$ Debye
Spheroid prolate
 $a = 6.46 \text{ \AA}$
 $b = d = 2.02 \text{ \AA}$
aprotic: $\mu^* - \mu = 2.98 \pm 0.49$ Debye
protic: $\mu^* - \mu = 4.99 \pm 0.37$ Debye

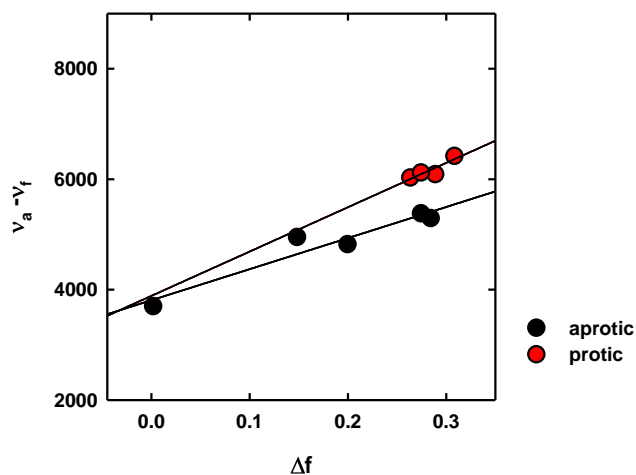
1c



Calculated $\epsilon_{366\text{nm}}$ in ethanol = $20488 \text{ M}^{-1} \text{ cm}^{-1}$

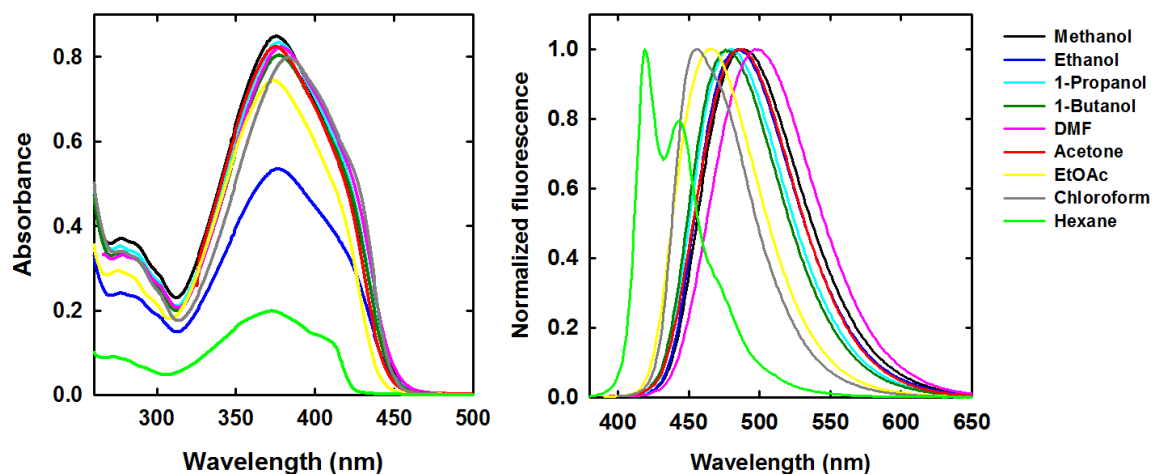
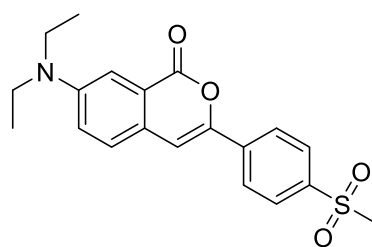
Measurements carried out at 20°C , $l=1 \text{ cm}$
 $[1c] = 55.35 \mu\text{M}$

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	364	475
Ethanol	366	471
1-Propanol	365	470
1-Butanol	365	468
1-Octanol	Not soluble	Not soluble
DMF	370	462
Acetone	366	454
Ethyl Acetate	365	443
Chloroform	368	450
Hexane	362	418



Molecular volume = 471.12 \AA^3
 $\mu = 4.12 \text{ Debye}$ (Discovery Studio)
Sphere
 Onsager cavity radius = 4.83 \AA
aprotic: $\mu^* - \mu = 7.94 \pm 0.65 \text{ Debye}$
protic: $\mu^* - \mu = 9.48 \pm 1.68 \text{ Debye}$
Spheroid prolate
 $a = 7.62 \text{ \AA}$
 $b = d = 2.02 \text{ \AA}$
aprotic: $\mu^* - \mu = 3.41 \pm 0.28 \text{ Debye}$
protic: $\mu^* - \mu = 4.07 \pm 0.72 \text{ Debye}$

1d

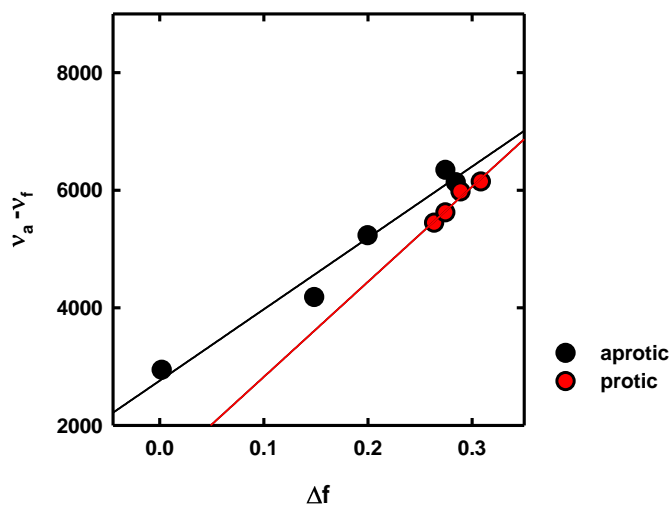


Calculated $\epsilon_{376\text{nm}}$ in ethanol = $9937 \text{ M}^{-1} \text{ cm}^{-1}$

Measurements carried out at 20°C , $l=1 \text{ cm}$

[1d] = $53.84 \mu\text{M}$

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	376	489
Ethanol	376	485
1-Propanol	378	480
1-Butanol	378	476
DMF	379	499
Acetone	375	487
Ethyl Acetate	374	465
Chloroform	383	456
Hexane	373	419



Molecular volume = 484.89 \AA^3
 $\mu = 3.02$ Debye (Discovery Studio)

Sphere

Onsager cavity radius = 4.87 \AA

aprotic: $\mu^* - \mu = 11.80 \pm 0.60$ Debye

protic: $\mu^* - \mu = 13.62 \pm 1.02$ Debye

Spheroid prolate

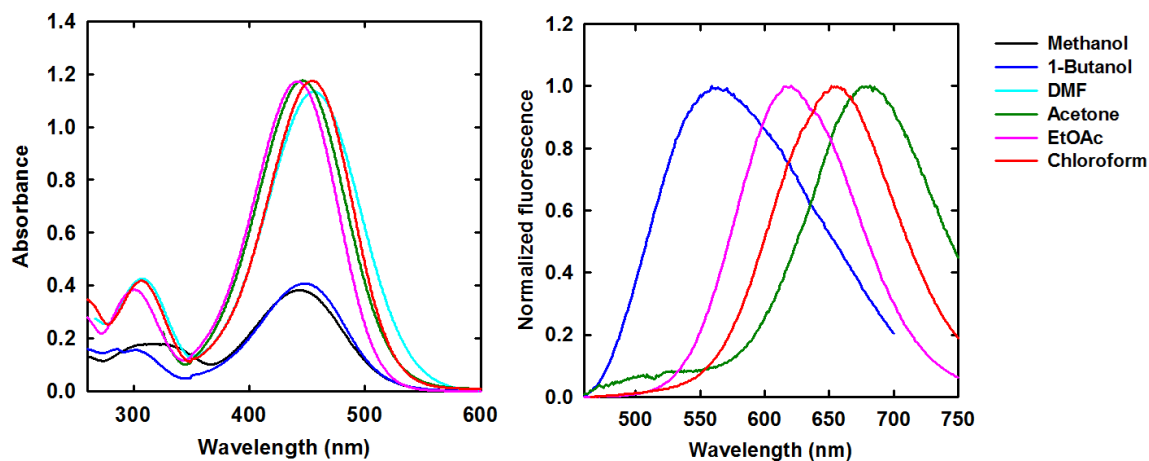
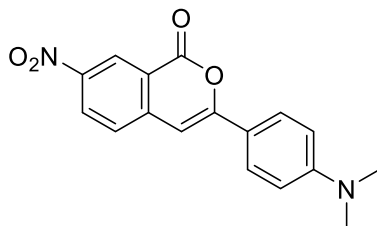
$a = 7.51 \text{ \AA}$

$b = d = 2.02 \text{ \AA}$

aprotic: $\mu^* - \mu = 4.96 \pm 0.25$ Debye

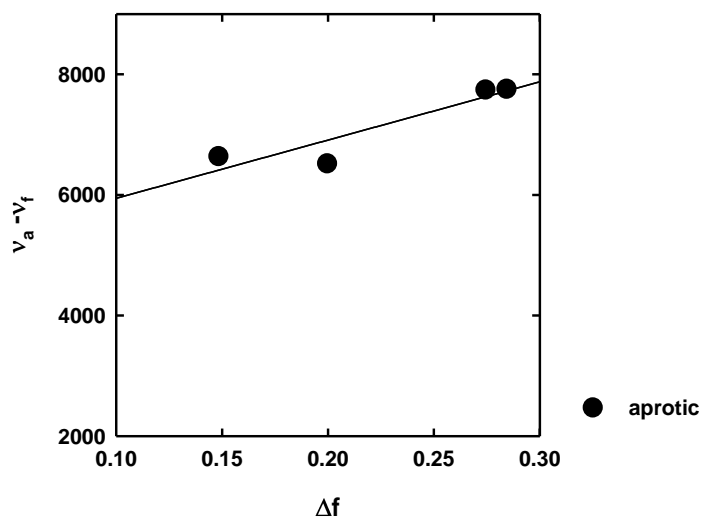
protic: $\mu^* - \mu = 5.73 \pm 0.43$ Debye

1e



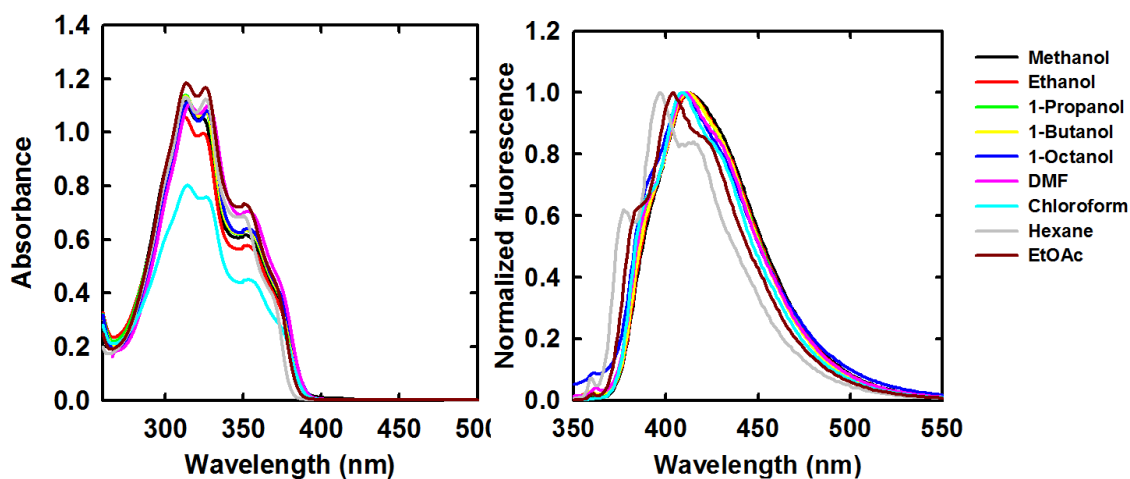
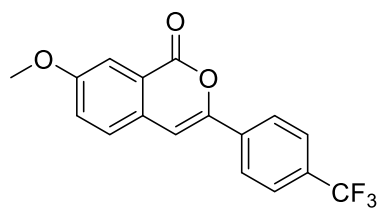
Measurements carried out at 20 °C, $l=1$ cm
 The molecule is poorly fluorescent in methanol and DMF

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	444	(605, poorly fluorescent)
Ethanol	Not soluble	Not soluble
1-Propanol	Not soluble	Not soluble
1-Butanol	448	559
1-Octanol	Not soluble	Not soluble
DMF	456	(605, poorly fluorescent)
Acetone	446	682
Ethyl Acetate	442	621
Chloroform	455	652
Hexane	Not soluble	Not soluble



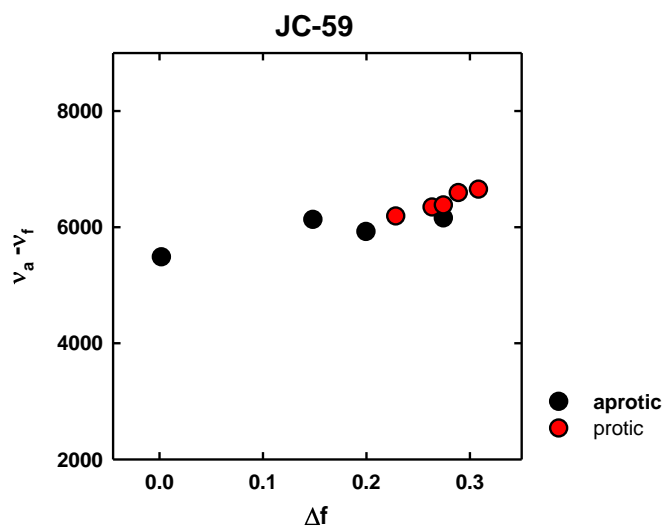
Molecular volume = 380.44 \AA^3
 $\mu = 6.17$ Debye (Discovery Studio)
Sphere
 Onsager cavity radius = 4.495 \AA
aprotic: $\mu^* - \mu = 9.33 \pm 1.42$ Debye
Spheroid prolate
 $a = 7.3 \text{ \AA}$
 $b = d = 2.57 \text{ \AA}$
aprotic: $\mu^* - \mu = 5.55 \pm 0.85$ Debye

1f



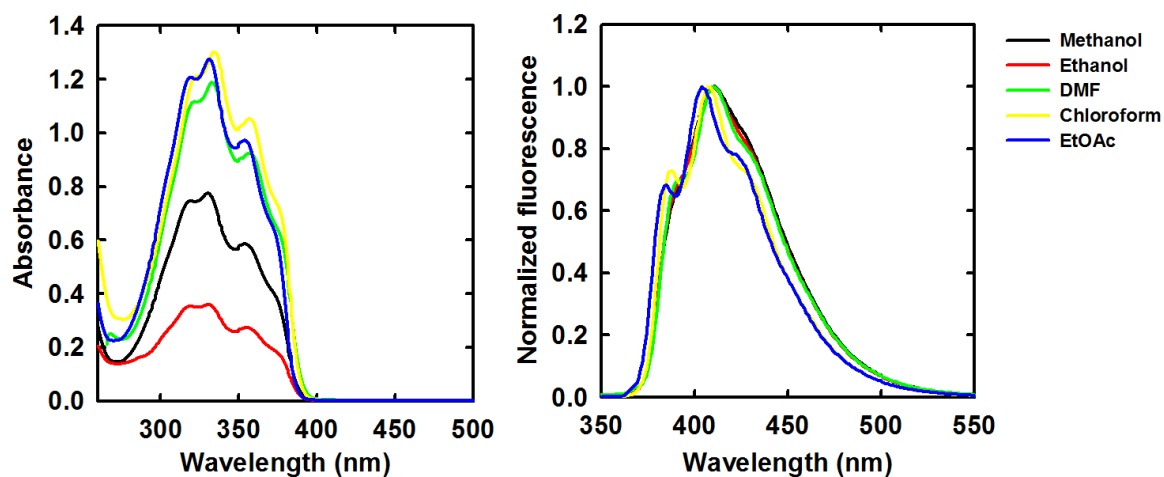
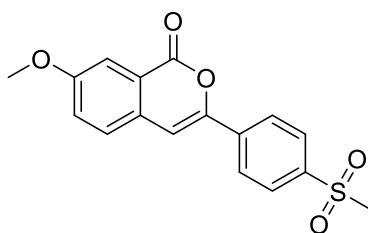
Calculated $\epsilon_{324\text{nm}}$ in ethanol = $15957 \text{ M}^{-1} \text{ cm}^{-1}$
 Measurements carried out at 20 °C, $l=1 \text{ cm}$
 $[1f]= 62.45 \mu\text{M}$

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	324	413
Ethanol	324	412
1-Propanol	325	410
1-Butanol	326	411
1-Octanol	327	410
DMF	328	411
Acetone	Solvent absorption	-
Ethyl Acetate	326	404
Chloroform	327	409
Hexane	326	397



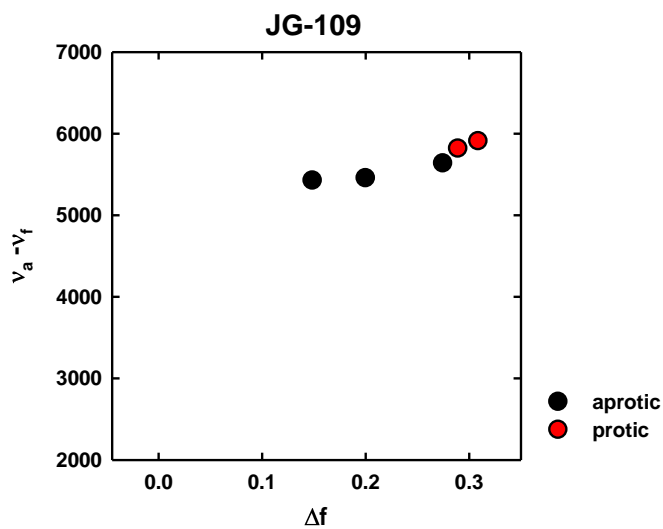
Molecular volume = 393.06 \AA^3
 $\mu = 4.984 \text{ Debye}$ (Discovery Studio)
 Onsager cavity radius = 4.54 \AA

1g



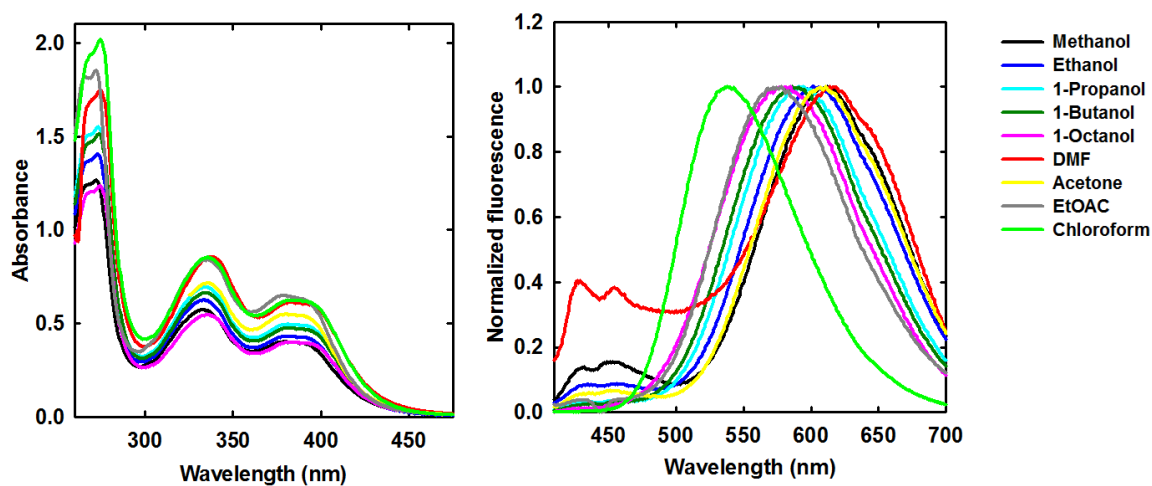
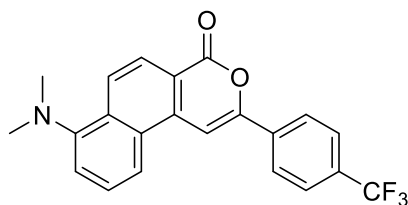
Measurements carried out at 20 °C, l=1 cm
[1g] = 60.54 μ M

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	330	410
Ethanol	331	410
1-Propanol	Not soluble	Not soluble
1-Butanol	Not soluble	Not soluble
1-Octanol	Not soluble	Not soluble
DMF	333	410
Acetone	Solvent absorption	-
Ethyl Acetate	331	404
Chloroform	334	408
Hexane	Not soluble	Not soluble



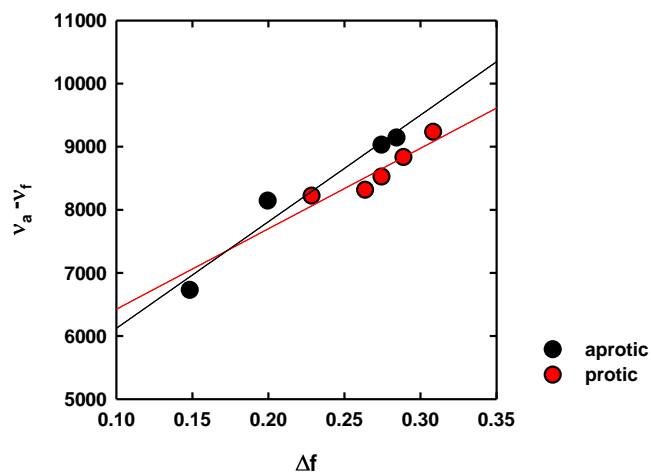
Molecular volume = 406.84 \AA^3
 μ = 6.323 Debye (Discovery Studio)
 Onsager cavity radius = 4.60 \AA

11c



Measurements carried out at 20 °C, l=1 cm

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	391	612
Ethanol	393	602
1-Propanol	393	591
1-Butanol	394	586
1-Octanol	395	585
DMF	395	614
Acetone	392	611
Ethyl Acetate	393	578
Chloroform	395	538
Hexane	Not soluble	Not soluble



Molecular volume = 472.93 \AA^3
 $\mu = 4.113$ Debye (Discovery Studio)

Sphere

Onsager cavity radius = 4.833 \AA

aprotic: $\mu^* - \mu = 13.77 \pm 1.06$ Debye

protic: $\mu^* - \mu = 11.96 \pm 1.45$ Debye

Spheroid prolate

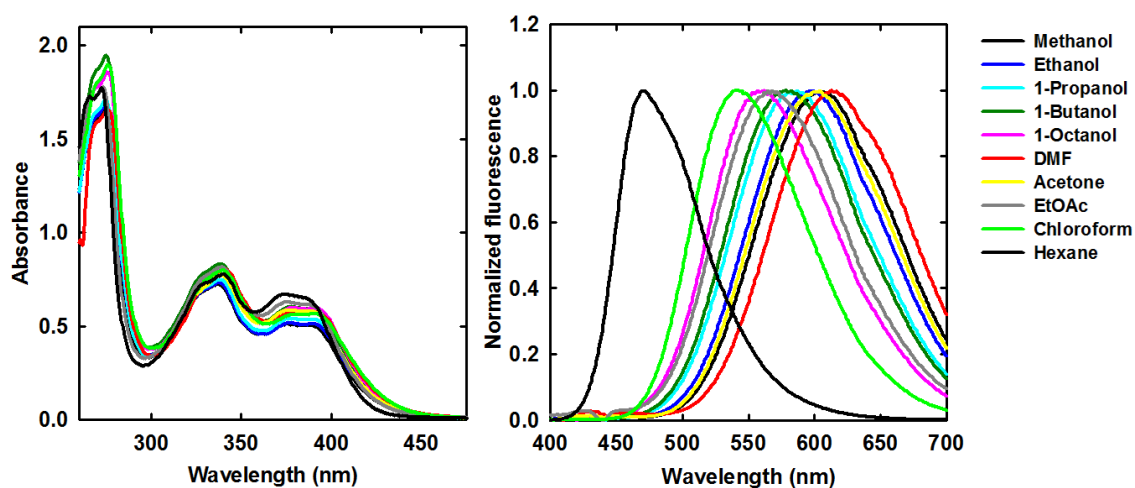
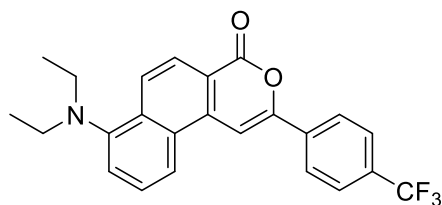
$a = 7.7 \text{ \AA}$

$b = d = 3.88 \text{ \AA}$

aprotic: $\mu^* - \mu = 11.39 \pm 0.88$ Debye

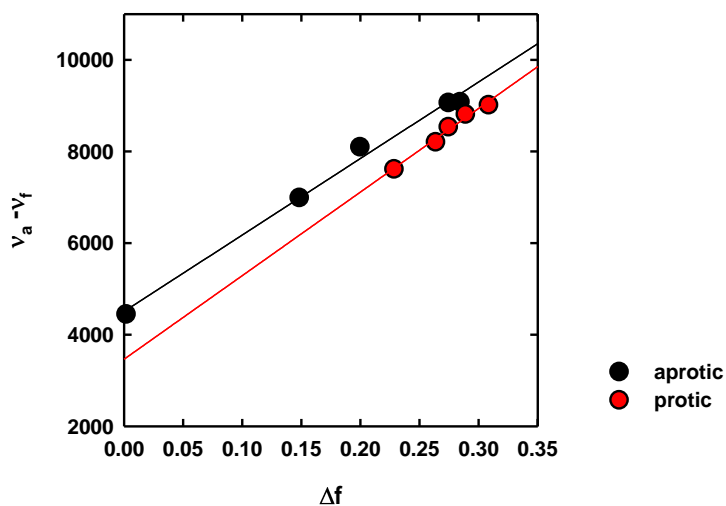
protic: $\mu^* - \mu = 9.90 \pm 1.20$ Debye

11d



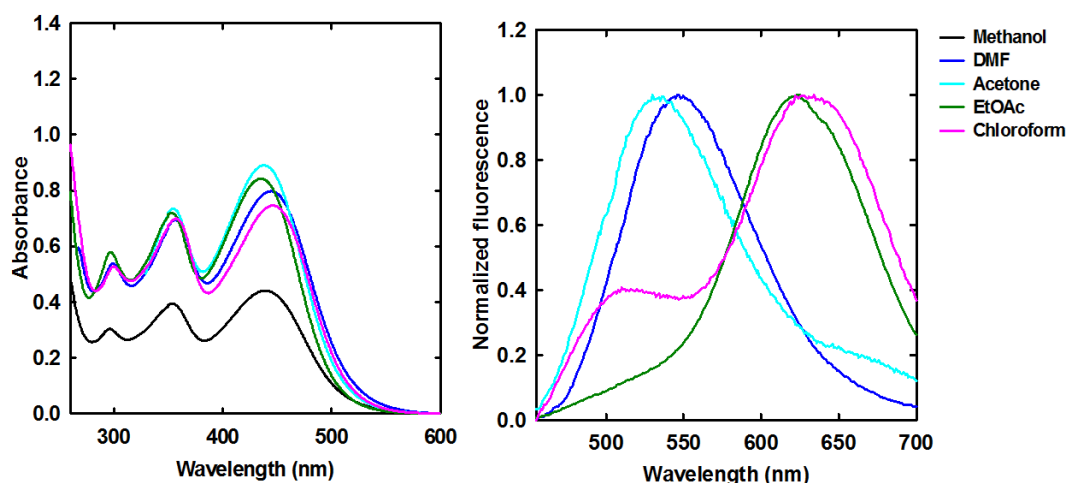
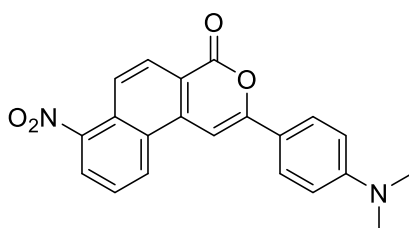
Measurements carried out at 20 °C, l=1 cm

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	391	604
Ethanol	392	599
1-Propanol	391	587
1-Butanol	392	578
1-Octanol	393	561
DMF	394	613
Acetone	390	604
Ethyl Acetate	389	568
Chloroform	393	542
Hexane	388	469



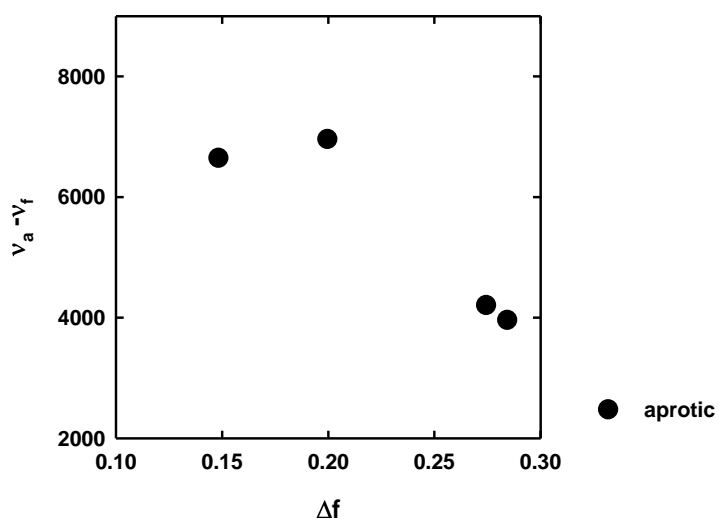
Molecular volume = 527.9 Å³
 $\mu = 4.267$ Debye (Discovery Studio)
Sphere
 Onsager cavity radius = 5.014 Å
aprotic: $\mu^* - \mu = 11.36 \pm 0.28$ Debye
protic: $\mu^* - \mu = 11.87 \pm 0.45$ Debye
Spheroid prolate
 $a = 8.3$ Å
 $b = d = 3.88$ Å
aprotic: $\mu^* - \mu = 11.76 \pm 0.29$ Debye
protic: $\mu^* - \mu = 12.29 \pm 0.47$ Debye

11f



Measurements carried out at 20 °C, l=1 cm
The molecule is poorly fluorescent in methanol.

Solvent	λ_{exc} (nm)	λ_{em} (nm)
Methanol	439	(548, poorly fluorescent)
Ethanol	Not soluble	Not soluble
1-Propanol	Not soluble	Not soluble
1-Butanol	Not soluble	Not soluble
1-Octanol	Not soluble	Not soluble
DMF	444	546
Acetone	438	530
Ethyl Acetate	435	624
Chloroform	446	634
Hexane	Not soluble	Not soluble

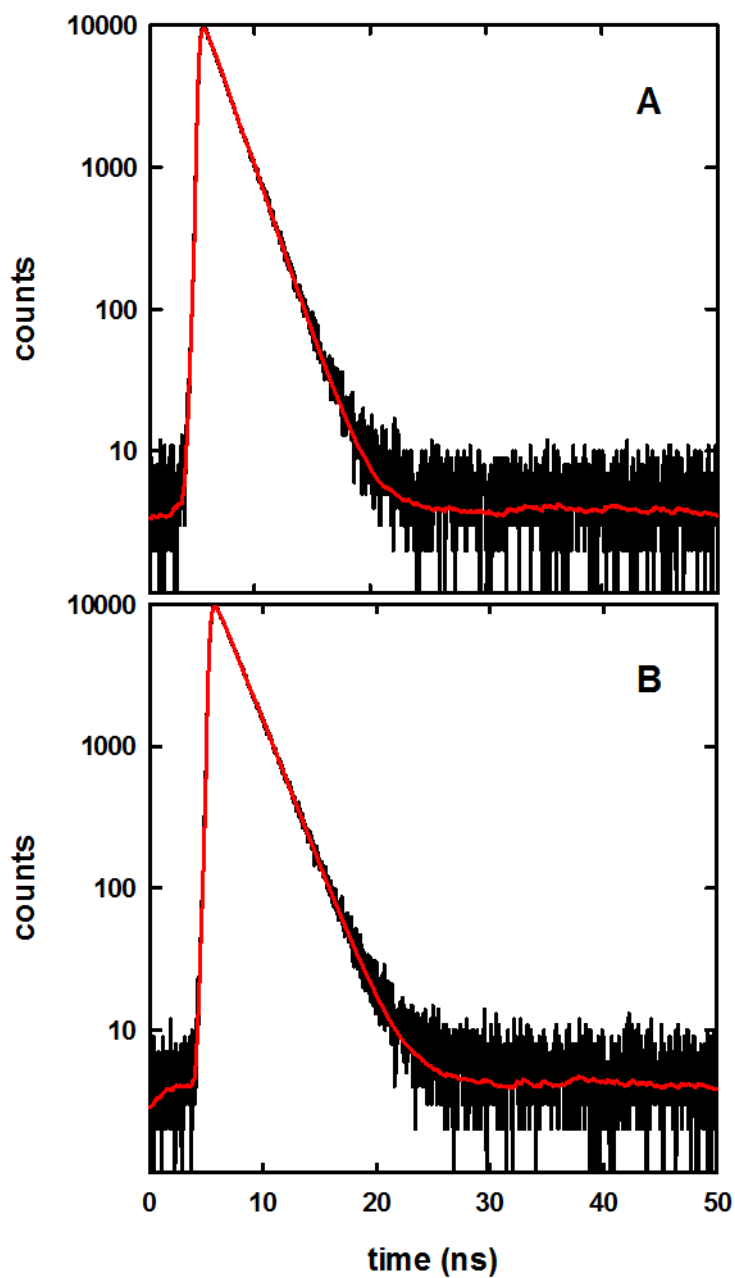


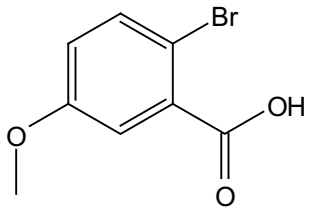
Molecular volume = 437.06 Å³
 μ = 6.236 Debye (Discovery Studio)
Sphere
 Onsager cavity radius = 4.708 Å

Spheroid prolate
 a = 7.85 Å
 b = d = 3.86 Å

Fluorescence lifetimes of **1d** and **11d**

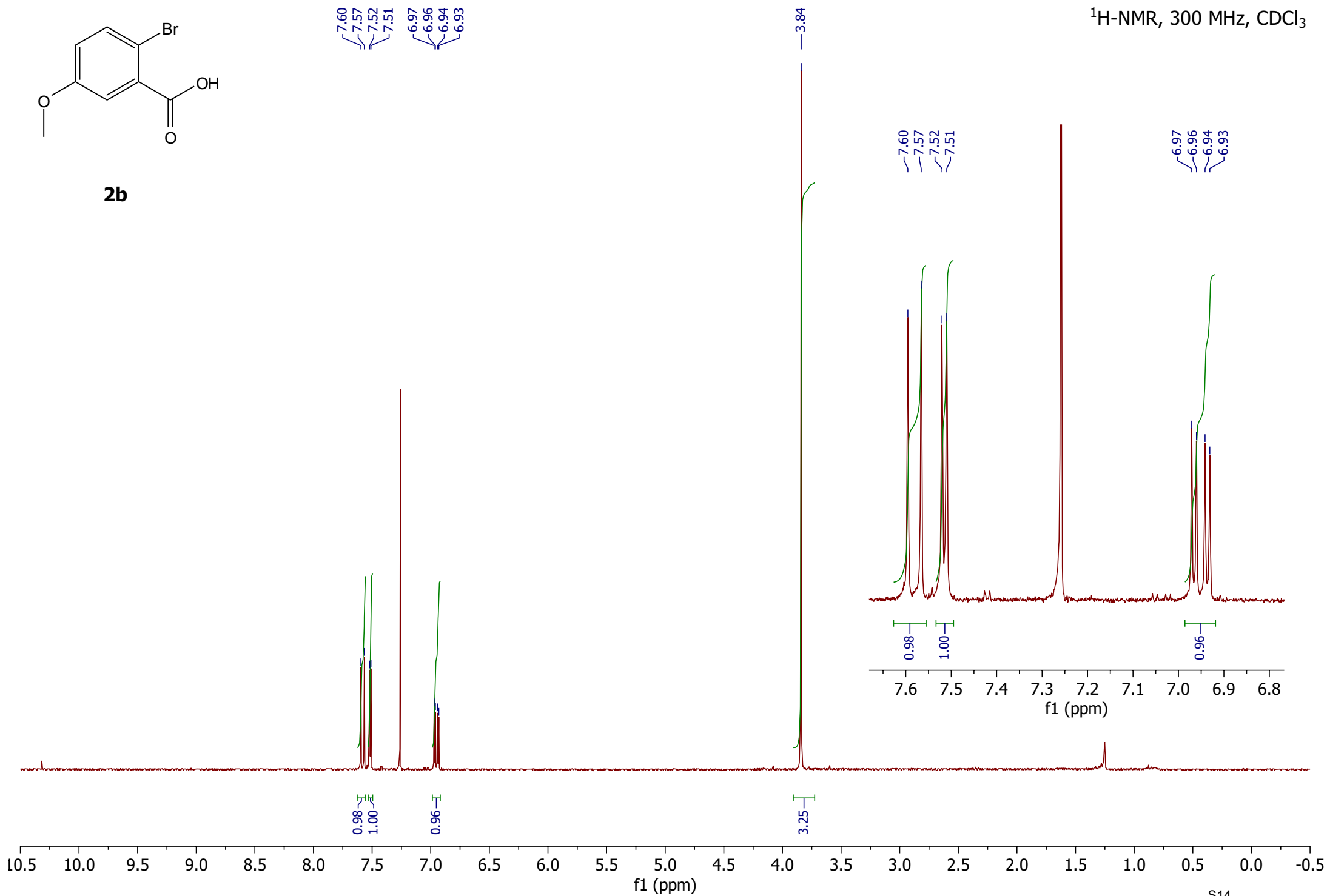
Fluorescence lifetime decays upon 375 nm excitation (0.2 μ s repetition rate) of **1d** (panel A) and **11d** (panel B) (black lines). Data, upon IRF correction were fitted to a single exponential decay (red lines).

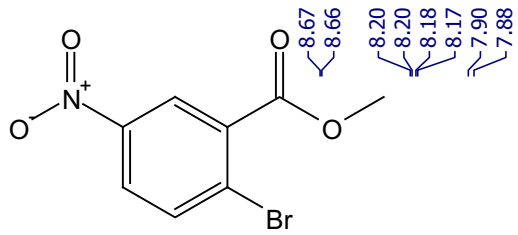




2b

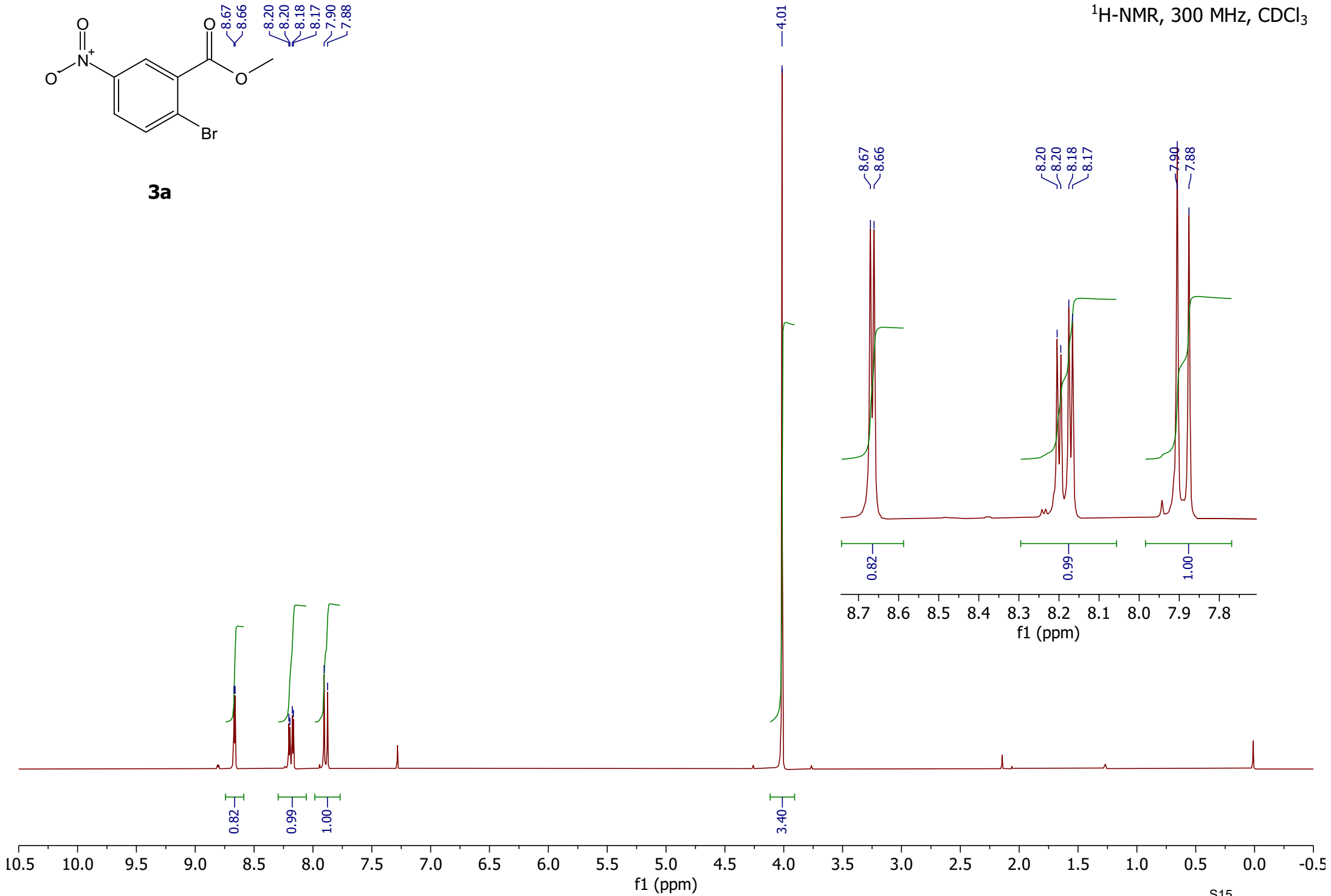
¹H-NMR, 300 MHz, CDCl₃

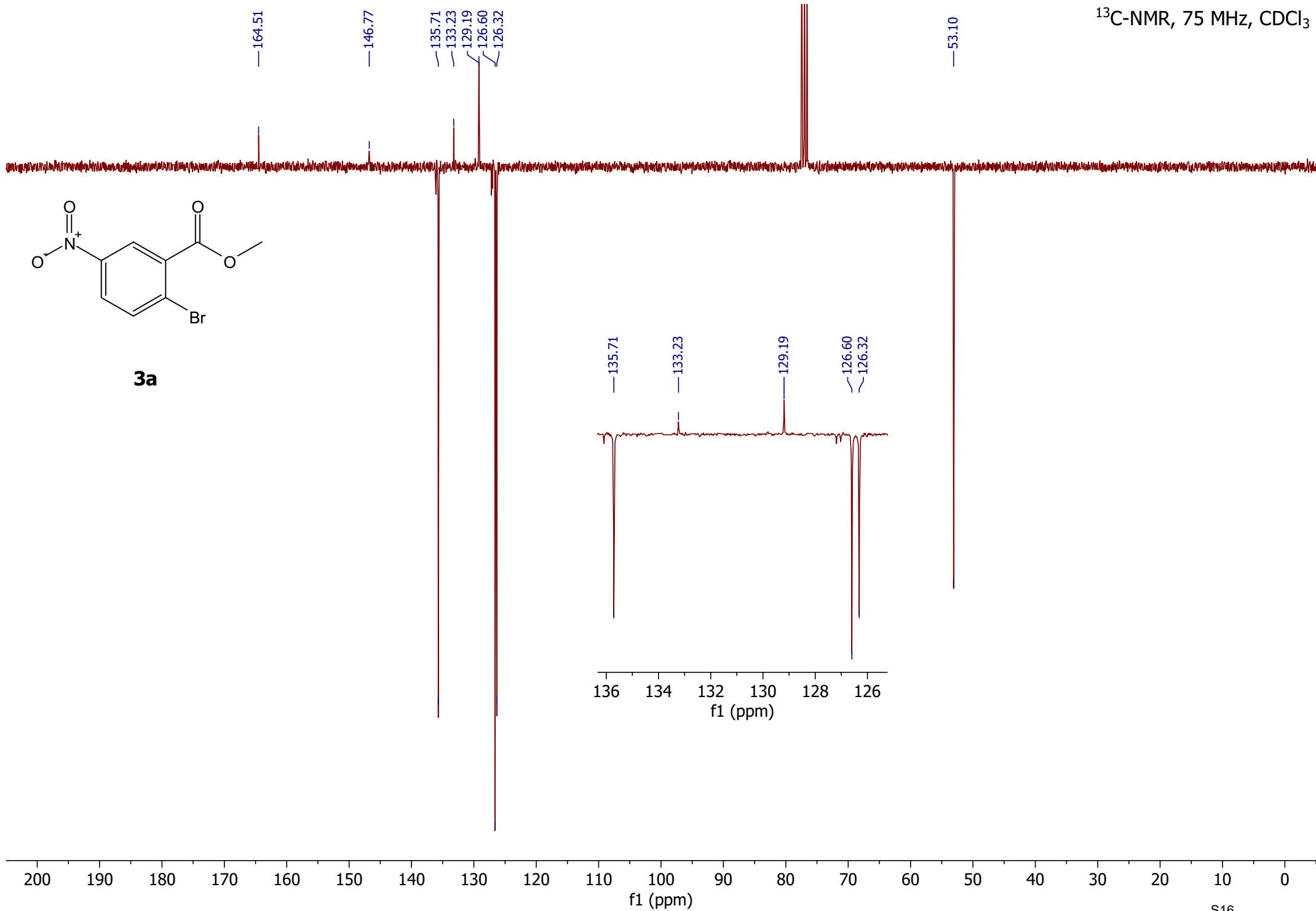


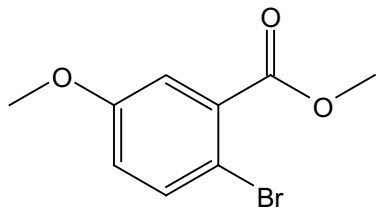


3a

¹H-NMR, 300 MHz, CDCl₃

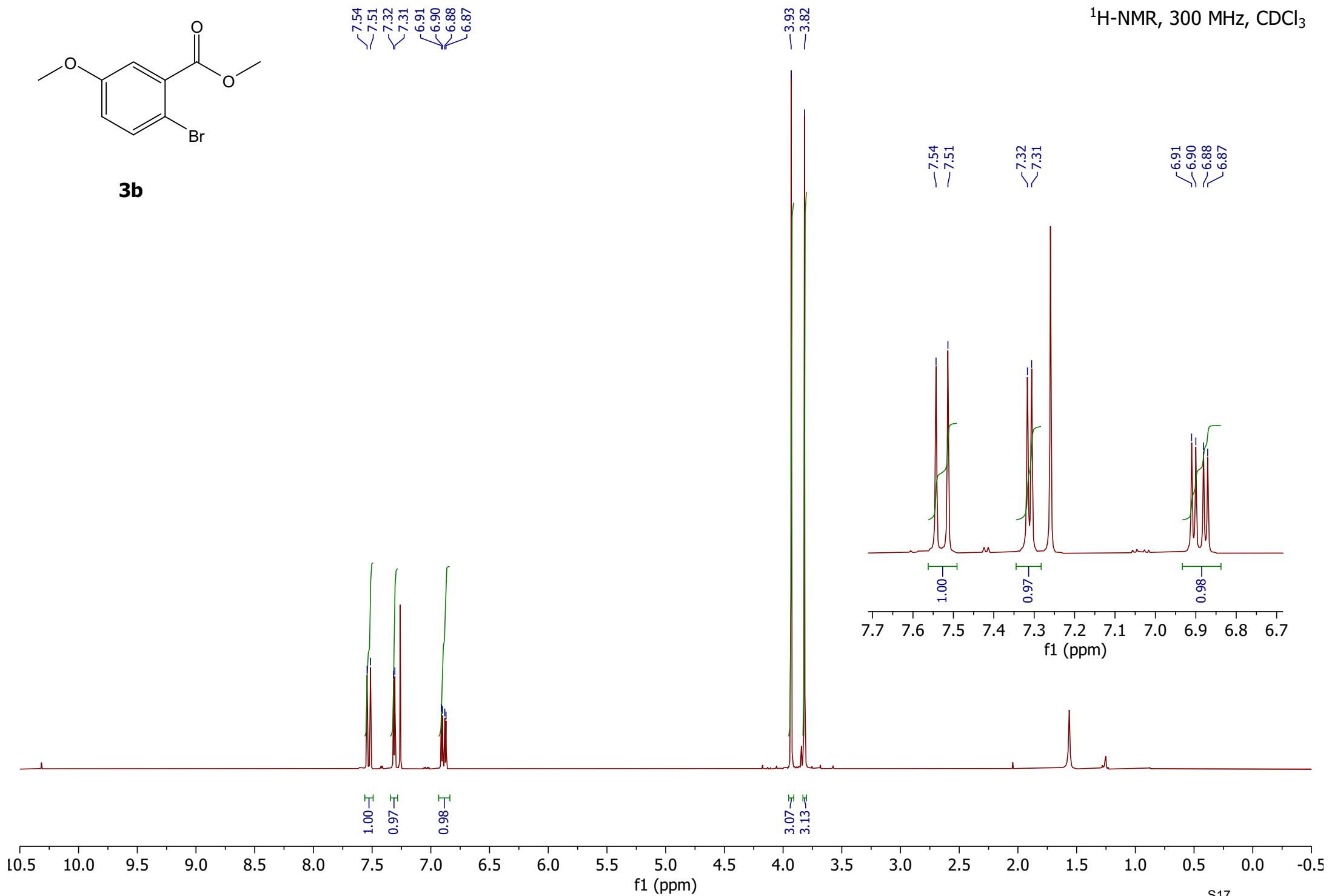


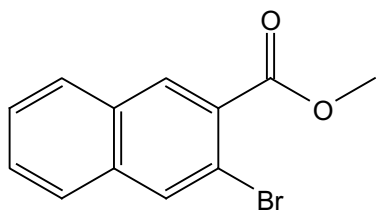




3b

¹H-NMR, 300 MHz, CDCl₃

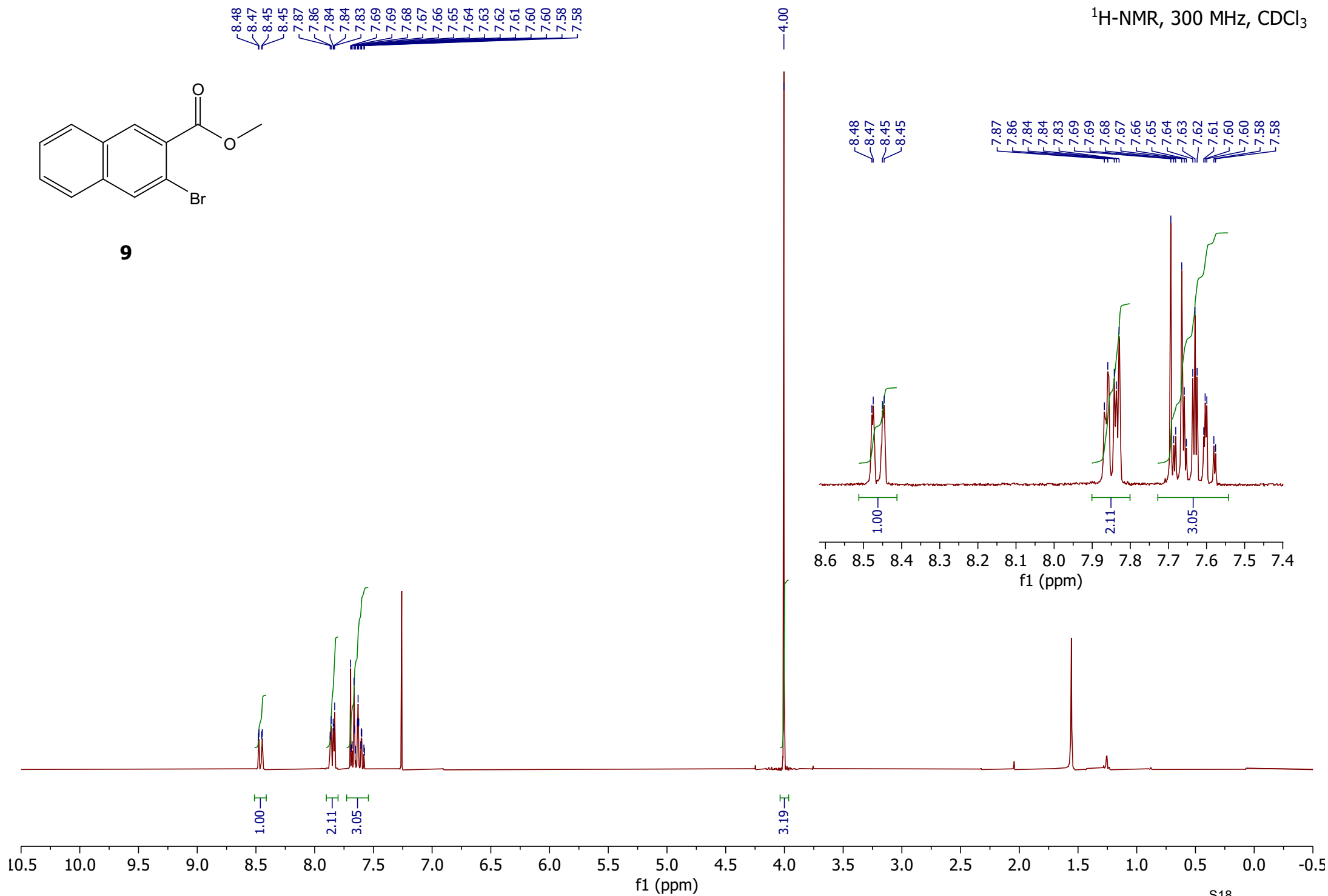


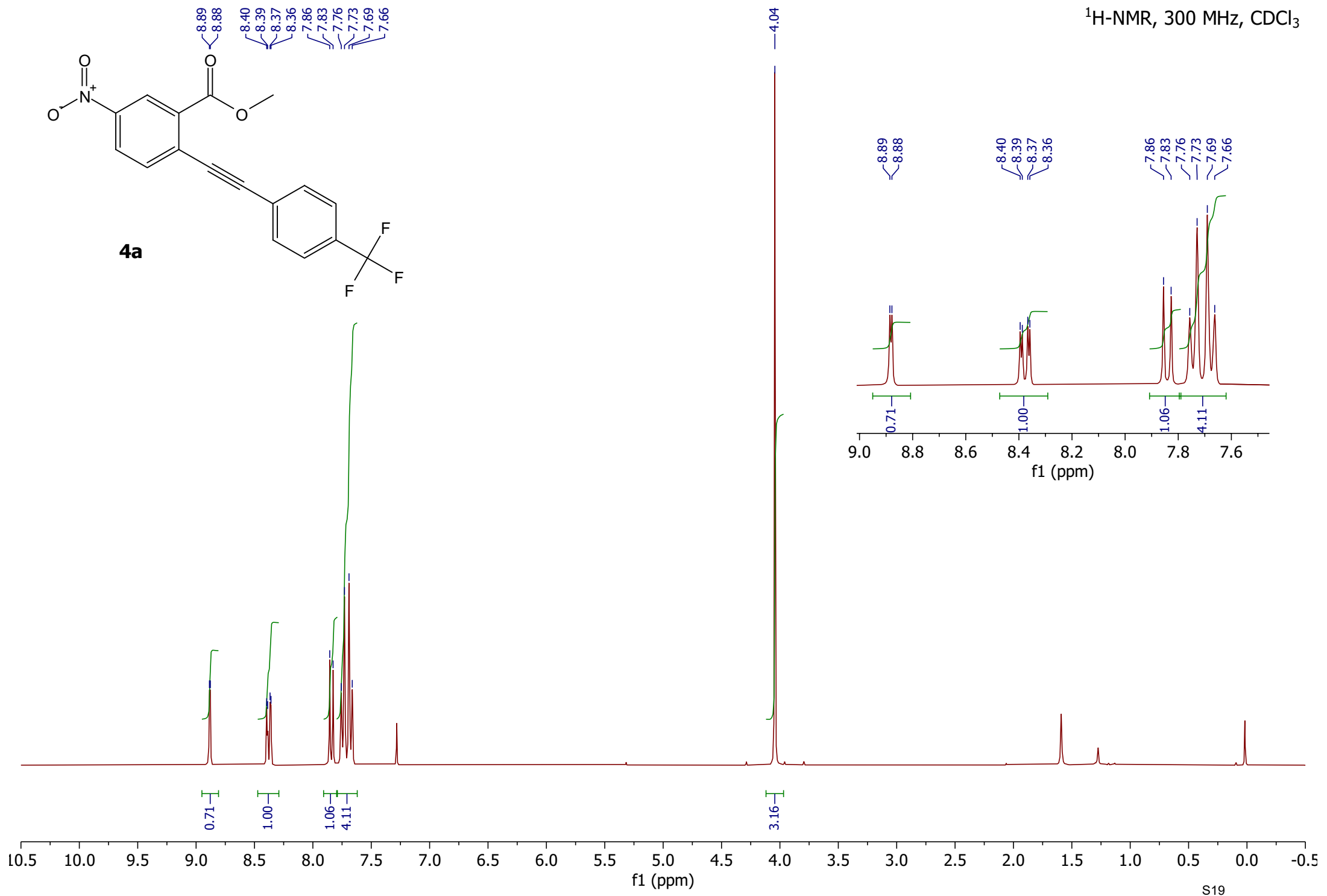
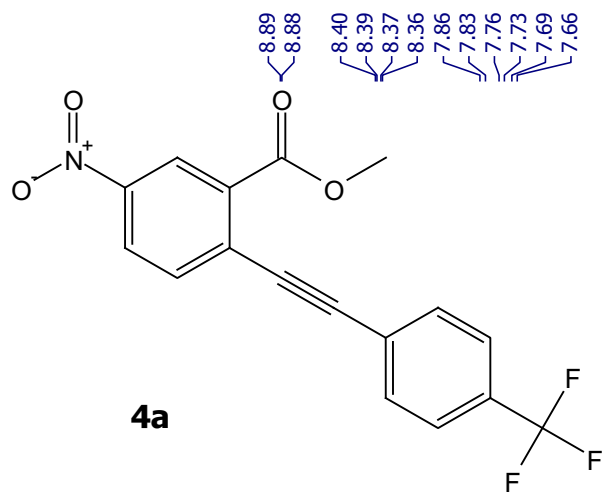


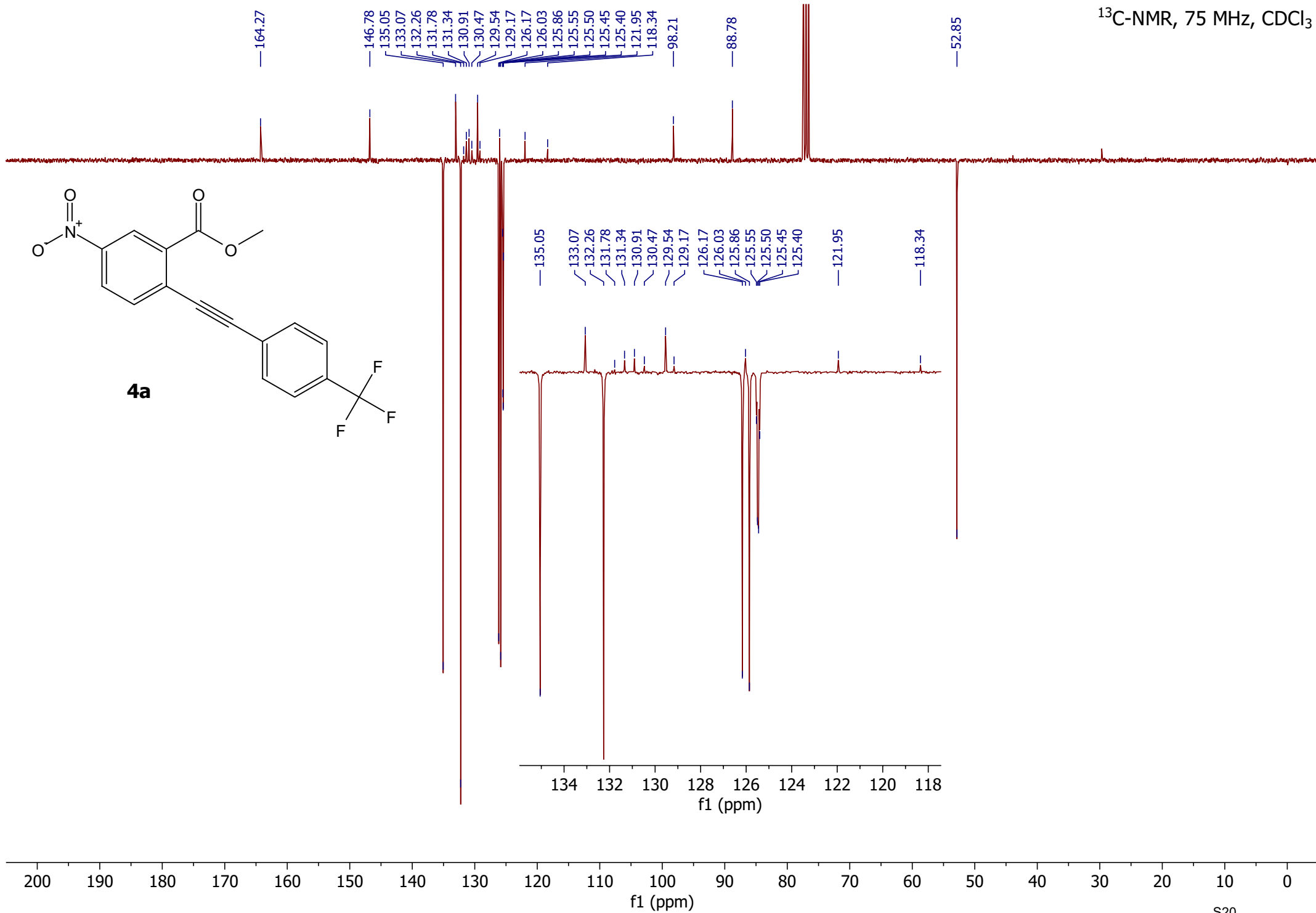
9

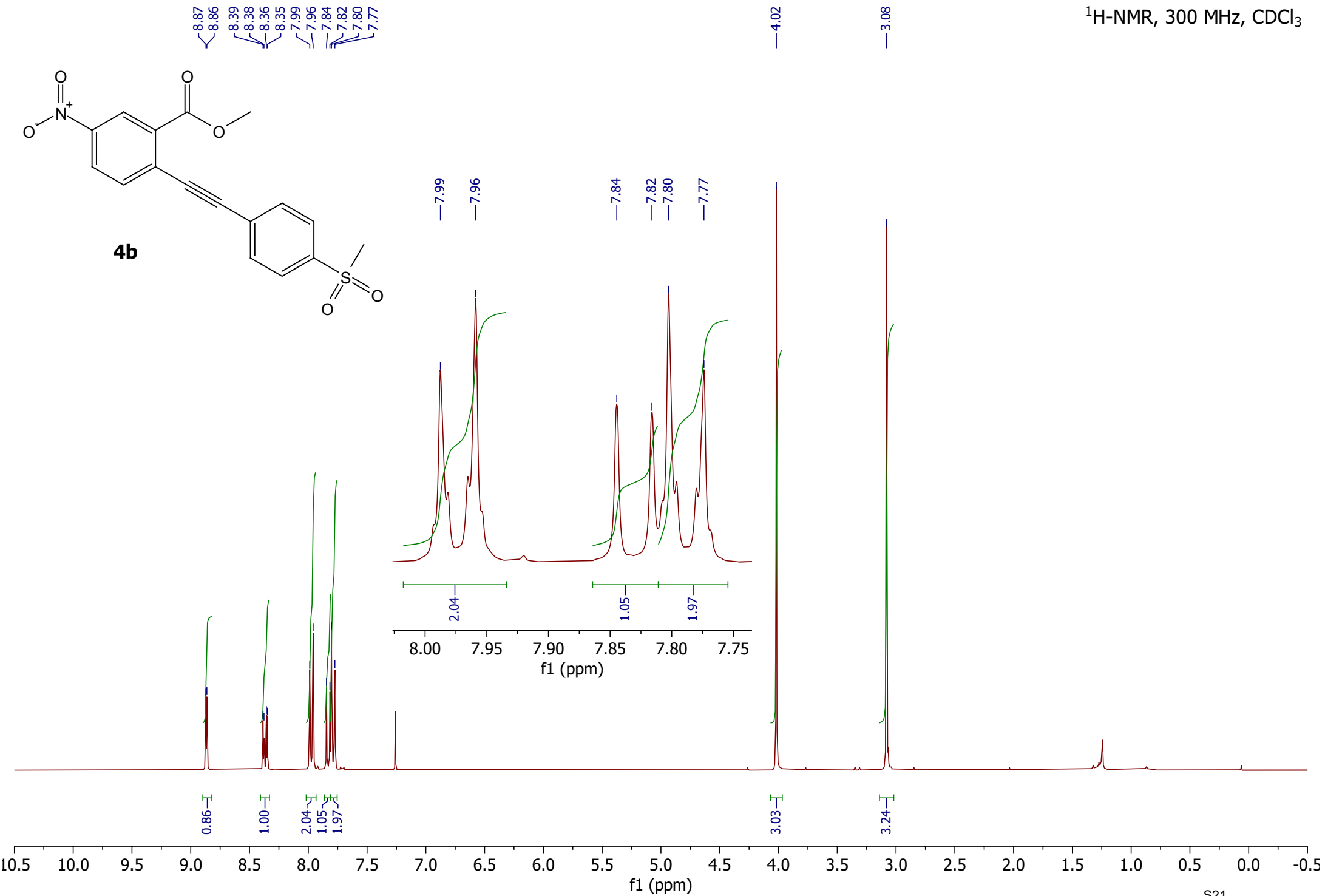
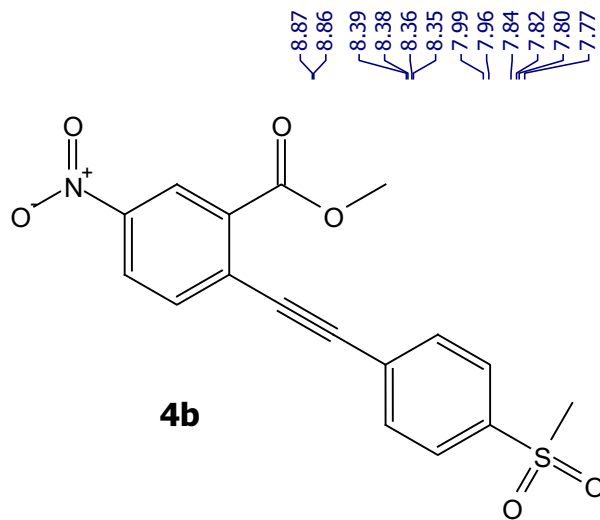
8.48
8.47
8.45
8.45
7.87
7.86
7.84
7.84
7.83
7.69
7.68
7.67
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7.64
7.63
7.62
7.61
7.60
7.60
7.58
7.58

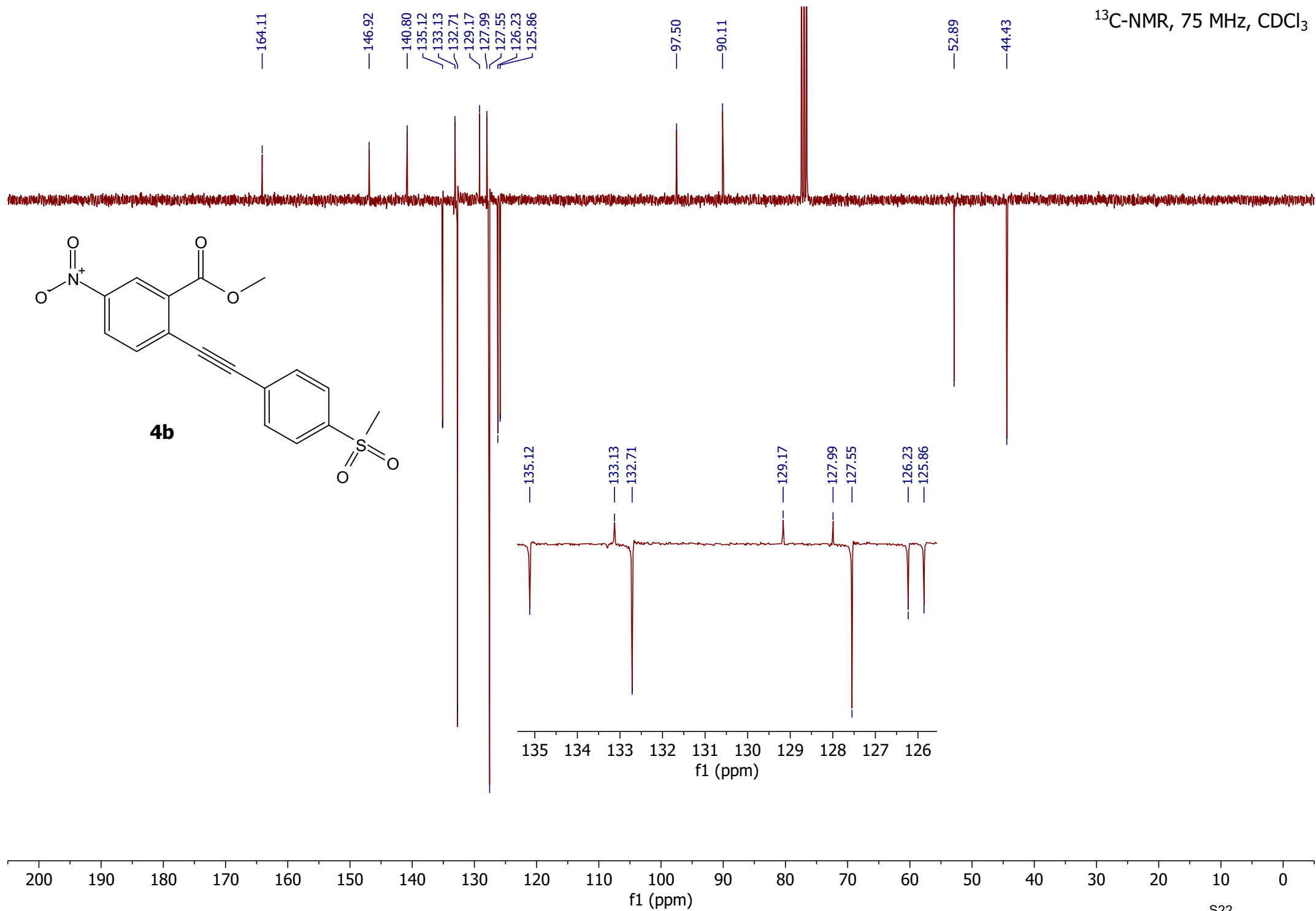
¹H-NMR, 300 MHz, CDCl₃

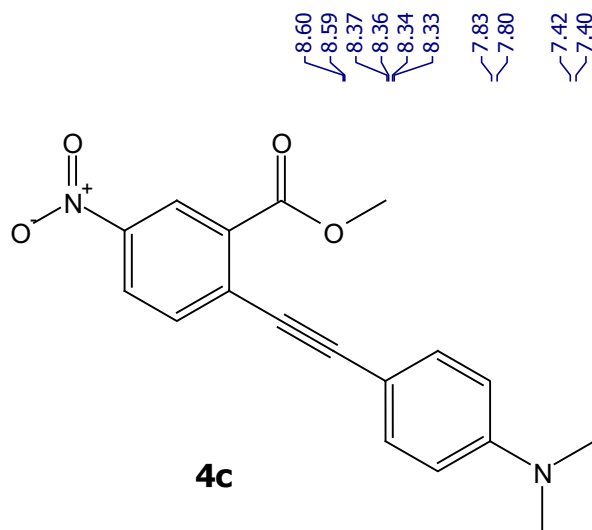








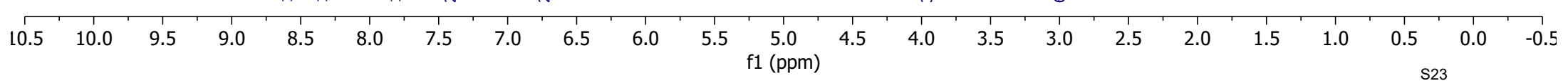




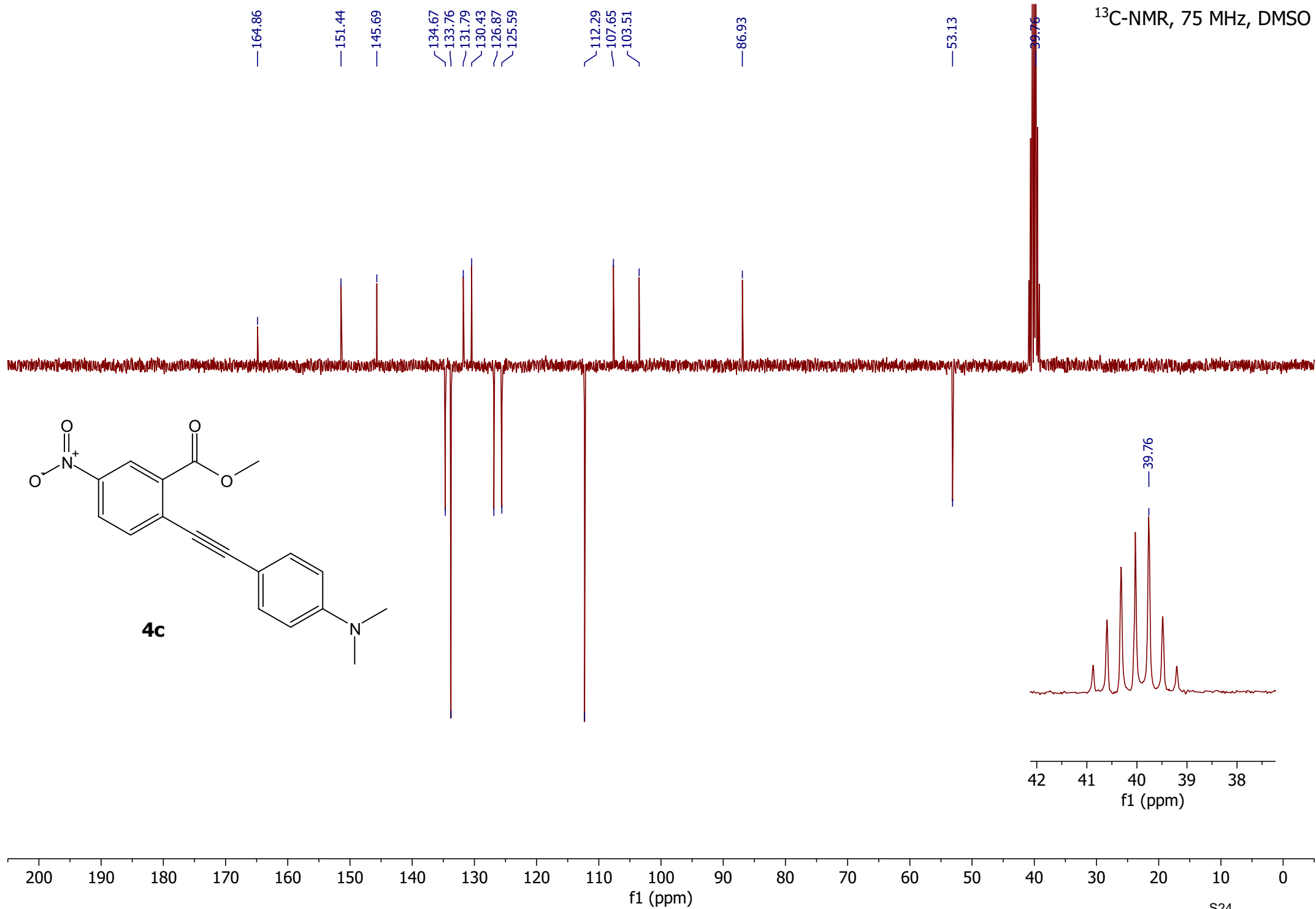
8.60
8.59
8.37
8.36
8.34
8.33
7.83
7.80
7.42
7.40
6.75
6.72

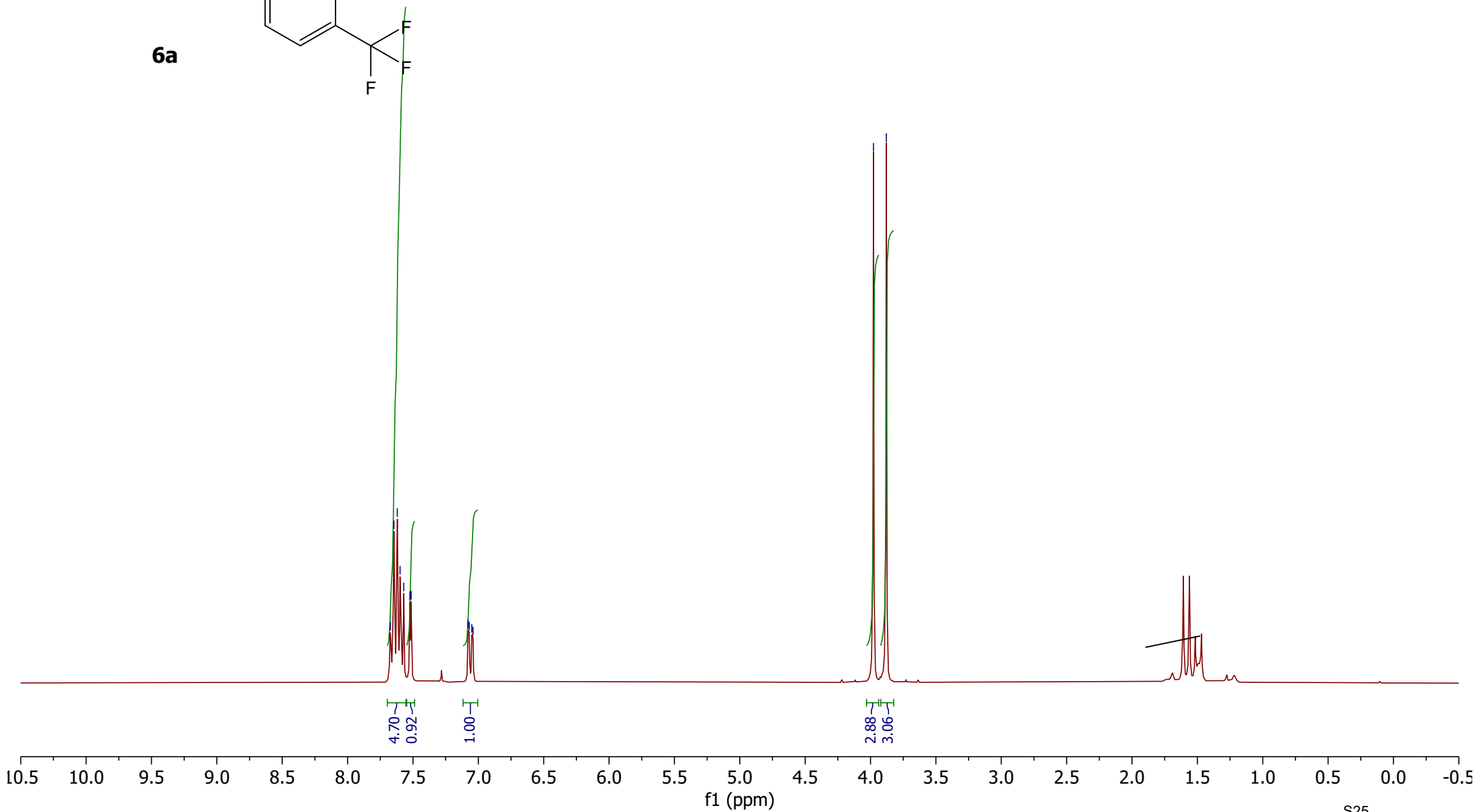
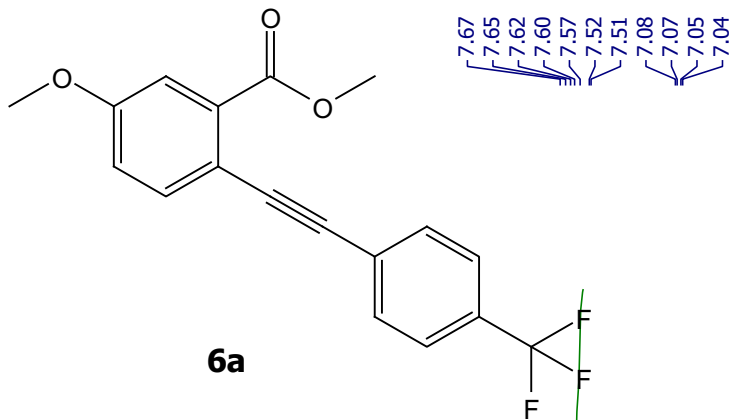
3.94

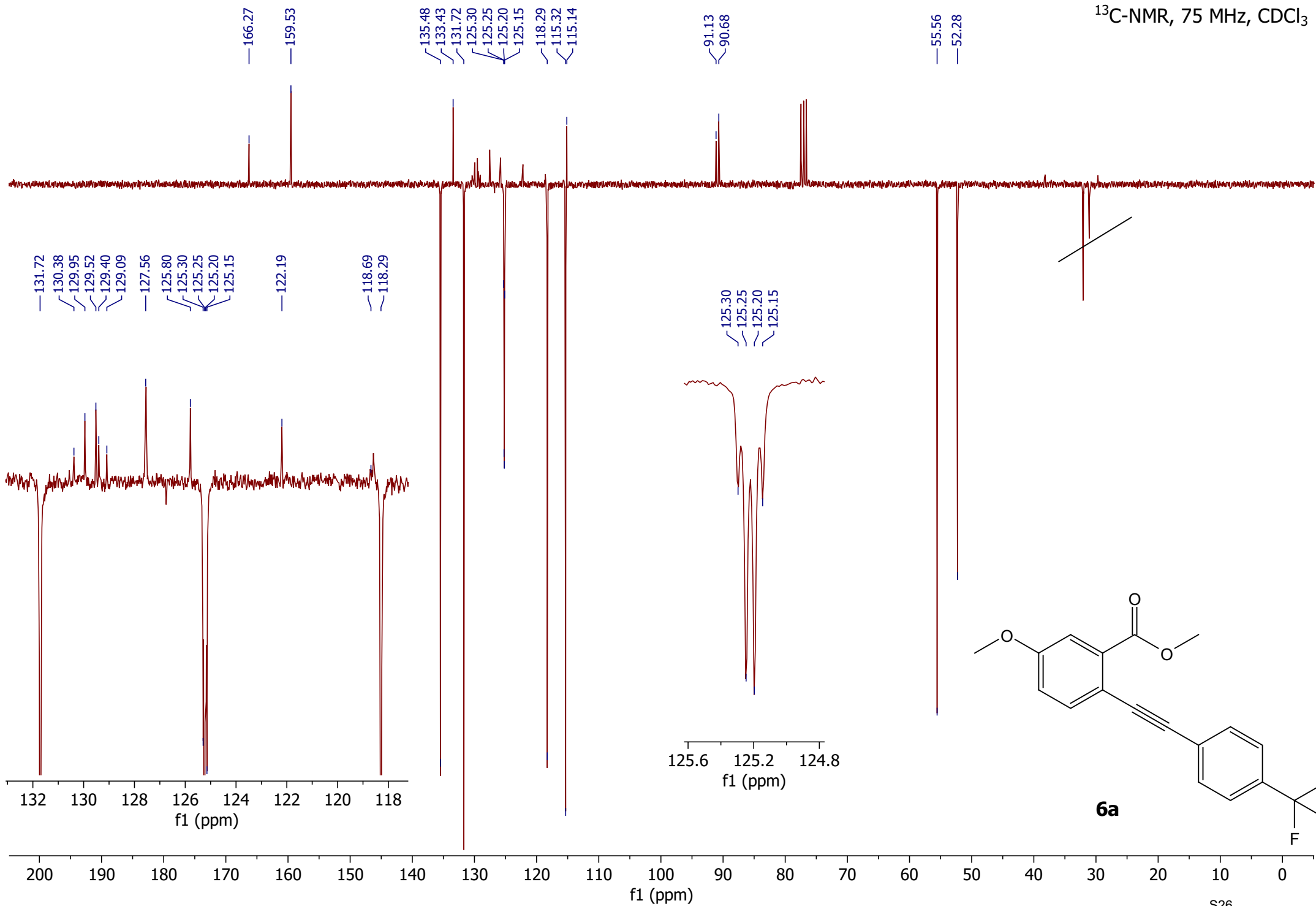
2.98



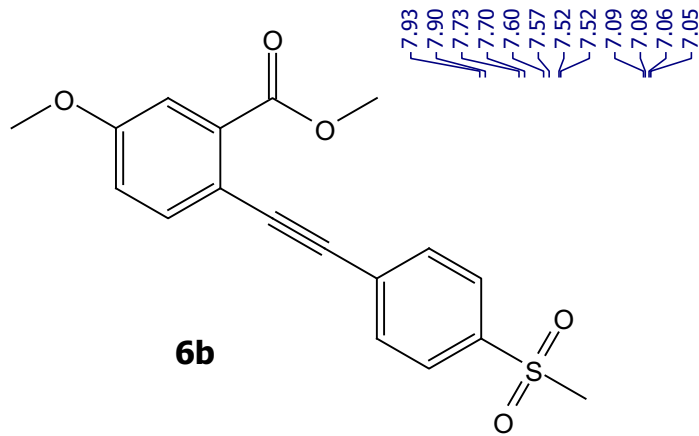
¹³C-NMR, 75 MHz, DMSO



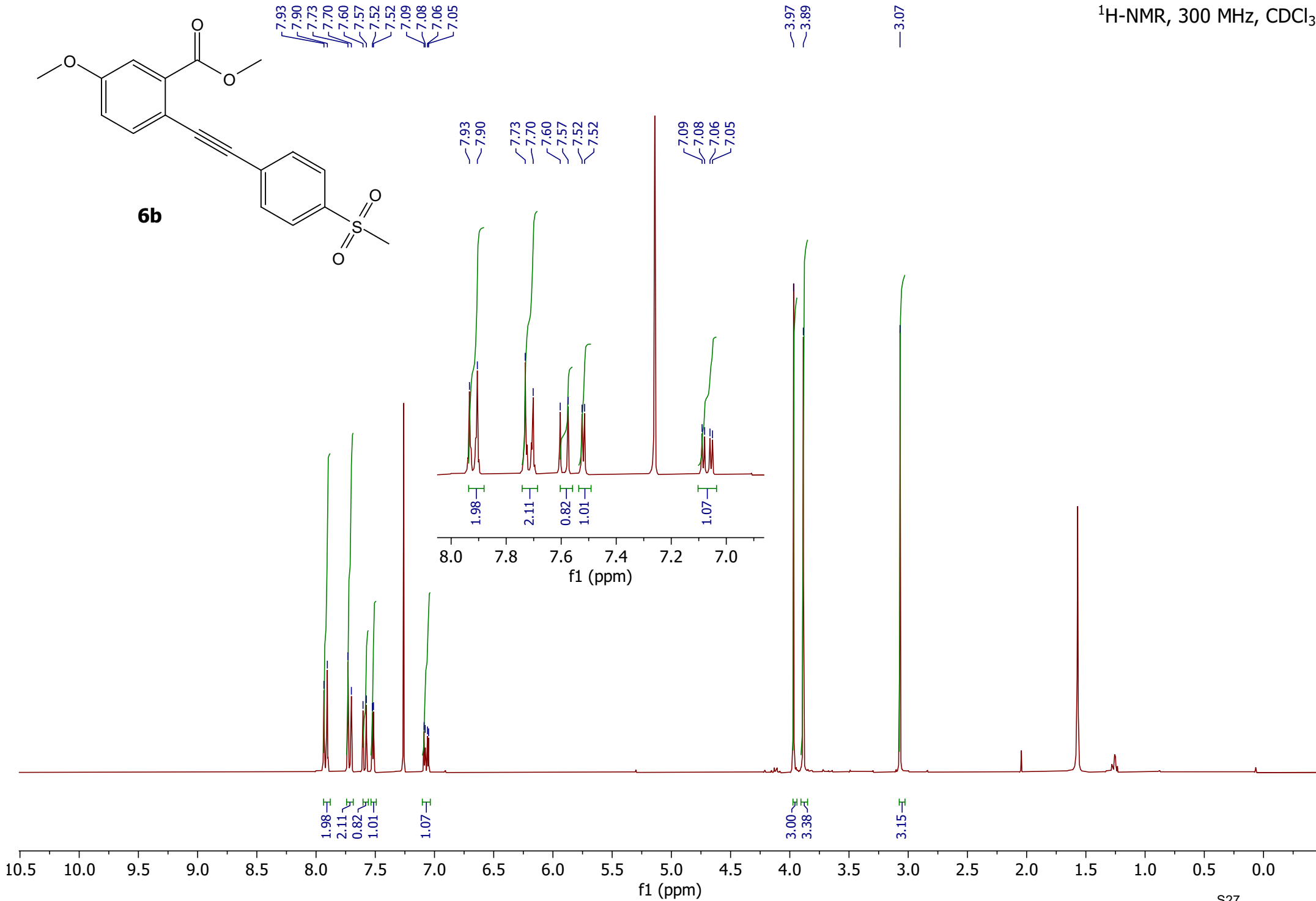


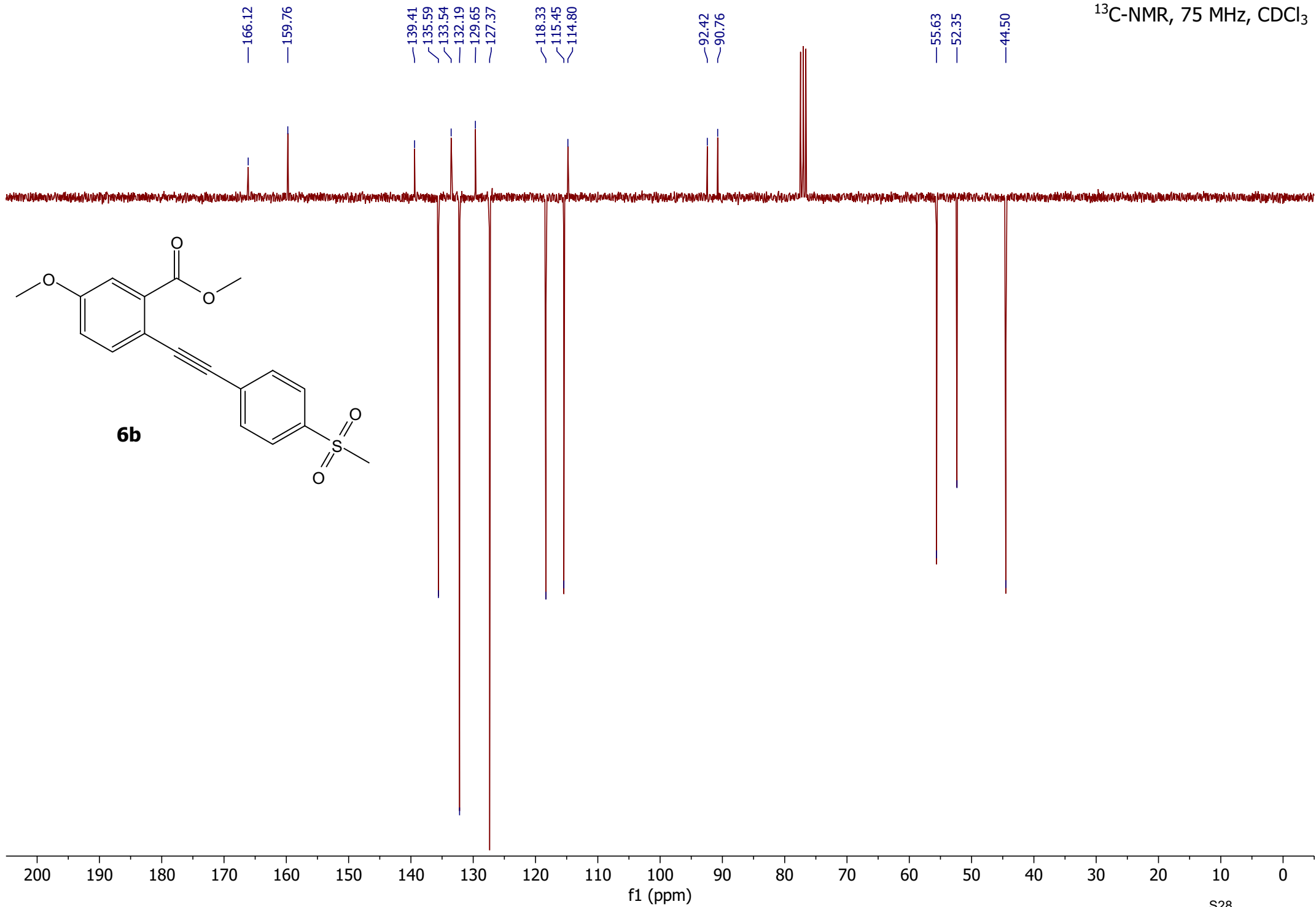


6a

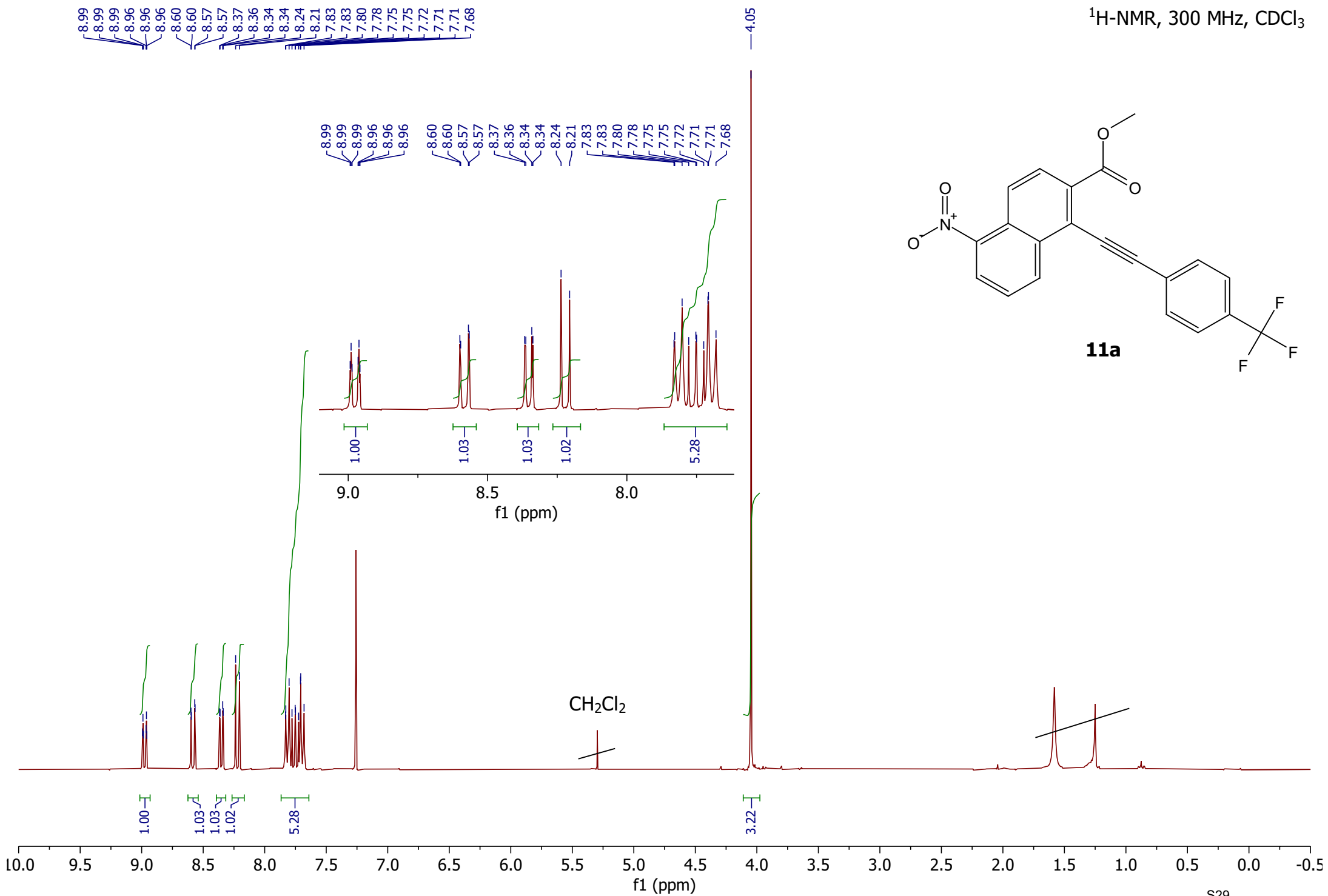
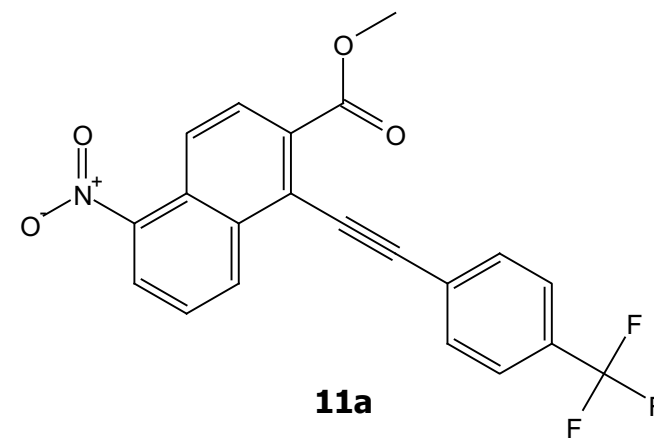


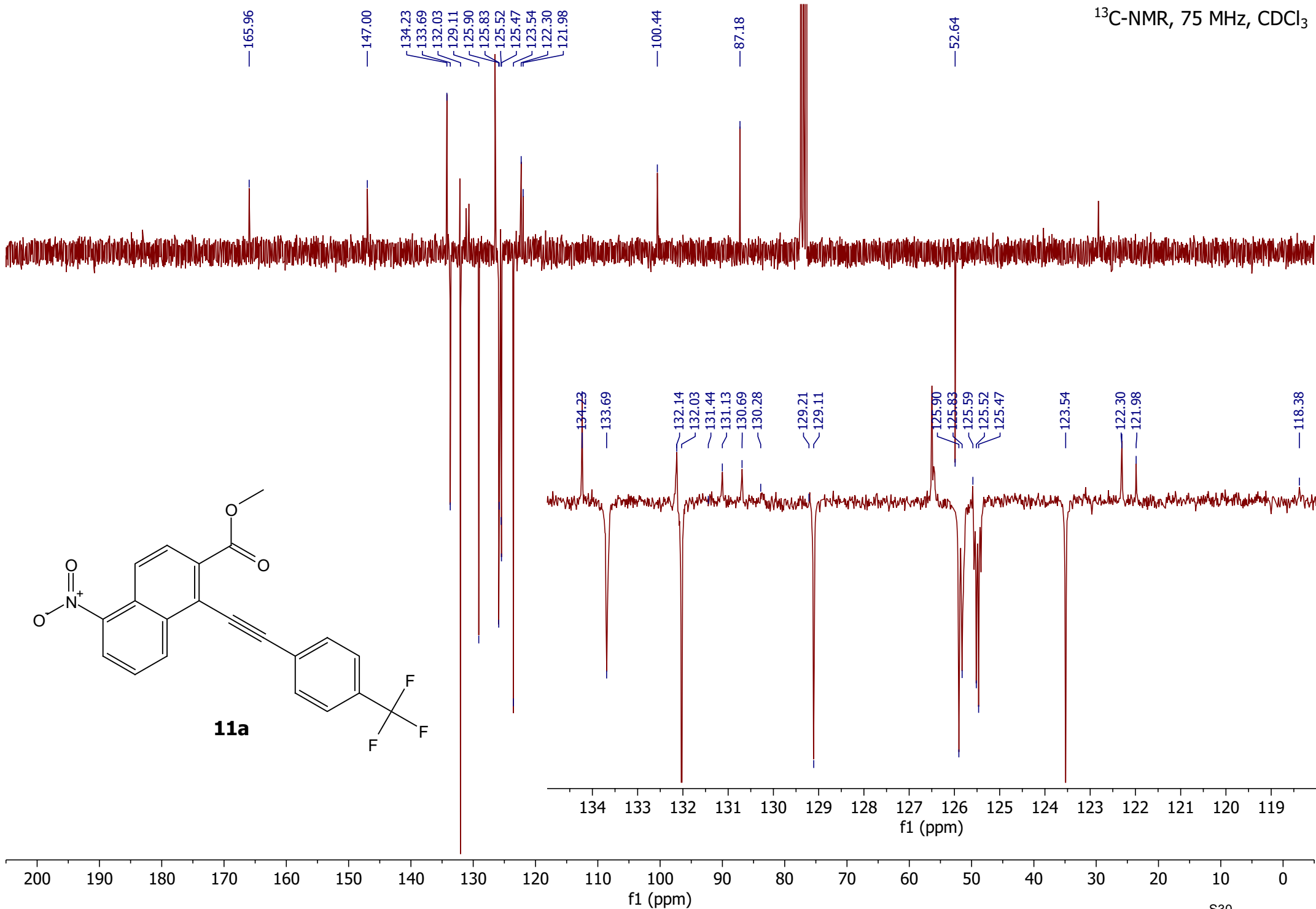
6b

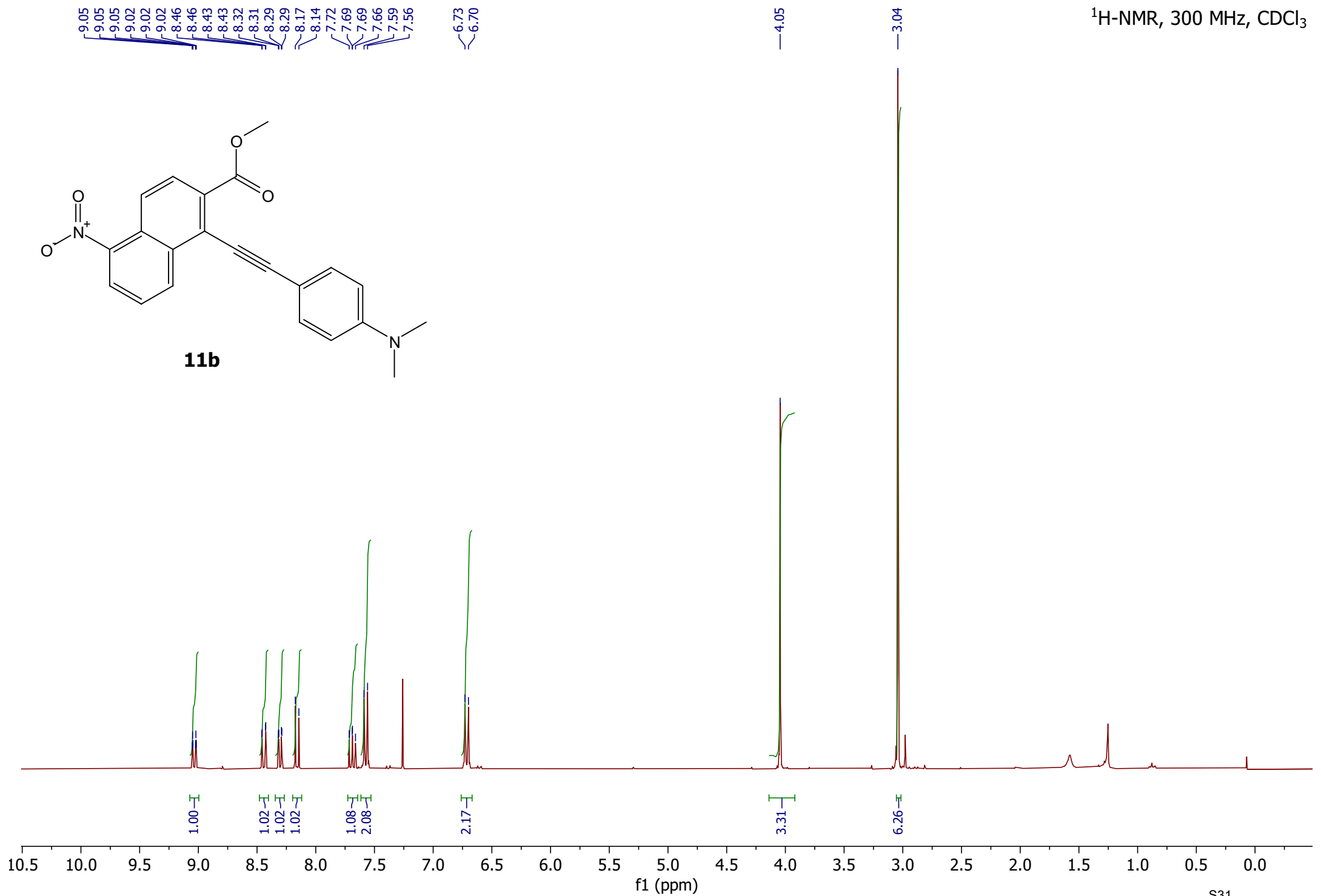
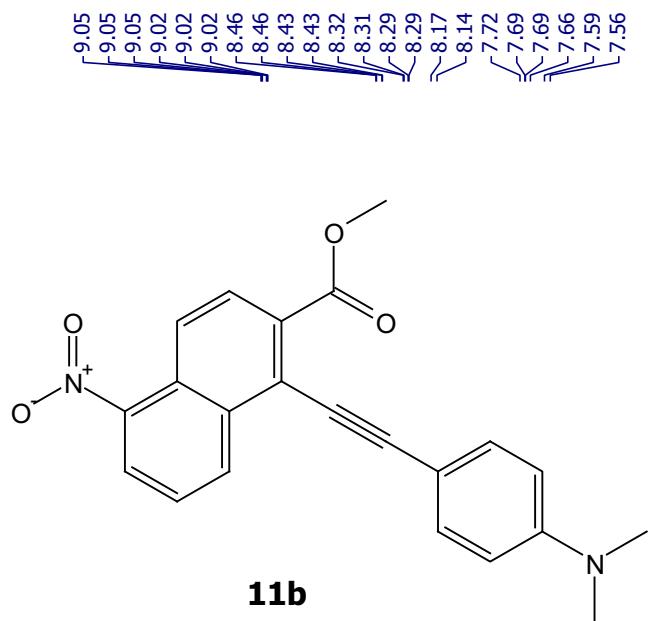


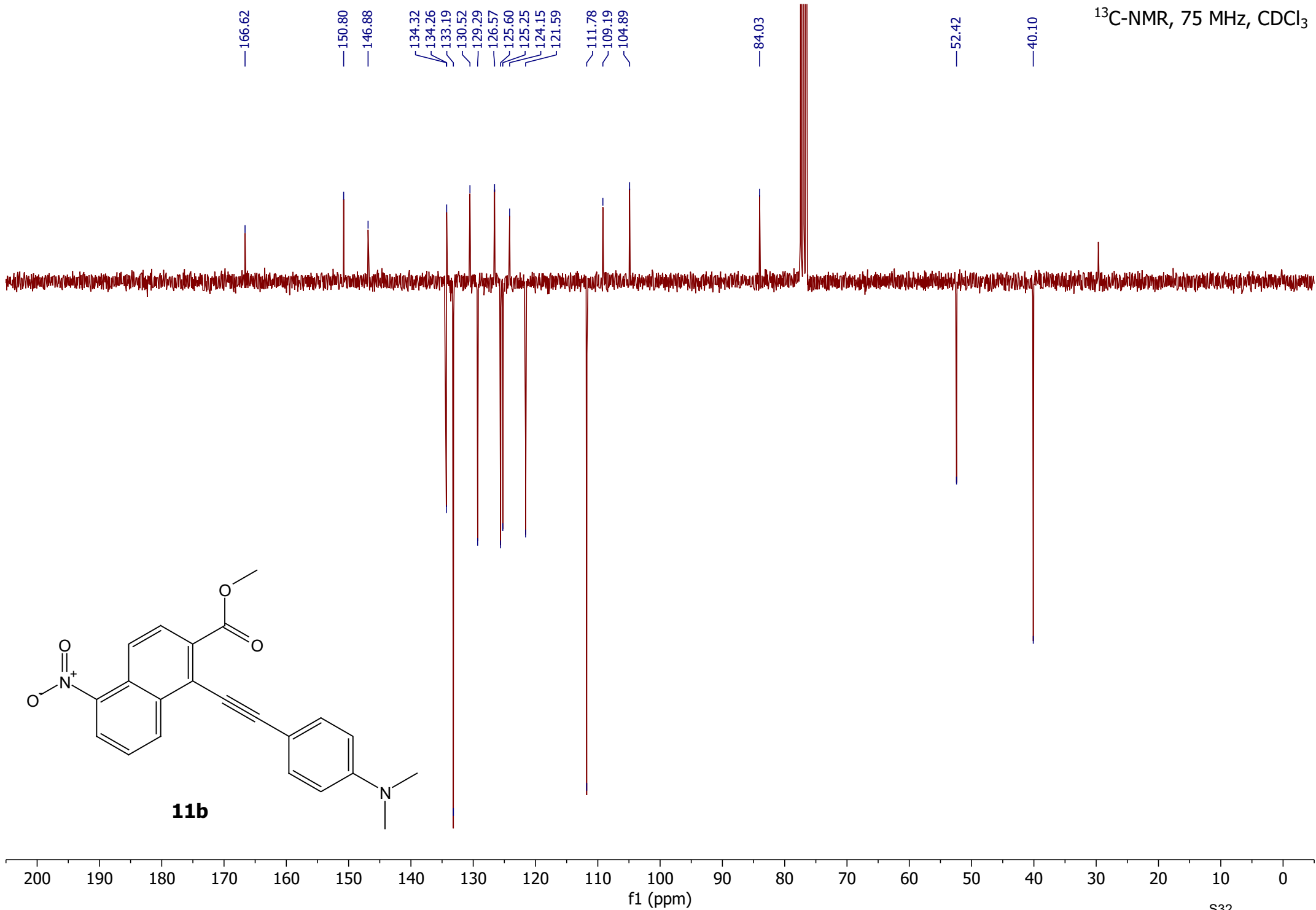


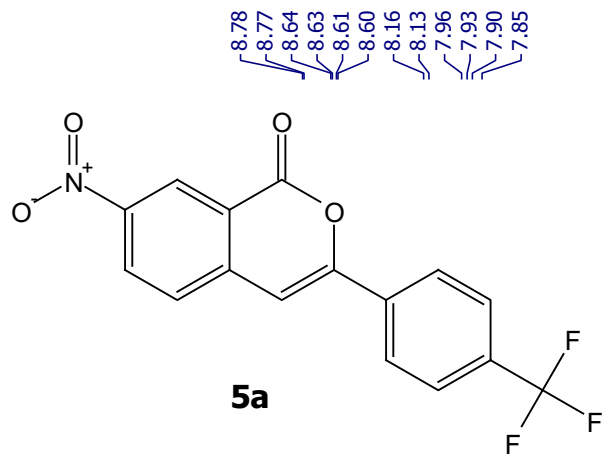
¹H-NMR, 300 MHz, CDCl₃



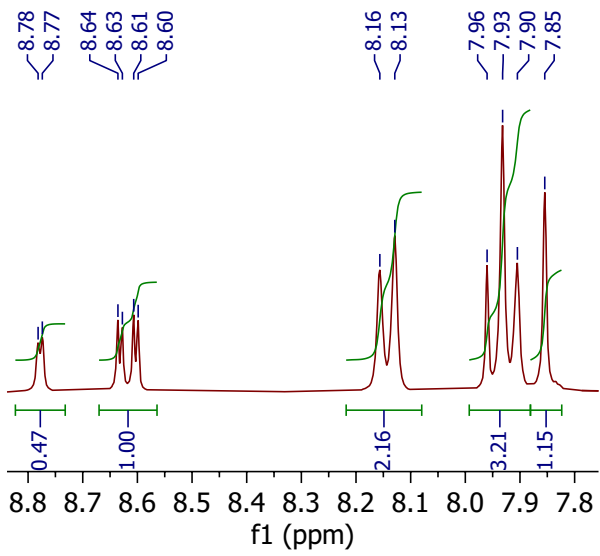




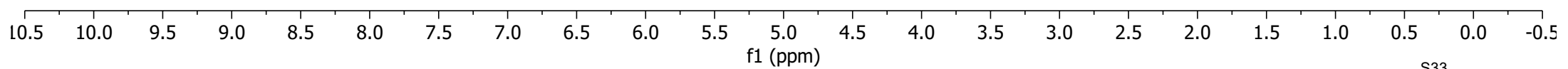


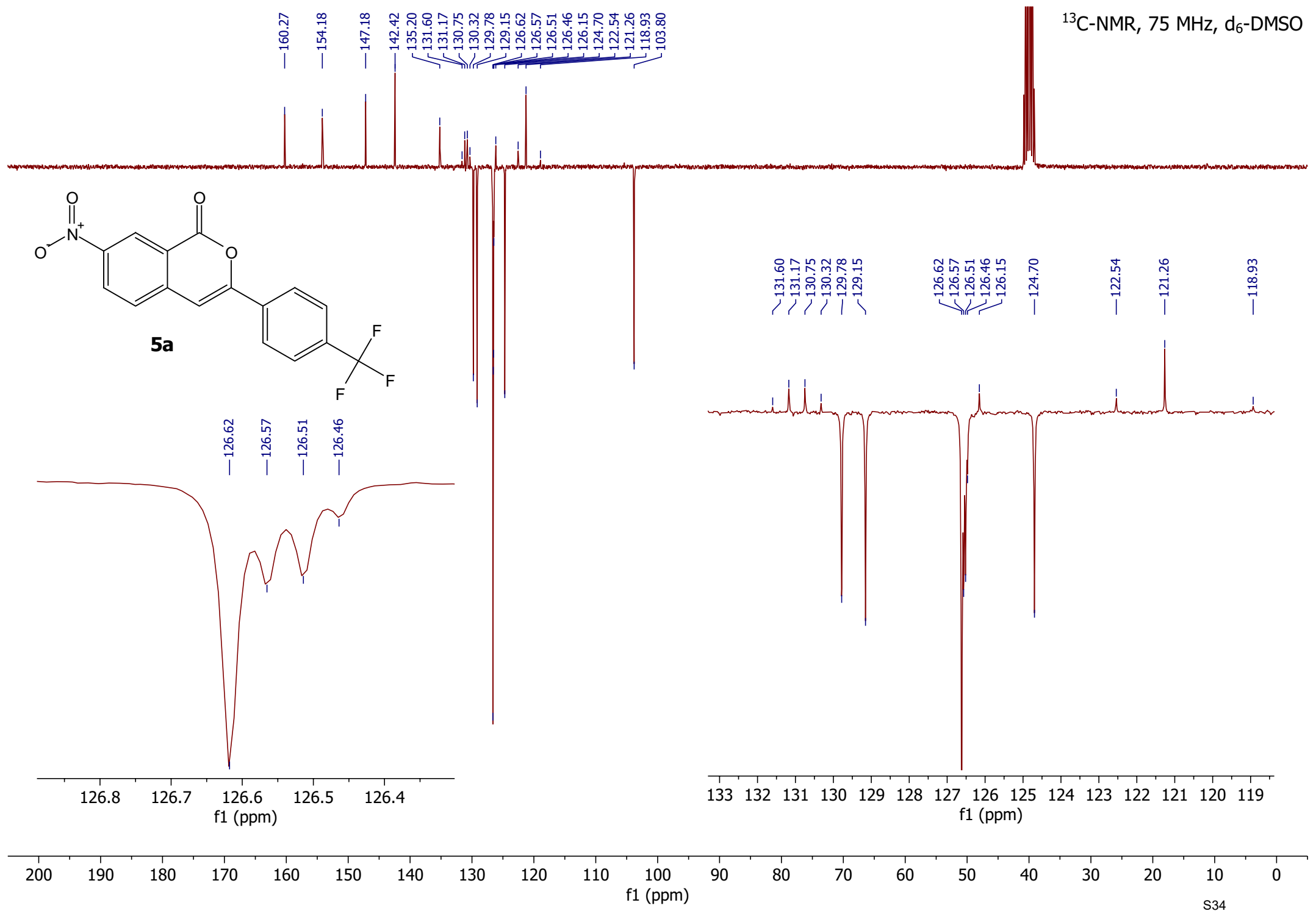
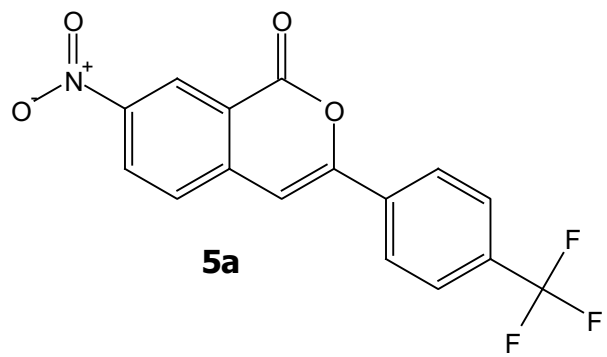


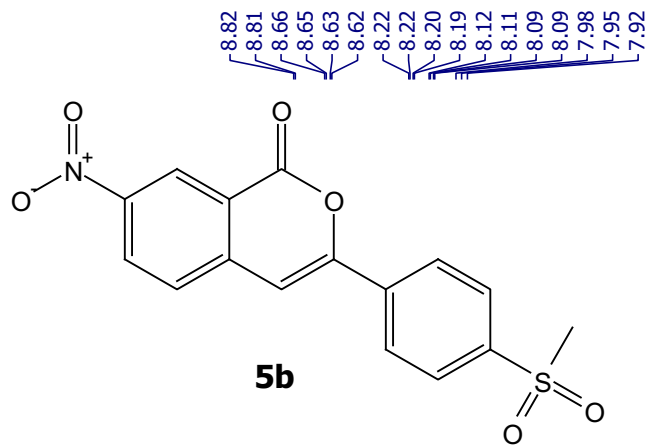
8.78
8.77
8.64
8.63
8.61
8.60
8.16
8.13
7.96
7.93
7.90
7.85



0.47
1.00
2.16
3.21
1.15

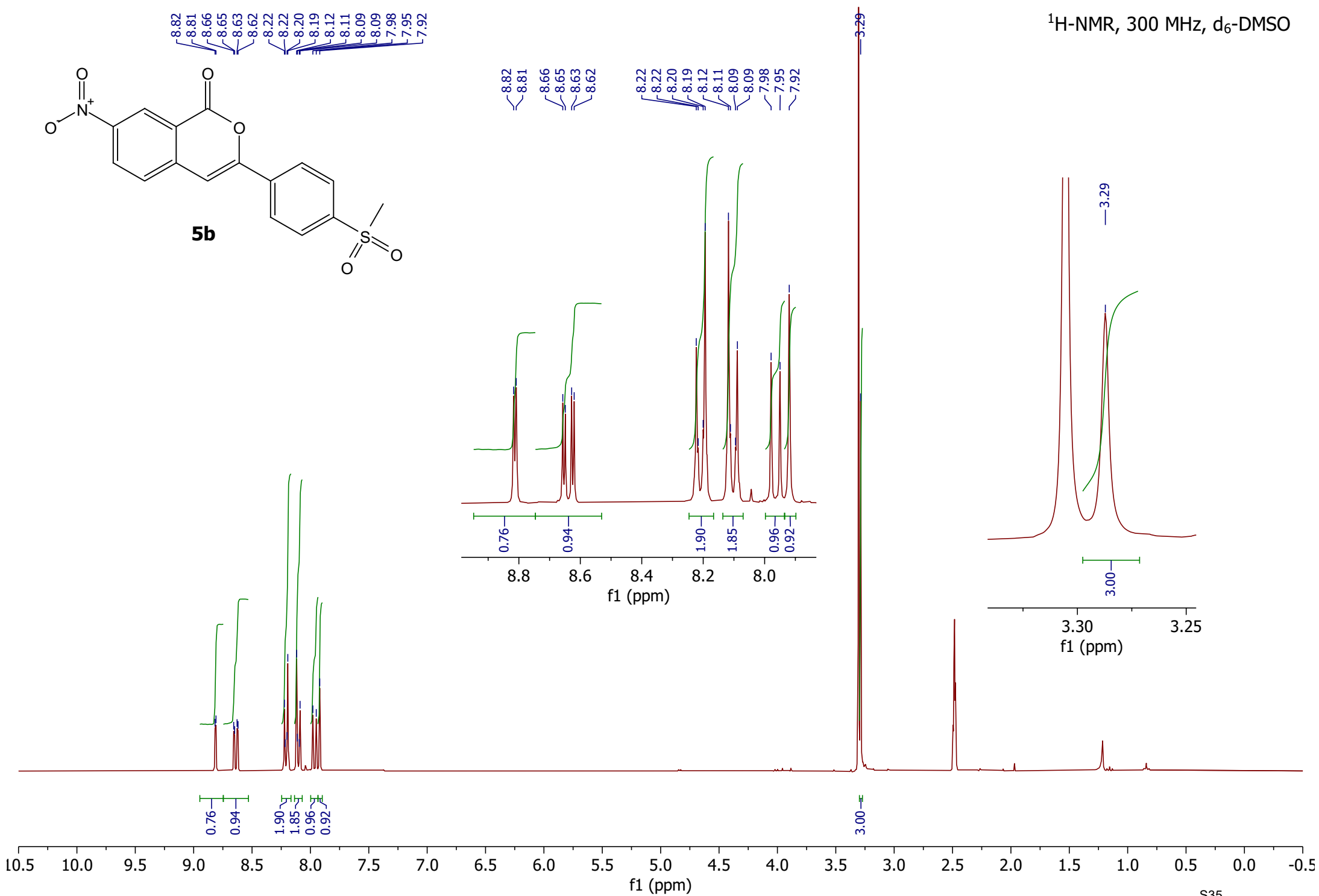


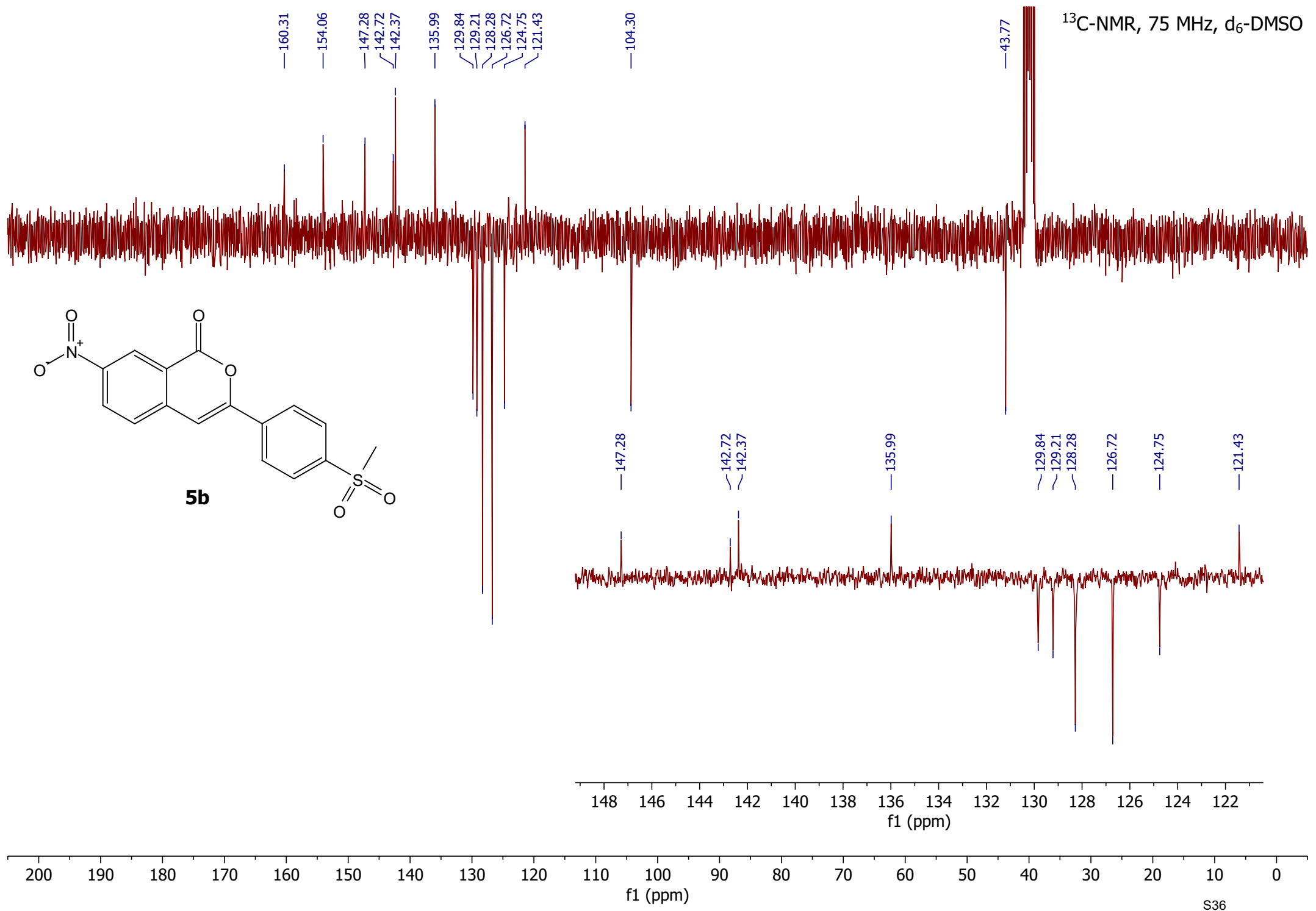


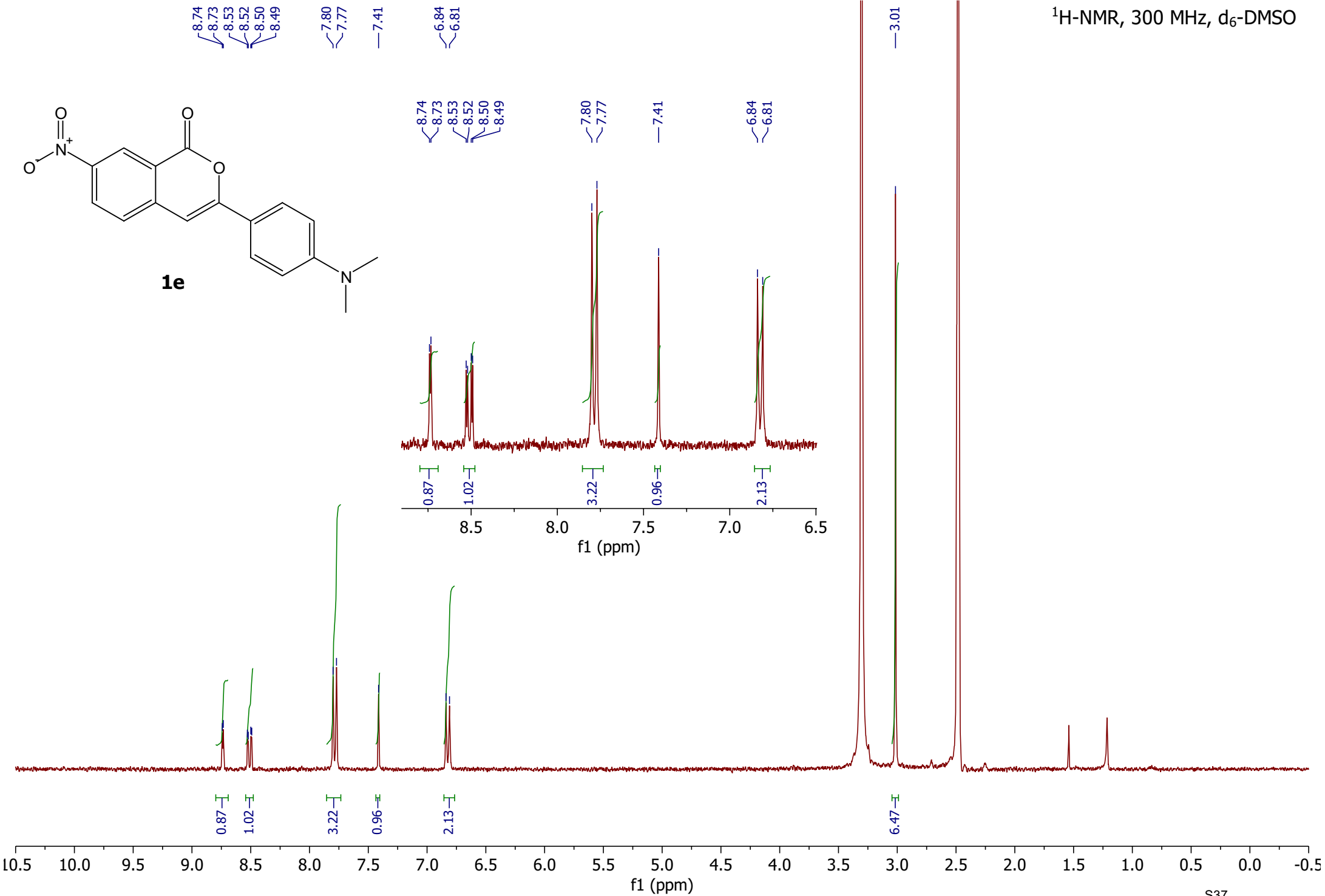
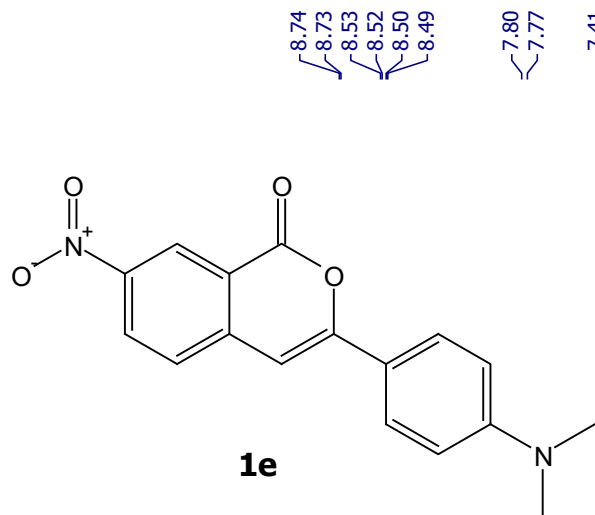


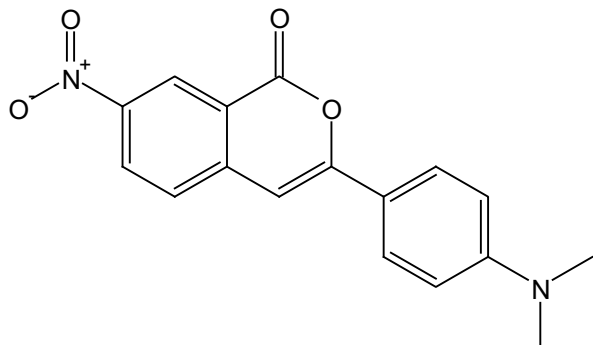
8.82
8.81
8.66
8.65
8.63
8.62
8.22
8.22
8.20
8.19
8.12
8.11
8.09
8.09
7.98
7.95
7.92

8.82
8.81
8.66
8.65
8.63
8.62
8.22
8.22
8.20
8.19
8.12
8.11
8.09
8.09
7.98
7.95
7.92



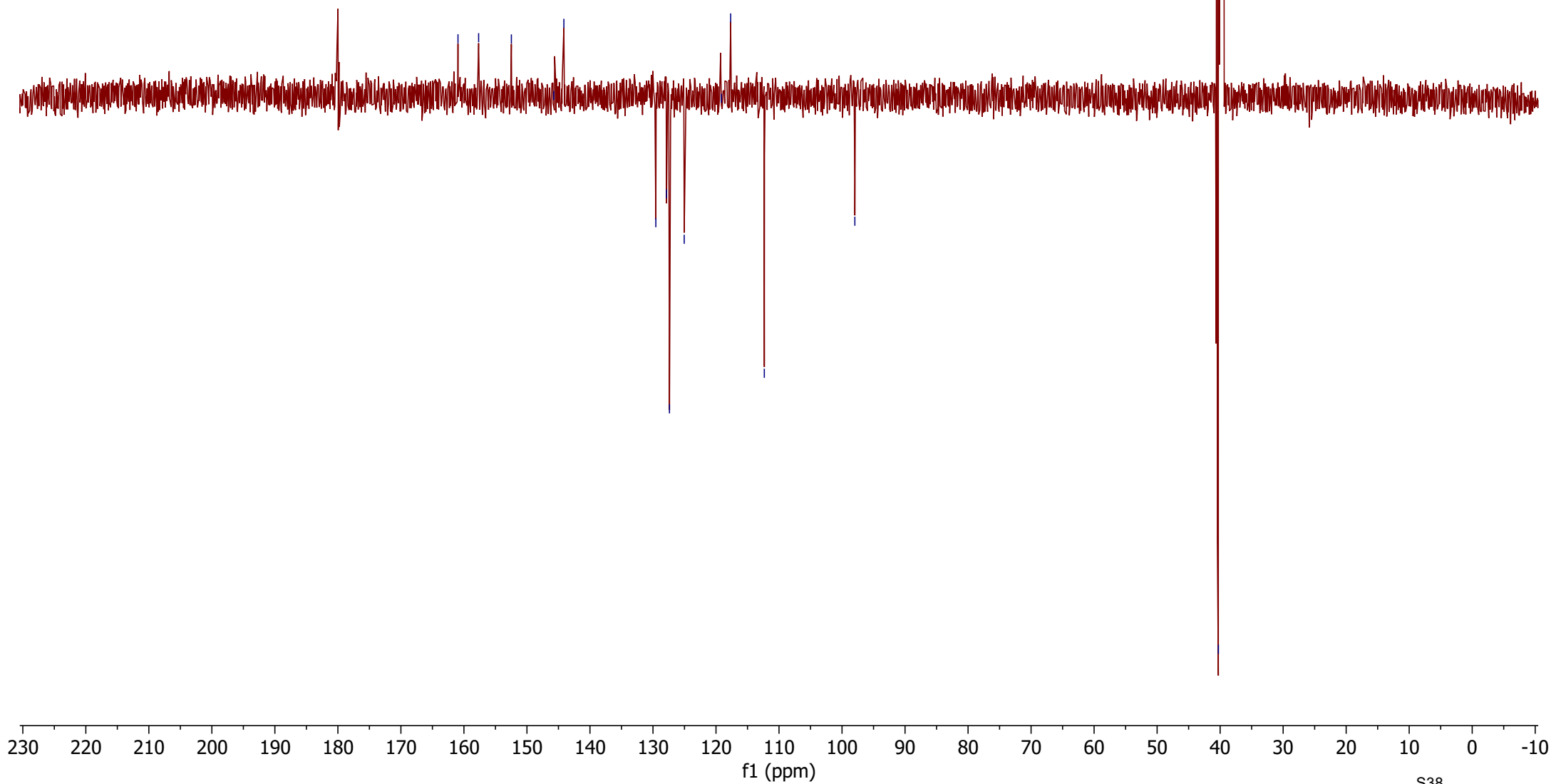


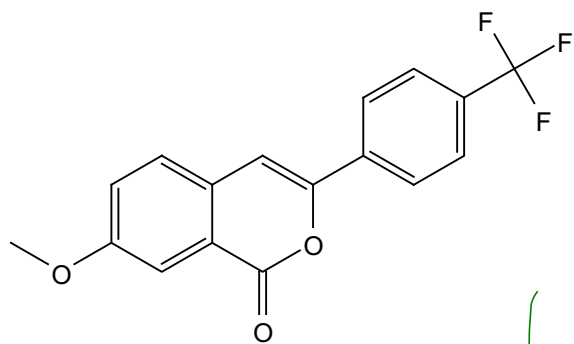




1e

$^{13}\text{C-NMR}$, 75 MHz, $\text{d}_6\text{-DMSO}$





1f

7.97
7.95
7.73
7.73
7.72
7.69
7.49
7.46
7.36
7.35
7.33
7.32
7.01

3.94

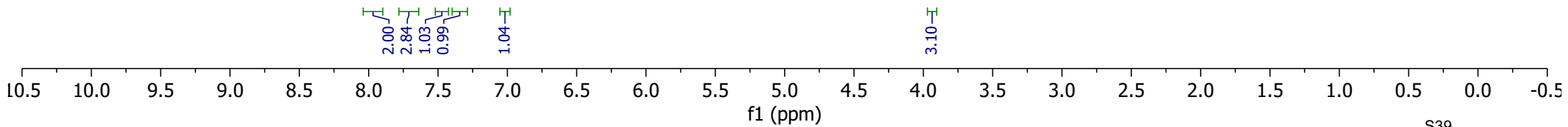
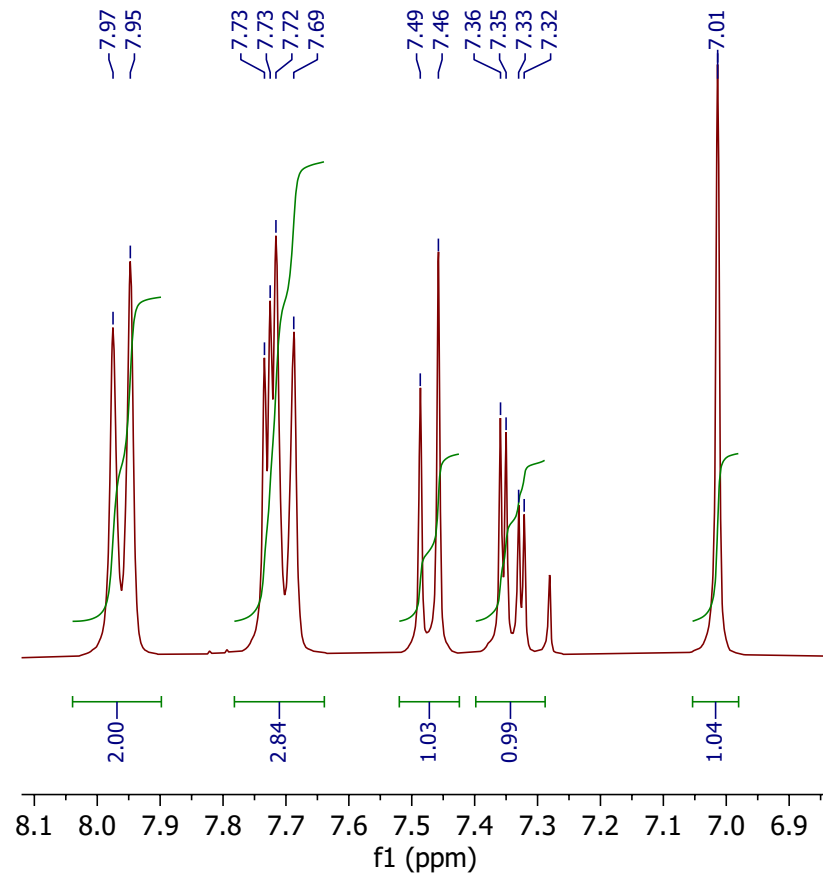
7.97
7.95

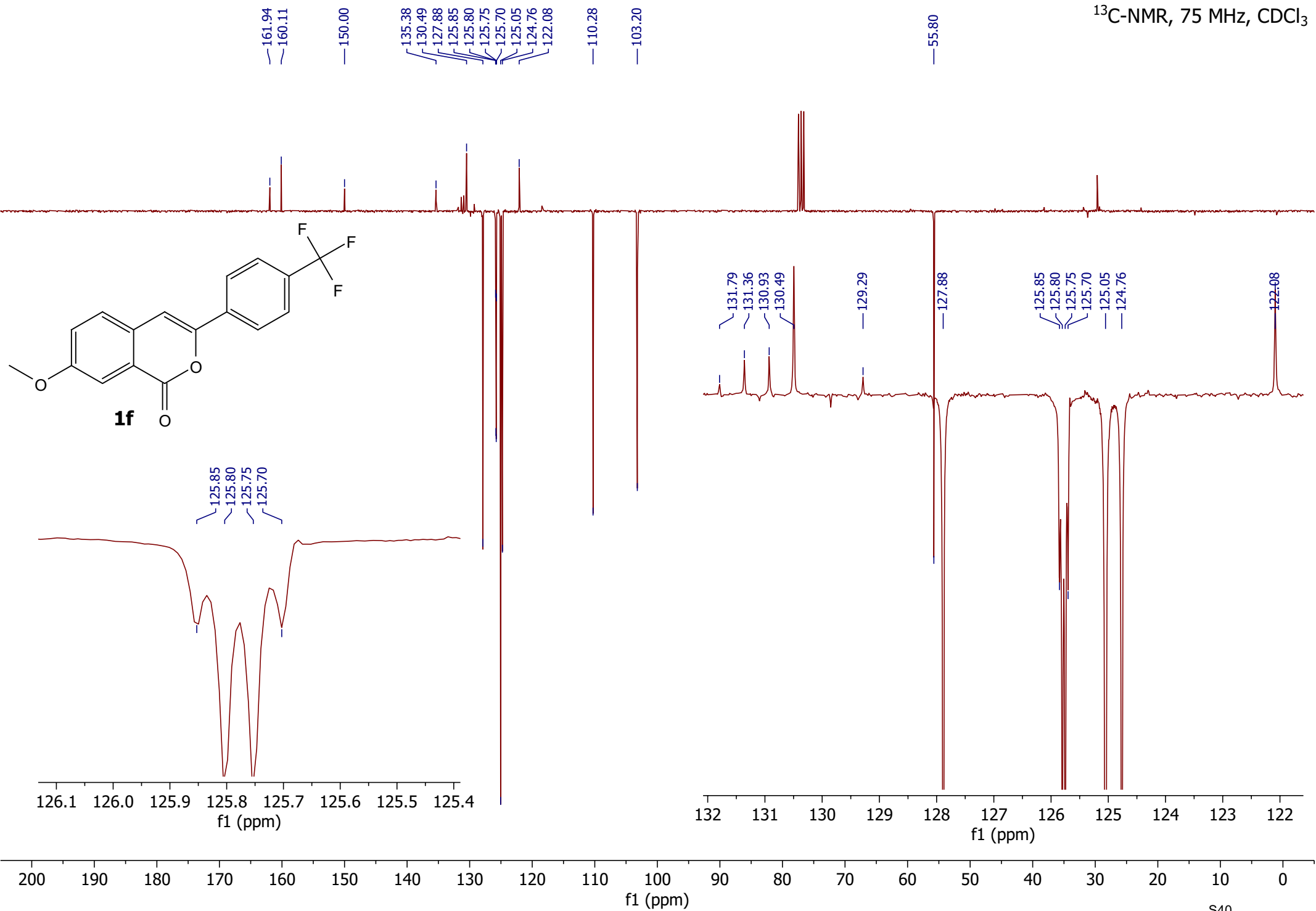
7.73
7.73
7.72
7.69

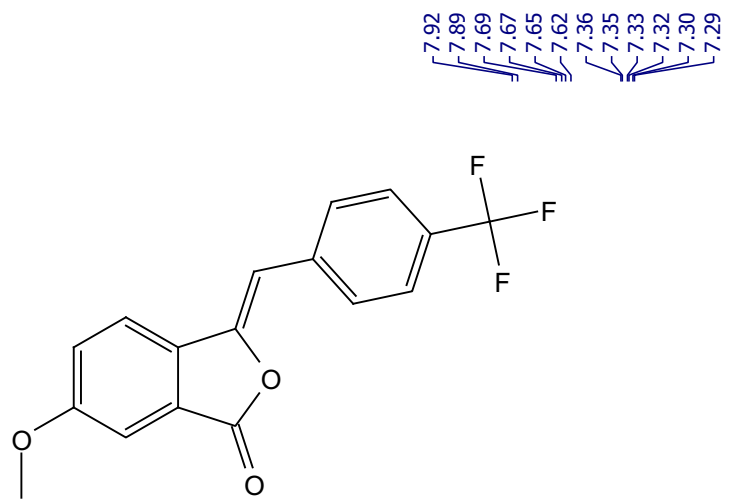
7.49
7.46

7.36
7.35
7.33
7.32

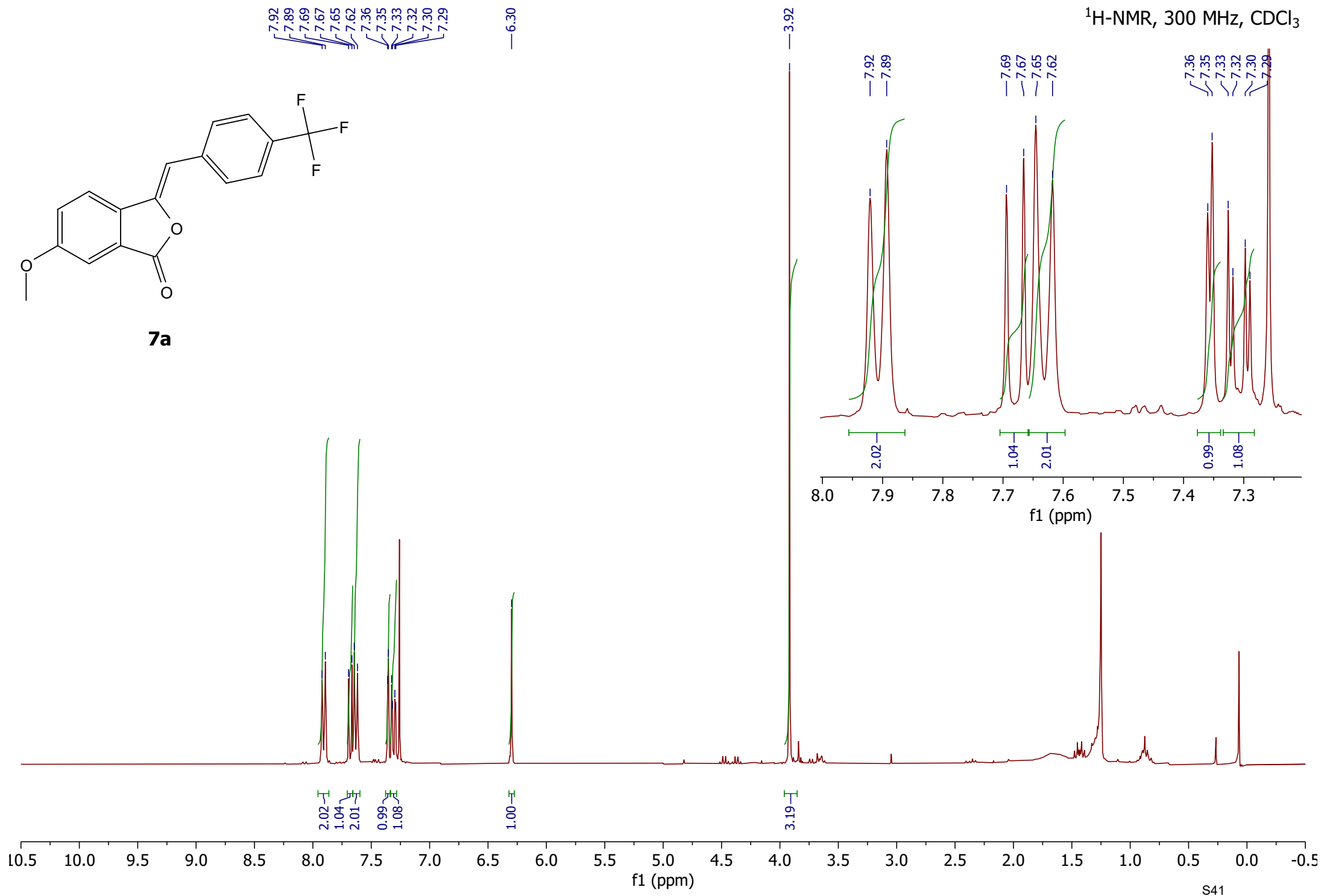
7.01

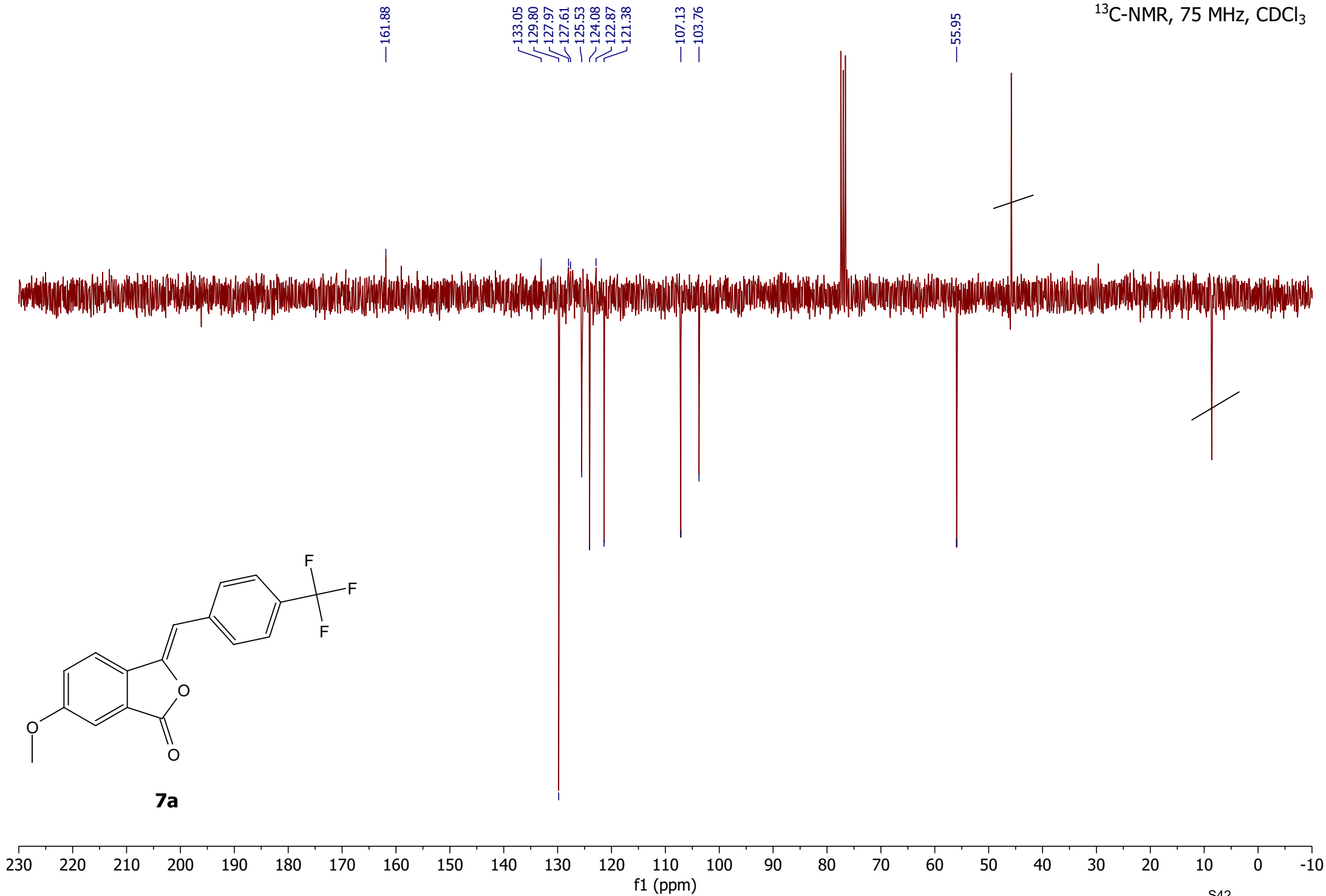


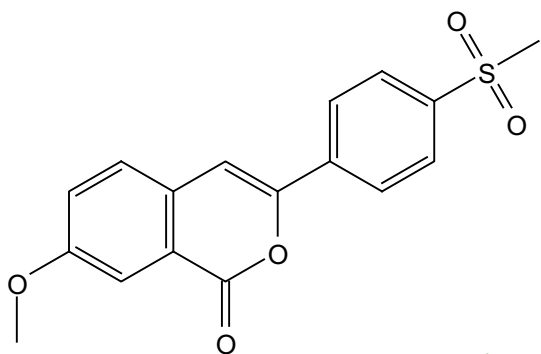




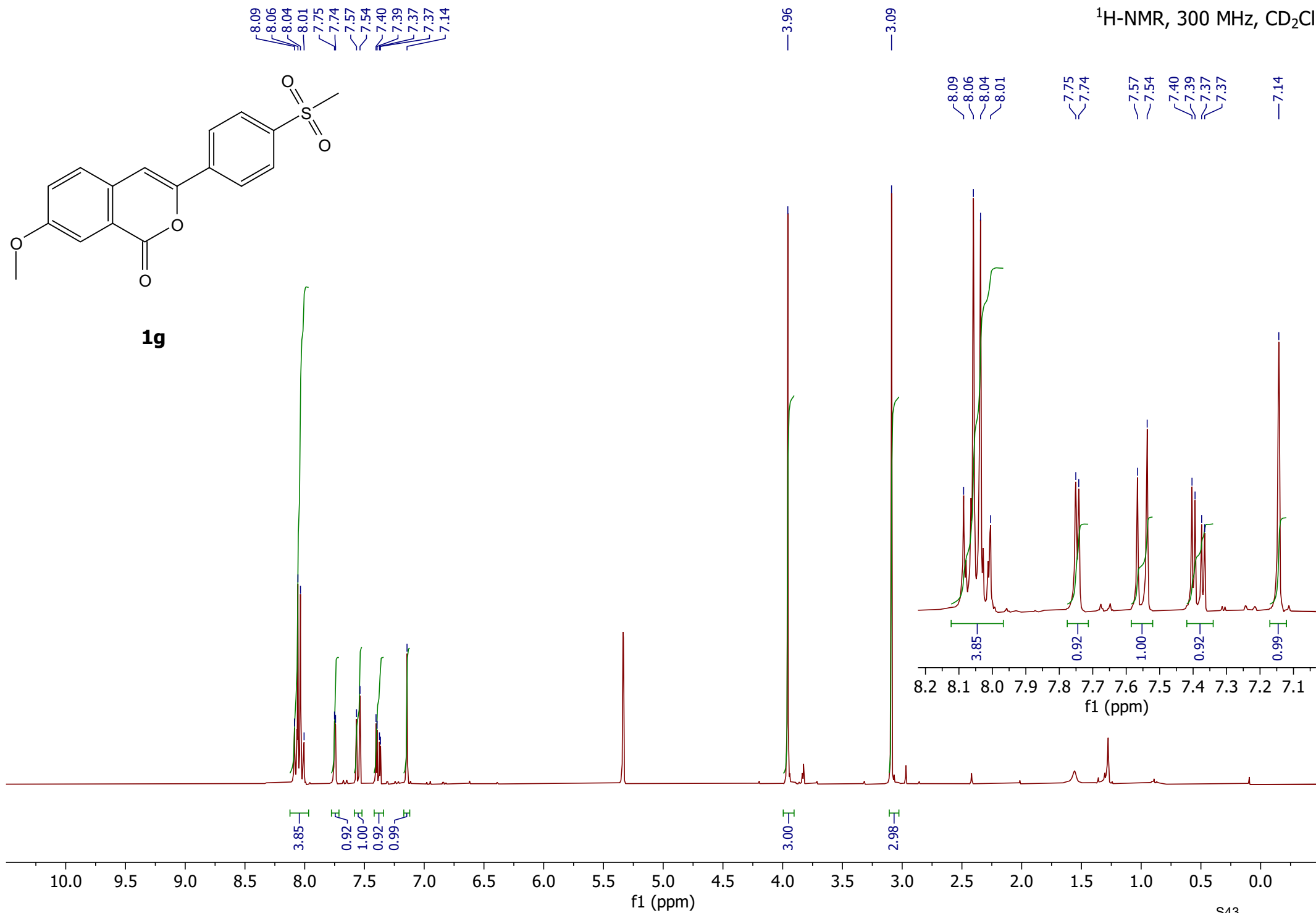
7a

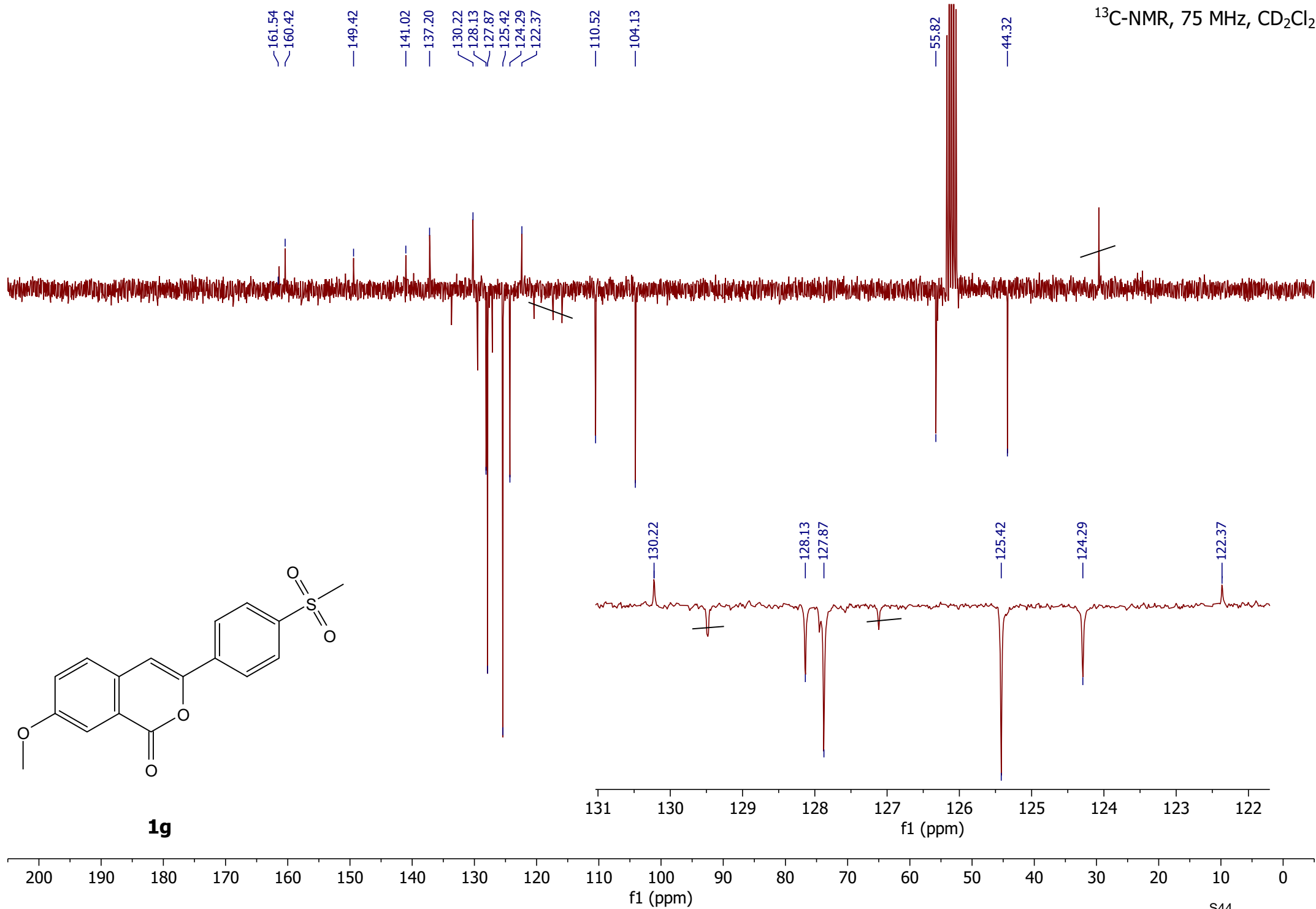


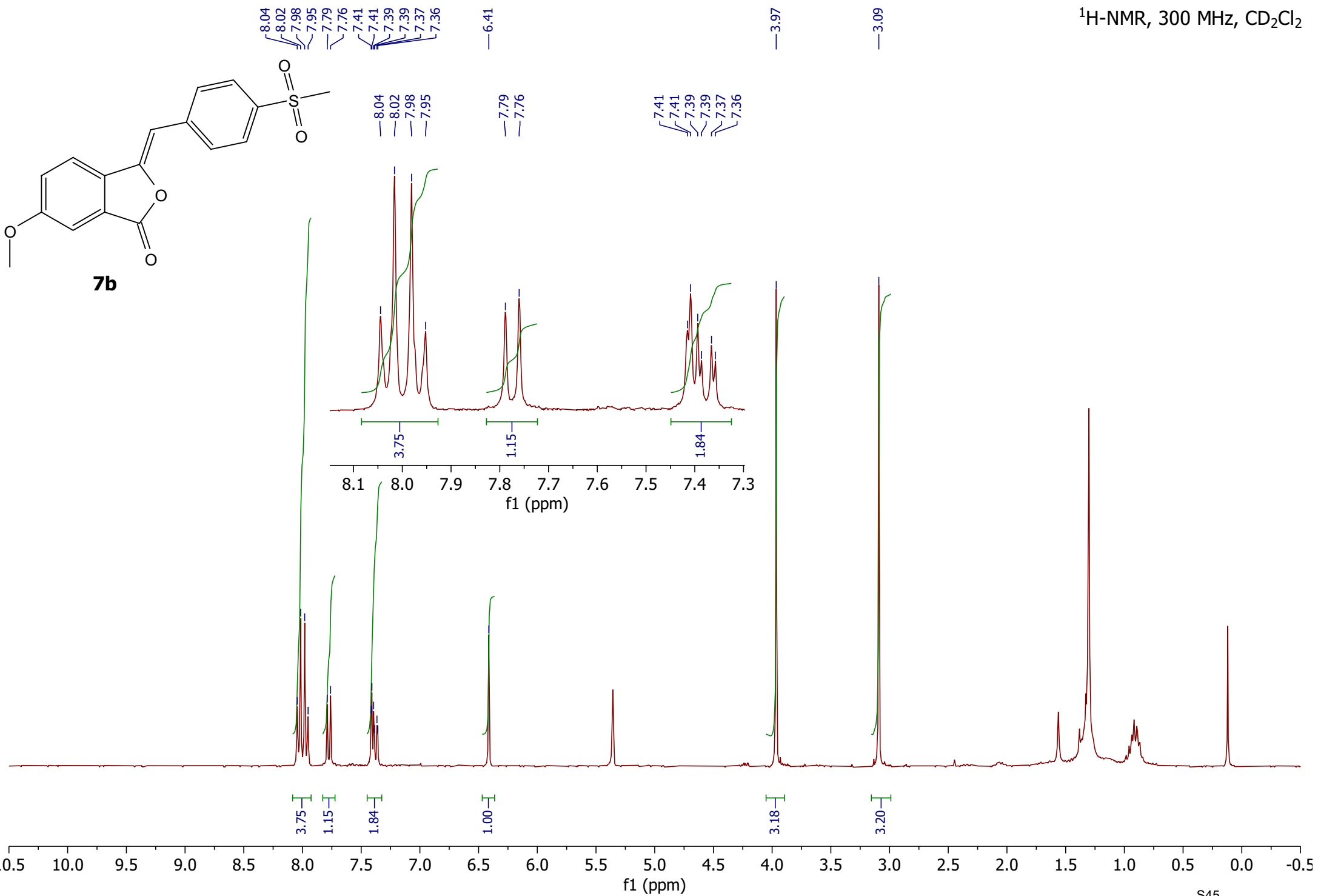


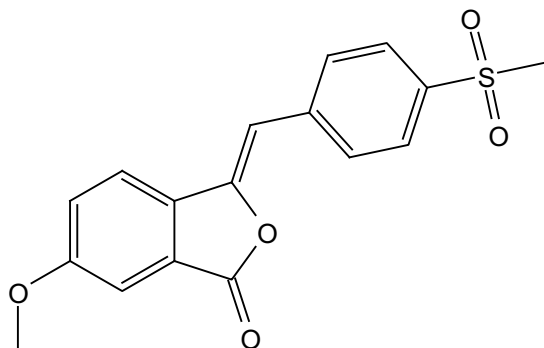


1g

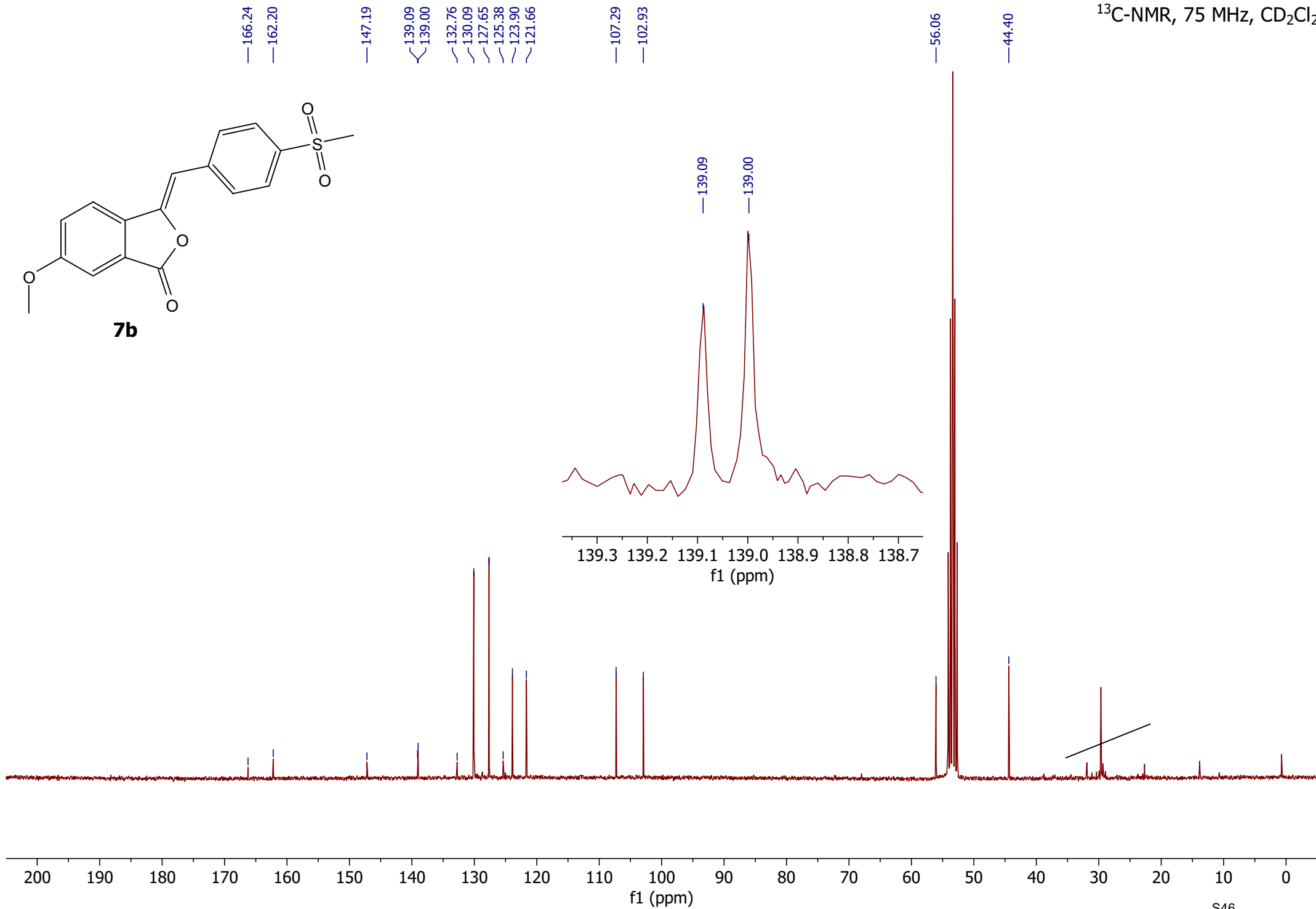


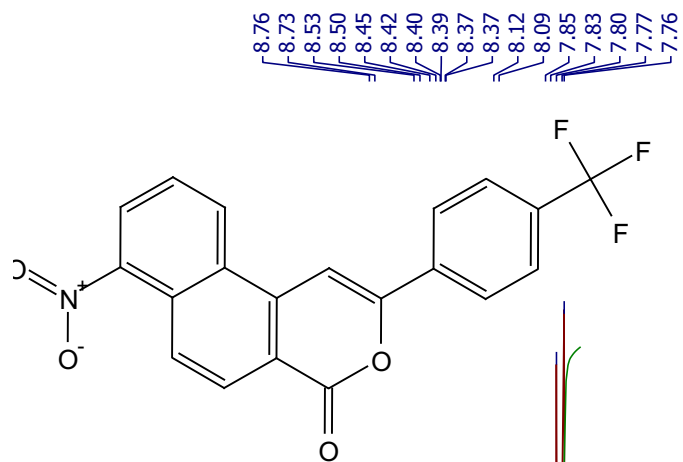






7b





11a

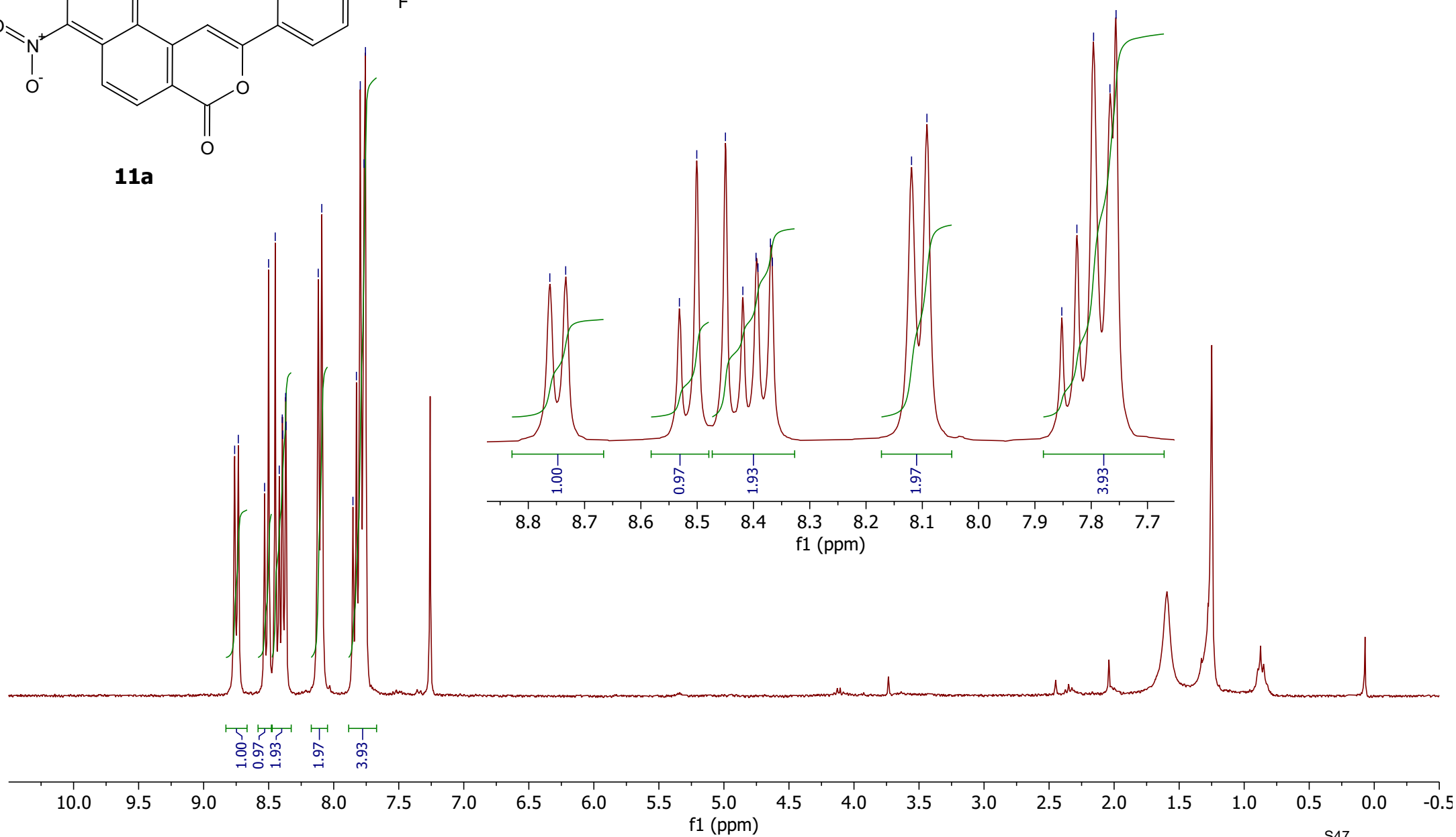
8.76
8.73
8.53
8.50
8.45
8.42
8.40
8.39
8.37
8.37
8.12
8.09
7.85
7.83
7.80
7.77
7.76

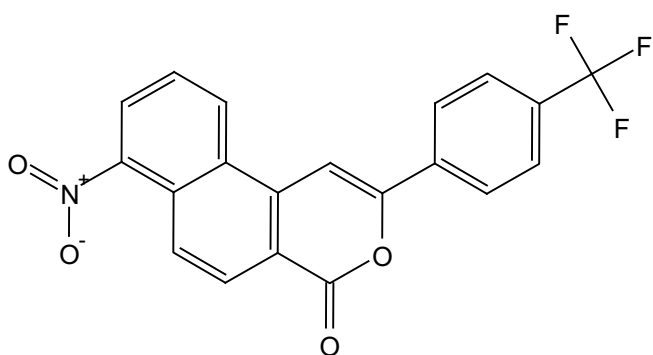
8.76
8.73

8.53
8.50
8.45
8.42
8.40
8.39
8.37
8.37

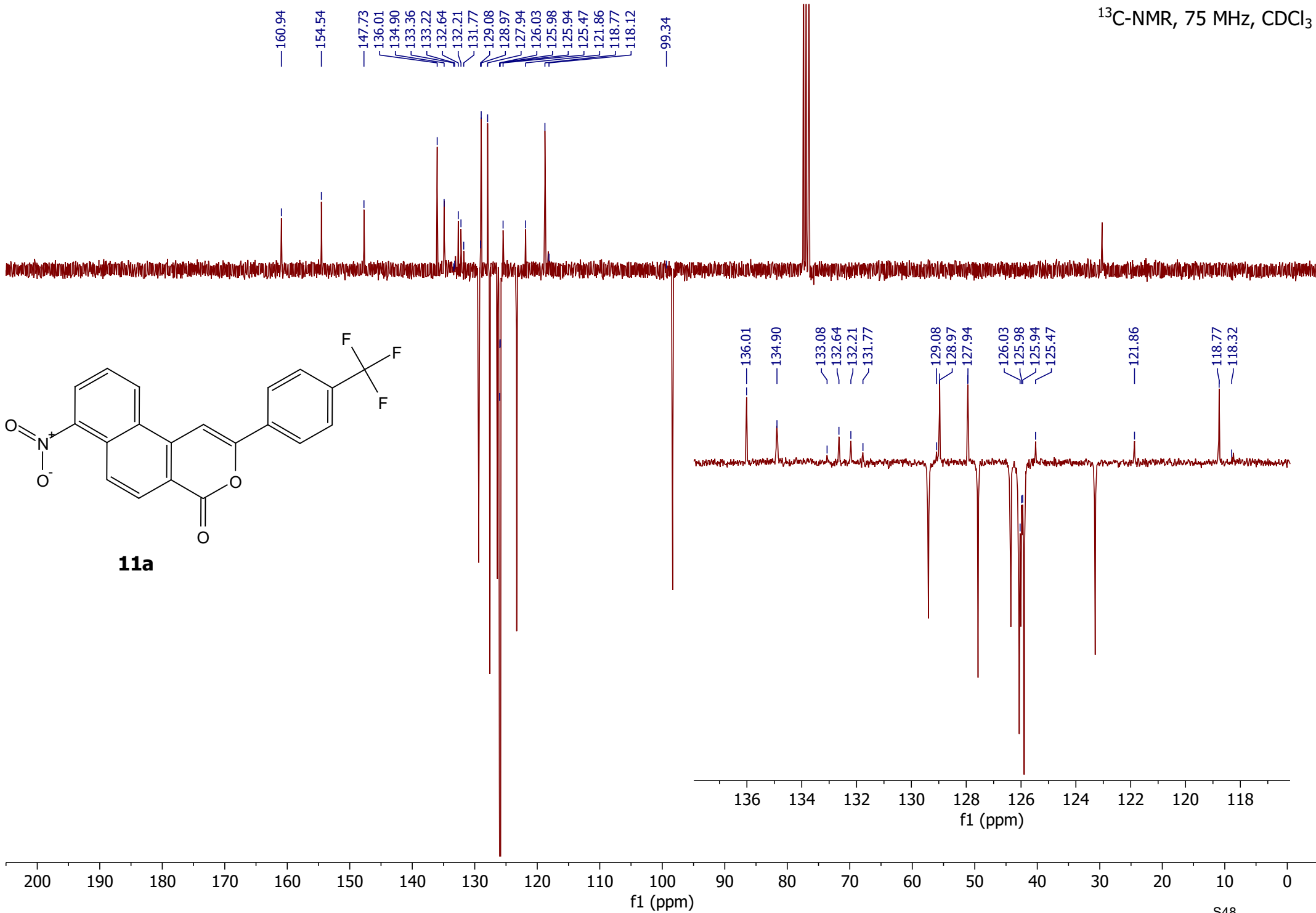
8.12
8.09

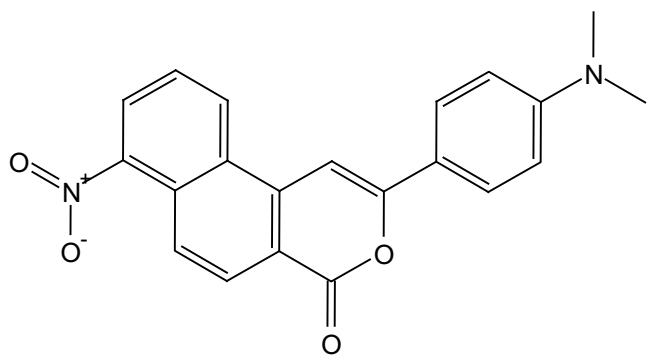
7.85
7.83
7.80
7.77
7.76





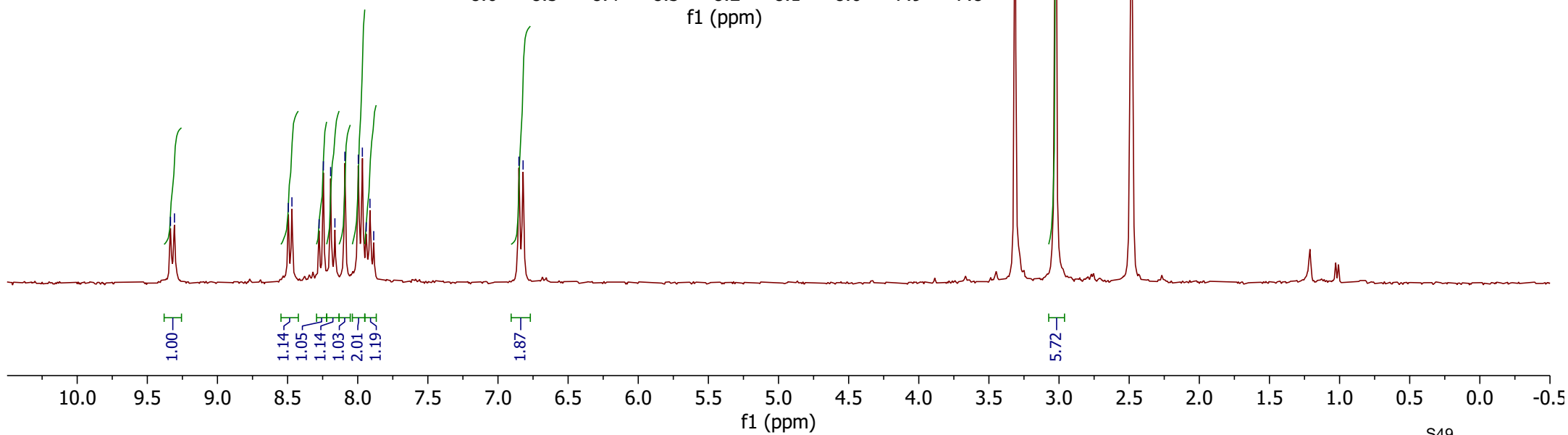
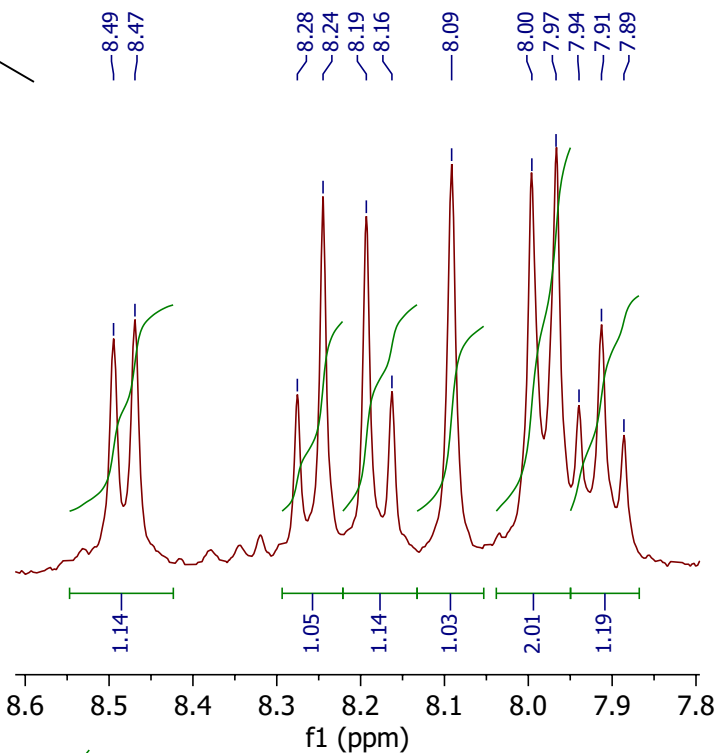
11a

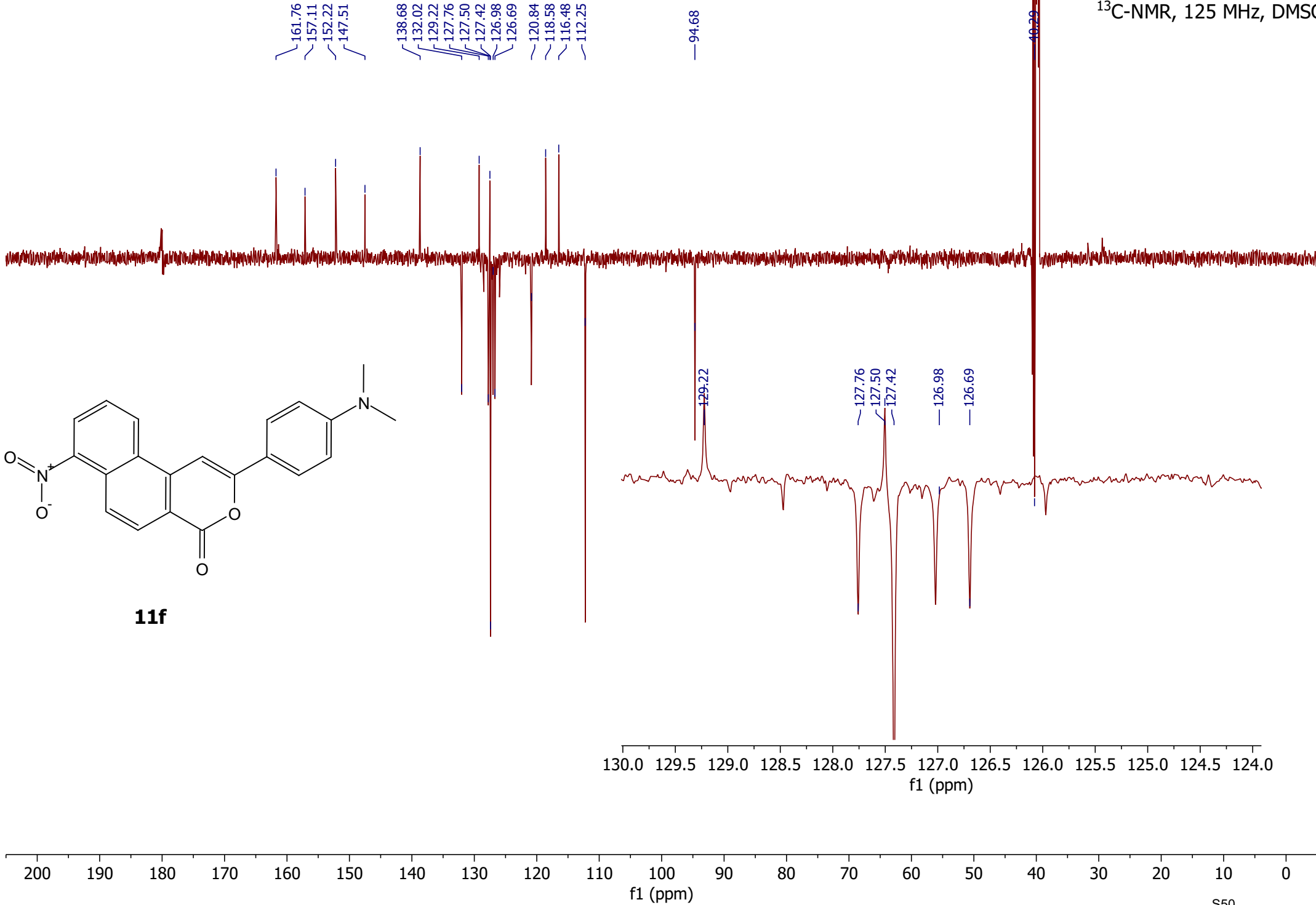


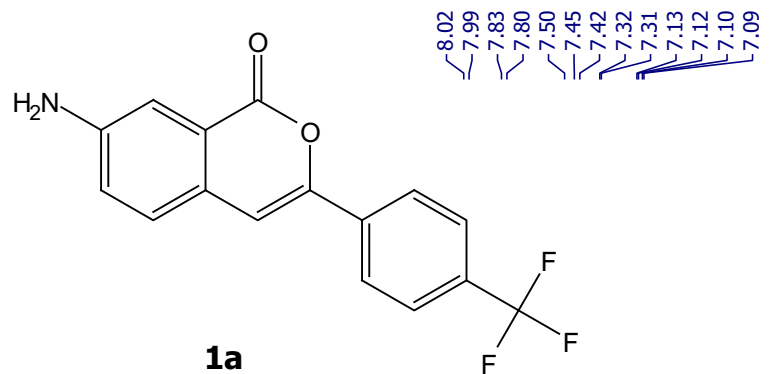


11f

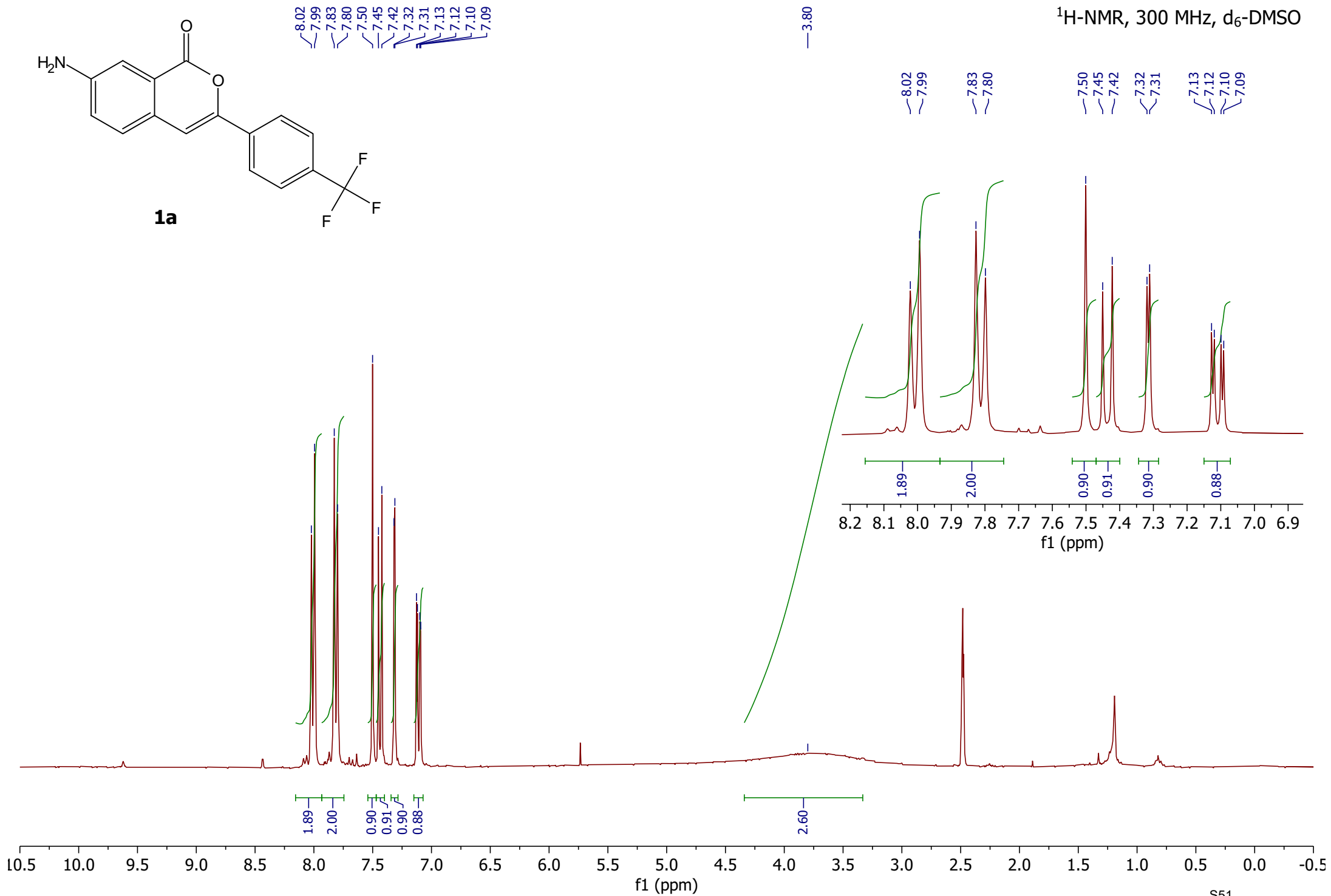
9.33
9.31
8.49
8.47
8.28
8.24
8.19
8.16
8.09
8.00
7.97
7.94
7.91
7.89
6.85
6.82



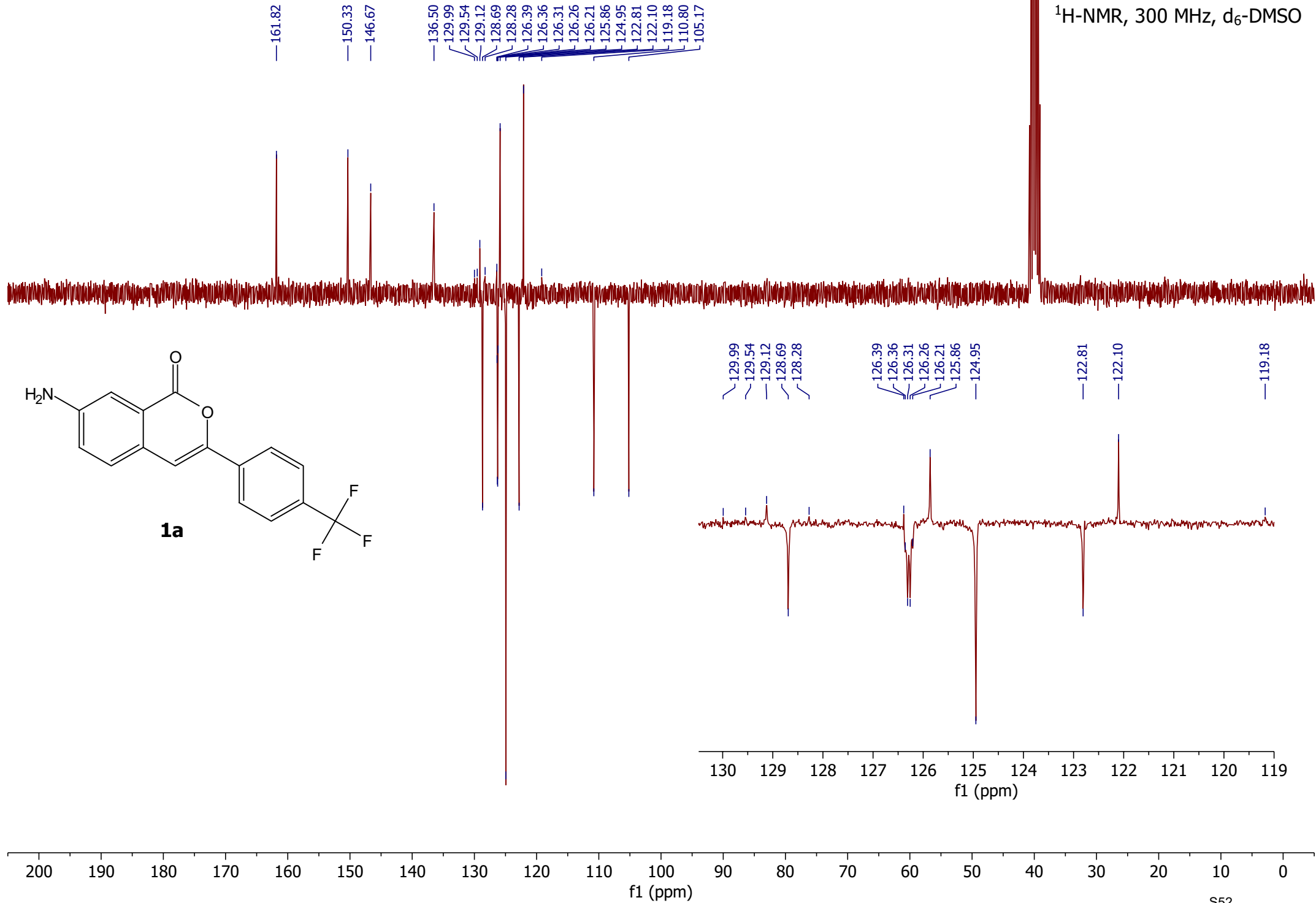


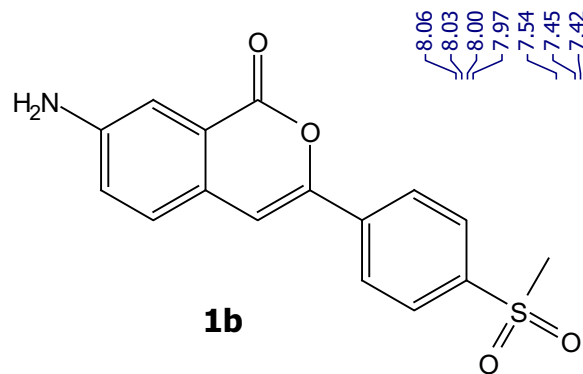


$^1\text{H-NMR}$, 300 MHz, $\text{d}_6\text{-DMSO}$



¹H-NMR, 300 MHz, d₆-DMSO



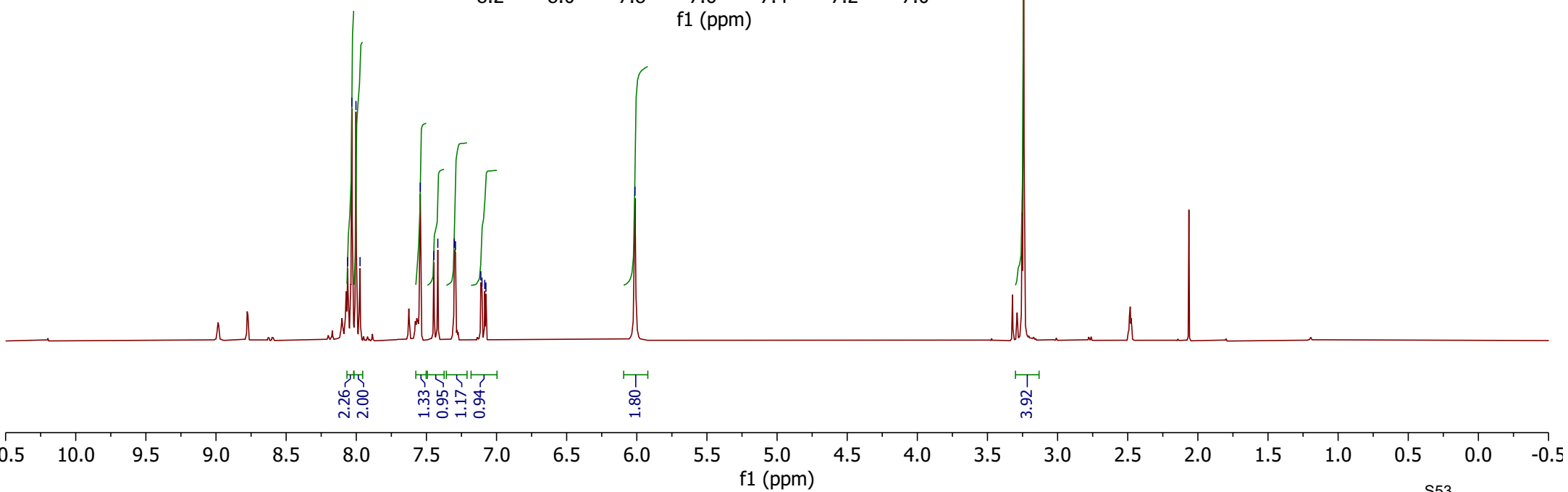
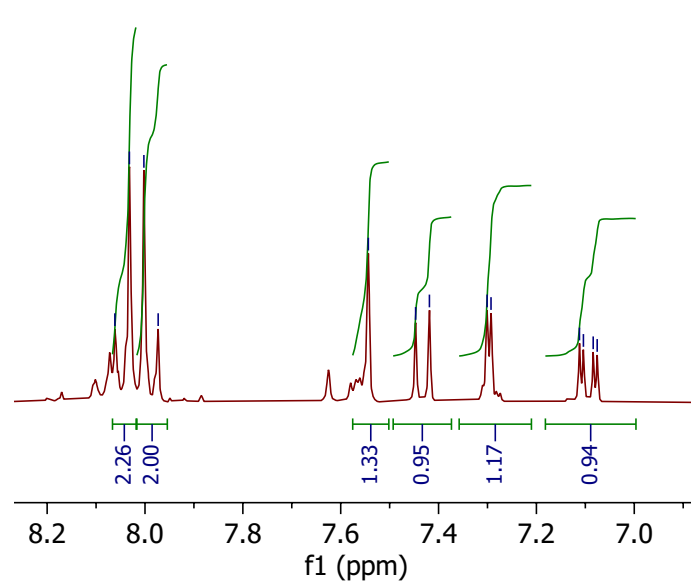


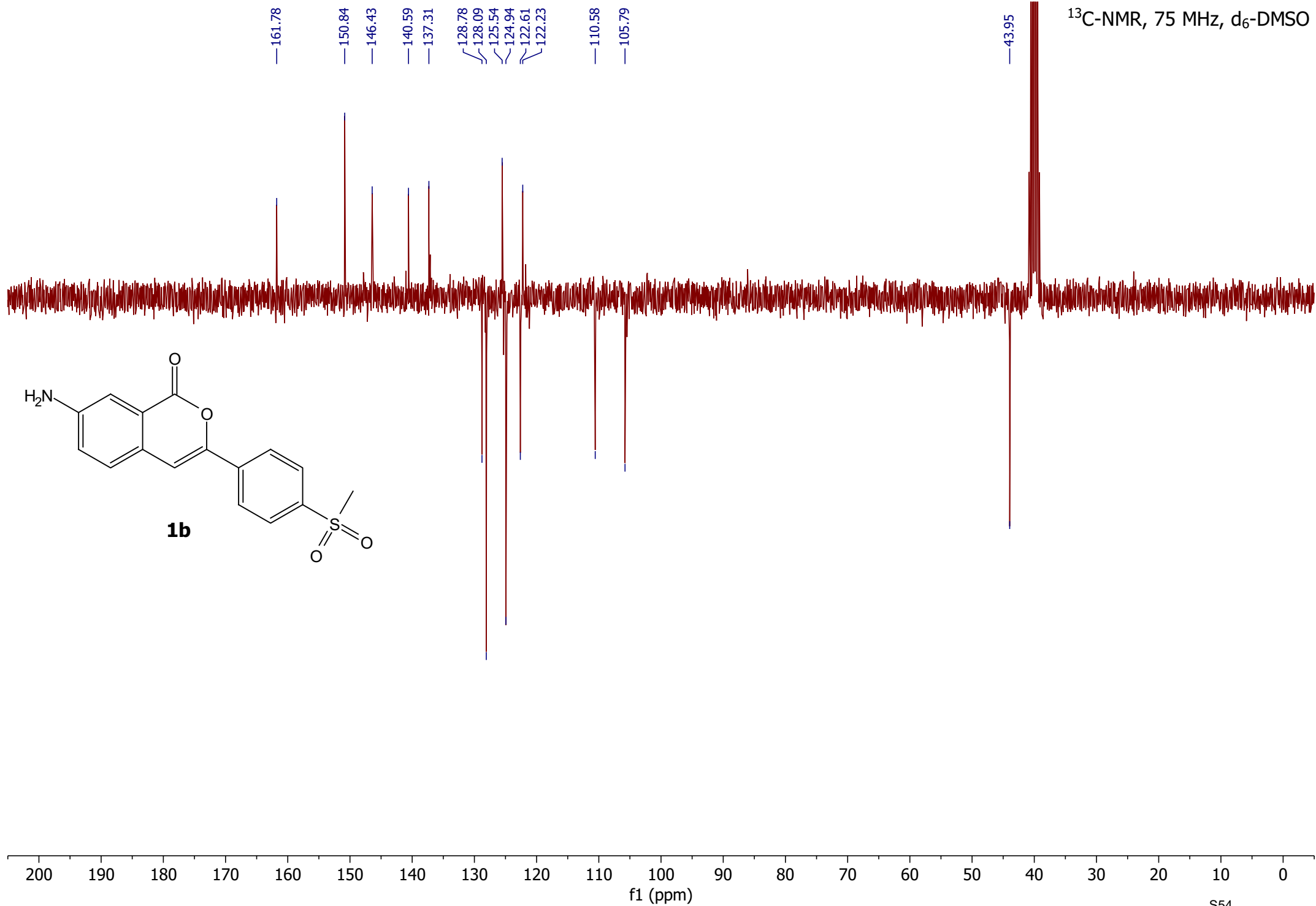
8.06
8.03
8.00
7.97
7.54
7.45
7.42
7.30
7.29
7.11
7.10
7.08

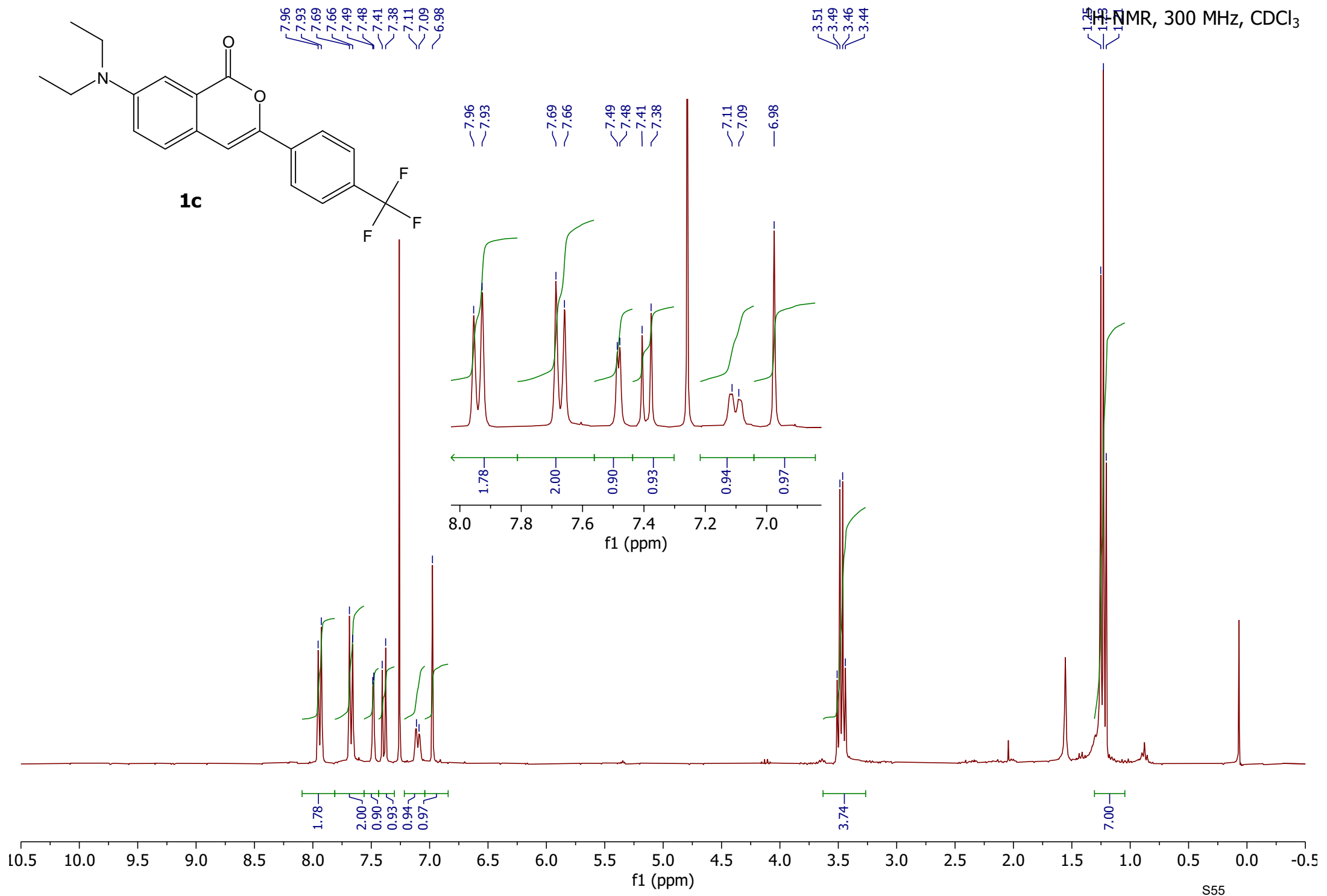
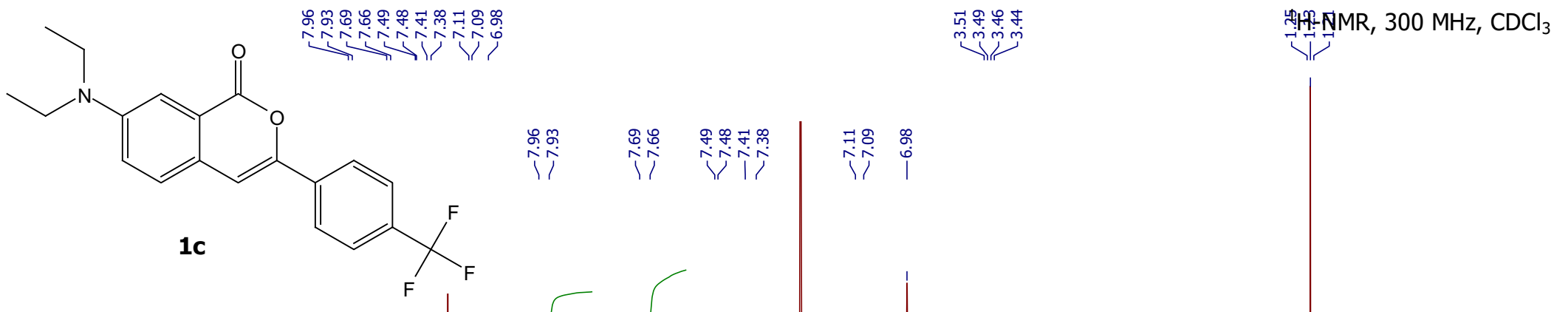
6.01

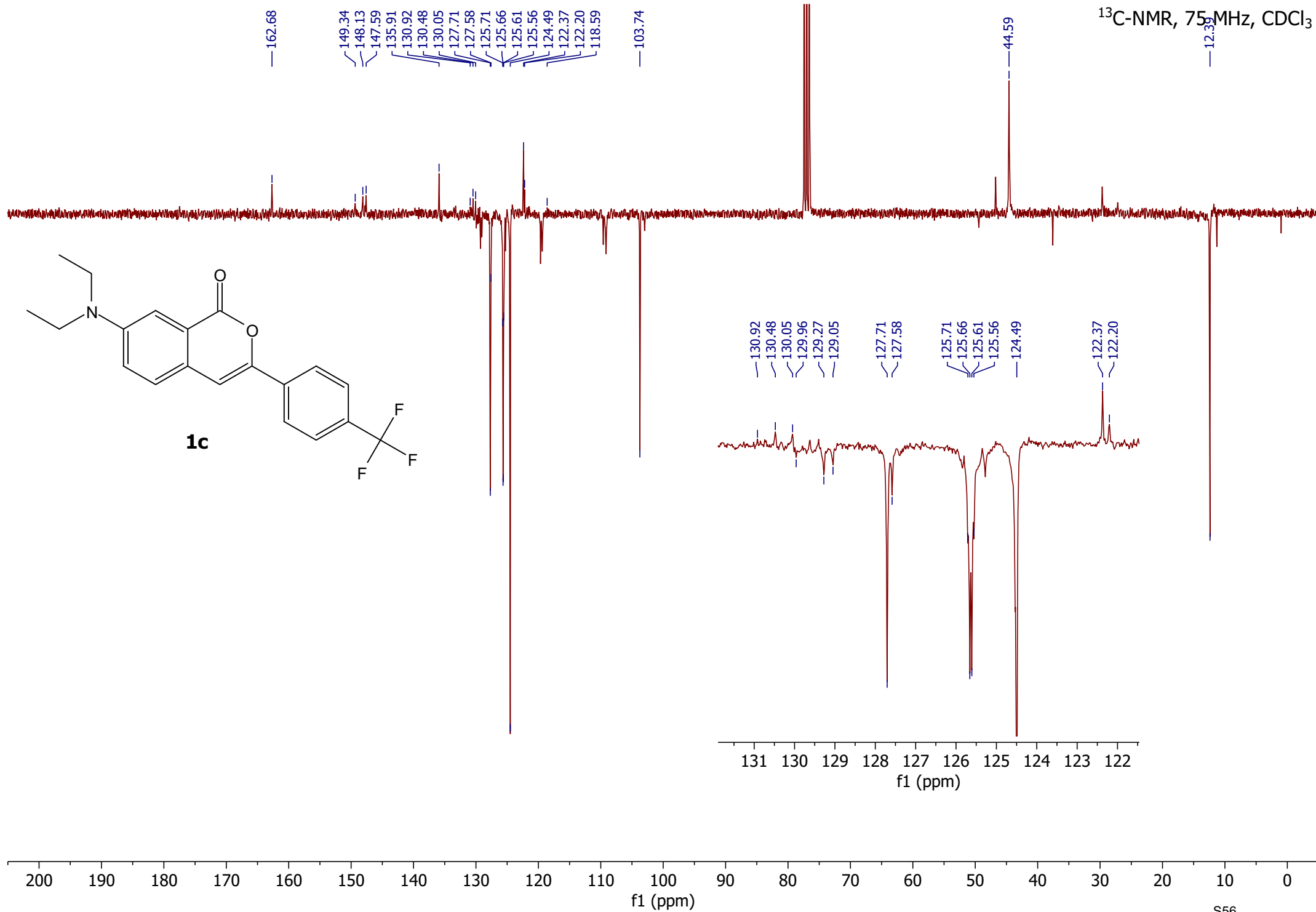
3.24

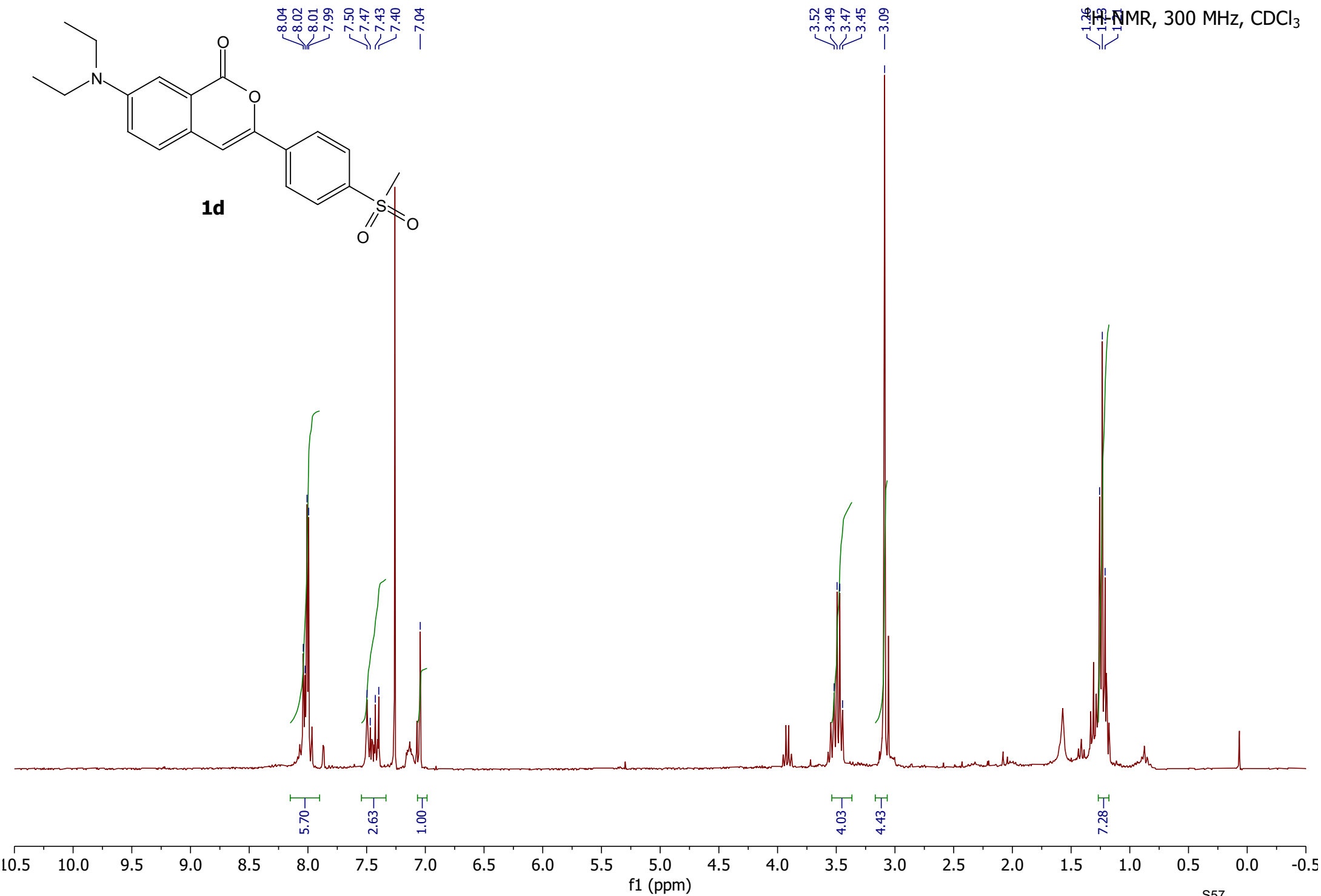
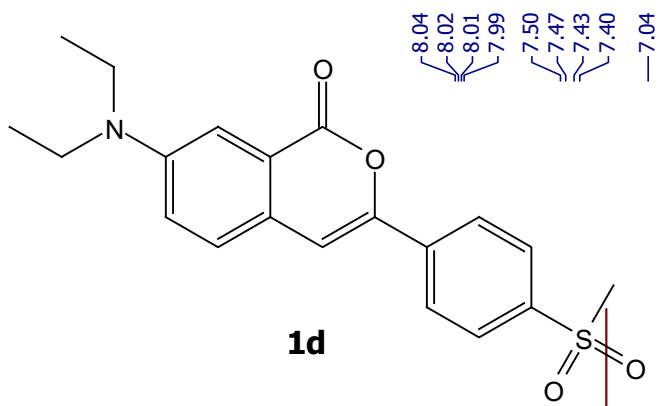
8.06
8.03
8.00
7.97
7.54
7.45
7.42
7.30
7.29
7.11
7.10
7.08

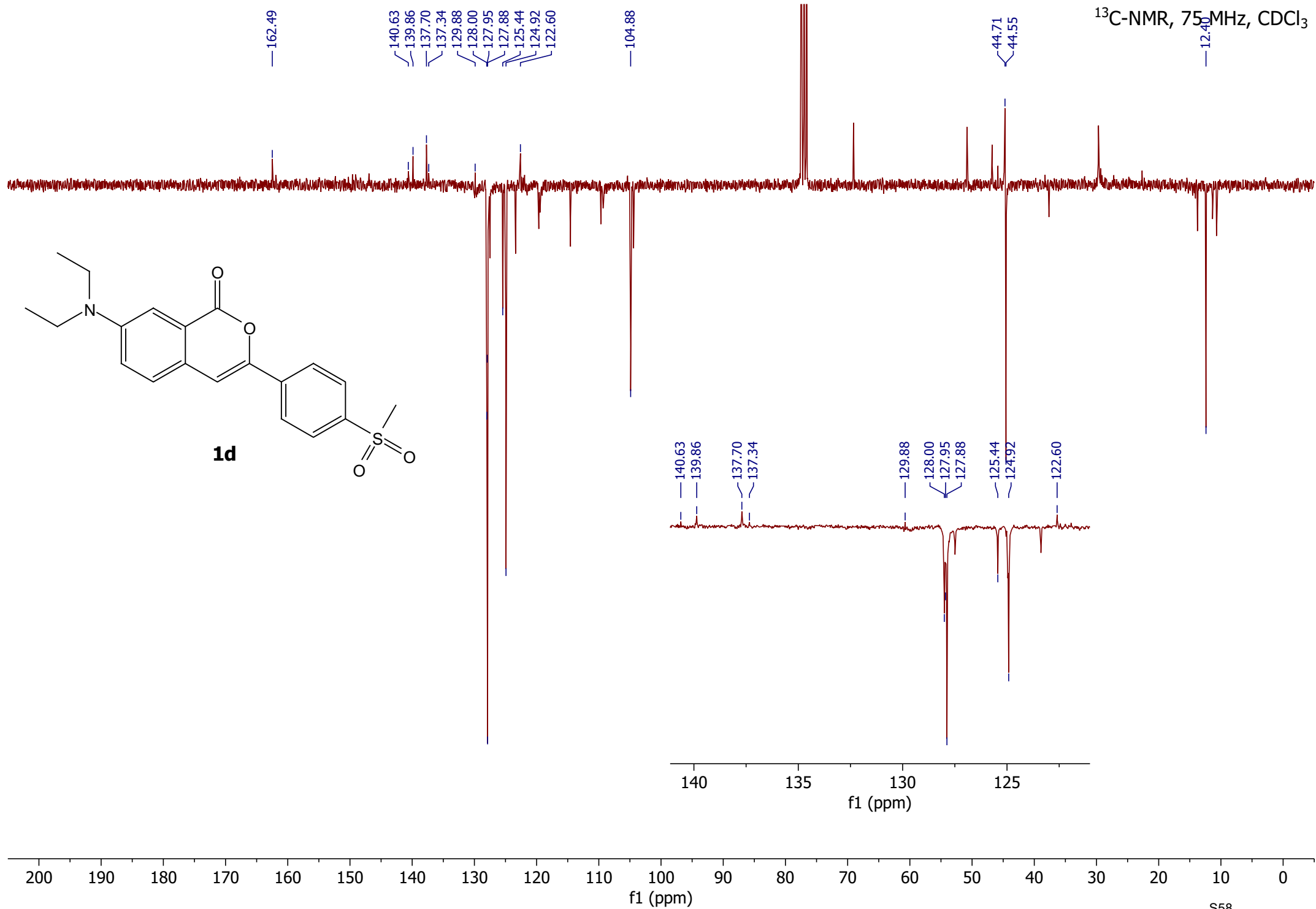


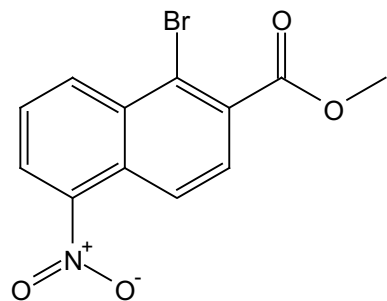




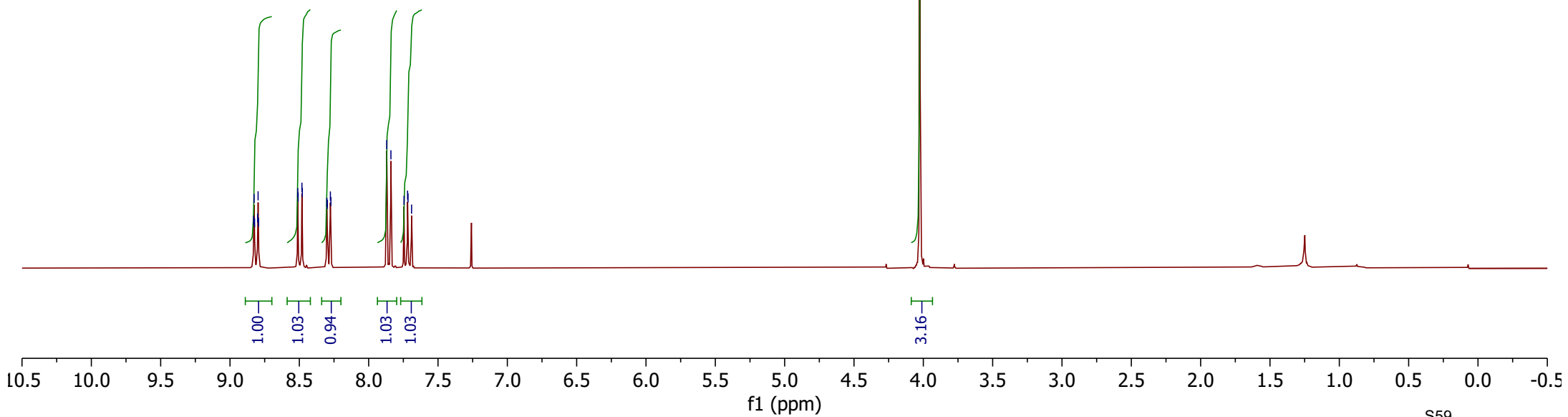
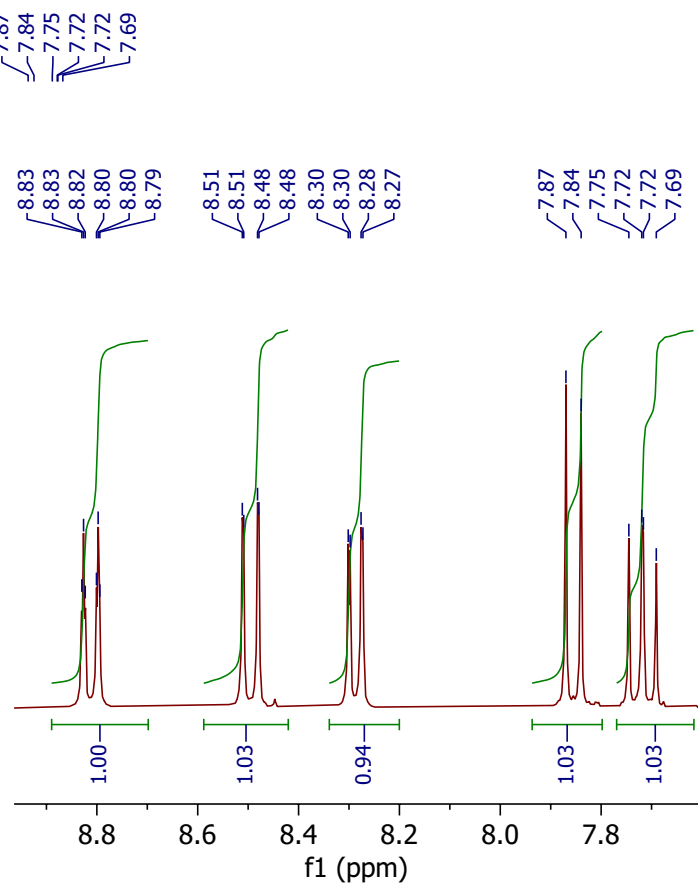


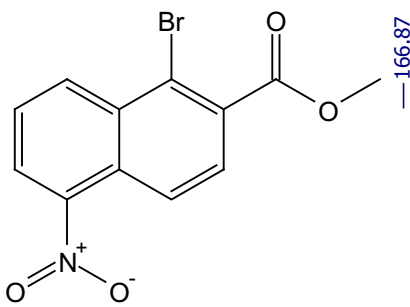




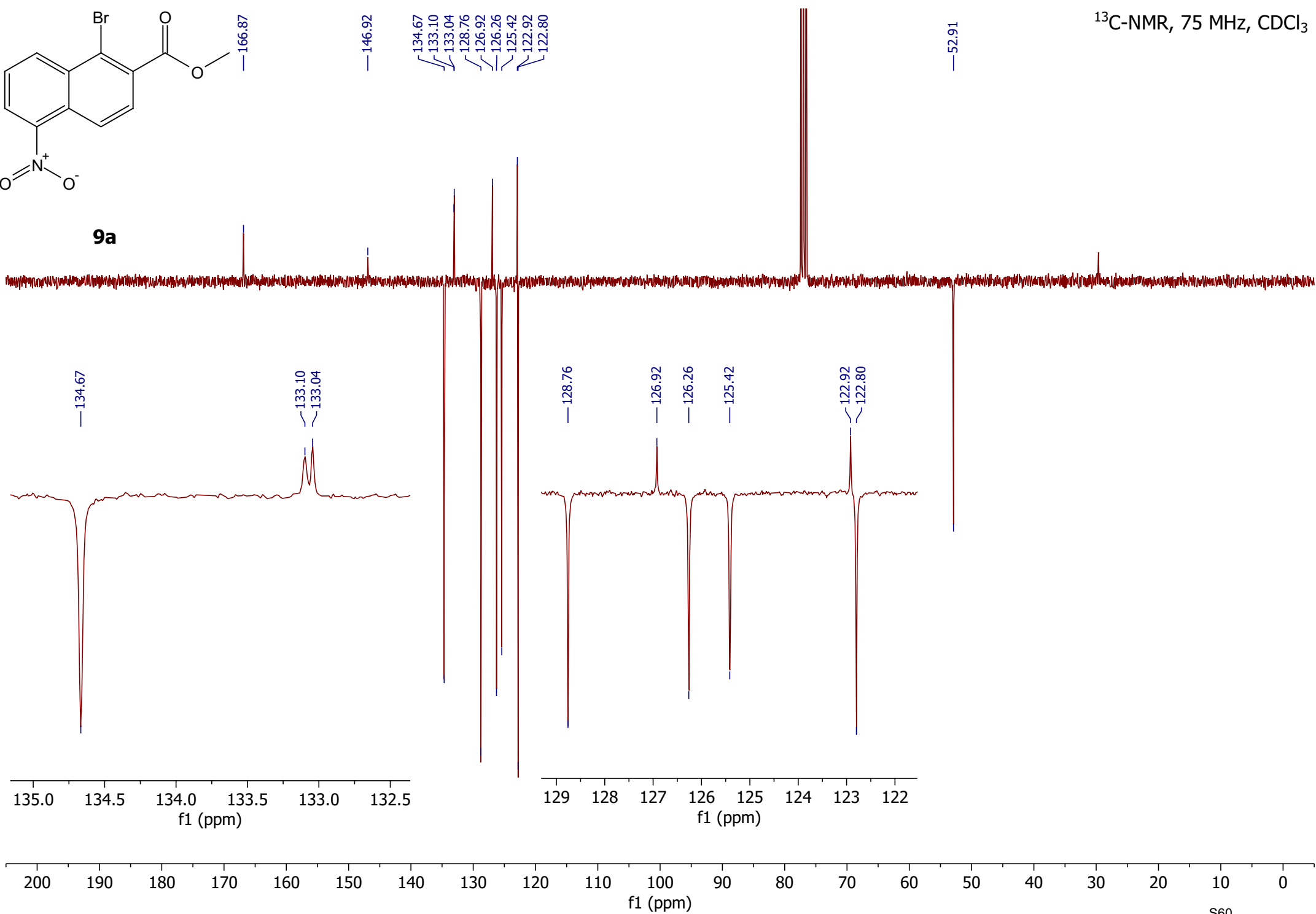


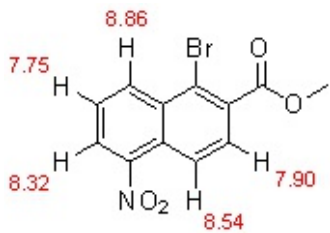
9a



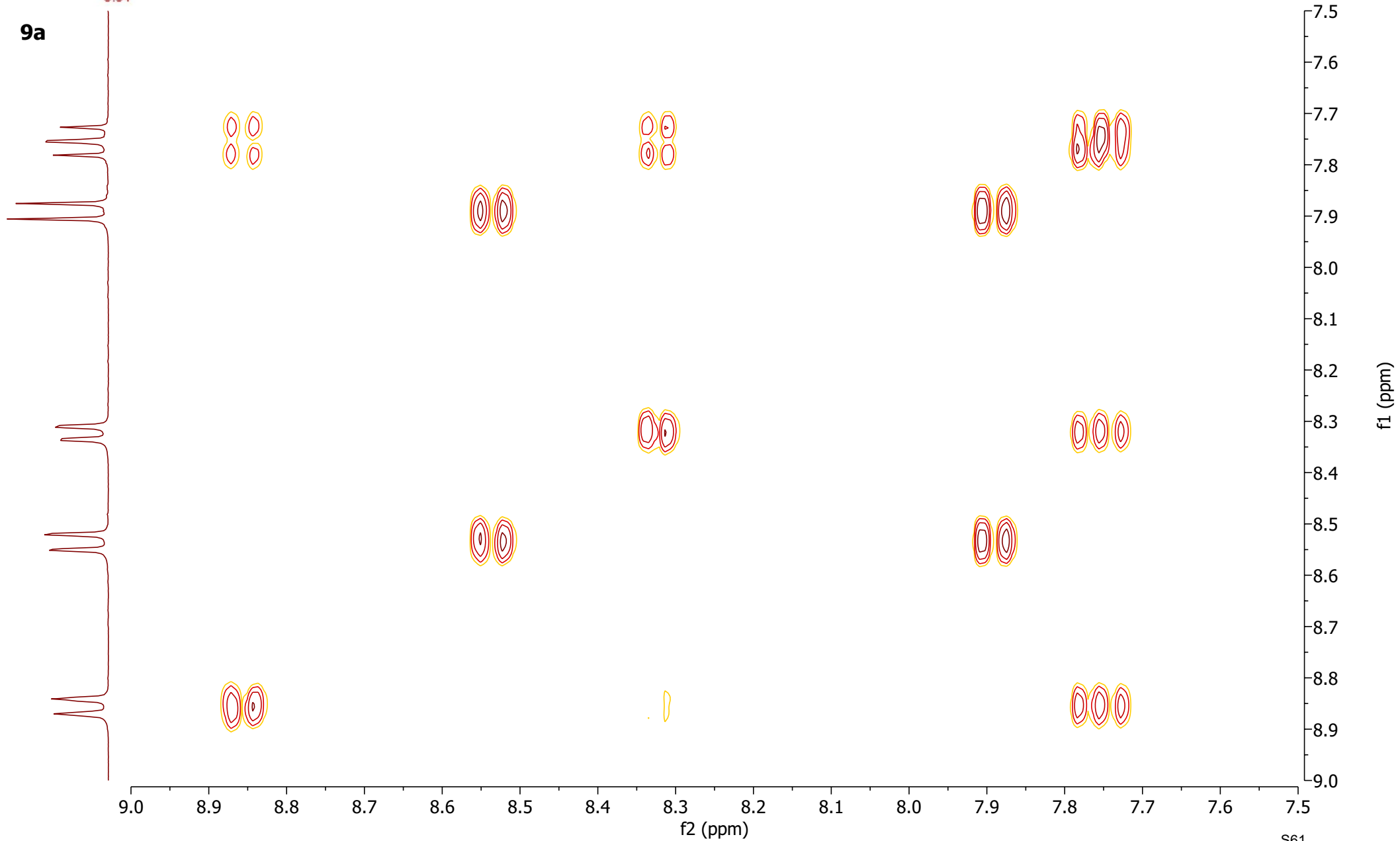


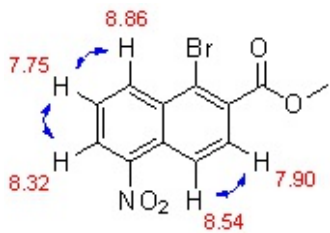
9a





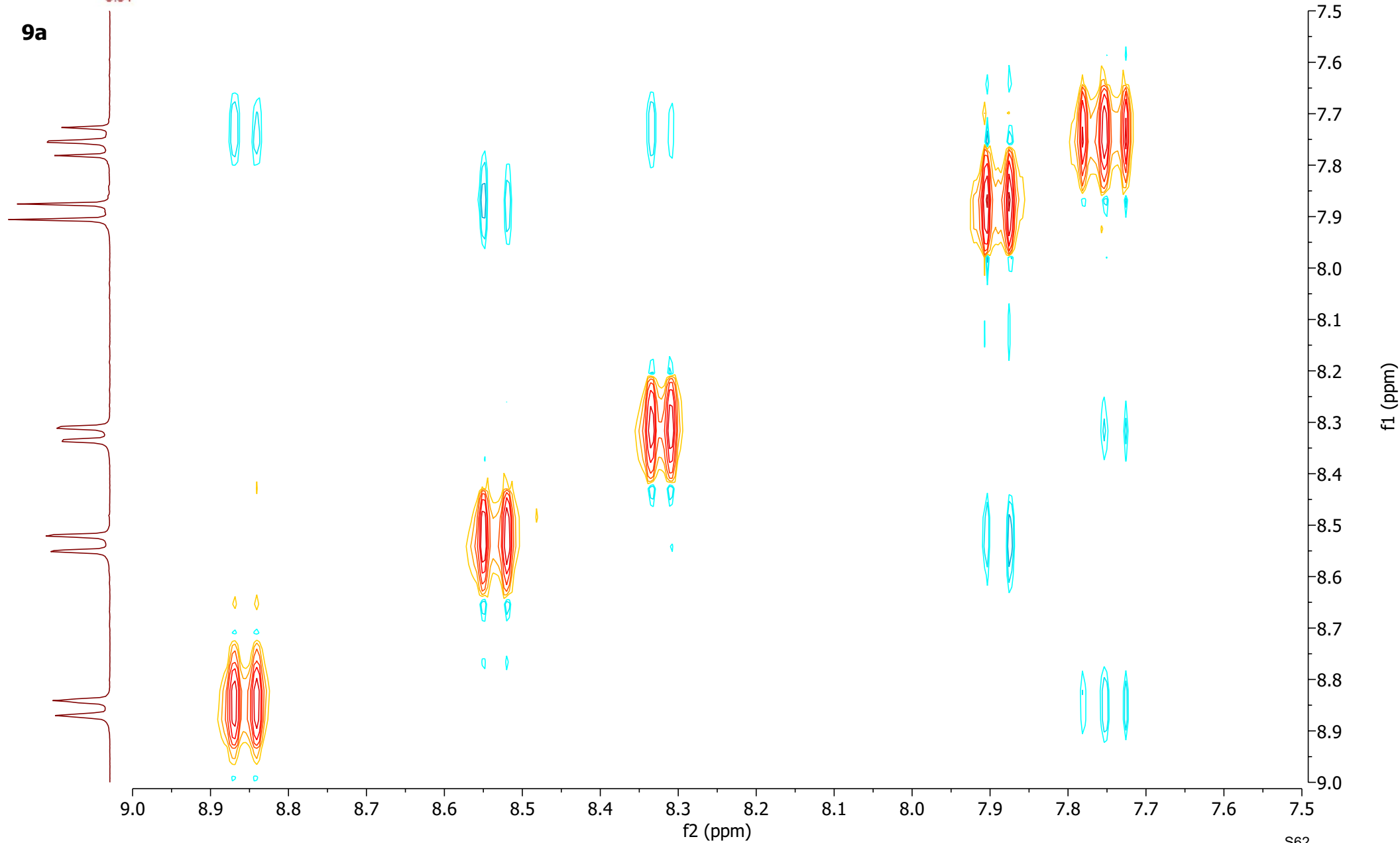
COSY
CDCl₃, T = 300K

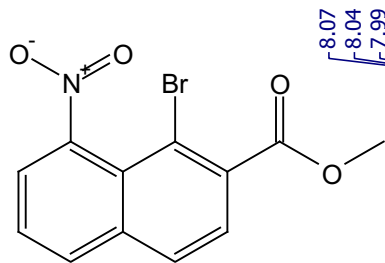




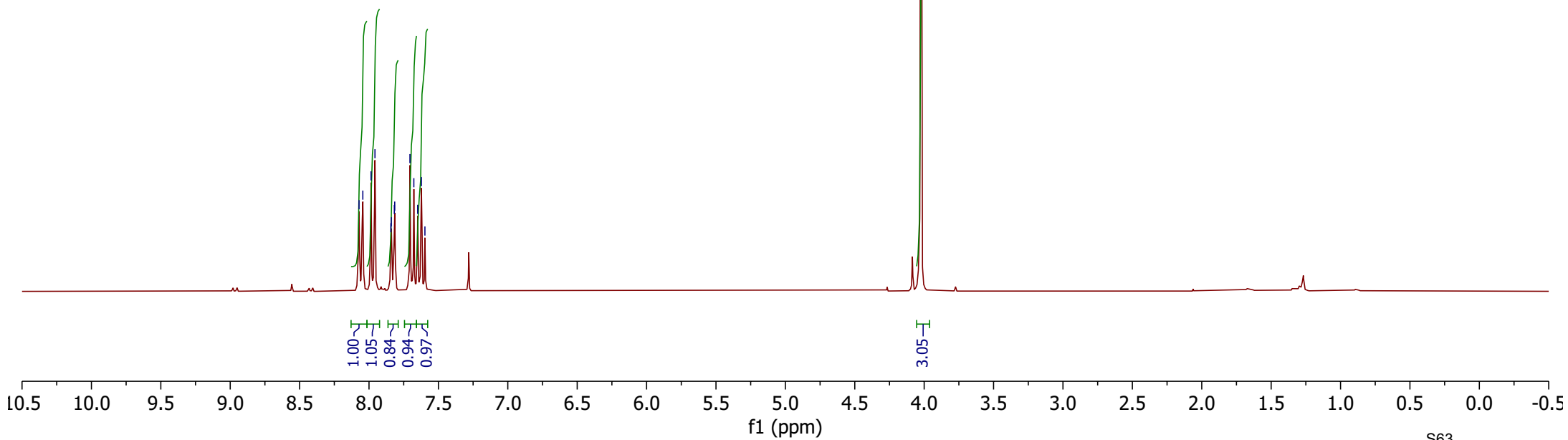
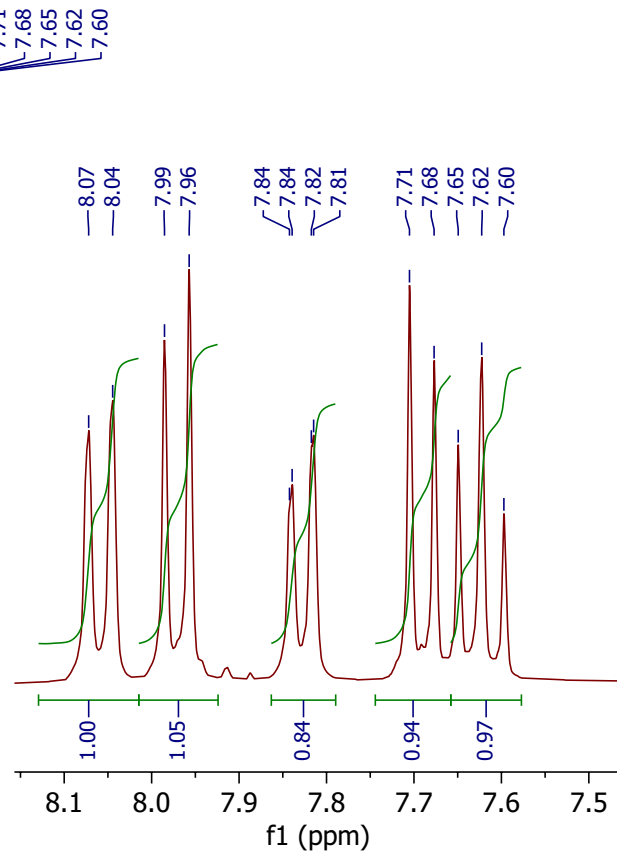
NOESY
CDCl₃, T = 300K
tmix = 2.2 sec

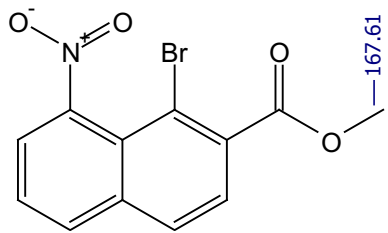
9a



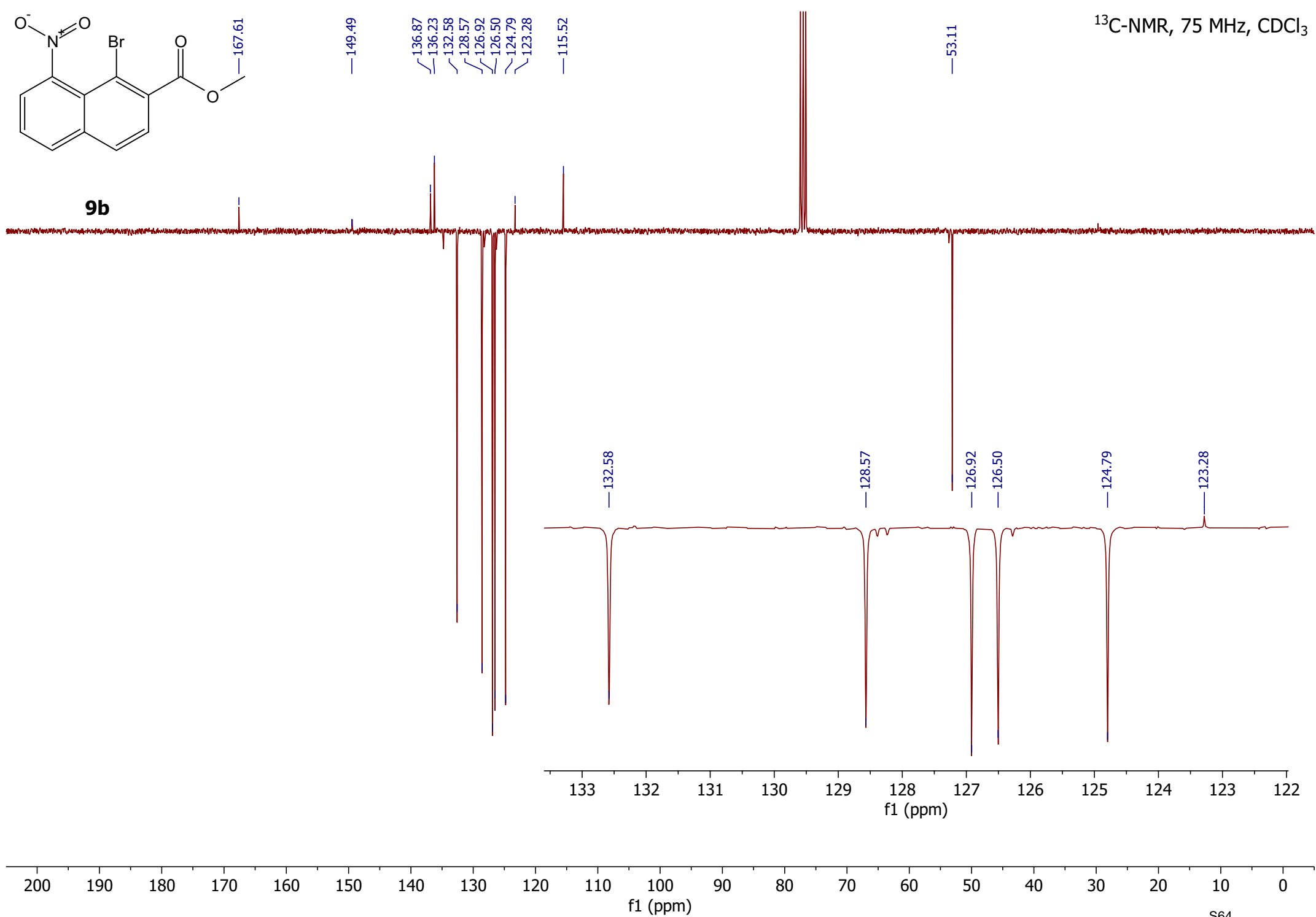


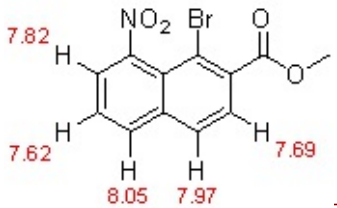
9b



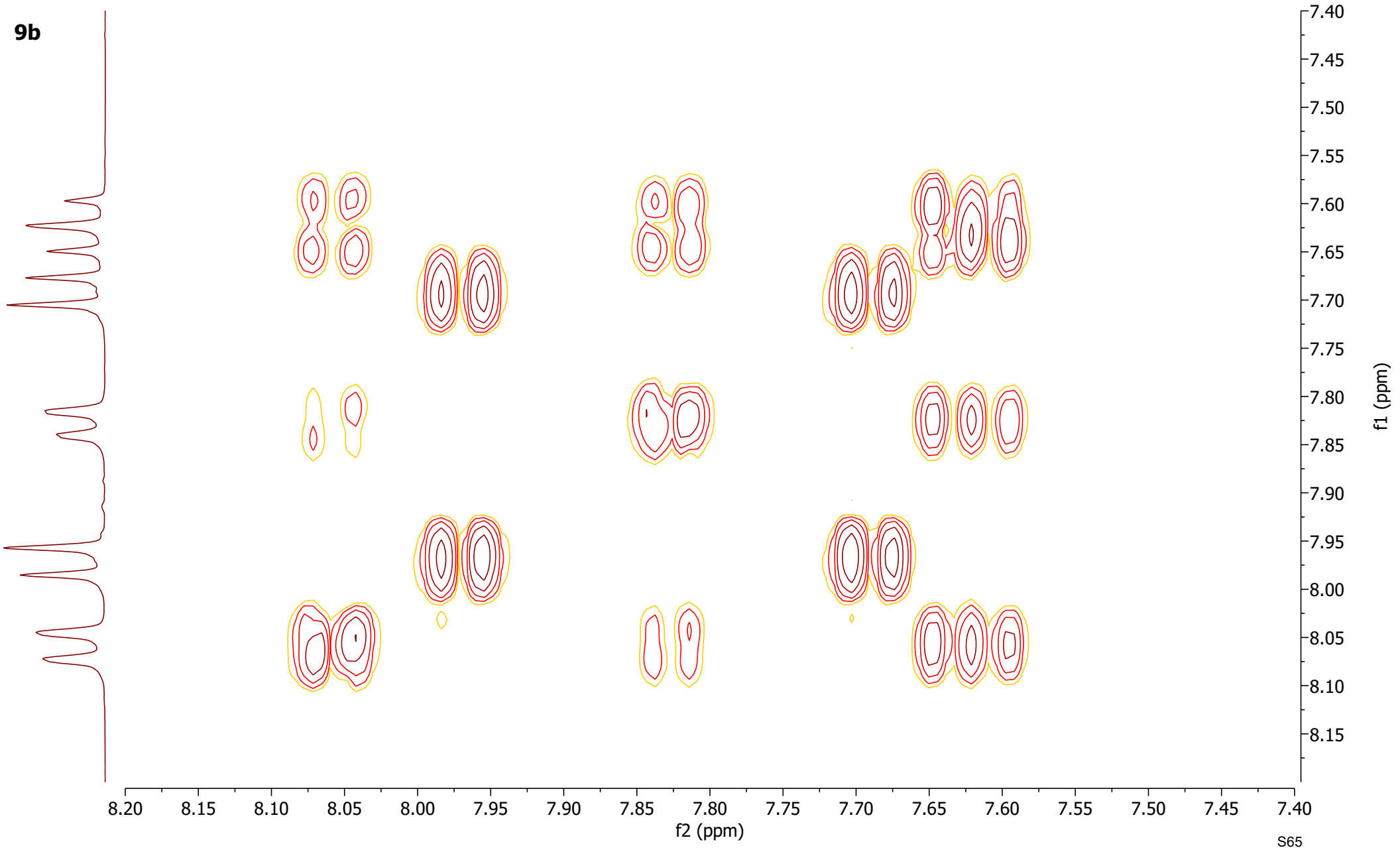


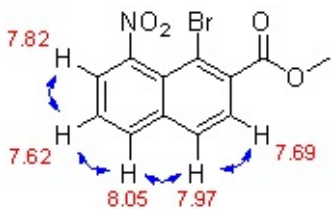
9b



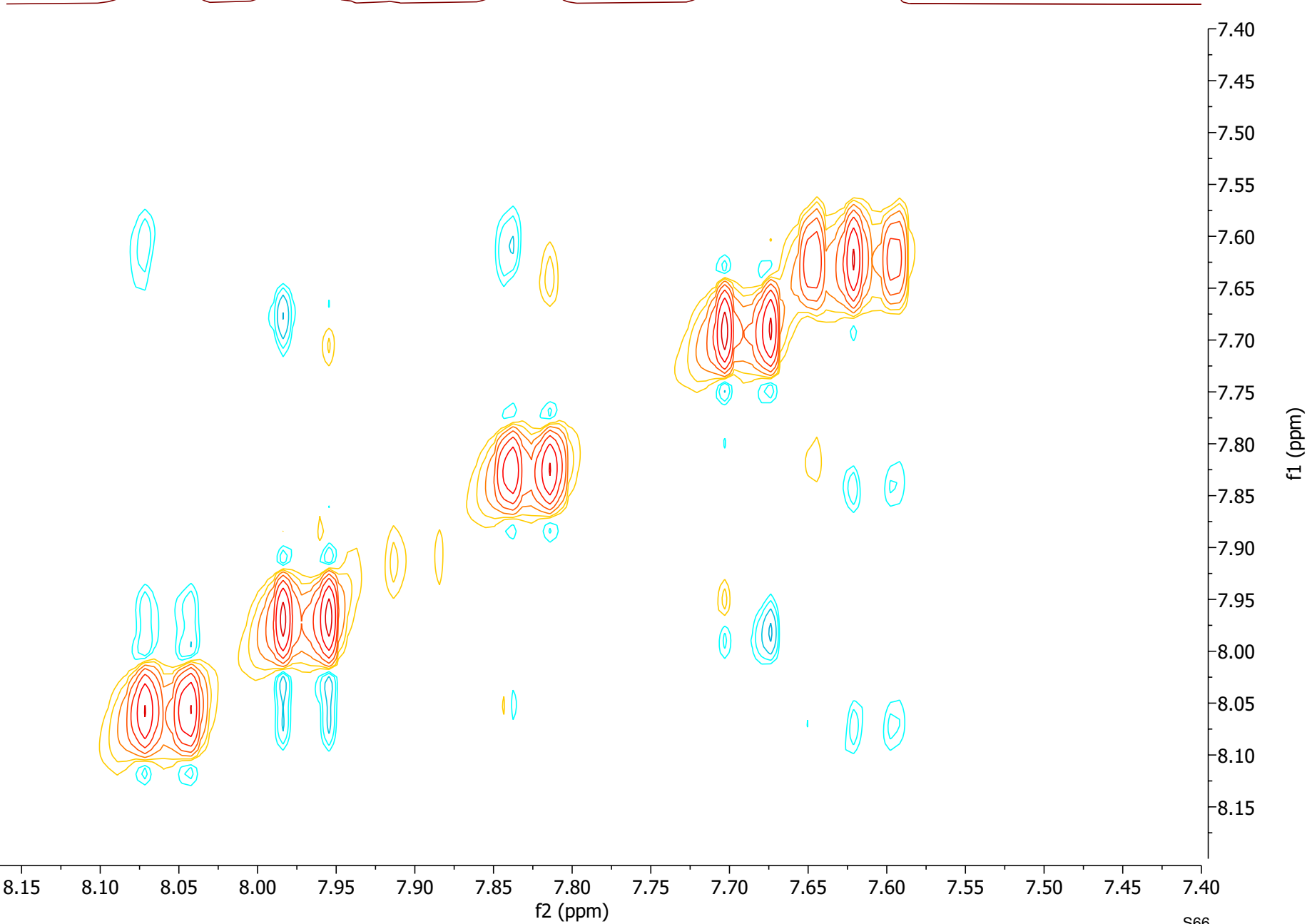


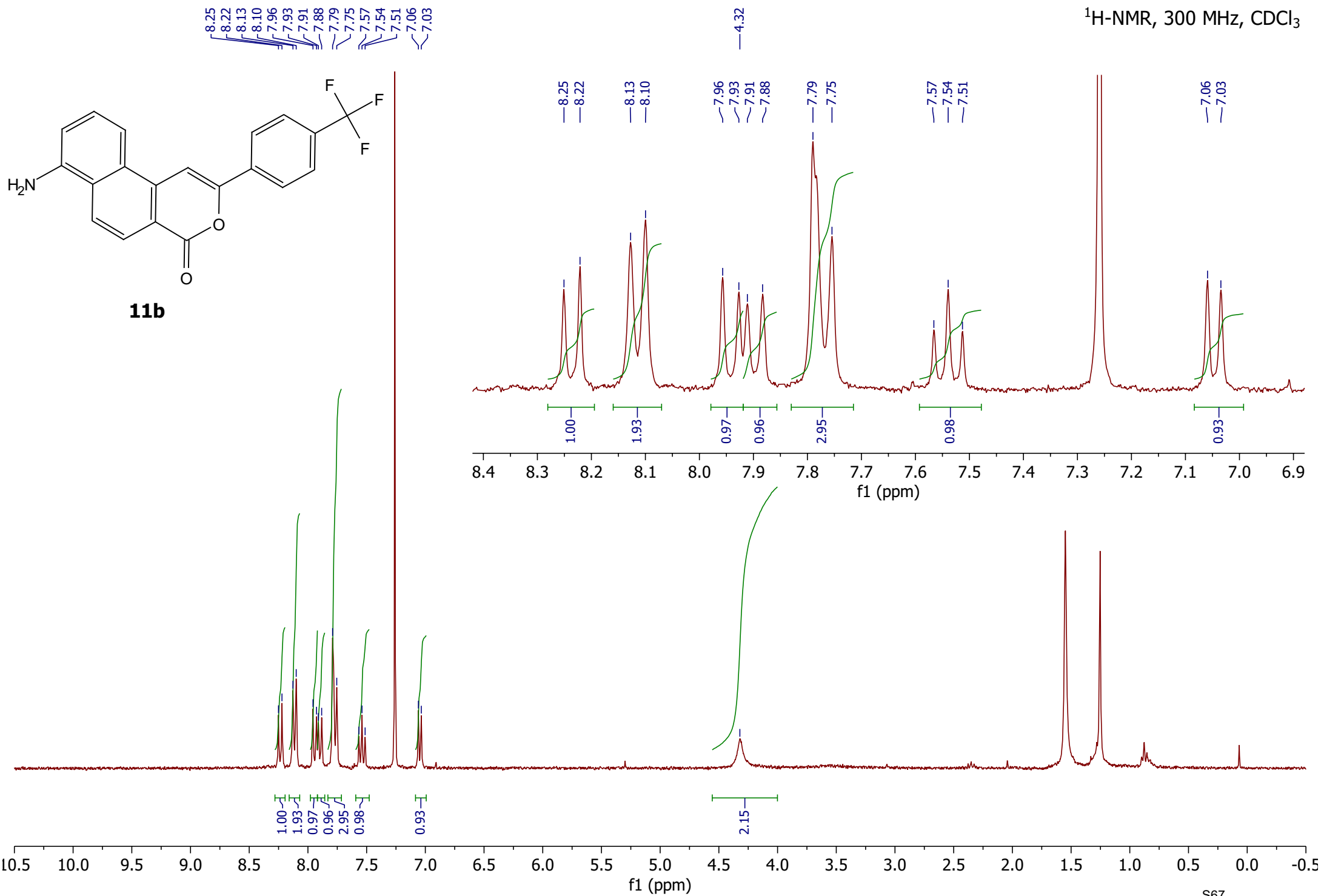
COSY
CDCl₃, T = 25 °C

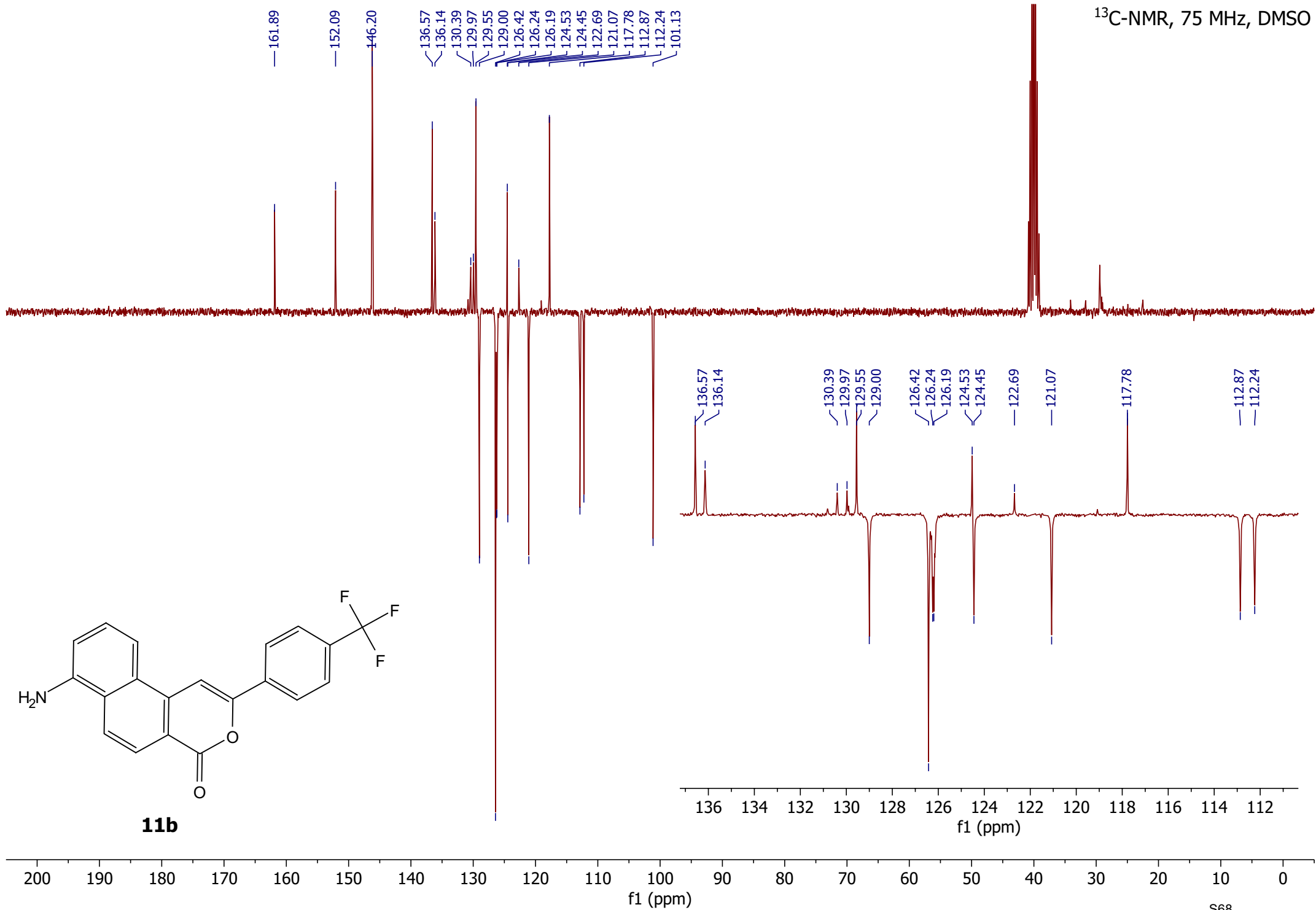


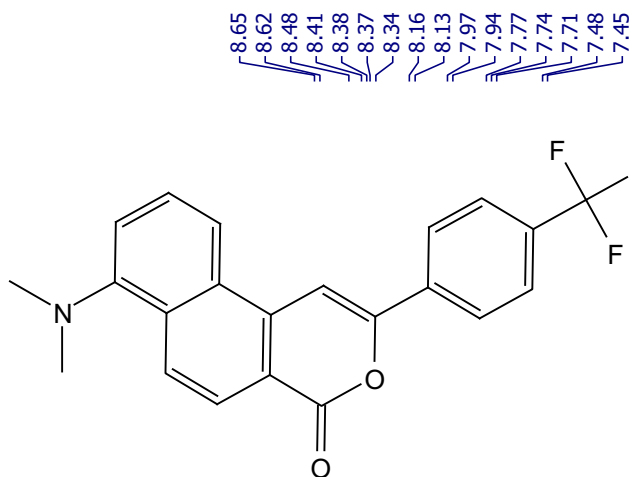


NOESY
CDCl₃, T = 25 °C
tmix = 1.4 sec

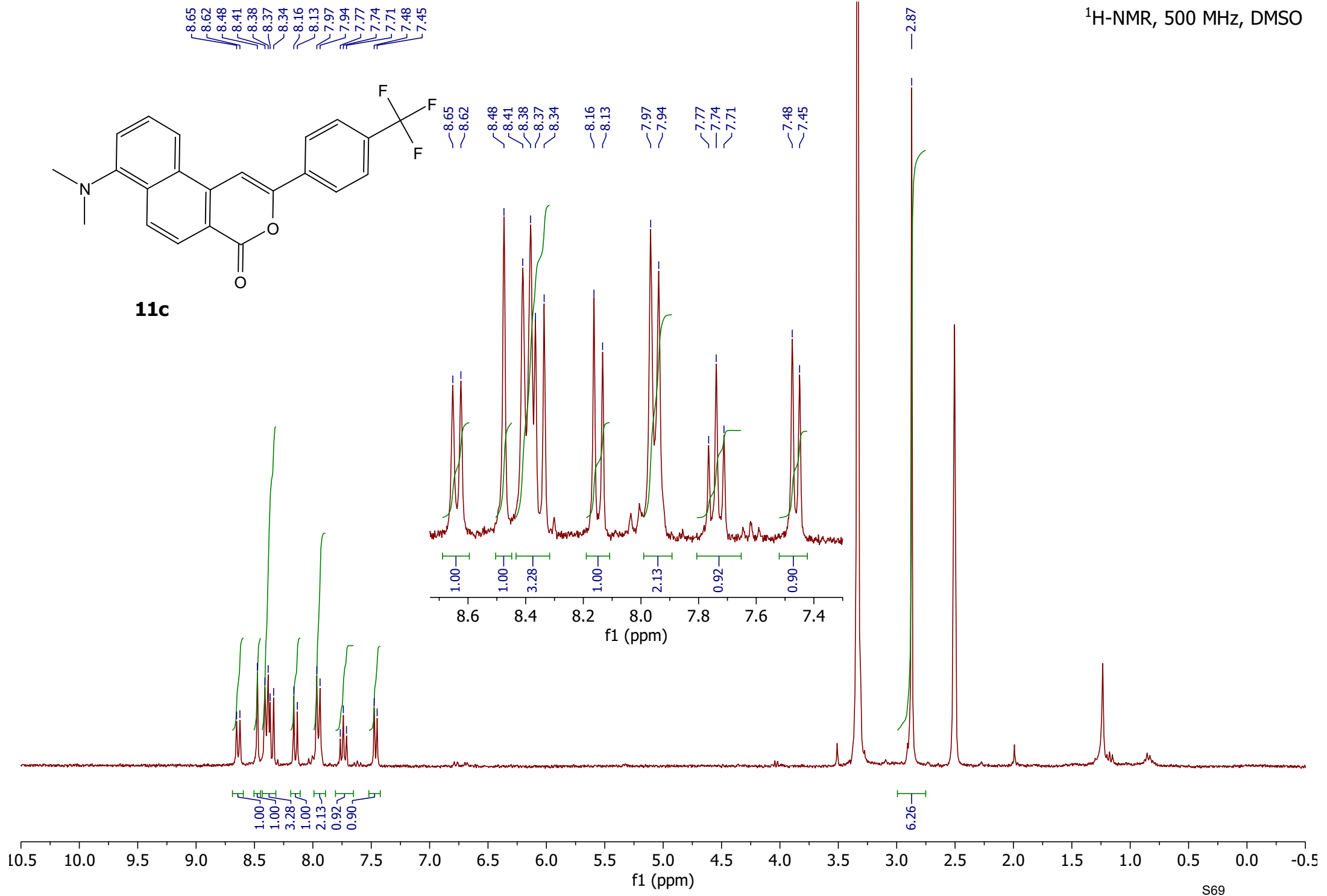


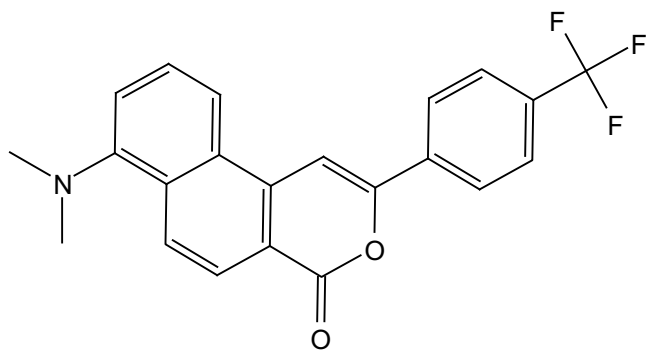




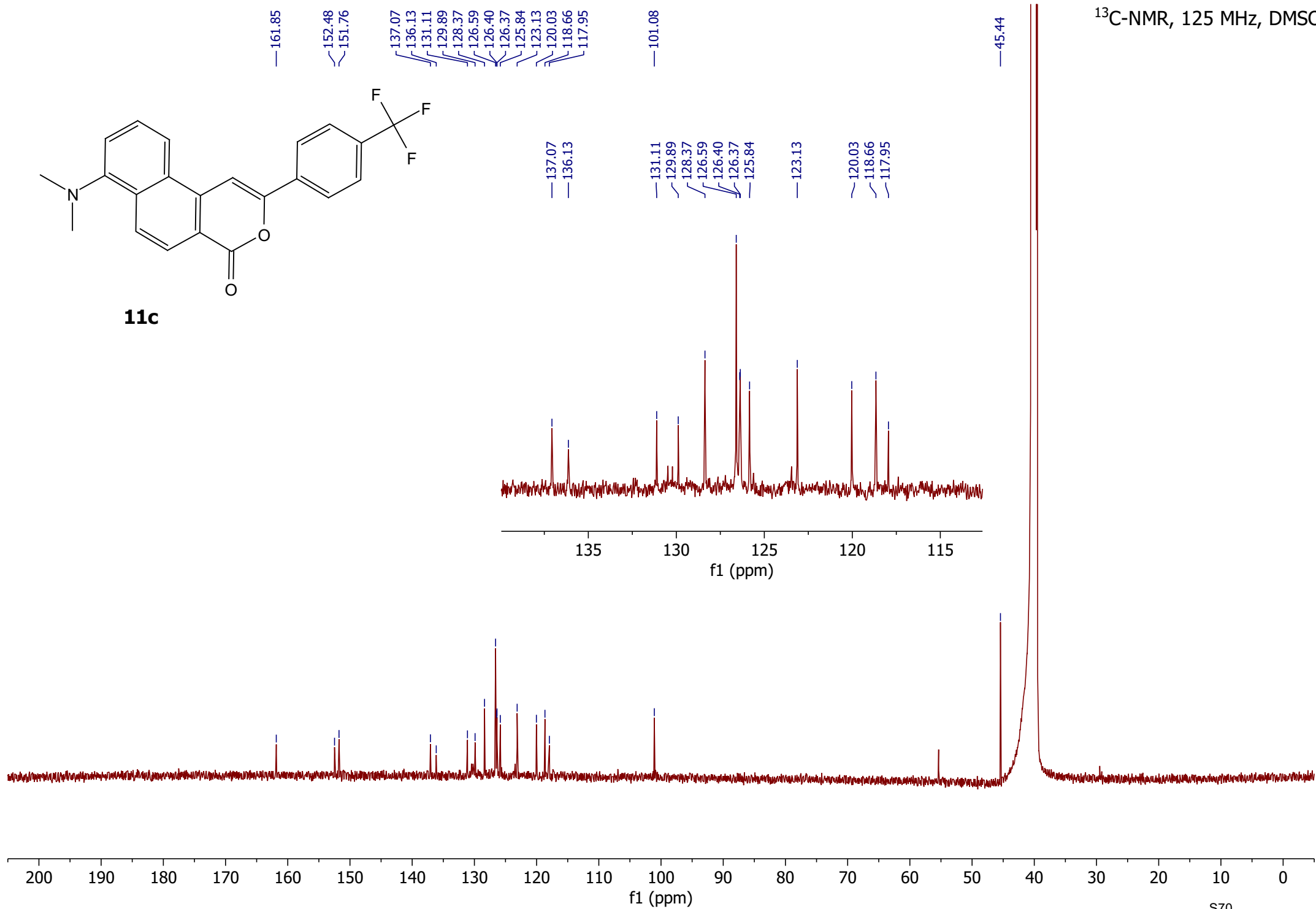


11c





11c

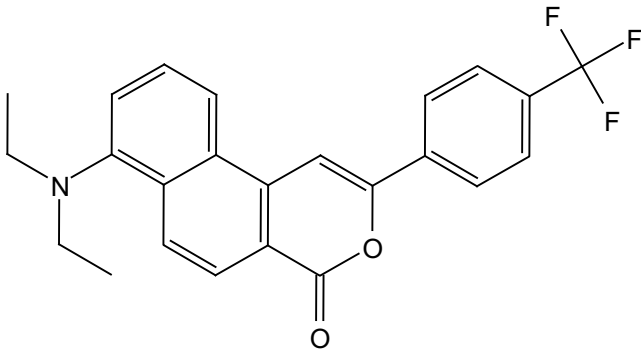


¹H-NMR, 300 MHz, DMSO

8.68
8.65
8.45
8.40
8.37
8.35
8.12
8.09
7.93
7.91
7.76
7.74
7.71
7.54
7.51

3.20
3.18
3.16
3.13

1.00
0.98
0.95



11d

