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The toxicity and bioavilability of pollutants could be enhanced by their interaction with nanomaterials. This new aspect could lead to unpredictable impacts for aquatic wildlife. We studied the possible carrier role of carbon nanopowder for benzo(α)pyrene on zebrafish embryos. Our results clearly show that the adsorption to nanopowder modifies the uptake and the distribution of the pollutant in the organism. The benzo(α)pyrene bound to nanopowder is able to interfere with important cellular targets in a different way in comparison to the pollutant alone. This study opens new views of nano-ecotoxicity highlighting the importance of the interaction between nanomaterials and environmental pollutants. Moreover, our study provides a new methodologic strategy to investigate their effects *in vivo*.

Adsorption of B(α)P on Carbon Nanopowder affects accumulation and toxicity on zebrafish (*Danio rerio*) embryos

Camilla Della Torre^{1*}, Marco Parolini¹, Luca Del Giacco¹, Anna Ghilardi¹, Miriam Ascagni¹, Nadia Santo¹, Daniela Maggioni², Stefano Magni¹, Laura Madaschi¹, Laura Prosperi¹, Caterina La Porta^{1,3}, Andrea Binelli^{1*}

¹Department of Biosciences, University of Milan, Via Celoria 26, 20133 Milan, Italy

²Department of Chemistry, University of Milan Via Golgi 19, 20133 Milan, Italy

³Centre for Complexity & Biosystems, University of Milan, Via Celoria 26, 20133 Milan, Italy

*corresponding authors

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The increasing use of nanomaterials opens several concerns regarding their potential risk for the environment and human health. In particular, the aquatic ecosystems appear highly susceptible. Our research wanted to breakthrough this aspect investigating the interplay between carbon nanopower (CNPW) and a common pollutant such as Benzo(α)pyrene (B(α)P) in zebrafish embryos. To this aim CNPW was contaminated with B(α)P, showing significant adsorption properties towards the hydrocarbon. Embryos were then exposed to CNPW (50 mg/L), B(α)P (0.2-6-20 µg/L) alone and to the CNPW doped with the three B(α)P concentrations. We demonstrated that CNPW helps B(α)P uptake by zebrafish embryos and we also demonstrated that the interaction between CNPW and the hydrocarbon affects the organism stress response pathways eliciting their toxic effect. In particular, the modulation of genes related to the cellular stress response (*cyp1a*, *hsp70*, *sod1*, *sod2*) and the measure of oxidative stress enzyme activities allowed us to identify critical molecular events, modulated by the pollutants alone and in co-exposure. Finally, to evaluate the toxic effects due to CNPW interaction with B(α)P, we analyzed biomarkers of cyto/genotoxicity. As far as no significant genotoxicity was induced by B(α)P and CNPW alone, the co-exposure leaded to an increase of cytotoxicity, and higher incidence of necrotic and apoptotic cells. Altogether our data show that nanomaterials, even if they are not toxic *per se*, could help common pollutants to enhance their toxicity.

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Introduction

Nanoscale science and technology are providing unprecedented and revolutionary advances in several technology and industry sectors. While benefits and improvements of nanotechnology are well established, several concerns have raised regarding the potential risk of nanomaterials (NMs) for the environment and human health.^{1,2} The aquatic ecosystems in particular are highly susceptible to such contamination, being the sink of NMs released from soil, wastewaters and aerial depositions, as well as from direct applications.³ Several studies on aquatic organisms have provided data which highlight the harmful effects of NMs.⁴⁻⁶ Carbon-based NMs (CNMs) in particular have the ability to impact biological systems due to their physico-chemical properties and biological reactivity.⁷ Besides acute toxicity, several sub-lethal effects as genotoxicity, cytotoxicity, oxidative injuries, inflammation, behavioural alterations and reproduction impairment have been documented in aquatic models.⁸⁻¹³

Another extremely important feature of NMs is the ability to interact with many environmental pollutants, such as metals and man-made chemicals. The interactive effects of either metal and carbon-based nanoparticles (NPs) with different environmental pollutants have been extensively investigated on aquatic organisms.¹⁴⁻²¹ As for carbon-based NPs (fullerenes and nanotubes), most of the studies showed a high sorption capacity towards hydrophobic chemicals.²²⁻²⁵ Nevertheless the ecotoxicological consequences of such interactions for natural aquatic environment are hardly predictable, as the studies undertaken so far provide inconsistent results. In some cases it is highlighted the ability of NPs to adsorb and sequester the contaminants, thereby reducing their bioavailability for organisms.^{26,27} On the contrary, other studies indicate that the adsorption of pollutants on NPs might increase their cellular uptake and accumulation.²⁸⁻²⁹

Such uncertainties call the need for a deep understanding of complex mechanisms that occur once CNMs and pollutants meet each other, starting from a thorough evaluation of chemico-physical interactions and how these interactions affect pollutants uptake and distribution inside the biological system and affect cellular pathways leading to toxicity. Therefore, this study was aimed at evaluating the interactions between Carbon Nanopowder (CNPW) -representative of carbon nanoparticulate amorphous form- and the carcinogen Benzo(α)pyrene (B(α)P) on zebrafish (D. rerio) embryos. We doped CNPW with three different concentrations of $B(\alpha)P$, whose sorption on CNPW has been measured by gas chromatography-mass spectrometry (GC-MS/MS) and exposed zebrafish embryos to single contaminants administered alone (CNPW; $B(\alpha)P$) or in co-exposure (CNPW+ $B(\alpha)P$). The novelty of our experimental is to assess the accumulation and biological effects only of the $B(\alpha)P$ bound to CNPW without any interference of free hydrocarbon. A suite of biomarkers was also applied based on the postulated mechanisms of toxicity described for CNMs which involved the generation of oxidative stress³⁰ and DNA damage.³¹ The expression of genes related to cell stress response such as cypla, sodl, sod2 and hsp70 and the activity of proteins involved in the antioxidant response machinery as glutathione-S-transferase (GST), catalase (CAT) and superoxide dismutase (SOD) were measured, to evaluate whether the combination of $B(\alpha)P$ with CNPW induces different molecular pathways respect to the pollutants alone. Finally, the cyto-genotoxicity of the pollutants alone and in combination was compared by the application of trypan blue exclusion method,

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single gel cell electrophoresis, DNA diffusion assay and Micronucleus test, confirming the enhancing effect of the CNPW on the pollutant toxicity.

Methods

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Chemicals and reagents

The CBNW was purchased by Sigma-Aldrich (CAS n. 7440-44-0). The advertised particle size was <50 nm (TEM) and the specific surface area >100 m²/g (BET). Bulk powder was observed by scanning electron microscopy (SEM Zeiss LEO 1430). The powder was mounted onto an aluminium stub and gold coated. B(α)P powder and PAHs' standard (PAH-Mix 14) were supplied by Dr. Ehrenstorfer. All solvents used for PAHs' determination were pesticide grade from Sigma-Aldrich. Reagents for biomarkers and immunostaining were purchased from Sigma-Aldrich. Reagents for RNA extraction and quantitative RT-PCR were from Promega and Biorad.

CNPW clean-up and characterization

An aliquot (5 g) of CNPW standard was cleaned-up with toluene in a Soxhlet apparatus (FALC Instruments, Lurano, Italy) up to 92 h. One hundred mL of toluene were collected after 24-48-52-90 h to monitor the clean-up process over time. We added 20 mL of isooctane to the samples, which were then concentrated to a volume of 1 mL using a rotary evaporator (RV 06-LR, IKA, Staufen, Germany) followed by a gentle nitrogen flow. The effective elimination of PAHs' impurities over time was evaluated by GC-MS/MS.

Purified CNPW was resuspended in milliQ water, stirred for 48 h and sonicated for 15 min and observed through transmission electron microscopy (TEM) for size and shape determination. A drop of CNPW dispersion was placed on 300 mesh formvar copper grids and the excess of water was gently blotted using filter paper. Images were acquired using a Zeiss LEO 912ab Energy Filtering TEM operating at 100 kV, at a magnification of 25-50,000 using a CCD-BM/1 K system. Size distribution of the CNPW suspended in water and surface charge (ζ potentials) were determined by Dynamic Light Scattering (DLS) at 298 K, by using a Malvern Zetasizer nano ZS instrument (Malvern instruments, UK), equipped with solid state He-Ne laser operating at a wavelength of 633 nm and detecting the scattered light at a scattering angle of 173°. Data were elaborated through the Zetasizer Nano Series software, version 7.02 (Particular Sciences, UK).

$B(\alpha)P$ adsorption

Three suspensions of purified CNPW were prepared by dispersing it in milliQ water at 1 g/L and by sonication (15 min) with a probe sonicator (ST.IM.IN Milan) at 12,000 Hz. Proper volumes of $B(\alpha)P$ dissolved in dimethylsulphoxide (DMSO) were added to reach the nominal concentrations of 1, 0.1 and 0.01 mg/L. The suspensions were then stirred for 72 h at 20 °C. At the end of the adsorption process, each suspension was centrifuged for 15 min at 3,000 x g. The CNPW pellets were dried in an oven for 24 h. An aliquot of about 70 mg of each contaminated CNPW was extracted with toluene in a Soxhlet extractor for 24 h and the B(α)P effective adsorption on CNPW was measured by GC-MS/MS.

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GC-MS/MS analysis of PAHs

An aliquot of 2 µL was injected into a GC chromatograph (TRACE GC, Thermo-Electron, Texas, USA) equipped with a programmed temperature vaporizer (PTV) injector, an AS 2000 autosampler (Thermo Electron) and an Rtx-5MS (Restek, Bellefonte, PA, USA) capillary column (30 m length, 0.25 mm I.D., 0.25 µm film thickness). The gas chromatograph was coupled with a PolarisQ Ion Trap mass spectrometer. Chromatographic conditions were as follows: PTV in solvent split mode (split flow= 50 mL min⁻¹) and splitless time of 2 min; carrier gas helium at 1.2 mL min⁻¹; surge pressure of 280 kPa; injection pressure of 67 kPa; transfer pressure of 134 kPa; injector temperature starting at 70 °C and held for 1.2 min, then ramped to 300 °C at 14 °C s⁻¹ and held for 1.2 min; oven temperature starting at 70 °C and held for 1.2 min, then ramped to 220 °C at 40 °C min⁻¹ and held for 1 min, and finally ramped to 290 °C at 4 °C min⁻¹ and held for 8 min. PAHs were quantified by GC/MS under the following instrumental conditions: Selective Ion Monitoring (SIM) mode after Electron Ionization (EI) with standard electron energy of 70 eV; transfer line at 280 °C; ion source at 260 °C and the damping gas at 1 mL min⁻¹. Quantitative analyses were performed using Excalibur software (Thermo Electron) with external calibration curves ranging from 10 to 50 µg/L. A blank sample was run in parallel. The most volatile PAHs, showing non-negligible concentrations, were corrected by the blanks.

Preparation of CNPW suspensions and hydrodynamic behaviour in exposure media

Suspensions of purified CNPW and CNPW contaminated with $B(\alpha)P$ at concentration of 50 mg/L were equilibrated in zebrafish water (ZFW) for one week by stirring in the dark at 20 °C. Hydrodynamic diameters and surface charges (ζ potentials) of all samples were determined by DLS as described above. Time-dependent variations of the size distributions of the CNPW suspensions in zebrafish water were also monitored, at 2-4-24 h recording each measurement in quadruplicate.

Zebrafish embryos exposure

Zebrafish embryos of the AB strain were obtained from adult fish reared in the zebrafish facility of the Department of Bioscience, University of Milan. Our facility strictly complies with the relevant Italian laws, rules and regulations (Legislative Decree No. 116/92), as confirmed by the authorization issued by the municipality of Milan (Art. 10 of Legislative Decree No. 116, dated 27.1.1992). The whole procedure was carried out in accordance with the relevant guidelines and regulations.

To avoid any physical interference with the uptake of CNPW, the chorion was removed by enzymatic reaction with pronase (0.5 mg/mL) at 24 hours post-fertilization (hpf), immediately prior to the exposure. Dechorionated embryos were exposed in Petri dishes (4 mL) to B(α)P alone (0.2, 6, 20 μ g/L), CNPW (50 mg/L) alone and contaminated with B(α)P (CNPW + B(α)P 0.2 μ g/L, CNPW + B(α)P 6 μ g/L, CNPW + B(α)P 20 μ g/L). B(α)P concentrations were defined based on the effective B(α)P sorption on CNPW (50 mg/L) measured at GC-MS/MS (see results, Tab. 2). Control animals were exposed to zebrafish water (ZFW) and to vehicle (0.1% DMSO) only.

Based on several preliminary range-finding experiments, the adopted CNPW and $B(\alpha)P$ concentrations did not produce mortality or teratogenic effects. Anyhow, these enpoints were checked at the end of each exposure experiment.

The exposure proceeded until 96 hpf under semistatic conditions, and the exposure solutions were changed every 24 h. For gene expression and biochemical analyses each batch of treated and control embryos was placed in 1.5 mL microtubes and stored at -80 °C until processing. For genotoxicity assessment, embryos were processed immediately at the end of the exposure as described below. A companion exposure aimed at assessing the uptake/accumulation of CNPW and B(α)P was performed exposing embryos in ZFW containing 0.003% 1-phenyl 2-thiourea (PTU) to prevent pigmentation. At the end of exposures, embryos were fixed in paraformaldehyde (4% in 0.1 M phosphate buffered solution at pH 7.2 - PB). With the aim to accurately localize B(α)P fluorescence signal inside embryos, we used the fli1:GFP transgenic line *Tg(fli1a:EGFP)*,³² which enables the visualization of the circulatory system in green. Experiments were run at least 3 times for each analysis.

Whole mount confocal microscopy

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59 60 To define the B(α)P uptake localization in embryos, we recorded the B(α)P fluorescence signal through confocal microscopy. This technique has been recognized as reliable indicator of hydrocarbon exposure in biological systems including fish embryos.^{33,34} Z-stacks of whole mount *Tg(fli1a:EGFP)* zebrafish embryos were collected using a Leica SP2 laser scanning confocal microscope (Leica Microsystems) with a Plan-Apochromat 10x/0.4 objective. To record the B(α)P-associated fluorescence, images were acquired using 405 nm excitation laser and the emission signal was collected in the range of 410-550nm. The maximum projection of fluorescence images was achieved for each embryo (n = 3) and was analized using ImageJ analysis software v.1.49k. To quantify B(α)P accumulation, the Integrated Density (Area x Mean Grey Level) was calculated.

CNPW localization

For CNPW localization embryos were cryoprotected in sucrose 15 % for 2h and sucrose 30 % overnight a 4 °C. Sections (10µm) were cut by a cryostat (Leica CM1850) and were sequentially incubated in 0.05 M NH₄Cl in PB for 30 min to quench free aldehyde groups, in 0.01 M phosphate-buffered saline pH 7.4 (PBS) containing 1% bovine serum albumin (BSA) and 0.2 % Triton X-100 (30 min). Sections were exposed to phalloidin (Cytoskeleton 555 actin stain, TRITC conjugate, PHDH1) which binds with the filamentous actin, for 2 h at room temperature, washed extensively in PBS and mounted in PBS/glycerol (1:2 v/v), with DNA-binding dye 4'-6-diamidino2-phenylindole (DAPI –specific fluorescent probe). Sections were observed with confocal microscope Leica SP2 microscope with He/Kr and Ar lasers (Leica, Wetzlar, Germany) and CNPW was visualized in reflection mode at 488 nm. For the localization of nanoscale CNPW by ultrastructural analysis, 10 embryos from each experimental group were fixed in a mixture of 4% paraformaldehyde and 2.5% glutaraldehyde in 0.1 M sodium cacodylate buffered solution at pH 7.4. After washes in the same

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buffer, embryos were postfixed in 1% OsO₄, dehydrated in a graded ethanol series and infiltrated in Araldite-Epon. Sectioning was performed using an Ultracut E microtome (Reichert, Austria). Ultrathin sections of about 70 nm were collected on 600-mesh uncoated copper grids. Sections were not counterstained to avoid interference with CNPW visualization. Digital images were acquired using a CCD-BM/1K system, and image elaboration was performed using the ESI vision software AnalySIS (Soft Imaging Systems, Muenster, Germany). The identity of CNPW was confirmed through Electron Spectroscopy Imaging, using a Zeiss LEO 912ab energy-filtering transmission electron microscope.

RNA extraction and quantitative RT-PCR (qRT-PCR)

Total RNA was isolated using SV Total RNA Isolation System (Promega, Madison, Wisconsin). Each sample was incubated with DNAse to eliminate any genomic DNA residue from the RNA preparation. First-strand cDNAs were synthesized employing the ImProm-II Reverse Transcription System (Promega), according to the manufacturer's protocol, using random oligonucleotides to prime the reverse transcription of 1 μ g of total RNA. qPCRs were performed using *cyp1a*, *hsp70*, *sod1* and *sod2* specific primers (see table S1 for sequences). For normalization purposes, 18S ribosomal RNA level was also tested in all the samples. Reactions were performed in a 96-well format iQ5TM Multicolor Real-Time PCR Detection System (Biorad) using the iQTM SYBR® Green Supermix (Biorad). Three independent qRT-PCR experiments from the same reverse transcribed sample were performed using a pair of gene-specific primers. The presence of a single PCR product was verified by melting-curve and agarose gel analyses. The gene expression levels were considered significantly different when 2-fold higher or lower than those measured in controls (0.5 < not significant <2).

Biochemical analysis

Pools of 60 embryos were homogenized using a pestle in 100 mM potassium phosphate buffer (KCl 100 mM, EDTA 1 mM, protease inhibitors 1:100 v/v, dithiothreitol 1 mM pH 7.4). The homogenates were centrifuged at 15,000 x g for 10 min at 4 °C. The supernatant was collected and GST, SOD and CAT activities were immediately measured in triplicate. The GST activity was measured by adding reduced glutathione (1 mM) in 100 mM phosphate buffer (pH 7.4) and using CDNB (1mM) as substrate. The reaction was monitored for 1 min at 340 nm. The CAT activity was determined by measuring the consumption of H₂O₂ (50 mM) in 100 mM potassium phosphate buffer (pH 7) at 240 nm. The SOD activity was determined by the superoxide anion generated by the xanthine oxidase (1.87 mU/mL)/hypoxanthine (50 μ M) reaction at 550 nm. The activity is given as SOD units (1 SOD unit=50% inhibition of the xanthine oxidase reaction).

The total protein content of each sample was measured according to the Bradford $(1976)^{35}$ method using bovine serum albumin as standard.

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Biomarkers of cyto-genotoxicity were performed on cells dissociated from a pool of ten zebrafish embryos (three pools per treatment). Since biomarker methods are described elsewhere,³⁶ a brief description was reported here. Cell viability was assessed by the trypan blue dye exclusion method. Embryo cell suspension (10 μ L) was mixed with 10 μ L of 0.4 % (w/v) trypan blue solution in PBS and transferred to a hemocytometer. Non-viable cells were stained deep blue, whereas viable cells were colorless. Three replicates per experimental group were performed. The alkaline (pH > 13) Single Cell Gel Electrophoresis (SCGE) assay was performed according to the method described in Koshmel et al. (2008).³⁷ One hundred cells per slide (n = 9; three slides per each pool) were analyzed using the Comet Score[®] image analysis software. The percentage of DNA in the comet tail was used as endpoints of primary genetic injuries. The apoptotic cell frequency (%) was assessed³⁸ analyzing three hundreds cells per slide (n=6; two slides per each pool). The frequency of micronuclei (MN‰) was calculated on 400 cells/slide (n=6; two slides per each pool) according to Pavlica et al. (2000).³⁹

Statistical analysis

To evaluate differences between control and treated groups the one-way analysis of variance (ANOVA) was applied to data from biomarker and fluorescence analysis. Specifically, the B(α)P groups were compared to the carrier group (DMSO), while groups treated with CNPW alone and in combination with B(α)P were compared to the control. The ANOVA was followed by a Fisher LSD *post-hoc* test to investigate significant differences taking *p*<0.05 as significance cut-off. Statistical analyses were carried out using the STATISTICA 7.0 software package.

Results and Discussion

This study evaluated a key issue related to ecotoxicity of carbon-based NMs such as their interactions with environmental pollutants and the toxicological consequences for aquatic organisms.

The interactive effects of CNPs and pollutants in fish have been investigated on early life stages and on adults.⁴⁰⁻⁴³ While these studies clearly highlighted the impacts of NPs on the uptake and relative accumulation of the adsorbed pollutants in the organisms, their effects on toxicity are still controversial since NPs can decrease the contaminants' bioavailability or act as Trojan-horse, raising and/or changing the toxicological behavior of associated pollutants, based also on the physiology of the biological model.⁴⁴

In this context our research aims to understand how the interplay between CNPW and B(α)P influences contaminants accumulation and related toxicity. First we assessed the sorption property of CNPW towards B(α)P and how it reflects on B(α)P uptake by zebrafish embryos. Secondly, we demonstrated that such interaction affects organism stress response pathways and toxic effects of the single pollutants.

CNPW characterization and behavior in exposure media

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59 60 The primary characterization by SEM showed that the bulk CNPW consisted of large aggregates up to 10 μ m (Fig 1). The superficial ultrastructure appeared to be made up of graphite-like sheets with variable dimensions well-packed each other. Furthermore, TEM observation highlighted that the CNPW consisted also of another component in particulate form with mean diameter of about 20 nm (Fig. 1). Concerning CNPW behavior in water media, the Z average measured for CNPW suspended in Milli-Q water (1 g/L), within the *range* 600-1000 nm, indicated a substantial aggregation state. Once in ZFW, a significant aggregation of CNPW and consequent sedimentation occurred, starting already after 2 h (Fig S1). This behavior was confirmed by DLS analysis, showing a reduction of size distribution of the CNPW left into the finest fraction of CNPW, and also with larger aggregates (Fig. S1). As to surface charge, the ζ potential value of CNPW suspended in Milli-Q water of -30.9 ± 3.05 mV moved to -20.7 ± 1.15 mV once CNPW was in ZFW. The sorption of B(α)P did not affect significantly the hydrodynamic behavior of CNPW (Table 1).

CNPW clean-up and $B(\alpha)P$ sorption

GC-MS/MS analysis showed a great amount of PAHs' impurities on CNPW up to 20 ppm, equal to 0.002% of total carbon mass. Fifteen PAHs (US EPA) were identified on the CNPW, at the end of the 92 h clean-up, the PAHs' impurities on CNPW were reduced to 1.4 ppm (Tab S1).

After 72 h of contamination with $B(\alpha)P$, a negligible fraction of the hydrocarbon remained into water, corresponding to 0.1% of the administered concentration for CNPW + B(α)P 0.01 mg/L, 0.02 % for CNPW + B(α)P 0.1 mg/L and 0.07 % for CNPW + B(α)P 1 mg/L (Tab. 2). A significant sorption of B(α)P on CNPW was observed, which covered two orders of magnitude, reflecting the three different concentrations of B(α)P administered to CNPW. In detail, about 100 % of B(α)P was adsorbed on CNPW at the B(α)P concentration of 0.1 mg/L. Instead, the percentage of $B(\alpha)P$ recovered on CNPW was 42 % and 33 % at 0.01 mg/L and 1 mg/L, respectively. The different extent of recovery is probably due to the particular characteristics of the CNPW, which was formed by two different components: one composed of nanometric graphite-like sheets and another constituted by nanometric particles. Thus, in spite of our attempt to homogenize the suspensions, they might have adsorbed the $B(\alpha)P$ differently. Despite the significantly different percentage of recovery of $B(\alpha)P$ sorbed on CNPW, our goal to obtain three CNPW doped with increasing amount of B(α)P was achieved. Indeed, the amount of B(α)P sorbed on CNPW doped with B(α)P 0.1 mg/L is lower than that adsorbed on CNPW doped with 1 mg/L (111 μ g/g vs 335 μ g/g), regardless the % of recovery. On the basis of these results, we set the $B(\alpha)P$ concentrations corresponding to suspensions containing 50 mg/L of CNPW (Tab. 2). The CNPW showed high adsorption properties towards organic compounds, as widely described for carbon-based NPs.²²⁻²⁵ The observed properties highlight that airborne NMs can contribute to PAHs transit into the atmosphere, as well as in the water column, representing a further PAHs source for aquatic environments. Besides, the high sedimentation rate observed within 24 h highlighted that CNPW could transport PAHs either into the water column and mainly in sediments.

CNPW and B(a)P uptake in embryos

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Images of whole mount embryos exposed to $B(\alpha)P$ alone showed a dose-dependent increase of $B(\alpha)P$ fluorescence signal in the trunk muscle (Fig. 2 d, e, f; Fig. 3 p < 0.001). At the highest $B(\alpha)P$ concentration (20 µg/L), an intense fluorescence signal was observed also into the yolk sac (fig. 2 f). A different pattern of $B(\alpha)P$ accumulation was shown in embryos treated with CNPW contaminated with $B(\alpha)P$ (Fig. 2 g, h, j), where $B(\alpha)P$ was localized in the trunk muscle but without observe a dose-dependent accumulation (Fig. 3). A higher fluorescence signal was detected at the lower $B(\alpha)P$ concentration (CNPW + $B(\alpha)P$ 0.2 µg/L and CNPW + $B(\alpha)P$ 6 µg/L) compared to the single exposure (p = 0.001, p = 0.006 respectively), while at the highest one (CNPW + $B(\alpha)P$ 20 µg/L) a lower accumulation of $B(\alpha)P$ was measured compared to the $B(\alpha)P$ alone (p < 0.001) and no fluorescence at all was observed in the yolk. The negligible fluorescence observed in all exposure conditions in the eye and cephalic structures is only due to the natural fluorescence of the surface pigment layer present at the head level.

These evidence highlight that, in embryos, the $B(\alpha)P$ distribution results different in dependence to the way of administration, that is alone or adsorbed on CNPW. Indeed, the $B(\alpha)P$ signal followed a dose-dependent increase upon single exposure, while in co-exposure a similar accumulation of the hydrocarbon seemed to occur independently from the administered $B(\alpha)P$ concentrations, confirming that the two contaminants entered together, being associated one to the other. Besides, the potential of CNPW to increase the bioavailability and accumulation of the hydrocarbon is suggested for lower $B(\alpha)P$ concentrations. Though, at the highest concentration, the strong $B(\alpha)P$ signal in the yolk sac, observed after single exposure, could not be detected in embryos exposed to CNPW contaminated with $B(\alpha)P$. This evidence suggests a different internalization and sorption of $B(\alpha)P$ on CNPW with respect to $B(\alpha)P$ alone dissolved into the exposure medium. When dissolved in water it is likely that $B(\alpha)P$ enters the embryo's body through the epidermis and accumulates into the yolk. When in co-exposure, however, the B(α)P is sequestered on CNPW and forced to follow the conventional uptake route of this physical contaminant, accumulating in other tissues. Indeed, reflection analysis highlighted the presence of CNPW aggregates -as clear white masses- at the gills level (Fig. 4 d, g) and into the digestive tract (Fig. 4 e, f, h, i). The uptake was observed in embryos exposed to CNPW either alone or combined with $B(\alpha)P$. TEM analysis localized also the nanometric fraction of CNPW in those organs (Fig. 5), confirming these systems as primary sites for the uptake of CNPW, as described for some NPs in other aquatic models.⁴⁵⁻⁴⁹ The beginning of feeding behavior in zebrafish embryos does not occur at fixed time-point, but only a rough estimate when ingestion of food starts could be established. Certain anatomical conditions have to occur to allow the external feeding: the mouth must have opened sufficiently wide, the digestive tract has to be fully differentiated, the anus must have opened and the swim bladder inflated, to allow the embryos to hunt.⁵⁰ At 96 hpf the occurrence of external feeding is unlikely as the digestive tract is not completely developed and the swim bladder is not inflated.⁵¹ Therefore, at this stage,

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the transit of CNPW in the digestive tract -revealed by microscopy analysis- has been likely due to passive entrance through the mounth.

Our results are in line with other studies describing a carrier role of CNPs for organic pollutants. The gastrointestinal tract, in particular, proved to be a key tissue for the bioaccessibility of organic pollutants adsorbed on CNTs, due to the desorption action of pepsin and bile salts.⁵² Accordingly, in the medaka (*Oryzias latipes*), single walled carbon nanotubes (SWCNTs) associated with PhE were first retained in the digestive tract of fish, and only later, a significant accumulation of PhE was recorded in other organs as liver and brain due to the release of the hydrocarbon from SWCNTs over time.²⁸ Nevertheless a study demonstrated that PhE bound to C_{60} resulted bioavailable and bioaccumulated in junior carp (*Cyprinus carpio*) but was not responsible for toxicity -evaluated as ethoxiresorufin-O-deethylase activity- which was, instead, generated exclusively by the free PhE.⁵³

Effects on genes' transcription and enzymes' activities

In order to assess whether the described interactions could interfere with cellular pathways of individual contaminants, we evaluated the expression of genes related to the cellular stress response and the activity of enzymes involved in the antioxidant response machinery.

The cytochrome P4501A (CYP1A) is the primary enzyme involved in the first phase of biotransformation/detoxification of toxic pollutants, such as PAHs.⁵⁴ The expression pattern of *cyp1a* has been extensively investigated in zebrafish embryos, either under physiological condition or exposed to environmental pollutants.^{55,56} Significant transcription of *cyp1a* mRNA has been observed in zebrafish embryos already at 24 hpf localizing in almost all tissues. The inducibility of *cyp1a* transcription by dioxin-like pollutants has been also established.

An up-regulation of *cyp1a* gene transcription (2.4-, 4.7-, 2.1-folds compared to DMSO) was observed in embryos exposed to all B(α)P concentrations (Fig. 6), confirming the involvement of the gene in the metabolism of B(α)P. As we noticed, the gene induction followed a bell shape trend with the highest up-regulation at B(α)P 6 µg/L. Its low induction at the highest tested concentration suggests that the B(α)P accumulated in the yolk did not undergo metabolic modification. The exposure to all three concentrations of B(α)P did not modulate *sod1* and *sod2* gene transcription, as well as *hsp70* expression (Fig. 6).

A different pattern of gene transcription was observed upon exposure to CNPW alone and combined with $B(\alpha)P$. The profile of genes' modulation suggests that the effects at molecular level mainly mirrored those shown by the physical contaminant rather than by $B(\alpha)P$, albeit with some differences.

CNPW did not affect *cyp1a* expression, which resulted even slightly down-regulated, as well as CNPW + $B(\alpha)P \ 6 \ \mu g/L$ and CNPW + $B(\alpha)P \ 20 \ \mu g/L$ (Fig. 6). On the contrary, an increased amount of *cyp1a* mRNA was observed upon exposure to CNPW + $B(\alpha)P \ 0.2 \ \mu g/L$, albeit with high variability.

A down-regulation of *sod1* and *sod2* was observed in embryos exposed to CNPW alone. Whilst no modulation of *sod1* was shown upon exposure to CNPW combined with $B(\alpha)P$, a lower level of *sod2* mRNA was found in embryos exposed to CNPW + $B(\alpha)P 0.2 \mu g/L$ (0.4- folds vs ctrl) and CNPW + $B(\alpha)P 6 \mu g/L$

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(0.4- folds vs ctrl), while *sod2* expression was restored to background level in embryos exposed to CNPW + $B(\alpha)P \ 20 \ \mu g/L$ (1.0 folds vs ctrl). Again, a lower *hsp70* mRNA level was noticed in embryos exposed to CNPW, CNPW + $B(\alpha)P \ 6 \ \mu g/L$ and CNPW + $B(\alpha)P \ 20 \ \mu g/L$ (0.2-, 0.3-, 0.3- folds vs ctrl), while an upregulation of this gene was found in the CNPW + $B(\alpha)P \ 0.2 \ \mu g/L$ group.

Concerning enzymes activities, exposure to B(α)P affected GST activity following a slight bell shape trend (Fig. 7), since the enzyme resulted significantly induced at 6 µg/L B(α)P (F=144.17; p<0.001), while GST was significantly lowered at the highest B(α)P compared to the corresponding control (p<0.001). GST is a family of phase II detoxification enzymes that catalyze the conjugation of glutathione to a wide variety of endogenous and exogenous electrophilic compounds, including environmental pollutants and oxidative stress by-products.⁵⁷ The increase of GST, which followed a similar trend of *cyp1a* gene, supports an efficient metabolism/excretion of B(α)P out of the cell. A significant increase of GST was observed also in embryos exposed to CNPW alone (p<0.001). On the contrary, a significant inhibition of GST activity was measured in all groups exposed to CNPW contaminated with B(α)P (p<0.001). Such result underlines that the exposure to combined pollutants could hinder the detoxifying capacity of the embryos. The same evidence was also found in zebrafish hepatocytes after C₆₀ and B(α)P exposure.²⁹ Fullerene enhanced B(α)P cellular uptake and decreased GST activity, with consequent induction of cell death.

The oxidative metabolism of pollutants, mediated by CYP1A, could determine the formation of hydroxy radicals and reactive oxygen species (ROS), leading to oxidative stress.⁵⁸ To counteract the rise of ROS, the organisms use diverse arrays of enzymes, such as SOD and CAT. A significant activation of CAT activity was found only in embryos exposed to the highest $B(\alpha)P$ concentration (F=9.57; p=0.03). An increase of SOD activity in zebrafish exposed to increasing $B(\alpha)P$ concentrations, when administered alone, was measured, following a clear bell-shape slope (Fig. 7). The observed increase of SOD and CAT activities confirmed an efficient protective mechanism displayed by the embryos upon exposure to $B(\alpha)P$, which prevent the onset of oxidative stress condition. On the contrary a significant inhibition of CAT was measured in embryos exposed to CNPW alone (p=0.004) and to CNPW + 20 μ g/L B(α)P (p=0.004). CNPW did not affect significantly SOD activity neither alone nor combined with all $B(\alpha)P$ concentrations. Oxidative stress is a widely described toxic effect of NMs, mainly due to ROS overproduction, though the cellular mechanisms are not completely understood, and might vary in relation to different chemico/physical properties of NMs.⁵⁹ In aquatic organisms exposed to NMs either induction or depression of SOD and CAT activities have been reported.⁶⁰⁻⁶³ Therefore, the observed misregulation of key genes/proteins related to the antioxidant response, suggest that CNPW alone and combined with $B(\alpha)P$ might induce toxicity through the imbalance of the antioxidant defense mechanism.

Overall, results highlight that, once entered in the organism, the $B(\alpha)P$ adsorbed on CNPW activates cell pathways in a different way with respect to $B(\alpha)P$ alone. In particular the adsorbed $B(\alpha)P$ overcomes the CYP450-mediated detoxification mechanisms and interferes with transcription of genes related to stress response pathways.

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Cyto-genotoxic effects

The different modulation of genes/proteins involved in the cellular protective mechanisms, elicited by CNPW contaminated with $B(\alpha)P$ with respect to the two pollutants alone, might have reduced the detoxifying capacity of the organism, reflecting in unexpected adverse consequences, such as for instance an increase of cyto-genotoxicity.

We did not found any significant cytotoxic effect at all $B(\alpha)P$ and CNPW experiments with respect to controls (DMSO for $B(\alpha)P$; ZFW for CNPW) (Fig. 8). Besides, MN analysis showed a very low frequency of micronucleated cells (< 3 ‰) and no effect on MN frequency was revealed upon all exposure conditions (F = 1.69; p = 0.12; data not shown).

The exposure to $B(\alpha)P$ determined a significant increase (F=3.46; p<0.01) of DNA fragmentation only at the intermediate concentration (6 µg/L; Fig. 8). Any fixed genetic damage was observed in embryos exposed to all $B(\alpha)P$ concentrations (Fig. 8). The low geotoxicity observed for $B(\alpha)P$, followed a trend similar to *cyp1a* gene modulation. The genotoxic properties of high-molecular-weight PAHs such as $B(\alpha)P$ are commonyl associated with their biotransformation/detoxification.⁶⁴ In particular CYP1A is able to convert PAHs to reactive intermediates which have the ability to interact with DNA, causing adducts.⁶⁵ In line with this evidence, our results suggested that a relationship could be established between CYP1A detoxifying activity and occurrence of primary DNA damage.

Anyhow the observed induction of CYP1A was not enough to produce a significant bioactivation of chemical. Besides, the short time of exposure (72 h) may have contributed to the absence of genotoxic effects, particularly the occurrence of micronucleated cells.

Any genetic injury was detected in embryos exposed to CNPW alone (Fig. 8). To date, the genotoxicity of carbon-based NPs have been investigated mostly on *in vitro* mammalian models,³² but barely assessed *in vivo* on aquatic organisms, and the mechanisms underlying genotoxicity are far to be understood. Any fixed genotoxic effect -measured as MN occurrence- has been observed in the amphibian *X. leavis* erythrocytes, upon exposure to CNTs (10 mg/L) for 12 days.⁶⁶ On the contrary, the occurrence of apoptotic and necrotic cells has been detected in zebrafish embryos exposed to C₆₀ (0.05-0.1-0.2 mg/L) for 12 h, affecting also the embryo development.^{67,68} The genotoxicity of CB (Printex 90) has been investigated in haemocytes of the bivalve *M. galloprovincialis*,⁸ in which the nano-CB increased the release of hydrolytic enzymes and mitochondrial damage, enhancing the onset of pre-apoptotic events. In line with this evidence, the down-regulation of genes related to the cellular stress response (*sod1, sod2, hsp70*) and the reduction of the antioxidant enzyme CAT, highlight the genotoxic potential of CNPW. Nevertheless, the experimental conditions might have hindered the onset of genetic damage as observed for B(α)P. Besides, the increase of GST activity could allow counteracting the occurrence of cyto-genotoxic effects upon single exposure.

Although the exposure to individual contaminants did not induce cyto-genotoxic damage to embryos, when molecules were administered in co-exposure, a significant increase in toxicity was observed. The embryos exposed to CNPW combined with $B(\alpha)P$ showed a dose-dependent reduction of cell viability (F=6.86; p<0.001), up to 25% at the highest $B(\alpha)P$ dose. Despite the observed cytotoxicity, the cell viability remained

higher than 70% in all exposures, which is the threshold required to perform further genotoxicity tests.⁶⁹ Embryos exposed to CNPW combined with all the three doses of $B(\alpha)P$ showed a significantly higher percentage of DNA in comet tail than that found in individuals treated with CNPW alone (about 7% p < 0.01). In addition, the onset of fixed genetic damage with respect to the controls was observed. Embryos treated with CNPW combined with $B(\alpha)P$ showed an increase in the % of apoptotic cells, which resulted significantly higher compared to control and to CNPW alone in CNPW + $B(\alpha)P$ 6 µg/L group (F = 22.47; p < 0.001). Similarly, a higher percentage of necrotic cells was measured in embryos exposed to CNPW + $B(\alpha)P$ 20 µg/L with respect to controls and CNPW alone (F = 5.07; p < 0.001).

These results confirm that the alteration of genes/proteins, related to the cellular stress response observed upon exposure to combined pollutants, reflect in higher cyto/genotoxicity for embryos.

The occurrence of cyto-genotoxic effects in embryos exposed to CNPW contaminated with $B(\alpha)P$ might be due to ROS production that overcomes the scavenging capacity of the antioxidative system, as suggested by the down-regulation of *sod* genes and the significant inhibition of GST. Moreover, the down-regulation of *hsp70* transcription might explain the occurrence of apoptotic events in co-exposed embryos, as this molecular chaperone is actively involved in cellular response related to stress resistance, providing cytoprotective function against apoptosis.⁷⁰

The cellular responses obtained in co-exposure conditions suggested that the $B(\alpha)P$ adsorbed contributes to increase the toxicity of CNPW, producing a greater toxic effect compared to the damage caused by the individual contaminants, as already described for carbon-based NPs combined with PAHs. For instance, the co-exposure of C₆₀ (0.10 to 1 mg/L) and fluoranthene (32-100 mg / L) for 72 hours, increased the DNA fragmentation compared to individual contaminants in *M. galloprovincialis*.⁹ The increase of the genotoxicity might also be related to higher bioavailability and retention of B(α)P, which is translocated inside the embryos by CNPW as shown by confocal images.

Conclusions

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58 59 60 Overall, our results open a new light in the possible relationship of NMs and pollutants in the aquatic environment. In particular, our findings highlighted the interaction with contaminants as key feature of NMs ecotoxicity. In fact the adsorption properties of CNPW for $B(\alpha)P$ suggested that NM act as vehicle of toxic pollutants in the aquatic ecosystems. CNPW is also able to interfere with cell stress/detoxification mechanisms, thereby affecting the organism susceptibility to chemical pollutants. Therefore, single or combination of CNPW and $B(\alpha)P$ induce different cellular metabolic responses respect to single exposure, enhancing the occurrence of cyto-genotoxic effects, affecting different intracellular pathways. Further studies focusing on specific molecular pathways are prompted, to get an in-depth knowledge of the relationship between CNPW and one of the most toxic environmental pollutants. The sorption properties of CNPW towards organic pollutants, pointed out in the present study, arouse concern also in terms of the possible negative effects on human health. Carbon-based nanopowder represents, in fact, one of the main components of PM 2.5 and therefore could contribute to enhance the uptake of airborne noxious contaminants in human,

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Author contribution

C.D.T. and M.P. designed and performed experiments, analysed data and wrote the paper. A.B. is the supervisor and leader of the study. L.D.G. supervised qRT-PCR analysis. A.G. designed and performed qRT-PCR analysis and edited the paper. M.A. designed and performed confocal microscopy analysis. N.S. supervised microscopy analysis. D.M. designed and performed DLS analysis and gave conceptual advice for characterization. L.M. performed confocal microscopy analysis. S.M. and L.P. gave technical support and conceptual advice. C.L.P. gave conceptual advice. A.B., L.D.G., N.S. and C.L.P. jointly conceived the study and edited the paper.

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Table 1. DLS measurements performed on CNPW alone in MILLIQ and CNPW contaminated with growing amounts of $B(\alpha)P$ in zebrafish water: ζ -potential (mV) and mean size distribution (nm).

Sample	ζ-potential mV	Z average nm (size range)		
		2 h	4 h	24 h
CNPW 50 mg/L	-20.7 ± 1.15	356 (271 – 482)	167 (133-224)	73.1 (23-130)
CNPW + B(α)P 0.2 μ g/L	-20.5 ± 3.58	408 (245 - 589)	186 (123 –275)	172 (26-447)
$CNPW + B(\alpha)P 6 \mu g/L$	-20.0 ± 1.59	330 (242 - 405)	231 (78-530)	202 (149-263)
$CNPW + B(\alpha)P \ 20 \ \mu g/L$	-20.1 ± 3.12	419 (171 – 445)	267 (163 - 495)	282 (97-396)

Table 2. Results of GC/MS analyses of supernatant and CNPW fraction after contamination of cleaned CNPW with increasing amounts of $B(\alpha)P$.

Sample	$\mu g B(\alpha) P / L$ water	$\mu g B(\alpha) P / g CNPW$	$\mu g B(\alpha) P / 50 mg CNPW$
$CNPW + B(\alpha)P 0.01 mg/L$	0.02	4.22	0.21
$CNPW + B(\alpha)P 0.1 mg/L$	0.02	111.34	5.57
$CNPW + B(\alpha)P \ 1 \ mg/L$	0.71	344.62	17.23



Characterization of CNPW at SEM and TEM

209x297mm (300 x 300 DPI)



Confocal microscopy showing accumulation of B(a)P (red) in zebrafish fli-embryos (96 hpf) showing the vascular system in green. (a) CTRL (b) DMSO (c) CNPW (50 mg/L) (d) B(a)P 0.2 μ g/L (e) B(a)P 6 μ g/L (f) B(a)P 20 μ g/L (g) CNPW + B(a)P 0.2 μ g/L (h) CNPW + B(a)P 6 μ g/L (i) CNPW + B(a)P 20 μ g/L. Scale bar: 200 μ m.

257x151mm (150 x 150 DPI)

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Quantitative fluorescence analysis of B(a)P accumulated in zebrafish embryos (96 hpf) exposed to CNPW (50 mg/L) and B(a)P (0.2, 6, 20 μ g/L) alone and in combination measured as mean (± SEM) of Grey Level x Area (pixel^2). Different letters indicate significantly different values (one-way ANOVA, Fischer's LSD post hoc test; p < 0.05).

163x111mm (300 x 300 DPI)



Images of cryostate sections showing the uptake of CNPW -indicated by arrows- in gills (A,D,G) and digestive apparatus (B,C,E,F,H,I) of zebrafish embryos. Controls (A-C), CNPW (D-F) and CNPW + B(a)P 20 μ g/L (G-H). DNA in blue (DAPI) (nuclei), musculature in red (phalloidin-TRITC).

129x134mm (150 x 150 DPI)



TEM images showing CNPW indicated by arrows, in gills (A,B) and gastrointestinal tract (C,D) of embryos exposed to CNPW (50 mg/L) (B,D) and CNPW + B(a)P 20 μ g/L (A,C). N = nucleus, sm = somatic muscle, m = mitochondrion, I = lumen, mv = microvilli.

209x297mm (300 x 300 DPI)



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Modulation of genes (cyp1a, hsp70, sod1, sod2) in zebrafish embryos (96 hpf) exposed to CNPW (50 mg/L) and B(a)P (0.2, 6, 20 µg/L) alone and in combination. Results are expressed as fold induction over control (ZFW or DMSO for CNPW alone and contaminated or B(a)P respectively) (± SEM). 18S is used as housekeeping gene. The threshold for biological significance 0.5 < > 2 is pointed out.

218x182mm (150 x 150 DPI)

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Effects of CNPW (50 mg/L) and B(a)P (0.2, 6, 20 μ g/L) alone and in combination on the activity (mean ± SEM) of glutathione S-transferase (GST); catalase (CAT) and superoxide dismutase (SOD) measured in zebrafish embryos (96 hpf) (n = 3; pool of 3 independent experiments). Different letters indicate significantly different values (one-way ANOVA, Fischer's LSD post hoc test, p < 0.05).

254x58mm (300 x 300 DPI)



Cyto-genotoxic effects of CNPW (50 mg/L) and B(α)P (0.2, 6, 20 μ g/L) alone and in combination, measured as mean (± SEM) cell viability, DNA strand breaks and occurrence of apoptotic and necrotic cells in zebrafish embryos (96 hpf) (n = 3; pool of 3 independent experiments). Different letters indicate significantly different values (one-way ANOVA, Fischer's LSD post hoc test; p < 0.05).

209x163mm (150 x 150 DPI)