1	Plant-derived bioactive compounds at sub-lethal concentrations: towards smart biocide-free
2	antibiofilm strategies
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14 Abstract

15 Biofilm resistance to biocides is becoming a global issue with an impact on many fields, including health care, agriculture, the environment, society and industry. Plants offer a virtually inexhaustible 16 and sustainable resource of very interesting classes of biologically active, low-molecular-weight 17 compounds (parvome). In the past, the plant parvomes were screened mainly for their lethal effects, 18 disregarding concentrations and ecologically relevant functions of these molecules in the natural 19 context. Testing sub-lethal concentrations of plant-derived compounds mimicking environmental 20 levels may be critical to reveal mechanisms subtler than the killing activity, e.g. those influencing 21 the multicellular behavior, offering an elegant way to develop novel biocide-free antibiofilm 22 strategies. In a cross-disciplinary fashion, we illustrated recent successes of sub-lethal 23 concentrations of plant-derived compounds, their ecological insight, pro et contra, future directions 24 and impacts, envisioning implications for policy making and resource management. 25 26

27 Keywords: biofilm; plant-derived compounds; biocide-free strategies; sub-lethal concentrations

29 Introduction

30 It has been estimated that at least 99% of the world's microbial biomass exists in form of biofilm, a complex differentiated surface-associated community embedded in a self-produced polymeric 31 matrix enabling microorganisms to develop coordinated and efficient survival strategies. Although 32 the inclination to colonize surfaces is advantageous from the microbial standpoint, it may cause 33 chronic infections (Cegelski et al. 2008; Estrela et al. 2009), parasitism phenomena in animals and 34 plants (Skamnioti and Gurr 2009), biodeterioration of historical and artistic objects (Giacomucci et 35 al. 2011; Cappitelli et al. 2012), biodeterioration of engineered systems (Zhang et al. 2012), and 36 fouling in food-processing equipments (Ranier et al. 2011). Furthermore, biofilm injury has a 37 38 profound socio-economic impact, incurring direct and indirect industrial costs that result in a huge financial burden for an already over-stretched economy. 39 For human societies, the most detrimental property of biofilms is the expression of specific 40 41 characters that make sessile microorganisms more resistance to antimicrobial agents (up to 1000fold) than their planktonic counterparts (Høiby et al. 2010; Flemming 2011). As climate conditions 42 43 change, natural and engineered ecosystems are increasingly reaching temperatures and humidity that are conducive to biofilm growth. Although increased biofilm biomass would lead to an 44 increased use of biocides, questions concerning the biodegradability of biocides, their risk to human 45 46 and animal health and their environmental impact, have increasingly discouraged biocide use. This is readily seen in the number of recent policies, directives, technical reports, strategies, 47 recommendations and regulatory decisions designed to reduce antimicrobial agents consumption, 48 ensuring the prudent use of these fragile strategies, and protect specific agents that are critically 49 important for human and animal health and wellbeing (Directive 98/8/EC; Recommendation 50 2002/77/EC; SCENIHR report 2009; EFSA Summary Report 2012). Finally, the antimicrobial 51 arena is experiencing a shortage of lead compounds progressing into both clinical and industrial 52 trials and growing negative consumer perception against synthetic compounds has led to the search 53 for natural-derived products (Lam 2007). 54

In the last few years, the efforts have been directed towards developing preventive strategies that 55 can be used to disarm microorganisms without killing them (Cegelski et al. 2008; Rasko and 56 Sperandio 2010). An innovative approach is the use of biocide-free antibiofilm agents with novel 57 targets, unique modes of action and proprieties that are different from those of the currently used 58 antimicrobials. In addition, as these substances do not exert their action by killing cells, they do not 59 impose a selective pressure causing the development of resistance (Rasko and Sperandio 2010). 60 Observing the processes of biofilm formation it is reasonable to expect that interfering with the key-61 steps that orchestrate genesis of virtually every biofilm could be a way for new preventive strategies 62 that do not necessarily exert lethal effects on cells but rather sabotage their propensity for a sessile 63 64 lifestyle (Figure 1). For instance, interfering with the surface sensing process and mystifying intercellular signals, the biofilm cascade might be hampered. 65

These strategies might bring new products to the market and cover methodologies and novel
approaches, making significant contributions to innovation and economic productivity in SMEs.
They provide support for cross-cutting actions while offering new tools for society and policy
makers.

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71 Ecological insight of plant-derived antibiofilm compounds

The need for innovative antibiofilm technologies has led to renewed interest in the ways thatorganisms protect themselves against microbial colonization.

Plants lacking cell-based inducible immune responses and that live in nutrient-rich environments are continuously exposed to a broad array of potentially deleterious microorganisms leading to increased weight and friction, impeded trans-epidermal exchanges, altered color, smell, and contour (Wahl et al. 2012). This provides the driving force behind the evolution of a variety of sophisticated strategies to enhance plant fitness via chemical defenses against biofilms (de Nys and Steinberg 2002; Qian and Fusetani 2010). In addition, one of the main advantages of plant-derived compounds with potential pharmaceutical and medical applications is the lack of shared pathogens
between plant and mammals (Cichocka et al. 2010).

Both aquatic and terrestrial plants offer very interesting classes of biologically active, low-82 molecular-mass (< 5 kDa) compounds ("parvome", parv=small, -ome= group), like alkaloids, 83 terpenoids, flavonoids and coumarins, peptides, glycosides, nucleosides and polyphenols. They may 84 act in a variety of ways: antibiotics, allosteric regulators, catalysis, catalytic cofactors, regulatory 85 86 activities at level of DNA, RNA and protein, pigments, mutagens, antimutagens, receptor agonists, antagonists, signal molecules, siderophores, detergents, metal complexing/transporting agents, 87 pheromones, toxins and other interesting activities (Davies and Ryan 2012). However, during the 88 89 intensive half-century of drug discovery, available natural compounds found in the plant parvome were screened mainly for their lethal effects, disregarding concentrations and ecologically relevant 90 functions of these molecules in the natural environments. All that mattered were compounds 91 92 effective in killing target microorganisms (inter alias Gibbons 2005; Puglisi et al. 2007; Quave et al. 2008; Mayavu et al. 2009; Tajkarimi et al. 2010; Artini et al. 2012; Falcão et al. 2012; Guedes et 93 94 al. 2012). In contrast, few papers address the inhibition of biofilm formation by using compounds at sublethal concentrations. 95

In many cases, the killing activity of a naturally-occurring compounds is primarily a laboratory 96 property, since the concentrations of these agents available in nature would be insufficient to exert 97 their lethal effects (Yim et al. 2007; Davies 2011). Several studies on marine plants highlighted a 98 lack of correlation between antimicrobial activities and abundance of surface-associated 99 microorganisms, suggesting that chemical defenses may function by mechanisms more subtle than 100 101 the simple killing activities like those influencing the multicellular behavior by manipulating the expression of specific phenotypes that represent different stages of the biofilm process (Harder 102 2009). 103

The optimal defense theory asserts that organisms allocate resources to chemical defenses in a way
 that maximizes fitness and preserves their primary biological functions such as homeostasis

maintenance, growth and reproduction (Ivanisevic et al. 2011). The production of toxic compounds 106 107 might impose: i) a significant metabolic burden to the plant in order to protect itself from autotoxicity (Heil and Baldwin 2002) and ii) ecological costs resulting from the myriad of 108 109 interactions that a plant has with its biotic and abiotic environment (Heil and Baldwin 2002). In fact, it has been estimated that a considerable percentage of bacterial genomes is dedicated to 110 shaping the organisms' habitat and maintaining their community and niche in the ecosystem 111 112 (Phelan et al. 2012). Thus, killing microorganisms is not advantageous for the plant as might affect local ecological relationship. Finally, sub-lethal concentration represents one mechanism by which 113 the host minimizes the risk of counter adaptation, which would be likely to occur if secondary 114 115 metabolites were toxic to associated microbes (Engel et al. 2002). Testing sub-lethal concentrations of plant-derived compounds mimicking environmental levels may 116 be critical to understand biological functions, highlighting different and valuable biological 117 118 activities far from killing activities. As a consequence, one of the most pressing issues is the estimation of the sub-lethal concentrations of secondary metabolites experienced by 119 120 microorganisms in nature. In the context of antibiofilm researches, this gap may be filled carrying out preliminary experiments to define the toxicological threshold zone for the selected model 121 systems and then screening a wide range of sub-lethal concentrations at frequent intervals in order 122 123 to identify the experimental space with the maximum antibiofilm activity. However, the efforts of industrial, academic, governmental actors are made to reduce time and costs of research 124 programmes by testing few concentrations at standard conditions, demanding carefully designed 125 experiments to explore in details and at reasonable cost the low-dose response and the cellular 126 behavior in complex scenarios. 127

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129 Determination of optimal sub-lethal concentrations

130 The design of experiments technique (DOE) could be successfully employed to clarify the131 antibiofilm performance of plant-derived compounds without testing many sub-lethal

concentrations, but just performing a limited number of experiments according to rigorously
formulated mathematical protocols (Franceschini and Macchietto 2008). With this multivariate
approach it is possible to simulate cellular behavior in complex scenarios, considering effective
factors, interactions and selecting optimum conditions that maximized the antibiofilm response
(Leardi 2009).

Although DoE methods have been around since the mid-20th century, their application in the 137 discovery of non-toxic antibiofilm compounds has only recently taken hold. DoE has been shown to 138 perform excellently in a wide range of applications: chemical kinetics, process control, drug 139 discovery, biological systems (e.g. fermentation and bio-kinetics), pharmacodynamics, process 140 141 engineering etc. (inter alias Akhbari et al. 2011; Hu et al. 2012; Papaneophytou and Kontopidis 2012; Jibrail and Keat Teong 2013). However, to the best of our knowledge, only three works 142 (carried out by the authors of the present paper) successfully modeled the antibiofilm performances 143 144 of plant-derived compounds at sub-lethal concentrations exploiting High Throughput Screening techniques. 145

146 By using DoE coupled with microtiter biofilm assay Villa and colleagues (2011) observed that the best anti-biofilm performance of sub-lethal concentrations of the phenolic compound zosteric acid 147 (secondary metabolite from the seagrass Zostera marina, figure 2a) against Candida albicans was 148 obtained at a specific threshold level, which corresponds to the minimum point of the response 149 surface model and not to the maximum concentration tested (Figure 3). At this level, zosteric acid 150 played a role in thwarting budded-to-hyphal-form transition, in reducing biofilm biomass and 151 thickness, in extending the performance of antimicrobial agents and showed cytocompatibility 152 towards soft and hard tissue (Figure 4). The non-linear response patterns depicted by the surface 153 response followed a parabola-like shape profile, resembling a hormetic property (the situation in 154 which the response to an environmental stressor varies with the level of exposure). However, a 155 biphasic profile is not new in the biofilm world: the biofilm mediators homoserine lactones act in a 156

157 concentration-dependent manner, where upper and lower threshold concentrations trigger the158 formation of a biofilm (Rickard et al. 2007).

Escherichia coli cells treated with zosteric acid were characterized by stress-associated (e.g. AhpC, 159 OsmC, SodB, GroES, IscU, DnaK), motility-related (FliC), quorum-sensing-associated (LuxS) and 160 metabolism/biosynthesis-related (e.g. PptA, AroA, FabD, FabB, GapA) proteins. This indicated that 161 the antibiofilm compound targeted key steps involved in biofilm formation by modulating the 162 163 threshold level of the extracellular signalling molecule autoinducer-2 (AI-2) and inducing a hypermotile phenotype unable to firmly adhere on surfaces (Villa et al. 2012b). The compound 164 seems to act as an environmental stimulus or chemical manipulator that provides advance warning 165 about environmental changes, allowing the microorganisms to prepare for adversity while 166 conditions are still favorable. From an ecological perspective, the mechanism of action of the 167 zosteric acid seems to portray the "xenohormesis theory". According to the xenohormesis, 168 169 heterotrophs (animals and microbes) are able to sense chemical stimuli synthesized by autotrophs (like plants) in response to stress to mount a preemptive defense response that increases their 170 171 chance of survival (Howitz and Sinclair 2008). Interestingly, the synthesis of phenolic compounds is induced in plants by a variety of environmental stresses and the planktonic phenotype represents 172 a life-extending physiological trait to escape from adversity improving the colonization of new 173 174 favorable habitat. In a similar way, reacting to zosteric acid would allow the bacterial response to begin ahead of any direct damage or energy deficit, and, more importantly, would not stake the life 175 of both the plant and the microorganism respecting the ecological relationships and leading to an 176 extended lifespan of the involved counterparts. 177

Thus, exploring the effects of sub-lethal concentrations of plant-derived compounds on microbial behavior (e.g. adhesion, chemotaxis, swimming and swarming motility) has the potential not only to demonstrate interesting xenohormetic-like responses and the extent and the modality to which microbial surface colonization is chemically mediated, but also to unveil potent biocide-free antibiofilm mechanisms. 183

184	Recent successes of antibiofilm compounds from plants at sub-lethal concentrations
185	Vattem et al. (2007) have suggested that spices with renowned antibiotic properties could also
186	possess antipathogenic activities, which may not be related to lethal effects on the target
187	microorganism. The plant-derived compounds icariin and resveratrol, used in traditional Chinese
188	medicine, were found potent antibiofilm molecules against Propionibacterium acnes (Coenye et al.
189	2012). Importantly, the antibiofilm activity was detected at sub-inhibitory concentrations. Similarly,
190	extracts from Commiphora leptophloeos, Bauhinia acuruana and Pityrocarpa moniliformis
191	demonstrated marked Staphylococcus epidermidis antibiofilm activity on polystyrene and glass
192	surfaces without causing bacterial death (Trentin et al. 2011). The extract 220D-F2 from the root of
193	Rubus ulmifolius was used to inhibit S. aureus biofilm formation to a degree that can be correlated
194	with increased antibiotic susceptibility without limiting bacterial growth (Quave et al. 2012).
195	Ursolic acid from the tree <i>Diospyros dendo</i> (Figure 2b) is completely non-toxic towards <i>E. coli</i> , <i>P.</i>
196	aeruginosa, Vibrio harveyi, and successfully inhibited the formation of these bacterial biofilms.
197	Transcriptome analyses showed the induction of chemotaxis and motility genes in E. coli treated
198	with the plant-derived compound, suggesting that ursolic acid may function as a signal that tells
199	cells to remain too motile hindering cell adhesion or destabilizing already formed biofilm (Ren et al.
200	2005).
201	The methanolic extract obtained from Cuminum cyminum, a traditional food ingredient in South
202	Indian dishes, was shown to act as quorum-sensing inhibitor. By interfering with the acyl-
203	homoserine lactone activity, it inhibited the production of violacein pigment, swimming and
204	swarming motility, production of the extracellular polymeric substances and biofilm formation in
205	several bacterial pathogens (Issac Abraham et al. 2012). Also the extract of Capparis spinosa
206	showed a high degree of anti-quorum sensing activity in a dose dependent manner without affecting
207	the bacterial growth of Serratia marcescens, P. aeruginosa, E. coli and Proteus mirabilis. It also

exhibited inhibition in swimming and swarming motility of the bacterial pathogens (Issac Abraham 208

et al. 2011). Two synthetic furanones based on those produced by the marine macroalga Delisea 209 210 pulchra (Figure 2c) were shown to attenuate bacterial virulence in the mouse models of chronic lung infection by targeting *Pseudomonas aeruginosa* quorum-sensing without directly killing 211 bacteria, not imposing a selective pressure for the development of bacterial resistance (Wu et al. 212 2004). A number of flavonoids found in citrus species, including naringenin (Figure 2d), 213 kaempferol (Figure 2e), apigenin (Figure 2f) and quercetin (Figure 2g), which are antagonists of 214 215 homoserine lactone and AI-2-mediated cell-cell signaling in V. harveyi, were able to inhibit biofilm formation by V. harvevi BB120 and E. coli O157:H7 in a dose-dependent manner (Vikrame et al. 216 2010). 217

218 Recently, members of the Transient Receptor Potential (TRP) channels have drawn large attention as versatile sensors to detect changes in the external environment being associated to sensation of 219 heat, cold, noxious chemicals, pain, osmotic force, touch, vibration, proprioception and axon 220 221 guidance (Vriens et al. 2008) in various animals and in man. Interestingly, fungal genomes present genes encoding a TRP-like structure. The mechanosensitive TRP channel in Saccharomyces 222 223 cerevesiae (Yvc1=TRPY1) has orthologs in other fungal genomes including TRPY2 of Kluyveromyces lactis and TRPY3 of C. albicans (Chang et al. 2010). Since several plant-derived 224 taste-active substances are able to modulate/interact with these sensing channels, they are 225 226 interesting bioactive molecules with new potential targets for the development of non-toxic strategies against biofilms. According to this chemosensory-based strategy, the efficacy of sub-227 lethal concentrations of Muscari comosum bulb extract in modulating yeast adhesion and 228 subsequent biofilm development on abiotic surfaces and its role as extracellular signal responsible 229 for biofilm dispersion was reported (Villa et al. 2012a) (Figure 1). 230

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232 Drawbacks in the advancement of plant-derived products production

Main reasons for the fact that plant-derived products research has not yet advanced to great lengthsin the last 20 years include the incompatibility of natural product libraries with high-throughput

screening, the marginal improvement in core technologies for natural product screening and natural
product structure elucidation (Lam 2007). In addition, chemists have been sometimes frustrated by
their inability to resolve complex mixtures at reasonable cost. However, an advantage of using
mixture is that effects may be additive and synergistic, through their ability to affect multiple targets
(Kirakosyan and Kaufman 2009), a smart strategy when dealing with the complex phenomenon
such as biofilm formation in which different pathways are involved.

241 Recently, the development of new methodologies has revolutionized the screening of natural products: bio-prospecting, development of a streamlined screening process, improved natural 242 product sourcing, advances in chemical methodologies, combinatorial biosynthesis and plant 243 244 genomics (Lam 2007; Bohlin et al. 2010). For instance, rapid and more cost-effective genome sequencing technologies coupled with advanced computational power permits extracting chemical 245 knowledge from genetic information more efficiently (Li et al. 2009). Less expensive DNA 246 247 sequencing allows the identification of gene clusters known to be associated with a production of small molecules. In addition to identify new natural products, genome mining may certainly have an 248 249 impact on the understanding the production of natural products (Clardy and Walsh 2004; Lam 2007). 250

When research leads to the commercialization of an agent, large quantities of the compound are 251 required. The preferred option is synthesis of the compound. Combinatorial chemistry approaches 252 are being applied based on phytochemical scaffolds to create screening libraries that closely 253 resemble antibiofilm-like compounds. In silico techniques like quantitative structure-activity 254 relationships (QSAR) analysis, pioneered by Hansch et al. (1962), helps to quantitatively correlate 255 the activity or properties of compounds with their measured or computed physiochemical 256 properties, playing crucial and rate accelerating steps for the better drug design in the modern era 257 (Lill 2007; Verma et al. 2010; Kar and Roy 2012; Yao 2012). QSAR approaches have been 258 developed and have demonstrated appealing advantages, including their low-cost and capability to 259 scale up easily (Yao 2012). The main assumption in the QSAR approaches is that the all properties 260

viz. physical, chemical and biological are purely depending on the molecular structure. QSAR is an 261 262 attempt to remove the element of luck from drug design by establishing a mathematical relationship in the form of an equation between biological activity and measurable/computed physicochemical 263 parameters. These equations may be used by the chemist to make a more informed choice as to 264 which analogues to prepare. Currently, QSAR approach has been successfully applied to many data 265 sets of plant-derived compounds (Wright et al. 2006; Chen and Li 2009; Nargotra et al. 2009; De-266 Eknamkul et al. 2011; Yao et al. 2011). Thus, by applying the QSAR technique, new organic 267 synthetic methodologies and biotransformation for the modification of natural product leads would 268 generate a novel, structurally diverse analogs with improved properties or new activities (Zhou et al. 269 2012). 270

However, owing to their structural complexity, some natural products are not currently produced on 271 an industrial scale by chemical synthesis. Thus, another drawback lies in the sustainability of the 272 273 use and management of plant resources, insuring that the population size and the availability of the extracted product do not decline as a result of harvesting (Gilliland et al. 2009). A solution is 274 275 represented by microbial hosts engineered to express plant metabolic pathways as reported by Ajikumar et al. (2010) and the developing of a platform technology to isolate and culture cambial 276 meristematic cells (CMCs, multipotent plant cells that give rise to the vascular tissues xylem and 277 phloem) in the laboratory and then harvesting the desired products from the media in which they 278 grow (Lee et al. 2010). Finally, tailoring efficient laboratory plant-systems to produce specific 279 compounds can be an efficient and sustainable source of plant-derived products. 280

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282 Concluding remarks

Plants represent a virtually inexhaustible and sustainable resource of biocide-free antibiofilm agents
with novel targets, unique modes of action and proprieties with potential for utilization in a plethora
of medical, agricultural, and industrial fields. On the one hand, realization of this possibility has so

far been hindered by insufficient fundamental research to comprehensively understand the 286 287 ecologically relevant functions of plant-derived compounds in the real natural environments. When testing the biocidal action of a naturally-occurring agent against biofilm-forming 288 microorganisms, we should keep in mind that this might not be the modality whereby this molecule 289 works in nature. The concept that the killing activity is not the only property of a compound can be 290 traced back to the 16th century when the Swiss chemist and physician Paracelsus wrote: "All things 291 are poison and nothing is without poison, only the dose permits something not to be poisonous". 292 Now the question is: what happens at sub-lethal concentrations? 293

This is a common failure of many studies in which the investigator is unaware of the microbial behavior at sub-inhibitory concentrations. Thus, it is possible that the use of plant-derived compounds as less toxic or non-toxic antibiofilm products has been neglected or even abandoned principally because the optimal sub-lethal concentrations and working conditions were not found and not because the agent was ineffective. This holistic approach provides risk managers and decision-makers with the evidence they need to prioritize their resources and efforts to develop new technologies to deal with the spread and recalcitrance of unwanted biofilms.

Sub-inhibitory concentrations of plant-derived compounds might offer an elegant way to interfere with specific key-steps that orchestrate biofilm formation, mitigating biofilm formation without affecting their existence, sidestepping drug resistance and extending the efficacy of the current arsenal of antimicrobial agents. This technology might pave the way to more innovative, resource efficient and competitive society that reconciles human wellbeing with the sustainable use of renewable resources for industrial purposes, while ensuring environmental protection.

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Figure 1: The biofilm life cycle in three main steps (1- reversible and irreversible attachment; 2maturation; 3- detachment) and action of some plant-derived bioactive compounds at sub-lethal
concentrations.

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Figure 2: Plant-derived compounds with antibiofilm activities at sub-lethal concentrations: (a)
zosteric acid, (b) ursolic acid, (c) synthetic furanones based on those produced by *Delisea pulchra*,
(d) naringenin, (e) kaempferol, (f) apigenin and (g) quercetin.

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Figure 3: Three-D response surface model displaying the hormetic properties of zosteric acid, a 512 secondary metabolite of the seagrass Zostera marina tested against Candida albicans biofilm. Plot 513 shows interaction between zosteric acid and pH when time and temperature were 12 hours and 25 514 °C respectively. The variables were coded in the range -1 (minimum selected value) and +1 515 516 (maximum selected value). Ranges in the legends represent the number of adhered cells. The graph shows that the best anti-biofilm performance of the plant-derived compound was obtained at a 517 specific threshold level, which corresponds to the minimum point of the response surface model. 518 519 Thus, the minimum number of adhered cells does not correspond to the high amount of zosteric acid. Minimum adhesion (that is the maximum response) corresponds to 10 mg/l of zosteric acid. 520 The maximum response is predicted to be a reduction of fungal spores adhesion by 70%. 521

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Figure 4: View of 3D reconstruction images of *Candida albicans* biofilm grown without (a) and
with sublethal dose of zosteric acid (b). Zosteric acid induces morphostructural alterations,
thwarting budded-to-hyphal-form transition. Biofilms were stained with FUN-1 yeast viability stain
(red-orange), indicating that zosteric acid treatment maintains metabolically active cells. Biofilm
samples were visualized using a Leica TCS-SP2 AOBS confocal laser scanning microscope with

- excitation at 488 nm, and emission \geq 530 nm (green and red channels). Images were captured with a
- 529 63X 0.9 NA w water immersion objective and analyzed with the software Imaris (Bitplane
- 530 Scientific Software, Zurich, Switzerland).
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