

1       **“Aerogels of enzymatically oxidized galactomannans from leguminous plants:**  
2       **versatile delivery systems of antimicrobial peptides and enzymes”**  
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48       **Running title:** *Aerogels from oxidized leguminous galactomannans as delivery systems*

## ***Abstract***

We describe aerogels obtained by laccase/TEMPO-oxidation and lyophilization of galactomannans (GM) from fenugreek, sesbania and guar. Enzymatic oxidation of GM in aqueous solution caused viscosity increase up to fifteen-fold, generating structured, elastic, stable hydrogels, presumably due to establishment of hemiacetalic bonds between newly formed carbonyl groups and free hydroxyl groups. Upon lyophilization, water-insoluble aerogels were obtained, whose mechanical properties and porosity were investigated.

Active principles were absorbed into the aerogels from aqueous solutions and, following rinsing, blotting, re-lyophilization, were released in an appropriate medium. The release of the antibiotic polymyxin B was tested against six different Gram-negative bacterial strains, of the antimicrobial peptide nisin against two Gram-positive and of the muraminidase lysozyme against one anaerobic strain. Protease and lipase release in solution from “enzyme loaded” aerogels was monitored by the increase in enzymatic activity.

These biomaterials could represent new versatile, biocompatible “delivery systems” of actives for biomedical and industrial applications.

**Keywords:** *galactomannans; polymyxin; nisin; lysozyme; aerogels; delivery systems.*

**Abbreviations:** *ABTS, 2,2'-azino-bis(3-ethyl-benzothiazoline-6-sulphonic acid); CFU, colony forming unit; CNF, cellulose nanofibrillated; DS, delivery system; EOLFG, enzymatically oxidized lyophilized fenugreek gum; EOLGG, enzymatically oxidized lyophilized guar gum; EOLSG, enzymatically oxidized lyophilized sesbania gum; FG, fenugreek gum; GG, guar gum; GO, galactose oxidase; Lcc, laccase; LFG, lyophilized fenugreek gum; LGG, lyophilized guar gum; LMS, laccase mediator system; LSG, lyophilized sesbania gum; MIP, mercury intrusion porosimetry; SG, sesbania gum; TEMPO, 2,2,6,6-Tetramethyl-1-piperidinyloxy radical.*

## 75 **1. Introduction**

76  
77 Galactomannans (GM) from leguminous plants are widely exploited to generate a considerable  
78 range of derivatives with several practical applications. GM are high molecular weight  
79 polysaccharides found in the seed endosperms of some *Leguminosae* (belonging to the family  
80 *Fabaceae*) where they serve as reserve source for carbon and energy upon germination (Prajapati et  
81 al., 2013). They have a branched polymeric structure composed of a backbone of mannose units  
82 linked by  $\beta$ -1,4 glycosidic bonds with side units of galactose bound to mannose by  $\alpha$ -1,6 glycosidic  
83 bonds. The average ratio of galactose to mannose (Gal: Man) is variable, depending on the plant  
84 species, and ranges from 1:4.5 in cassia (*Cassia tora*) to 1:1 in fenugreek (*Trigonella foenum-*  
85 *graecum*) (Crescenzi et al., 2004; Daas, Schols & de Jongh, 2000; Daas, Grolle, van Vliet, Schols &  
86 de Jongh, 2002; Daniel, Whistler, Voragen & Pilnik, 1994; Liyanage, Abidia, Auldb & Moussa,  
87 2015; McCleary, Clark, Dea & Rees, 1985; Merlini, Boccia, Mendichi & Galante, 2015;  
88 Sittikijyothin, Torres & Gonçalves, 2005; Wei et al., 2015). The species-dependent monomers ratio  
89 makes them, to various extents, soluble in water at different temperatures, chemically/biochemically  
90 quite reactive and flexible in application (Cheng, Prud'homme, Chick & Rau, 2002).

91 GM and their derivatives are used as rheology modifiers, thickening and suspending agents in food,  
92 feed and manufacturing industries (Mathur, 2011). They find application as excipients and co-  
93 formulants in the biomedical field, such as in pharmaceutical formulations of tablets and in orally  
94 controlled drug delivery systems (ODDS), but also as binders, disintegrants, suspending,  
95 thickening, gelling, stabilizing and protective agents, to add cohesiveness to drug powder, as they  
96 are susceptible to microbial degradation in the large intestine (Meghwa & Goswami, 2012). GM are  
97 also increasingly consumed as dietary fibers with atoxic bioactivities, to lower calories intake and  
98 for weight reduction, to control blood glucose, cholesterol and insulin levels, to reduce the risks of  
99 heart diseases and colon cancer, as texture modifiers and stabilizers in “specialty” foods (Murthy,  
100 Moorthy, Prabhu & Puri, 2010).

101 In the present paper, we describe the production of aerogels from enzymatically-oxidized,  
102 lyophilized (EOL) sesbania and guar gums, following a protocol similar to the one previously  
103 reported for fenugreek gum by Rossi et al. (2016). They are referred to as: EOLSG (for sesbania  
104 gum), EOLGG (for guar gum) and EOLFG (for fenugreek gum). **Because of their biodegradability,**  
105 **biocompatibility, high surface area and porosities with open pore structures, nanostructured**  
106 **aerogels, particularly from polysaccharides (e.g., starch, cellulose, alginates), constitute an**  
107 **emerging platform as “delivery systems” (DS) of active principles, such as drugs, peptides and**  
108 **enzymes (Ulker & Erkey, 2014). However, to our knowledge, GM-based aerogels DS have not yet**

109 **been reported by others.** We have absorbed into these GM aerogels the following **model** actives  
110 from aqueous solutions: the antibiotic polymyxin B; the antimicrobial peptide nisin; the enzymes  
111 lysozyme, protease and lipase, followed in all cases by repeated cycles of water rinsing, to eliminate  
112 surface-absorbed compounds, dry blotting and re-lyophilization of the hydrogels to generate  
113 “loaded” aerogels. The gradual release of the incorporated actives was monitored either by  
114 measuring enzyme activity in solution or in Petri dish on agar culture medium, seeded with different  
115 bacterial strains, by evaluating the inhibition halo of cell growth.

116 Polymyxins, a group of polypeptide antibiotics that consists of 5 chemically different compounds  
117 (polymyxins A–E), are secondary metabolites produced by the soil bacterium *Paenibacillus*  
118 *polymyxa*, were discovered in 1947 (Velkov, Thompson, Nation, & Li, 2010), but only polymyxin  
119 B (PMB) and polymyxin E (colistin) have been used in clinical practice (Payne, Gwinn, Holmes &  
120 Pompliano, 2007; Talbot et al., 2006). They share a common primary sequence of a cationic cyclic  
121 decapeptide linked to a fatty acid chain through an alfa-amide linkage (Li, Rayner & Nation, 2007),  
122 with a molecular weight of 1,750 Da, the only difference being at position 6, which is occupied by  
123 D-Phe in PMB and D-Leu in colistin. Polymyxins have been used against Gram-negative bacteria  
124 responsible for nosocomial infections until the early seventies when they were abandoned, because  
125 of their human nephro- and neuro-toxicity. More recently, there has been a revival of interest in  
126 polymyxins considering the emergence of multidrug resistant (MDR) Gram-negative bacteria to  
127 almost all classes of antibiotics. Polymyxin B is active against MDR bacteria, such as *Pseudomonas*  
128 *aeruginosa*, *Acinetobacter baumannii* and *Klebsiella pneumoniae*, also called “superbugs” (Falagas,  
129 Kasiakou & Saravolatz, 2005; Velkov et al., 2010). **Thus, innovative, biocompatible,**  
130 **polysaccharide-derived DS could contribute to a more rational and controlled clinical management**  
131 **of PMB.** We have absorbed polymyxin B from an aqueous solution onto plugs carved from wafers  
132 of EOLFG, EOLSG and EOLGG, followed by rinsing and re-lyophilization to form “loaded”  
133 aerogel plugs and shown its release based on the inhibition halo of cell growth of: *P. aeruginosa*,  
134 *Serratia marcescens*, *Escherichia coli*, *Salmonella enterica* serovar Typhimurium, *Hafnia alvei*,  
135 *Enterobacter cloacae*. Controls, i.e., non “loaded” aerogel plugs, had no effect on cell proliferation,  
136 while a positive control of free PMB was always present in a separate well of each experimental  
137 Petri dish.

138 Nisin is a natural antimicrobial peptide produced by *Lactococcus lactis*, composed of 34 amino  
139 acids, some of which have a D rather than L configuration (de Arauz, Jozala, Mazzola, & Penna,  
140 2009). It inhibits Gram-positive bacteria belonging to genera such as: *Lactococcus*, *Streptococcus*,  
141 *Staphylococcus*, *Micrococcus*, *Pediococcus*, *Lactobacillus*, *Listeria* and *Mycobacterium*, and also  
142 vegetative cells and spores of *Bacillus* and *Clostridium* (Sahl, Jack, & Bierbaum, 1995). If

143 combined with a chelating agent (such as EDTA), nisin is able to destabilize the bacterial outer  
144 membrane (Vaara, 1992), thus can also be effective against Gram-negative bacteria like *E. coli* and  
145 *P. aeruginosa* (Bozariis & Adams, 1999; Fang & Tsai, 2003). In 2006, the European Food Safety  
146 Authority approved the use of nisin as an antimicrobial food additive (E234). Currently in the EU,  
147 the use of nisin is allowed in: semolina and tapioca puddings, in ripened and processed cheese, in  
148 clotted cream, mascarpone and pasteurized liquid egg products, at concentrations ranging from 3 to  
149 12.5 mg/L (2010/69/EU). It is also used in combination with antibiotics to prevent nosocomial  
150 infections caused by drug-resistant bacteria (Tong et al., 2014). **A control release of nisin into the**  
151 **medium could contribute to its long term stability and effectiveness.** Following the same protocol as  
152 with PMB, described in details in Methods, release of nisin from the three aerogels was evaluated  
153 on culture agar Petri dish from the growth inhibition of the Gram-positive *Enterococcus faecalis*  
154 and *Clostridium tyrobutyricum*, the latter under anaerobic conditions.

155 Finally, a few enzymes were also studied for absorption and release from the three GM aerogels.  
156 Indeed, enzymes often show low stability in industrial applications under storage or use conditions,  
157 and in free form can lead to sensitization and allergies of operators and users. Three different  
158 commercial enzymes of great practical importance were selected: egg lysozyme, a commercial  
159 protease and a lipase.

160 Lysozyme (LSZ, EC 3.2.1.17) is a hydrolytic enzyme obtained from hen's egg white (HEW),  
161 composed of 129 amino acids and with muraminidase activity (Silvetti et al., 2010). It has the  
162 ability to hydrolyze the  $\beta$ -1,4-glycosidic bond between N-acetyl-muramic acid and N-acetyl-  
163 glucosamine in the peptidoglycans of Gram-positive bacteria (Brasca et al., 2013). It is active on  
164 bacterial species, such as: *Staphylococcus aureus*, *Micrococcus lysodeikticus*, *Listeria*  
165 *monocytogenes* and *L. innocua*, *Bacillus cereus* and *B. stearothermophilus*, *Clostridium*  
166 *thermosaccharolyticum* and *C. tyrobutyricum*, etc. (Ávila, Gómez-Torres, Hernández & Garde, 2014;  
167 Takahashi et al., 2011). Lysozyme is non-toxic to humans and is authorized by EU legislation (EU  
168 No. 1129/2011) as a food additive (E1105) at “*quantum satis*” in ripened cheeses and milk products  
169 to prevent butyric acid fermentation which causes the “late blowing” of cheese wheels (Brasca et  
170 al., 2013).

171 Proteases constitute a fundamental group of hydrolytic enzymes, mainly employed as detergent  
172 additives, but also in several other applications. (Fu et al., 2015; Gupta, Beg & Lorenz, 2002; Karn  
173 & Kumar, 2015; Souza et al., 2015). Proteases can be unstable, as they are prone to autolysis (thus  
174 they are commercially formulated in high-glycol containing gels and reversibly inhibited with boron  
175 salts, whose toxicology is however under scrutiny, see TOXICOLOGICAL PROFILE FOR  
176 BORON; U.S. DEPARTMENT OF HEALTH AND HUMAN SERVICES; Public Health Service

177 Agency for Toxic Substances and Disease Registry; November 2010). Alternatives to boron are  
178 widely sought to be used as reversible protease inhibitors. Not least, protease dust inhalation can  
179 lead to irritation and pulmonary harm in humans, such as workers and users.

180 Lipases are increasingly used for natural fat degradation, esterification, transesterification and in the  
181 production of high value compounds, such as biopolymers, biodiesel, enantiopure pharmaceuticals,  
182 flavors and agrochemicals (Borrelli & Trono, 2015; Jaeger & Eggert, 2002; Reetz, 2002;). It would  
183 thus be valuable, for both basic studies and applications, to have a readily available, pure and stable  
184 lipase entrapped and released from a solid support before use.

185 Therefore, a great deal of research has been performed on the encapsulation or immobilization of  
186 hydrolytic enzymes, e.g.: in gel beads (Verma, Puri & Barrow, 2015) and nanoparticles (Lin, Chen  
187 & Liu, 2016; Vaghari et al., 2016;) **to improve their long term stability against inactivation.**

188 In this work, we describe the absorption, retention and release in active form of lysozyme, protease  
189 and lipase from GM aerogel plugs. Released lysozyme activity was demonstrated microbiologically  
190 by actual growth inhibition of *C. tyrobutyricum* (i.e., whose vegetative cells and spores represent its  
191 target when added to milk during cheesemaking) under anaerobic conditions. Protease and lipase  
192 release from similarly “loaded” and re-lyophilized aerogel plugs was estimated from the increase in  
193 their respective enzymatic activity, as a function of time, of solutions in which the plugs were  
194 incubated.

195 To better characterize these materials, compressive tests were also performed and their porosity was  
196 determined by Hg absorption under high pressure.

197 From all of the above, we suggest that aerogels made from enzymatically oxidized and lyophilized  
198 GM from leguminous plants, could represent very innovative, versatile, biocompatible delivery  
199 systems of various active principles, with potential applications in the biomedical field, as well as in  
200 food packaging, feed and industrial products.

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202

## 203 **2. Materials and Methods**

### 204 **2.1. Materials**

205 Laccase from *Trametes versicolor*, in powder form, was a generous gift of Amano Ltd, UK, and  
206 Amano, Japan, with a measured activity of 1600 U/g on ABTS as substrate. It was dissolved with  
207 mild stirring in MilliQ water.

208 The strains of *E. coli* (ATCC 8739) and *S. Typhimurium* (ATCC 14028) were from the American  
209 Type Culture Collection (U.S.A.). The other strains were obtained from the bacterial collection of the  
210 ISPA-CNR (Institute of Science of Food Production of the Italian National Research Council, Milan,

211 Italy) and included: the Gram-positive *C. tyrobutyricum* (IN15b) and *E. faecalis* (VS485); the Gram-  
212 negative *P. aeruginosa* (PS20), *S. marcescens* (S91), *H. alvei* (PS58) and *E. cloacae* (PS25). All  
213 these strains were isolated from contaminated dairy samples, identified through partial 16S rRNA  
214 sequencing using the universal primer set p8FPL and p806R (McCabe, Zhang, Khan, Mason &  
215 McCabe, 1995). The Gram-negative strains were also subjected to *rpoB* sequencing (Sajben,  
216 Manczinger, Nagy, Kredicsa & Vágvölgyia, 2011). Sequences were analyzed by NCBI BLAST  
217 search (<http://www.ncbi.nlm.nih.gov/BLAST>, Altschul, Gish, Miller, Myers & Lipman, 1990).

218 Lysozyme hydrochloride was supplied by Sacco (Cadorago (CO), Italy). Nisin (with a potency of  $10^6$   
219 IU/g) and polymyxin B were purchased from Sigma-Aldrich (St. Louis, MO, USA).

220 The industrial protease (Deterzyme® APY 560 in granular form from *Bacillus alcalophilus*) was from  
221 Enmex (Tlalnepantla, Mexico). As for the lipase, preliminary experiments were performed with the  
222 Type VII enzyme from *Candida rugosa* (Sigma lot 107H1024) in powder form and later with the  
223 commercial lipase (Greasex® Ultra L from *Candida lipolitica*) in liquid form (from Novozymes,  
224 Bagsvaerd, DK).

225 TEMPO and all other chemicals were from Sigma-Aldrich or Fluka.

226

## 227 **2.2. Laccase assay.**

228 Laccase activity was determined using as substrate 2.48 mM ABTS in 100 mM sodium acetate at pH  
229 5 (Niku-Paavola, Karhunen, Salola & Raunio et al., 1988).

230 One laccase unit is defined as the amount of enzyme that catalyzes the oxidation of one  $\mu$ mole of  
231 ABTS in one min at 25 °C and pH 5.

232

## 233 **2.3. Purification and viscosity measurements of GM solutions.**

234 Non purified gum powder from guar (GG), sesbania (SG) and fenugreek (FG), with a Brookfield  
235 viscosity at 1% (w/v) in aqueous solution at 20 rpm and 25°C of 5000, 3000 and 1500-2500 mPa\*s,  
236 respectively, were from a commercial source and kindly supplied by Lamberti S.p.A. Actual GM  
237 content of unpurified gum varied between 76 and 80% (w/w), the remaining components being  
238 residual aleuronic proteins, seed coat residues, low mol wt sugars, ashes. Before performing  
239 oxidation and other experiments, all GM were purified by dispersion (at 10% w/w) in a 3:7 solution  
240 of H<sub>2</sub>O/ethanol, by stirring at room temperature for 30 min, followed by vacuum filtration. The  
241 recovered GM were dispersed (at 10% w/w) in acetone, stirred as before, and finally recovered by  
242 vacuum filtration. Before use, they were oven-dried at 60°C overnight. Polysaccharide yield of this  
243 procedure was 85-90% (w/w), while the residual 10-15%, composed of proteins and other minor  
244 components, was discarded.

245 “Purified” GG and FG were dissolved in MilliQ water at room temperature at 1200 rpm with an  
246 IKA overhead stirrer for 30 min. SG was firstly dispersed using an IKA Ultraturrax in MilliQ water  
247 at room temperature for a min and finally at 1200 rpm with an IKA overhead stirrer for 30 min. The  
248 solution was kept standing overnight at room temperature without stirring before any further  
249 manipulation. Compared to “non purified” GM, viscosity increased by about 10-15% at equal gum  
250 concentration in water, indicating a higher polymer amount in the “purified” material. GM solutions  
251 for all experiments were prepared the same way.

252 Viscosity measurements were performed in a volume of 300-400 ml in a beaker at room temperature  
253 using a Brookfield DV-I Prime, at 20 rpm, mounted with the appropriate spindle.

254

#### 255 **2.4. Preparation of the aerogels.**

256 TEMPO-mediated, laccase oxidation of FG, SG and GG was carried out following the procedure  
257 previously described by Merlini et al. (2015) with slight modifications. Purified FG, SG and GG  
258 were dissolved, as described in § 2.3, at 1.075% (w/w) in 100 ml of MilliQ water at room  
259 temperature for 30 min and the solution was kept standing overnight at room temperature. The  
260 mediator TEMPO was added to a final concentration of 0.64 mM (10 mg); laccase, 160 U/g<sub>(GM)</sub> (eq.  
261 to about 110 mg), was dissolved in 2 ml of MilliQ water, centrifuged for 4 min at 4000 rpm and the  
262 supernatant added to the solution. The reaction was continued for 3 h at 35°C, with constant  
263 mechanical stirring at 500 rpm, after which Brookfield viscosity was measured and the mixture was  
264 left standing at room temperature. Viscosity was measured again after 6 and 24 h from the start of the  
265 reaction.

266 To obtain the aerogels, whole preparations of enzymatically oxidized FG, SG or GG hydrogels were  
267 distributed either into 24 well plates, to obtain 16-20 x 12-16 mm cylindrically shaped samples (see  
268 also Rossi et al., 2016), or in Petri dishes (90x12 mm), to give a flat, round aerogel wafer of about  
269 80x5 mm, and frozen for 12 h at -80°C, followed by freeze-drying at -55 °C for 48 h (these materials  
270 are referred to as EOLFG, EOLSG and EOLGG, respectively).

271

#### 272 **2.5. Preparation of actives solutions.**

273 A 1 g/L of polymyxin B stock solution was prepared in sterile distilled water, filter-sterilized and  
274 stored at 4°C. A working solution of 300 mg/L was prepared by diluting the stock solution in  
275 distilled, sterile water. Nisin working solution (50,000 IU/ml) was prepared by dissolving the  
276 appropriate amounts of powder in sterile 0.02 N HCl, sterilized by filtration through a 0.20 µm  
277 membrane filter (Corning, USA) and stored at 4 °C. Lysozyme powder was dissolved in distilled,  
278 sterile water to give a stock solution of 2 g/L, which was freshly prepared before each set of



279 experiments and filter-sterilized as before. A lysozyme working solution of 1.25 g/L was prepared by  
280 diluting the stock solution with distilled, sterile water.

281 Before use, 100 mg of protease were dissolved in 1 ml of MilliQ water and centrifuged at 11,200 x g  
282 at RT for 5 min, to eliminate insoluble material present in the formulation, the pellet was discarded  
283 and the “loading” experiments were carried out with the supernatant.

284 In the case of lipase, 125 mg of the *C. rugosa* lipase in powder form were likewise dissolved in 1 ml  
285 of MilliQ water and centrifuged at 11,200 x g at RT for 5 min, the pellet was discarded and the  
286 supernatant used for “loading” experiments. The commercial lipase from *C. lipolitica*, in liquid form,  
287 was used as such.

288 Protein concentration was measured by the Bradford assay (Bradford, 1976) and turned out to be:  
289 33 mg/g for the protease, about 5 mg/g for the *C. rugosa* lipase and 23 mg/ml for the *C. lipolitica*  
290 lipase.

291

## 292 **2.6. Absorption of the active principles in the aerogels.**

293 Plugs (Ø 8 mm) of EOLFG (from fenugreek gum), EOLSG (from sesbania gum) and EOLGG (from  
294 guar gum) were carved with a cork borer as described by Rossi et al. (2016). Polymyxim B, nisin,  
295 lysozyme, protease or lipase were absorbed unto the aerogels by immersing one to three plugs for 1 h  
296 at room temperature in an Eppendorf tube containing 0.5-1.0 ml of solution of active at a specified  
297 concentration. The plugs were then rinsed three times in 1 ml of distilled, sterile water and blotted on  
298 UV-sterilized filter paper to remove eventual active solution present on the surface. The “theoretical”  
299 amount of active absorbed was calculated from the weight difference of the plug(s) (initial weight: 7-  
300 9 mg), measured on each plug before and after immersion in solution, which gave an average weight  
301 increase of 13-15-fold.

302 Protease and lipase protein concentration in the incubation solutions were further checked with the  
303 Bradford assay, before and after absorption of the enzymes as explained above, and found to be  
304 equal. In all cases, the “loaded” hydrogels were re-lyophilized in order to obtain the corresponding  
305 “loaded” aerogels to be tested.

306 The effect of lyophilization on lipase and protease was evaluated by measuring specific activity of  
307 both enzymes after freeze-drying of their respective aqueous solutions, re-dissolution in water and  
308 was negligible.

309

## 310 **2.7. Bacterial strains and culture conditions**

311 All bacterial strains were routinely grown under standard conditions. Working cultures of *P.*  
312 *aeruginosa*, *S. marcescens*, *E. coli*, *S. Typhimurium*, *H. alvei* and *E. cloacae* were propagated in

313 Brain Heart Infusion broth (Scharlau Microbiology, Barcelona, Spain) and incubated at 30 °C (at 37  
314 °C for *E. coli* and *S. Typhimurium*) for 18 h. *C. tyrobutyricum* was cultured anaerobically  
315 (Anaerocult A Merck, Darmstadt, Germany) in Reinforced Clostridial Medium (VWR Chemical,  
316 Leuven) at 37 °C for 48 h. *E. faecalis* was grown aerobically in M17 broth (Biolife Italiana, Milano,  
317 Italy) at 37 °C for 18 h. Before each set of experiments, all enriched cultures were diluted in Ringer  
318 solution (Scharlau Microbiology) to reach a concentration of 10<sup>5</sup> CFU/ml. For solid media, 15 g/L  
319 agar (Scharlau Microbiology) was added to the broth before sterilization.

320

## 321 **2.8. Evaluation of release and activity of active principles**

322 The detection of polymyxin B, nisin and lysozyme released from the aerogels was assessed by a  
323 modified method described by Rossi et al. (2016). For anti-bacterial activity, the poured plate method  
324 was used. Briefly, each Petri dish was layered with a selective agar medium, seeded with the bacterial  
325 inoculum at a concentration of 10<sup>5</sup> CFU/ml. After solidification of the agar, equidistant wells (Ø 8  
326 mm) were carved with a sterilized cork borer and filled with: i) different concentrations of the free  
327 active; ii) plugs of “loaded” EOLFG, EOLSG or EOLGG containing the appropriate amount of  
328 active; iii) a control of “unloaded” EOLFG, EOLSG or EOLGG plug. Inoculated plates were  
329 incubated at 30 °C, or 37 °C in the case of *E. coli*, *S. Typhimurium* and *C. tyrobutyricum*, for 24 h.  
330 Release of actives from aerogels was evaluated from the inhibition halo zones formed around each  
331 well in comparison to the control aerogel and to the diffusion of the free active substance. All manual  
332 operations were performed under a vertical laminar flow hood in sterile conditions and experiments  
333 were run in duplicate.

334 Release of protease or lipase in water was evaluated by withdrawing aliquots of the incubation  
335 solution and determining enzyme activity.

336 Proteolytic activity was calculated from the increase in absorbancy at 440 nm in a 1.0 ml reaction  
337 volume composed of 0.25% w/v azocasein, 0.3 mM calcium chloride, 25 mM Trizma base (pH 8) for  
338 10 min at 40°C. The reaction was stopped by adding 1.0 ml of 20% trichloroacetic acid, followed by  
339 centrifugation at 11,200 x g at RT for 5 min . Protease activity is expressed as Δ O.D. at 440 nm/10  
340 min/ml and one Unit of protease activity is the amount of enzyme causing a Δ O.D of 1.0 under those  
341 conditions.

342 Lipolytic activity was determined by measuring initial linear absorbance increase at 410 nm produced  
343 by p-nitrophenol upon hydrolysis of 1 mM p-nitrophenyl laurate (dissolved in isopropanol) in 10 mM  
344 ammonium acetate pH 7.2 and 0.5 % Triton X-100, in 1.0 ml reaction volume (Santambrogio et al.,  
345 2013). Lipase activity is expressed as Δ O.D. at 410 nm/min/ml and one Unit of lipase activity is the  
346 amount of enzyme causing a Δ O.D of 1.0 under those conditions.

347 All measurements were performed in triplicate.

348

349

350

### 351 **2.9 Determination of pore size**

352 Total porosity, average and modal pore size, pore size distribution, and bulk density of the aerogels  
353 were measured by mercury intrusion porosimetry (MIP), which consists in forcing mercury into the  
354 sample pores at increasing pressures. The output data is the volume of mercury intruded into the  
355 sample as a function of the pressure applied, which is inversely proportional to the pores size  
356 (Santos et al., 2015). Pascal 140 coupled with Pascal 240 porosimeters (Thermo Fisher Scientific,  
357 Waltham, MA) were used for the analysis. Pascal 140 module allows to reach low pressures (up to  
358 400 kPa), thereby measuring only large pores (macropores). Pascal 240 reaches pressures up to 200  
359 MPa, thus allowing measurement of the porosity into the mesopore region. Cylindrical aerogel  
360 corks weighing  $40 \pm 3$  mg were used in the analyses. To overcome the potential limitations  
361 associated to MIP (Majda et al., 2016) freeze-dried samples were placed in a desiccator containing  
362  $\text{CaCl}_2$  for 3 weeks to remove residual moisture. The residual moisture content measured with a  
363 moisture analyzer HB43-S (Mettler Toledo, Greifensee, Switzerland) was as low as  $0.5 \pm 0.02$  %.  
364 During analysis, air inside the samples was removed by three degassing cycles. Mercury intrusion  
365 was performed according to the PASCAL (Pressurization with Automatic Speed-up by Continuous  
366 Adjustment Logic) system, which minimizes crushes and deformations of the samples. Pore  
367 diameter was calculated according to the Washburn equation (Washburn, 1921) assuming a  
368 cylindrical shape of the pores (Webb and Orr, 1997), as in (1):

$$369 \quad P = -2\sigma \cos \frac{\theta}{r} \quad (1)$$

370 where  $P$  is the applied pressure,  $r$  is the radius of the pore,  $\sigma$  is the surface tension of the mercury  
371 and  $\theta$  is the contact angle between mercury and polymer. A surface tension of 480 mN/m and a  
372 contact angle of  $140^\circ$  (Kim & Chu, 2000; Joshi et al., 2015) were used for all measurements.  
373 SOL.I.D (SOLver of Intrusion Data) software was used for data elaboration.

374

### 375 **2.10 Determination of mechanical features.**

376 Cylindrically shaped samples ( $\sim 20$  mm height  $\times$  10 mm diameter) of both pristine non-oxidized,  
377 lyophilized gums (LFG, LSG, and LGG) and the corresponding laccase-oxidized, lyophilized  
378 aerogels (EOLFG, EOLSG, and EOLGG) were evaluated by a compressive test according to the  
379 procedure described by Deszczynski, Kasapis & Mitchell (2003) and more recently applied by

380 Rossi et al. (2016). Briefly, two consecutive cycles of compression were performed with a  
381 dynamometer (mod. Z005, Zwick Roell, Ulm, Germany) fitted with a 100 N load cell and  
382 connected to two plates (30 mm diameter), placed at a distance of 22 mm apart. Each compression  
383 cycle accounted for a maximum deformation of the sample of 2 mm, at a crosshead speed of 2 mm  
384  $\times s^{-1}$ . Stress–strain and force–time plots allowed for the calculation of compressive modulus,  
385 maximum compressive force, cohesiveness, springiness, and resilience using the software  
386 TestXpert V10.11 Master (see also Ghafar et al., 2015). All tests were carried out at  $23 \pm 0.5$  °C and  
387  $40 \pm 2.5$  % relative humidity (RH). At least ten replicates were performed with each sample.  
388 Statistical differences between mean values were determined by Student’s *t*-test, with a significance  
389 level ( $p$ ) < 0.05, using Statgraphic Plus 4.0 software.

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391

### 392 **3. Results and Discussion**

393

#### 394 **3.1 Release from aerogels and activity of polymyxin B against Gram-negative bacteria.**

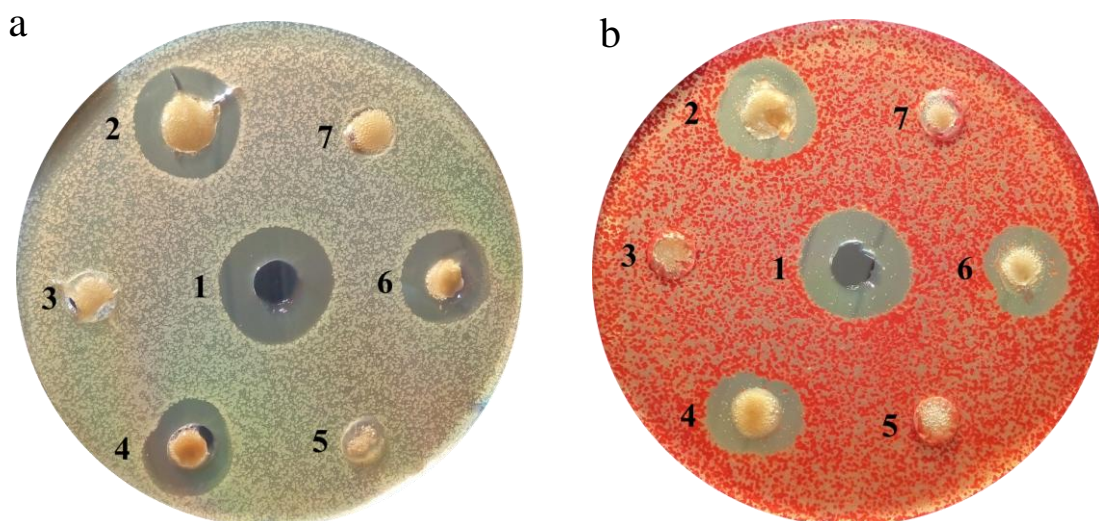
395 The properties of EOLFG, EOLSG and EOLGG aerogels as delivery systems were firstly  
396 investigated with the antibiotic polymyxin B against six Gram-negative bacterial strains. All the six  
397 bacterial species selected are responsible for nosocomial infections. Indeed, *P. aeruginosa* is one of  
398 the major multidrug-resistant (MDR) bacterial species and almost all the currently available  
399 antibiotics have no effect on it, with the exception of polymyxins (Velkov et al., 2010). Therefore,  
400 since the mid nineties, the use of polymyxins has seen a renewed interest, in spite of their suspected  
401 nephrotoxicity and neurotoxicity. Thanks to the wide activity spectrum of polymyxin B, we were  
402 able to test five other bacterial species of *Enterobacteriaceae*, namely: *S. marcescens*, *E. coli*, *S.*  
403 *Typhimurium*, *H. alvei* and *E. cloacae*; all responsible for nosocomial infections that can lead to life-  
404 threatening bacteremia (Alexopoulou et al., 2016; Kim et al., 2015; Steinbach, Töpper, Adam, &  
405 Kees, 2015).

406 The activity of polymyxin B was firstly demonstrated by cell growth inhibition of *P. aeruginosa*.

407 Fig.1a shows the inhibition zone by polymyxin B released from “loaded” plugs of :EOLFG (2),  
408 EOLSG (4) and EOLGG (6) on live cells of *P. aeruginosa*.

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413 **Fig. 1-a,b-** Inhibition by polymyxin B released from aerogels of Gram-negative bacterial cell growth.  
414 Central well (1), 30  $\mu\text{g}$  (equivalent to 22 nanomoles) of free polymyxin B. Lateral wells: plugs of  
415 loaded EOLFG (2), control EOLFG (3); loaded EOLSG (4), control EOLSG (5); loaded EOLGG (6),  
416 control EOLGG (7). Fig. 1a: *P. aeruginosa*. Fig. 1b: *S. marcescens*.

417

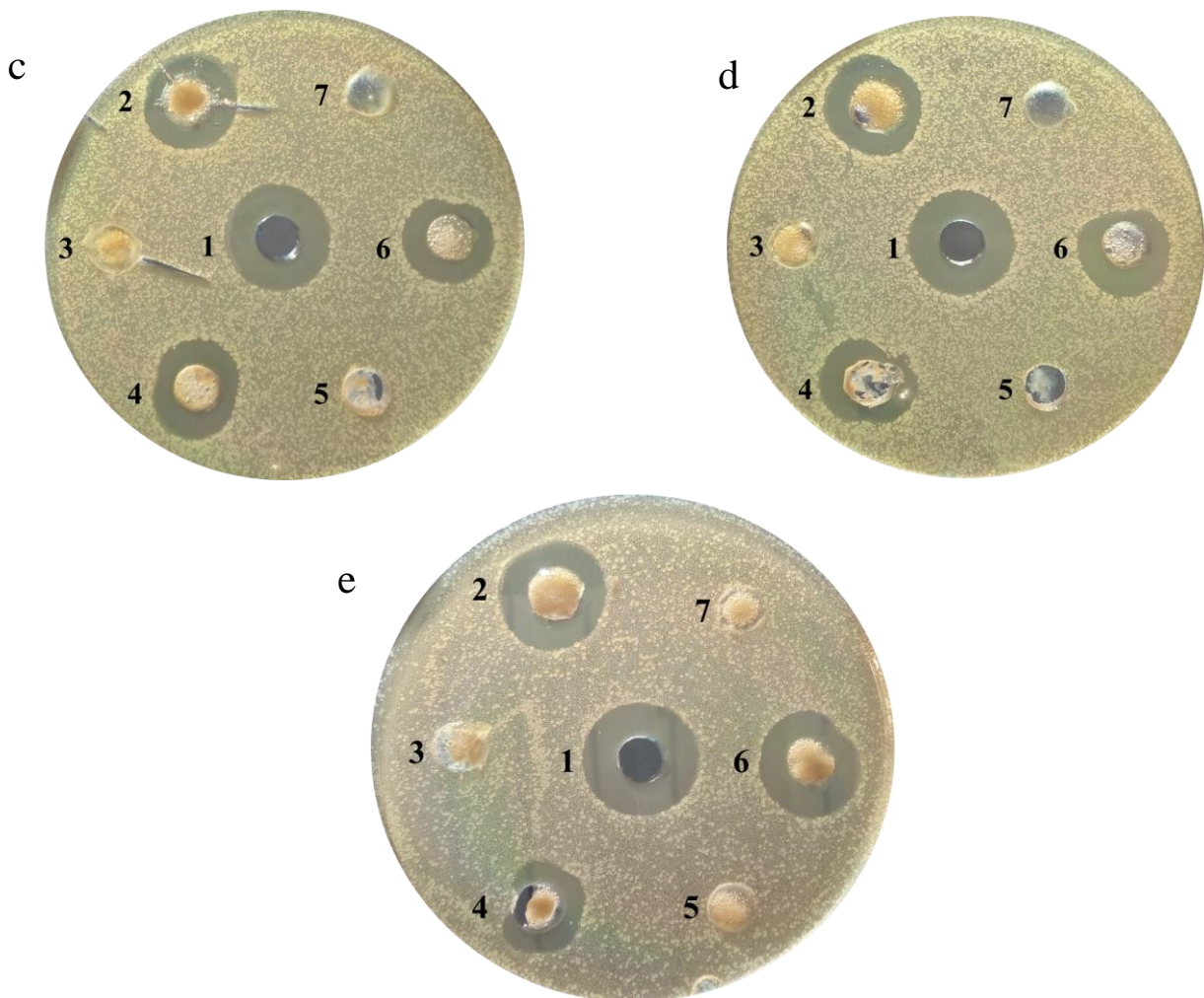
418 The central well (1) was filled with 100  $\mu\text{l}$  of 300 mg/l of polymyxin B (equivalent to 30  $\mu\text{g}$  or 22  
419 nanomoles of active substance). Control plugs of “unloaded”: EOLFG (3), EOLSG (5) and EOLGG  
420 (7) were fitted in the corresponding wells. The amount of incorporated polymyxin B in aerogel plugs  
421 was calculated from the individual average weight increase, performed in triplicate, after 1 h  
422 immersion in a polymyxin B solution, rinsing, blotting, and was estimated to be  $28.9 \pm 8.3 \mu\text{g}$  for  
423 EOLFG,  $23.7 \pm 2.2 \mu\text{g}$  for EOLSG and  $32.4 \pm 6.8 \mu\text{g}$  for EOLGG.

424 As detailed in § 2.6, “loaded” plugs were rinsed three times in water in order to remove any  
425 polymyxin B loosely absorbed on the surface of the aerogels. 100  $\mu\text{l}$  of each rinse water was  
426 evaluated for cell growth inhibition on a similarly seeded Petri dish. While the first rinse water  
427 showed a very slight inhibition halo, the third one had none (see Fig. 1.S in “Supplementary  
428 Material”). Therefore, we conclude that the large halos around the plugs of EOLFG, EOLSG and  
429 EOLGG in Fig. 1a are largely or exclusively due to the release of polymyxin B absorbed and retained  
430 within the aerogels.

431 A similar experiment was carried out on a strain of *S. marcescens*, a species recognized to cause  
432 human clinical diseases, with a multidrug-resistance profile (Zavascki, Goldani, Li & Nation, 2007).  
433 Fig 1b shows the inhibition zone by polymyxin B released from “loaded” plugs of: EOLFG (2),  
434 EOLSG (4) and EOLGG (6) of *S. marcescens* cell growth. As for *P. aeruginosa*, the third rinse water  
435 of all plugs developed no halos (see Fig 2.S in “Supplementary Material”).

436 Following the same protocol as above, we have evaluated four other Gram-negative bacterial  
437 species involved in nosocomial infections, that can be fatal, mostly if infected patients are elderly,  
438 young, or people with depressed immune systems (Alexopoulou et al., 2016; Kim et al., 2015;  
439 Steinbach et al., 2015). The bacterial strains were the following: *S. Typhimurium* (Fig. 1c), *E. coli*  
440 (Fig. 1d), *E. cloacae* (Fig. 1e) and *H. alvei* (Fig. 3.S in Supplementary Material). All strains  
441 analyzed were susceptible to polymyxin B cell growth inhibition, which was absorbed and released  
442 from the aerogel plugs.

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449 **Fig. 1-c-e-** Inhibition by polymyxin B released from aerogels of Gram-negative bacterial cell growth.  
450 Central well (1), 30 µg (equivalent to 22 nanomoles) of free polymyxin B. Lateral wells: plugs of  
451 loaded EOLFG (2), control EOLFG (3); loaded EOLSG (4), control EOLSG (5); loaded EOLGG (6),  
452 control EOLGG (7). Fig. 1c: *S. Typhimurium*. Fig. 1d: *E. coli*. Fig. 1e: *E. cloacae*.

453

454 It is difficult to make a quantitative evaluation of polymyxin B released from Figs.1, a-e and Fig.  
455 3.S. However, by comparing the diameters of the halos around the “loaded” plugs to the central  
456 wells filled with the free active substance, the amounts of polymyxin B diffused appear to be fairly  
457 equal.

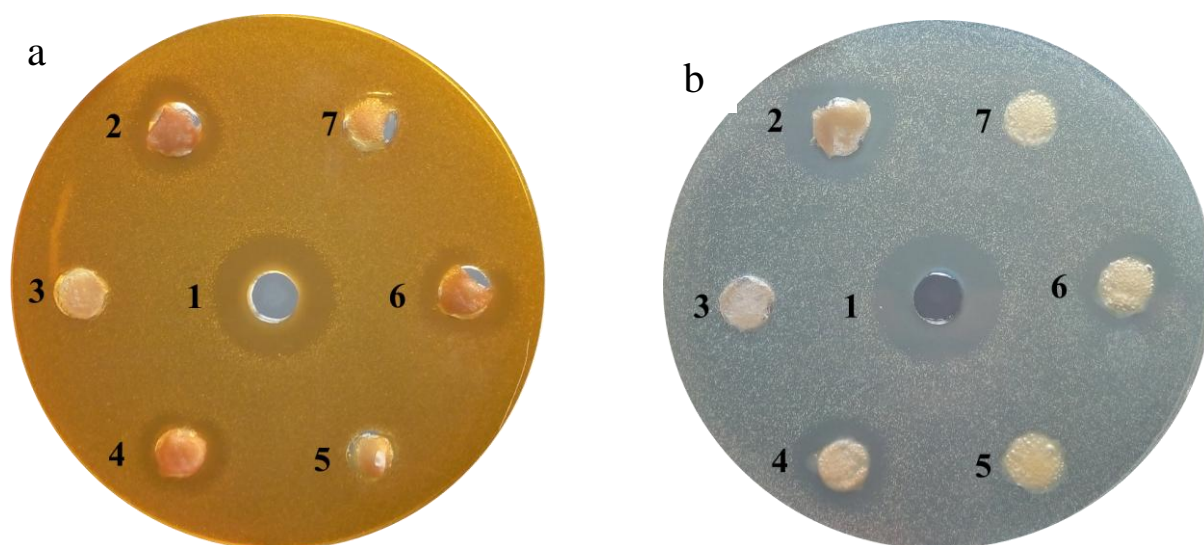
458

### 459 3.2. Release from aerogels and activity of nisin against Gram-positive bacteria.

460 The activity of nisin loaded and released from the three GM aerogels was evaluated against Gram-  
461 positive bacteria in a similar way as for polymyxin B.

462 Fig. 2a shows the growth inhibition by nisin released from “loaded” plugs of: EOLFG (2), EOLSG  
463 (4) and EOLGG (6) on cells of *E. faecalis* isolated from contaminated dairy samples and Fig. 2b on  
464 a strain of *C. tyrobutyricum*, cultured under anaerobic conditions. The amount of incorporated nisin  
465 was calculated from the average weight increase of individual aerogel plugs, performed in triplicate,  
466 after 1 h of immersion in a nisin solution (50,000 IU/g) and was estimated to be  $120.5 \pm 34.6 \mu\text{g}$  for  
467 EOLFG,  $98.9 \pm 9.3 \mu\text{g}$  for EOLSG and  $135.1 \pm 28.5 \mu\text{g}$  for EOLGG. All plugs were rinsed three  
468 times, as before, in order to remove nisin eventually present on the outer surface. The third rinse  
469 water did not develop any halo (see Fig. 4.S in “Supplementary Material”).

470



471

472

473 **Fig. 2a,b** - Inhibition by nisin released from aerogels of Gram-positive bacterial cell growth.  
474 Central well (1) 125  $\mu\text{g}$  of free nisin (equivalent to 37 nanomoles). Lateral wells: plugs of loaded  
475 EOLFG (2), control EOLFG (3); loaded EOLSG (4), control EOLSG (5); loaded EOLGG (6),  
476 control EOLGG(7). Fig. 2a: *E. faecalis*. Fig. 2b: *C. tyrobutyricum*.

477

### 478 3.3. Release and activity of lysozyme against *C. tyrobutyricum*

479 The muraminidase lysozyme is used in cheese and milk products to prevent butyric acid  
480 fermentation prevalently caused by *Clostridium* species. In a previous work (Rossi et al., 2016) we  
481 have described the release of lysozyme from EOLFG aerogel, evaluated by a standard biochemical  
482 test, based on the hydrolytic activity of this enzyme on agar plates layered with lyophilized cells of  
483 *M. lysodeikticus*, while in the present work a microbiological test was used.  
484 Fig. 5.S shows the inhibition zone by lysozyme released from a “loaded” plug of EOLFG (2) on live  
485 cells of *C. tyrobutiryicum*, grown under anaerobic conditions. The well on top (1) was filled with 100  
486 µl of 1.25g/L of lysozyme (i.e., 125 µg of protein equivalent to 8.7 nanomoles of active substance). A  
487 control plug of EOLFG (3) was fitted in the corresponding well. The amount of incorporated  
488 lysozyme was calculated from the average weight increase of individual aerogels, performed in  
489 triplicate, after 1 h of immersion in a lysozyme solution and was estimated to be  $120.5 \pm 34.6$  µg of  
490 lysozyme for EOLFG. Similar experiments with EOLSG and EOLGG gave somewhat unclear  
491 results, as those two aerogels turned out to be too brittle under the anaerobic and temperature  
492 conditions used for the test. However, this aspect is being investigated.  
493 Therefore, it appears that with the three selected antimicrobials and under the experimental  
494 conditions applied, the release of active was able to control microbial growth *in vitro*, even in the  
495 presence of a relatively high cell concentration.

#### 496 497 **3.4 Uptake and release of protease and lipase from GM aerogels.**

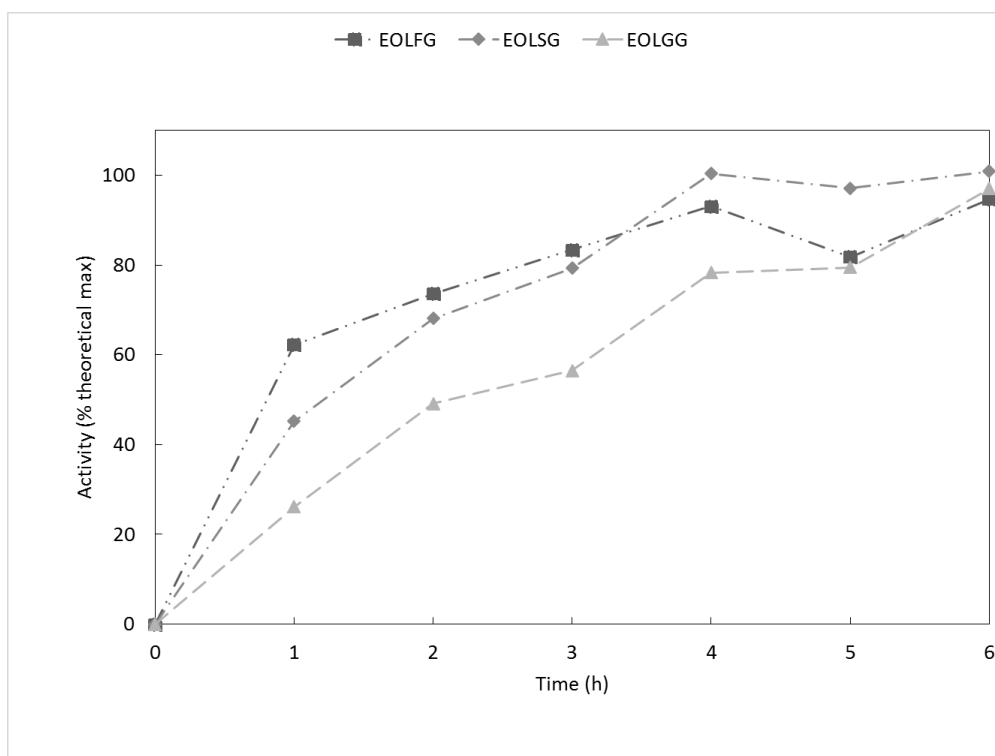
498 The following preliminary controls were performed on the two industrial enzymes tested in order to  
499 substantiate our experimental approach:

- 500 i. The effect of the lyophilization procedure on the lipase and protease was tested by activity  
501 determination of the enzymes, before and after lyophilization from water solution and re-  
502 hydration. Results indicated no significant difference caused by the treatment (*data not*  
503 *shown*).
- 504 ii. Eventual preferential uptake or exclusion of enzymes by GM gels was tested by measuring  
505 protein concentration before and after immersion of the plugs in the solutions of actives. In  
506 both cases of protease and lipase, protein concentration was unchanged, which is taken to  
507 mean that there is no selective protein exclusion or binding by the hydrogels (*data not*  
508 *shown*).

509 Enzyme release from “loaded” aerogel plugs in water was monitored by measuring protein  
510 concentration and increase in enzyme activity as a function of time.



511 Fig. 3 shows the kinetics of protease release at RT from the three hydrogels, measured every hour,  
512 for 6 h, expressed as percentage of the “theoretical” total protease activity loaded. The mean values  
513 and standard deviations are reported in Tab. 1.S (see Supplementary Material).  
514 Protease release appears to be faster in the first 2 h and then gradually reaches a plateau after 6 h.  
515



516  
517

518 **Fig. 3** - Time course of protease release in water solution from “loaded” hydrogel plugs of EOLFG,  
519 EOLSG and EOLGG. Each time point was performed in triplicate on separately prepared plugs and  
520 is expressed as the average percentage of the “theoretical” total protease activity loaded.

521

522 It can be calculated that this represents, in all three cases, an almost complete release of enzyme  
523 loaded into the aerogels (Tab. 1.S.).

524 In the case of lipase, preliminary evaluations were performed with a *C. rugosa* enzyme obtained from  
525 Sigma for research purposes, containing over 90% of non protein material, which had to be discarded  
526 before further experiments. Loading and release were performed only on EOLFG, as described  
527 above, and it was found that almost all the calculated amount of enzyme absorbed was released after  
528 5 h incubation in water at RT (*not shown*). However, in consideration of the little amount of enzyme  
529 available, the other two aerogels were not tested.

530 Therefore, for practical applications (e.g., in detergency), we considered more meaningful to try with  
531 an industrial enzyme commercially available, hence the choice of Novozymes’ Greasex Ultra L from

532 *C. lipolitica* with an activity of 15 U/ml, used as such with no further purification or pre-treatments.  
 533 In this case, as shown in Fig. 6.S for the three aerogels, the release of lipase in solution was limited to  
 534 about 30% of the “theoretical” maximum, even after a very long incubation period, as if the  
 535 stabilizing agents contained in the commercial formulation of this enzyme (e-g., glycols, salts, etc.),  
 536 prevented or limited its full release. Purification experiments of this lipase are underway to verify the  
 537 hypothesis, focusing mainly on EOLFG which preserves its structure upon prolonged incubation  
 538 periods in solution.

539 The mean values and standard deviations of the experimental data points of Fig. 6.S are reported in  
 540 Tab. 2.S.

541

### 542 **3.5. Determination of aerogels pore size**

543 Data from MIP tests are reported in Tab. 1.

544

<b>Sample</b>	<b>Bulk density (kg x m<sup>-3</sup>)</b>	<b>Porosity (%)</b>	<b>Pore surface area (m<sup>2</sup> x g<sup>-1</sup>)</b>	<b>Average pore diameter (µm)</b>	<b>Modal pore diameter (µm)</b>
<b>EOLFG</b>	35.9	99.91	11.32	9.83	42.44
<b>EOLSG</b>	32.8	96.19	16.89	6.94	3.59
<b>EOLGG</b>	31.3	93.28	15.54	7.68	102.71

545

546 **Tab. 1 - MIP parameters of the three GM aerogels**

547

548 The aerogel from fenugreek (EOLFG) exhibited higher density (35.9 kg x m<sup>-3</sup>) compared to guar  
 549 gum (EOLGG) and sesbania (EOLSG) with 31.3 and 32.8 kg x m<sup>-3</sup>, respectively. Because a number  
 550 of parameters may affect the bulk density of aerogels (e.g., type of polymer, initial polymer  
 551 concentration, pH, ionic strength, air inclusion method, and instrumental technique for the  
 552 measurements), a strict comparison among different materials is relatively hard to make. For  
 553 example, sodium carboxymethyl cellulose aerogels were reported to have bulk density of 56.8 kg x  
 554 m<sup>-3</sup> (Lin, Li, Lu, & Cao, 2015), whereas the bulk density of wheat gluten-based aerogels ranged  
 555 between 58 and 343 kg m<sup>-3</sup> (Blomfeldt et al., 2010), therefore a wide range of values is to be  
 556 expected. On the other hand, previously reported open-cell CNF foams and CNF aerogels had  
 557 densities of 15 and 7–20 kg x m<sup>-3</sup>, respectively (Jiang & Hsieh, 2014; Svagan, Samir, & Berglund,  
 558 2008), lower than the values we have found, so that different thermal values are to be expected.

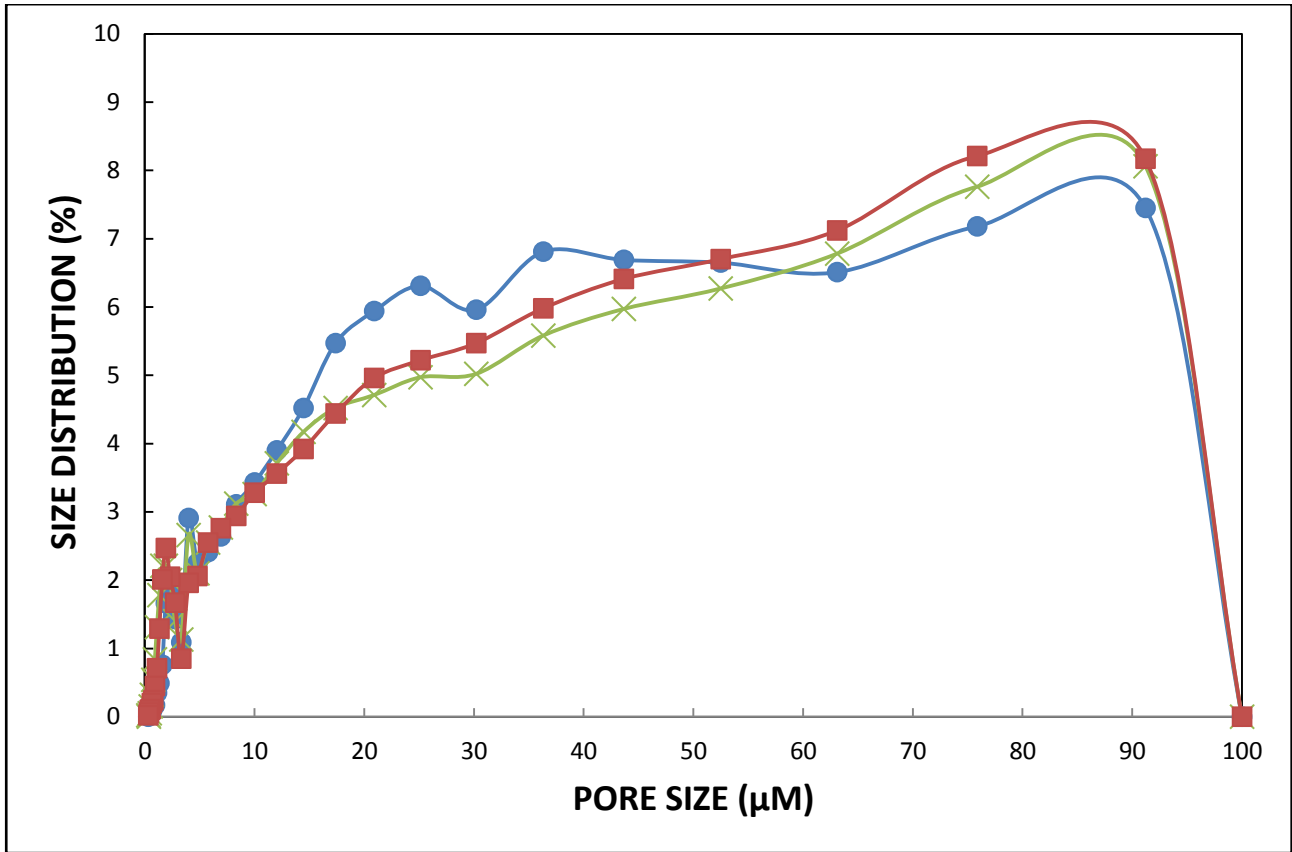
559 From a practical point of view, it is interesting to notice that expanded and extruded polystyrene  
560 closed-cell foams, for thermal insulation applications, have densities in the range of 16–45 kg x m<sup>-3</sup>,  
561 whereas those of polyurethane are in the range of 40–55 kg x m<sup>-3</sup> (Al-Homoud, 2005). Thus, it will  
562 be interesting to study the thermal insulation properties of the plant-derived materials we have  
563 obtained.

564 The three aerogels described in this study showed very high open porosity values, with an  
565 outstanding 99.9 % measured for the EOLFG sample. This value is in line with a recent work on  
566 lightweight foams based on nano-cellulose, which exhibited an open porosity of 99.6% (Wicklein et  
567 al., 2015). In our studies, differences in pore size and distribution were observed. EOLFG showed  
568 pores of larger size compared to the two other aerogels, as demonstrated by both pore surface area  
569 (11.32 m<sup>2</sup> x g<sup>-1</sup>) and average pore diameter (9.83 μm) (see Tab. 1). EOLSG showed the smallest  
570 pores, with pore surface area and average diameter of 16.89 m<sup>2</sup> x g<sup>-1</sup> and 6.94 μm, respectively.  
571 While EOLGG stood halfway in terms of pore size. There was also a difference in pore size  
572 distribution according to the nature of the GM aerogel.

573 As indicated by the modal pore diameter, most of the pores were centered at ~ 3.6 μm diameter for  
574 the EOLSG (meaning that the smallest pores are more largely represented), ~ 42 μm for EOLFG,  
575 and ~ 103 μm for the EOLGG. This can be seen in the pore size distribution plots of Fig. 7, a-c.S  
576 (see Supplementary Material). While the pore diameter frequency distribution of the sample of  
577 EOLFG (Fig. 7,a.S) includes several classes of the same height between 10 and 50 μm, the  
578 frequency distribution of samples EOLSG (Fig. 7,b.S) and EOLGG (Fig. 7,c.S) is somehow skewed  
579 toward high pore diameters. Moreover, with EOLGG and EOLSG the occurrence of a second peak  
580 can also be observed, which is broader for EOLSG between 1 and 5 μm. These results indicate that,  
581 although with some differences, all the three aerogels tested in this work had more open macropores  
582 than mesopores and micropores in their final structure, suggesting a preferential practical  
583 application where rapid capillary suction is of utmost importance, e.g., for “fast” delivery systems  
584 and in liquid cleaning/removing devices (Blomfeldt et al., 2010).

585 Finally, Fig. 4 (which is derived from the data of Fig. 7, a-c.S) indicates that in the case of  
586 fenugreek there is a wider distribution of pores with intermediate size (20-50 μm) than in guar and  
587 sesbania, presumably because it is the most extensively cross-linked aerogel.

588



589

590

591 **Fig. 4-** Pore size distribution in diameter (abscissa) vs their % distribution (ordinate) of the three  
 592 GM aerogels (blu curve (●), EOLFG; green curve (x), EOLSG; red curve (■), EOLGG.

593

594 However, if we consider the distribution of larger pores (i.e., between 50-100 μm), guar appears to  
 595 have the most, closely followed by sesbania.

596

597 **3.6. Mechanical properties**

598 Large deformation tests allowed to quantify the effect of enzymatic oxidation on the mechanical  
 599 properties of EOLFG, EOLSG and EOLGG. Tab. 2 reports the mechanical parameters obtained  
 600 from compressive tests.

601

Sample	E-mod (kPa)	F <sub>max</sub> (N)	Cohesiveness	Springiness	Resilience
<b>LFG</b>	0.502 <sup>a</sup> (± 0.085)	0.081 <sup>a</sup> (± 0.014)	0.713 <sup>a</sup> (± 0.040)	0.651 <sup>ab</sup> (± 0.042)	0.393 <sup>ab</sup> (± 0.033)
<b>EOLFG</b>	22.775 <sup>b</sup> (± 5.792)	2.376 <sup>b</sup> (± 0.548)	0.685 <sup>a</sup> (± 0.034)	0.611 <sup>a</sup> (± 0.038)	0.316 <sup>c</sup> (± 0.037)

<b>LSG</b>	0.249 (± 0.083) <sup>a</sup>	0.039 (± 0.012) <sup>a</sup>	0.728 (± 0.033) <sup>a</sup>	0.665 (± 0.033) <sup>b</sup>	0.434 (± 0.034) <sup>b</sup>
<b>EOLSG</b>	12.479 (± 5.165) <sup>c</sup>	1.332 (± 0.231) <sup>c</sup>	0.712 (± 0.057) <sup>a</sup>	0.641 (± 0.063) <sup>ab</sup>	0.373 (± 0.066) <sup>a</sup>
<b>LGG</b>	0.433 (± 0.070) <sup>a</sup>	0.081 (± 0.023) <sup>a</sup>	0.694 (± 0.043) <sup>a</sup>	0.632 (± 0.049) <sup>ab</sup>	0.386 (± 0.048) <sup>a</sup>
<b>EOLGG</b>	18.577 (± 5.527) <sup>d</sup>	2.075 (± 0.607) <sup>b</sup>	0.627 (± 0.049) <sup>b</sup>	0.542 (± 0.047) <sup>c</sup>	0.257 (± 0.032) <sup>d</sup>

602

603 **Table 2** -Mechanical parameters from compressive tests on pristine, lyophilized and laccase-  
604 oxidized, lyophilized fenugreek (LFG, EOLFG), sesbania (LSG, EOLSG) and guar gum  
605 (GG, EOLGG) aerogels.

606

607 Results are expressed as mean values and standard deviation (in brackets). Different superscripts  
608 within a group (i.e., within each parameter) refers to a statistically significant difference ( $p < 0.05$ ).

609 The three non-oxidized, lyophilized GM aerogels (i.e., LFG, LSG, LGG) did not exhibit  
610 appreciable differences in terms of  $F_{max}$  and E-mod. However, after laccase oxidation it was  
611 possible to observe statistically significant differences between samples. In absolute terms, EOLFG  
612 aerogel was stiffer and harder than EOLSG and EOLGG, with EOLSG showing the lowest mean  
613 values of both  $F_{max}$  and E-mod. Nevertheless, if we consider the relative increase of the above  
614 parameters, the largest occurred for EOLSG ( $F_{max}$  and E-mod increased  $\sim 50$  and 34 times,  
615 respectively). This observation seems to indicate that the oxidation, and resulting internal  
616 crosslinking, was somewhat more “efficient” in the case of SG than FG and GG, though SG  
617 ultimately yielded a relatively weaker aerogel.

618 Concerning the other mechanical parameters, cohesiveness represents the ability of the sample to  
619 stand a second deformation in relation with the sample behavior during the first deformation cycle.  
620 Springiness indicates the capability of the sample to spring back after the first compression.  
621 Resilience is a measure of the ability of the sample to recover its original shape. After oxidation,  
622 cohesiveness and springiness decreased significantly only for the GG samples. Resilience decreased  
623 significantly upon oxidation for all the three samples, consistent with the inverse relationship  
624 between this parameter on one hand, and the compressive modulus and maximum force on the other  
625 (Ghafar et al., 2015).

626 Although the results of the mechanical tests unambiguously demonstrate positive effect of  
627 enzymatic oxidation on gel structuring, it is clear that the crosslinking efficiency depends on the  
628 specific polysaccharide and the formation of hemiacetal bonds at inter- and intra- molecular level.

#### 629 **4. Conclusions**

630  
631 There is an increasing interest in the development of new biomaterials from renewable resources,  
632 possibly by means of sustainable biochemical reactions. Indeed, plant polysaccharides represent a  
633 valuable and sustainable alternative to traditional synthetic polymers produced from monomers of  
634 fossil, non-renewable origin and are increasingly applied in a growing number of industrial fields,  
635 either in their natural or chemically/biochemically modified forms. Enzyme modifications of  
636 polysaccharides is still an open field of investigation and product development. However, while  
637 glycohydrolases (e.g., amylases, cellulases, mannanases, etc.) are currently widely used enzymes for  
638 depolymerization, the family of redox enzymes (e-g., dehydrogenases and oxidases) have so far been  
639 relatively less studied and applied to generate “functional” polymers and new biomaterials (see also  
640 Karaki, Aljawish, Hueau, Muniglia & Jasniewski, 2016). Galactomannans are employed to produce a  
641 considerable range of derivatives with numerous applications in food, feed and industrial fields.  
642 These are versatile and quite reactive polysaccharides, that, from an agronomic point of view, offer  
643 the advantage of being reserve polysaccharides of leguminous plants grown in sub-tropical and semi-  
644 arid areas of the world, requiring little or no fertilizing and limited artificial watering, compared to  
645 starch-containing cereals. In the present work, we have applied to three different GM the laccase-  
646 mediator oxidation system previously described (Lavazza et al, 2011), followed by lyophilization,  
647 and have shown that oxidation is a key factor in the generation of aerogels with enhanced overall  
648 features and performance. However, differences among the oxidized GM suggest that the chemical  
649 composition of the polysaccharides (i.e., the galactose : mannose ratio) appears to play an important  
650 role. Fenugreek exhibited the best performance in terms of mechanical properties over guar and  
651 sesbania gums, which is likely due to its highest amount of galactose along the mannose backbone  
652 (Gal : Man = 1:1).

653 We suggest that the findings reported might be useful in the designing of novel nanostructured  
654 biomaterials with tailored functional properties as biocompatible, flexible delivery systems of various  
655 actives (for a comprehensive review, see Ulker & Erkey, 2014), as well as for food packaging or  
656 insulating devices.

657 More studies are needed to establish their real potential and applicability. Few lines of investigation  
658 are ongoing to clarify various aspects of these systems, such as: freeze-drying protocols to influence  
659 pore and size formation, range of actives that can be uploaded and released, kinetics of release, role

660 of temperature and humidity on aerogels performance, long term activity preservation, stability in  
661 order to ensure long-term antimicrobial protection.

662

663 ***Acknowledgements***

664 We wish to thank Dr. Sergio Riva for constant support, suggestions and criticism throughout this  
665 project, Dr. Lucia Galimberti for performing the MIP and Prof Alessandra Polissi for suggesting the  
666 testing and the gift of a polymyxin B sample.

667 The supply of GM gums from guar, sesbania and fenugreek by Lamberti SpA is gratefully  
668 acknowledged.

669 This project was supported by the program “**Suschem Lombardia: prodotti e processi chimici**  
670 **sostenibili per l’industria lombarda**”. Accordo Quadro Regione Lombardia-CNR, 16/07/2012  
671 (protocol no. 18096/RCC), by **Cariplo Foundation** (grant 2014-0478).

672



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849

**Table 1** - MIP parameters of the three GM aerogels

<b>Sample</b>	<b>Bulk density</b> <b>(kg x m<sup>-3</sup>)</b>	<b>Porosity</b> <b>(%)</b>	<b>Pore surface area</b> <b>(m<sup>2</sup> x g<sup>-1</sup>)</b>	<b>Average pore diameter</b> <b>(μm)</b>	<b>Modal pore diameter</b> <b>(μm)</b>
<b>EOLFG</b>	35.9	99.91	11.32	9.83	42.44
<b>EOLSG</b>	32.8	96.19	16.89	6.94	3.59
<b>EOLGG</b>	31.3	93.28	15.54	7.68	102.71

**Table 2** - Mechanical parameters from compressive tests on pristine, lyophilized and laccase-oxidized, lyophilized fenugreek (LFG, EOLFG), sesbania (LSG, EOLSG) and guar gum (GG, EOLGG) aerogels.

<b>Sample</b>	<b>E-mod (kPa)</b>	<b>F<sub>max</sub> (N)</b>	<b>Cohesiveness</b>	<b>Springiness</b>	<b>Resilience</b>
LFG	0.502 (± 0.085) <sup>a</sup>	0.081 (± 0.014) <sup>a</sup>	0.713 (± 0.040) <sup>a</sup>	0.651 (± 0.042) <sup>ab</sup>	0.393 (± 0.033) <sup>ab</sup>
EOLFG	22.775 (± 5.792) <sup>b</sup>	2.376 (± 0.548) <sup>b</sup>	0.685 (± 0.034) <sup>a</sup>	0.611 (± 0.038) <sup>a</sup>	0.316 (± 0.037) <sup>c</sup>
LSG	0.249 (± 0.083) <sup>a</sup>	0.039 (± 0.012) <sup>a</sup>	0.728 (± 0.033) <sup>a</sup>	0.665 (± 0.033) <sup>b</sup>	0.434 (± 0.034) <sup>b</sup>
EOLSG	12.479 (± 5.165) <sup>c</sup>	1.332 (± 0.231) <sup>c</sup>	0.712 (± 0.057) <sup>a</sup>	0.641 (± 0.063) <sup>ab</sup>	0.373 (± 0.066) <sup>a</sup>
LGG	0.433 (± 0.070) <sup>a</sup>	0.081 (± 0.023) <sup>a</sup>	0.694 (± 0.043) <sup>a</sup>	0.632 (± 0.049) <sup>ab</sup>	0.386 (± 0.048) <sup>a</sup>
EOLGG	18.577 (± 5.527) <sup>d</sup>	2.075 (± 0.607) <sup>b</sup>	0.627 (± 0.049) <sup>b</sup>	0.542 (± 0.047) <sup>c</sup>	0.257 (± 0.032) <sup>d</sup>



Fig. 1-a,b

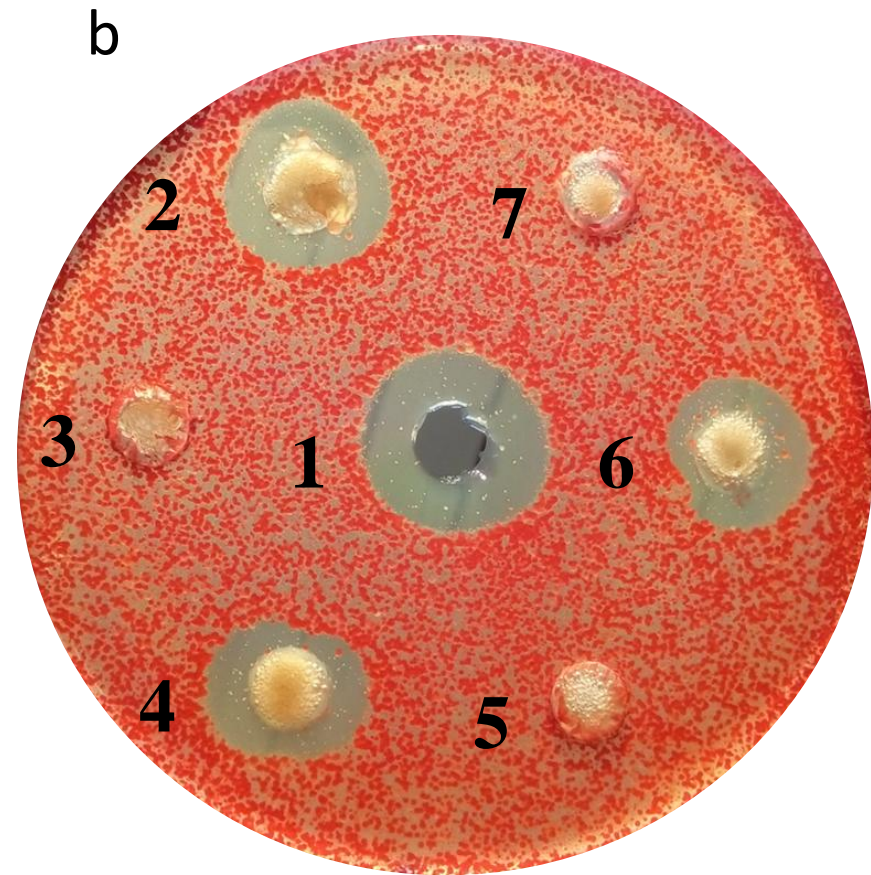
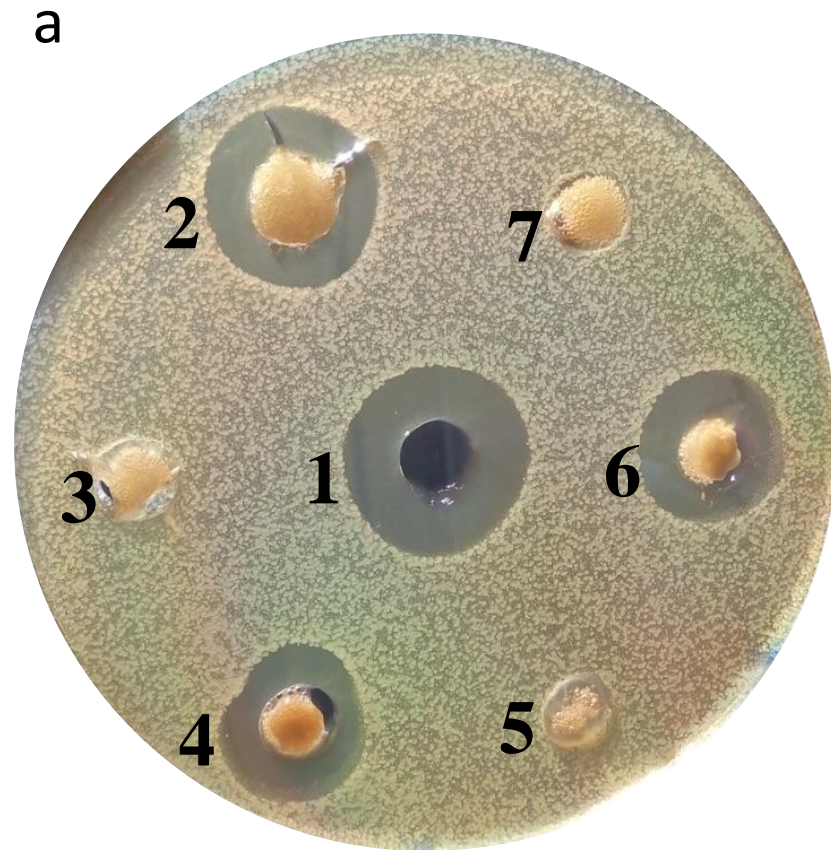


Fig. 1-c,d,e

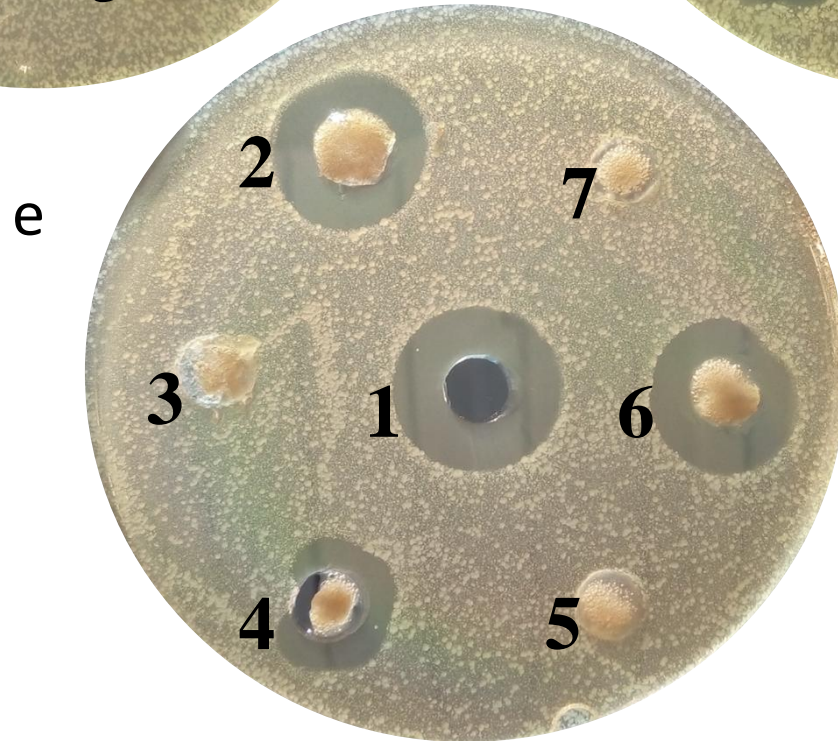
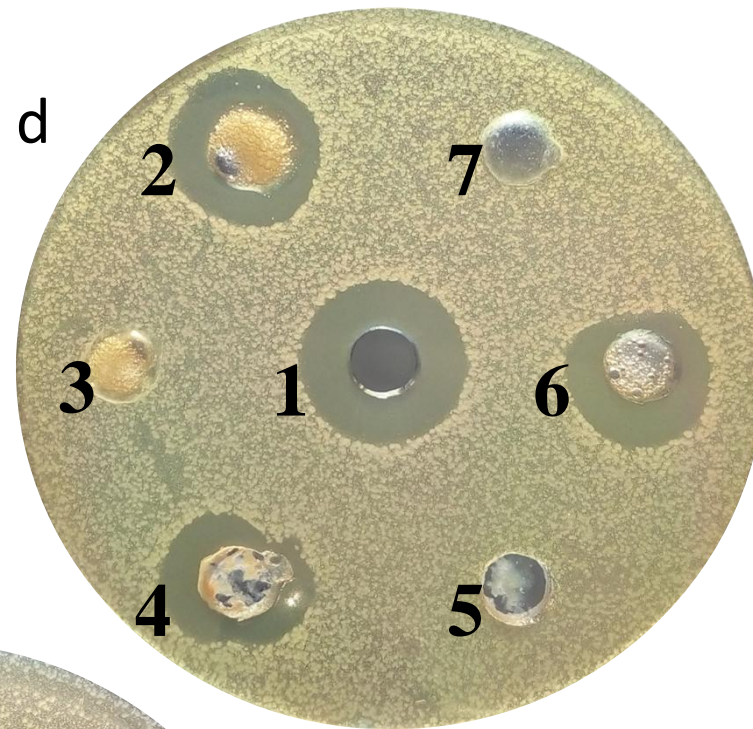
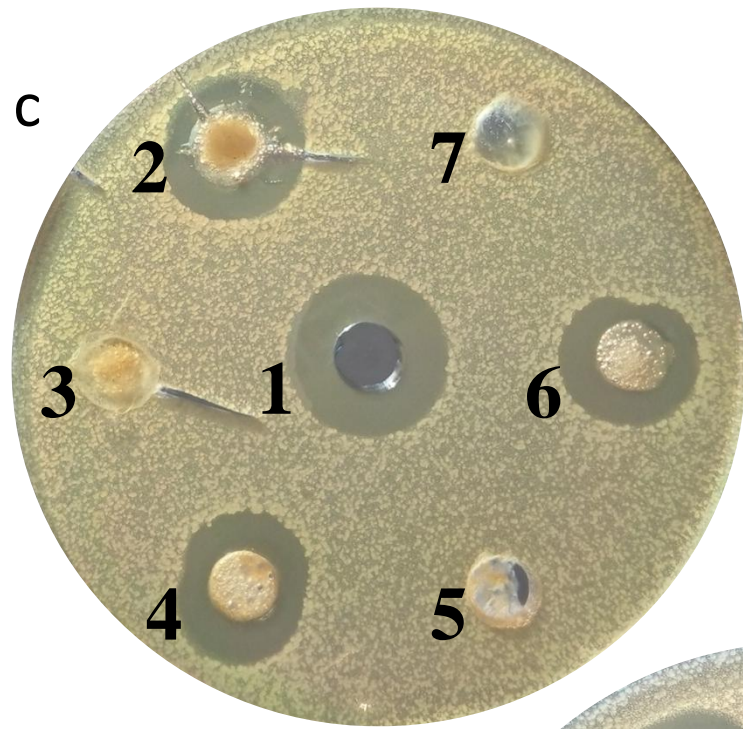


Fig. 2-a,b

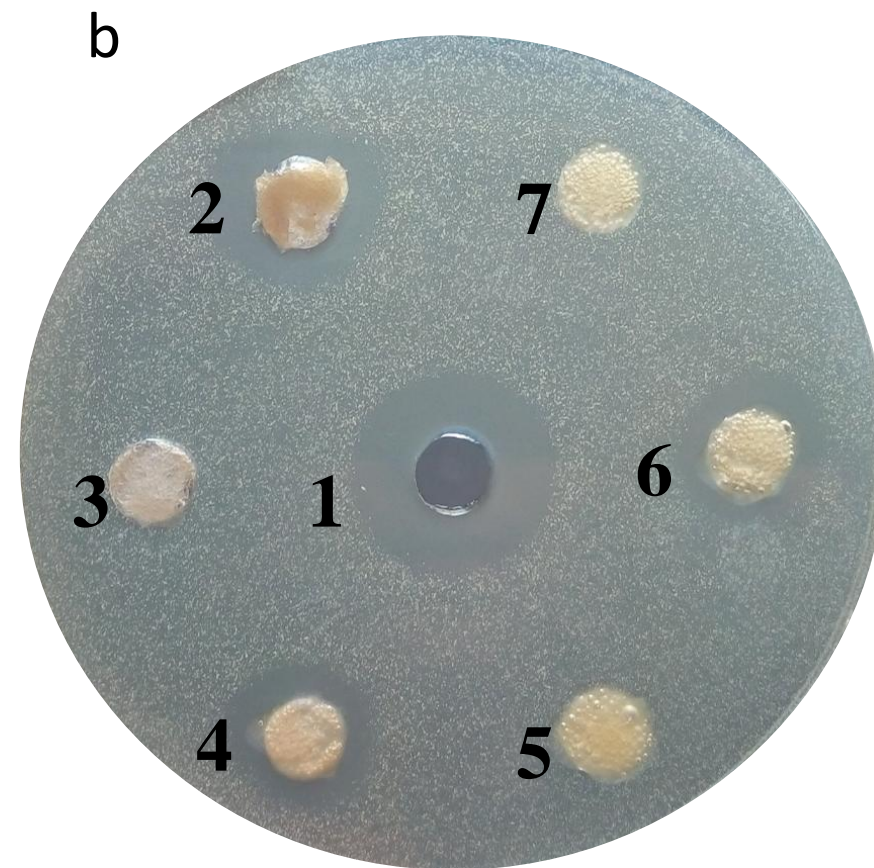
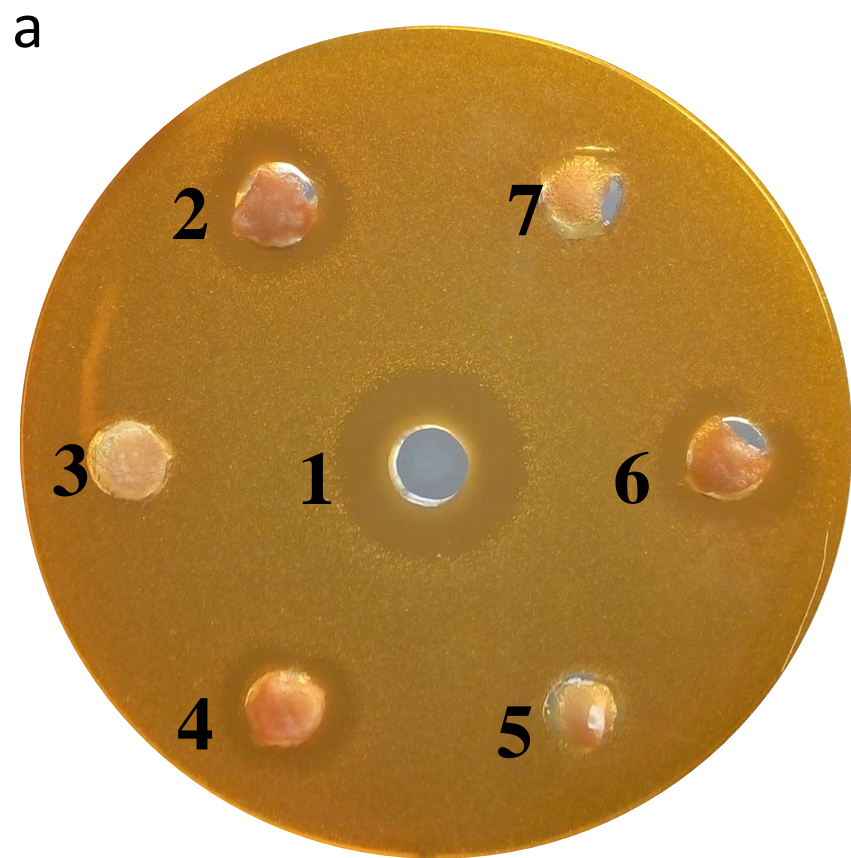


Fig. 3

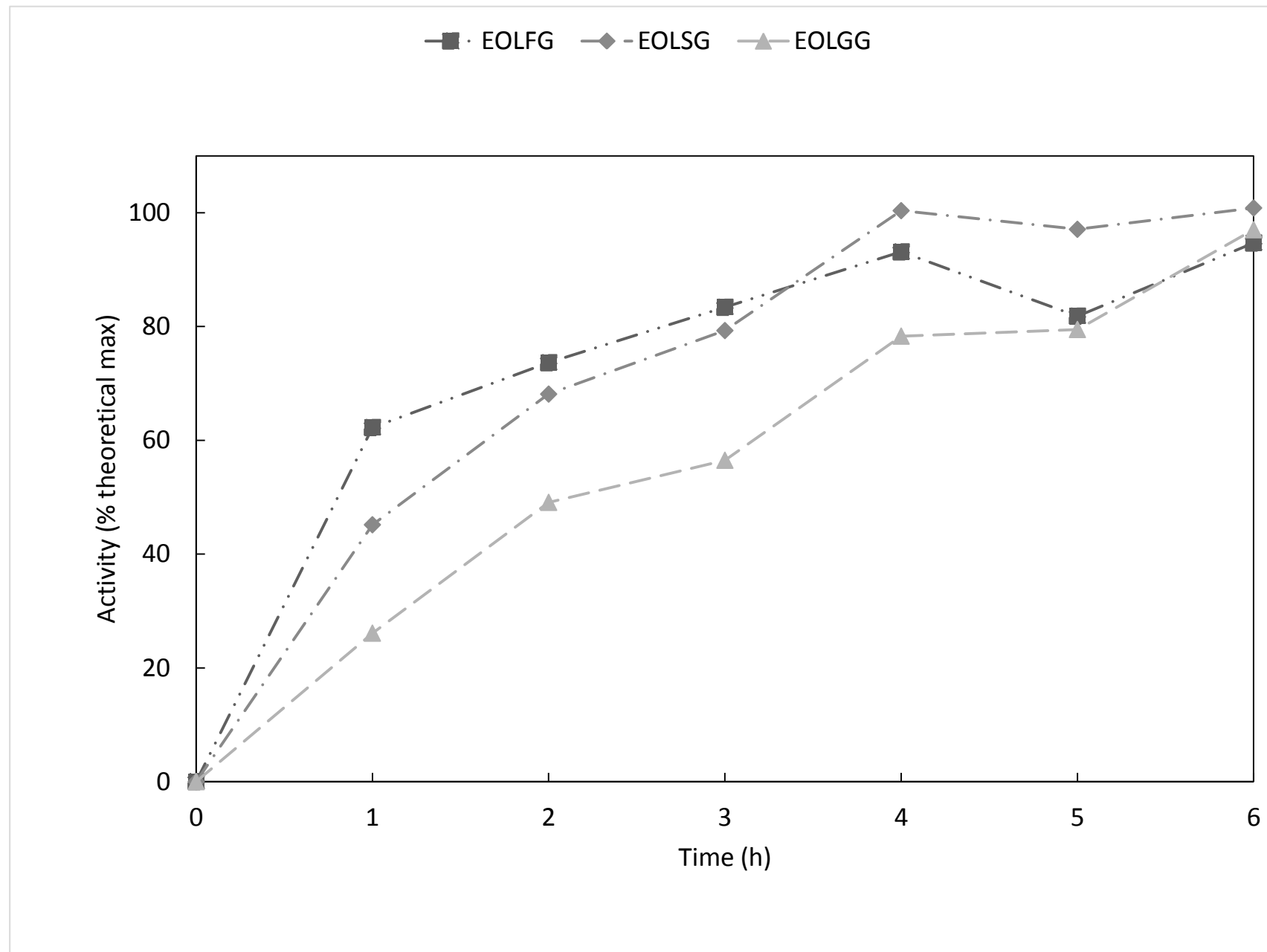
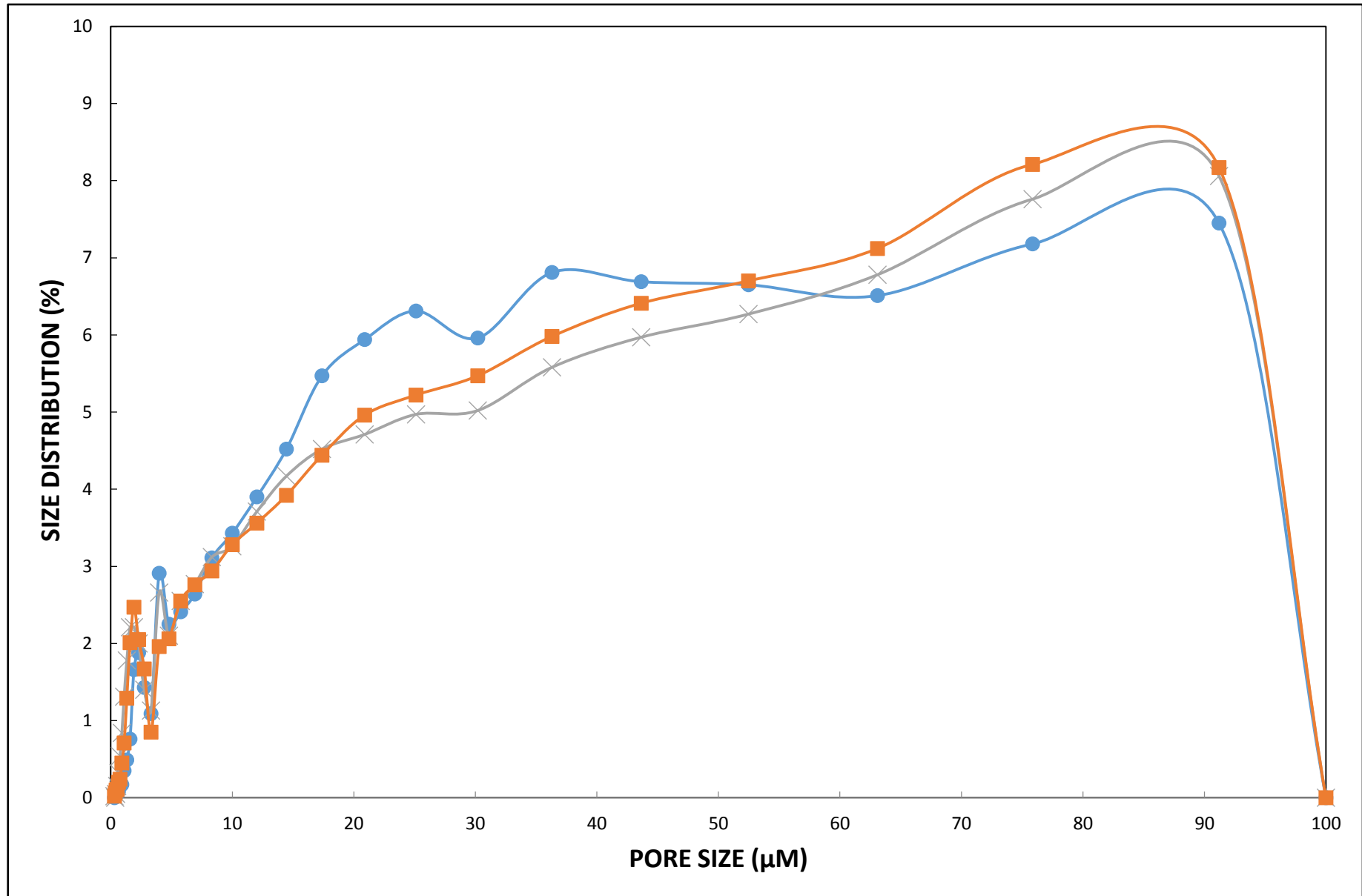


Figure 4



## Figure legends

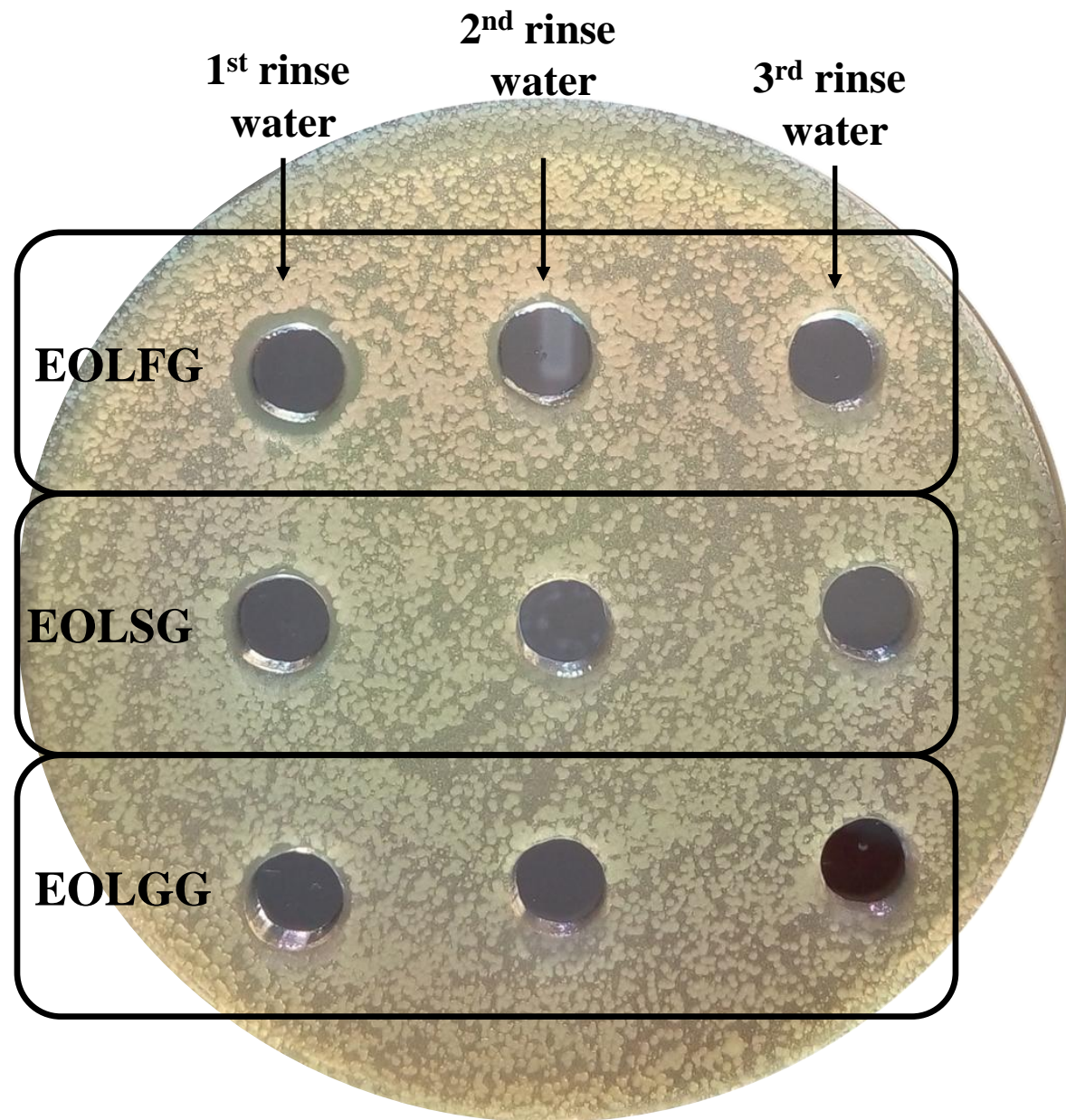
**Fig. 1-** Inhibition by polymyxin B released from aerogels of Gram-negative bacterial cell growth. Central well (1), 30  $\mu\text{g}$  (equivalent to 22 nanomoles) of free polymyxin B. Lateral wells: plugs of loaded EOLFG (2), control EOLFG (3); loaded EOLSG (4), control EOLSG (5); loaded EOLGG (6), control EOLGG (7). Fig. 1a: *P. aeruginosa*. Fig. 1b: *S. marcescens*. Fig. 1c: *S. Typhimurium*. Fig. 1d: *E. coli*. Fig. 1e: *E. cloacae*.

**Fig. 2 -** Inhibition by nisin released from aerogels of Gram-positive bacterial cell growth. Central well (1) 125  $\mu\text{g}$  of free nisin (equivalent to 37 nanomoles). Lateral wells: plugs of loaded EOLFG (2), control EOLFG (3); loaded EOLSG (4), control EOLSG (5); loaded EOLGG (6), control EOLGG(7). Fig. 2a: *E. faecalis*. Fig. 2b: *C. tyrobutirycum*.

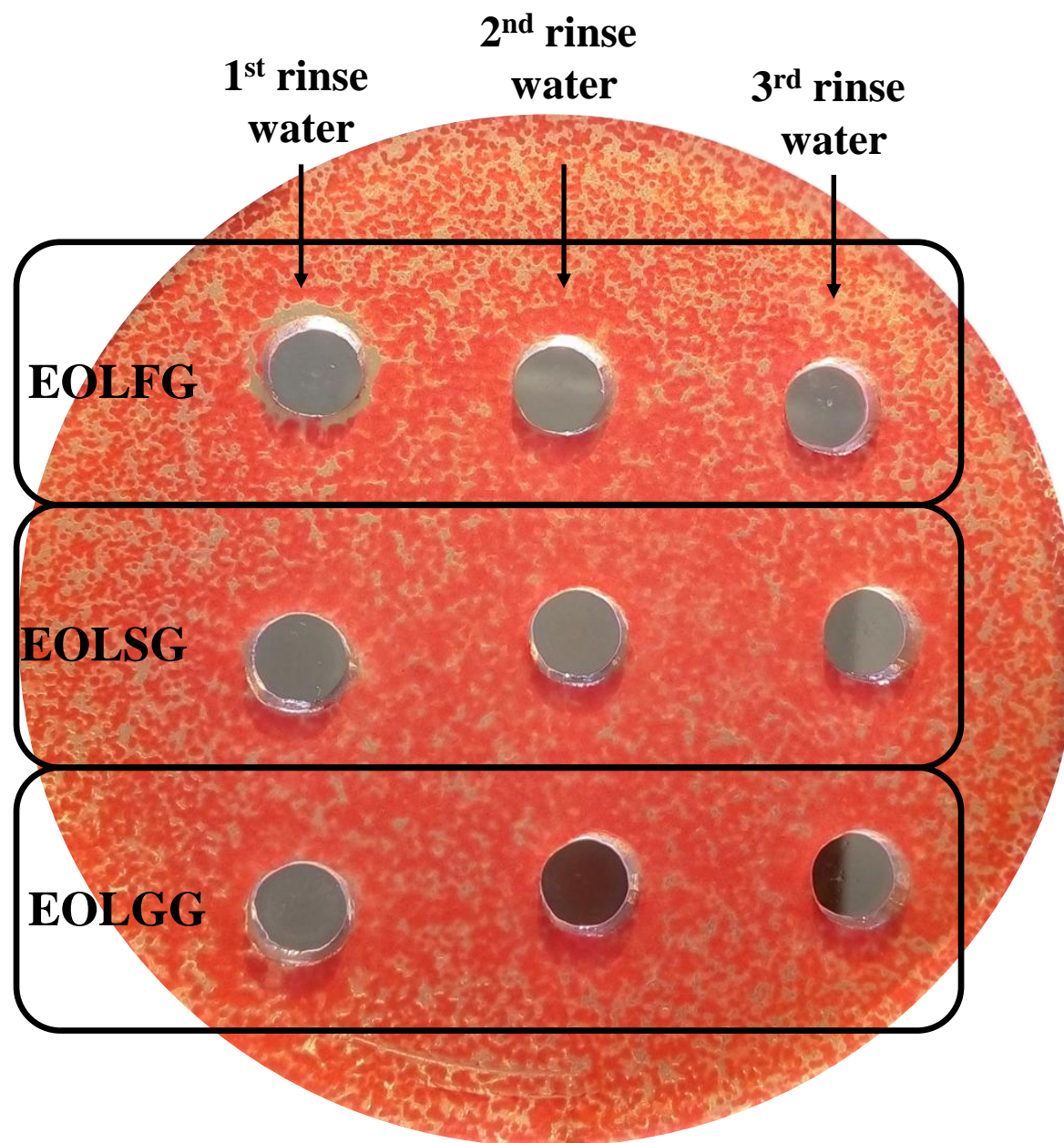
**Fig. 3 -** Time course of protease release in water solution from “loaded” hydrogel plugs of EOLFG, EOLSG and EOLGG. Each time point was performed in triplicate on separately prepared plugs and is expressed as the average percentage of the “theoretical” total protease activity loaded.

**Fig. 4-** Pore size distribution in diameter (abscissa) vs their % distribution (ordinate) of the three GM aerogels (blue curve (●), EOLFG; green curve (x), EOLSG; red curve (■), EOLGG).

**Fig. 1.S**

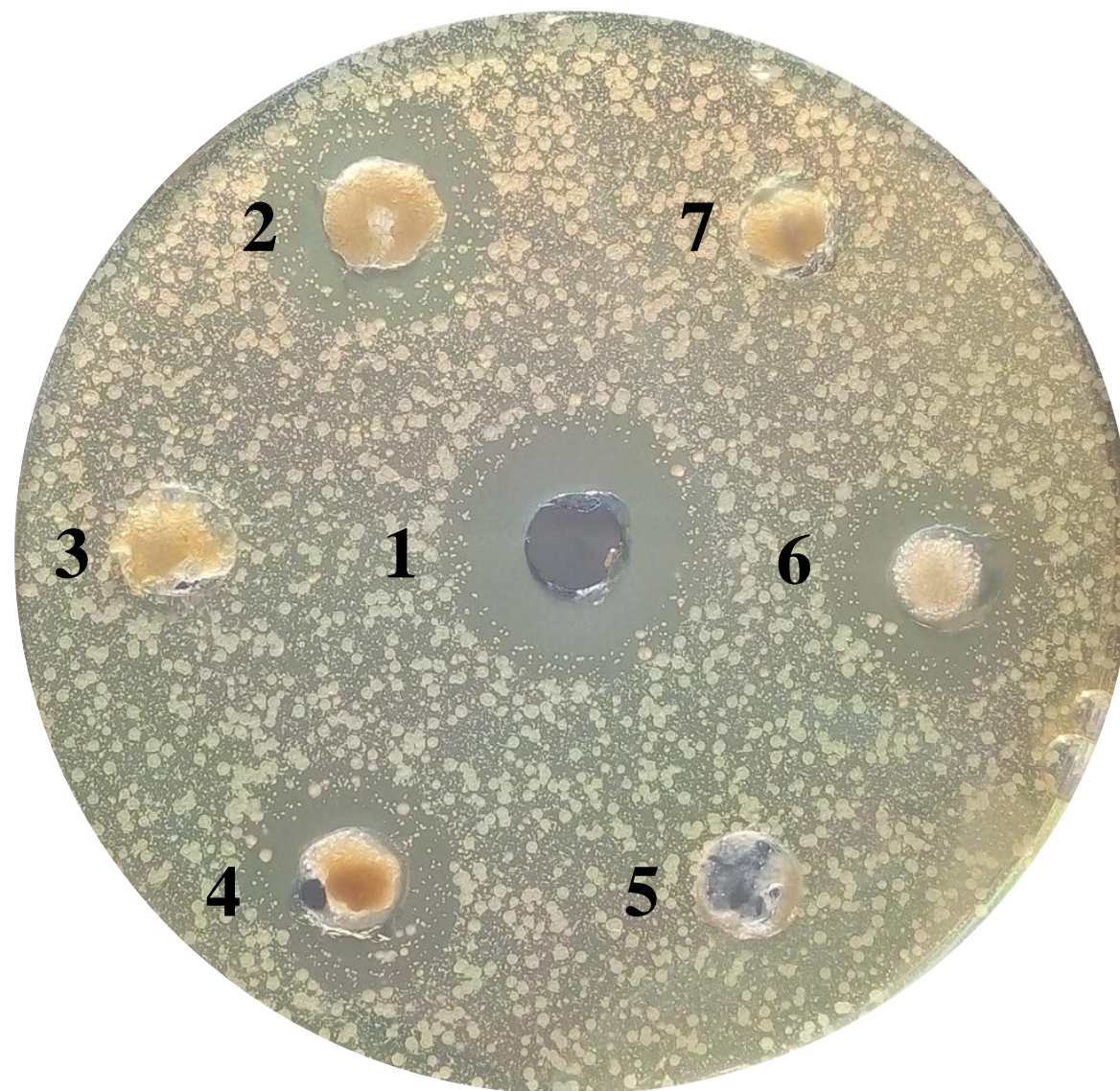


**Fig. 2.S**

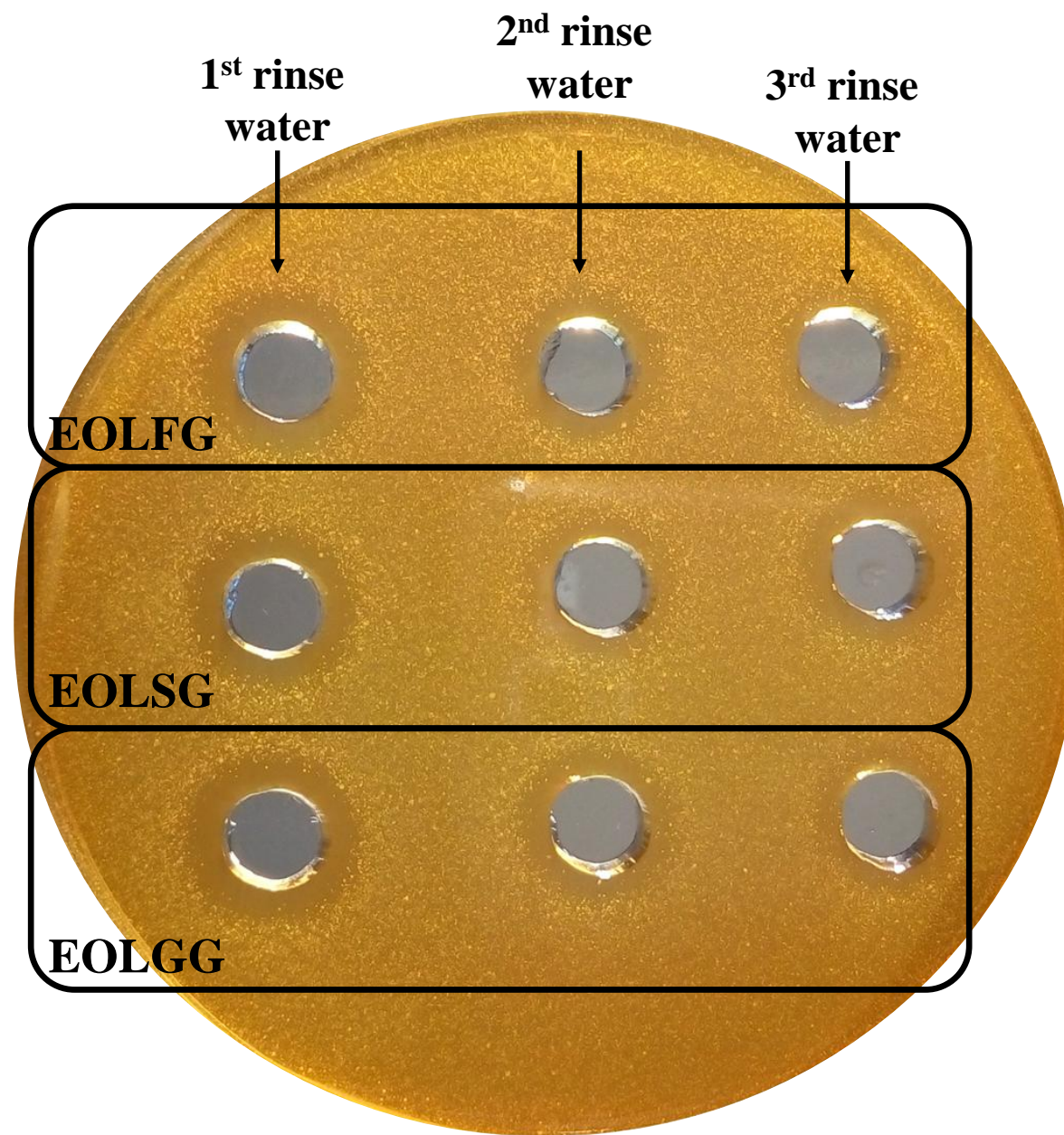




**Fig. 3.S**



**Fig. 4.S**



**Fig. 5.S**

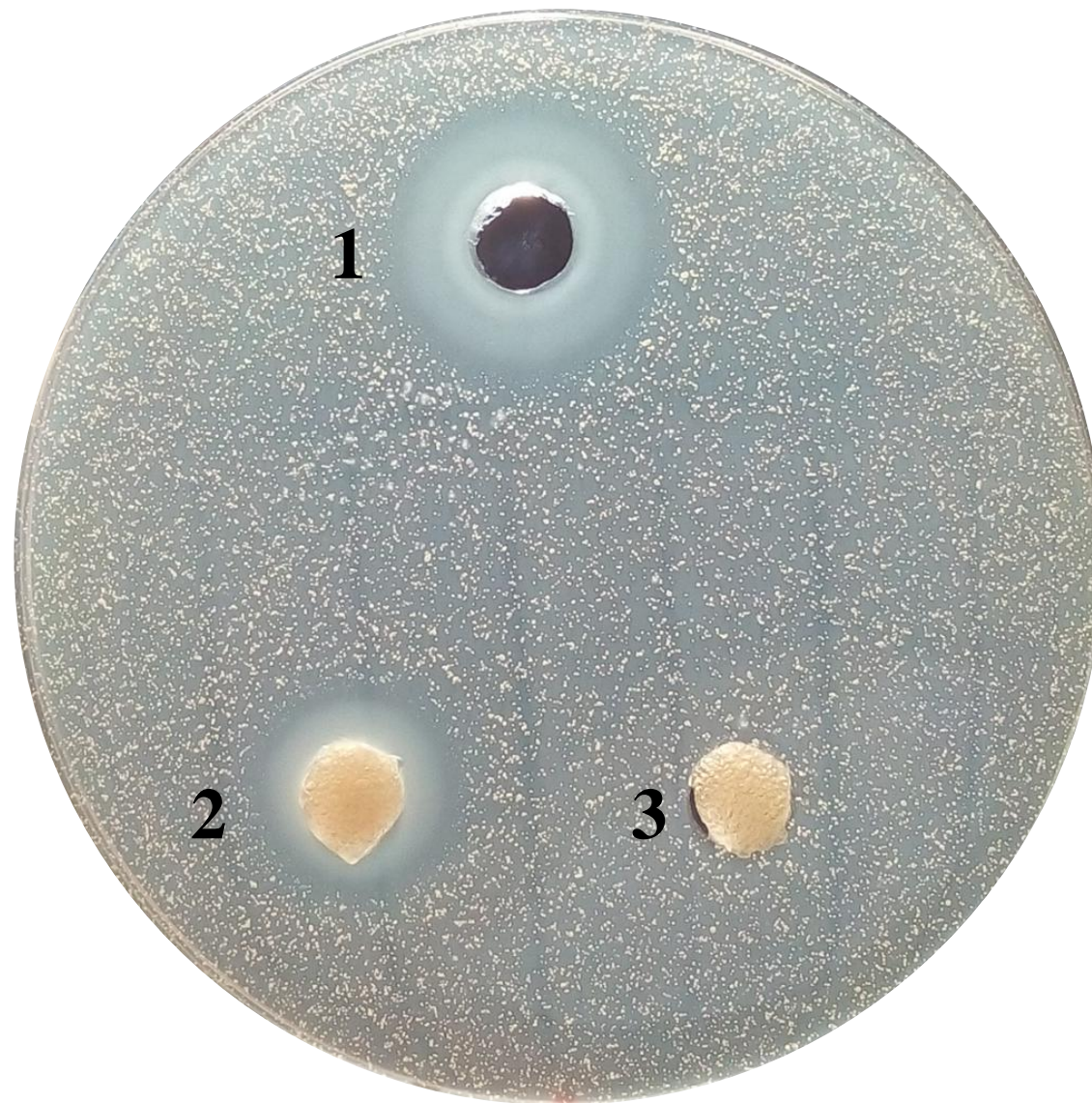


Fig. 6.S

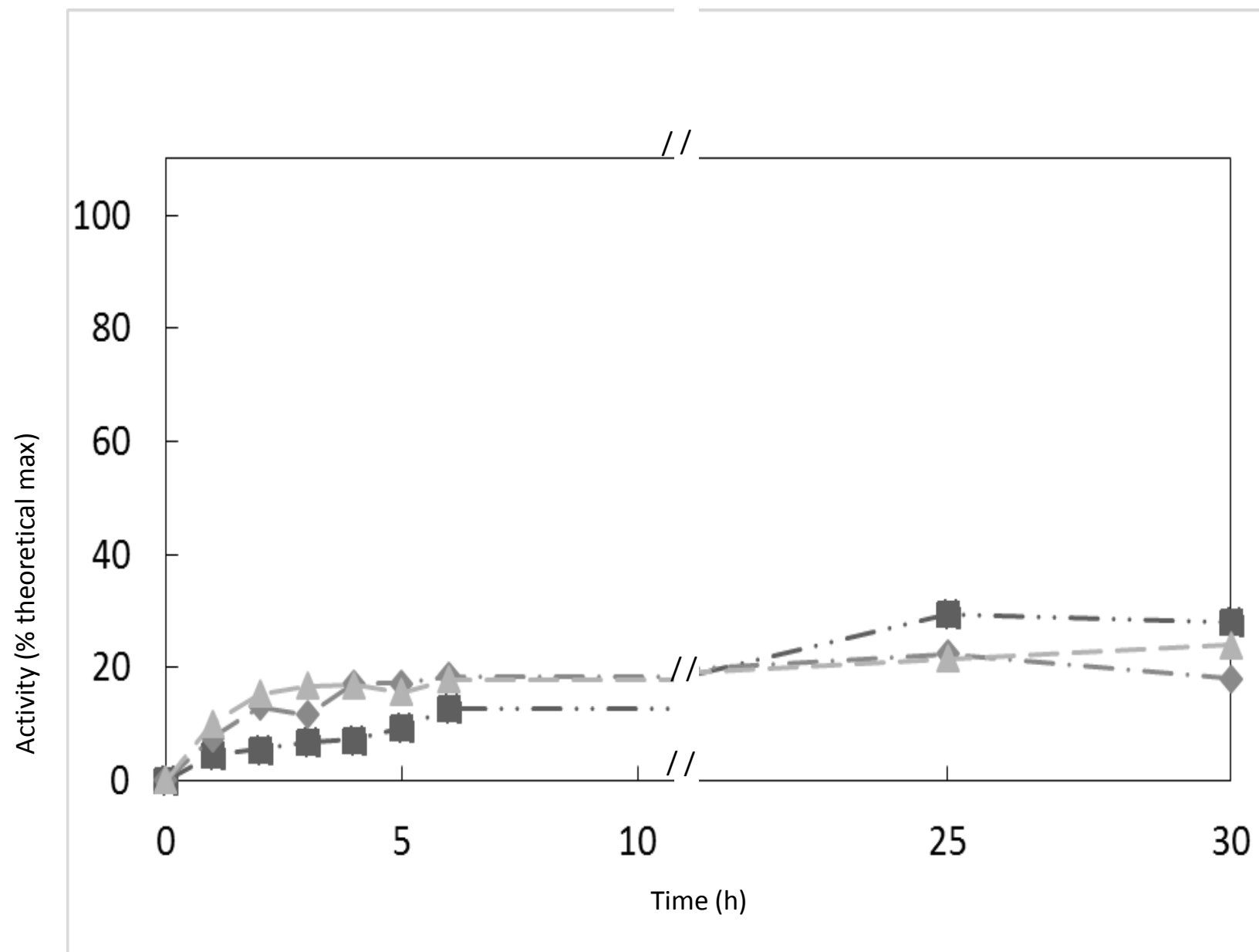
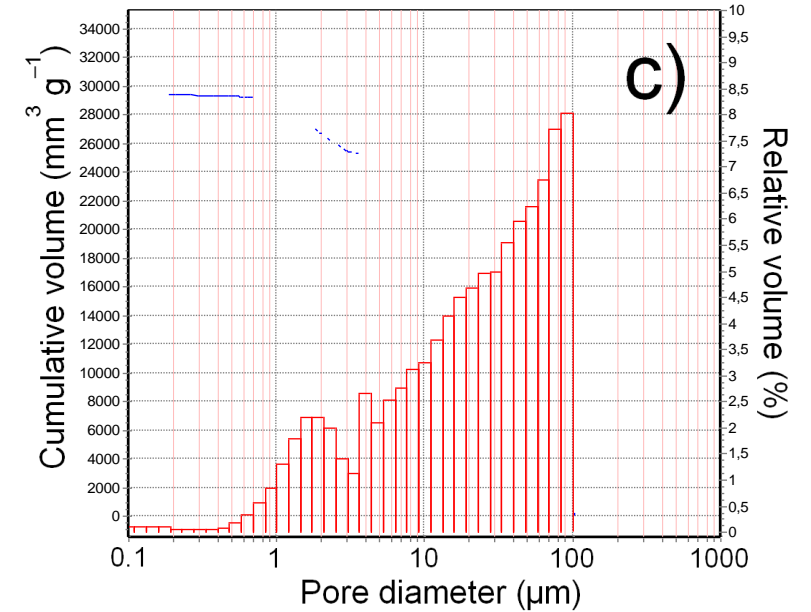
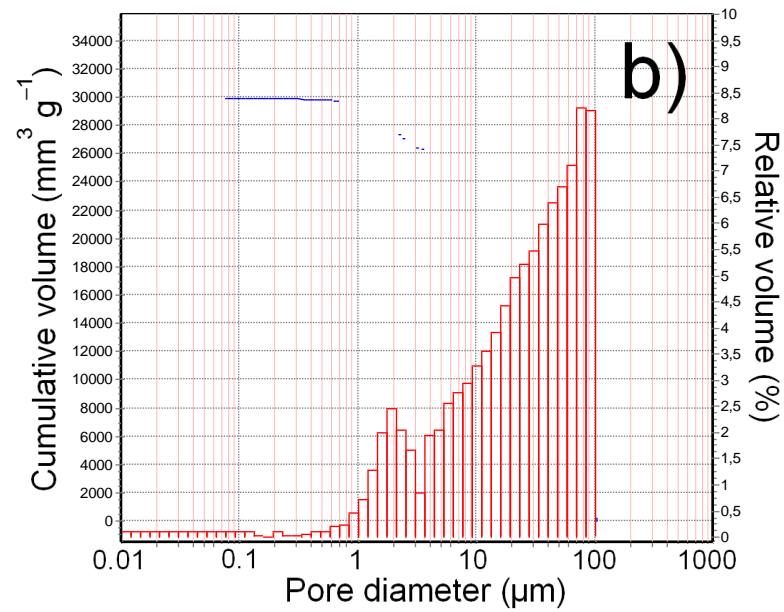
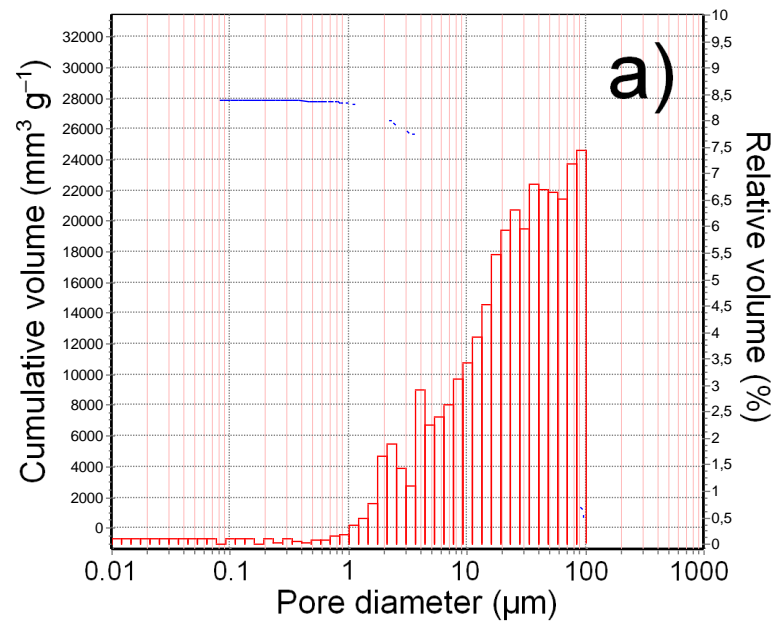


Figure 7.S



## SUPPLEMENTARY MATERIALS

## Legends to Figures

- Fig. 1.S** - Rinse water of EOLFG, EOLSG, EOLGG plugs loaded with polymyxin B (from Fig. 1.a) and checked for cell growth inhibition of *P. aeruginosa*. 100  $\mu$ l of (from left to right): 1<sup>st</sup>, 2<sup>nd</sup>, 3<sup>rd</sup> rinse water were added to each well.
- Fig. 2.S** - Rinse water of EOLFG, EOLSG, EOLGG plugs loaded with polymyxin B (from Fig. 1.b) and checked for cell growth inhibition of *S. marcescens*. 100  $\mu$ l of (from left to right): 1<sup>st</sup>, 2<sup>nd</sup>, 3<sup>rd</sup> rinse water were added to each well.
- Fig. 3.S**- Inhibition by polymyxin B released from aerogels of Gram-negative *Hafnia alvei* cell growth. Central well (1), 30  $\mu$ g (equivalent to 22 nanomoles) of free polymyxin B. Lateral wells: plugs of loaded EOLFG (2), control EOLFG (3); loaded EOLSG (4), control EOLSG (5); loaded EOLGG (6), control EOLGG (7).
- Fig. 4.S** - Rinse water of EOLFG, EOLSG, EOLGG plugs loaded with nisin (from Fig. 2.a) and checked for cell growth inhibition of *E. faecalis*. 100  $\mu$ l of (from left to right): 1<sup>st</sup>, 2<sup>nd</sup>, 3<sup>rd</sup> rinse water were added to each well.
- Fig. 5.S** - Inhibition of *C. tyrobutirycum* growth by lysozyme released from EOLFG aerogel: (1) 125  $\mu$ g (equivalent to 8.7 nanomoles) of free lysozyme. Plugs of: (2) loaded EOLFG, (3) control EOLFG.
- Fig. 6.S** - Time course of commercial *C. lipolitica* release in water solution from “loaded” hydrogel plugs of EOLFG, EOLSG and EOLGG. Each time point was performed in triplicate on separately prepared plugs and is expressed as the average percentage of the “theoretical” total lipase activity loaded.
- Fig. 7.S** - Pore size distribution plots of EOLFG (a), EOLSG (b), EOLGG (c)



**Tab. 1.S** – Time course of protease release in water from “loaded” hydrogel plugs of EOLFG, EOLSG and EOLGG. Experiments were performed in triplicate on independently prepared plugs. Average activity values and percentage of the ”theoretical” maximum activity loaded into the aerogels are reported for each time point, with standard deviations. Protease activity is expressed as  $\Delta$  O.D. at 440 nm/10 min/ml.

Time, h	EOLFG		EOLSG		EOLGG	
	Protease activity	Theoretical max, %	Protease activity	Theoretical max, %	Protease activity	Theoretical max, %
1	23.8±1.2	54.3±6.0	17.5±13.1	45.1±25.8	11.2±4.0	26.1±9.4
2	26.7±2.8	60.5±3.2	25.2±8.7	68.1±8.7	21.0±4.7	49.1±11.0
3	34.0±4.7	76.9±1.6	29.5±12.0	79.3±15.2	24.2±6.4	56.5±14.9
4	35.8±4.5	81.2±2.8	28.2±9.7	76.3±9.4	33.5±7.3	78.3±17.1
5	39.7±0.9	90.6±12.3	35.0±4.7	97.1±8.8	34.0±4.7	79.5±11.0
6	43.2±5.4	97.8±3.3	36.8±9.2	100.8±2.6	41.5±6.4	97.0±14.9



**Tab. 2.S** – Time course of *C. lipolitica* release in water from “loaded” hydrogel plugs of EOLFG, EOLSG and EOLGG. Experiments were performed in triplicate on independently prepared plugs. Average activity values and percentage of the ”theoretical” maximum activity loaded into the aerogels are reported for each time point, with standard deviations. Lipase activity is expressed as  $\Delta$  O.D. at 410 nm/min/ml.

Time, h	EOLFG		EOLSG		EOLGG	
	Lipolytic activity	Theoretical max, %	Lipolytic activity	Theoretical max, %	Lipolytic activity	Theoretical max, %
1	0.7±0.1	4.5±2.4	1.4±0.4	7.5±1.9	2.2±1.2	10.1±5.5
2	0.8±0.4	5.5±3.1	2.4±1.8	13.0±10.0	3.3±1.8	15.3±8.6
3	1.0±0.5	6.8±4.1	2.2±0.7	11.6±4.2	3.6±1.8	16.7±7.8
4	1.1±0.6	7.3±4.7	3.2±0.8	17.2±5.6	3.6±1.6	16.9±6.5
5	1.4±0.8	9.4±5.8	3.2±1.0	17.1±6.2	3.3±0.9	15.6±4.2
6	1.9±1.0	12.8±7.8	3.4±1.3	18.5±7.6	3.8±0.9	17.9±3.3
25	4.8±0.7	29.5±8.0	4.3±0.9	22.4±2.3	4.4±0.6	21.5±4.9
30	4.3±0.1	28.8±11.6	3.4±0.7	18.0±2.0	4.9±0.7	24.0±4.7